

Supplementary Information

A new series of magnetic and luminescent layered hybrid materials obtained from thianthrene phosphonic acid: $M(H_2O)PO_3-S_2C_{12}H_7$ ($M = Cu, Zn$) and $M(H_2O)_2(PO_2OH-S_2C_{12}H_7)_2$ ($M = Mn, Co$)

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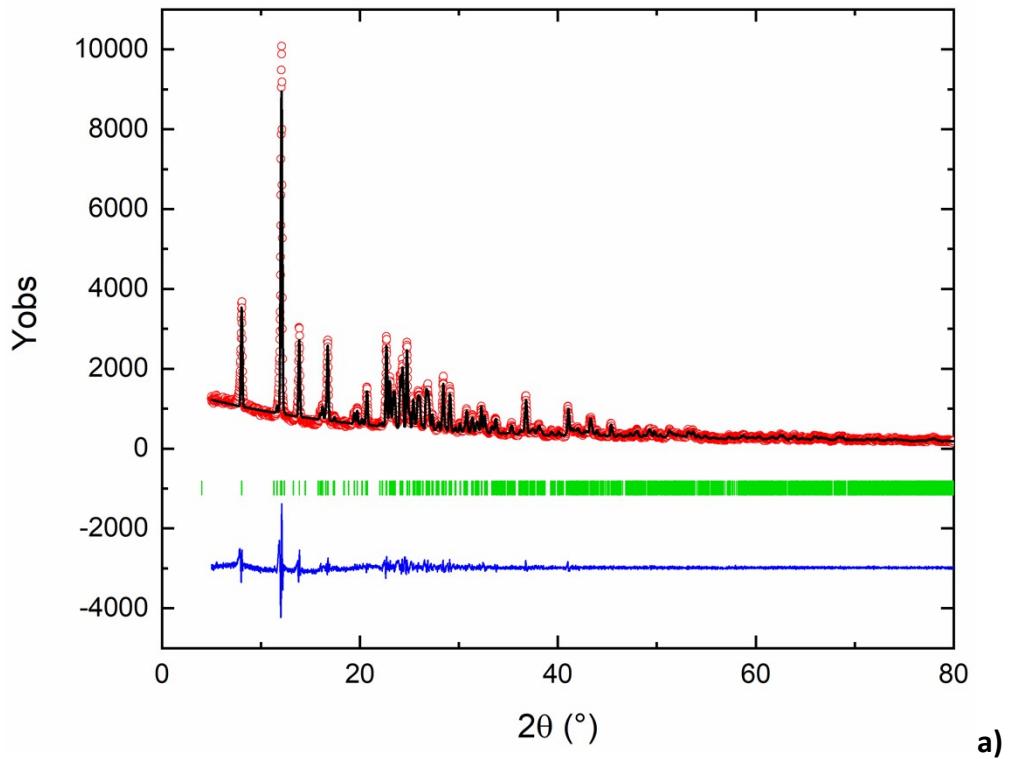
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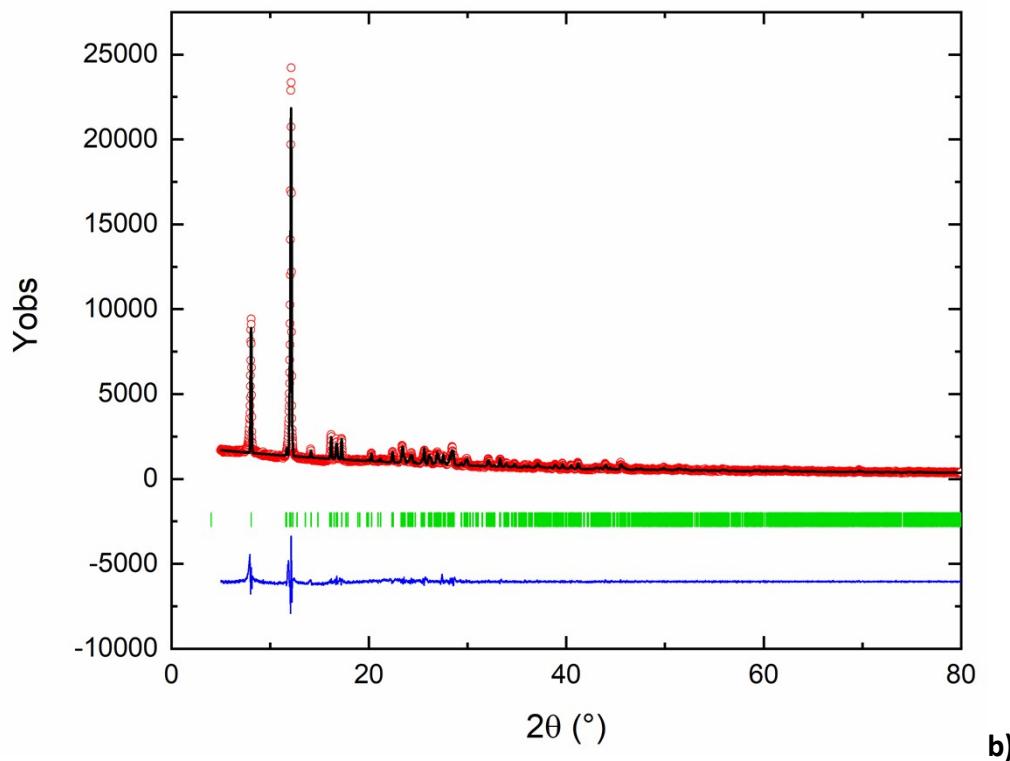
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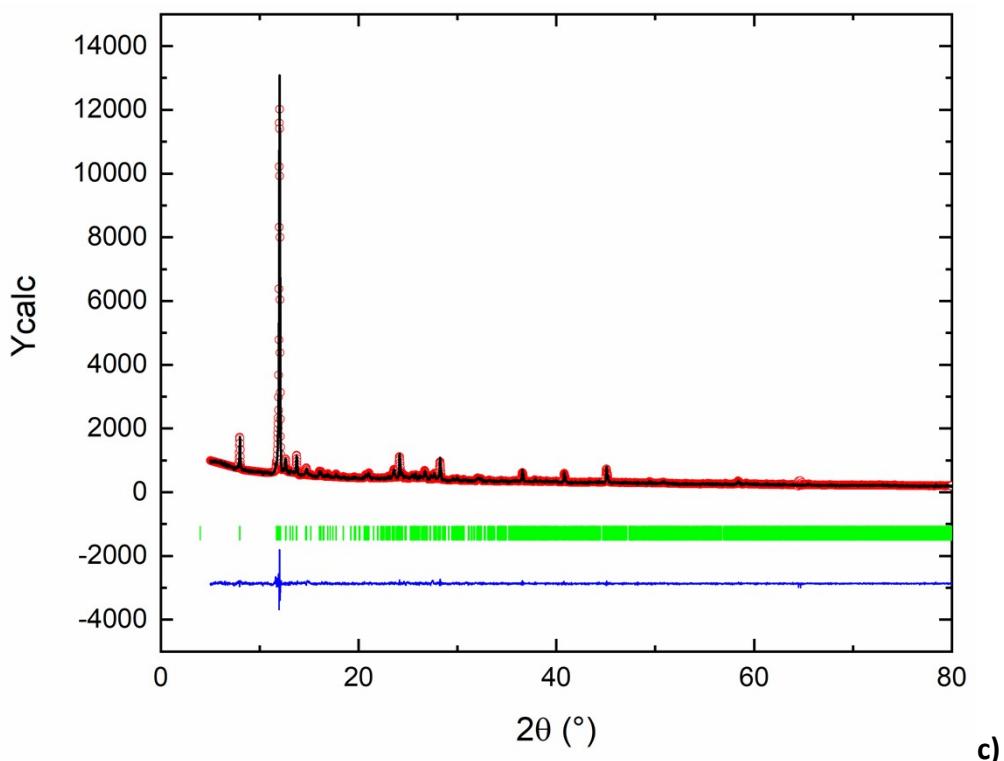
