

Cation Doped Sea Urchin like MnO₂ for Electrocatalysis Overall Water Splitting

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#contributed equally to this work.

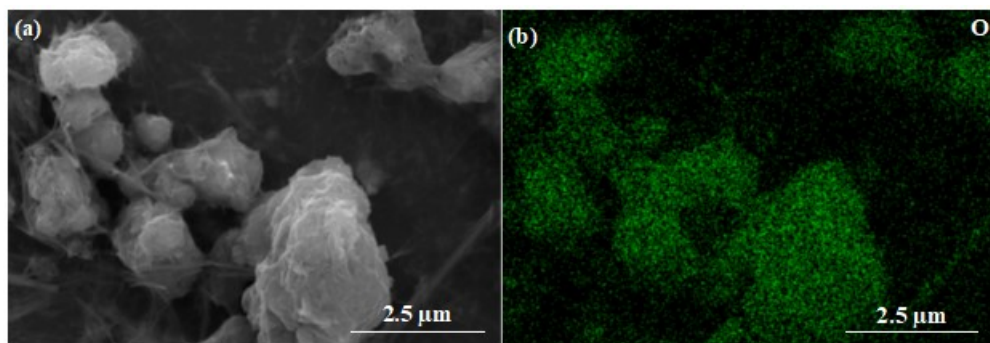


Figure S1. (a) SEM image, (b) elemental mapping image of O of MnO_2 .

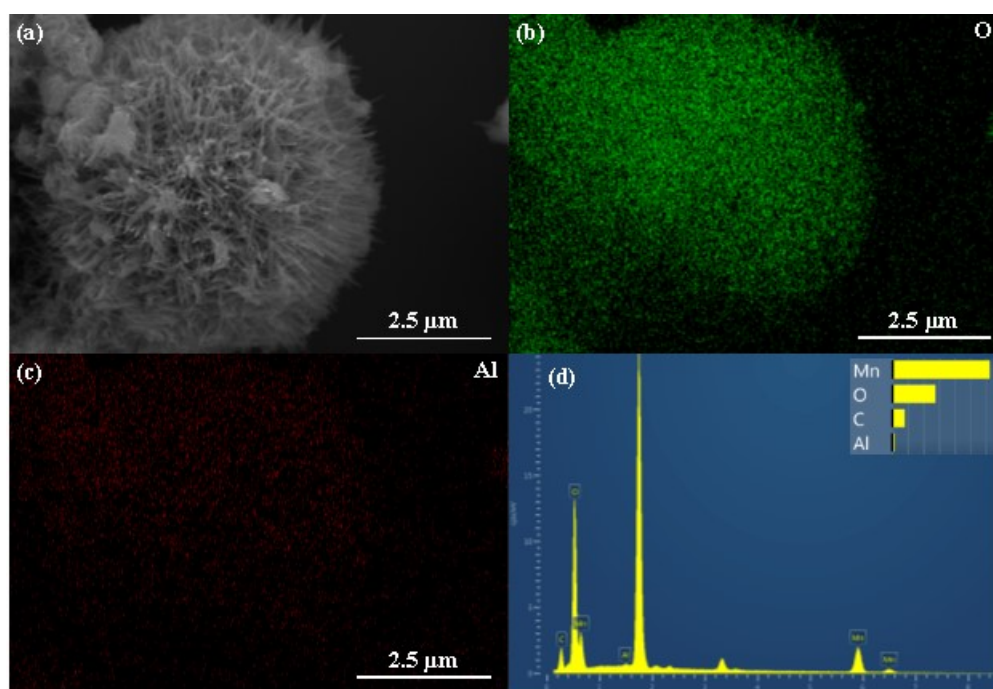


Figure S2. (a) SEM image, (b) elemental mapping images of (b) O, (c) Al and (d) EDS spectrum of Al-MnO_2 .

