

Supporting Information

Regulating luminescence thermal quenching based on the synergistic effect of energy transfer and energy gap modulation

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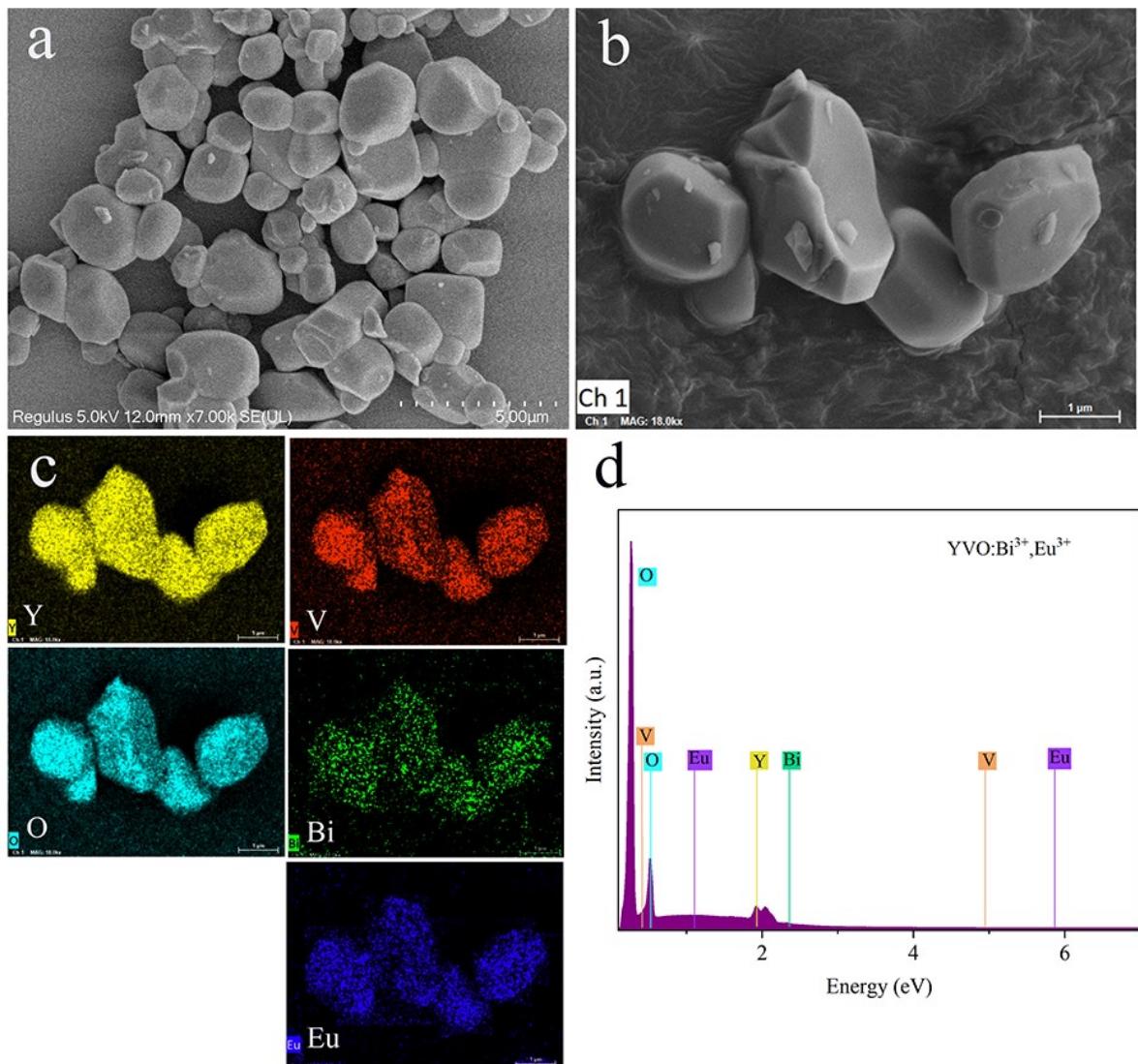


Fig. S1 (a) Typical SEM image of YVO₄:Bi³⁺, Eu³⁺solid solution. (b) Mapping diagram of YVO₄:Bi³⁺, Eu³⁺solid solution. (c) The corresponding elemental mapping analysis for the Y, V, O, Bi and Eu elements. (d) EDS spectra of YVO₄:Bi³⁺, Eu³⁺.