FIGURE SI 1 : Powder X ray diffraction recorded on compounds a) Zn(H ₂ O)(PO ₃ S ₂ C ₁₂ H ₇) (4); b) Cu(H ₂ O)(PO ₃ S ₂ C ₁₂ H ₇) (5), c) Co(H ₂ O) ₂ (PO ₂ OHS ₂ C ₁₂ H ₇) ₂ (6) and d) Mn(H ₂ O) ₂ (PO ₂ OHS ₂ C ₁₂ H ₇) ₂ (7). Experimental diagram (red circles), calculated position of the peaks (green dashes), calculated pattern from the single cif file (black line), difference between experimental and calculated pattern (blue line)	4
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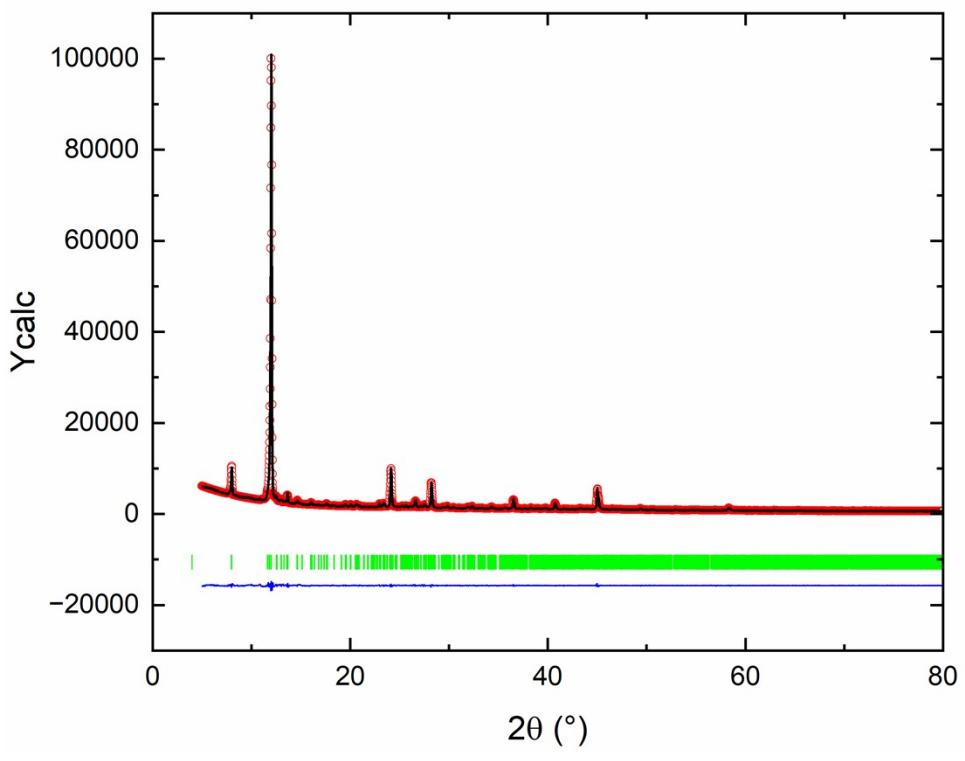
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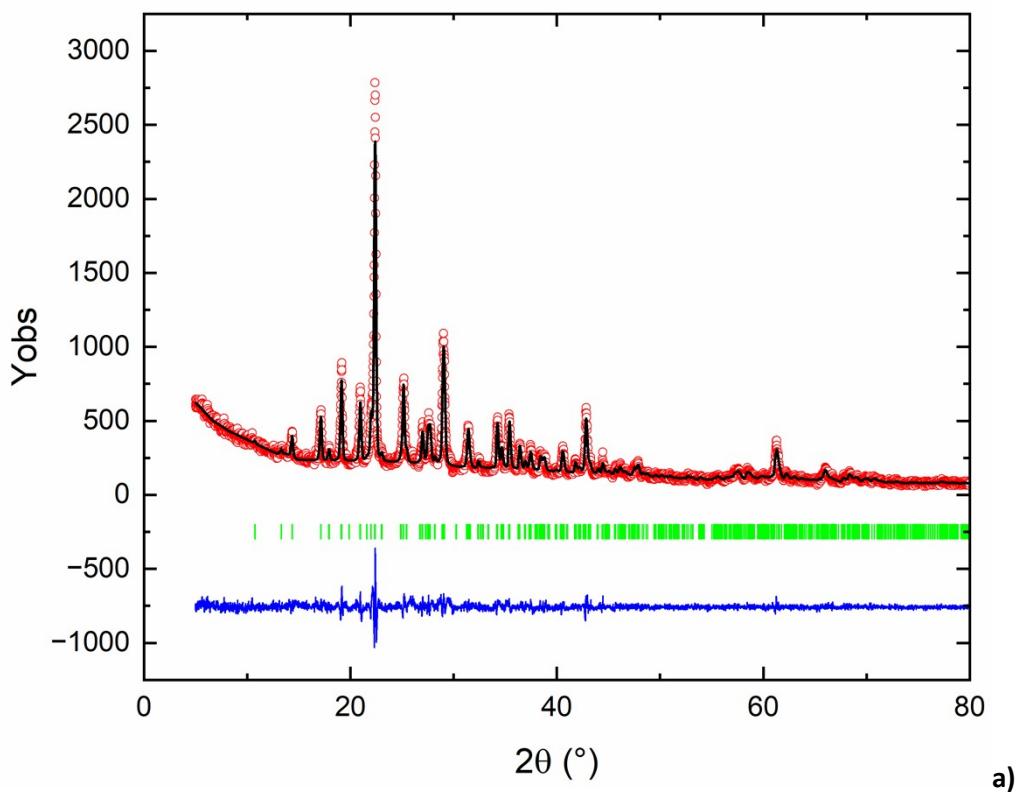


