

Electronic supporting information

Resolving a structural issue in cerium-nickel-based oxide: single compound or a two-phase system?

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Table S1. Amounts of precursors for the synthesis of compounds.

Compound	Element	Amount (mmol)
RB1 (CeNiO ₃)	Ce	1
	Ni	
CeO ₂	Ce	1
NiO	Ni	1

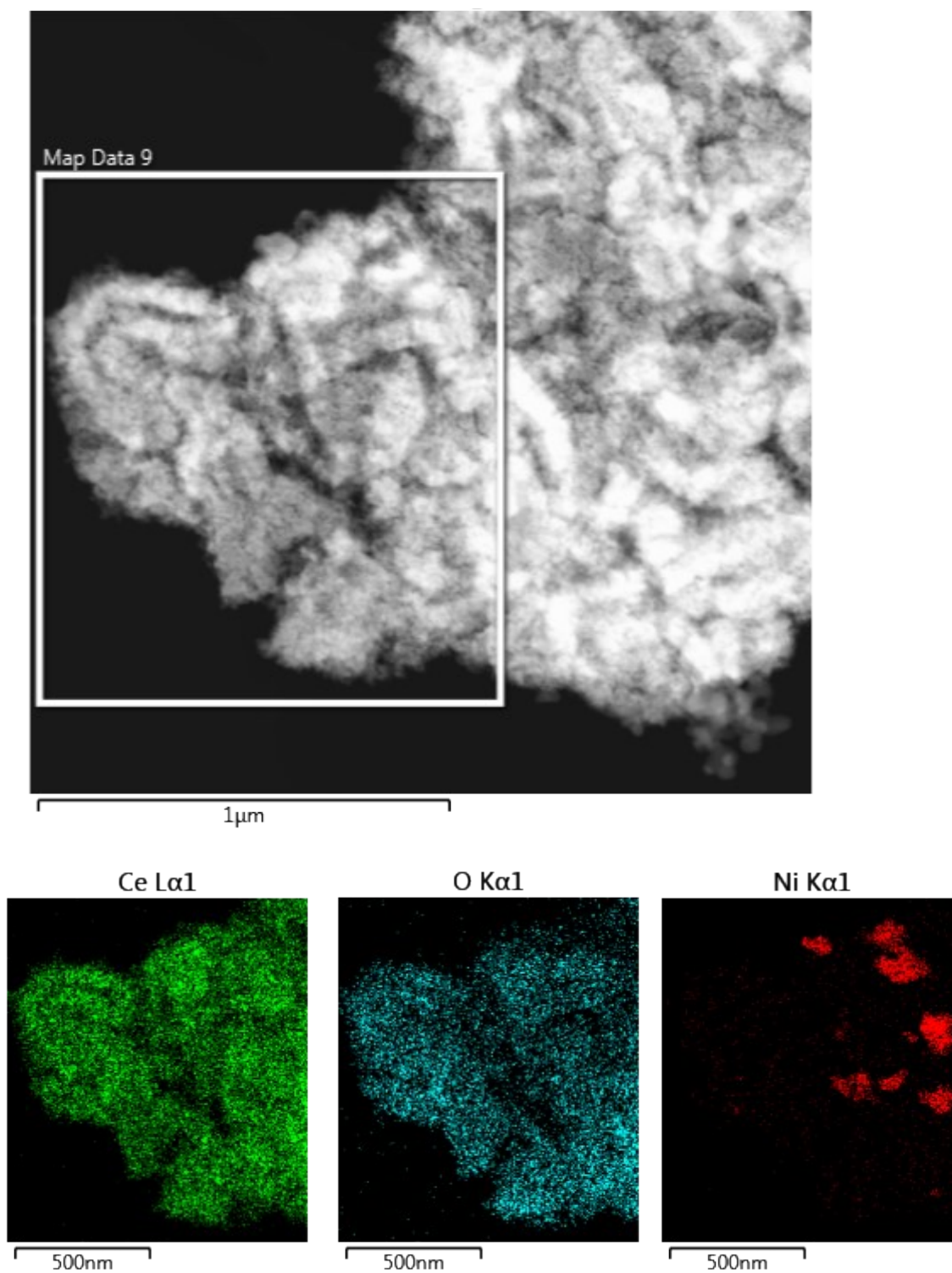


Figure S1. STEM-EDS image of a mixture $\text{CeO}_2 + \text{NiO}$ ($n : n = 1:1$) at one spot showing a non-uniform distribution of Ce, Ni and O. The measured spot contains more cerium than nickel.