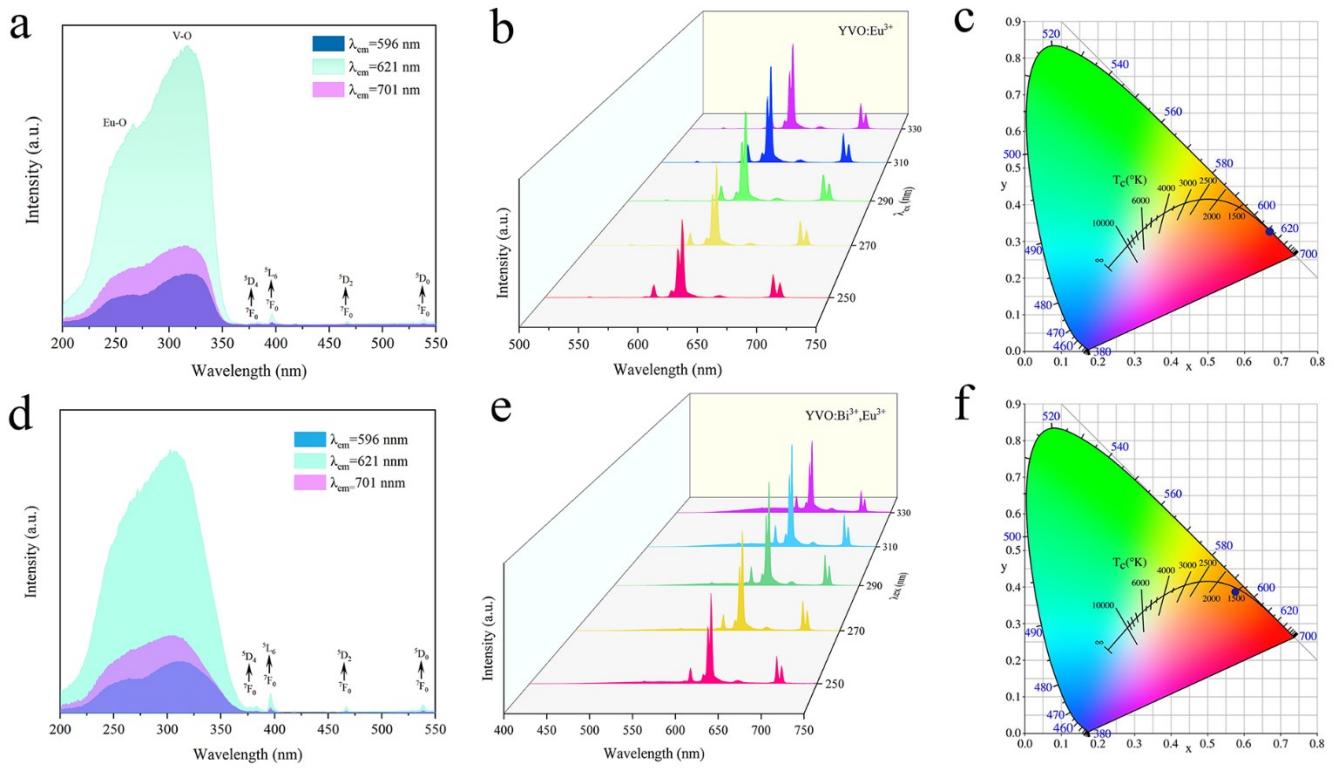


Fig. S2 (a) PLE spectra of YVO₄: Eu³⁺ ($\lambda_{\text{em}} = 596 \text{ nm}, 621 \text{ nm}, 621 \text{ nm}$). (b) PL spectra of YVO₄: Eu³⁺ ($\lambda_{\text{ex}} = 250 \text{ nm}, 270 \text{ nm}, 290 \text{ nm}, 310 \text{ nm}, 330 \text{ nm}$). (c) CIE chromaticity diagram ($\lambda_{\text{ex}} = 310 \text{ nm}$). (d) PLE spectra of YVO₄:Bi³⁺, Eu³⁺ ($\lambda_{\text{em}} = 596 \text{ nm}, 621 \text{ nm}, 621 \text{ nm}$). (e) PL spectra of YVO₄:Bi³⁺, Eu³⁺ ($\lambda_{\text{ex}} = 250 \text{ nm}, 270 \text{ nm}, 290 \text{ nm}, 310 \text{ nm}, 330 \text{ nm}$). (f) CIE chromaticity diagram ($\lambda_{\text{ex}} = 310 \text{ nm}$).