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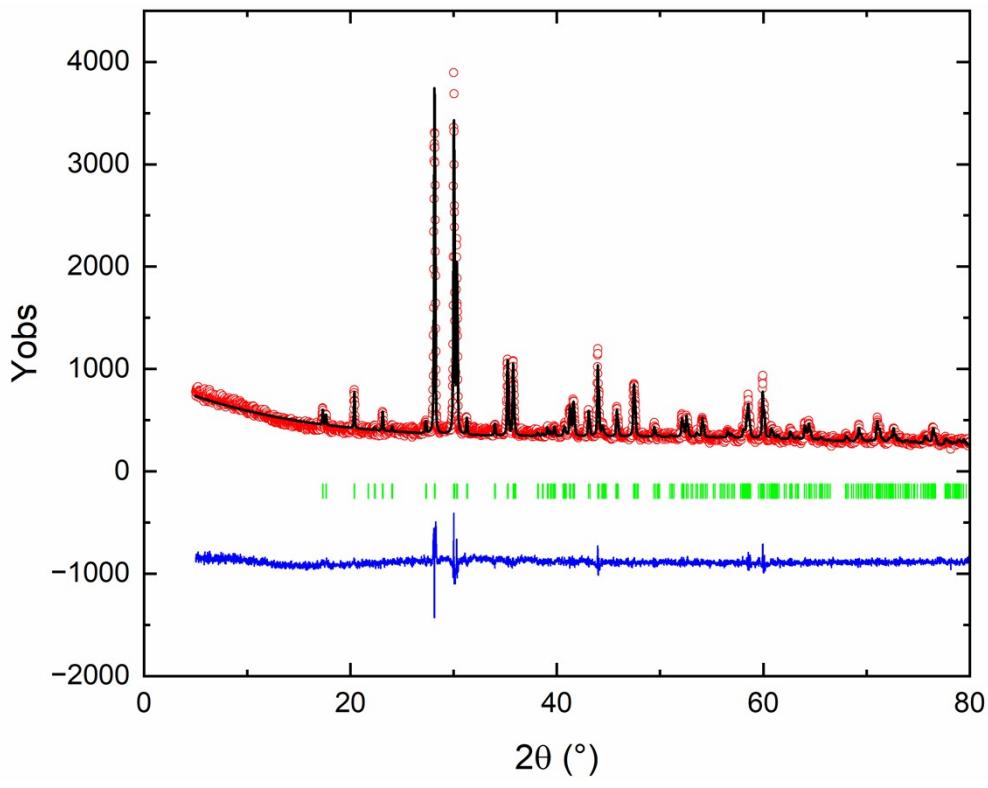


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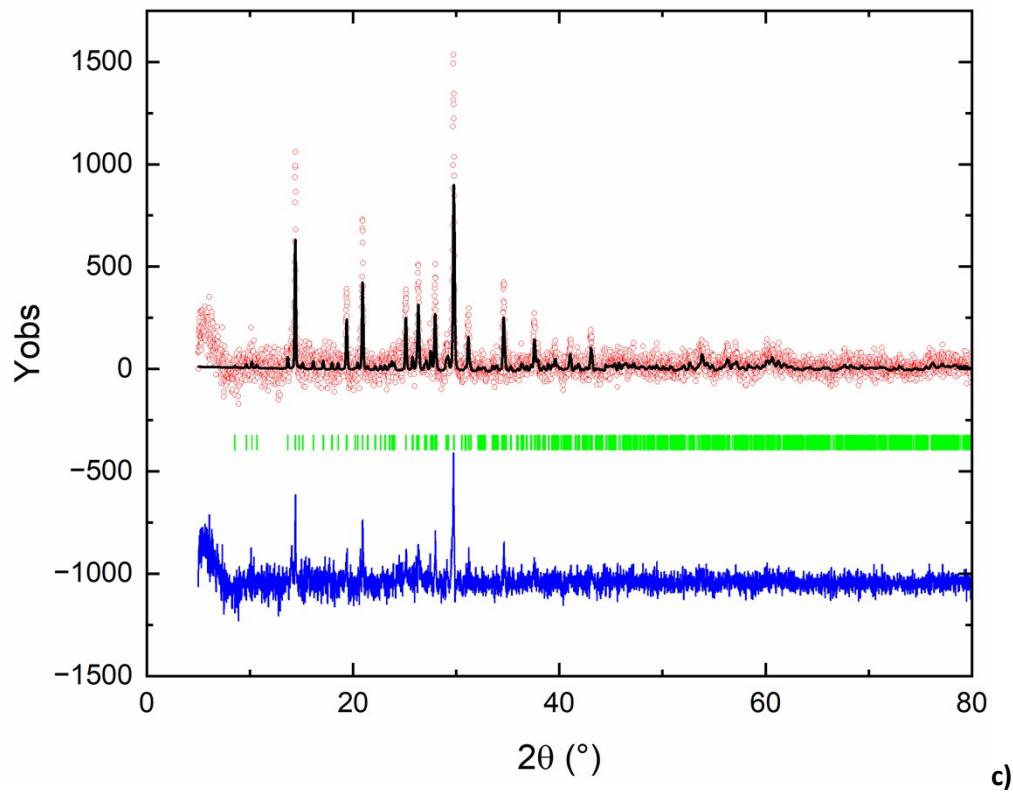
FIGURE SI 1 : Powder X ray diffraction recorded on compounds a) $\text{Zn}(\text{H}_2\text{O})(\text{PO}_3\text{S}_2\text{C}_{12}\text{H}_7)$ (**4**); b) $\text{Cu}(\text{H}_2\text{O})(\text{PO}_3\text{S}_2\text{C}_{12}\text{H}_7)$ (**5**), c) $\text{Co}(\text{H}_2\text{O})_2(\text{PO}_2\text{OHS}_2\text{C}_{12}\text{H}_7)_2$ (**6**) and d) $\text{Mn}(\text{H}_2\text{O})_2(\text{PO}_2\text{OHS}_2\text{C}_{12}\text{H}_7)_2$ (**7**). Experimental diagram (red circles), calculated position of the peaks (green dashes), calculated pattern from the single cif file (black line), difference between experimental and calculated pattern (blue line)



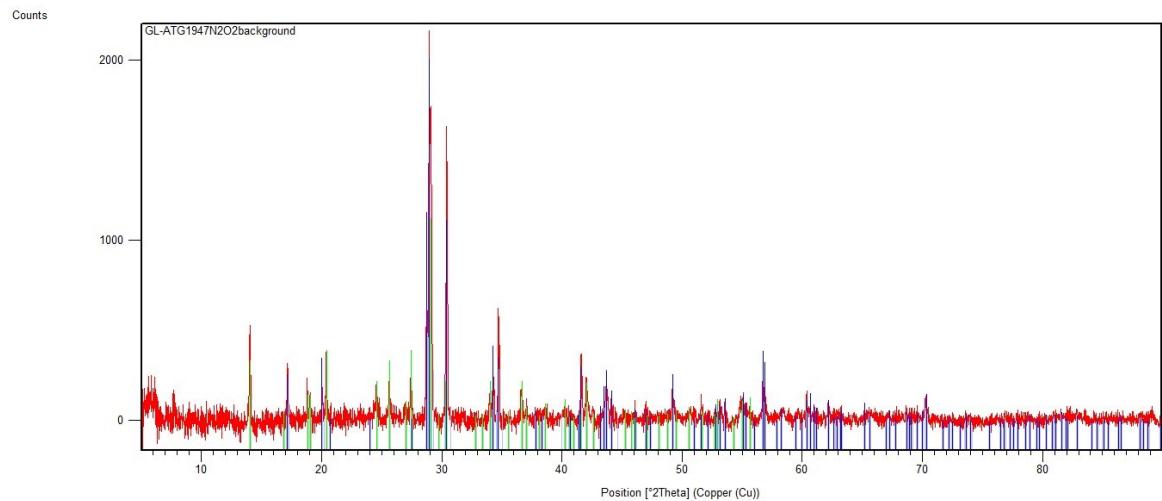
a)



b)



c)



d)

