Supporting Information for

CAAC-stabilised 9,10-diboraanthracene: an electronically and structurally flexible platform for small-molecule activation and metal complexation

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Methods and materials

All manipulations were performed either under an atmosphere of dry argon or *in vacuo* using standard Schlenk line or glovebox techniques. Deuterated solvents were dried over molecular sieves and degassed by three freeze-pump-thaw cycles prior to use. All other solvents were distilled and degassed from appropriate drying agents. Both deuterated and non-deuterated solvents were stored under argon over activated 4 Å molecular sieves. Liquid-phase NMR spectra were acquired on a Bruker Avance 500 (¹H: 500.1 MHz, ¹¹B: 160.5 MHz, ¹³C: 125.8 MHz) spectrometer. Chemical shifts (δ) are reported in ppm and internally referenced to the carbon nuclei (¹³C{¹H}) or residual protons (¹H) of the solvent. Heteronuclei NMR spectra are referenced to external standards (¹¹B: BF₃·OEt₂). Resonances are identified as singlet (s), doublet (d), triplet (t), septet (sept), multiplet (m) or broad (br). Coupling constants are ¹H-¹H coupling constants unless specified otherwise. High-resolution mass spectrometry (HRMS) data were obtained from a Thermo Scientific Exactive Plus spectrometer. Solid-state IR spectra were acquired on a Bruker FT-IR spectrometer ALPHA II inside a glovebox. UV-vis spectra were acquired on a METTLER TOLEDO UV-vis-Excellence UV5 spectrophotometer inside a glovebox.

Solvents and reagents were purchased from Sigma-Aldrich, abcr or Alfa Aesar. $[(CAAC)_2(C_{12}H_8B_2)] = II (CAAC = 1-(2,6-di iso propylphenyl)-3,3,5,5-tetramethyl-pyrrolidin-2-ylidene)^1$ and $[(MeCN)_3M(CO)_3]$ (M = Cr, Mo, W)^2 were synthesised using literature procedures.

Synthetic procedures

Synthesis of 1

A solution of II (60.0 mg, 80.6 µmol) in benzene (4 mL) was placed under an atmosphere of dihydrogen (1 atm) via three freeze-pump-thaw cycles. The solution was stirred at room temperature for 30 min, resulting in a colourless solution. All volatiles were removed in vacuo and the residue was washed with hexane $(3 \times 1 \text{ mL})$ and dried, yielding 1 as a colourless solid (48.0 mg, 64.3 µmol, 80%). Single crystals suitable for X-ray diffraction analysis were obtained by slow evaporation of a saturated benzene/pentane (2:1) solution at room temperature. ${}^{1}H{}^{11}B{}$ NMR (500.1 MHz, C_6D_6): $\delta = 8.20-8.12$ (m, 1H, DBA-H), 7.60–7.53 (m, 1H, DBA-H), 7.42– 7.34 (m, 2H, DBA-H), 7.31–7.22 (m, 3H, m-Ar-H + p-Ar-H), 7.18–6.96 (m, 6H, m-Ar-H + p-Ar-H + DBA-H), 6.83–6.74 (m, 1H, DBA-H), 5.25 (s, 1H, NCH), 4.23 (sept, ${}^{3}J$ = 6.6 Hz, 1H, *i*Pr-CH), 4.05 (sept, ${}^{3}J = 6.6$ Hz, 1H, *i*Pr-CH), 3.22 (br s, 1H, BH), 3.18 (sept, ${}^{3}J = 6.6$ Hz, 1H, *i*Pr-CH), 3.05 (sept, ${}^{3}J = 6.6$ Hz, 1H, *i*Pr-CH), 1.84 (d, ${}^{2}J = 12.3$ Hz, 1H, CH₂), 1.78 (d, ${}^{2}J = 12.3$ Hz, 1H, CH₂), 1.58 (d, ${}^{2}J = 12.9$ Hz, 1H, CH₂), 1.45 (d, ${}^{3}J = 6.6$ Hz, 3H, *i*Pr-CH₃), 1.40 (d, ${}^{3}J = 6.6$ Hz, 3H, *i*Pr-CH₃), 1.37 (d, ${}^{2}J = 12.9$ Hz, 1H, CH₂), 1.33 (d, ${}^{3}J = 6.6$ Hz, 3H, *i*Pr-CH₃), 1.32 (s, 6H, NC(CH₃)₂), 1.23 (s, 3H, C(CH₃)₂), 1.22 (d, ${}^{3}J$ = 6.6 Hz, 3H, *i*Pr-CH₃), 1.18 (d, ${}^{3}J = 6.6$ Hz, 3H, *i*Pr-CH₃), 1.16 (d, ${}^{3}J = 6.6$ Hz, 3H, *i*Pr-CH₃), 0.95 (s, 6H, NC(CH₃)₂) $+ C(CH_3)_2$, 0.91 (d, ${}^{3}J = 6.6$ Hz, 3H, *i*Pr-CH₃), 0.91 (s, 3H, C(CH₃)₂), 0.89 (s, 3H, NC(CH₃)₂), 0.85 (s, 3H, C(CH₃)₂), 0.74 (d, ${}^{3}J = 6.6$ Hz, 3H, *i*Pr-CH₃) ppm. ${}^{13}C{}^{1}H$ NMR (125.8 MHz, C₆D₆): $\delta = 243.5$ (C_{carbene}, *identified by HMBC*), 153.0 (o-Ar-C), 151.0 (DBA-C_B), 150.5 (o-Ar-C), 146.8 (DBA-C_B), 146.5 (o-Ar-C), 145.3 (o-Ar-C), 143.9 (i-Ar-C), 135.1 (DBA-C), 134.3 (DBA-C), 134.1 (DBA-C + *i*-Ar-C), 133.3 (DBA-C), 129.4 (*p*-Ar-C), 127.1 (DBA-C), 126.4 (p-Ar-C), 126.1 (DBA-C), 125.3 (m-Ar-C), 125.3 (m-Ar-C), 124.9 (m-Ar-C), 124.9 (m-Ar-C), 123.5 (DBA-C), 123.3 (DBA-C), 77.7 (NC(CH₃)₂), 73.7 (NCH), 63.5 (NC(CH₃)₂), 60.1 (CH₂), 54.5 (C(CH₃)₂), 52.7 (CH₂), 44.2 (C(CH₃)₂), 33.2 (C(CH₃)₂), 31.4 (C(CH₃)₂), 31.2 (C(CH₃)₂), 30.5 (C(CH₃)₂), 30.3 (NC(CH₃)₂), 29.6 (*i*Pr-CH), 29.6 (*i*Pr-CH), 29.3 (*i*Pr-CH), 29.1 (NC(CH₃)₂), 28.3 (*i*Pr-CH), 28.1 (NC(CH₃)₂), 27.5 (*i*Pr-CH₃), 27.4 (*i*Pr-CH₃), 27.0 (*i*Pr-CH₃), 26.2 (*i*Pr-CH₃), 25.8 (NC(CH₃)₂), 25.1 (*i*Pr-CH₃), 24.8 (*i*Pr-CH₃), 23.9 (*i*Pr-CH₃), 23.7 (*i*Pr-CH₃) ppm. ¹¹B NMR (160.5 MHz, C₆D₆): $\delta = -14.4$ (d, ¹J_{B-H} = 71.1 Hz) ppm. Note: The second ¹¹B resonance could not be detected. FT-IR (solid-state): $\tilde{\nu}$ (B–H) = 2353 cm⁻¹. HRMS LIFDI for $[C_{52}H_{72}B_2N_2]^+ = [M]^+$: *m/z*: calcd. 746.5876; found 746.5845.

