

# Supplementary Information

to

## **Novel Stable Ytterbium Acetylacetonate-Quinaldinate Complexes as Single-Molecule Magnets and Surprisingly Efficient Luminophores**

by

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**Table S1.** Crystal data and structure refinement for **1**, **2**, **2\_1**.

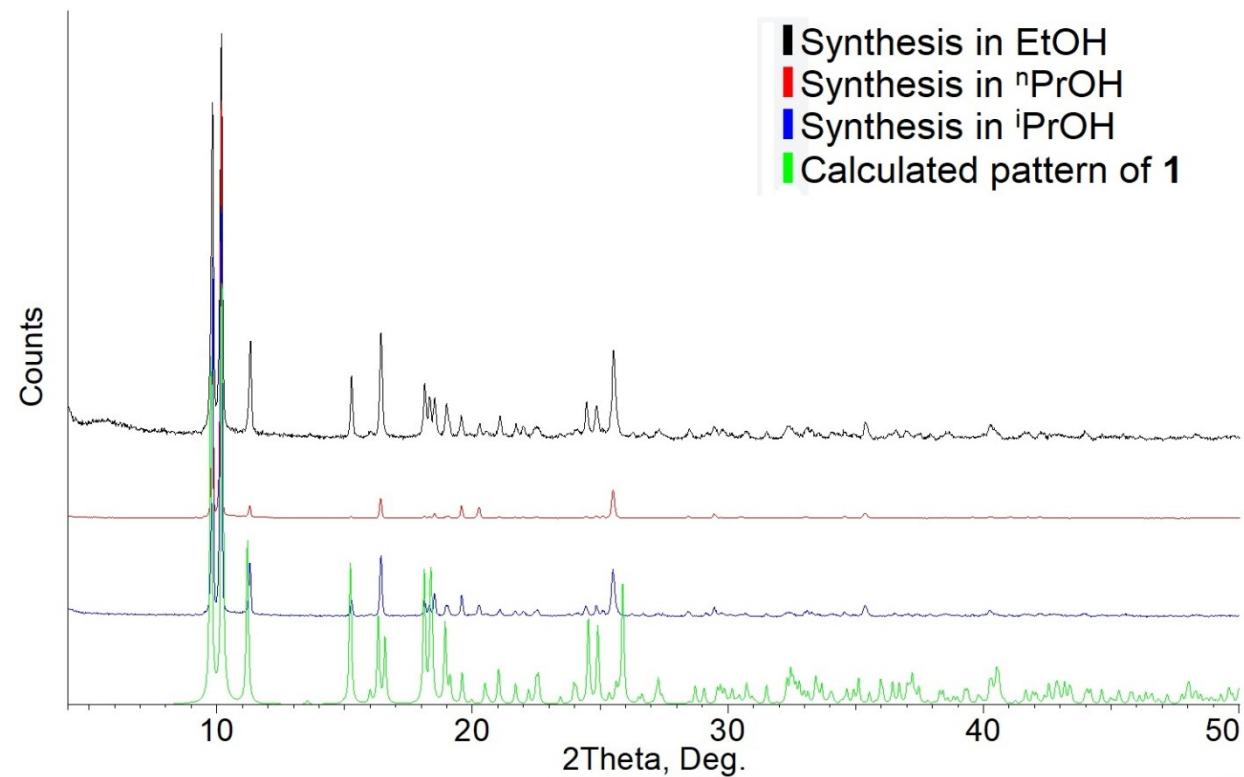
Identification code	<b>1</b>	<b>2</b>	<b>2_1</b>
Empirical formula	C <sub>20</sub> H <sub>20</sub> NO <sub>6</sub> Yb	C <sub>32</sub> H <sub>28</sub> N <sub>3</sub> O <sub>6</sub> Yb	C <sub>35</sub> H <sub>37</sub> N <sub>3</sub> O <sub>7.50</sub> Yb
Formula weight	543.41	723.61	792.71
Temperature, K	100(2)	150(2)	100(2)
Wavelength, Å	0.71073	0.71073	0.71073
Crystal system	Monoclinic	Triclinic	Triclinic
Space group	P2 <sub>1</sub> /n	P-1	P-1
a, Å	9.3312(2)	10.1267(4)	11.9925(4)
b, Å	11.6285(3)	12.5770(5)	12.1587(4)
c, Å	18.2162(5)	12.5907(5)	26.1155(8)
α, °	90	83.2460(10)	101.6180(10)
β, °	96.5070(10)	71.9510(10)	92.0670(10)
γ, °	90	67.4820(10)	118.4930(10)
Volume, Å <sup>3</sup>	1963.87(9)	1408.44(10)	3239.33(18)
Z	4	2	4
D (calc), Mg/m <sup>3</sup>	1.838	1.706	1.625
μ, mm <sup>-1</sup>	4.798	3.371	2.942
F(000)	1060	718	1592
Crystal size, mm	0.28 x 0.20 x 0.18	0.28 x 0.22 x 0.20	0.20 x 0.12 x 0.04
θ range, °	2.082, 36.130	2.274, 30.138	2.057, 33.159
Index ranges	-14<=h<=14 -17<=k<=18 -28<=l<=28	-13<=h<=14 -16<=k<=17 -17<=l<=17	-18<=h<=18 -18<=k<=18 -37<=l<=40
Reflections collected	43043	30295	97701
Independent reflections, Rint	8524, 0.0364	7753, 0.0561	23599, 0.0284
Completeness to θ = 25.242°	99.9 %	99.9 %	99.8 %
Absorption correction	Semi-empirical from equivalents	Semi-empirical from equivalents	Semi-empirical from equivalents
Max., min. transmission	0.5675, 0.4024	0.746, 0.6579	0.7465, 0.6129
Refinement method	Full-matrix least-squares on F <sup>2</sup>	Full-matrix least-squares on F <sup>2</sup>	Full-matrix least-squares on F <sup>2</sup>
Data / restraints / parameters	8524 / 0 / 253	7753 / 0 / 383	23599 / 0 / 849
Goodness-of-fit	1.037	1.077	1.013
R1, wR2 [I>2sigma(I)]	0.0269, 0.0455	0.0348, 0.0533	0.0226, 0.0478

R1, wR2 (all data)	0.0354, 0.0474	0.0478, 0.0582	0.0297, 0.0498
Largest diff. peak and hole, e.Å <sup>-3</sup>	0.959, -2.147	1.442, -1.378	0.850, -0.954
CCDC	2292786	2292787	2292788

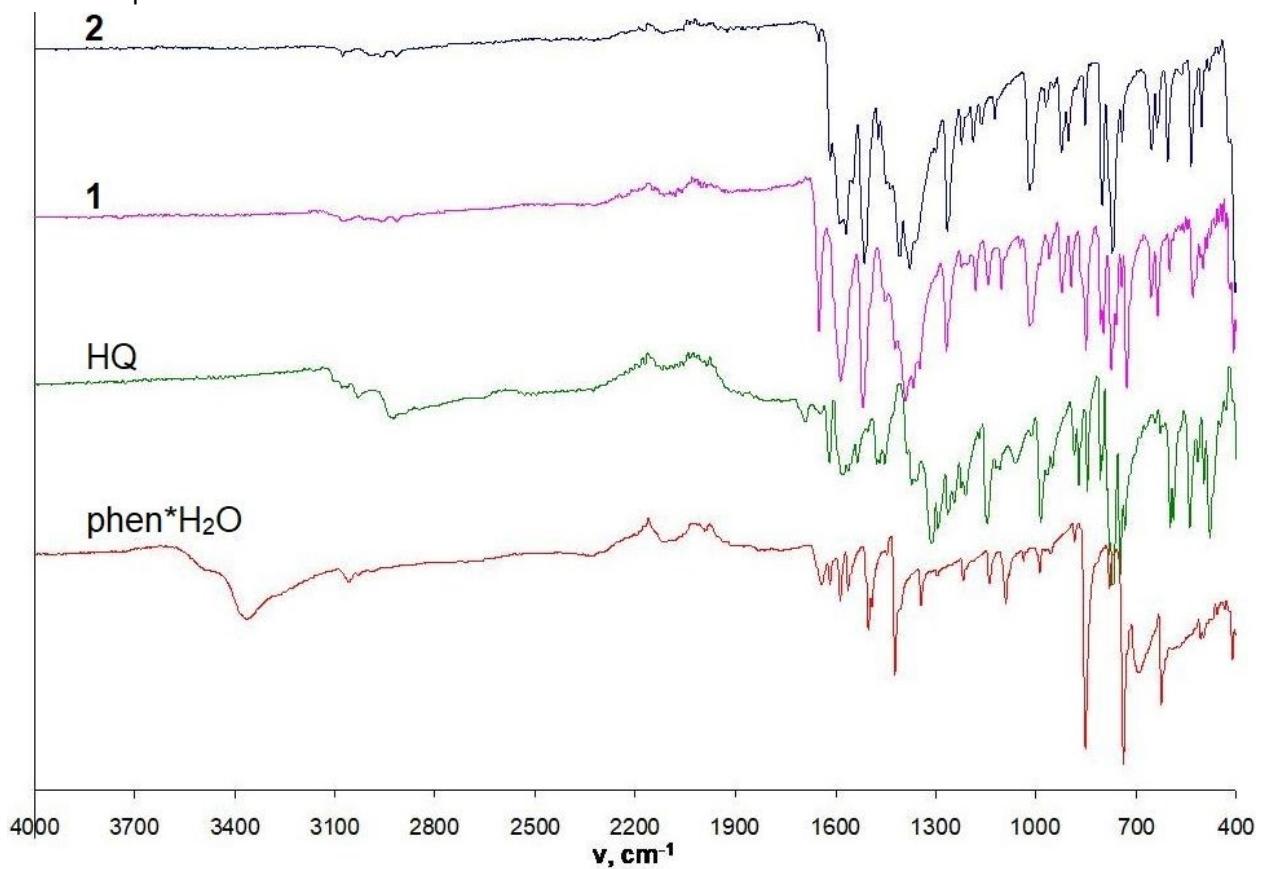
**TableS2.** Fragment of .prp file for the single crystal of **2\_1**.

SPACE GROUP DETERMINATION

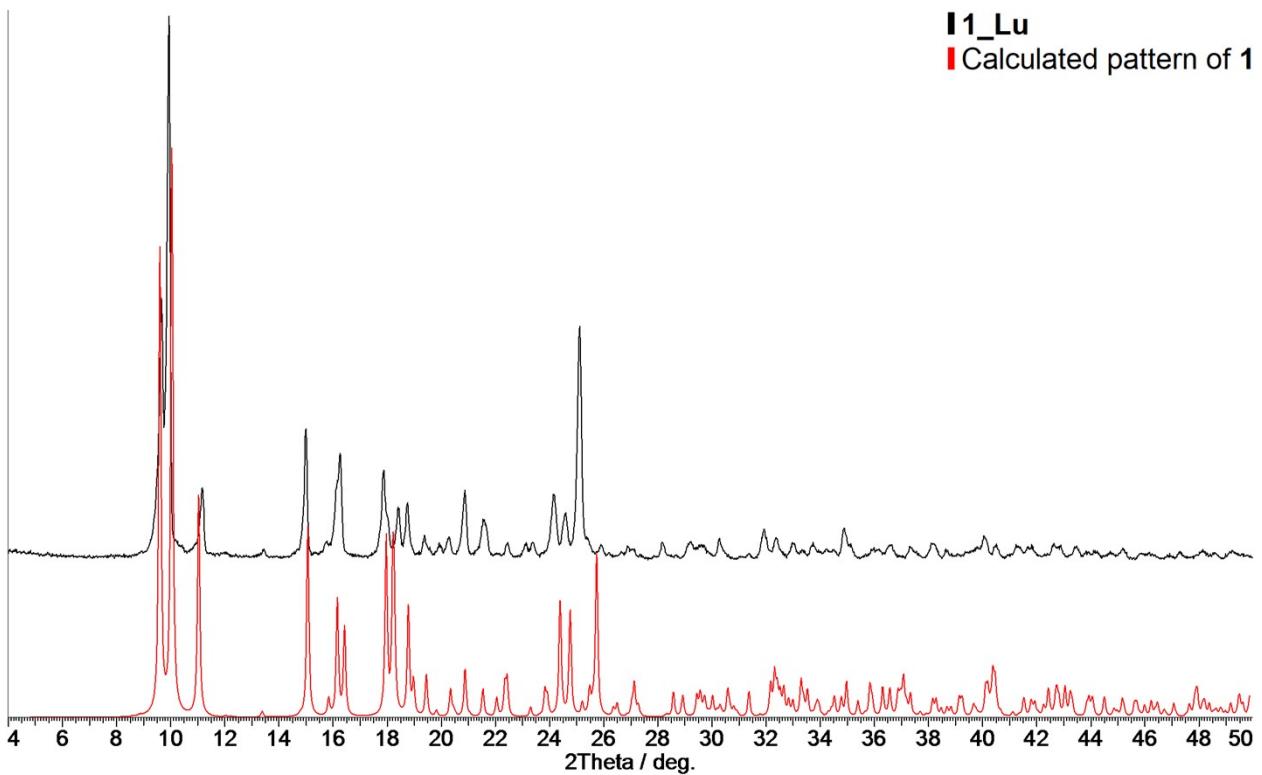
Lattice exceptions:	P	A	B	C	I	F	Obv	Rev	All
N (total) =	0	48804	49064	48886	48872	73377	65126	65148	97728
N (int>3sigma) =	0	38960	34978	39280	39183	56609	52235	52138	78149
Mean intensity =	0.0	32.5	11.4	32.4	32.7	25.4	32.8	32.6	32.6
Mean int/sigma =	0.0	15.0	10.2	15.0	15.0	13.4	15.0	15.0	15.0



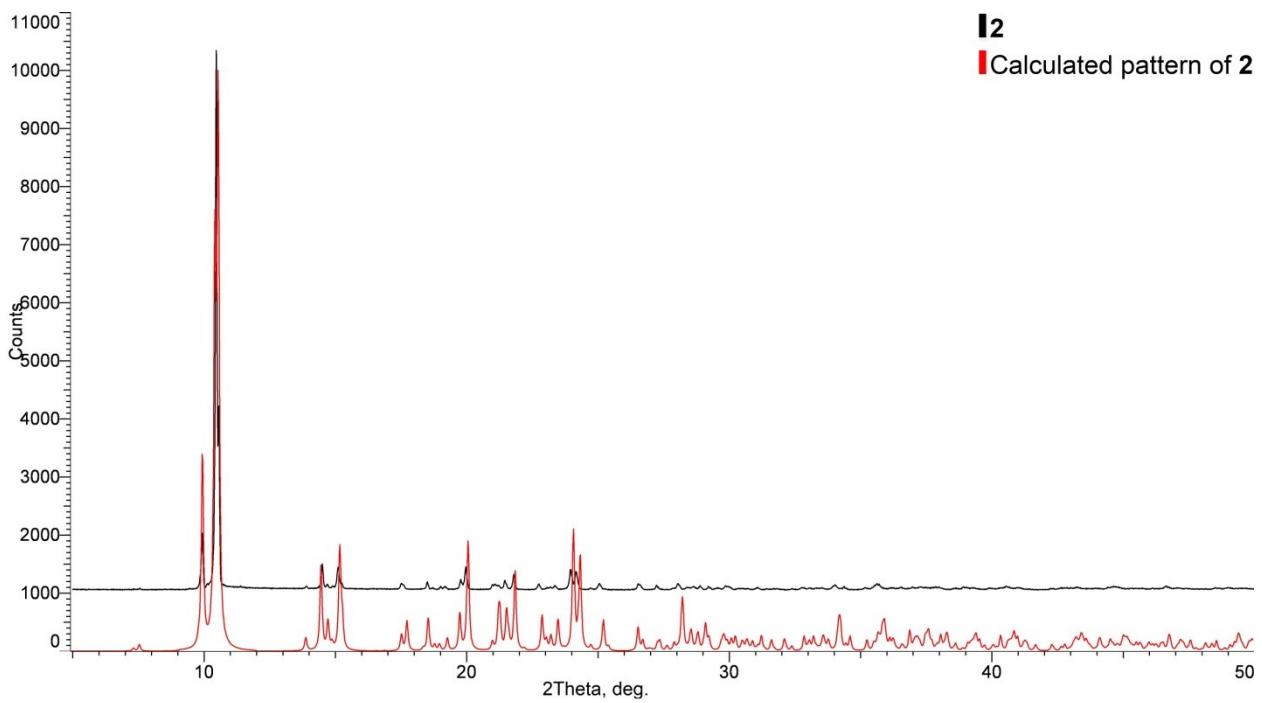
**Fig. S1.** Powder XRD patterns of products of syntheses of **1** in various alcohols compared to the calculated pattern of **1** structure.



