

## *Electronic Supplementary Information*

### **Photochromism and single-component white light emission from a metalloviologen complex based on 1,5-naphthyridine**

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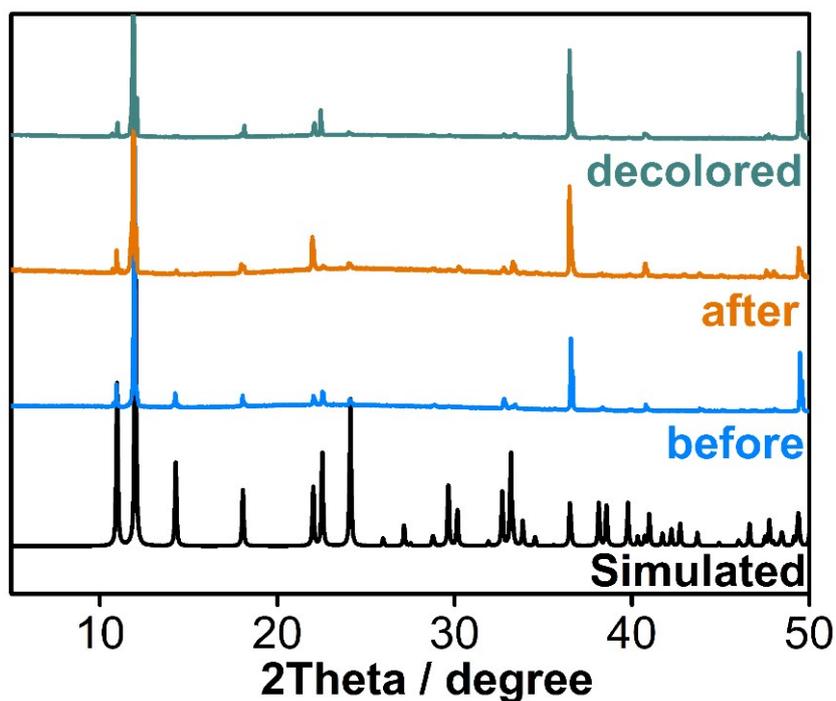
## Additional graphics

**Table S1.** Crystal data and structural refinements for compound **1**.

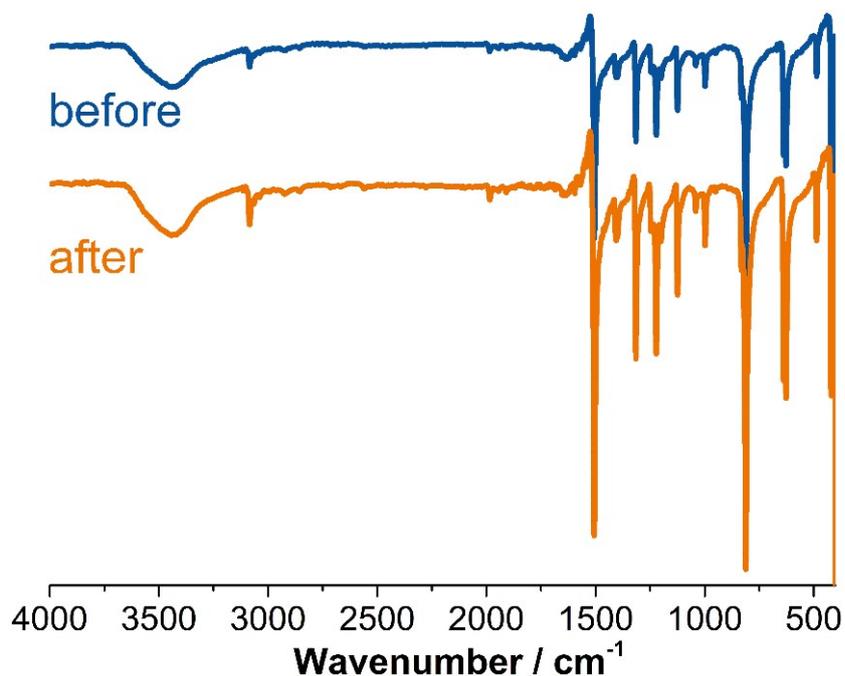
	<b>Compound 1</b>
<b>CCDC</b>	2307757
<b>Formula</b>	CdCl <sub>2</sub> C <sub>8</sub> H <sub>6</sub> N <sub>2</sub>
<b>Mr</b>	313.45
<b>Crystal size (mm<sup>3</sup>)</b>	0.42*0.17*0.1
<b>Crystal system</b>	monoclinic
<b>Space group</b>	C2/m
<b><i>a</i> (Å)</b>	15.1638(14)
<b><i>b</i> (Å)</b>	3.8071(4)
<b><i>c</i> (Å)</b>	8.2899(7)
<b><math>\alpha</math> (deg)</b>	90
<b><math>\beta</math> (deg)</b>	103.338(5)
<b><math>\gamma</math> (deg)</b>	90
<b><i>V</i> (Å<sup>3</sup>)</b>	465.67(8)
<b><i>D</i><sub>calcd</sub> (g/cm<sup>3</sup>)</b>	2.236
<b><i>Z</i></b>	2
<b><i>F</i>(000)</b>	300
<b>Abs coeff (mm<sup>-1</sup>)</b>	2.865
<b><i>R</i><sub>1</sub><sup>a</sup></b>	0.0331(478)
<b><math>\omega R</math><sub>2</sub><sup>b</sup></b>	0.0903(480)
<b>GOF on <i>F</i><sup>2</sup></b>	1.177

