

Electronic Supplementary Information (ESI)

Immobilization of Ferrocene and Its Derivatives within Metal-Organic Frameworks with High Loadings toward Efficient Oxygen Evolution Reaction

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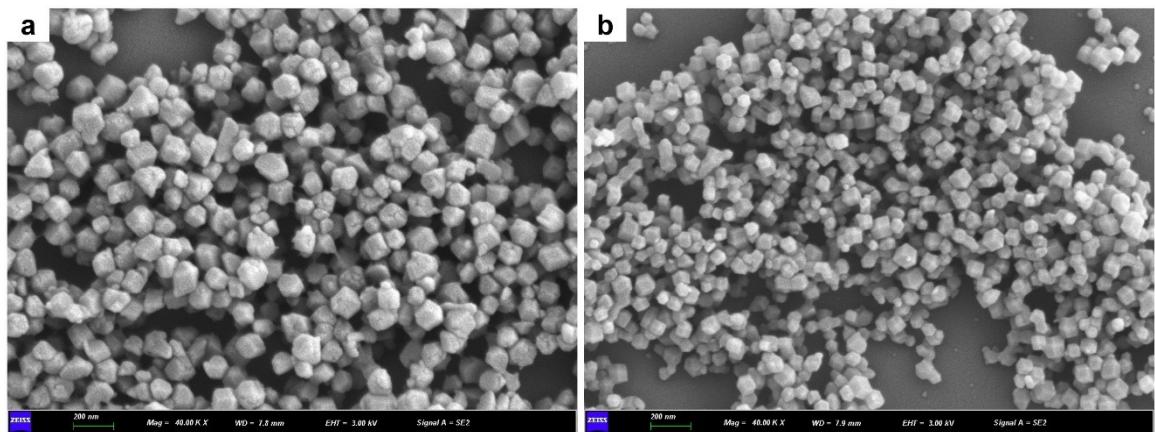


Fig. S1 SEM images of ZIF-8 (a) and Fc/ZIF-8-2:1 (b).

Table S1 Pore structure of ZIF-8 and Fc/ZIF-8-2:1.

Sample	S _{BET} (m ² g ⁻¹) ^a	S _{micro} (m ² g ⁻¹) ^b	V _{total} (cm ³ g ⁻¹) ^c	V _{micro} (cm ³ g ⁻¹) ^d
ZIF-8	1785.5	1672.2	1.75	0.62
Fc/ZIF-8-2:1	938.9	770.7	1.09	0.38

^a Brunauer-Emmett-Teller (BET) surface area estimated at $P/P_0 = 0.005\text{-}0.05$, ^b microporous surface area determined by *t*-plot method at $P/P_0 = 0.4\text{-}0.6$, ^c total pore volume calculated at $P/P_0 = 0.995$, ^d micropore volume determined by *t*-plot method.

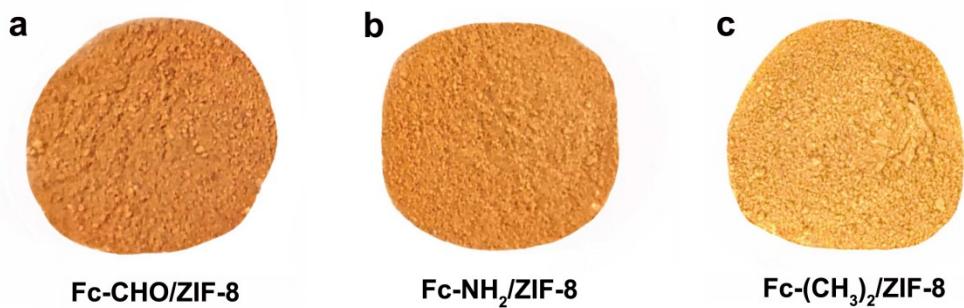


Fig. S2 Optical photos of Fc-CHO/ZIF-8 (a), Fc-NH₂/ZIF-8 (b) and Fc-(CH₃)₂/ZIF-8 (c).

Table S2 Content of Fe determined by ICP-AES in Fc-CHO/ZIF-8, Fc-NH₂/ZIF-8 and Fc-(CH₃)₂/ZIF-8 composites.