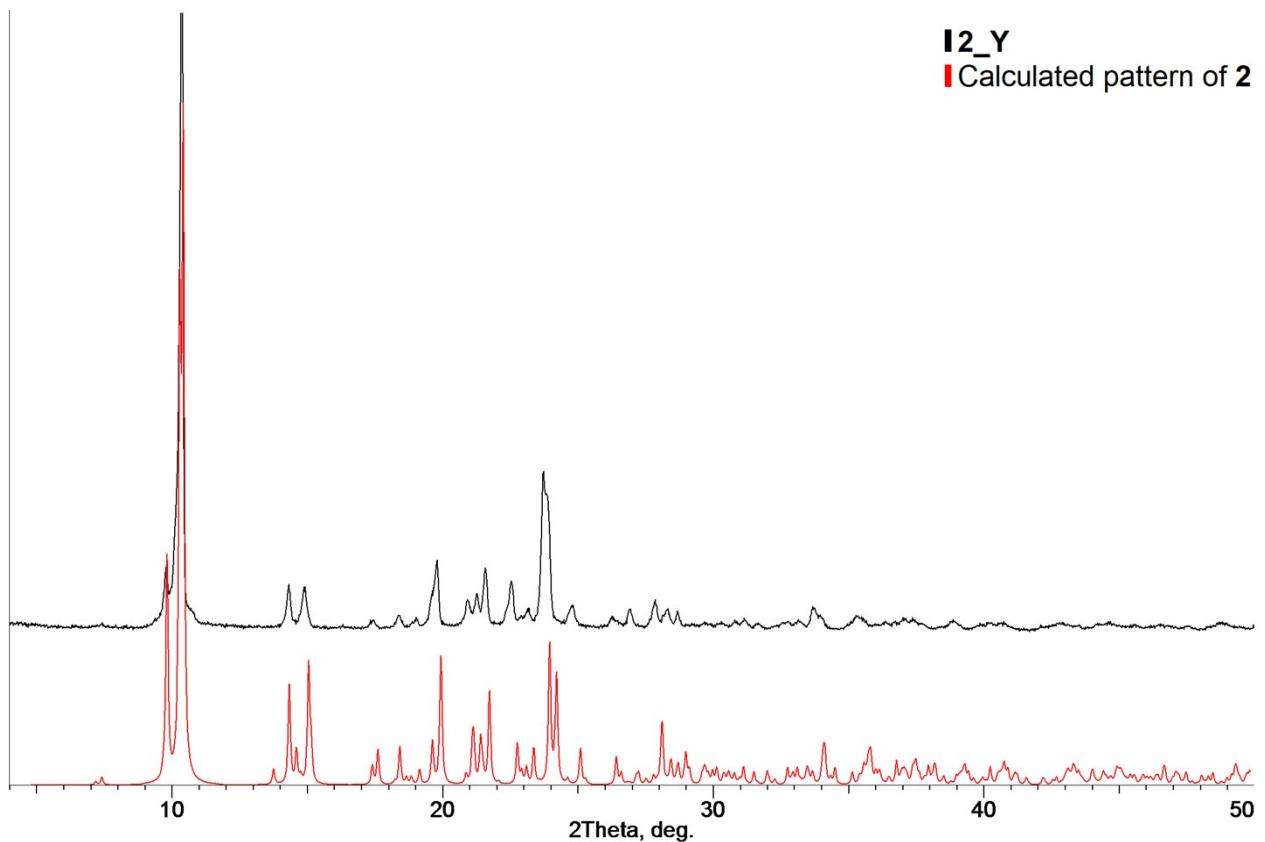
**Fig. S2.** IR spectra of complexes **1** and **2** compared with those of HQ and Phen $\cdot$ H<sub>2</sub>O.



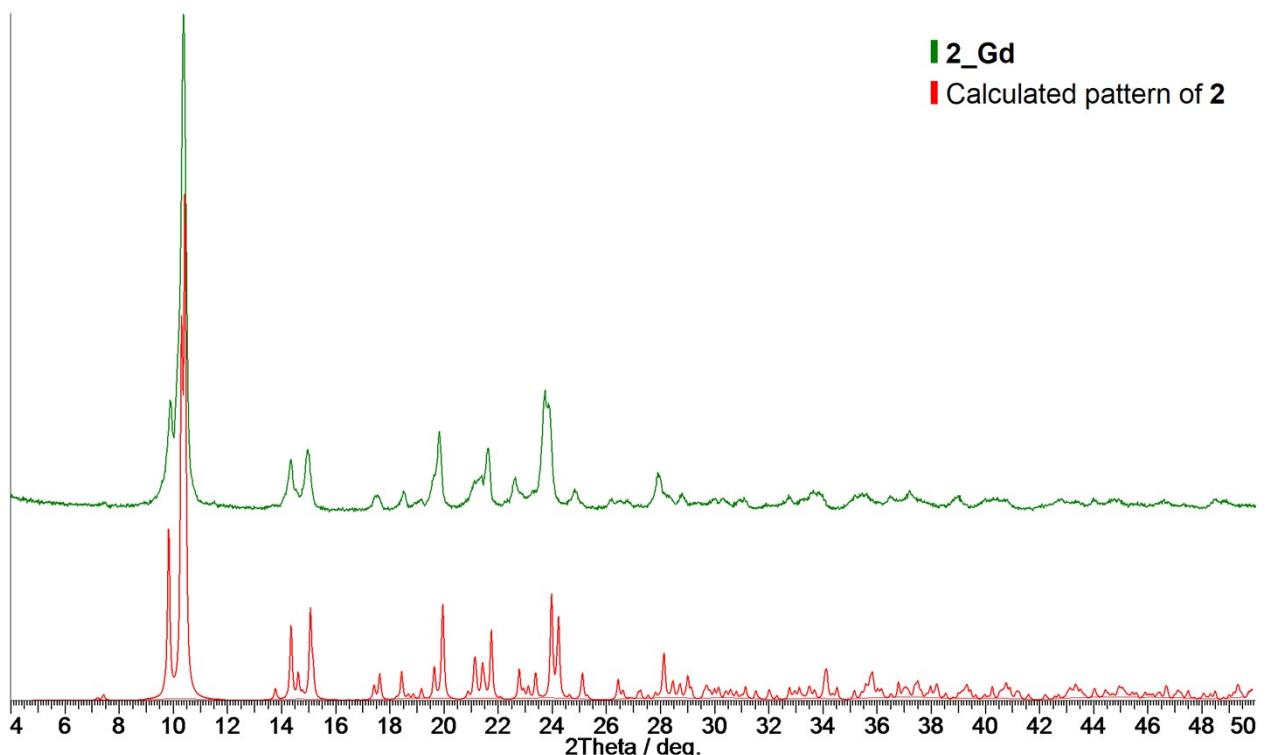
**Fig. S3.** Powder XRD pattern of the product of the synthesis of **1\_Lu** compared to calculated pattern of **1** structure.



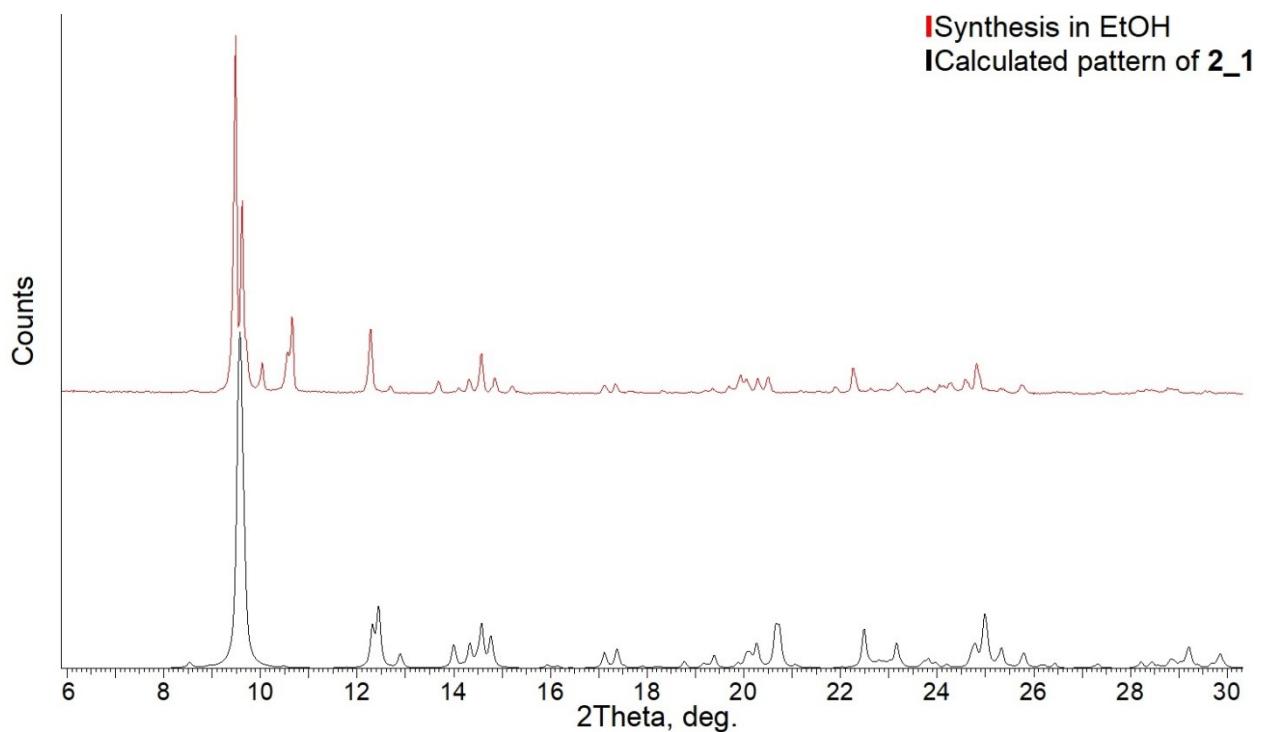
**Fig. S4.** Powder XRD pattern of the product of the synthesis of **2** compared to calculated pattern of **2** structure.



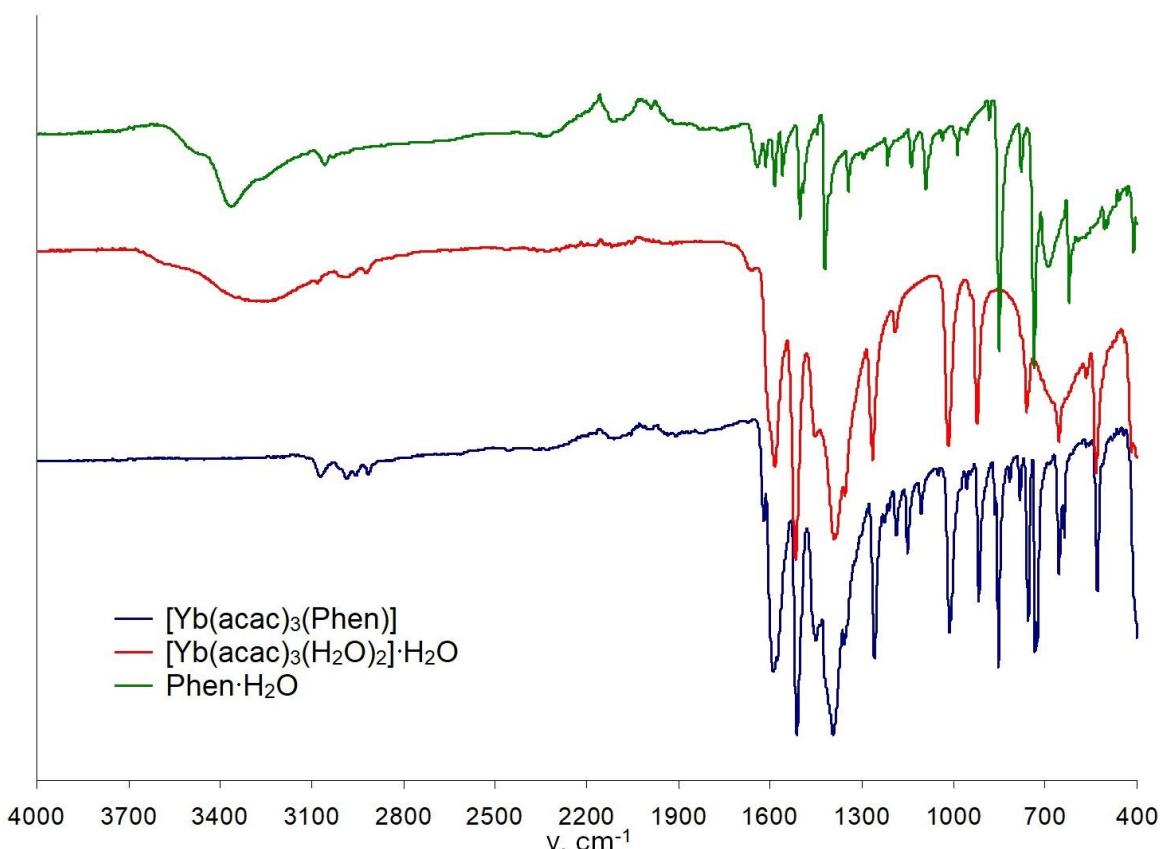
**Fig. S5.** Powder XRD pattern of the product of the synthesis of **2\_Y** compared to calculated pattern of **2** structure.