Synthesis of 2

To a solution of II (50.0 mg, 67.1 μ mol) in benzene (2 mL) was added phenyl azide (9.58 mg, 80.5 μ mol) and the mixture stirred at room temperature for 5 min, resulting in a greenish yellow solution. All volatiles were removed *in vacuo* and the residue was washed with hexane (3 × 2 mL) and dried, yielding *anti*-2 as a pale yellow solid (42.3 mg, 49.0 μ mol, 73%). Single crystals suitable for X-ray diffraction analysis were obtained by slow evaporation of a saturated benzene solution at room temperature. *Note: Heating a solution of anti*-2 *at 60* °C *for 4 d led to quantitative isomerisation to syn*-2.

Analytical data for *anti*-2: ${}^{1}H{}^{11}B{}$ NMR (500.1 MHz, THF- d_8): $\delta = 7.61$ (t, ${}^{3}J = 7.8$ Hz, 1H, *p*-Ar-*H*), 7.51 (t, ${}^{3}J$ = 7.7 Hz, 1H, *p*-Ar-*H*), 7.47 (dd, ${}^{3}J$ = 7.8 Hz, ${}^{4}J$ = 1.5 Hz, 1H, *m*-Ar-*H*), 7.46 (dd, ${}^{3}J = 7.7$ Hz, ${}^{4}J = 1.7$ Hz, 1H, *m*-Ar-*H*), 7.45 (dd, ${}^{3}J = 7.8$ Hz, ${}^{4}J = 1.5$ Hz, 1H, *m*-Ar-*H*), 7.28 (dd, ${}^{3}J = 7.7$ Hz, ${}^{4}J = 1.7$ Hz, 1H, *m*-Ar-*H*), 7.10–7.02 (m, 4H, *o*-Ph-H + *m*-Ph-H), 6.91 (d, ${}^{3}J = 7.1$ Hz, 1H, DBA-H), 6.76 (d, ${}^{3}J = 7.1$ Hz, 1H, DBA-H), 6.67 (tt, ${}^{3}J = 6.9$ Hz, ${}^{4}J = 1.5$ Hz, 1H, *p*-Ph-H), 6.19 (dt, ${}^{3}J = 7.1$ Hz, ${}^{4}J = 1.0$ Hz, 1H, DBA-H), 6.17 (dt, ${}^{3}J = 7.1$ Hz, ${}^{4}J = 1.0$ Hz, 1H, DBA-H), 5.87 (t, ${}^{3}J = 7.3$ Hz, 2H, DBA-H), 4.69 (d, ${}^{3}J = 7.3$ Hz, 1H, DBA-*H*), 4.39 (sept, ${}^{3}J = 6.5$ Hz, 1H, *i*Pr-C*H*), 4.21 (d, ${}^{3}J = 7.3$ Hz, 1H, DBA-*H*), 3.52 (sept, ${}^{3}J = 6.6$ Hz, 1H, *i*Pr-CH), 3.42 (sept, ${}^{3}J = 6.6$ Hz, 1H, *i*Pr-CH), 3.34 (sept, ${}^{3}J = 6.7$ Hz, 1H, *i*Pr-CH), 2.27 (d, ${}^{2}J = 11.7$ Hz, 1H, CH₂), 2.21 (d, ${}^{2}J = 12.6$ Hz, 1H, CH₂), 2.14 (d, ${}^{2}J = 12.6$ Hz, 1H, CH₂), 1.85 (s, 3H, NC(CH₃)₂), 1.76 (s, 3H, NC(CH₃)₂), 1.73 (d, ${}^{2}J$ = 11.7 Hz, 1H, CH₂, overlapping with THF-d₈), 