Sample	Fe (%)
Fc-CHO/ZIF-8	3.6%
Fc-NH ₂ /ZIF-8	3.5%
Fc-(CH ₃) ₂ /ZIF-8	3.9%

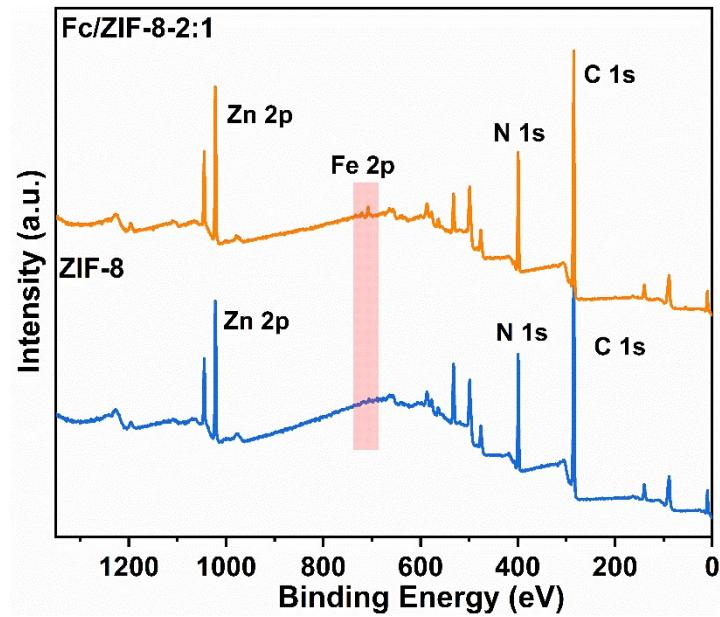


Fig. S3 XPS survey spectra of Fc/ZIF-8-2:1 and ZIF-8.

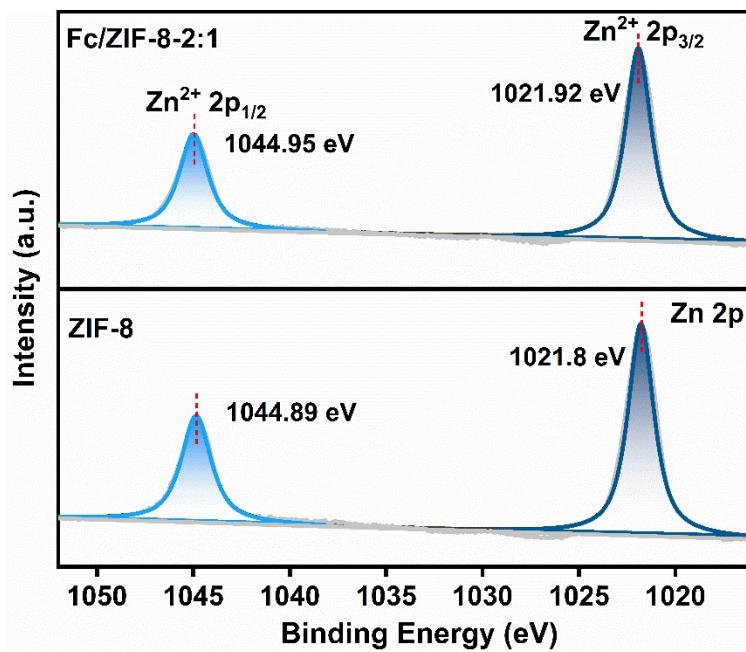


Fig. S4 High-resolution XPS spectra of Zn 2p in pristine Fc/ZIF-8-2:1 and ZIF-8.