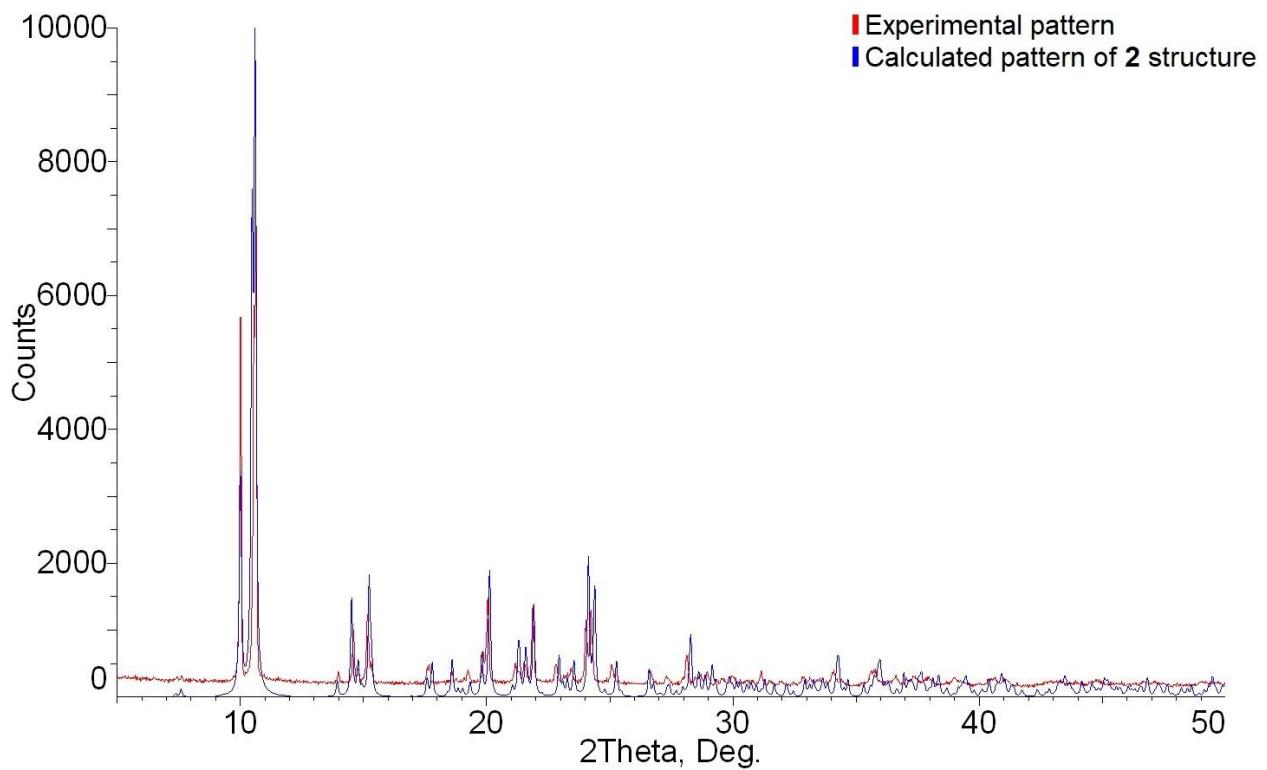
**Fig. S6.** Powder XRD pattern of the product of the synthesis of **2\_Gd** compared to calculated pattern of **2** structure.



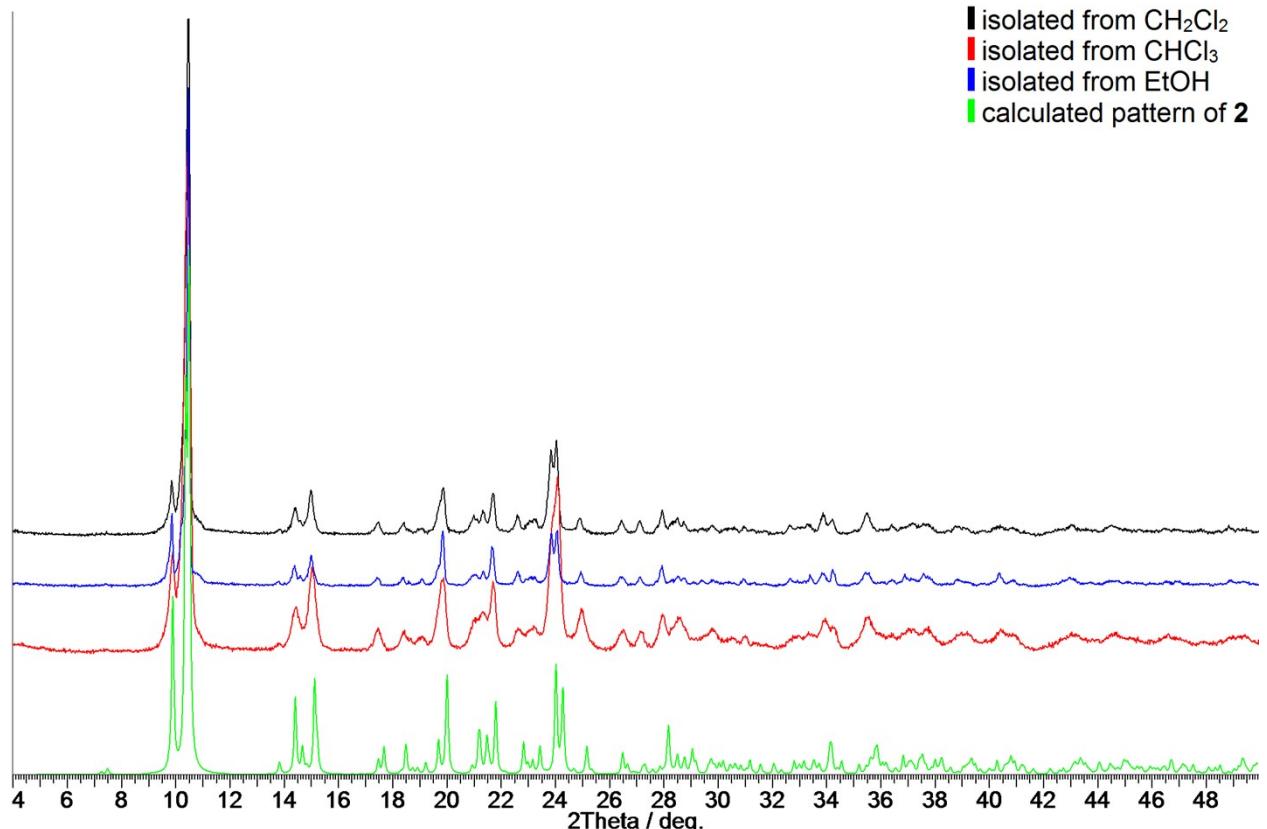
**Fig. S7.** Powder XRD pattern of the product of the synthesis of **2\_1** from EtOH compared to calculated pattern of **2\_1** structure.



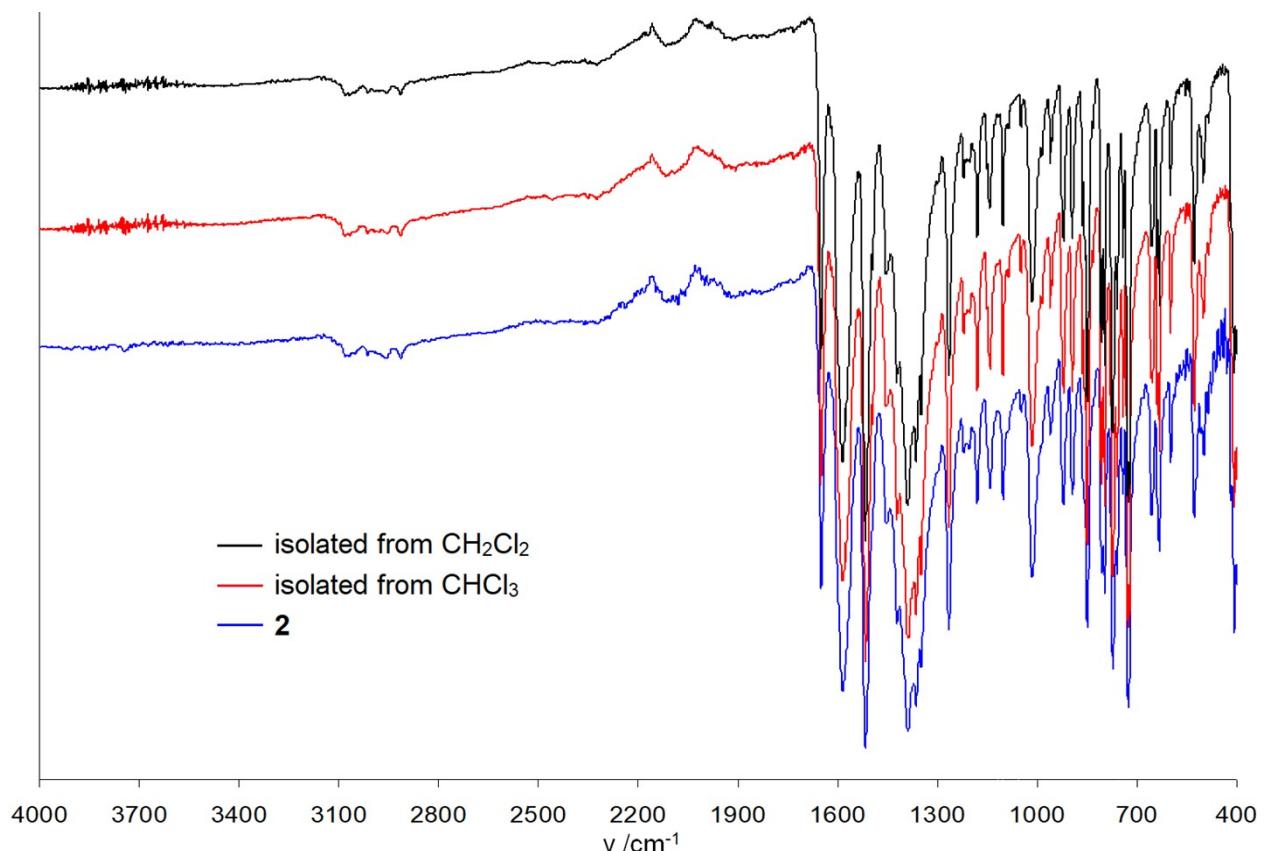
**Fig. S8.** IR spectrum of the intermediate product precipitated during the alternative synthesis of **2** (blue line, see p. 3.1. of the main text) compared with the spectra of  $\text{Yb}(\text{acac})_3 \cdot 3\text{H}_2\text{O}$  (red line) and  $\text{Phen} \cdot \text{H}_2\text{O}$  (green line).



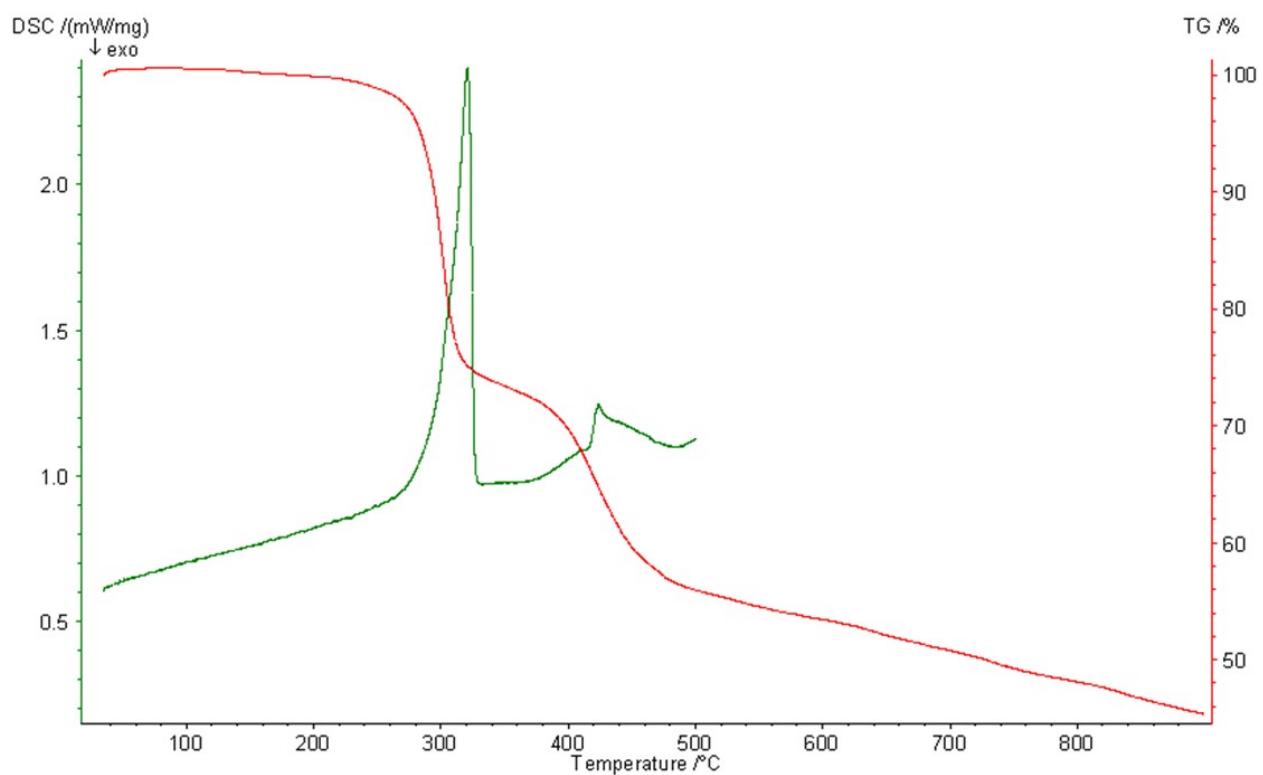
**Fig. S9.** Powder XRD pattern of the product isolated from the alternative synthesis of **2**(see p. 3.1. of the main text) compared to calculated pattern of **2**structure.