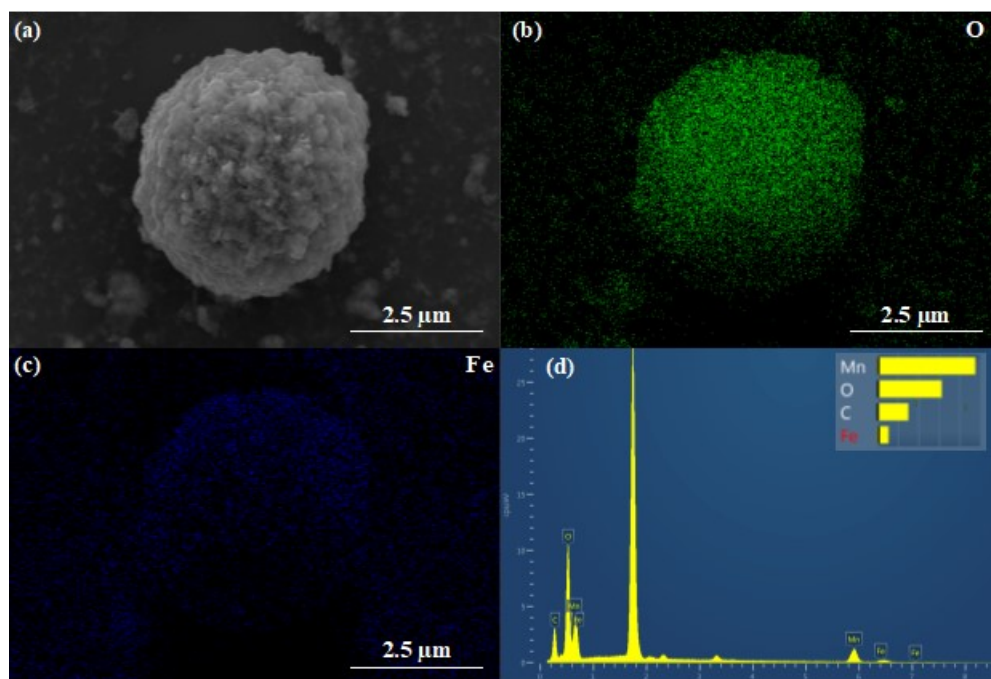


Figure S3. (a) SEM image, (b) elemental mapping images of (b) O, (c) Fe and (d) EDS spectrum of Fe-MnO₂.

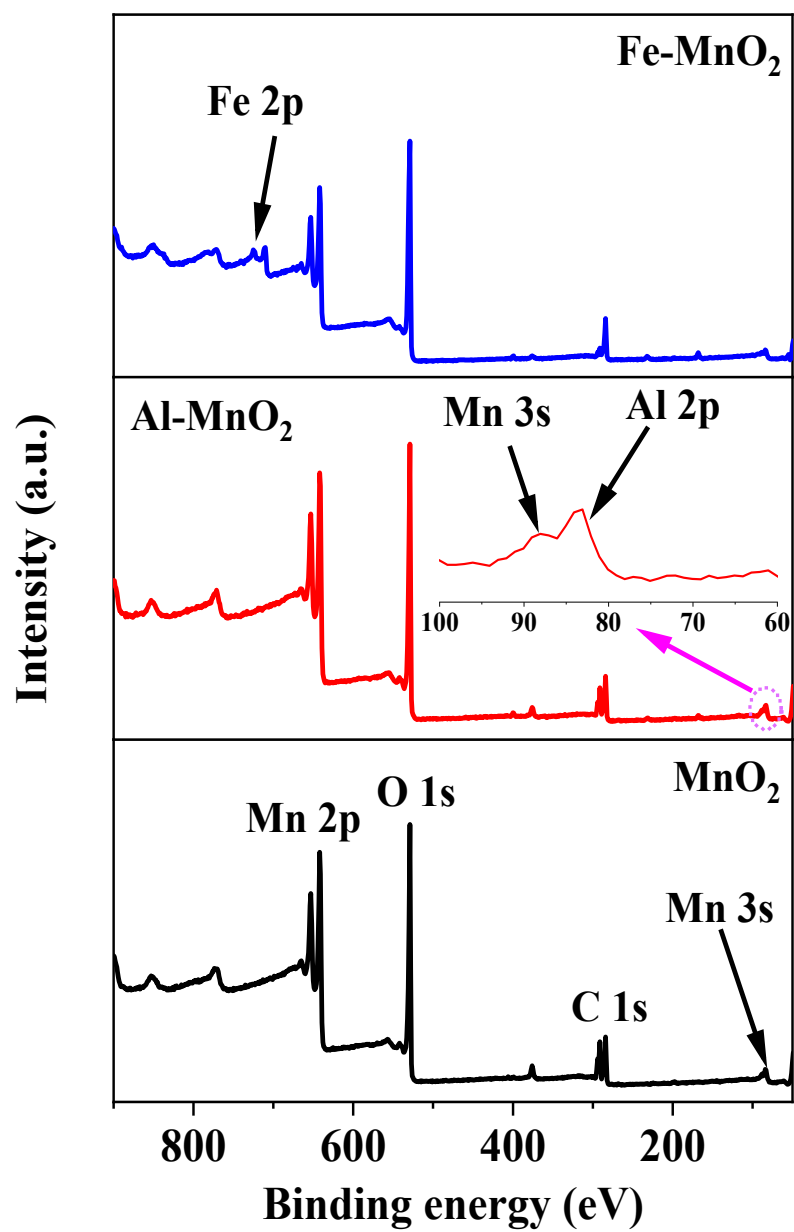


Figure S4. The survey XPS spectra of MnO₂, Al-MnO₂ and Fe-MnO₂.