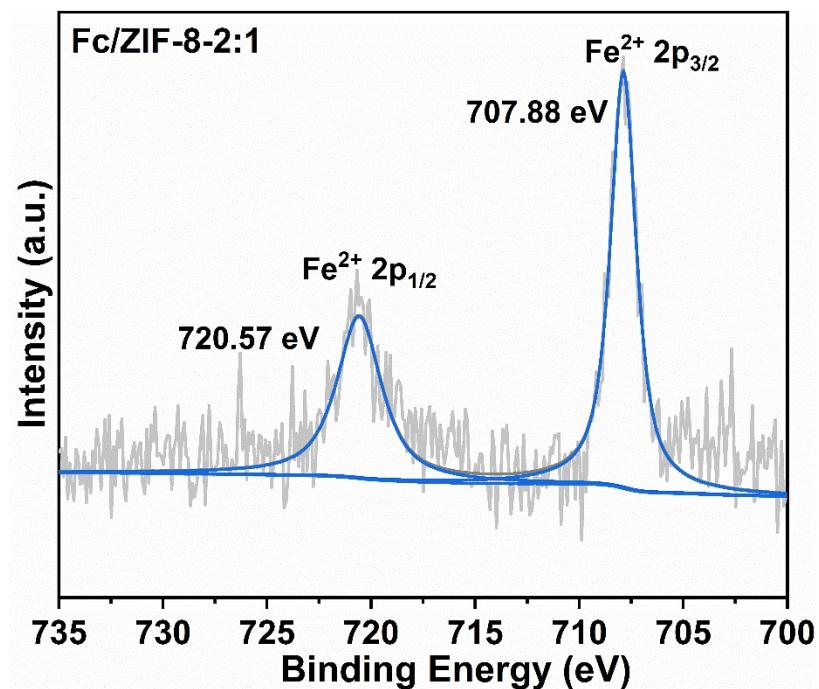


Fig. S5 High-resolution XPS spectra of Fe 2p in pristine Fc/ZIF-8-2:1.

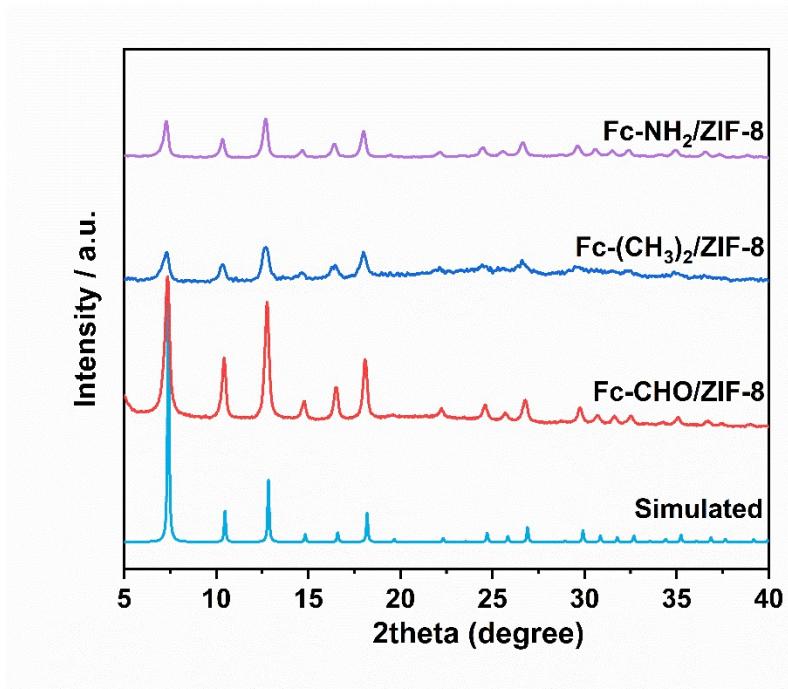


Fig. S6 XRD patterns of simulated ZIF-8, Fc-CHO/ZIF-8, Fc-(CH₃)₂/ZIF-8 and Fc-NH₂/ZIF-8.

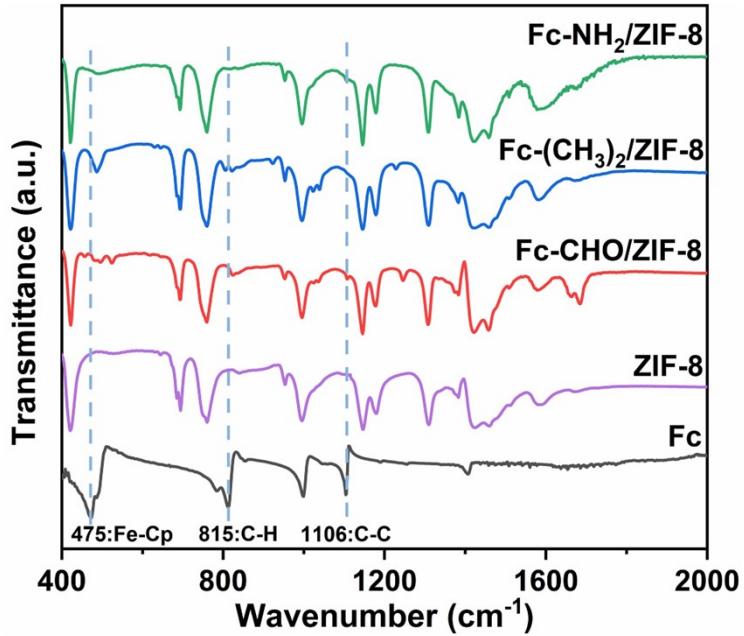


Fig. S7 FT-IR spectra of Fc, ZIF-8, Fc-CHO/ZIF-8, Fc-(CH₃)₂/ZIF-8 and Fc-NH₂/ZIF-8.