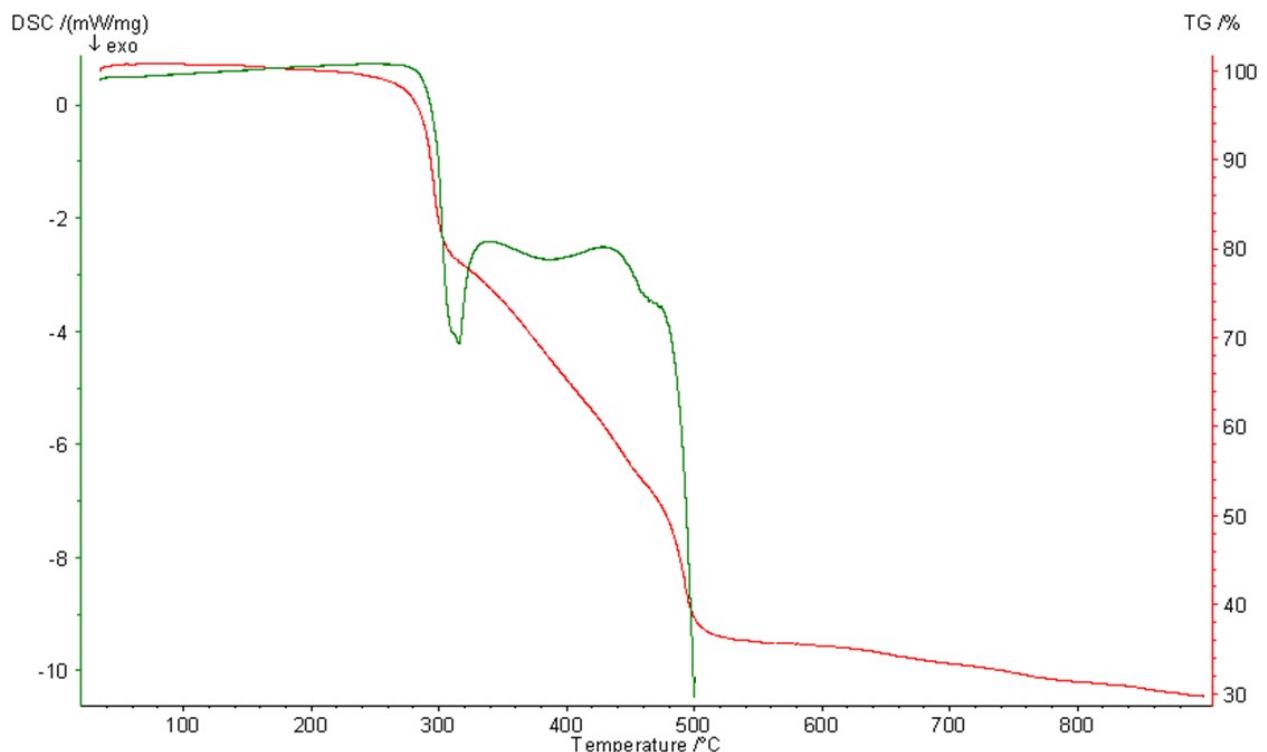
**Fig. S10.** Powder XRD patterns of the products isolated from the solutions of **2** in CH<sub>2</sub>Cl<sub>2</sub>, CHCl<sub>3</sub>and EtOH compared to calculated pattern of **2** structure.



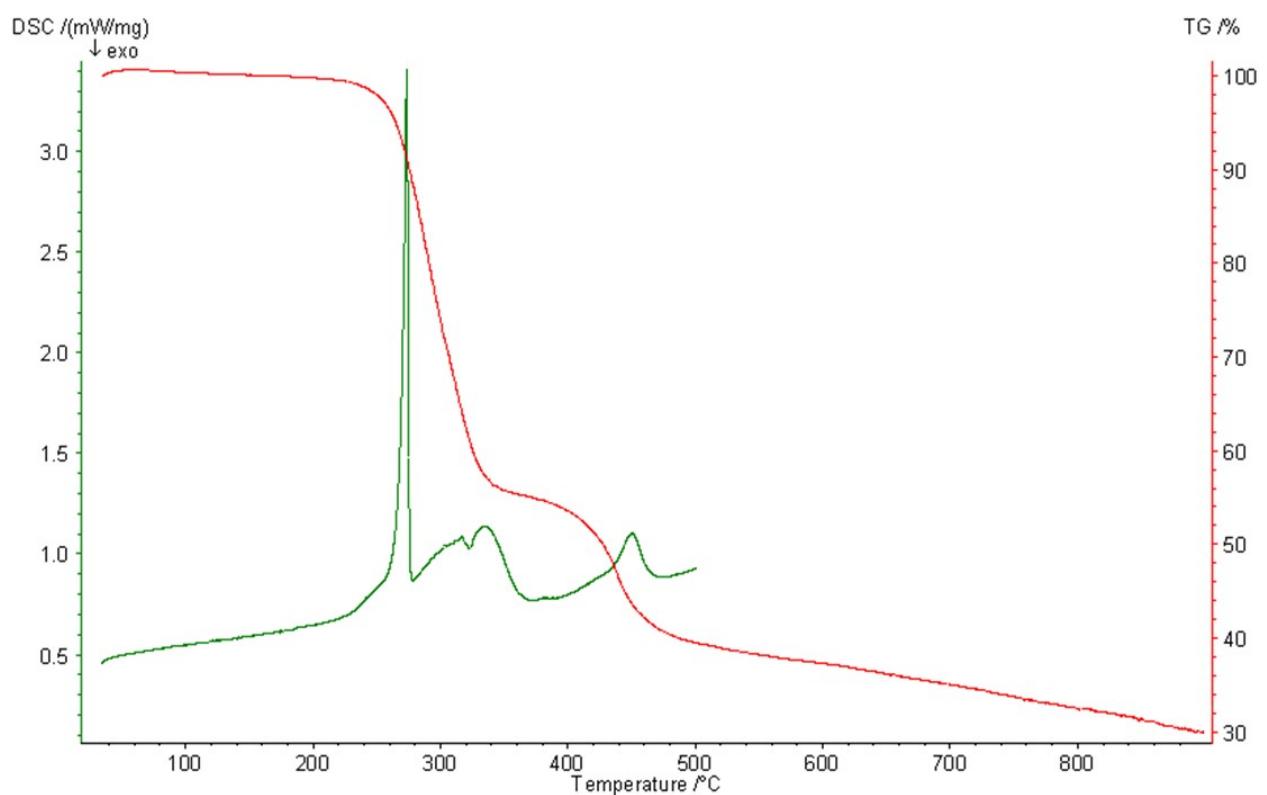
**Fig. S11.** IR spectra of the products isolated from the solutions of **2** in  $\text{CH}_2\text{Cl}_2$  and  $\text{CHCl}_3$ .



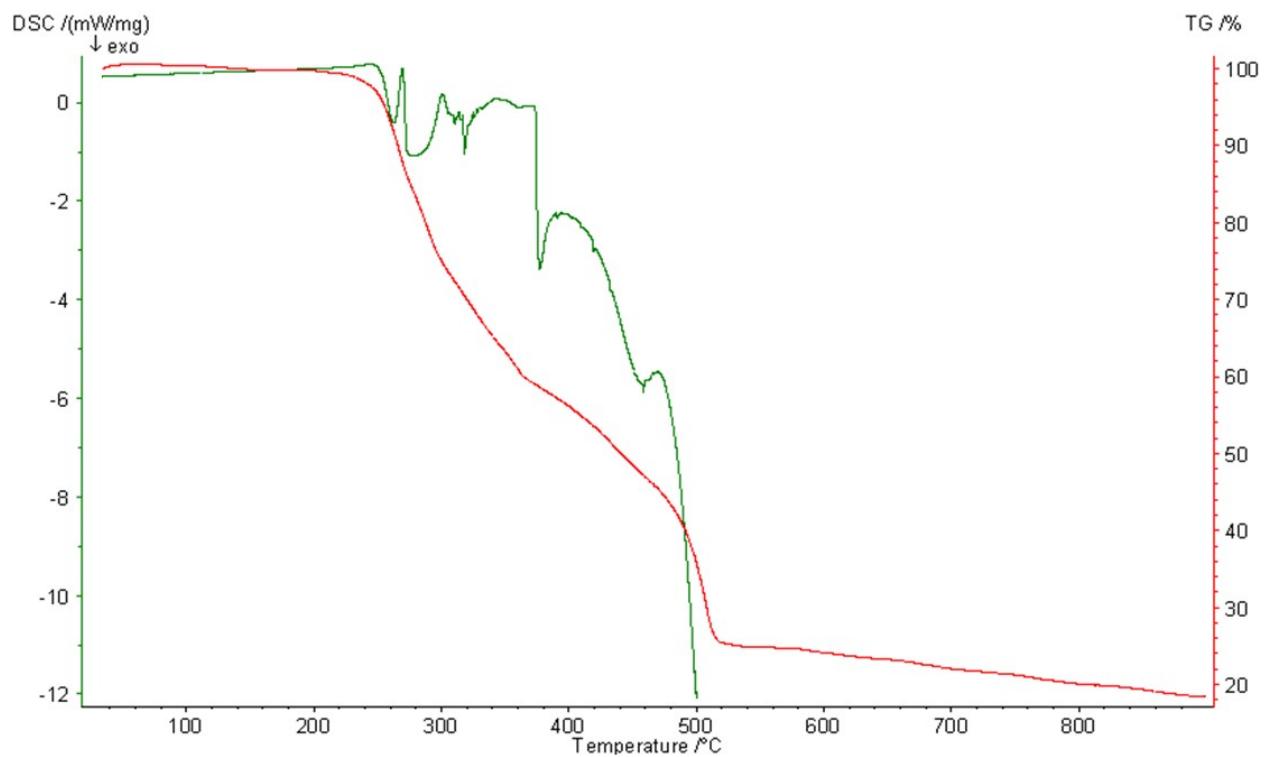
**Fig. S12.** TG (red) and DSC (green) curves of complex **1** on heating under Ar flow.



**Fig. S13.** TG (red) and DSC (green) curves of complex **1** on heating under artificial air flow.



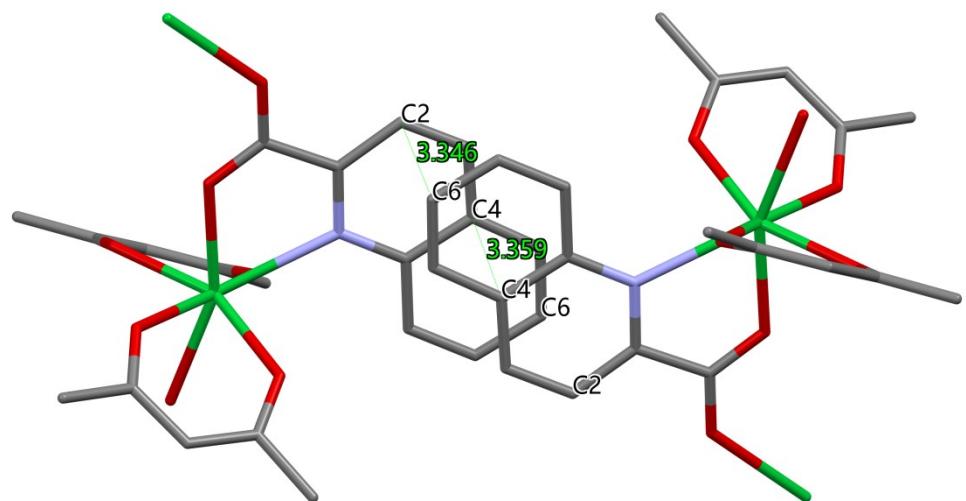
**Fig. S14.** TG (red) and DSC (green) curves of complex **2** on heating under Ar flow.



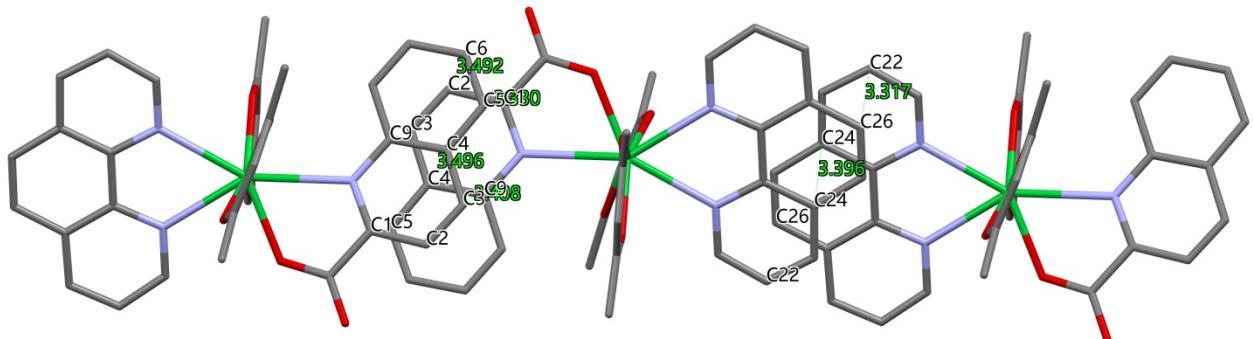
**Fig. S15.** TG (red) and DSC (green) curves of complex **2** on heating under artificial air flow.

**TableS3.** Shortest C...C contacts ( $\text{\AA}$ ) in stacking interactions within **1**, **2** and **2\_1** structures.