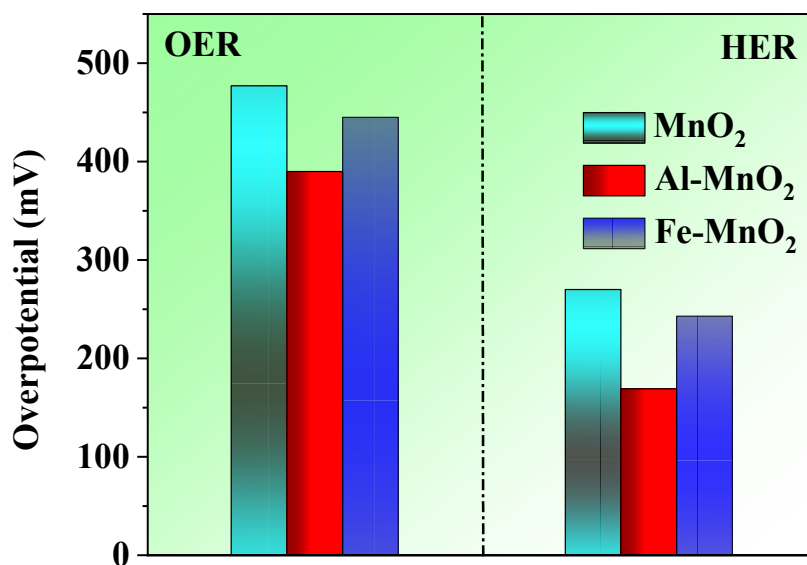


Figure S5. Overpotentials during OER and HER at 10 mA cm⁻² of MnO₂, Al-MnO₂ and Fe-MnO₂.

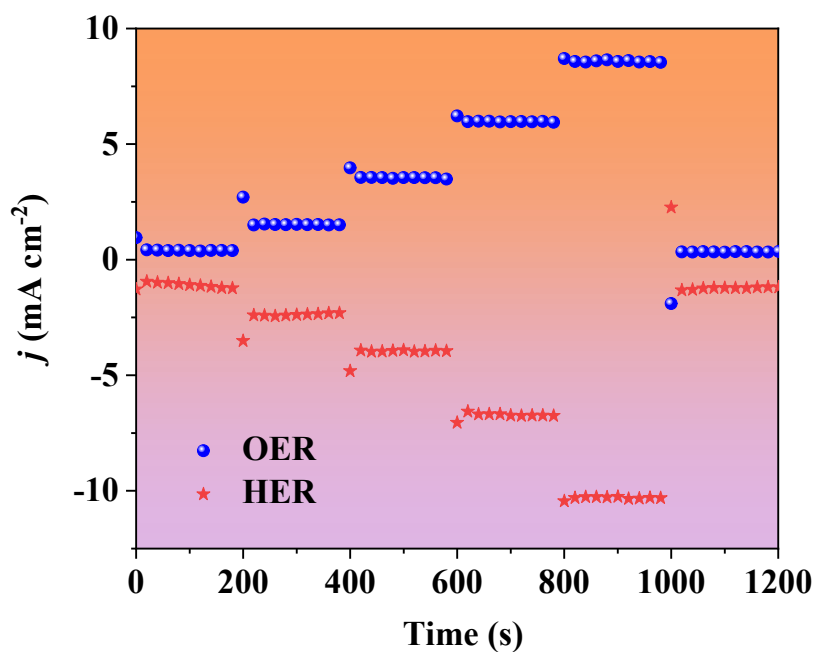


Figure S6. Time dependence of current densities of Al-MnO₂ recorded at different potentials for OER and HER, the potential change of 30 mV at each stage.