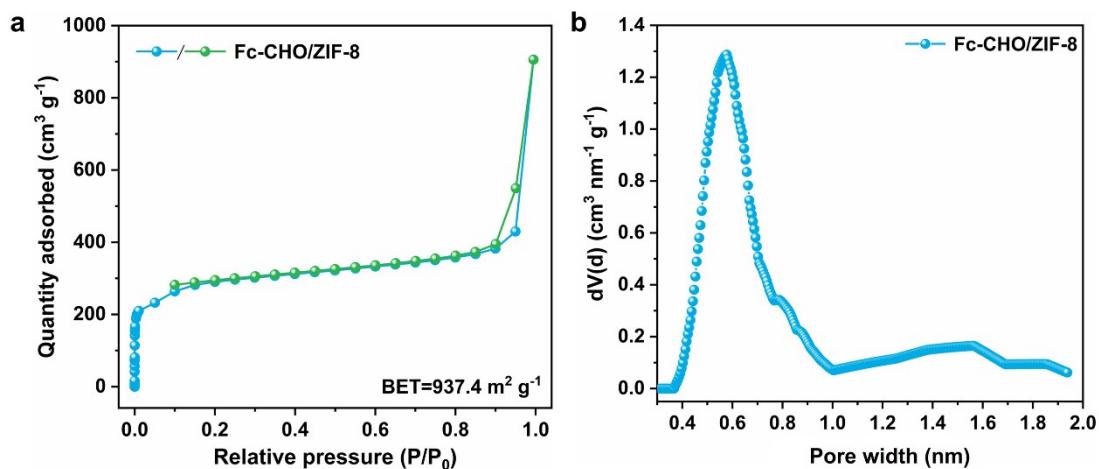


Fig. S8 N₂ adsorption-desorption isotherms (a) and the pore size distribution using the Horvath-Kawazoe model (b) of Fc-CHO/ZIF-8.

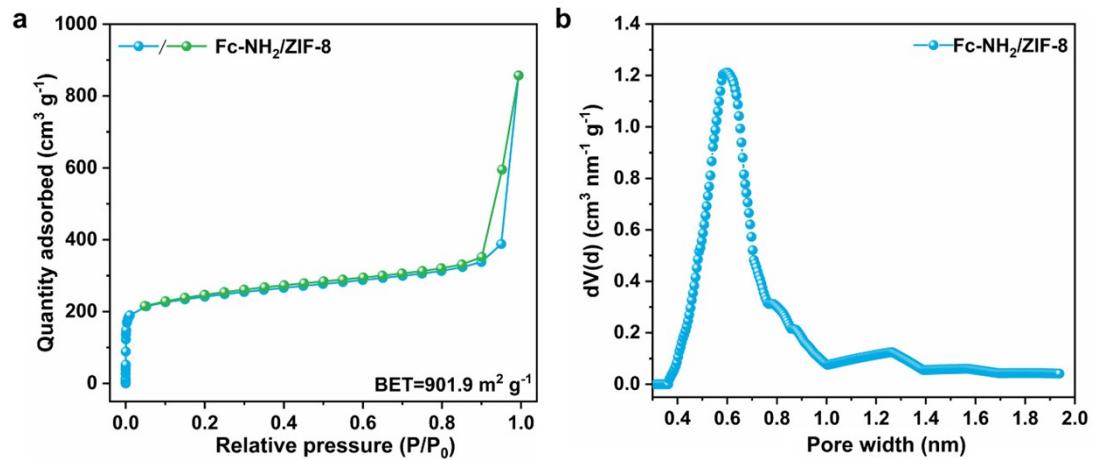


Fig. S9 N₂ adsorption-desorption isotherms (a) and the pore size distribution using the Horvath-Kawazoe model (b) of Fc-NH₂/ZIF-8.