FIGURE SI 2 : X-ray diffraction pattern of the decomposition products of the four hybrid materials (a) Zn₂P₂O₇ (b) Cu₂P₂O₇ (c) CoP₂O₆ and (d) a mix of Mn₂P₂O₇ (blue) and MnP₂O₆ (green)

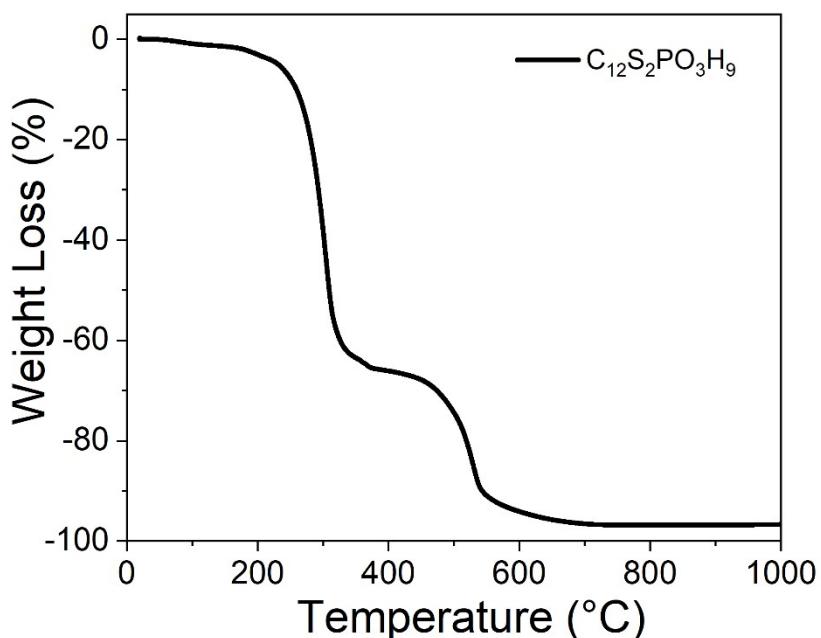


FIGURE SI 3 : thermogravimetric curve of $\text{PO}(\text{OH})_2\text{S}_2\text{C}_{12}\text{H}_7$ (**3**)

O\cdotsH-O hydrogen bonds between the phosphonic acid functions of $\text{PO}(\text{OH})_2\text{S}_2\text{C}_{12}\text{H}_7$ (blue dashed lines)			
O16-H16 \cdots O18	3.2623(17) Å	O18-H18 \cdots O16	3.180(2) Å
O18-H18 \cdots O18	3.217(2) Å	O18-H18 \cdots O18	3.165(2) Å
O18-H18 \cdots O17	1.7999(17) Å	O16-H16 \cdots O17	1.834(2) Å
C\cdotsH-C interaction between the C_6H_4 rings of $\text{PO}(\text{OH})_2\text{S}_2\text{C}_{12}\text{H}_7$ in one layer (light green dashed lines)			
C5-H5 \cdots C4	3.158(3) Å	C5-H5 \cdots C5	2.813(4) Å
C5-H5 \cdots C6	3.160(4) Å		
C\cdotsH-C interaction between two adjacent layers (dark green dashed lines)			
C11-H11 \cdots C10	3.299(3) Å	C11-H11 \cdots C8	3.285(3) Å
C12-H12 \cdots C6	2.970(3) Å	C7-H7 \cdots C12	3.205(3) Å
C12-H12 \cdots C7	2.709(4) Å		
S\cdotsH-C interaction between two adjacent layer (pink dashed lines)			
C14-H14 \cdots S2	2.9836(8) Å	C11-H11 \cdots S9	3.2007(8) Å

Table SI 1 : Intermolecular: i) O \cdots H-O hydrogen bonds ranging from 1.8 to 3.3 Å; ii) C \cdots H-C interactions ranging from 2.7 to 3.3 Å; iii) S \cdots H-C interactions ranging from 2.9 to 3.4 Å.

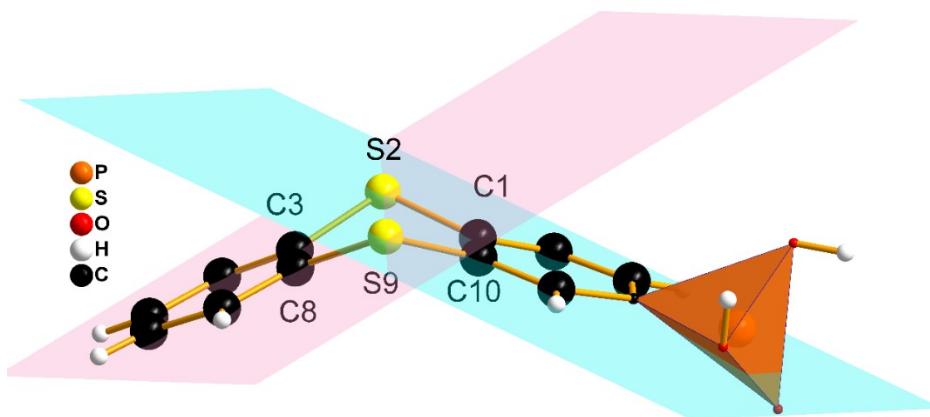


FIGURE SI 4: Chemical structure of 3-thianthrenephosphonic acid precursor.