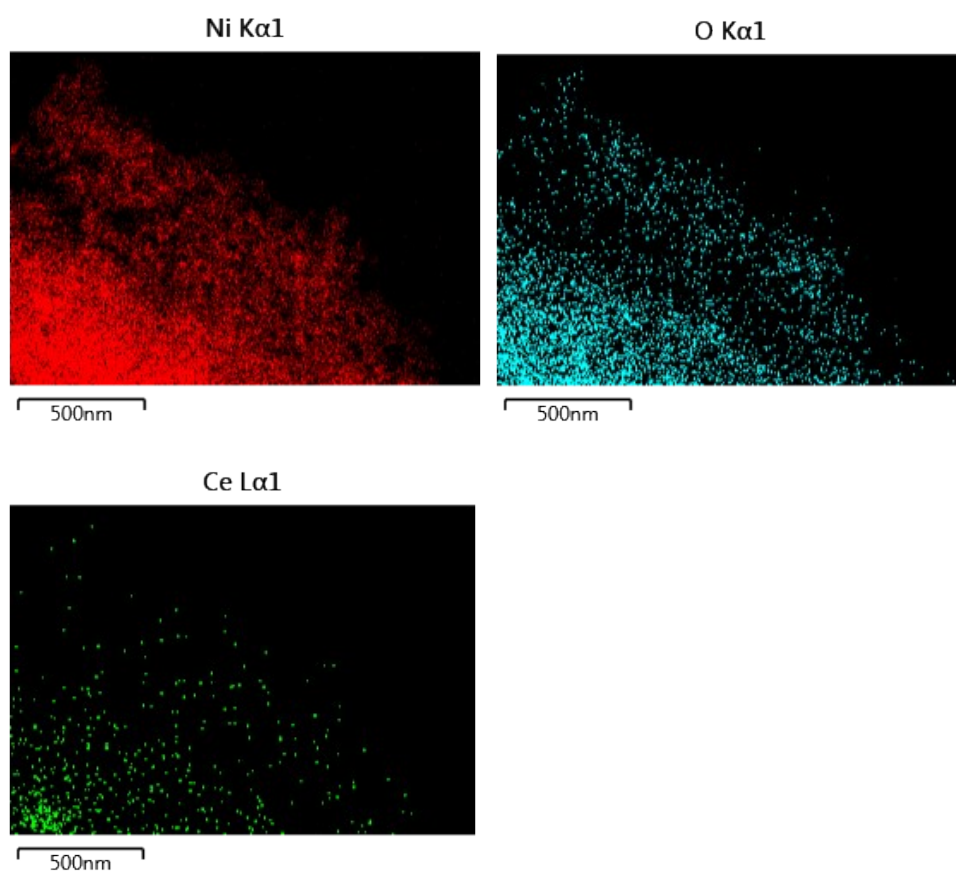
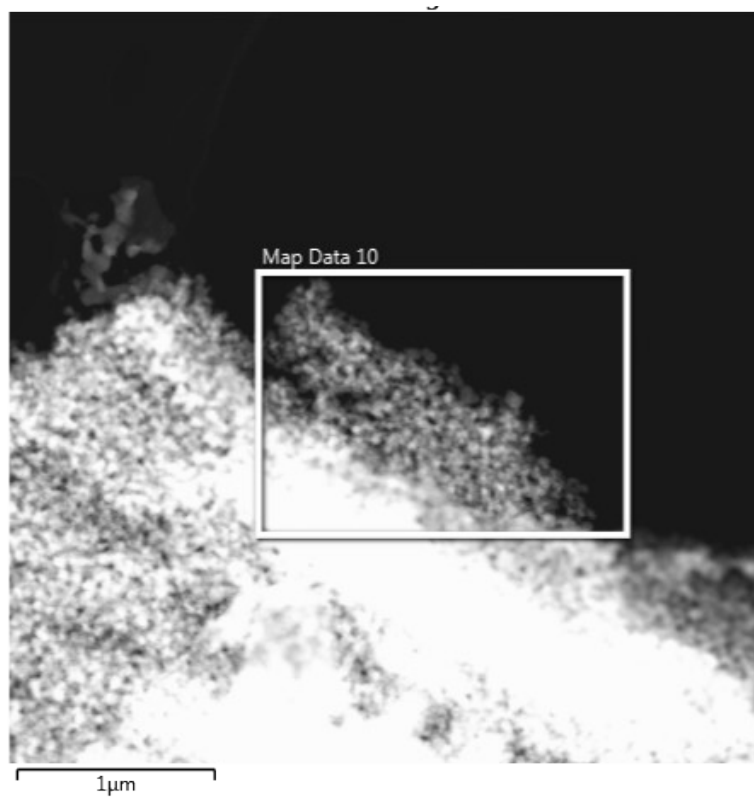