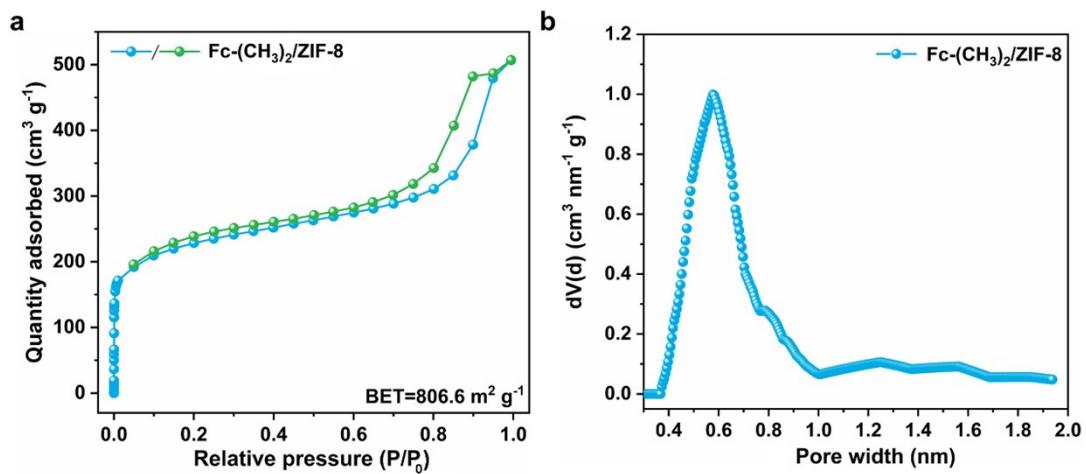


Fig. S10 N₂ adsorption-desorption isotherms (a) and the pore size distribution using the Horvath-Kawazoe model (b) of Fc-(CH₃)₂/ZIF-8.

Table S3 OER performances for this work and reported Fe/Co/Ni-based OER catalysts under alkaline conditions.

	<i>Overpotential</i> (mV) at 10 mA cm ⁻²	<i>Tafel slope</i> (mV dec ⁻¹)	<i>References</i>
Fc/ZIF-8	246	52	This Work
Fc-CHO/ZIF-8	238	44.4	This Work
BMM-10-Fe-H	260	137.4	1
NiFe-MOF NSs	240	73.44	2
G-FeNi-Co-ZIF-L/NF	248	49.5	3
Fe-Ni MOF NSs/NF	258	40.8	4
Co₃(HITP)	254	86.5	5
NiCoFeP	273	35	6
CoFe-MOF-NK	268	77.5	7
MCCF/NiMn-MOFs	280	86	8
CoNi1@C	276	55.6	9
CoCu-MOF	271	63.5	10
Ni-Co LDH@ZIF-67-V_o/NF	290	58	11
CoFeMOF	265	44	12
I³O⁰-type Co-MOF	361	28	13
Ni@NC	280	46	14
NiFc-MOF	195	44.1	15
FeTAPP-NiTCPP-POP	338	52	16
Co porphyrins 1/CNTs	407	60.3	17

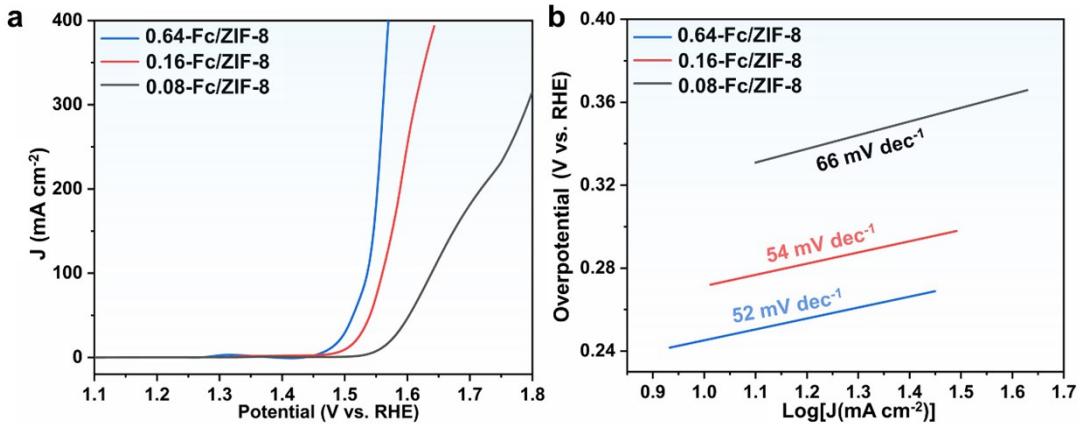


Fig. S11 LSV curves (a) and corresponding Tafel slopes (b) of Fc/ZIF-8 with different Fc loading.