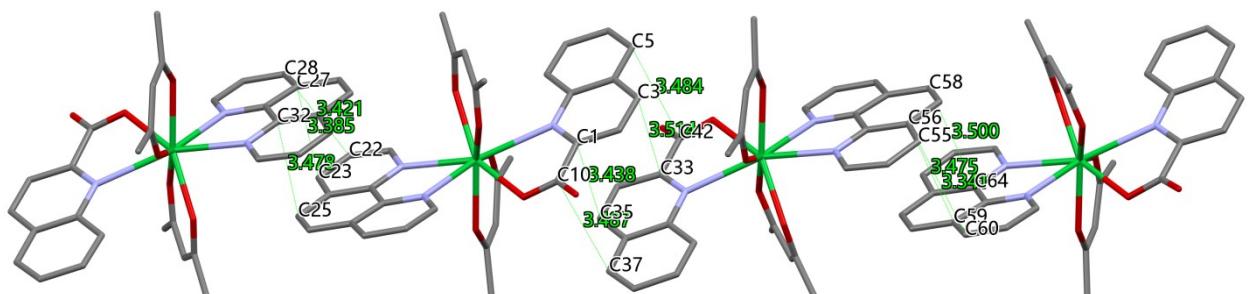
<b>1</b>	
C(2)...C(6) (-x, -y+1, -z)	3.346
C(4)...C(4) (-x, -y+1, -z)	3.359
<b>2</b>	
C(1)...C(5) (-x, -y, -z+1)	3.331
C(2)...C(6) (-x, -y, -z+1)	3.492
C(3)...C(9) (-x, -y, -z+1)	3.407
C(4)...C(4) (-x, -y, -z+1)	3.496
C(22)...C(26) (-x+1, -y+1, -z)	3.311
C(24)...C(24) (-x+1, -y+1, -z)	3.395
<b>2_1</b>	
C(1)...C(35)	3.438
C(3)...C(33)	3.514
C(5)...C(42)	3.484
C(10)...C(37)	3.487
C(22)...C(28) (-x, -y, -z+1)	3.421
C(23)...C(27) (-x, -y, -z+1)	3.385
C(23)...C(28) (-x, -y, -z+1)	3.410
C(25)...C(32) (-x, -y, -z+1)	3.478
C(55)...C(59) (-x+3, -y+2, -z+2)	3.479
C(55)...C(60) (-x+3, -y+2, -z+2)	3.341
C(56)...C(59) (-x+3, -y+2, -z+2)	3.475
C(58)...C(64) (-x+3, -y+2, -z+2)	3.500



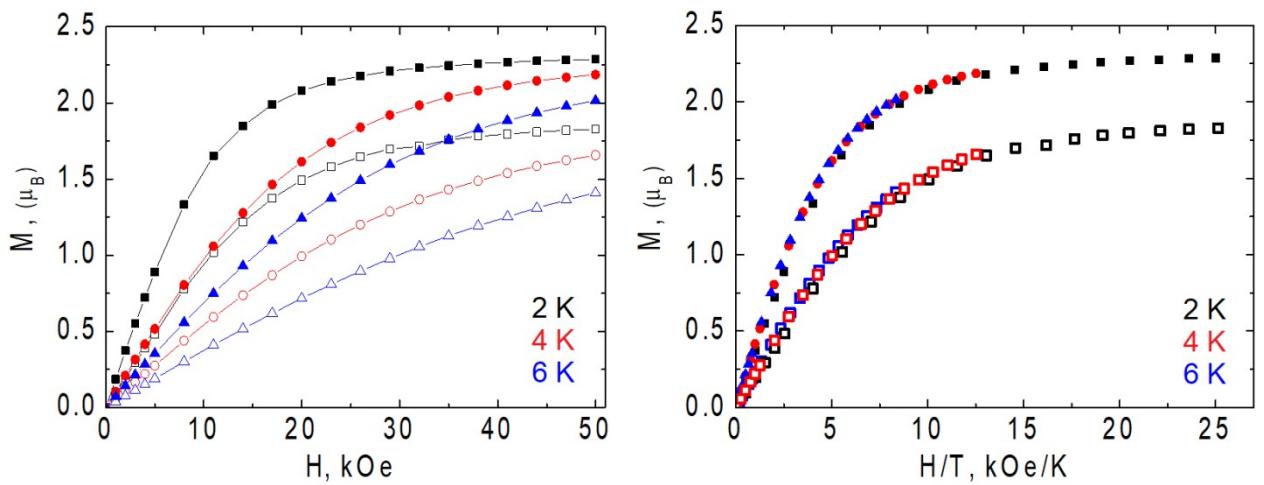
**Fig.S16.** Stacking interactions in the structure **1**.



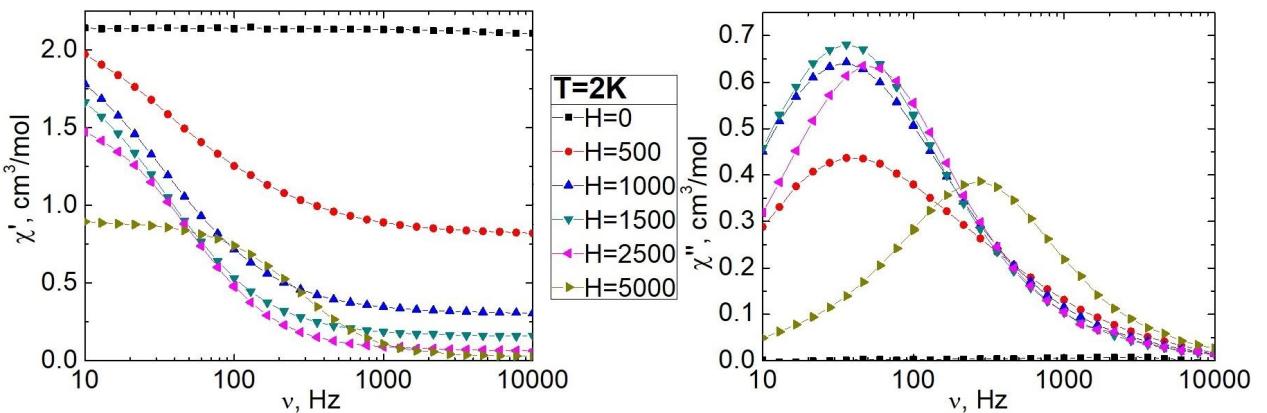
**Fig.S17.** Stacking interactions in the structure **2**.



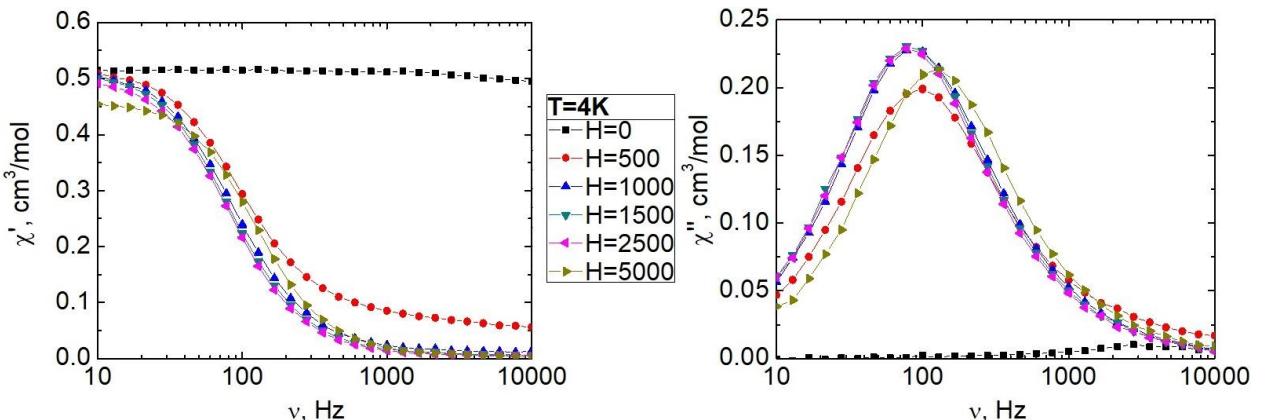
**Fig.S18.** Stacking interactions in the structure **2\_1**.



**Fig. S19.** Low-temperature magnetization vs.  $H$  (left) and vs.  $HT^{-1}$  (right) for complex **1** (per one  $\text{Yb}^{3+}$  ion, empty symbols) and **2** (filled symbols).

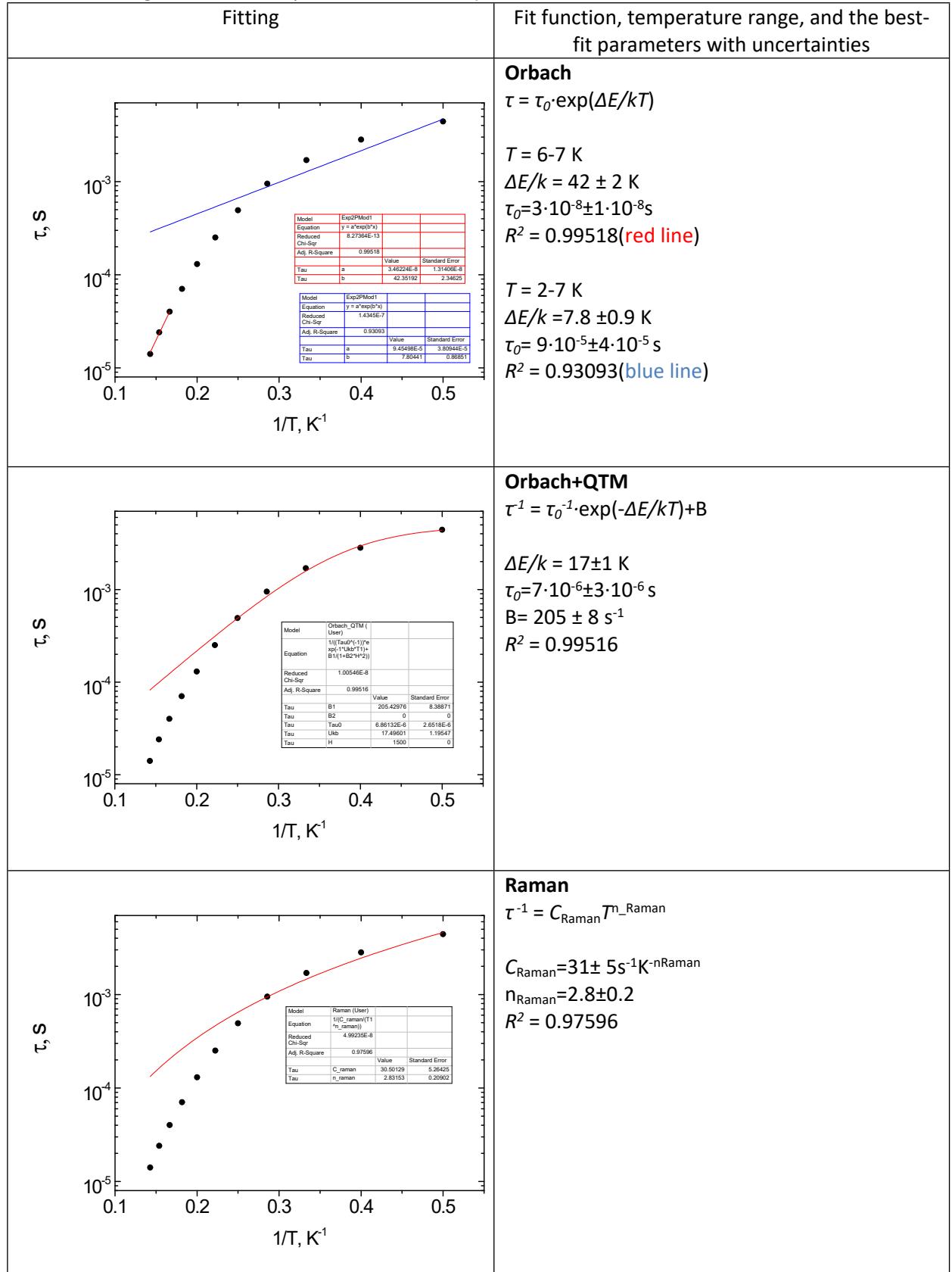


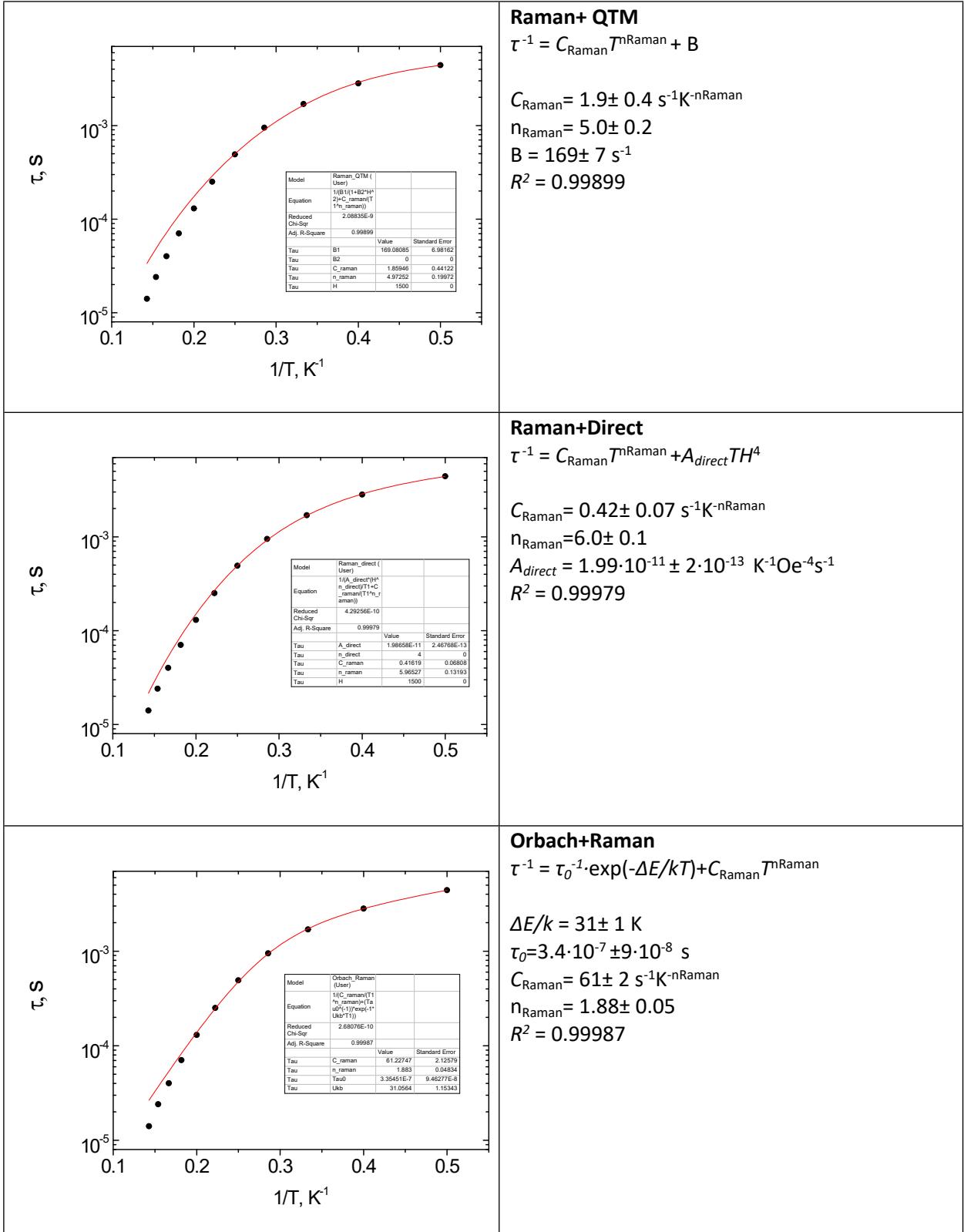
**Fig. S20.** Frequency dependencies of the in-phase ( $\chi'$ ) and out-of-phase ( $\chi''$ ) components of the dynamic magnetic susceptibility of **1** at varied strength of the external static magnetic field and  $T=2\text{ K}$ . Solid lines are visual guides.

