

Supporting information

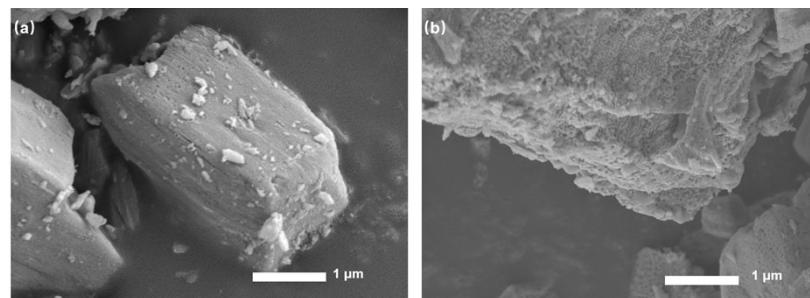


Fig. S1 (a) Intermediate products of Mn-BDC and NH₄VO₃ annealed at 200°C. (b) SEM image of MVO@C .

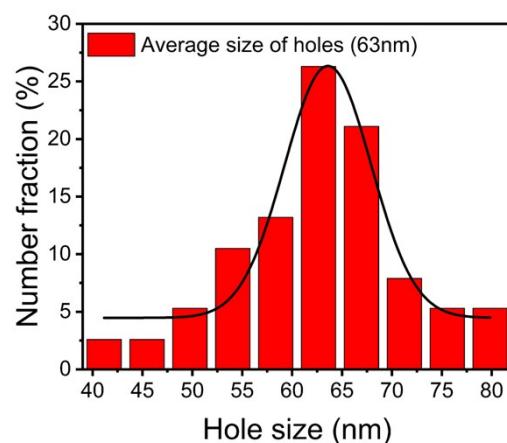


Fig. S2 Pore size distribution on MVO@C layered nanoplates.

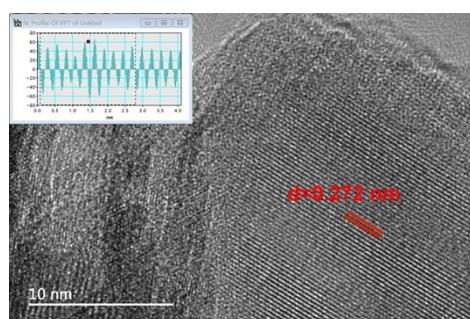


Fig. S3 The HRTEM image of MVO@C.

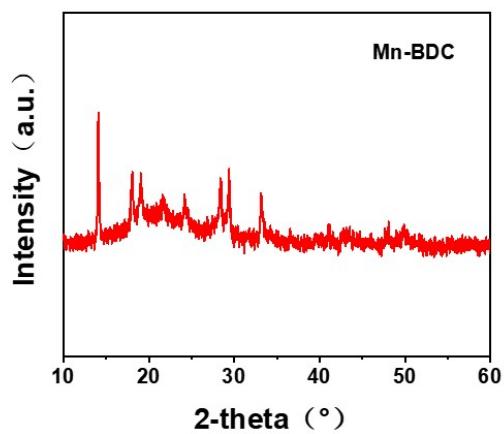


Fig. S4 The XRD pattern of Mn-BDC.

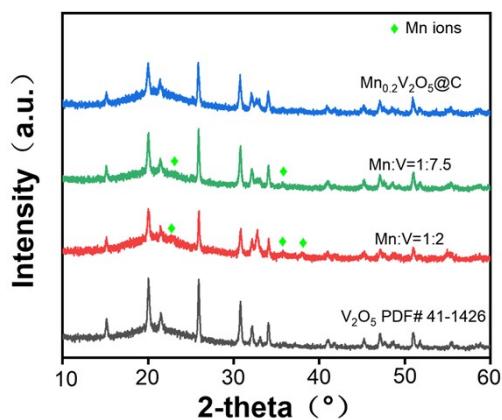


Fig. S5 XRD patterns of MVO@C cathode at different V and Mn ratios.