1.57 (s, 3H, C(CH₃)₂), 1.52 (s, 3H, C(CH₃)₂), 1.50 (s, 3H, C(CH₃)₂), 1.48 (s, 3H, C(CH₃)₂), 1.45 (d, ${}^{3}J$ = 6.6 Hz, 3H, *i*Pr-CH₃), 1.38 (d, ${}^{3}J$ = 6.7 Hz, 3H, *i*Pr-CH₃), 1.36 (d, ${}^{3}J = 6.6$ Hz, 3H, *i*Pr-CH₃), 1.31 (d, ${}^{3}J = 6.5$ Hz, 3H, *i*Pr-CH₃), 1.28 (d, ${}^{3}J = 6.6$ Hz, 3H, *i*Pr-C*H*₃), 1.14 (d, ³*J* = 6.7 Hz, 3H, *i*Pr-C*H*₃), 1.07 (s, 3H, NC(C*H*₃)₂), 0.95 (s, 3H, NC(C*H*₃)₂), 0.57 (d, ${}^{3}J = 6.6$ Hz, 3H, *i*Pr-CH₃), 0.47 (d, ${}^{3}J = 6.5$ Hz, 3H, *i*Pr-CH₃) ppm. ${}^{13}C{}^{1}H{}$ NMR (125.8 MHz, THF-d₈): δ = 235.7 (*C*_{carbene}, *identified by HMBC*), 230.5 (*C*_{carbene}, *identified by* HMBC), 168.4 (DBA-C_B, identified by HMBC), 164.4 (DBA-C_B, identified by HMBC), 161.3 (DBA-C_B, identified by HMBC), 158.7 (DBA-C_B, identified by HMBC), 156.1 (i-Ph-C), 150.5 (o-Ar-C), 148.1 (o-Ar-C), 147.6 (o-Ar-C), 147.4 (o-Ar-C), 136.2 (i-Ar-C), 136.2 (i-Ar-C), 130.8 (p-Ar-C), 130.2 (p-Ar-C), 129.7 (DBA-C), 129.2 (DBA-C), 128.8 (DBA-C), 128.6 (o-Ph-C), 127.2 (m-Ar-C), 126.8 (DBA-C), 126.5 (m-Ar-C), 126.2 (m-Ar-C), 125.6 (m-Ar-C), 122.0 (DBA-C), 120.9 (DBA-C), 119.9 (m-Ph-C), 119.8 (DBA-C), 119.7 (DBA-C), 119.6 (p-Ph-C), 80.1 (NC(CH₃)₂), 78.3 (NC(CH₃)₂), 56.6 (C(CH₃)₂), 56.2 (C(CH₃)₂), 53.3 (CH₂), 52.7 (CH₂), 36.4 (C(CH₃)₂), 32.9 (C(CH₃)₂), 31.7 (NC(CH₃)₂), 31.5 (C(CH₃)₂), 30.8 (*i*Pr-CH), 30.7 (NC(CH₃)₂), 29.6 (*i*Pr-CH), 29.5 (*i*Pr-CH), 29.2 (*i*Pr-CH), 28.8 (*i*Pr-CH₃), 28.5 (*i*Pr-CH₃), 27.5 (*i*Pr-CH₃), 27.3 (NC(CH₃)₂), 27.2 (NC(CH₃)₂), 26.5 (*i*Pr-CH₃), 25.3 (*C*(CH₃)₂, *overlapping with THF-d*₈, *identified by HSQC*), 24.7 (*i*Pr-CH₃), 23.9 (*i*Pr-CH₃), 23.9 (*i*Pr-CH₃), 23.8 (*i*Pr-CH₃) ppm. ¹¹B NMR (160.5 MHz, THF-*d*₈): $\delta = -1.3$ (br s), -4.4 (br s) ppm. FT-IR (solid-state): $\tilde{\nu}$ (N=N=N) = 1423, 1251 cm⁻¹. UV-vis (THF): $\lambda_{max} = 348$ nm, $\lambda_2 = 296$ nm. HRMS LIFDI for [C₅₈H₇₅B₂N₅]⁺ = [M]⁺: *m/z*: calcd. 863.6203; found 863.6186.

