

## Supporting Information

# ***In-Situ* Growth of NiCo-MOF and the Derived NiCo<sub>2</sub>O<sub>4</sub>/NiCo<sub>2</sub>O<sub>4</sub>/Ni foam Composite with Wire-Penetrated- Cage Hierarchical Architectures for Efficient Oxygen Evolution Reaction**

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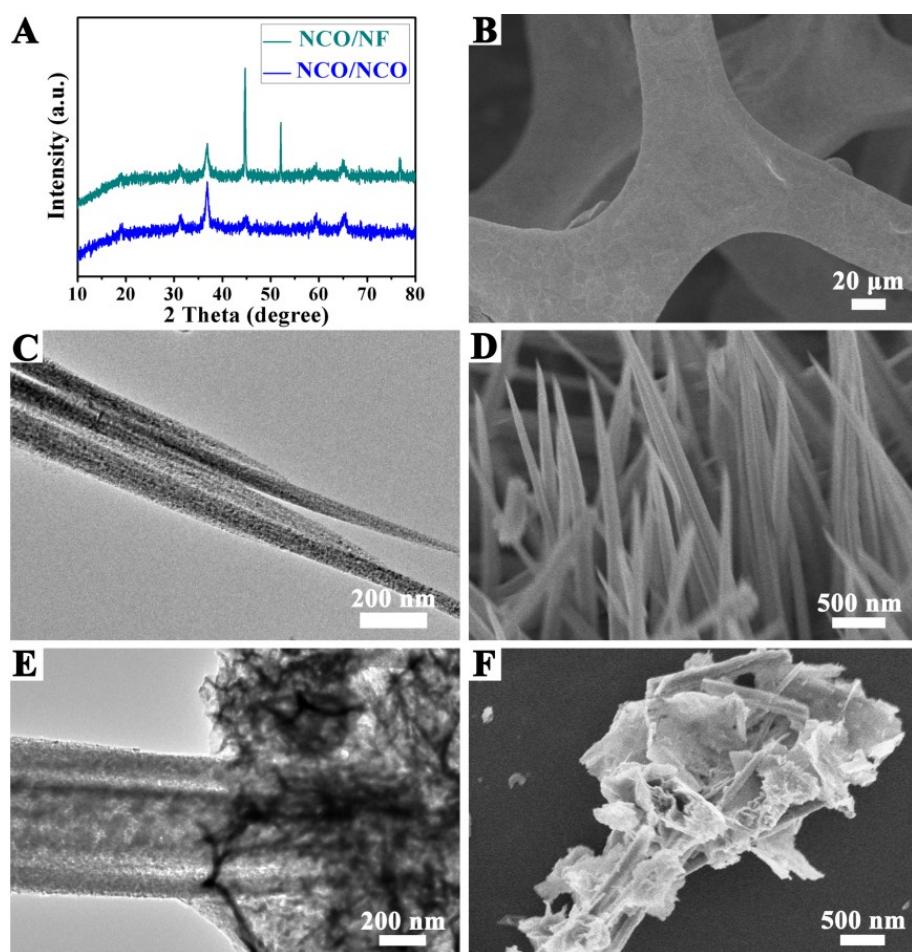
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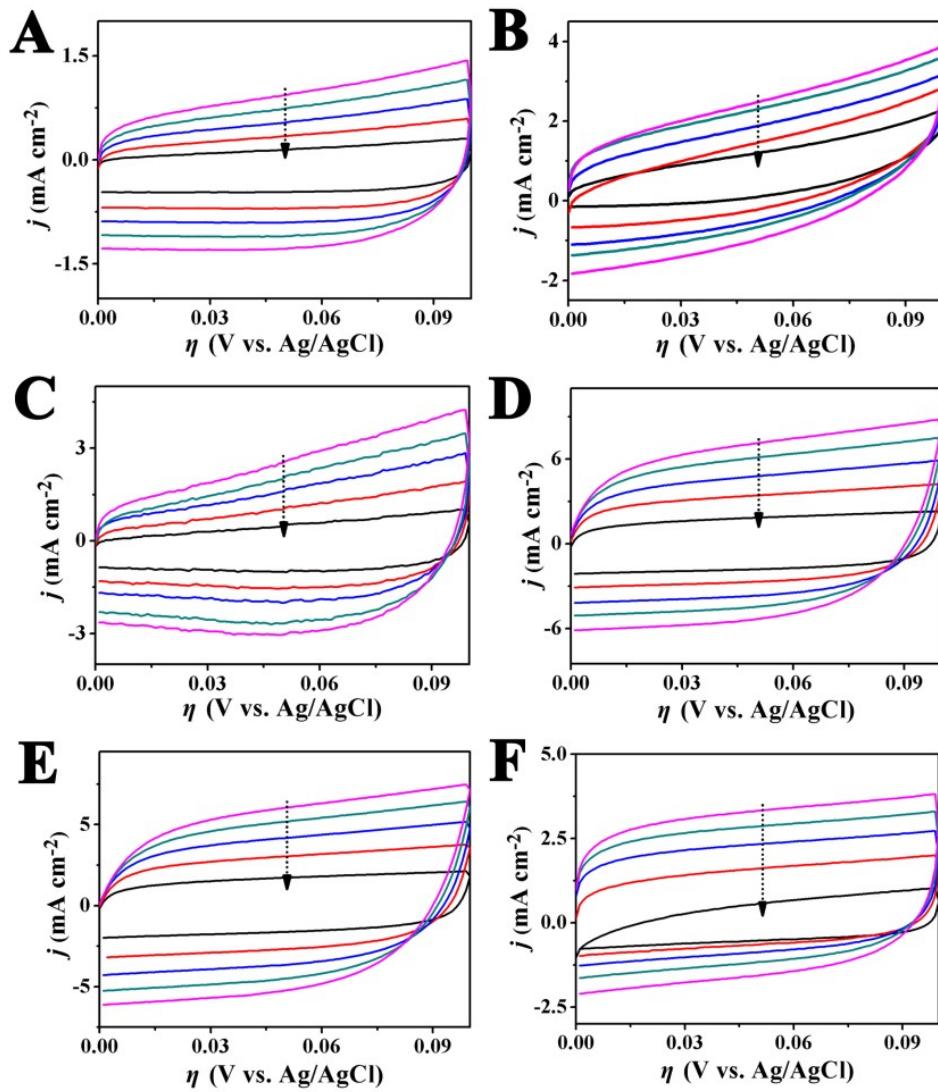
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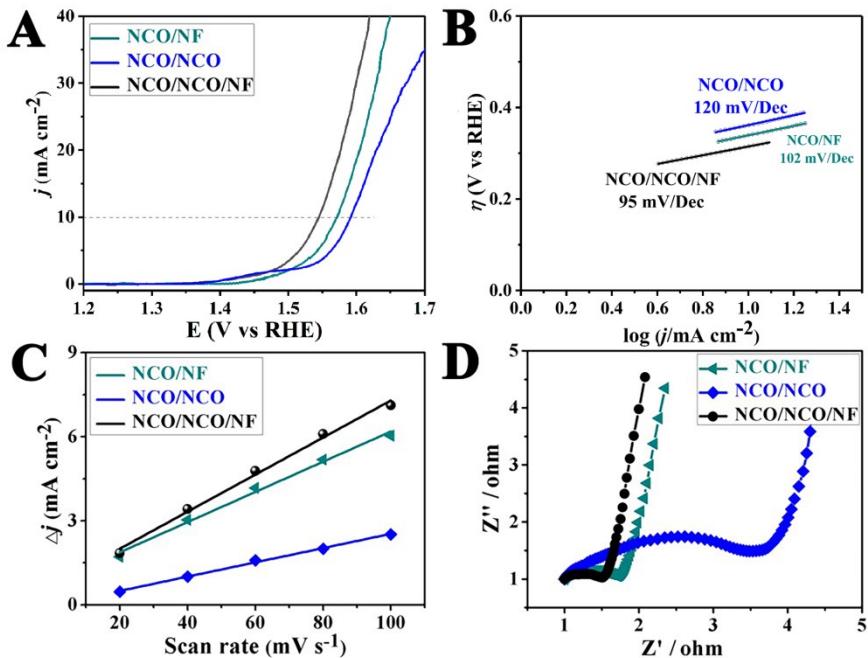
**Figure S1** Optical image of NF, NC-LDH/NF, NC-MOF/LDH/NF and NCO/NCO/NF electrode.