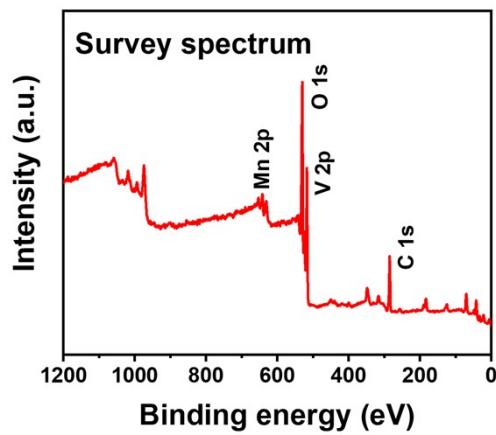


Fig. S6 Total XPS spectrum of MVO@C.

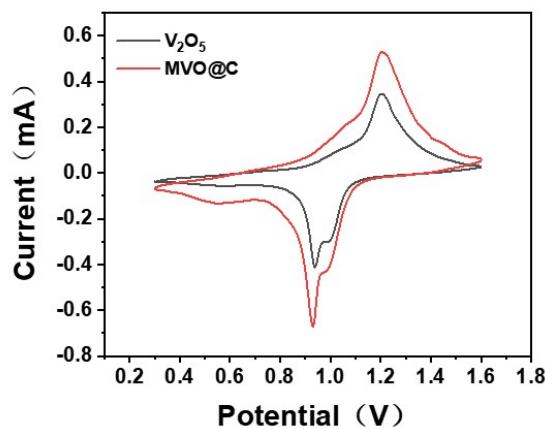


Fig. S7 CV comparison of MVO@C and V_2O_5 at 0.1 mV s^{-1} scan rate.

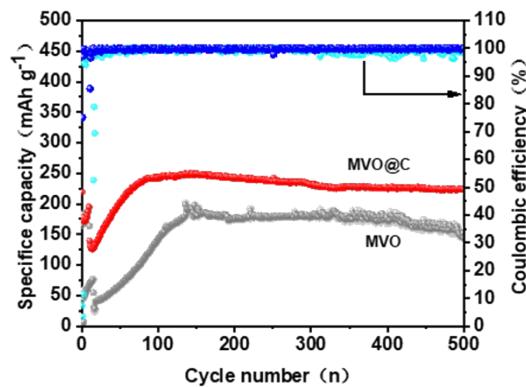


Fig. S8 Cycle performance of MVO@C and MVO at 4 A g^{-1} .

Table S1. $R_s (\Omega)$ and $R_{ct} (\Omega)$ values on electrochemical impedance spectroscopy of MVO@C.

Cycle number	MVO@C $R_s (\Omega)$	MVO@C $R_{ct} (\Omega)$
0	4.962	432.6
50	2.651	59.62
250	1.24	13.9

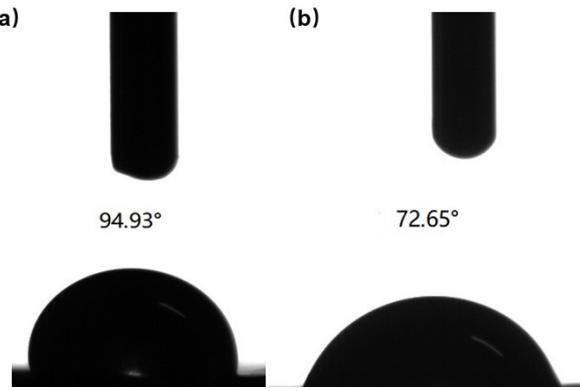


Fig. S9 (a) Instantaneous contact angle of V_2O_5 cathode, and (b) Instantaneous contact angle of $\text{MVO}@\text{C}$ cathode.

Table S2 Electrode performances of reported vanadium-based materials in AZIBs.