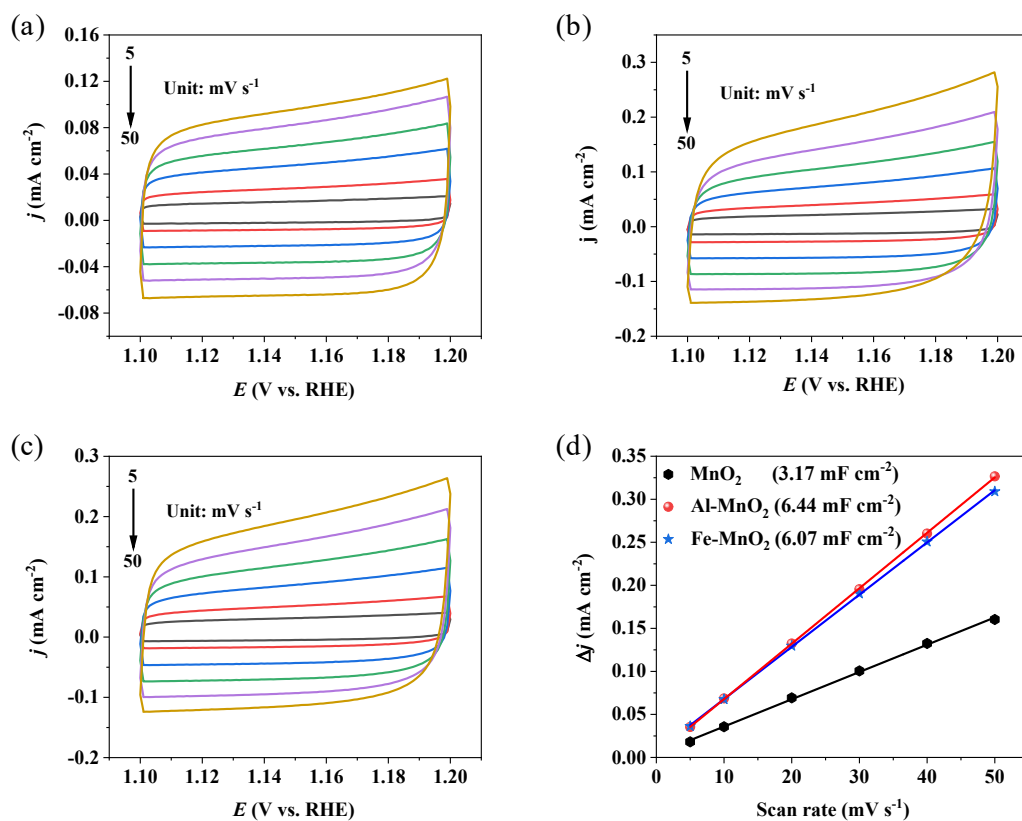


Figure S7. CVs of (a) MnO₂, (b) Al-MnO₂ and (c) Fe-MnO₂ run in a non-Faradic potential region at different scan rates (5, 10, 20, 30, 40 and 50 mV s⁻¹), (d) Double-layer capacitance of MnO₂, Al-MnO₂ and Fe-MnO₂.

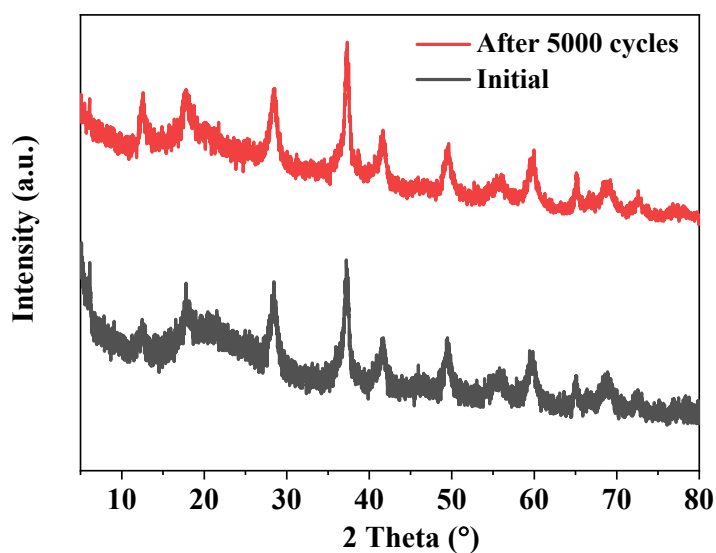


Figure S8 XRD patterns of initial and after stability measured Al-MnO₂.

Table S1 The electrochemical performances of Al-MnO₂ compared with the similar catalysts.

Sample	Electrolyte	OER			HER			Ref.
		E ₁₀ (mV)	Tafel (mV dec ⁻¹)	Stability	E ₁₀ (mV)	Tafel (mV dec ⁻¹)	Stability	
Al-MnO ₂	1 M KOH	390	107.9	5000 cycles	170	148.0	5000 cycles	This work
MnO ₂	1 M KOH	410	67	80 h	150	64	80 h	[1]
MnO ₂ @C	1 M NaOH	270	38.55	/	460	142.15	/	[2]
MnO ₂	1 M KOH	530	173.06	5000 cycles	/	/	/	[3]
MnO ₂ -NF-48h	1 M KOH	450	123	40000 s	140	110	40000 s	[4]

Reference

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