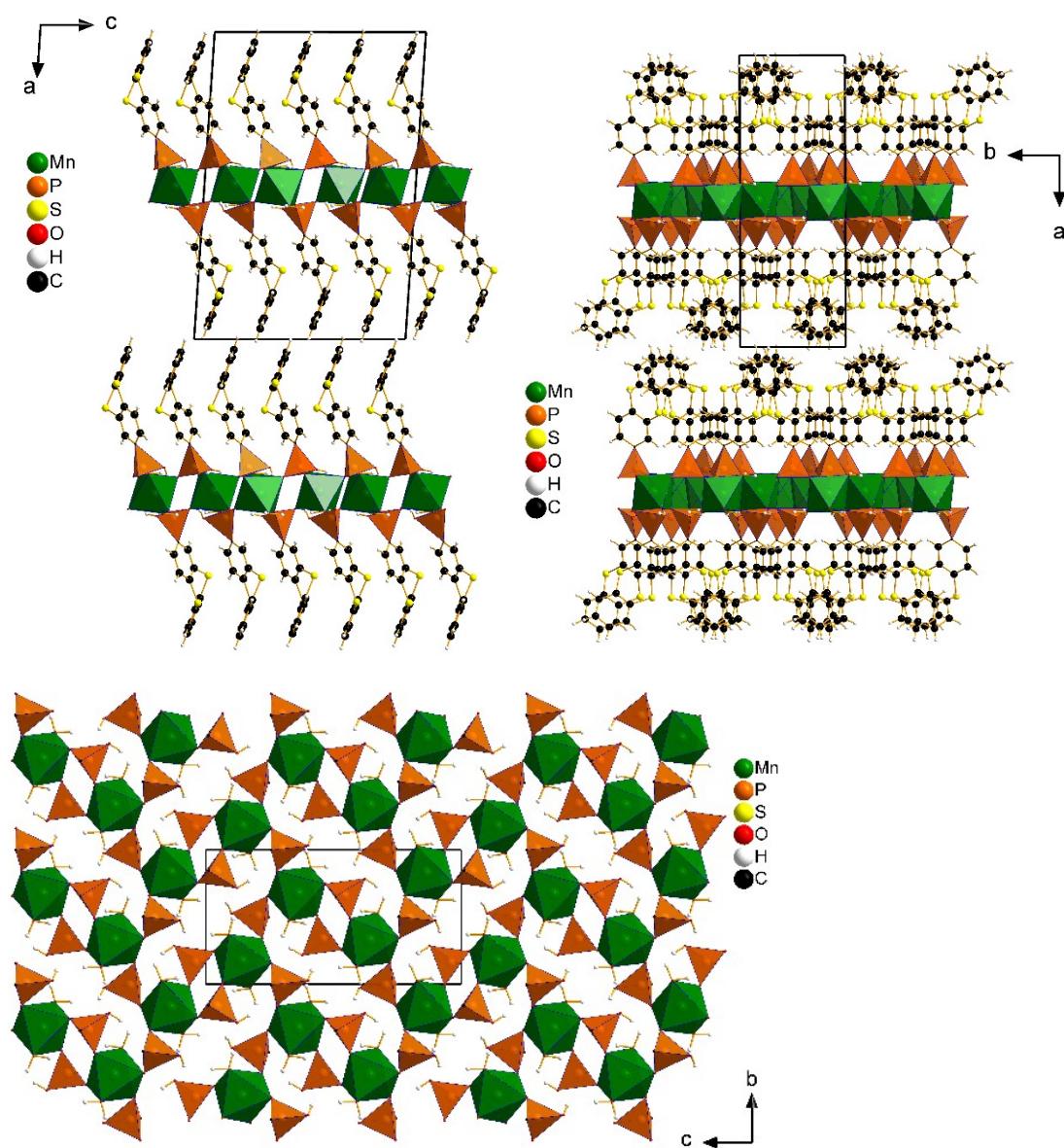


FIGURE SI 5 : View of the lamellar structure of $\text{Mn}(\text{H}_2\text{O})_2(\text{PO}_2\text{OHS}_2\text{C}_{12}\text{H}_7)_2$ (7).

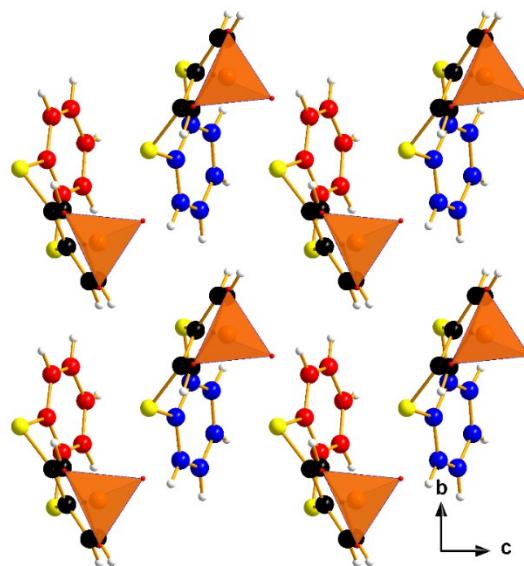


FIGURE SI 6 View of the orientation of the C_6H_4 ring along the b and $-b$ direction of $Zn(H_2O)(PO_3S_2C_{12}H_7)$ (**4**) also present in $Cu(H_2O)(PO_3S_2C_{12}H_7)$ (**5**).

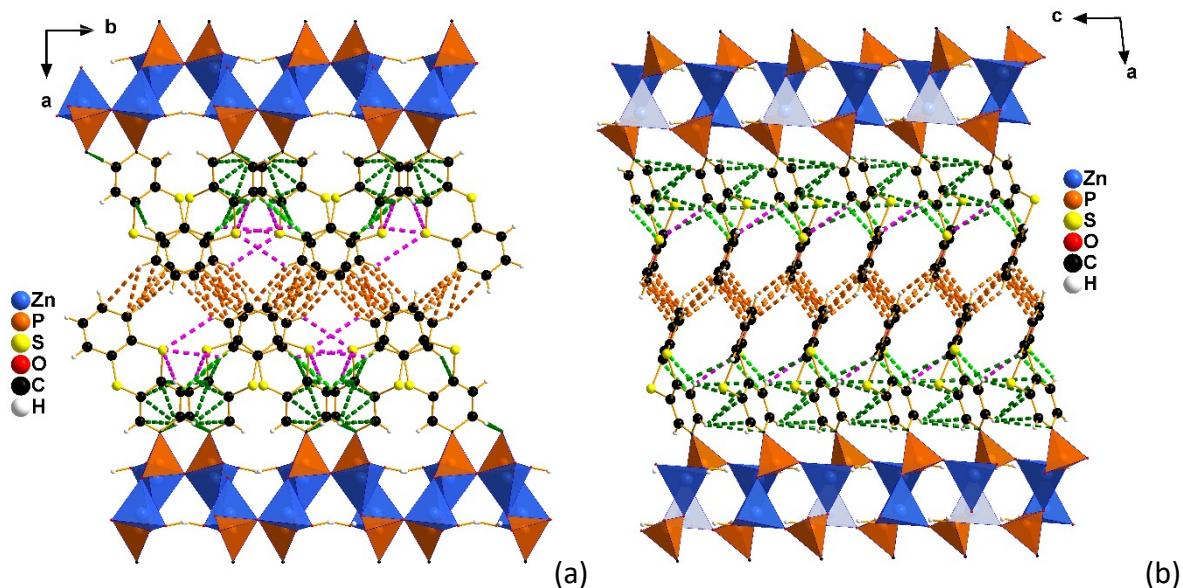


FIGURE SI 7: (a) Representation of the $C\bullet\bullet\bullet H-C$ (light and green dashed lines) and $S\bullet\bullet\bullet H-C$ (pink dashed lines) interactions in the structure of $Zn(H_2O)(PO_3S_2C_{12}H_7)$ (**4**) (a) view along the c direction; (b) view along the b direction.