Cathode	Electrolyte	Maximum capacity	Cycle performance	Ref.
$\text{MgxV}_2\text{O}_5 \cdot \text{nH}_2\text{O}$	3 M $\text{Zn}(\text{CF}_3\text{SO}_3)_2$	350 mAh g ⁻¹ at 0.1 A g ⁻¹	90 mAh g ⁻¹ at 5 A g ⁻¹ after 2000 cycles.	1
$\text{C}@\text{VO}_2 @\text{V}_2\text{O}_5$	3 M $\text{Zn}(\text{CF}_3\text{SO}_3)_2$	376 mAh g ⁻¹ at 0.05 A g ⁻¹	160 mAh g ⁻¹ at 5 A g ⁻¹ after 2000 cycles	2
$\text{Cu}_{0.26}\text{V}_2\text{O}_5 @\text{C}$	3 M $\text{Zn}(\text{CF}_3\text{SO}_3)_2$	328.8 mAh g ⁻¹ at 0.2 A g ⁻¹	173.5 mAh g ⁻¹ at 2 A g ⁻¹ after 500 cycles	3
ZnVOH/rGO	3 M $\text{Zn}(\text{CF}_3\text{SO}_3)_2$	306 mAh g ⁻¹ at 0.1 A g ⁻¹	210 mAh g ⁻¹ at 0.1 A g ⁻¹ after 100 cycles	4
$\text{a-V}_2\text{O}_5$	3 M $\text{Zn}(\text{CF}_3\text{SO}_3)_2$	348 mAh g ⁻¹ at 0.1 A g ⁻¹	\sim 80 mAh g ⁻¹ at 5 A g ⁻¹ after 2000 cycles	5
$\text{P-V}_2\text{O}_5$	3 M $\text{Zn}(\text{CF}_3\text{SO}_3)_2$	322 mAh g ⁻¹ at 0.1 A g ⁻¹	80 mAh g ⁻¹ at 2 A g ⁻¹ after 2000 cycles	6
$\text{V}_2\text{O}_5 @\text{Graphene}$	3 M $\text{Zn}(\text{CF}_3\text{SO}_3)_2$	\sim 375 mAh g ⁻¹ at 0.1 A g ⁻¹	305 mAh g ⁻¹ at 0.1 A g ⁻¹ after 100 cycles	7
V_2O_5 nanofiber	3 M $\text{Zn}(\text{CF}_3\text{SO}_3)_2$	226 mAh g ⁻¹ at 0.294 A g ⁻¹	166 mAh g ⁻¹ at 0.588 A g ⁻¹ after 500 cycles	8
$\text{Ag}_{0.33}\text{V}_2\text{O}_5$	2 M $\text{Zn}(\text{CF}_3\text{SO}_3)_2$	200 mAh g ⁻¹ at 0.2 A g ⁻¹	\sim 75 mAh g ⁻¹ at 3 A g ⁻¹ after 700 cycles	9

$\text{Mn}_{0.18}$	3 M	380 mAh g^{-1} at 0.1 A g^{-1}	161 mAh g^{-1} at 6 A g^{-1}	This work
$\text{V}_2\text{O}_5@\text{C}$	$\text{Zn}(\text{CF}_3\text{SO}_3)_2$		after 2000 cycles	

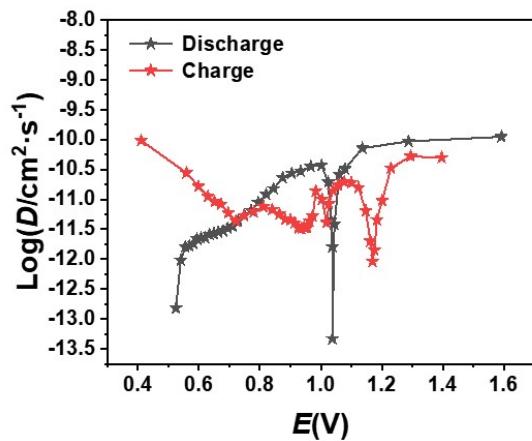


Fig. S10 The GITT curves and the corresponding Zn^{2+} diffusion coefficient of the V_2O_5 electrode.

References:

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