Figure S2. STEM-EDS image of a mixture $\text{CeO}_2 + \text{NiO}$ ($n : n = 1:1$) at one spot showing a non-uniform distribution of Ce, Ni and O. The measured spot contains more nickel than cerium.

EXAFS analysis of CeO₂ and RB1 at Ce L₃-edge

Structural parameters of the average local Ce neighbourhood (the type and the average number of neighbours; the radii and Debye–Waller factor of neighbour shells) in CeO₂ are quantitatively resolved from the EXAFS spectra by comparing the measured EXAFS signal with the model signal. The FEFF model for the crystalline CeO₂ nanoparticles is based on the cubic crystal structure of CeO₂ with the space group *Fm*–*3m* with the lattice constant $a = 5.411$ Å (PDF4+ database card #00-004-0593), where Ce is coordinated with 8 oxygen atoms at a distance of 2.34 Å, 12 Ce atoms at 3.83 Å and 24 oxygen atoms at 4.47 Å. The FEFF model comprised three single scattering and ten significant multiple scattering paths up to 4.6 Å, with 7 variable parameters: coordination shell distance (R) for the first (O) and second (Ce) neighbour shell, Debye–Waller factor (σ^2) of the single scattering path for (Ce–O) whereas Debye–Waller factors of other paths are evaluated using Debye model via Debye temperature (θ_d). The amplitude reduction factor (S_0^2) and shift of the energy origin of the photoelectron (ΔE_0), common to all scattering paths, are introduced. The shell coordination numbers were fixed to the crystallographic values in the fit, and the structural parameters of multiple scattering paths are constrained to those of the corresponding single scattering paths. A very good EXAFS fit (**Fig. S3**) is obtained in the k range of 2–9.9 Å⁻¹ and the R -range of 1.3–4.6 Å. The best-fit structural parameters are given in **Table S2**. For the RB1 sample, the EXAFS spectra were fitted with the FEFF model as it was done for CeO₂ (cubic *Fm*–*3m*) and the FEFF model for CeO₂ with the space group *P4*₂/*nmc* with the lattice constant $a = 3.818$ Å, $b = 3.818$ Å and $c = 5.426$ Å (PDF4+ database card #04-025-2756). For fitting the RB1, the amplitude reduction factor (S_0^2) was fixed to the value obtained for CeO₂ and shell coordination numbers were used as variable parameters. The best-fit structural parameters are given in **Table S2**. The results show that EXAFS cannot resolve between cubic and tetragonal structures for RB1 sample. The obtained structural parameters for RB1 are the same for both FEFF models and in agreement with those obtained for CeO₂.