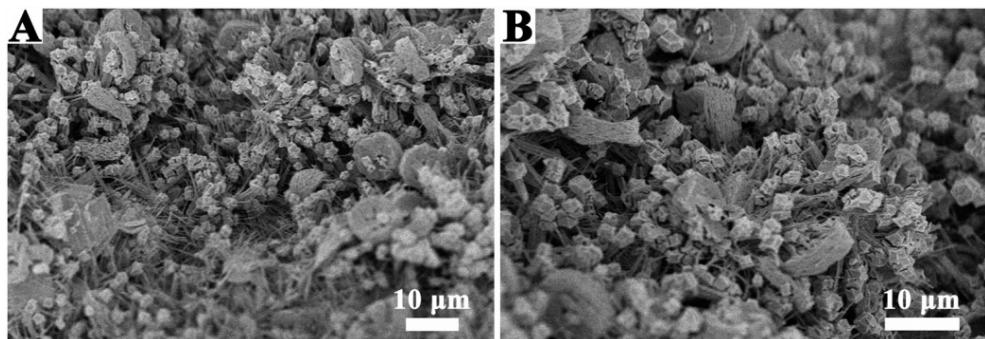
**Figure S2** (A) XRD patterns of the NCO/NF and NCO/NCO. (B) SEM image of NF. TEM and SEM images of (C, D) NCO/NF and (E, F) NCO/NCO.



