## **Electronic Supplementary Information**

## Preparation and electrochemical properties of NiMn-LDH with

## petal-like lamellar structure derived from Mn MOF-74

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Fig. S1 The full XPS spectra of NiMn-LDH 3 and O-LDH samples.



Fig. S2 (a)  $N_2$  adsorption/desorption isotherms of Mn MOF-74, NiMn-LDH 3 and O-LDH; (b) corresponding pore size distribution curves of Mn MOF-74, NiMn-LDH 3 and O-LDH.

Binding energy (eV) Relative peak area	530.8 O <sup>2–</sup>	531.6 OH <sup>-</sup>	532.9 Н <sub>2</sub> О
NiMn-LDH	2196.6	14645.2	10916.8
O-LDH	2728.7	11437.8	10036.9

Table S1 O 1s peak for NiMn-LDH 3 and O-LDH materials.

 Table S2 Mn 2p peak for NiMn-LDH 3 and O-LDH materials.

Binding energy	641.3	642.6	643.9
Relative peak area	Mn <sup>2+</sup>	Mn <sup>3+</sup>	Mn <sup>4+</sup>
NiMn-LDH	686.3	527.6	2216.6
O-LDH	1444.2	1105.1	2411.7

	Relative	Relative		Relative	Relative	Relative
	content	content	OH <sup>-</sup> /O <sup>2-</sup>	content	content	content
	of O <sup>2–</sup>	of OH⁻		of $Mn^{2+}$	of Mn <sup>3+</sup>	of Mn <sup>4+</sup>
NiMn-LDH	7.9%	52.8%	6.7	20.0%	15.4%	64.6%
O-LDH	11.3%	47.3%	4.2	29.1%	22.3%	48.6%

 Table S3 The analysis results of XPS for NiMn-LDH 3 and O-LDH materials.

Table S4 Pore structure parameters of Mn MOF-74, NiMn-LDH 3 and O-LDH materials.

Materials	$S_{BET} (m^2/g)$	V <sub>total</sub> (cm <sup>3</sup> /g)	Average pore diameter (nm)
Mn MOF-74	20.25	0.04	13.43
NiMn-LDH 3	46.10	0.19	27.16
O-LDH	111.33	0.35	20.40



Fig. S3 Cyclic Voltammetry (CV) curves of Mn MOF-74/NF (a), NiMn-LDH 1/NF(b), NiMn-LDH 2/NF (c), NiMn-LDH 4/NF (d) and NiMn-LDH 5/NF (e) electrodesatvariousscanrates(10-40mV $s^{-1}$ ).



**Fig. S4** Galvanostatic charge-discharge (GCD) curves of Mn MOF-74/NF (a), NiMn-LDH 1/NF (b), NiMn-LDH 2/NF (c), NiMn-LDH 4/NF (d) and NiMn-LDH 5/NF (e) electrodes at various current densities (1-30 A  $g^{-1}$ ).



Fig. S5 the equivalent circuit fitting diagram of the EIS data.

Electrode materials	Specific capacitance	Current density	Rate capability	Reference
NiMn-LDH/PC	1634 F g <sup>-1</sup>	1 A g <sup>-1</sup>	60.5% at 10 A g <sup>-1</sup>	1
Ni/Mn LDHs microspheres	1379 F g <sup>-1</sup>	1 A g <sup>-1</sup>	75.1% at 50 mV s <sup>-1</sup>	2
Ni-Mn LDH/Co <sub>3</sub> O <sub>4</sub> on carbon paper	1327 F g <sup>-1</sup>	1 A g <sup>-1</sup>	70.9% at 50 mV s <sup>-1</sup>	3
Ni-Mn LDH@Co <sub>3</sub> O <sub>4</sub> on nickel foam	607.9 F g <sup>-1</sup>	0.5 A g <sup>-1</sup>	73% at 5 A g <sup>-1</sup>	4
Mn <sub>3</sub> O <sub>4</sub> /NiMoO <sub>4</sub> @NiCo LDH on carbon cloth	815 F g <sup>-1</sup>	1 A g <sup>-1</sup>	76.92% at 5 A g <sup>-1</sup>	5
NiMn-LDH/Mxene	1575 F g <sup>-1</sup>	0.5 A g <sup>-1</sup>	91.8% at 2 A g <sup>-1</sup>	6
ZIF-9@CoAl LDH	702.7 F g <sup>-1</sup>	1 A g <sup>-1</sup>	75% at 8 A g <sup>-1</sup>	7

**Table S5** Comparison of supercapacitor electrochemical performance between thepreviously reported materials and the active material O-LDH.

O-LDH	1875 F g <sup>-1</sup>	1 A g <sup>-1</sup>	71.7% at 3 A $g^{\text{-}1}$	This work
NiMn-LDH/CNTs/rGO	1268 F g <sup>-1</sup>	1 A g <sup>-1</sup>	69.4% at 10 A g <sup>-1</sup>	10
KCu <sub>7</sub> S <sub>4</sub> @NiMn LDHs	733.8 F g <sup>-1</sup>	1 A g <sup>-1</sup>	84.8% at 2.5 A g <sup>-1</sup>	9
NiMnMg-LDH	1772 F g <sup>-1</sup>	1 A g <sup>-1</sup>	74% at 3 A g <sup>-1</sup>	8

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