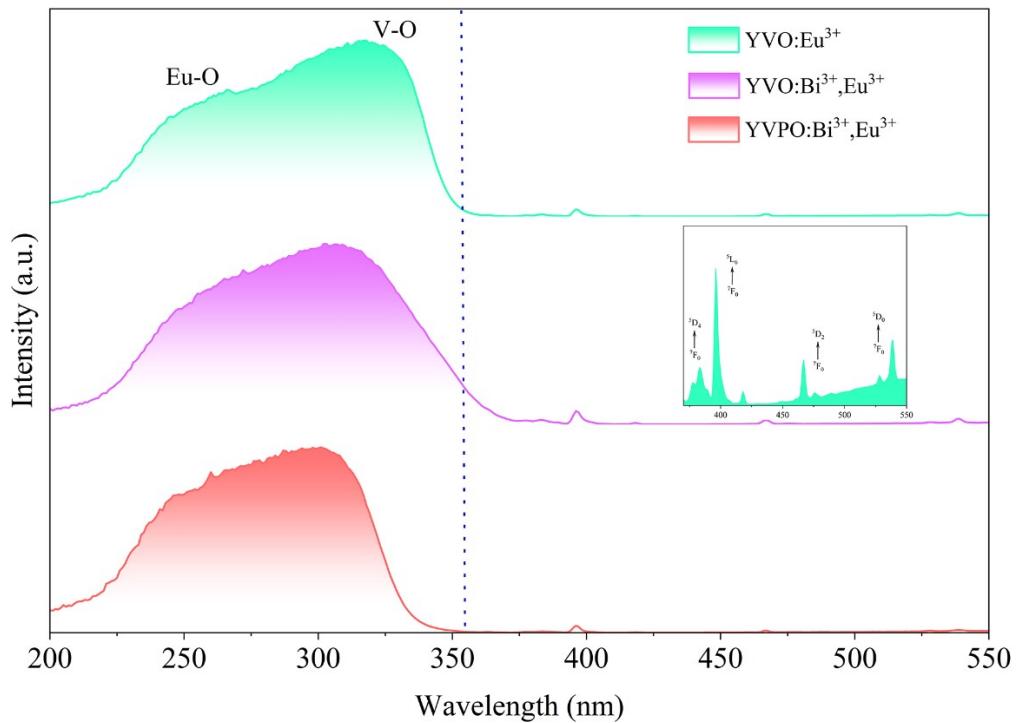


Fig. S3 PLE spectra of YVO₄: Eu³⁺, YVO₄:Bi³⁺, Eu³⁺ and YV_{0.25}P_{0.75}O₄:Bi³⁺, Eu³⁺ at $\lambda_{\text{em}}=621\text{nm}$.