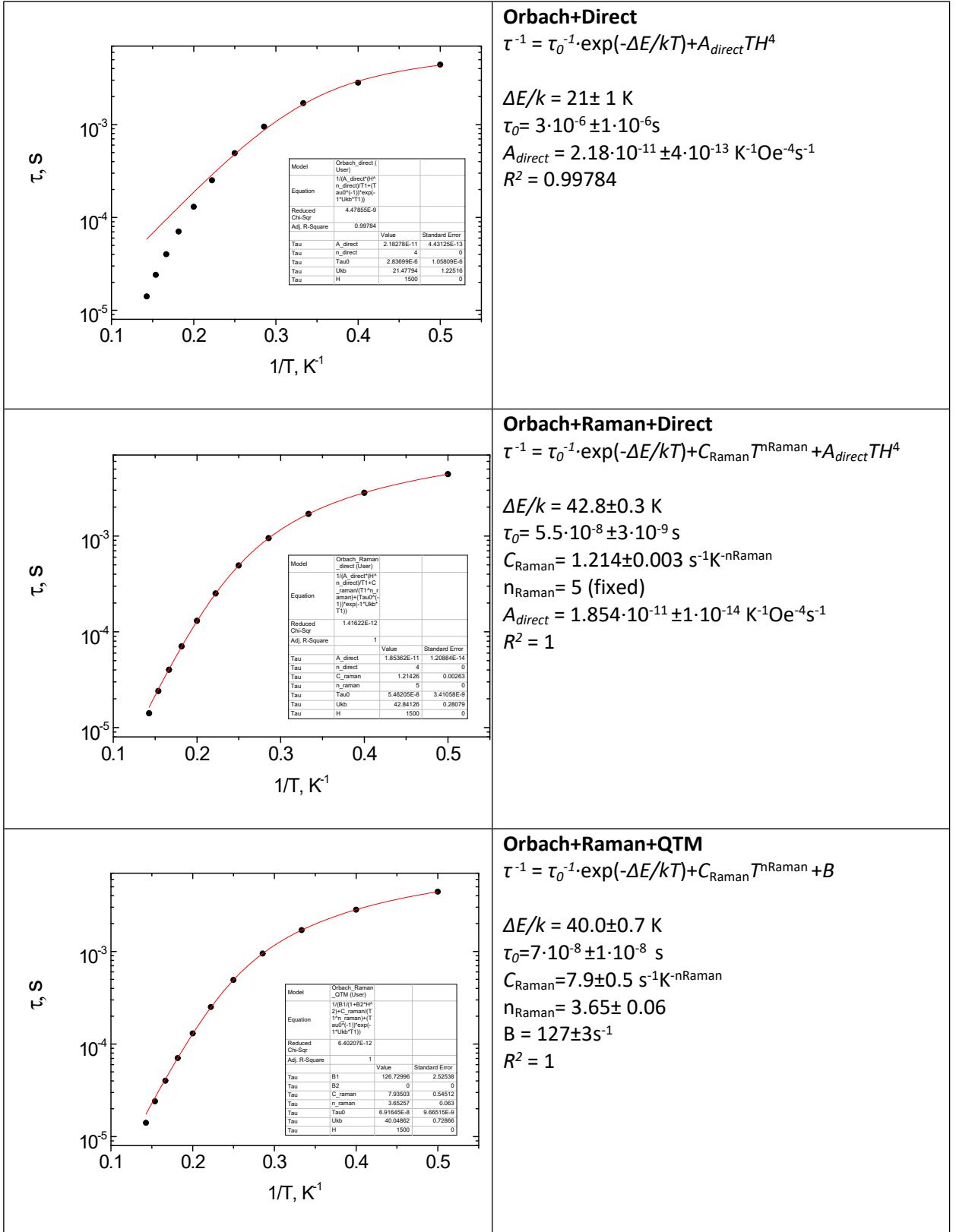


**Fig. S21.** Frequency dependencies of the in-phase ( $\chi'$ ) and out-of-phase ( $\chi''$ ) components of the dynamic magnetic susceptibility of **2** at varied strength of the external static magnetic field and  $T=4\text{ K}$ . Solid lines are visual guides.

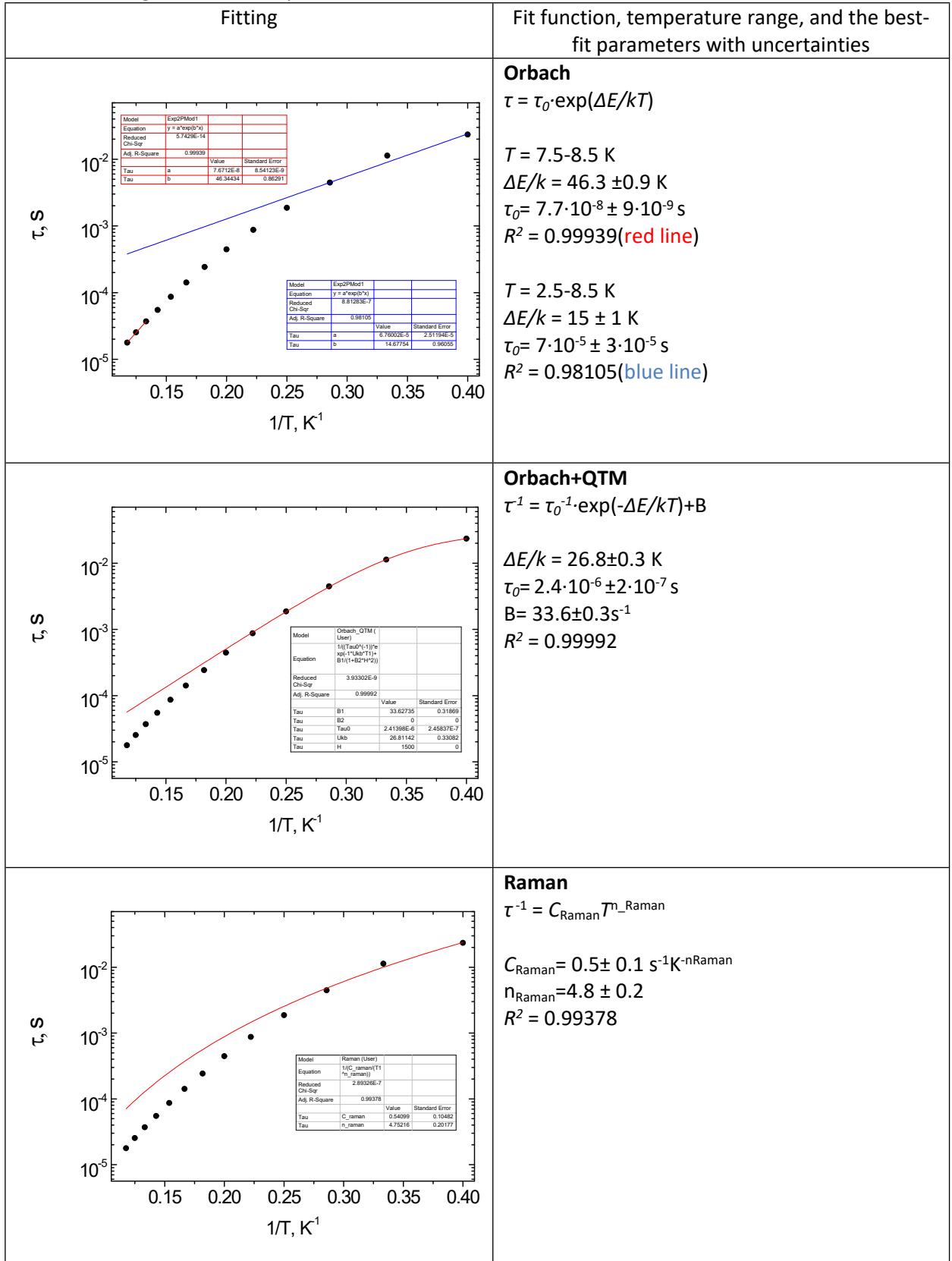
**Table S4.** Fitting of the  $\tau$  vs.  $T$  dependences for **1** ( $H_{DC} = 1.5$  kOe,  $T = 2\text{-}7$  K).