$${}^a R_1 = \sum ||F_o| - |F_c|| / \sum |F_o|;$$

$${}^b \omega R_2 = \{ \sum \omega [(F_o)^2 - (F_c)^2]^2 / \sum \omega [(F_o)_2]^2 \}^{1/2}.$$



**Fig. S1** PXRD patterns of compound **1**: **simulated**, simulated data using single-crystal data; **before**, measured data for as-synthesized samples; **after**, measured data for colored samples; **decolorized**, measured data for its heated sample at 200 °C for 1 day.



**Fig. S2** IR spectra of compound **1**: **before**, measured data for as-synthesized samples; **after**, measured data for colored samples.

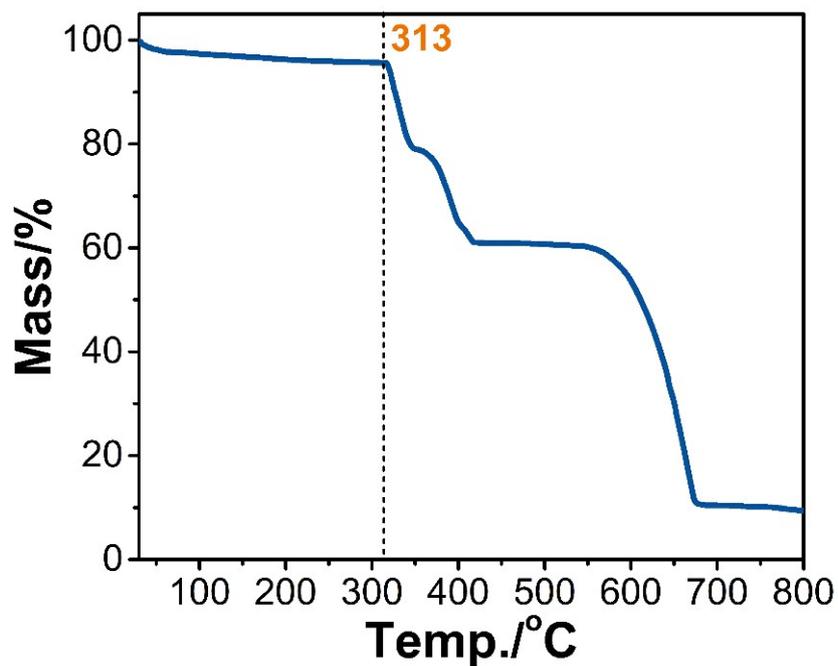


Fig. S3 Thermogravimetric curve of compound **1** under N<sub>2</sub> with heating rate of 10 °C/min.