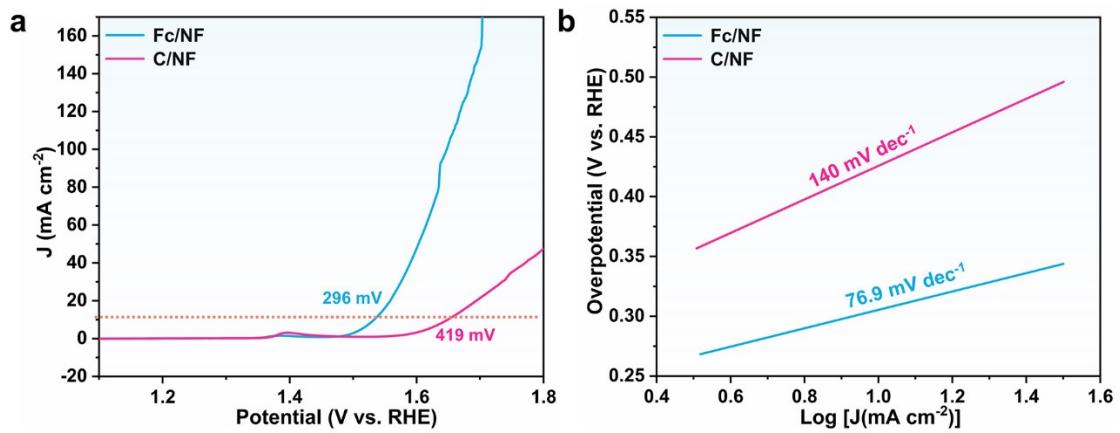


Fig. S12 LSV curves (a) and corresponding Tafel slopes (b) of Fc/NF and C/NF.

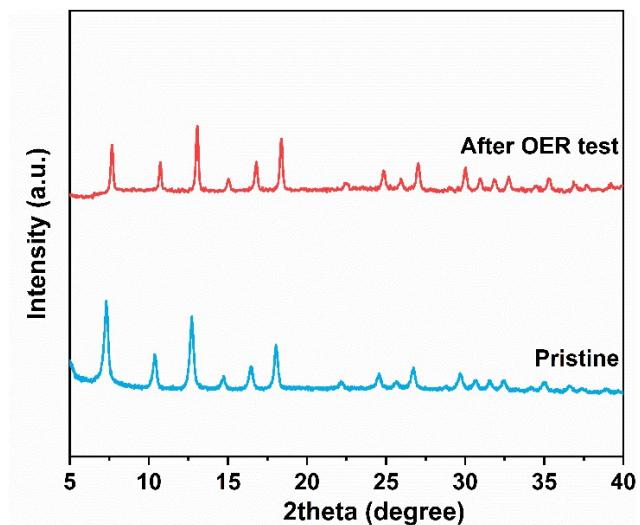


Fig. S13 XRD patterns of of Fc/ZIF-8 before and after OER test.

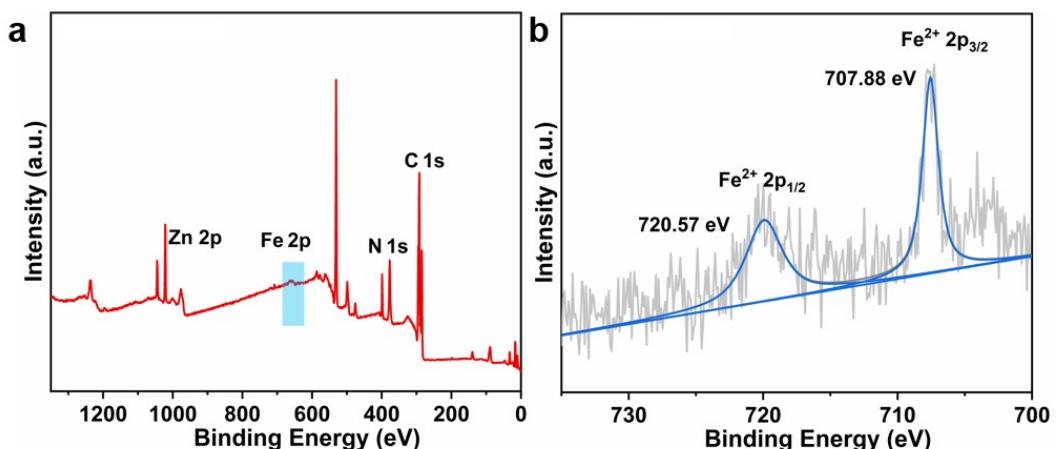


Fig. S14 XPS survey spectra (a) and Fe 2p (b) of Fc/ZIF-8-2:1 after OER test.