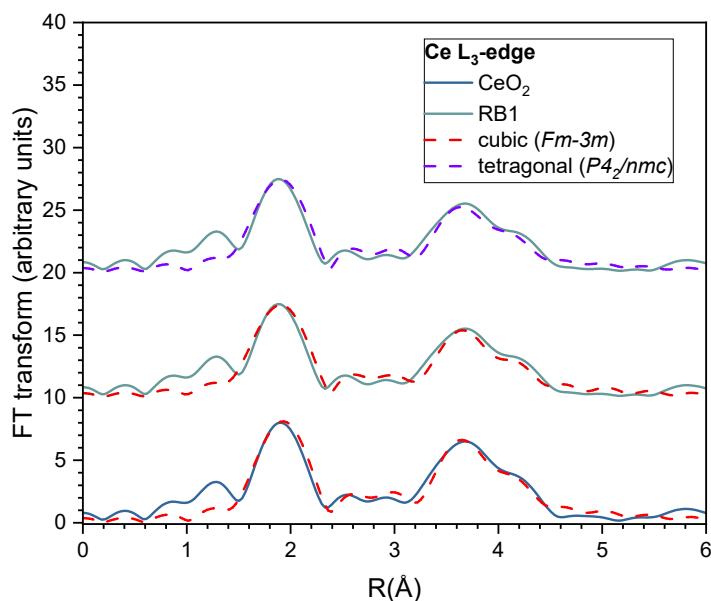


Figure S3. Fourier transform magnitude of the k^3 -weighted Ce L₃-edge EXAFS spectra of the CeO₂ and RB1 samples calculated in the k range of 2–9.9 Å. Experiment – (solid line); the best fit EXAFS model calculated in the R -range of 1.3 to 4.6 Å – (red dashed line for cubic system, violet dashed line for tetragonal system). Graph curves shifted vertically for clarity.

Table S2. Parameters of the nearest coordination shells around Ce cations in the CeO₂ and RB1 sample: coordination number (N), distance (R), and Debye-Waller factor (σ^2). Uncertainty of the last digit is given in parentheses. The best fit is obtained with the amplitude reduction factor $S_0^2 = 0.66$ and the shift of the energy origin $\Delta E_0 = 8(1)$ eV.

CeO ₂				
Neighbour	N (fixed)	R (Å)	σ^2 (Å ²)	R -factor
O	8	2.34(1)	0.004(3)	0.028
Ce	12	3.86(1)	0.004(3)	
O	24	4.41(1)	0.02(1)	
RB1 - fit with cubic $Fm\text{-}3m$ structure model				
Neighbour	N	R (Å)	σ^2 (Å ²)	R -factor
O	8(2)	2.33(1)	0.006(3)	0.041
Ce	10(3)	3.85(1)	0.004(3)	
O	21(7)	4.41(2)	0.02(1)	
RB1 - fit with tetragonal $P4_2/nmc$ structure model				
Neighbour	N	R (Å)	σ^2 (Å ²)	R -factor
O	8(2)	2.35(1)	0.005(3)	0.042
Ce	8(3)	3.86(1)	0.004(3)	
O	20(8)	4.3(1)	0.03(1)	

Table S3. High-resolution Ce 3d spectra peak position analysis for CeO₂ and RB1.

CeO ₂	RB1	Assignment	Orbital splitting
Binding energy (eV)	Binding energy (eV)		
883.1	883.3	Core level	3d _{5/2}
889.8	890.1	Satellite 1	
899.3	899.3	Satellite 2	
902.3	901.9	Core level	3d _{3/2}
908.8	908.7	Satellite 3	
917.8	918.4	Satellite 4	

Table S4. High-resolution Ni 2p spectra peak position analysis for NiO and RB1.