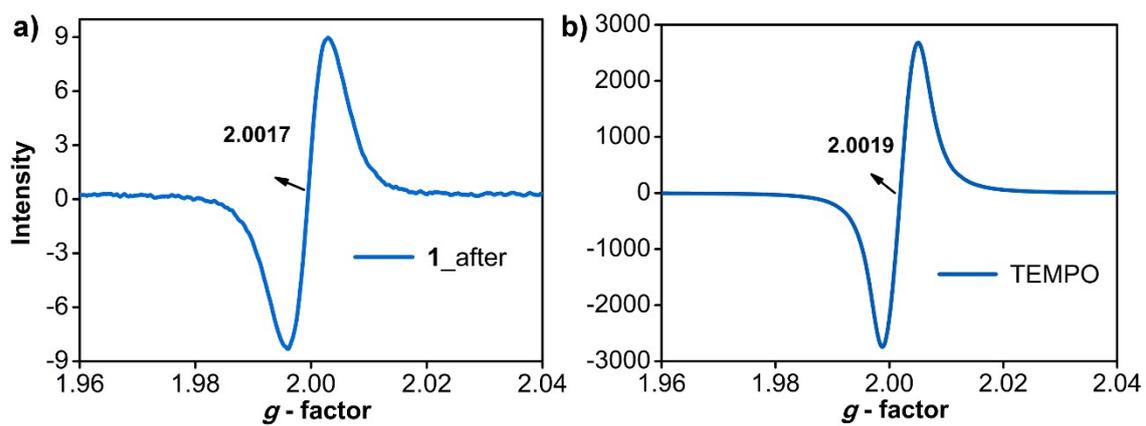
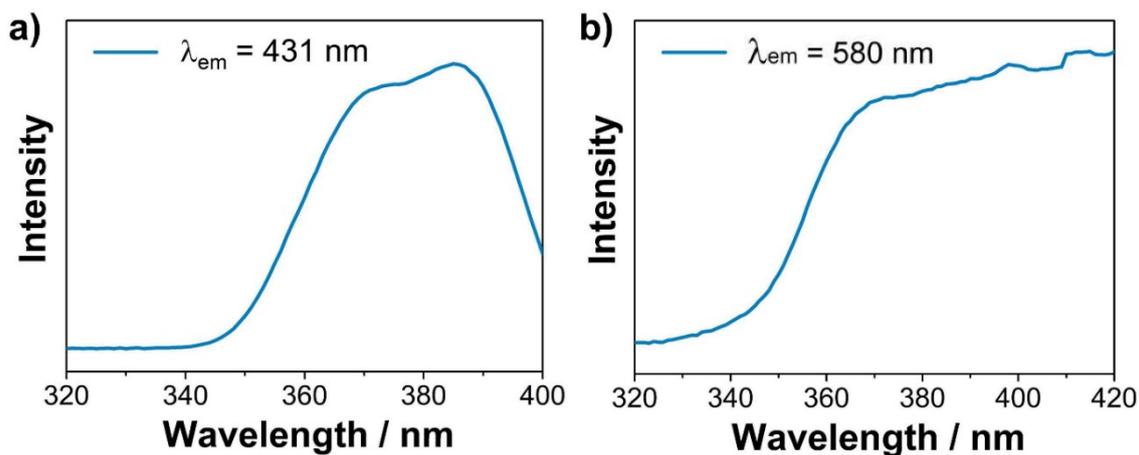
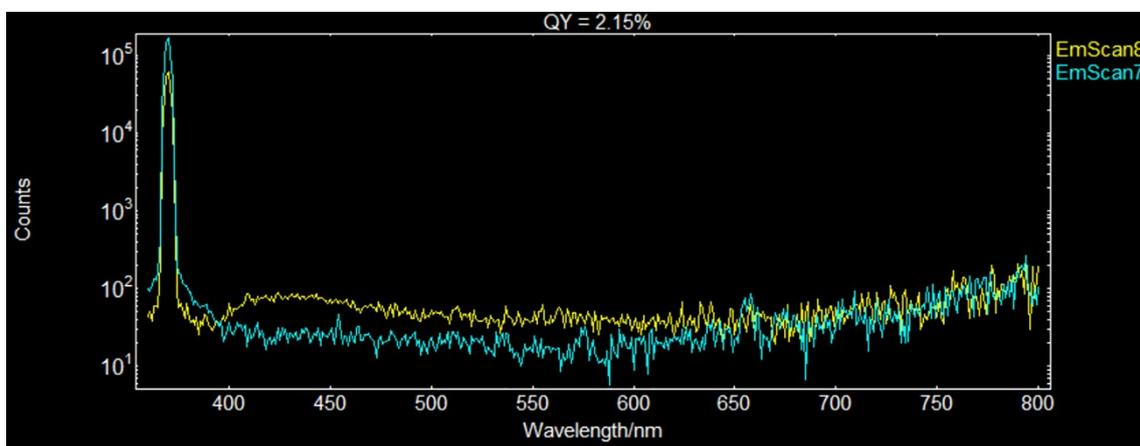