NMR data for syn-2: ${}^{1}H{}^{11}B{}$ NMR (500.1 MHz, THF-d₈): $\delta = 7.50$ (t, ${}^{3}J = 7.8$ Hz, 1H, *p*-Ar-*H*), 7.43 (dd, ${}^{3}J$ = 7.8 Hz, ${}^{4}J$ = 1.5 Hz, 1H, *m*-Ar-*H*), 7.39 (t, ${}^{3}J$ = 7.7 Hz, 1H, *p*-Ar-*H*), 7.37 (dd, ${}^{3}J = 7.8$ Hz, ${}^{4}J = 2.3$ Hz, 1H, *m*-Ar-*H*), 7.30 (dd, ${}^{3}J = 7.8$ Hz, ${}^{4}J = 1.5$ Hz, 1H, *m*-Ar-*H*), 7.27 (dd, ${}^{3}J = 8.5$ Hz, ${}^{4}J = 1.0$ Hz, 2H, *o*-Ph-H), 7.15–7.11 (m, 1H, DBA-*H*), 7.08– 7.03 (m, 3H, *m*-Ar-*H* + *m*-Ph-H), 7.03–7.01 (m, 1H, DBA-*H*), 6.67 (tt, ${}^{3}J = 7.2$ Hz, ${}^{4}J = 1.0$ Hz, 1H, p-Ph-H), 6.47–6.42 (m, 2H, DBA-H), 5.44 (dt, ${}^{3}J = 7.3$ Hz, ${}^{4}J = 1.4$ Hz, 1H, DBA-H), 5.39 (dt, ${}^{3}J = 7.3$ Hz, ${}^{4}J = 1.4$ Hz, 1H, DBA-H), 4.39 (dd, ${}^{3}J = 7.3$ Hz, ${}^{4}J = 0.8$ Hz, 1H, DBA-*H*), 4.17 (dd, ${}^{3}J = 7.3$ Hz, ${}^{4}J = 0.8$ Hz, 1H, DBA-*H*), 3.61 (sept, ${}^{3}J = 6.6$ Hz, 1H, *i*Pr-CH, overlapping with THF-d₈), 3.47 (sept, ${}^{3}J$ = 6.6 Hz, 1H, *i*Pr-CH), 3.42 (sept, ${}^{3}J$ = 6.6 Hz, 1H, *i*Pr-CH), 2.72 (d, ${}^{2}J$ = 12.1 Hz, 1H, CH₂), 2.40 (d, ${}^{2}J$ = 12.7 Hz, 1H, CH₂), 2.22 (d, ${}^{2}J$ = 12.7 Hz, 1H, CH₂), 2.11 (d, ${}^{2}J$ = 12.1 Hz, 1H, CH₂), 1.98 (s, 3H, C(CH₃)₂), 1.77 (s, 3H, C(CH₃)₂), 1.63 (s, 3H, C(CH₃)₂), 1.60 (s, 3H, NC(CH₃)₂), 1.56 (s, 3H, NC(CH₃)₂), 1.55 (s, 3H, C(CH₃)₂), 1,55 (s, 3H, NC(CH₃)₂), 1.53 (s, 3H, NC(CH₃)₂), 1.46 (d, ${}^{3}J = 6.6$ Hz, 3H, *i*Pr-CH₃), 1.45 (d, ${}^{3}J = 6.6$ Hz, 3H, *i*Pr-CH₃), 1.32 (d, ${}^{3}J = 6.6$ Hz, 3H, *i*Pr-CH₃), 1.18 (d, ${}^{3}J = 6.6$ Hz, 3H, *i*Pr-CH₃), 0.90 (d, ${}^{3}J = 6.6$ Hz, 6H, *i*Pr-CH₃), 0.57 (d, ${}^{3}J = 6.6$ Hz, 3H, *i*Pr-CH₃), -0.11 (d, ${}^{3}J = 6.6$ Hz, 3H, *i*Pr-CH₃) ppm. ${}^{13}C{}^{1}H$ NMR (125.8 MHz, THF-d₈): $\delta = 233.8$ (C_{Carbene}, identified by HMBC), 230.6 (C_{Carbene}, identified by HMBC), 164.2 (DBA-C_B, identified by HMBC), 163.8 (DBA-C_B, identified by HMBC), 163.6 (DBA-C_B, identified by HMBC), 162.8 (DBA-C_B, identified by HMBC), 155.3 (i-Ph-C), 150.7 (o-Ar-C), 147.9 (o-Ar-C), 146.9 (o-Ar-C), 146.8 (o-Ar-C), 136.6 (i-Ar-C), 136.0 (i-Ar-C), 131.3 (DBA-C), 130.7 (p-Ar-C), 130.2 (p-Ar-C), 128.7 (DBA-C), 128.6 (m-Ph-C), 128.4 (DBA-C), 128.3 (DBA-C), 127.3 (m-Ar-C), 126.9 (m-Ar-C), 126.0 (m-Ar-C), 125.6 (m-Ar-C), 121.2 (DBA-C), 121.0 (DBA-C), 120.3 (p-Ph-C), 120.1 (DBA-C), 119.9 (DBA-C), 119.3 (o-Ph-C), 80.3 (NC(CH₃)₂), 78.5 (NC(CH₃)₂), 56.7 (C(CH₃)₂), 56.7 (C(CH₃)₂), 54.1 (CH₂), 53.2 (CH₂), 35.1 (C(CH₃)₂), 34.2 (C(CH₃)₂), 34.1 (C(CH₃)₂), 31.0 (*i*Pr-CH), 30.8 (NC(CH₃)₂), 30.7 (NC(CH₃)₂), 29.6 (*i*Pr-CH), 29.6 (*i*Pr-CH), 29.5 (*i*Pr-CH₃), 29.4 (*i*Pr-CH + *i*Pr-CH₃), 28.5 (NC(CH₃)₂), 27.9 (NC(CH₃)₂), 27.2 (*C*(CH₃)₂), 27.1 (*i*Pr-CH₃), 26.1 (*i*Pr-CH₃), 24.9 (*i*Pr-CH₃), 24.7 (*i*Pr-CH₃), 24.2 (*i*Pr-CH₃), 24.1 (*i*Pr-*C*H₃) ppm. ¹¹B NMR (160.5 MHz, THF-*d*₈): $\delta = -1.2$ (br s), -4.1 (br s) ppm.