C\cdotsH-C interaction between the PO₃S₂C₁₂H₉ moieties of Zn(H₂O)(PO₃S₂C₁₂H₇) (4) in one column (light green dashed lines)			
C8-H8 \cdots C5	3.258(4) Å	C5-H5 \cdots C8	3.263(5) Å
C\cdotsH-C interaction between the PO₃S₂C₁₂H₉ moieties of (4) in two adjacent columns (dark green dashed lines)			
C6-H6 \cdots C1	3.283(4) Å	C6-H6 \cdots C2	3.099(4) Å
C5-H5 \cdots C3	3.278(4) Å	C6-H6 \cdots C3	3.097(4) Å
C5-H5 \cdots C4	3.045(4) Å	C8-H8 \cdots C4	3.395(4) Å
C6-H6 \cdots C4	3.351(3) Å		
C\cdotsH-C interaction between two adjacent organic layers of (4) (orange dashed lines)			
C11-H11 \cdots C10	3.259(5) Å	C10-H10 \cdots C9	3.283(5) Å
C11-H11 \cdots C11	3.310(4) Å	C11-H11 \cdots C9	3.113(5) Å
C9-H9 \cdots C11	3.127(4) Å	C9-H9 \cdots C10	3.343(5) Å
S\cdotsH-C interaction between the PO₃S₂C₁₂H₉ moieties of (4) (pink dashed lines)			
C9-H9 \cdots S1	3.2446(10) Å	C8-H8 \cdots S1	3.1232(10) Å
C5-H5 \cdots S1	3.0354(10) Å		

Table SI 2 Intermolecular: i) C \cdots H-C interactions between 2.7 to 3.4 Å; ii) S \cdots H-C interactions between 2.9 to 3.4 Å.

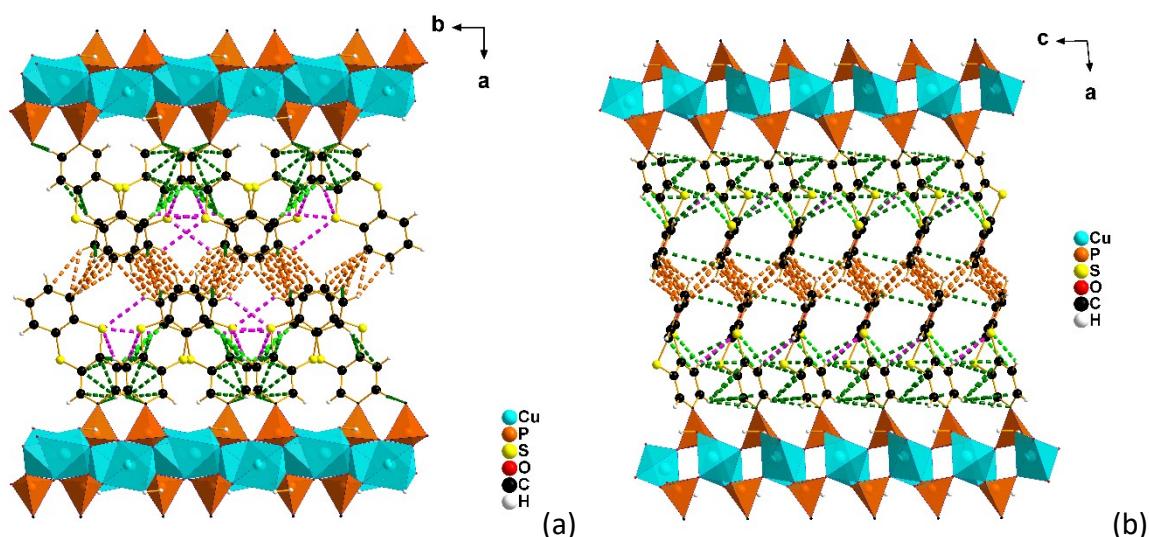


FIGURE SI 8 : (a) Representation of the C \cdots H-C (light and green dashed lines) and S \cdots H-C (pink dashed lines) interactions in the structure of Cu(H₂O)(PO₃S₂C₁₂H₇) (**5**) (a) view along the c direction; (b) view along the b direction.

C•••H-C interaction between the $\text{PO}_3\text{S}_2\text{C}_{12}\text{H}_9$ moieties of $\text{Cu}(\text{H}_2\text{O})(\text{PO}_3\text{S}_2\text{C}_{12}\text{H}_7)$ (5) in one column (light green dashed lines)			
C11-H11•••C3	3.304 (5) Å	C3-H3•••C11	3.348 (5) Å
C•••H-C interaction between the $\text{PO}_3\text{S}_2\text{C}_{12}\text{H}_9$ moieties of (5) in two adjacent columns (dark green dashed lines)			
C8-H8•••C10	3.270 (6) Å	C2-H2•••C5	3.069 (4) Å
C3-H3•••C4	2.856 (4) Å	C11-H11•••C4	3.291 (4) Å
C2-H2•••C2	3.288 (5) Å	C2-H2•••C1	3.056 (4) Å
C2-H2•••C6	2.949 (5) Å	C3-H3•••C5	3.075 (4) Å
C2-H2•••C4	3.304 (4) Å	C11-H11•••C3	3.300 (5) Å
C•••H-C interaction between two adjacent organic layers of (5) (orange dashed lines)			
C10-H10•••C9	3.255 (6) Å	C10-H10•••C8	3.114 (5) Å
C8-H8•••C9	3.203 (6) Å	C8-H8•••C10	3.094 (5) Å
C9-H9•••C10	3.243 (6) Å	C9-H9•••C8	3.388 (5) Å
C8-H8•••C8	3.341 (5) Å		
S•••H-C interaction between the $\text{PO}_3\text{S}_2\text{C}_{12}\text{H}_9$ moieties of (5) (pink dashed lines)			
C10-H10•••S1	3.1677 (12) Å	C11-H11•••S1	2.9494 (12) Å
C3-H3•••S1	2.9395 (11) Å		

Table SI 3 : Intermolecular: i) C•••H-C interaction between 2.7 to 3.4 Å; ii) S•••H-C interaction between 2.9 to 3.4 Å.