NiO	RB1	Assignment	Orbital splitting
Binding energy (eV)	Binding energy (eV)		
853.9	854.4	Core level	2p _{3/2}
855.7	855.8	Core level	
861.0	857.3	Satellite 1	
864.4	861.4	Satellite 2	
866.6	864.1	Core level	2p _{1/2}
871.4	Not possible to read	Core level	
873.2	874.0	Core level	
879.3	Not possible to read	Satellite 3	

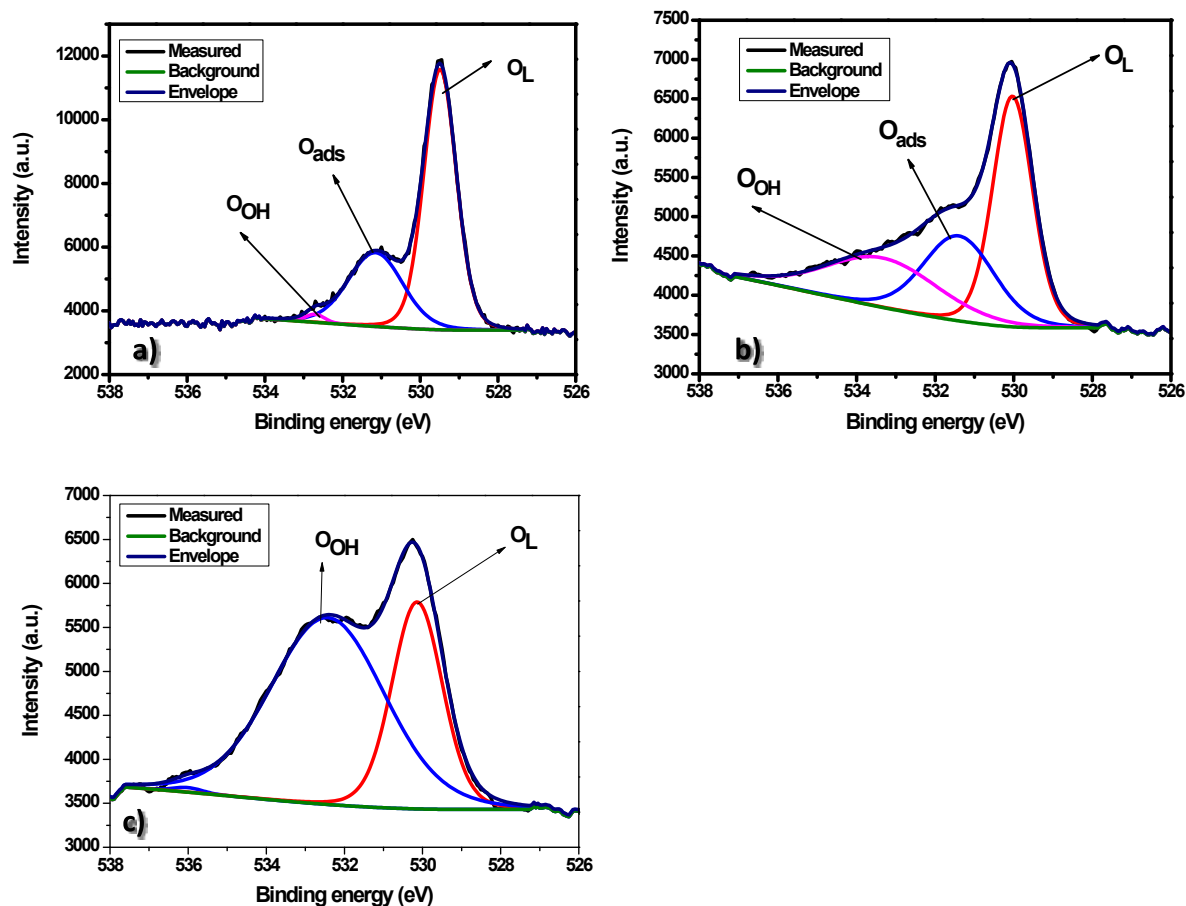


Figure S4. Deconvoluted high-resolution O 1s spectrum of a) NiO b) CeO₂ and c) RB1.

Table S5. Deconvoluted O 1s spectra report for CeO₂, NiO and RB1.