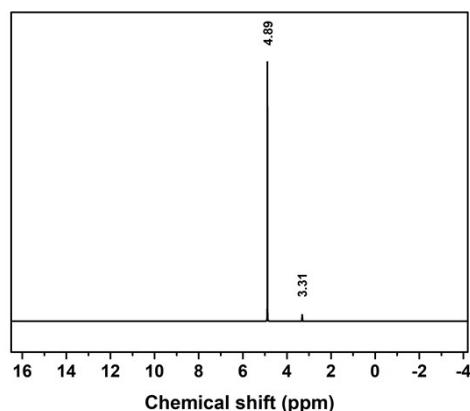


Fig. S15 ¹H NMR spectra of the electrolyte solution in methanol-d₄ reagent.

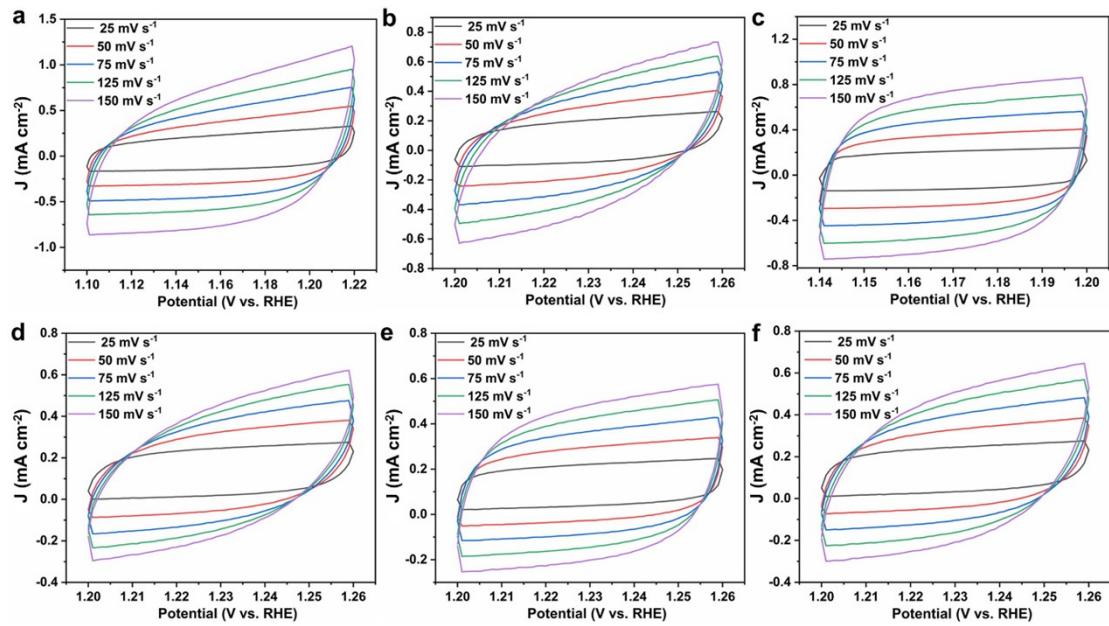


Fig. S16 CV curves measured at different scan rates of Fc/ZIF-8-0:1 (a), Fc/ZIF-8-1:1 (b), Fc/ZIF-8-2:1 (c), Fc/ZIF-8-3:1 (d), Fc/ZIF-8-4:1 (e) and ZIF-8 (f).

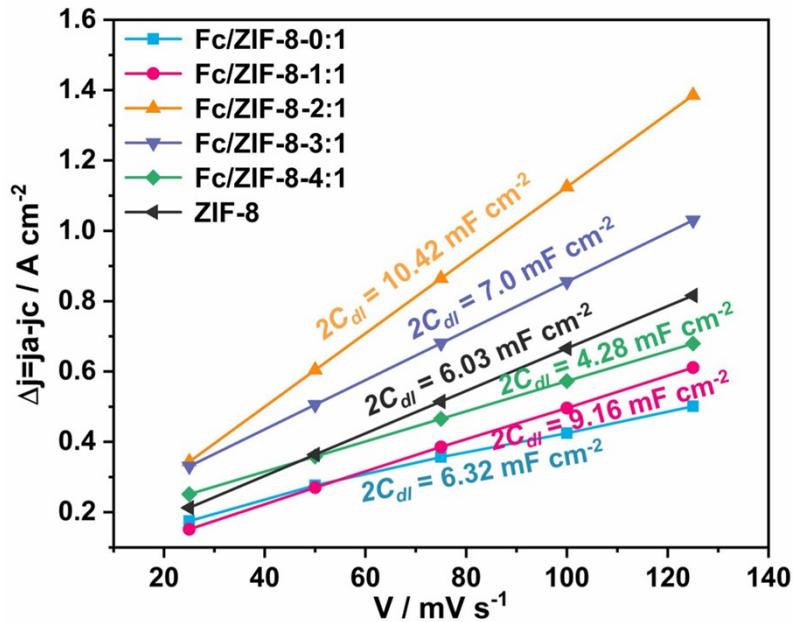


Fig. S17 C_{dl} values of Fc/ZIF-8-0:1, Fc/ZIF-8-1:1, Fc/ZIF-8-2:1, Fc/ZIF-8-3:1, Fc/ZIF-8-4:1 and ZIF-8.