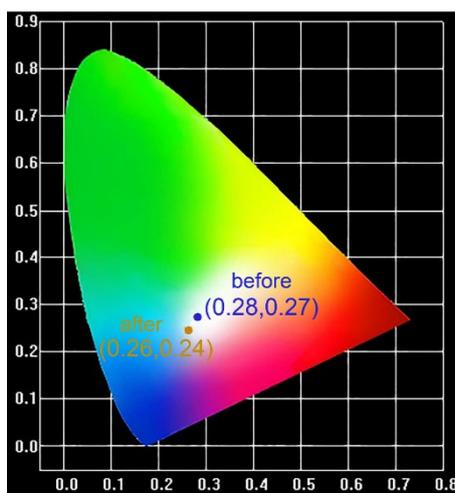
Fig. S4 ESR spectra of irradiated **1** (26 mg, a) and **TEMPO** (0.3 mg, b) in the solid state under the same test conditions.



**Fig. S5** The excitation spectra of fresh solid-state sample **1** monitored at 431 nm (a) and 580 nm (b), respectively.



**Fig. S6** Spectral diagram of quantum yield test of **1** at  $\lambda_{ex} = 370 \text{ nm}$ .



**Fig. S7** The CIE maps of **1** before and after irradiation at excitation wavelength of 365 nm.

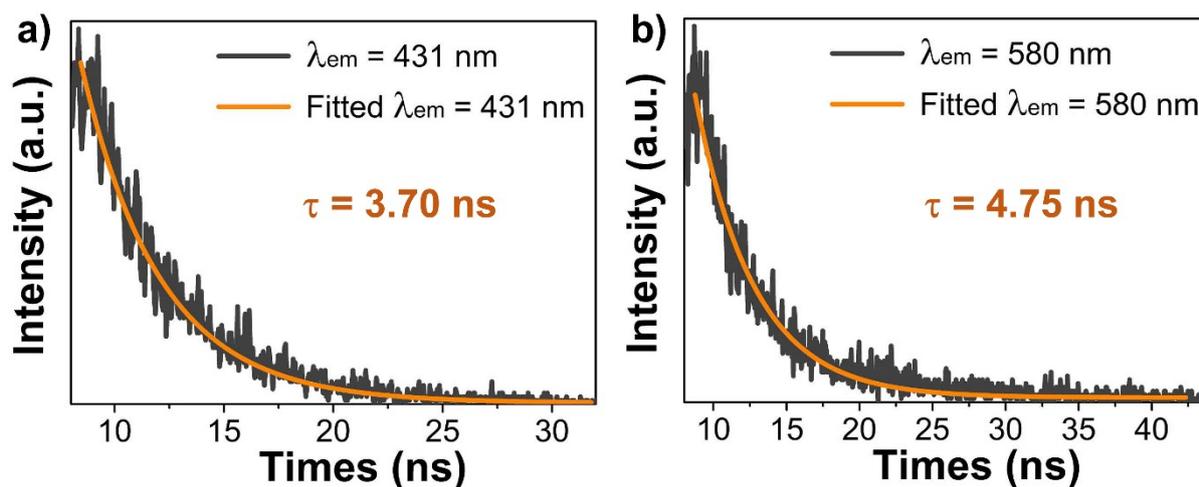


Fig. S8 Lifetime curves of **1** monitored at 431 nm (a) and 580 nm (b), respectively.

**Table S2.** The CIE coordinates, Color Temperature (CT) and Color-rendering Index (CRI) of fresh solid-state sample **1** at different excitation wavelength from 340 to 370 nm.

$\lambda_{\text{ex}}$ of Complex <b>1</b>	CIE	CT (T/K)	CRI
340 nm	(0.401, 0.393)	3615	82.54
350 nm	(0.371, 0.366)	4191	87.08
360 nm	(0.300, 0.293)	7981.	<b>94.04</b>
365 nm_before	(0.283, 0.272)	10640	<b>92.94</b>
365 nm_after	(0.262, 0.244)	19603	<b>92.49</b>
370 nm	(0.269, 0.252)	15718	<b>92.00</b>

**Table 3.** Typical part single-component white light emitting compounds and CRI values at specific excitation wavelengths ( $\lambda_{\text{ex}}$ ).

Compounds <sup>ref</sup>	CRI values	$\lambda_{\text{ex}}$ (nm)