**Figure S3** CV curves of (A) NF, (B) NC-LDH/NF, (C) NC-MOF/LDH/NF, (D) NCO/NCO/NF, (E) NCO/NF and (F) NCO/NCO in the potential range of 0.00 to 0.10 V vs. Ag/AgCl (non-Faradaic region) at different scan rates (20 - 100 mV s $^{-1}$ ).



**Figure S4** (A) Linear scan voltammograms (LSV) plots, (B) Tafel plots, (C) the differences in current density plotted against scan rate for estimation of double-layer capacitance ( $C_{dl}$ ), and (D) Nyquist plots of the NCO/NF, NCO/NCO and NCO/NCO/NF in 0.1 M KOH.



**Figure S5** SEM images of NCO/NCO/NF composite after 100 h of the catalytic oxidation reaction (A) and 2000 cycles (B), respectively.

**Table S1.** Comparison of OER activities for various NiCo<sub>2</sub>O<sub>4</sub> NWs based composites.

Catalyst	Substrate <sup>[c]</sup>	$j$ (mA cm <sup>-2</sup> )	$\eta$ (mV)	Tafel slope (mV Dec <sup>-1</sup> )	Loading amount of catalyst (mg cm <sup>-2</sup> )	Ref.
NiCo@NiCoO <sub>2</sub> /C PMRAs <sup>[a]</sup>	CFC	10	340	83.97	3.2	[1]
NiCo <sub>2</sub> O <sub>4</sub> /NiCo <sub>2</sub> O <sub>4</sub> NWs <sup>[a]</sup>	GO	10	373	63.1	--	[2]
MnO <sub>2</sub> /NiCo <sub>2</sub> O <sub>4</sub> /NF <sup>[a]</sup>	Ni foam	10	340	139	--	[3]
NiCo <sub>2</sub> O <sub>4</sub> /NiO/GO <sup>[a]</sup>	GO	10	350	66	--	[4]
NiCo <sub>2</sub> O <sub>4</sub> NWs <sup>[a]</sup>	Ni foam	9.46	550	68	--	[5]
Fe <sub>0.5</sub> Ni <sub>0.5</sub> Co <sub>2</sub> O <sub>4</sub> NWs <sup>[a]</sup>	Ni foam	10	350	27	--	[6]
NiCo <sub>2</sub> O <sub>4</sub> NWs <sup>[a]</sup>	Ti foil	0	500	62	0.33	[7]
NiCo <sub>2</sub> O <sub>4</sub> NA/CC <sup>[a]</sup>	CC	0	320	90	0.43	[8]
NiCo <sub>2</sub> O <sub>4</sub> nanoneedles <sup>[a]</sup>	FTO	10	565	292	0.53	[9]
CFP/NCO/ Co <sub>0.57</sub> Ni <sub>0.43</sub> <sup>[b]</sup>	CFP	10	340	63	1.477	[10]
NiMn-LDH/ NiCo <sub>2</sub> O <sub>4</sub> NWs <sup>[a]</sup>	Ni foam	10	310	99	5.4	[11]
NiCo <sub>2</sub> O <sub>4</sub> NWs <sup>[a]</sup>	Ni foam	10	363	121	2.5	[11]
<b>NCO/NCO/NF<sup>[b]</sup></b>	<b>Ni foam</b>	<b>10</b>	<b>320</b>	<b>80</b>	<b>1</b>	<b>This work</b>

All experiments are performed in 1.0 [a] and 0.1 [b] M KOH electrolyte, respectively.

[c] CFC = Carbon Fiber Cloth GO = Graphite Oxide, GCE = Glassy Carbon Electrode, CC = Carbon Cloth, FTO = Fluorine doped Tin Oxide, CB = Carbon Black, CFP = Carbon Fiber Paper.

## References

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