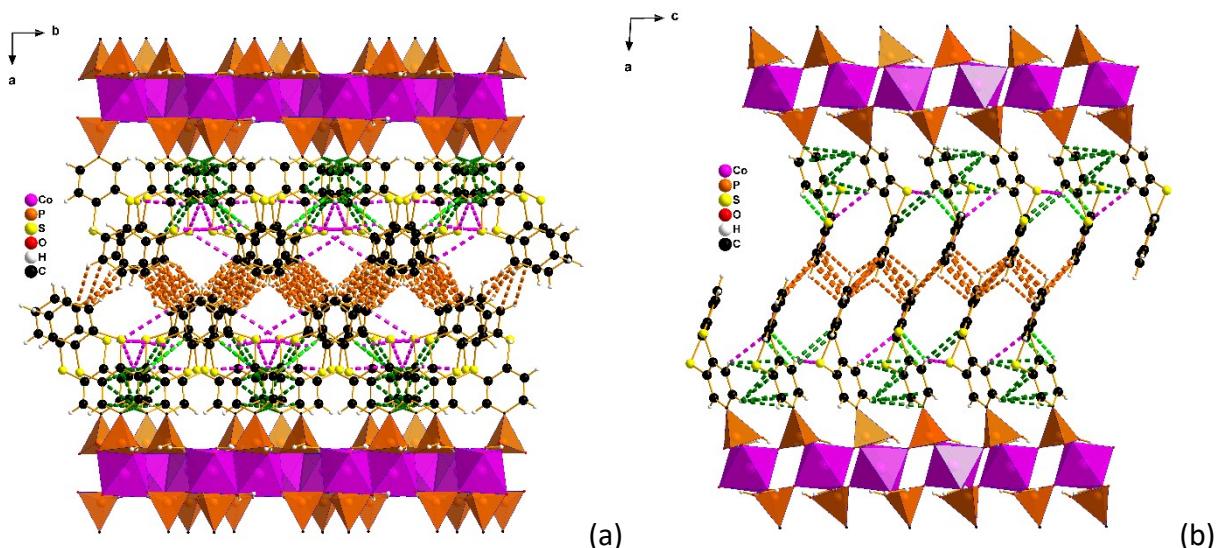


FIGURE SI 9: (a) Representation of the C•••H-C (light and green dashed lines) and S•••H-C (pink dashed lines) interactions in the structure of $\text{Co}(\text{H}_2\text{O})_2(\text{PO}_2\text{OHS}_2\text{C}_{12}\text{H}_7)_2$ (6) (a) view along the c direction; (b) view along the b direction.

C\cdotsH-C interaction between the PO₃S₂C₁₂H₉ moieties of Co(H₂O)₂(PO₂OHS₂C₁₂H₇)₂ (6) in one column (light green dashed lines)			
C20-H20 \cdots C17	3.18(1) Å	C17-H17 \cdots C20	3.20(2) Å
C\cdotsH-C interaction between the PO₃S₂C₁₂H₉ moieties of (6) in two adjacent columns (dark green dashed lines)			
C6-H6 \cdots C13	3.165(8) Å	C6-H6 \cdots C14	3.062(8) Å
C5-H5 \cdots C16	3.001(8) Å	C6-H6 \cdots C18	3.37(1) Å
C6-H6 \cdots C16	3.341(9) Å	C5-H5 \cdots C15	3.240(8) Å
C6-H6 \cdots C15	3.118(8) Å	C17-H17 \cdots C8	3.32(1) Å
C8-H8 \cdots C17	3.36(1) Å		
C\cdotsH-C interaction between two adjacent organic layers of (6) (orange dashed lines)			
C10-H10 \cdots C11	3.36(2) Å	C22-H22 \cdots C9	3.21(2) Å
C11-H11 \cdots C23	3.18(1) Å	C11-H11 \cdots C9	3.23(2) Å
C9-H9 \cdots C11	3.39(1) Å	C11-H11 \cdots C10	3.09(2) Å
C23-H23 \cdots C11	3.19(1) Å	C10-H10 \cdots C21	3.25(2) Å
C23-H23 \cdots C23	2.96(1) Å	C23-H23 \cdots C22	3.14(2) Å
C9-H9 \cdots C22	3.03(2) Å	C9-H9 \cdots C21	3.24(2) Å
C21-H21 \cdots C10	3.05(2) Å	C21-H21 \cdots C9	3.25(2) Å
S\cdotsH-C interaction between the PO₃S₂C₁₂H₉ moieties of (6) (pink dashed lines)			
C17-H17 \cdots S1	3.102(3) Å	C21-H21 \cdots S4	3.317(3) Å
C5-H5 \cdots S4	2.995(3) Å	C8-H8 \cdots S2	3.159(3) Å
C20-H20 \cdots S4	3.207(3) Å		

Table SI 4 : Intermolecular: i) C \cdots H-C interaction between 2.7 to 3.4 Å; ii) S \cdots H-C interaction between 2.9 to 3.4 Å for Co(H₂O)₂(PO₂OHS₂C₁₂H₇)₂ (6).

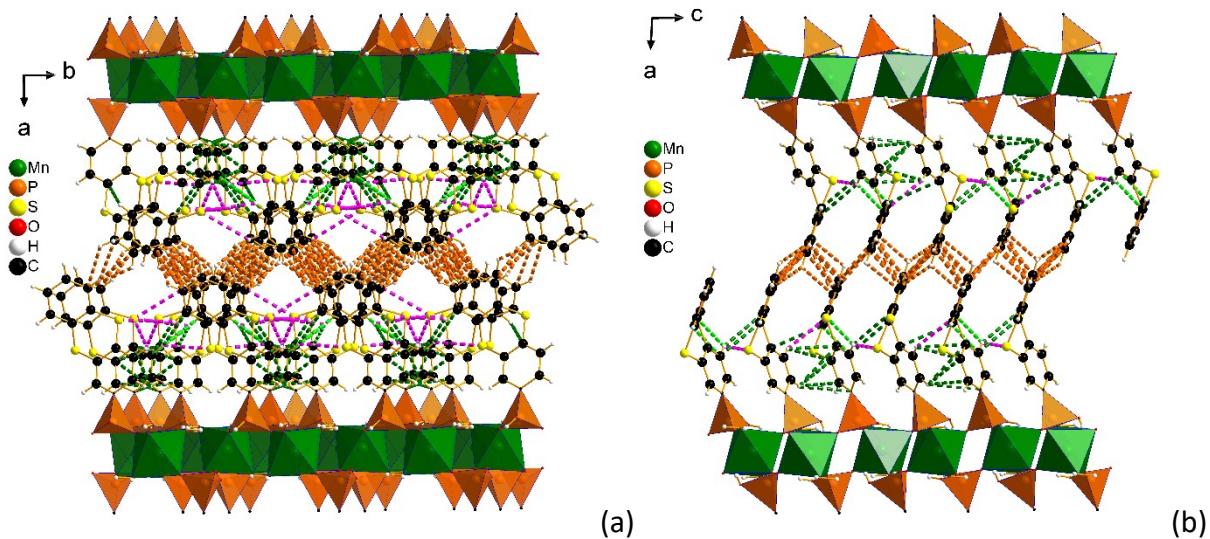


FIGURE SI 10 : (a) Representation of the C \cdots H-C (light and green dashed lines) and S \cdots H-C (pink dashed lines) interactions in the structure of Mn(H₂O)₂(PO₂OHS₂C₁₂H₇)₂ (7) (a) view along the *c* direction; (b) view along the *b* direction.