NiO		CeO ₂		RB1		Assignment
Binding energy (eV)	Area	Binding energy (eV)	Area	Binding energy (eV)	Area	
529.5	8909.8	530.03	3902.3	530.1	3977.2	Lattice oxygen (O _L)
531.2	4252.8	531.4	2554.4	-	-	Adsorbed oxygen species (O _{ads})
532.7	161.7	533.3	2347.0	532.4	8113.3	Hydroxides (O _{OH})
Total area	13324.2		8803.7		12142.1	
Oxygen species concentration	31.9 %		29.0 %		-	
Hydroxide concentration	1.2 %		26.7 %		66.8 %	

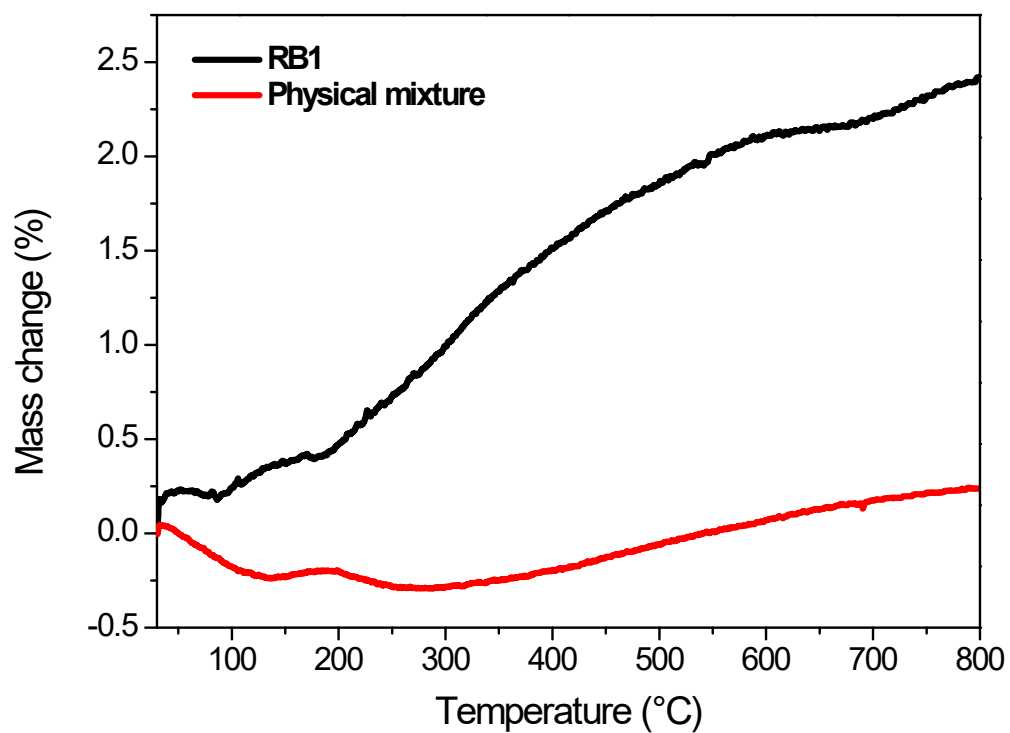


Figure S5. Thermogravimetric curves of RB1 (black) and a physical mixture of $\text{CeO}_2 + \text{NiO}$ (red) during heating in an oxidative atmosphere (10 °C/min) from 30 °C to 800 °C .

Table S6. Crystallographic data obtained from the Rietveld refinement of 3% Ni doped CeO₂.

Compound	CeO ₂ (3% doped Ni)
Chemical formula	Ce _{0.97} Ni _{0.03} O ₂
Crystal system	Tetragonal
Space group	<i>P 4₂ / n m c</i>
<i>Z</i>	2
Calculated density (g/cm ³)	7.076
Unit cell parameters (Å)	a= 3.8319(2) c= 5.4183(1)
Unit cell volume (Å ³)	79.5600(1)
Phase content (wt. %)	100
Average crystallite size (nm)	8.5
Average apparent microstrain (×10 ⁻⁴)	0.95
<i>R_B</i>	5.58
<i>R_p, R_{wp}, R_c</i>	13.2, 10.7, 8.23
χ^2	1.68