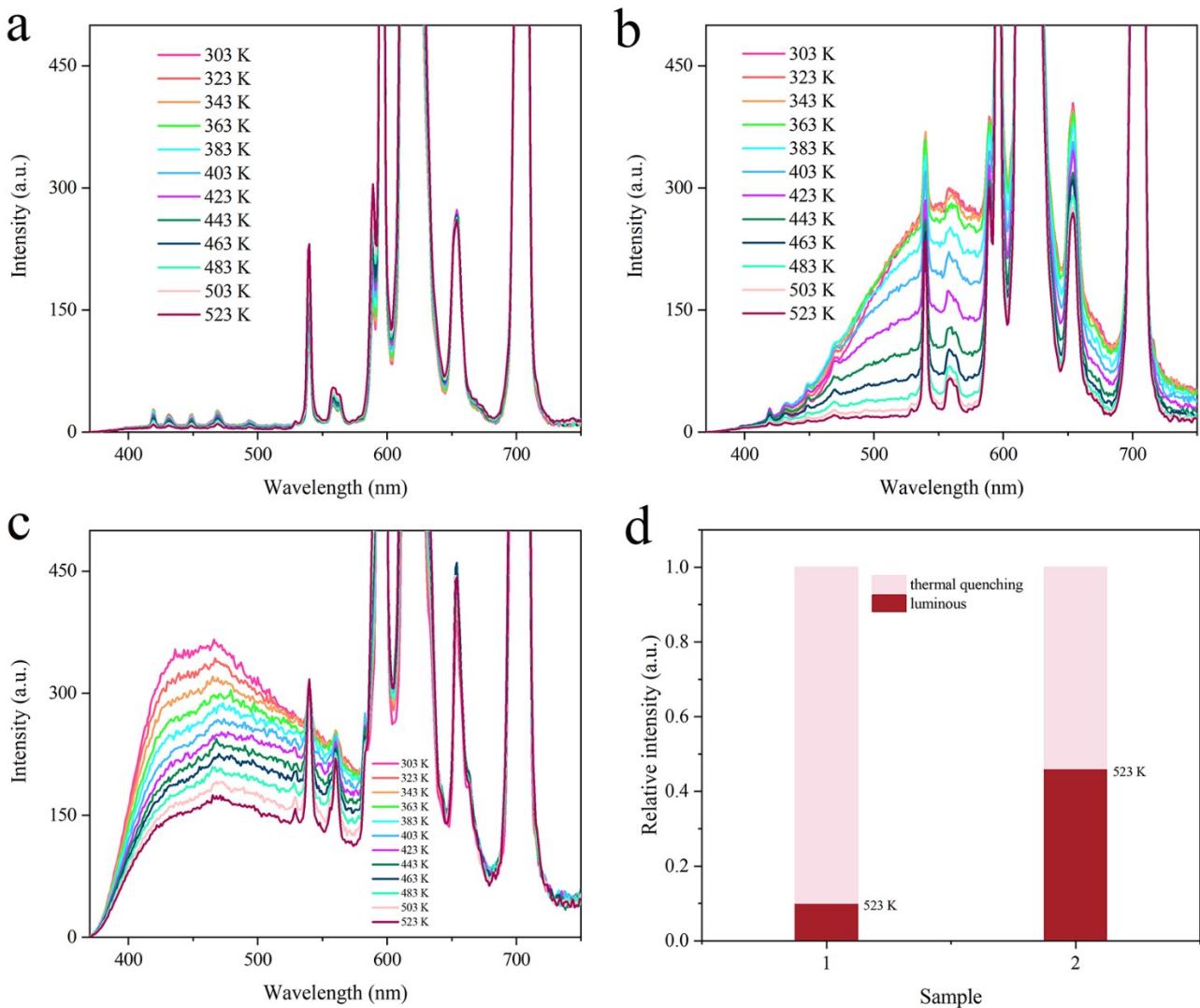


Fig. S4 PL spectra of (a) YVO_4 : Eu^{3+} ($\lambda_{\text{ex}}=320$ nm), (b) PL spectra of $\text{YVO}_4:\text{Bi}^{3+}, \text{Eu}^{3+}$ ($\lambda_{\text{ex}}=310$ nm), (c) PL spectra of $\text{YV}_{0.25}\text{P}_{0.75}\text{O}_4:\text{Bi}^{3+}, \text{Eu}^{3+}$ ($\lambda_{\text{ex}}=305$ nm) at various temperatures. (d) Relative emission intensity of Bi^{3+} at various temperature.

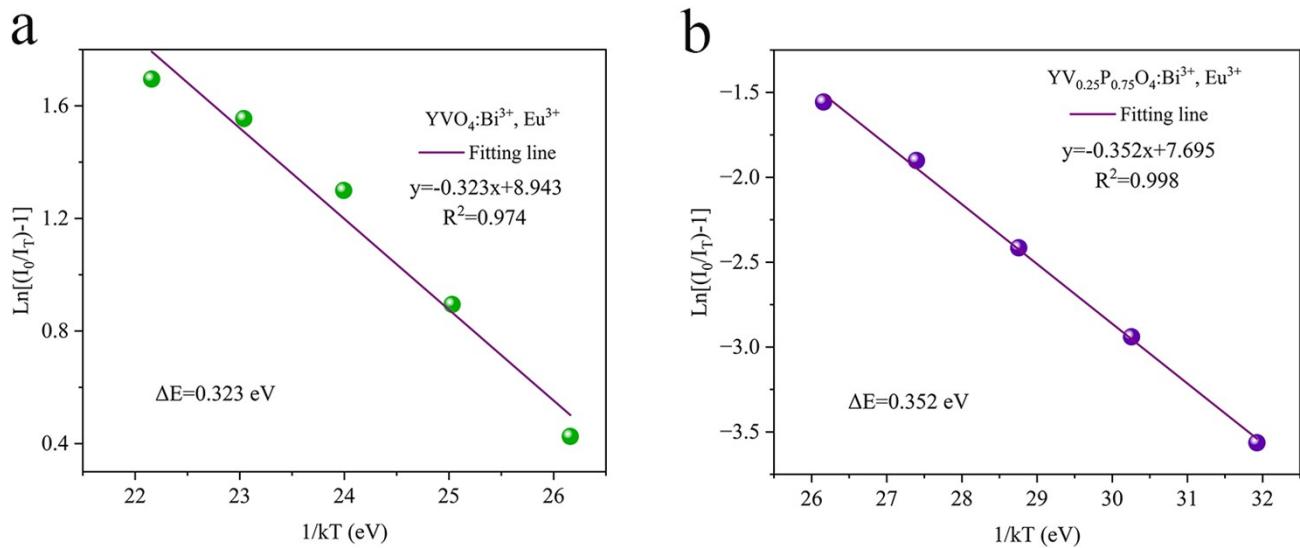


Fig. S5 Thermal quenching activation energy (ΔE) of (a) YVO₄:Bi³⁺, Eu³⁺ and (b) YV_{0.25}P_{0.75}O₄:Bi³⁺, Eu³⁺.