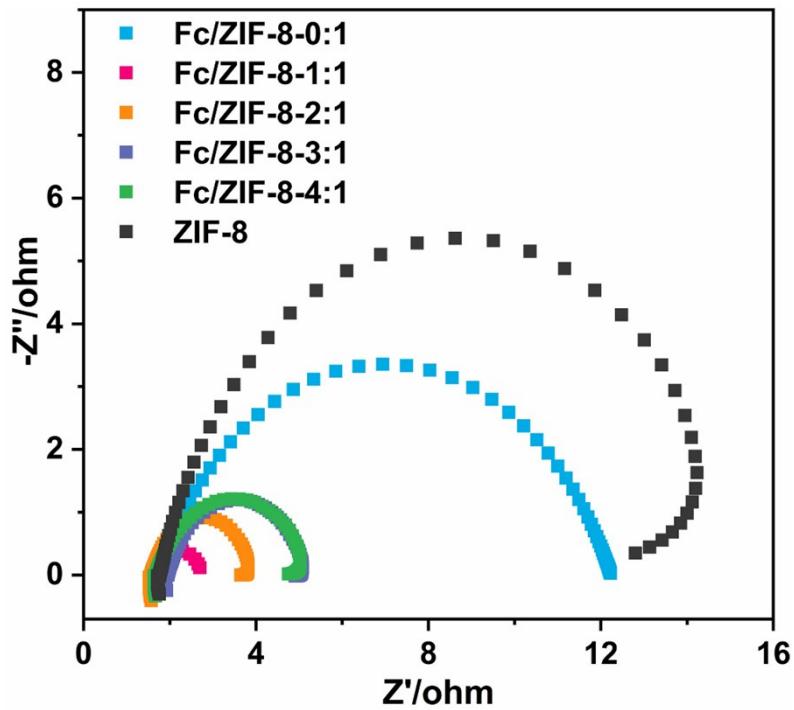


Fig. S18 Nyquist plots of Fc/ZIF-8-0:1, Fc/ZIF-8-1:1, Fc/ZIF-8-2:1, Fc/ZIF-8-3:1, Fc/ZIF-8-4:1 and ZIF-8.

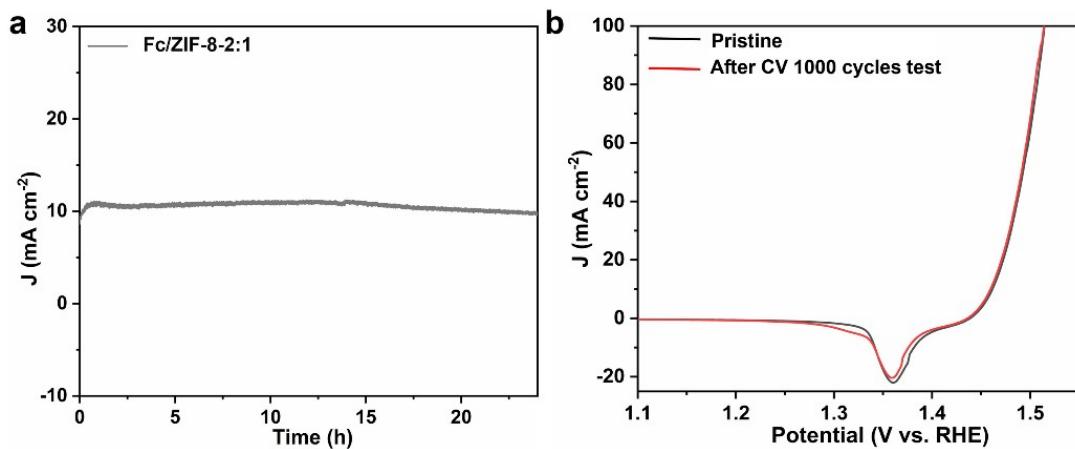


Fig. S19 The chronoamperometry curve at 10 mA cm $^{-2}$ of Fc/ZIF-8-2:1 (a), and LSV curves of Fc/ZIF-8-2:1 before and after 1000 CV cycles (b).

Supplementary References

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