C\cdotsH-C interaction between the PO₃S₂C₁₂H₉ moieties of Mn(H₂O)₂(PO₂OHS₂C₁₂H₇)₂ (7) in one column (light green dashed lines)			
C14-H14 \cdots C24	3.233(3) Å	C24-H24 \cdots C14	3.247(4) Å
C\cdotsH-C interaction between the PO₃S₂C₁₂H₉ moieties of (7) in two adjacent columns (dark green dashed lines)			
C7-H7 \cdots C19	3.008 (3) Å	C8-H8 \cdots C19	3.369 (3) Å
C14-H14 \cdots C2	3.375 (3) Å	C24-H24 \cdots C9	3.302 (3) Å
C24-H24 \cdots C14	3.247 (4) Å	C7-H7 \cdots C20	3.269 (3) Å
C9-H9 \cdots C24	3.353 (3) Å	C8-H8 \cdots C22	3.208 (2) Å
C\cdotsH-C interaction between two adjacent organic layers of (7) (orange dashed lines)			
C8-H8 \cdots C21	3.105 (3) Å	C8-H8 \cdots C20	3.155 (3) Å
C17-H17 \cdots C17	2.971 (3) Å	C10-H10 \cdots C16	3.035 (4) Å
C15-H15 \cdots C11	3.081 (4) Å	C12-H12 \cdots C11	3.096 (4) Å
C17-H17 \cdots C12	3.163 (3) Å	C17-H17 \cdots C16	3.181 (4) Å
C10-H10 \cdots C15	3.249 (4) Å	C11-H11 \cdots C15	3.277 (4) Å
C15-H15 \cdots C10	3.281 (3) Å	C11-H11 \cdots C12	3.367 (4) Å
C10-H10 \cdots C12	3.391 (3) Å	C16-H16 \cdots C10	3.224(3)
C12-H12 \cdots C10	3.229(4) Å	C12-H12 \cdots C17	3.163 (3) Å
S\cdotsH-C interaction between the PO₃S₂C₁₂H₉ moieties of (7) (pink dashed lines)			
C7-H7 \cdots S4	3.0047 (7) Å	C24-H24 \cdots S1	3.1634 (6) Å
C15-H15 \cdots S4	3.3384 (7) Å	C14-H14 \cdots S4	3.3044 (7) Å
C9-H9 \cdots S2	3.2463 (6) Å		

Table SI 5 : Intermolecular: i) C \cdots H-C interaction between 2.7 to 3.4 Å; ii) S \cdots H-C interaction between 2.9 to 3.4 Å for Mn(H₂O)₂(PO₂OHS₂C₁₂H₇)₂ (7).

Table A1: Tabulated the peak fitting information of Absorbance with Gaussian function

Compounds		Energy (eV)	Wavelength (nm)	Reduced chi ²	FWHM
PO(OH) ₂ S ₂ C ₁₂ H ₇ (3)	AI	4.58	270	1.10765E-4	1.49406
	AII	3.71	333		0.61949
	AIII	3.40	364		0.32972
	AIIV	2.44	506		0.93805
Zn(H ₂ O)(PO ₃ S ₂ C ₁₂ H ₇) (4)	AI	4.66	265	1.52964E-4	1.45058
	AII	3.76	329		0.55431
	AIIV	2.89	429		1.49267
Cu(H ₂ O)(PO ₃ S ₂ C ₁₂ H ₇) (5)	AI	4.72	262	1.12961E-4	1.49879
	AII	3.70	334		0.68241
	AIII	3.07	403		0.72344
	AIIV	1.66	742		0.41236
Co(H ₂ O) ₂ (PO ₂ OHS ₂ C ₁₂ H ₇) ₂ (6)	AI	4.71	262	7.92081E-5	1.18879
	AII	3.72	333		0.79531
	AIII	3.44	360		0.27453
	AIIV	2.75	450		0.6635
	AV	2.34	528		0.5312

$Mn(H_2O)_2(PO_2OHS_2C_{12}H_7)_2$ (7)	AI	4.60	269	3.06447E-5	1.2357
	All	3.80	325		0.62495
	AllIII	3.51	352		0.34333
	AllIV	2.96	418		2.13826

Table A2: Tabulated the peak fitting information of photoluminescence with Gaussian function

Compounds		Energy (eV)	Wavelength (nm)	Reduced chi ²	FWHM
$PO(OH)_2S_2C_{12}H_7$ (3)	LIII	2.92	425	3.91115E-6	0.54649
	LIV	2.47	501		0.29712
	LIII	2.99	414		0.29879
	LIV	2.46	503		0.88208
	LV	1.82	681		0.29341
	LI	3.97	312	4.09451E-9	0.64828
	LIII	2.95	420		0.35654
	LIV	2.39	518		0.53454
	LV	1.97	627		1.30235
$Cu(H_2O)(PO_3S_2C_{12}H_7)$ (5)	LI	3.61	343	2.49327E-9	0.12156
	LII	3.40	364		0.13526
	LIII	2.98	415		0.42864
	LIV	2.43	510		0.41104
	LV	2.00	610		0.42307
$Co(H_2O)_2(PO_2OHS_2C_{12}H_7)_2$ (6)	LIII	2.98	414	3.42801E-8	0.29525
	LIV	2.46	502		0.92081
	LV	1.93	642		0.4988

Table SI 6 : Absorbance (Table A1) and photoluminescence (Table A2) peak fitting information with a Gaussian function.

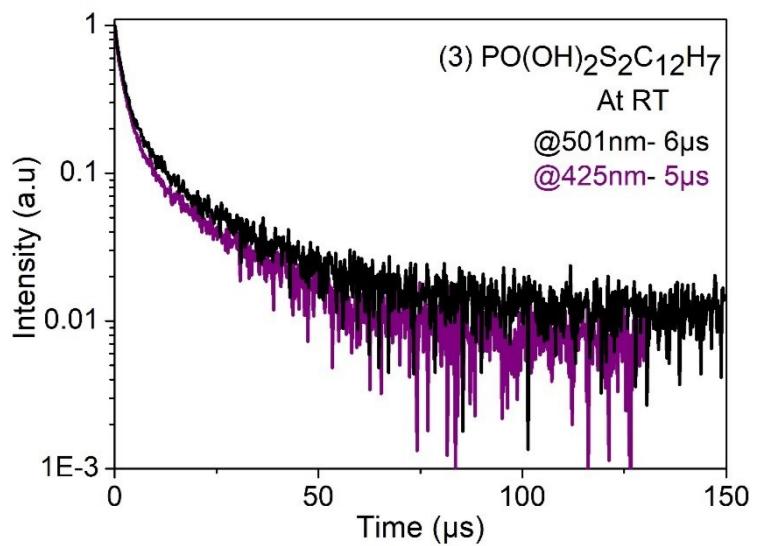


FIGURE SI 11: Thianthrene phosphonic acid $\text{PO(OH)}_2\text{S}_2\text{C}_{12}\text{H}_7$ (**3**) decay time at 425 nm (**LIII**) (black) and 501 nm (**LIV**) (Purple) under 266 nm excitation wavelength performed at room temperature.