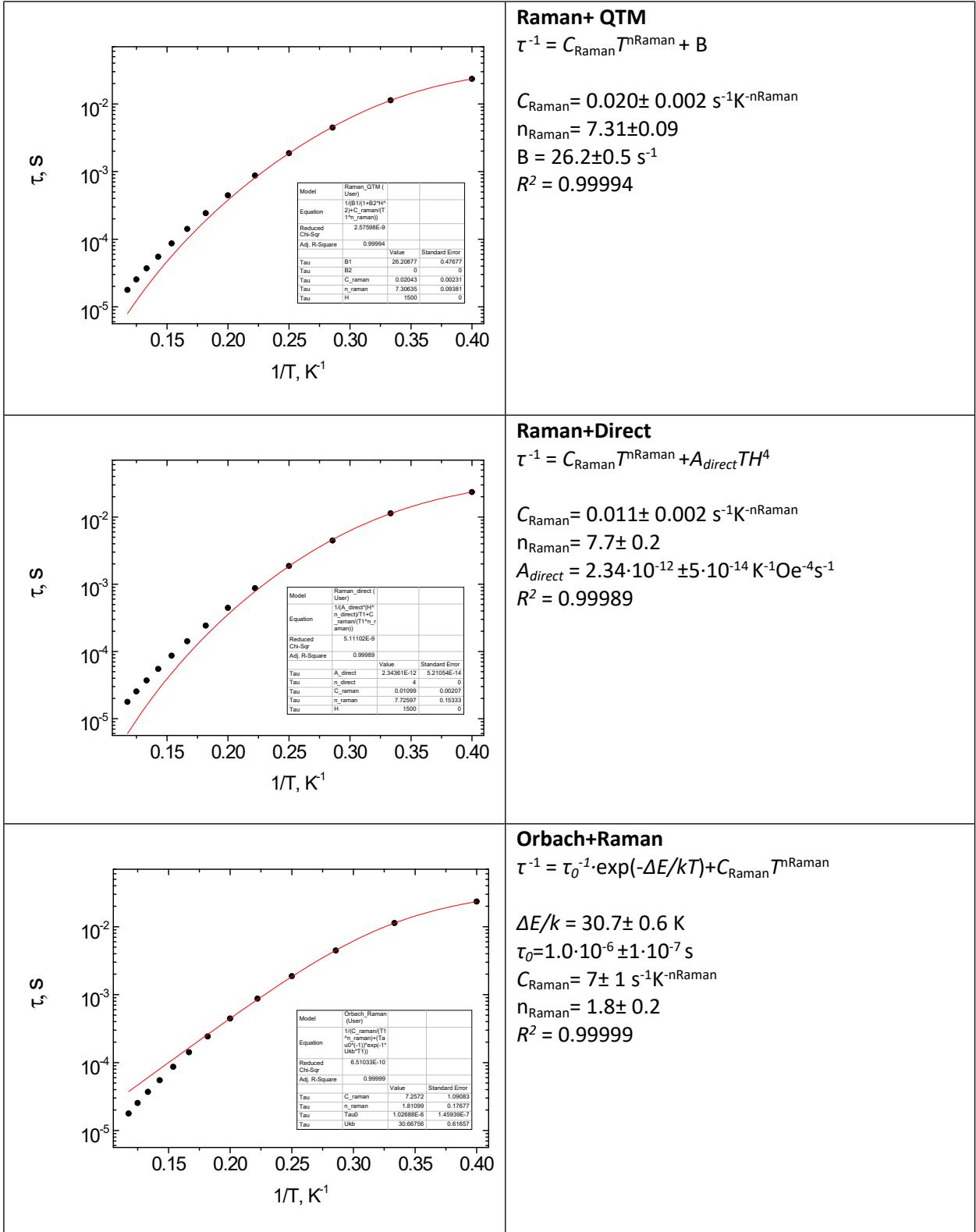


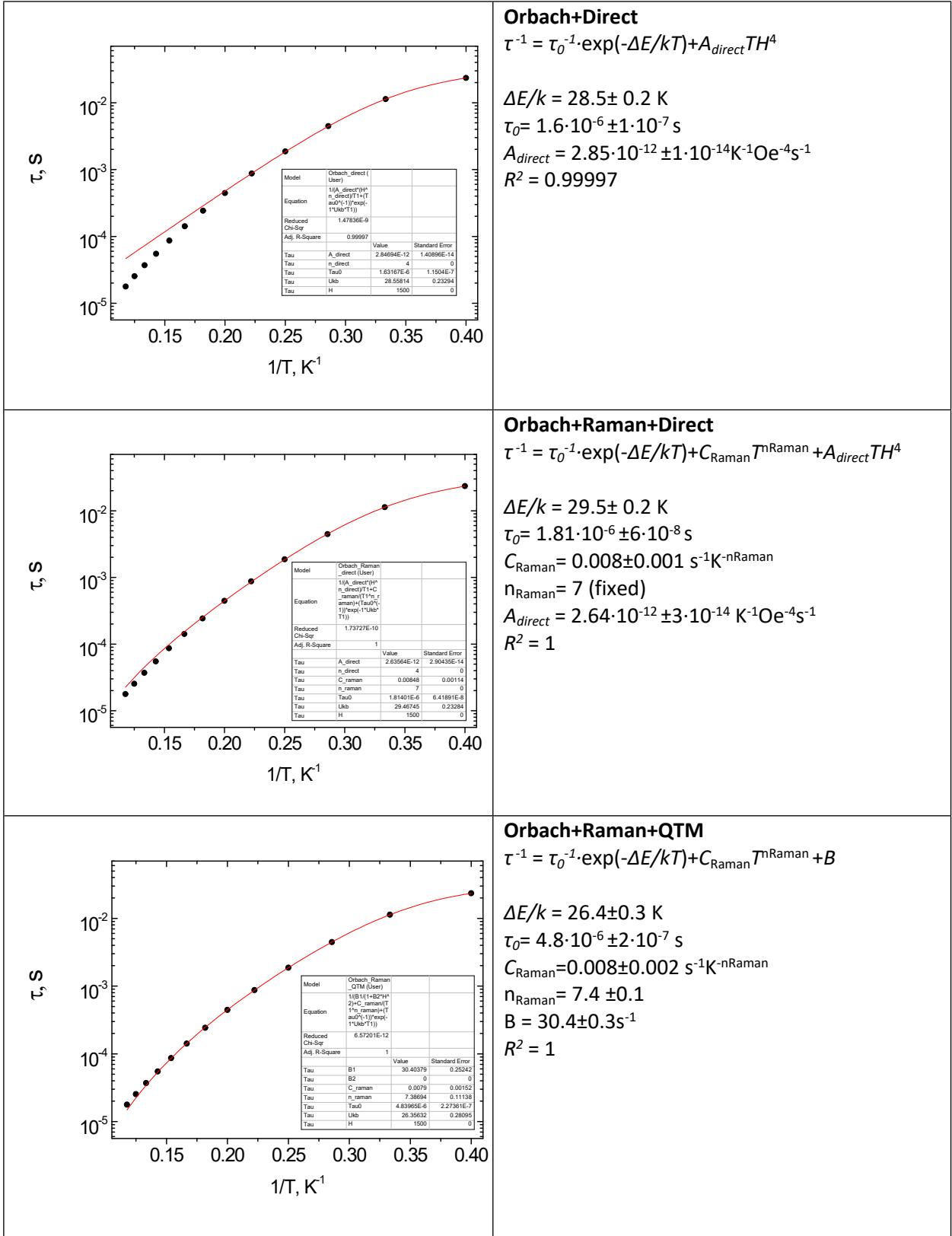


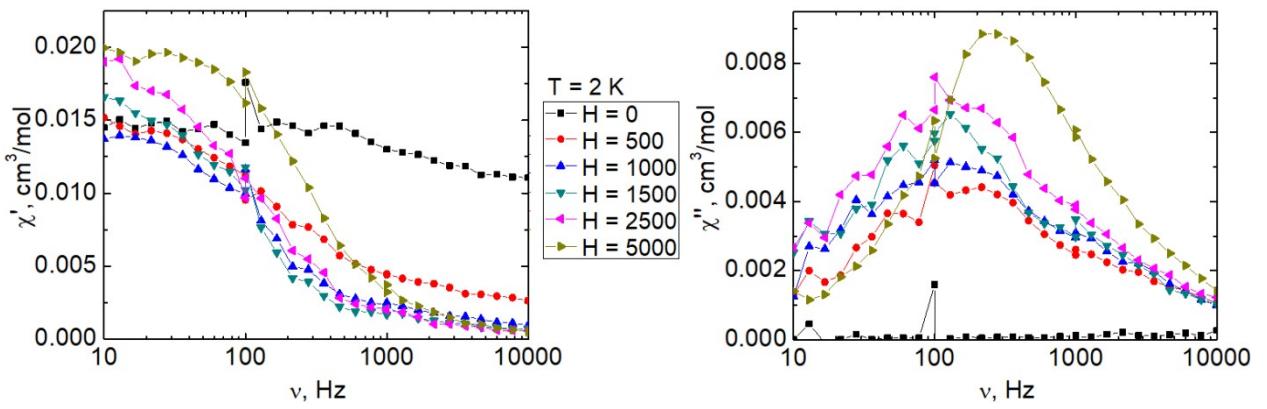


**Table S5.** Fitting of the  $\tau$  vs.  $T$  dependences for **2** ( $H = 1.5$  kOe,  $T = 2.5\text{--}8.5$  K).

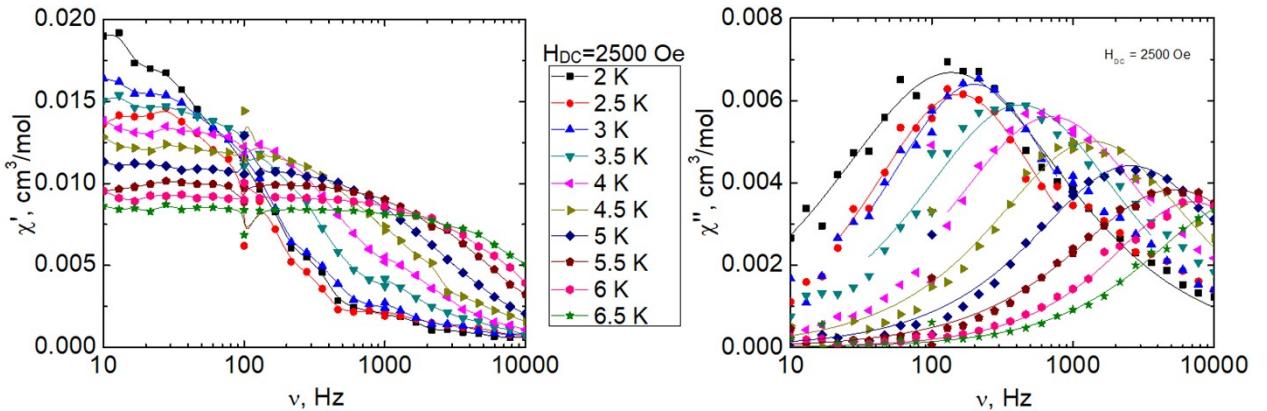




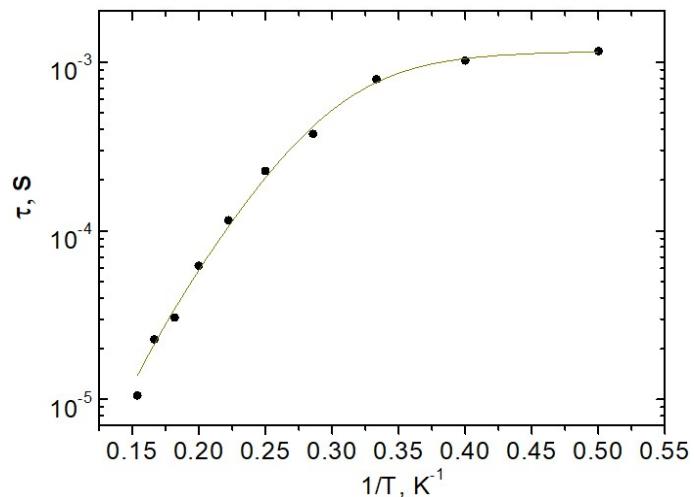




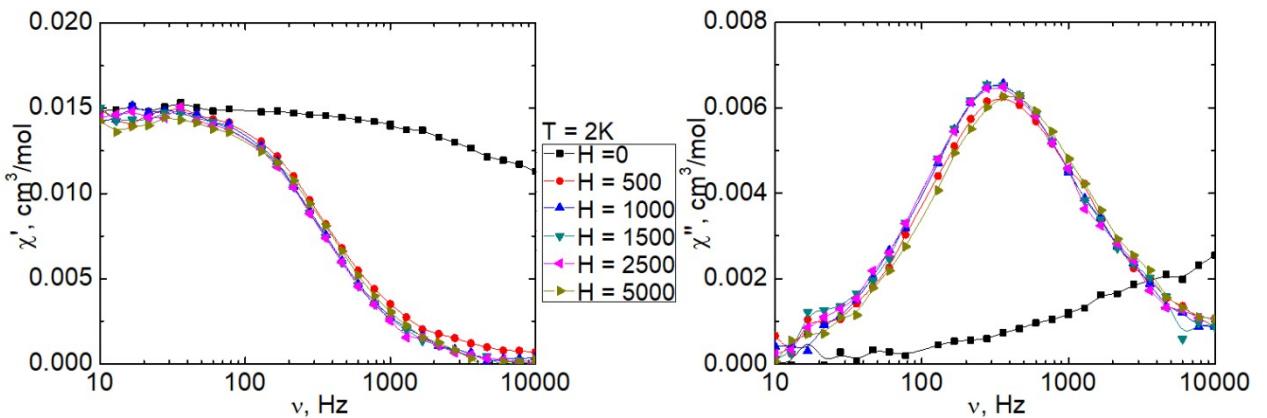
**Fig. S22.** Frequency dependencies of the in-phase ( $\chi'$ ) and out-of phase ( $\chi''$ ) components of the dynamic magnetic susceptibility of **1\_Lu** at varied strength of the external static magnetic field and  $T = 2 \text{ K}$ . Solid lines are visual guides.



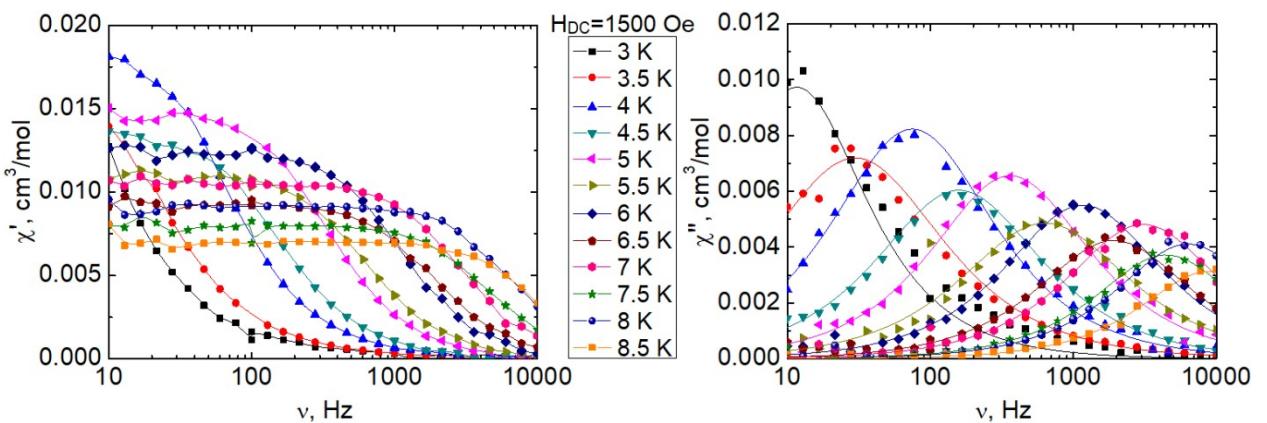
**Fig. S23.** Frequency dependencies of the in-phase ( $\chi'$ , a) and out-of-phase ( $\chi''$ , b) components of the dynamic magnetic susceptibility of **1\_Lu** under optimal magnetic field 2500 Oe at various temperatures. Solid lines are visual guides (a), approximations by the generalized Debye model (b).



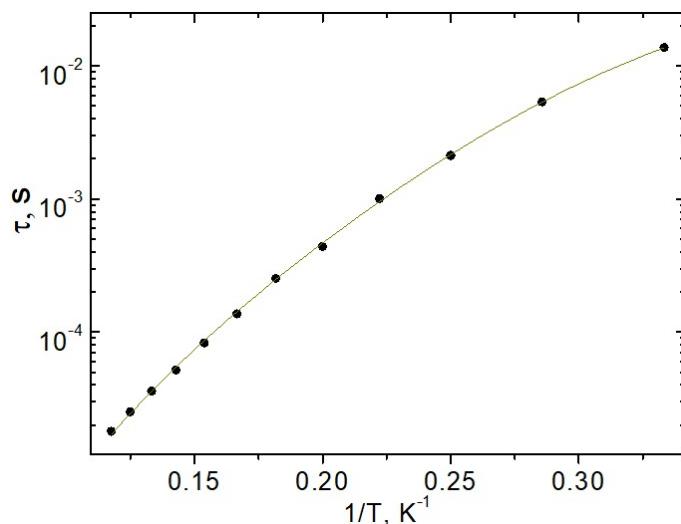
**Fig. S24.**  $\tau$  vs.  $T^{-1}$  plots for **1\_Lu** in 2500 Oe DC field. Solid line represents best fit by the sum of Orbach, Raman and QTM relaxation mechanisms.



**Fig. S25.** Frequency dependencies of the in-phase ( $\chi'$ ) and out-of-phase ( $\chi''$ ) components of the dynamic magnetic susceptibility of **2\_Y** at varied strength of the external static magnetic field and  $T=2\text{K}$ . Solid lines are visual guides.



**Fig. S26.** Frequency dependencies of the in-phase ( $\chi'$ , a) and out-of phase ( $\chi''$ , b) components of the dynamic magnetic susceptibility of **2\_Y** under optimal magnetic field  $1500 \text{ Oe}$  at various temperatures. Solid lines are visual guides (a), approximations obtained by the use of generalized Debye model (b).



**Fig. S27.**  $\tau$  vs.  $T^{-1}$  plots for **2\_Y** in  $1500 \text{ Oe}$  DC field. Solid line represents best fit by the sum of Orbach, Raman and QTM relaxation mechanisms.

**Table S6.**Fitting parameters of magnetization relaxation for **3** and **4**.

Complex (dc-field, Oe)	Orbach+Raman+QTM				
	$\Delta_{\text{eff}}/k_B, \text{K}$	$\tau_0, \text{s}$	$C_{\text{Raman}}, \text{s}^{-1}\text{K}^{-n_{\text{Raman}}}$	$n_{\text{Raman}}$	$B, \text{s}^{-1}$
<b>1_Lu</b> (2500)	$27 \pm 3$	$4.4 \cdot 10^{-7} \pm 4 \cdot 10^{-7}$	$0.1 \pm 0.1$	7 (fixed)	$860 \pm 28$
<b>2_Y</b> (1500)	$35 \pm 3$	$1 \cdot 10^{-6} \pm 5 \cdot 10^{-7}$	$0.040 \pm 0.008$	6.5 (fixed)	$18 \pm 5$