(H <sub>2</sub> DABCO)(Pb <sub>2</sub> Cl <sub>6</sub> ) <sup>1</sup>	96	300
Ba <sub>2</sub> [Sn(OH) <sub>6</sub> ][B(OH) <sub>4</sub> ] <sub>2</sub> <sup>2</sup>	94.1	283
<b>Compound 1 [this work]</b>	<b>94.04</b>	<b>360</b>
[Mg <sub>3</sub> (OH) <sub>2</sub> (1,4-NDC) <sub>2</sub> (dppe)(H <sub>2</sub> O)] <sup>3</sup>	93.12	380
{[Zn(bpdo)(fum)(H <sub>2</sub> O) <sub>2</sub> ]} <sub>n</sub> <sup>4</sup>	92.1	370
(C <sub>6</sub> H <sub>5</sub> C <sub>2</sub> H <sub>4</sub> NH <sub>3</sub> ) <sub>2</sub> PbBr <sub>2</sub> Cl <sub>2</sub> <sup>5</sup>	91	385

$(C_5H_{14}N_2)PbBr_4^6$	90	330
$(C_7H_{16}N)PbBr_3^6$	89	330
$(C_6H_{14}N)PbBr_3^6$	88	330
1-(4-carboxyphenyl)-1,2,3-triazole <sup>7</sup>	88	370
$(C_5H_{14}N_2)_2Pb_3Br_{10}^6$	86	330
2-MOP <sup>8</sup>	86	383
$[H_2DABCO][Ag_2Br_4(DABCO)]^9$	85	376
$(3APr)PbCl_4^{10}$	85	330
$(C_6H_{16}N_2)PbBr_4^6$	84	330
$(2meptH_2)PbCl_4^{11}$	84	330
1-(4-acetylphenyl)-1,2,3-triazole <sup>7</sup>	83	376
$(3APr)PbBr_4^{10}$	83	330
$[DMEDA]PbCl_4^{12}$	78	365
$(C_6H_{16}N_2)_3Pb_2Br_{10}^6$	77	330
$(3APr)PbI_4^{10}$	77	330
$(C_6H_{16}N_2)PbBr_4^6$	76	330
$[DMPDA]PbCl_4^{12}$	75	377

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