Synthesis of 3

II (30.0 mg, 40.3 µmol) and TEMPO (18.9 mg, 80.6 µmol) were combined in benzene (1 mL) and the suspension was stirred at room temperature for 30 min, resulting in a colourless reaction mixture. All volatiles were removed in vacuo and the residue was washed with hexane $(2 \times 0.5 \text{ mL})$ and dried, yielding **3** as a colourless solid (16.6 mg, 34.1 µmol, 85%). Single crystals suitable for X-ray diffraction analysis were obtained by slow evaporation of a saturated benzene solution at room temperature. Note: The NMR spectra showed the formation of a 77:23 mixture of anti-3 and syn-3. The NMR resonances of the two atropisomers were assigned on the basis of COSY, HSQC and HMBC spectra. ¹H{¹¹B} NMR (500.1 MHz, CD₂Cl₂) for anti-**3** (77%): $\delta = 9.41-9.36$ (m, 2H, DBA-H), 8.26-8.22 (m, 2H, DBA-H), 7.51-7.44 (m, 4H, DBA-H), 1.85–1.64 (m, 12H, CH₂), 1.34 (s, 12H, CH₃), 1.09 (s, 12H, CH₃) ppm; for syn-3 (23%): $\delta = 9.49-9.44$ (m, 2H, DBA-H), 8.17-8.13 (m, 2H, DBA-H), 7.51-7.44 (m, 4H, DBA-H), 1.85–1.64 (m, 6H, CH₂), 1.55–1.47 (m, 6H, CH₂), 1.33 (s, 12H, CH₃), 1.09 (s, 12H, CH₃) ppm. ¹³C{¹H} NMR (125.8 MHz, CD₂Cl₂) for *anti*-**3**: δ = 138.2 (Ar-*C*), 132.9 (Ar-*C*), 130.9 (Ar-C), 129.9 (Ar-C), 60.6 (NC(CH₃)₂), 40.0 (CH₂), 32.2 (CH₃), 20.4 (CH₃), 17.7 (CH₂) ppm; for *syn*-**3**: δ = 138.1 (Ar-*C*), 132.3 (Ar-*C*), 130.6 (Ar-*C*), 129.9 (Ar-*C*), 60.6 (N*C*(CH₃)₂), 40.0 (CH₂), 32.2 (CH₃), 20.4 (CH₃), 17.7 (CH₂) ppm. ¹¹B NMR (160.5 MHz, CD₂Cl₂): δ = 41.1 (br s) ppm. HRMS LIFDI for $[C_{30}H_{44}B_2N_2O_2]^+ = [M]^+$: *m/z*: calcd. 486.3583; found 486.3578.

Synthesis of 4-Cr

To a mixture of II (30.0 mg, 40.3 µmol) and [(MeCN)₃Cr(CO)₃] (10.4 mg, 40.3 µmol) was added THF (1 mL) and the reaction mixture was stirred at 60 °C for 5 d, resulting in a colour change to dark green. The reaction mixture was filtered and layered with hexane, resulting in crystallisation of the product. The crystals were collected by filtration, washed with benzene (2 × 1 mL) and hexane (3 × 1 mL) and dried *in vacuo*, yielding 4-Cr as black crystals (24.1 mg, 27.4 µmol, 68%). Single crystals suitable for X-ray diffraction analysis were obtained by slow evaporation of a saturated THF solution at room temperature. ¹H{¹¹B} NMR (500.1 MHz, THF-*d*₈): δ = 7.25–7.20 (m, 2H, DBA-*H*), 7.13 (t, ³*J* = 7.8 Hz, 2H, *p*-Ar-*H*), 6.84 (d, ³*J* = 7.8 Hz, 4H, *m*-Ar-*H*), 6.29–6.24 (m, 4H, DBA-*H*), 2.79 (sept, ³*J* = 6.4 Hz, 4H, *i*Pr-C*H*), 2.52 (s, 4H, C*H*₂), 2.41 (s, 12H, C(C*H*₃)₂), 1.51 (s, 12H, NC(C*H*₃)₂), 1.14 (d, ³*J* = 6.4 Hz, 12H, *i*Pr-C*H*₃), 0.18 (d, ³*J* = 64 Hz, 12H, *i*Pr-C*H*₃) ppm. ¹³C{¹H} NMR (125.8 MHz, THF-*d*₈): δ = 241.5 (CO), 235.2 (*C*_{carbene}, *identified by HMBC*), 146.8 (*o*-Ar-*C*), 137.8 (DBA-*C*), 134.0

(*i*-Ar-C), 129.5 (*p*-Ar-C), 126.3 (*m*-Ar-C), 119.2 (DBA-C), 108.8 (DBA-C_B), 82.1 (NC(CH₃)₂), 58.4 (C(CH₃)₂), 52.4 (CH₂), 31.4 (C(CH₃)₂), 30.8 (NC(CH₃)₂), 29.3 (*i*Pr-CH), 28.0 (*i*Pr-CH₃), 26.1 (*i*Pr-CH₃) ppm. ¹¹B NMR (160.5 MHz, THF-*d*₈): δ = 14.6 (br s) ppm. FT-IR (solid-state): $\tilde{\nu}$ (C=O) = 1882, 1777 cm⁻¹. UV-vis (THF): λ_{max} = 340 nm, λ_2 = 392 nm (shoulder), λ_3 = 602 nm (broad). HRMS LIFDI for [C₅₅H₇₀B₂CrN₂O₃]⁺ = [M]⁺: *m/z*: calcd. 880.4972; found 880.4968.