**Table S7.** Known 7-coordinated Yb SMMs.

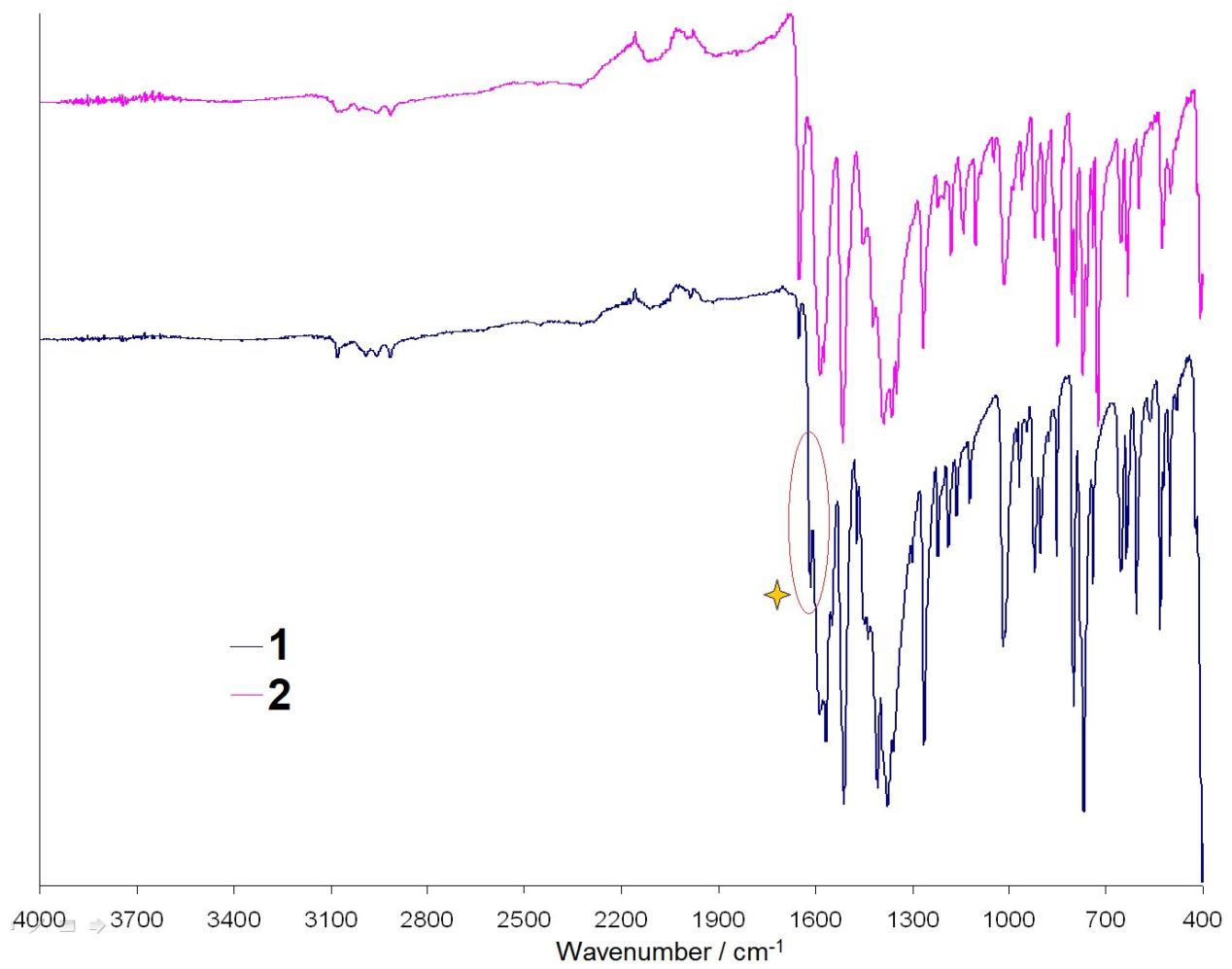
Complex / Nuclearity	Polyhedron	$\Delta_{\text{eff}}/k_B, K /$ (H <sub>dc</sub> )	[Ref.]
[Yb(acac <sup>[2]</sup> ) <sub>2</sub> (Q <sup>[3]</sup> )] <sub>n</sub> (1)	Capped octahedron	40 / (1.5 kOe)	This work
YbL <sup>[I]</sup> .4H <sub>2</sub> O / Mononuclear		16	2
Yb(trensal <sup>[II]</sup> ) <sup>‡</sup> / Mononuclear		54.7 <sup>†</sup>	5
[Yb <sub>2</sub> (H <sub>2</sub> cht <sup>[III]</sup> ) <sub>2</sub> Cl <sub>4</sub> (H <sub>2</sub> O)(MeCN)]·MeCN / Dinuclear		19.5(2)	8
YbL <sup>[IV]</sup> / Mononuclear		-§	9
YbL <sup>*[V]</sup> .3MeOH / Mononuclear			
{[Yb <sub>2</sub> (DTE <sup>[VI]</sup> )(H-DTE)(MeOH) <sub>2</sub> ]·2H <sub>2</sub> O} <sub>n</sub> <sup>‡</sup> / 1D-polynuclear	Capped trigonal prism	38.9 <sup>†</sup>	1
[Yb(H <sub>3</sub> L <sup>1,1,4<sup>[VII]</sup>)<sub>2</sub>]·2MeOH / Mononuclear</sup>		20.9 <sup>†</sup>	6
[Yb{Ir(ppy <sup>[VIII]</sup> ) <sub>2</sub> (dcbpy <sup>[IX]</sup> ) <sub>2</sub> (NO <sub>3</sub> ) (H <sub>2</sub> O) <sub>4</sub> }·Solv <sup>#</sup> / 1D-polynuclear	Capped trigonal pris m↔Pentagonal bip yramide	24.4	4
[Yb <sub>2</sub> (OH){Ir(ppy) <sub>2</sub> (dcbpy)} <sub>4</sub> (NO <sub>3</sub> )(H <sub>2</sub> O) <sub>4</sub> }·Solv <sup>#</sup> / 2D-polynuclear		22.2	
Yb <sub>2</sub> L <sup>[X]</sup> <sub>2</sub> (depma <sup>[XI]</sup> <sub>2</sub> )Cl <sub>2</sub> / Dinuclear	Pentagonal bipyramide	23.5	3
[Yb <sub>2</sub> (NMP <sup>[XII]</sup> ) <sub>12</sub> (PW <sub>12</sub> O <sub>40</sub> )][PW <sub>12</sub> O <sub>40</sub> ] <sup>#‡</sup> / Dinuclear		11.84 <sup>Δ</sup>	7
[Yb(BcrCOO <sup>[XIII]</sup> )(acac <sup>[XIV]</sup> ) <sub>2</sub> (H <sub>2</sub> O)] <sub>n</sub> <sup>#</sup> / 1D-polynuclear		36	10
[Yb(H <sub>3</sub> Bmshp <sup>[XV]</sup> )(DMF) <sub>2</sub> Cl <sub>2</sub> ]·DMF·1.5H <sub>2</sub> O / Mononuclear		14.5 <sup>†</sup>	11
[Yb(H <sub>3</sub> Bmshp)(DMF) <sub>2</sub> Cl <sub>2</sub> ]·H <sub>4</sub> Bmshp / Mononuclear		38.3 <sup>†</sup>	

<sup>[I]</sup>L = fully (triply) deprotonated N[(CH<sub>2</sub>)<sub>2</sub>N=CH-R-CH=N-(CH<sub>2</sub>)<sub>2</sub>]<sub>3</sub>N cryptand (R = m-C<sub>6</sub>H<sub>5</sub>OH-2-Me-5); <sup>[II]</sup>trensal = fully (triply) deprotonated 2,2',2"-tris(salicylideneimino)triethylamine; <sup>‡</sup>No powder XRD data were provided for the studied samples; <sup>†</sup>Recalculated from cm<sup>-1</sup>; <sup>[III]</sup>H<sub>3</sub>cht = mono deprotonated 1,3,5-cyclohexanetriol; <sup>[IV]</sup>L = tris(((3-formyl-5-methylsalicylidene)amino)ethyl)amine; <sup>§</sup> $\Delta_{\text{eff}}/k_B$  values were not determined since Orbach mechanism was not applied to fit relaxation times; <sup>[V]</sup>L<sup>\*</sup> = triply deprotonated product of condensation of tris(((3-formyl-5-methylsalicylidene)amino)ethyl)amine with benzylamine; <sup>[VI]</sup>DTE = 1,2-bis(5-carboxyl-2-methyl-3-thienyl)perfluorocyclopentene; <sup>[VII]</sup>H<sub>3</sub>L<sup>1,1,4</sup>= triply deprotonated 2-[{(2-[2-hydroxybenzyl]amino)ethyl}{2-[3-(2-hydroxybenzyl)-2,2-dimethylimidazolidin-1-yl]ethyl}amino)methyl]phenol; <sup>[VIII]</sup>ppy = 2-phenylpyridine; <sup>[IX]</sup>dcbpy = 2,2'-bipyridine-4-carboxyl-4'-carboxylate; <sup>#</sup>Magnetic behavior of these complexes is governed by Yb<sup>3+</sup> due to diamagnetism of the involved cations of heterometals; <sup>[X]</sup>L = doubly deprotonated N<sup>1,N<sup>3</sup></sup>-bis(salicylideneimino)diethylenetriamine; <sup>[XI]</sup>depma<sub>2</sub> = dimerized 9-diethyl-phosphonomethylanthracene; <sup>[XII]</sup>NMP = N-methyl pyrrolidone; <sup>Δ</sup>This value was obtained for the Lu-diluted sample; <sup>[XIII]</sup>BrcCOO = (n<sub>6</sub>-benzoate)tricarbonylchromium; <sup>[XIV]</sup>acac = acetylacetone (pentane-2,4-dionate) anion; <sup>[XVII]</sup>H<sub>3</sub>Bmshp = mono deprotonated (2,6-bis[(3-methoxysalicylidene)hydrazinecarbonyl]-pyridine).

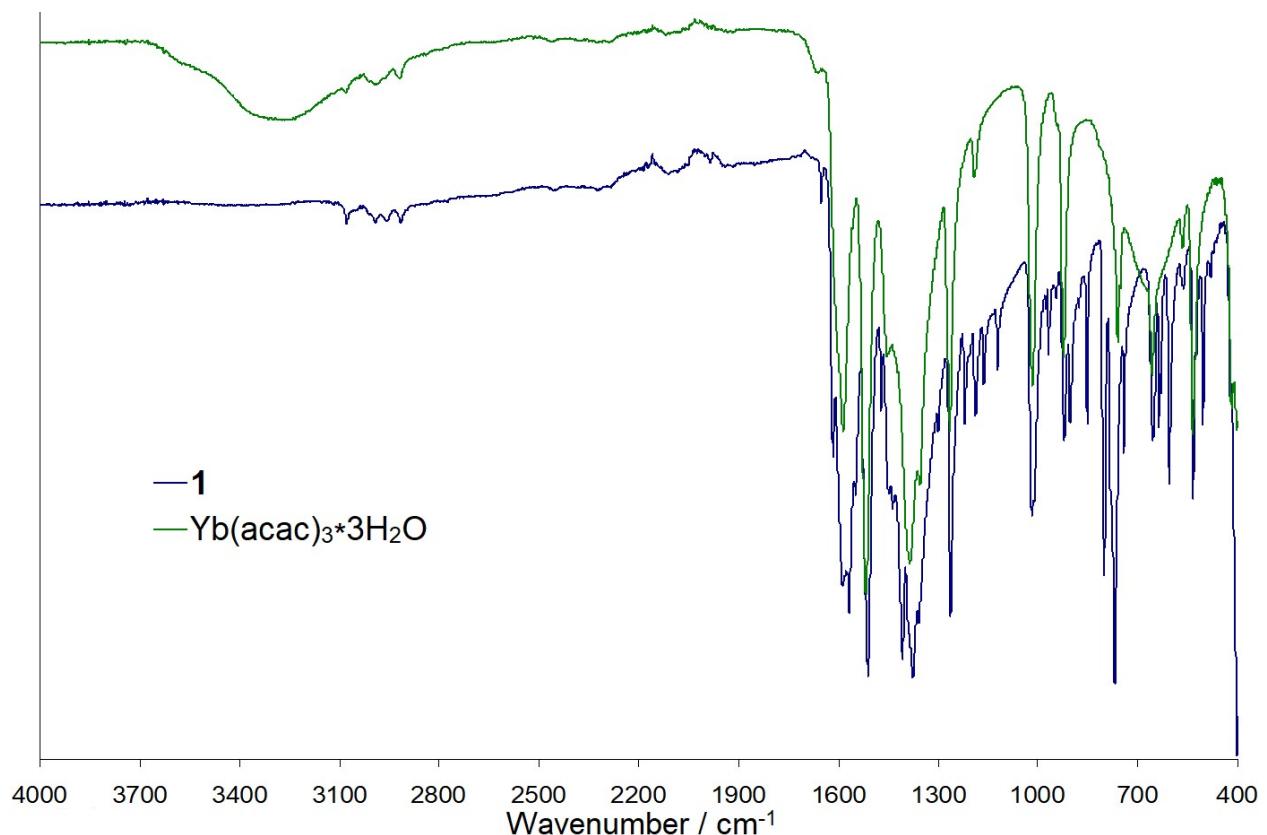
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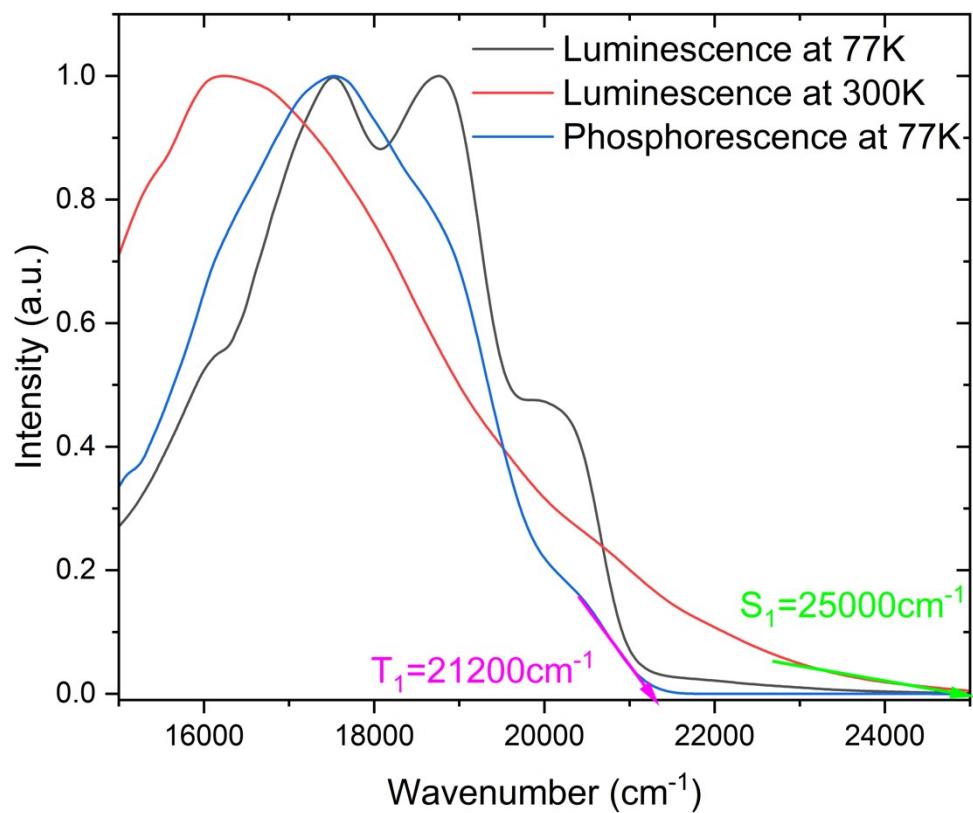
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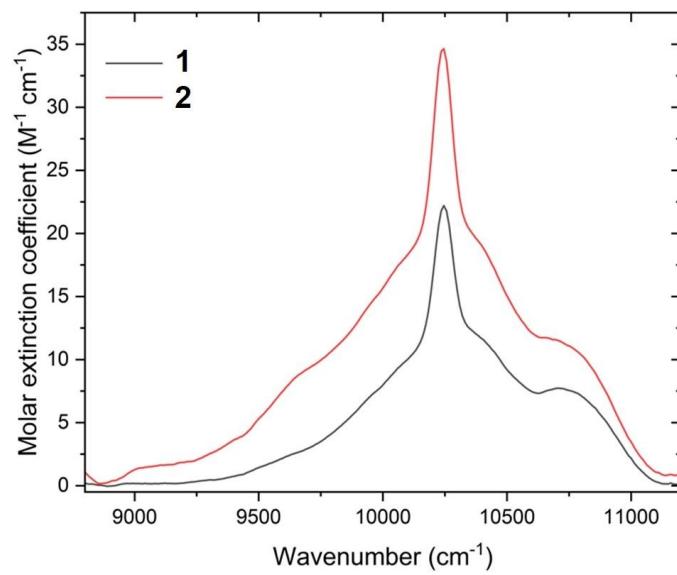
**Fig. S28.** IR spectra of **1** and **2** with specified band of high-energy asymmetric stretching vibration ofCOO(Q) groups.



**Fig. S29.** IR spectrum of **1** compared to that of initial  $\text{Yb}(\text{acac})_3 \cdot 3\text{H}_2\text{O}$ .



**Fig. S30.** Luminescence spectra in the energy representation for the **2\_Gd** at temperatures of 77K and 300K.



**Fig. S31.** Optical absorption spectra of solutions of **1** and **2** in DMSO.