Synthesis of 4-Mo

To a mixture of II (30.0 mg, 40.3 µmol) and [(MeCN)₃Mo(CO)₃] (12.2 mg, 40.3 µmol) was added THF (1 mL) and the reaction mixture was stirred at 60 °C for 5 d, resulting in a colour change to dark green. The reaction mixture was filtered and layered with hexane, resulting in crystallisation of the product. The crystals were collected by filtration, washed with benzene $(2 \times 1 \text{ mL})$ and hexane $(3 \times 1 \text{ mL})$ and dried *in vacuo*, yielding **4-Mo** as black crystals. (21.3 mg, 23.0 µmol, 57%). Single crystals suitable for X-ray diffraction analysis were obtained by slow evaporation of a saturated THF solution at room temperature. ¹H{¹¹B} NMR (500.1 MHz, THF- d_8): $\delta = 7.31-7.26$ (m, 2H, DBA-H), 7.15 (t, ${}^{3}J = 7.8$ Hz, 2H, p-Ar-H), 6.86 $(d, {}^{3}J = 7.8 \text{ Hz}, 4\text{H}, m\text{-Ar-}H), 6.29-6.24 (m, 4\text{H}, \text{DBA-}H), 2.80 (sept, {}^{3}J = 6.4 \text{ Hz}, 4\text{H}, i\text{Pr-C}H),$ 2.50 (s, 4H, CH₂), 2.36 (s, 12H, C(CH₃)₂), 1.52 (s, 12H, NC(CH₃)₂), 1.14 (d, ${}^{3}J$ = 6.4 Hz, 12H, *i*Pr-CH₃), 0.18 (d, ${}^{3}J = 64$ Hz, 12H, *i*Pr-CH₃) ppm. ${}^{13}C{}^{1}H$ NMR (125.8 MHz, THF-d₈): δ = 234.5 (*C*_{Carbene}, *identified by HMBC*), 232.8 (CO), 146.8 (*o*-Ar-*C*), 137.1 (DBA-*C*), 134.1 (*i*-Ar-*C*), 129.5 (*p*-Ar-*C*), 126.3 (*m*-Ar-*C*), 118.9 (DBA-*C*), 112.6 (DBA-*C*_B), 82.2 (N*C*(CH₃)₂), 58.2 (C(CH₃)₂), 51.8 (CH₂), 31.9 (C(CH₃)₂), 30.7 (NC(CH₃)₂), 29.3 (*i*Pr-CH), 27.8 (*i*Pr-CH₃), 26.1 (*i*Pr-*C*H₃) ppm. ¹¹B NMR (160.5 MHz, THF-*d*₈): δ = 14.8 (br s) ppm. FT-IR (solid-state): $\tilde{\nu}$ (C=O) = 1892, 1784 cm⁻¹. UV-vis (THF): λ_{max} = 347 nm, λ_2 = 393 nm (shoulder), $\lambda_3 = 490 \text{ nm}$ (shoulder), $\lambda_4 = 593 \text{ nm}$ (broad). HRMS LIFDI for $[C_{55}H_{70}B_2MoN_2O_3]^+ = [M]^+$: *m*/*z*: calcd. 924.4651; found 924.4639.

Synthesis of 4-W

To a mixture of II (30.0 mg, 40.3 μ mol) and [(MeCN)₃W(CO)₃] (15.8 mg, 40.3 μ mol) was added THF (1 mL) and the reaction mixture was stirred at 60 °C for 5 d, resulting in a colour change to dark green. The reaction mixture was filtered and layered with hexane, resulting in

crystallisation of the product. The crystals were collected by filtration, washed with benzene (2 × 1 mL) and hexane (3 × 1 mL) and dried *in vacuo*, yielding **4-W** as black crystals (28.6 mg, 28.2 µmol, 70%). Single crystals suitable for X-ray diffraction analysis were obtained by slow evaporation of a saturated THF solution at room temperature. ¹H{¹¹B} NMR (500.1 MHz, THF-*d*₈): δ = 7.20–7.15 (m, 6H, DBA-*H* + *p*-Ar-*H*), 6.89 (d, ³*J* = 7.8 Hz, 4H, *m*-Ar-*H*), 6.21–6.16 (m, 4H, DBA-*H*), 2.79 (sept, ³*J* = 6.4 Hz, 4H, *i*Pr-C*H*), 2.49 (s, 4H, C*H*₂), 2.30 (s, 12H, C(C*H*₃)₂), 1.52 (s, 12H, NC(C*H*₃)₂), 1.15 (d, ³*J* = 6.4 Hz, 12H, *i*Pr-C*H*₃), 0.23 (d, ³*J* = 6.4 Hz, 12H, *i*Pr-C*H*₃) ppm. ¹³C{¹H} NMR (125.8 MHz, THF-*d*₈): δ = 233.8 (*C*_{carbene}, *identified by HMBC*), 222.3 (s + satellites, ¹*J*_{W-C} = 191.2 Hz, CO), 146.8 (*o*-Ar-C), 137.1 (DBA-C), 134.1 (*i*-Ar-C), 129.6 (*p*-Ar-C), 126.4 (*m*-Ar-C), 119.2 (DBA-C), 108.8 (DBA-C_B), 82.4 (NC(CH₃)₂), 58.4 (*C*(CH₃)₂), 51.8 (CH₂), 32.0 (C(CH₃)₂), 30.7 (NC(CH₃)₂), 29.3 (*i*Pr-CH), 27.9 (*i*Pr-CH₃), 26.1 (*i*Pr-CH₃) ppm. ¹¹B NMR (160.5 MHz, THF-*d*₈): δ = 13.5 (br s) ppm. FT-IR (solid-state): $\tilde{\nu}$ (C=O) = 1886, 1771 cm⁻¹. UV-vis (THF): λ_{max} = 342 nm, λ_2 = 387 nm (shoulder), λ_3 = 595 nm (broad). HRMS LIFDI for [C₅₅H₇₀B₂N₂O₃W]⁺ = [M]⁺; *m/z*: calcd. 1012.5102; found 1012.5090.

NMR spectra of isolated compounds



Figure S1. ${}^{1}H{}^{11}B{}$ NMR spectrum of 1 in C₆D₆.



Figure S2. ¹³C{¹H} NMR spectrum of 1 in C₆D₆.



Figure S3. ¹¹B NMR spectrum of 1 in C_6D_6 .



Figure S4. ${}^{1}H{}^{11}B{}$ NMR spectrum of *anti*-2 in THF-d₈. The additional smaller resonances are those of the *syn* isomer, which is formed very slowly at rt.



Figure S5. ¹³C{¹H} NMR spectrum of *anti*-2 in THF-d₈. The additional smaller resonances are those of the *syn* isomer, which is formed very slowly at rt.



Figure S6. ¹¹B NMR spectrum of *anti-2* in THF-d₈.



Figure S7. ${}^{1}H{}^{11}B{}$ NMR spectrum of *syn*-2 in THF-d₈.



Figure S8. ${}^{13}C{}^{1}H$ NMR spectrum of *syn*-2 in THF-d₈.



Figure S9. ¹¹B NMR spectrum of *syn-*2 in THF-d₈.



Figure S12. ${}^{1}H{}^{11}B{}$ NMR spectrum of 3 in CD₂Cl₂ (• *anti*-3, • *syn*-3).



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Figure S14. ¹¹B NMR spectrum of 3 in CD₂Cl₂.



Figure S15. ${}^{1}H{}^{11}B{}$ NMR spectrum of 4-Cr in THF-d₈.



Figure S16. ¹³C{¹H} NMR spectrum of 4-Cr in THF-d₈.



Figure S17. ¹¹B NMR spectrum of 4-Cr in THF-d₈.



Figure S18. ${}^{1}H{}^{11}B{}$ NMR spectrum of 4-Mo in THF-d₈.



Figure S19. ${}^{13}C{}^{1}H$ NMR spectrum of 4-Mo in THF-d₈.



Figure S20. ¹¹B NMR spectrum of 4-Mo in THF-d₈.



Figure S21. ${}^{1}H{}^{11}B{}$ NMR spectrum of 4-W in THF-d₈.



Figure S22. ${}^{13}C{}^{1}H$ NMR spectrum of 4-W in THF-d₈.



Figure S23. ¹¹B NMR spectrum of 4-W in THF-d₈.

IR spectra



Figure S24. Solid-state IR spectrum of 1.



Figure S25. Solid-state IR spectrum of *anti-2*.



Figure S26. Solid-state IR spectrum of 4-Cr.



Figure S27. Solid-state IR spectrum of 4-Mo.



Figure S28. Solid-state IR spectrum of 4-W.

UV-vis spectra



Figure S29. UV-vis absorption spectrum of *anti-2* in THF at 23 °C. $\lambda_{max} = 348$ nm, $\lambda_2 = 296$ nm.



Figure S30. Overlay of UV-vis absorption spectra of **4-Cr** (normal), **4-Mo** (dashed) and **4-W** (dotted) in THF at 23 °C. **4-Cr**: $\lambda_{max} = 340$ nm, $\lambda_2 = 392$ nm (shoulder), $\lambda_3 = 602$ nm (broad); **4-Mo**: $\lambda_{max} = 347$ nm, $\lambda_2 = 393$ nm (shoulder), $\lambda_3 = 490$ nm (shoulder), $\lambda_4 = 593$ nm (broad); **4-W**: $\lambda_{max} = 342$ nm, $\lambda_2 = 387$ nm (shoulder), $\lambda_3 = 595$ nm (broad).

X-ray crystallographic data

The crystal data of 1, *anti*-2 and 4-Cr were collected on a *XtaLAB Synergy Dualflex HyPix* diffractometer with a Hybrid Pixel array detector and multi-layer mirror monochromated Cu_{Ka} radiation. The crystal data of *anti*-3, 4-Mo and 4-W were collected on a *Bruker D8 Quest* diffractometer with a CMOS area detector and multi-layer mirror monochromated Mo_{Ka} radiation. The structures were solved using the intrinsic phasing method,³ refined with the ShelXL program⁴ and expanded using Fourier techniques. All non-hydrogen atoms were refined anisotropically. Hydrogen atoms were included in structure factor calculations. All hydrogen atoms were assigned to idealised geometric positions.

Crystallographic data have been deposited with the Cambridge Crystallographic Data Centre as supplementary publication numbers CCDC-2306744 (1), 2306745 (*anti-2*), 2306746 (*anti-3*), 2306747 (**4-Cr**), 2306748 (**4-Mo**), 2306749 (**4-W**). These data can be obtained free of charge from The Cambridge Crystallographic Data Centre *via* www.ccdc.cam.ac.uk/data_request/cif.

Crystal data for 1: $C_{52}H_{72}B_2N_2$, $M_r = 746.73$, clear colourless block, $0.120 \times 0.100 \times 0.030 \text{ mm}^3$, orthorhombic space group *Pbca*, a = 13.89720(10) Å, b = 15.60870(10) Å, c = 41.3288(3) Å, $\alpha = 90^\circ$, $\beta = 90^\circ$, $\gamma = 90^\circ$, V = 8964.93(11) Å³, Z = 8, $\rho_{calcd} = 1.107 \text{ g} \cdot \text{cm}^{-3}$, $\mu = 0.460 \text{ mm}^{-1}$, F(000) = 3264, T = 100(2) K, $R_1 = 0.0595$, $wR_2 = 0.1287$, 8508 independent reflections $[2\theta \le 140.15^\circ]$ and 524 parameters.



Figure S31. Solid-state structure of **1**. Atomic displacement ellipsoids represented at 50%. Ellipsoids of ligand periphery and hydrogen atoms except B–H and CAAC–H omitted for clarity.

Crystal data for *anti-2*: C₇₉H₉₆B₂N₅, $M_r = 1137.22$, yellow block, $0.220 \times 0.200 \times 0.080$ mm³, triclinic space group $P\overline{1}$, a = 11.47910(10) Å, b = 13.02450(10) Å, c = 22.38830(10) Å, $\alpha = 79.3390(10)^{\circ}$, $\beta = 86.6270(10)^{\circ}$, $\gamma = 86.6760(10)^{\circ}$, V = 3280.00(4) Å³, Z = 2, $\rho_{calcd} = 1.151$ g·cm⁻³, $\mu = 0.495$ mm⁻¹, F(000) = 1230, T = 100(2) K, $R_l = 0.0383$, $wR_2 = 0.0962$, 12419 independent reflections $[2\theta \le 140.142^{\circ}]$ and 791 parameters.



Figure S32. Solid-state structure of *anti-2*. Atomic displacement ellipsoids represented at 50%. Ellipsoids of ligand periphery and hydrogen atoms except DBA–H…Dipp omitted for clarity.

Crystal data for *anti-***3**: C₃₆H₅₀B₂N₂O₂, $M_r = 564.40$, colourless block, 0.513×0.226×0.215 mm³, triclinic space group P $\overline{1}$, a = 7.7792(17) Å, b = 10.300(3) Å, c = 10.4076(19) Å, $\alpha = 78.833(8)^\circ$, $\beta = 82.411(11)^\circ$, $\gamma = 78.433(9)^\circ$, V = 797.8(3) Å³, Z = 1, $\rho_{calcd} = 1.175$ g·cm⁻³, $\mu = 0.071$ mm⁻¹, F(000) = 306, T = 100(2) K, $R_l = 0.0477$, $wR_2 = 0.0986$, 3138 independent reflections [2 $\theta \le 52.042^\circ$] and 194 parameters.



Figure S33. Solid-state structure of *anti*-**3**. Atomic displacement ellipsoids represented at 50%. Ellipsoids of ligand periphery and hydrogen atoms omitted for clarity.

Crystal data for 4-Cr: C₅₉H₇₈B₂CrN₂O₄, $M_r = 952.85$, clear brown block, 0.350×0.130×0.090 mm³, monoclinic space group $P2_1/n$, a = 11.25260(10) Å, b = 22.8684(2) Å, c = 19.7806(2) Å, $\beta = 93.8830(10)^\circ$, V = 5078.44(8) Å³, Z = 4, $\rho_{calcd} = 1.246$ g·cm⁻³, $\mu = 2.229$ mm⁻¹, F(000) = 2048, T = 100(2) K, $R_1 = 0.0375$, $wR_2 = 0.0981$, 9635 independent reflections [$2\theta \le 140.148^\circ$] and 629 parameters.



Figure S34. Solid-state structure of **4-Cr**. Atomic displacement ellipsoids represented at 50%. Ellipsoids of ligand periphery and hydrogen atoms omitted for clarity.

Refinement details for 4-Mo: Refined as a two-component twin. The BASF parameter was refined to 11.8%. Two outlying reflections affected by the beamstop were omitted: 1 1 0 and - 8 0 12. The ADPs of the boron atoms, distorted by their coordination to the heavy Mo atom, were restricted to similarity with the neighbouring carbon atoms using SIMU.

Crystal data for 4-Mo: C₅₉H₇₈B₂MoN₂O₄, $M_r = 996.79$, black block, 0.350×0.130×0.090 mm³, monoclinic space group Pc, a = 19.1301(8) Å, b = 10.9095(4) Å, c = 24.8499(11) Å, $\beta = 93.027(2)^\circ$, V = 5178.9(4) Å³, Z = 4, $\rho_{calcd} = 1.278$ g·cm⁻³, $\mu = 0.302$ mm⁻¹, F(000) = 2120, T = 100(2) K, $R_1 = 0.0804$, $wR_2 = 0.1433$, Flack parameter = 0.05(4), 13327 independent reflections [$2\theta \le 54.264^\circ$] and 1258 parameters.



Figure S35. Solid-state structure of **4-Mo**. Atomic displacement ellipsoids represented at 50%. Ellipsoids of ligand periphery and hydrogen atoms omitted for clarity.

Refinement details for 4-W: Refined as a two-component inversion twin. The BASF parameter was refined to 43.2%. Ten low-resolution reflections that were affected by the beamstop were removed from refinement. The ADPs of the boron atoms, distorted by their coordination to the heavy W atom, were restrained to similarity with the neighbouring carbon atoms using SIMU. An ISOR restraint was additionally applied to B8_10. In each of the two complexes the ADP of the carbonyl carbon atom C4, distorted by its coordination to the heavy W atom, were restrained to SIMU.

Crystal data for 4-W: C₅₉H₇₈B₂N₂O₄W, $M_r = 1084.70$, black block, $0.462 \times 0.304 \times 0.210 \text{ mm}^3$, monoclinic space group Pc, a = 19.1470(6) Å, b = 10.9198(4) Å, c = 24.8088(10) Å, $\beta = 93.0270(10)^\circ$, V = 5179.8(3) Å³, Z = 4, $\rho_{calcd} = 1.391 \text{ g} \cdot \text{cm}^{-3}$, $\mu = 2.279 \text{ mm}^{-1}$, F(000) = 2248, T = 100(2) K, $R_1 = 0.0234$, $wR_2 = 0.0482$, Flack parameter = 0.432(5), 20179 independent reflections $[2\theta \le 52.044^\circ]$ and 1258 parameters.



Figure S36. Solid-state structure of **4-W**. Atomic displacement ellipsoids represented at 50%. Ellipsoids of ligand periphery and hydrogen atoms omitted for clarity.

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