

Supporting Information (SI) for Publication

Bimolecular sinks of Criegee intermediates derived from Hydrofluoroolefins – A Computational Analysis

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S1. Breakdown of different types of rate constants

S1.1 Conditions Applied to Calculations determined using MESMER

The purpose of this section is to provide a summary of **all MESMER calculations used** in this *HFO-sCI + co-reactants* study and this section outlines: the different *grainsizes* used for that reaction system; what channels in that system are barrierless; what channels do not use Eckart Functions; and the temperatures and pressures applied to that system.¹ Determination of the rate constants (k_{ME}) and product branching ratio (Γ_{THEO}) values are vital in determining the atmospheric chemistry of each reaction. For reactions that do not proceed via intermediate product(s) (referred to a single-step reactions) only one MESMER calculation, with a single grain size, is required to determine both the k_{ME} & Γ_{THEO} values. One examples of a reaction series which does not produce intermediate products is the sCI + MeOH reactions. However, the determination of k_{ME} & Γ_{THEO} values of multi-step reaction systems (e.g. sCI + HCHO reactions) are more complex and often require splitting MESMER calculations into two operations: a full calculation involving all stationary point structures of the reaction system to determine the Γ_{THEO} values, which are calculated using all stationary point structures of the potential energy surface (reactants, pre-reaction complexes, transition states, the intermediate products and post-reaction complexes); and those used to calculate the k_{ME} values, which only contain the stationary points in the first step of the reaction prior to the fragmentation of any intermediate product. Systems which involve larger numbers of atoms, and/or a larger number of stationary points are more computationally costly to calculate and so the grainsize for MESMER calculation is adjusted to provide greater computational efficiency. The grainsize used to calculate the k_{ME} & Γ_{THEO} values for each reaction system in this study is found in *Table S1*.

Several other factors are also noted in *Table S1* including at what temperatures and pressures these calculations were ran and whether any *barrierless* pathways and/or Eckart Functions were included into these MESMER calculations. Some systems do include bimolecular association and/or unimolecular dissociation channels that proceed between reactant(s) and product(s) via a asymptotic minimum energy pathway, with no local maximum. To determine whether these *barrierless* unimolecular dissociations contribute significantly to the fragmentation of intermediate products, a *reverse ILT method* needs to be included in the MESMER calculations of the Γ_{THEO} values. The reason that the dipole-dipole capture limits of these processes are noted in *Table S1* is because they are required to determine the Γ_{THEO} values of final products that emerge from unimolecular dissociation channels.^{2,3} For more details on the dipole-dipole capture limit (k_{d-d}) see Section S3. Often it becomes impractical to calculate an Eckart function for certain very low transition state barriers and so, in *Table S1*, it is noted whether a *Eckart* tunneling function has been excluded from MESMER calculations and to which transition state barriers this *Eckart* tunneling function would have applied. The author also notes, in *Table S1*, under what range of temperature and pressure conditions these results were obtained. To determine the accuracy of the results calculated here, some additional temperature and pressure conditions are only applied to certain reactions, where existing k_{EXP} & Γ_{EXP} results from the literature have been obtained using equivalent conditions.

Table S1: The conditions applied to each reaction system calculated using MESMER: Including the grain sizes used for calculations; Transition States that do not apply an Eckart function; Reaction channels without TS minima referred to as “Barrierless”; and the Temperatures and Pressures applied to each system.

Reaction system	Grain size to determine k_{ME}	TSs without Eckart Functions	Barrierless channels and their k_{d-d} value	Grainsize to determine Γ_{THEO}	Temperatures (K) and Pressures (Torr)
sCI 1 + HCHO	grainsize 10	N/A	N/A	grainsize 15	At 760 Torr: 200, 275, 298, 325 & 400 K
sCIs 2 & 3 + HCHO, and sCI 1 + CF ₃ CHO	grainsize 10 (grainsize 5 sCI 3 + HCHO)	N/A	sCI 1 + CF ₃ CHO TSc 1 & 2 ($k_{d-d} \sim 6.50 \times 10^{-10} \text{ cm}^3 \text{ s}^{-1}$)	grainsize 40.	At 760 Torr: 200, 275, 298, 325 & 400 K
sCIs 2 & 3 + HCHO, and sCI 1 + CF ₃ CFO	sCI 5 + HCHO – grainsize 5 sCI 1 + CF ₃ CFO – grainsize 10	sCI 26 + HCHO TSc	sCI 4 + HCHO TSc ($k_{d-d} \sim 7.20 \times 10^{-10} \text{ cm}^3 \text{ s}^{-1}$)	grainsize 40.	At 760 Torr: 200, 275, 298, 325 & 400 K
sCI 1 + SO ₂	N/A	N/A	sCI 1 + SO ₂ ($k_{d-d} \sim 7.25 \times 10^{-10} \text{ cm}^3 \text{ s}^{-1}$)	grainsize 20	At 760 Torr: 200, 275, 298, 325 & 400 K
sCIs 2 & 3 + SO ₂	grainsize 10	sCI 2 + SO ₂	sCI 3 + SO ₂ ($k_{d-d} \sim 4.08 \times 10^{-10} \text{ cm}^3 \text{ s}^{-1}$)	sCI 2 + SO ₂ : grainsize 50 sCI 3 + SO ₂ : grainsize 40.	At 760 Torr: 200, 275, 298, 325 & 400 K
sCIs 4 & 5 + SO ₂	N/A	N/A	sCIs 4 & 5 + SO ₂ ($k_{d-d} \sim 4.51 \text{ & } 4.42 \times 10^{-10} \text{ cm}^3 \text{ s}^{-1}$)	grainsize 50	At 760 Torr: 200, 275, 298, 325 & 400 K
sCI 1 + HNO ₃	grainsize 10	N/A	N/A	N/A	At 760 Torr: 200, 275, 298, 325 & 400 K
sCI 2 + HNO ₃	grainsize 10	N/A	N/A	N/A	At 760 Torr: 200, 275, 298, 325 & 400 K
sCI 3 + HNO ₃	grainsize 10	N/A	N/A	N/A	At 760 Torr: 200, 275, 298, 325 & 400 K
sCI 4 + HNO ₃	grainsize 10	N/A	N/A	N/A	At 760 Torr: 200, 275, 298, 325 & 400 K
sCI 5 + HNO ₃	grainsize 10	N/A	N/A	N/A	At 760 Torr: 200, 275, 298, 325 & 400 K
sCI 1 + TFA	N/A	N/A	sCI 1 + TFA TS ($k_{d-d} \sim 7.54 \times 10^{-10} \text{ cm}^3 \text{ s}^{-1}$)	N/A	-295 K Pressures: 27, 31, 35 & 760 Torr -At 760 Torr: 200, 275, 298, 325 & 400 K
sCI 2 + TFA	N/A	N/A	sCI 2 + TFA TS ($k_{d-d} \sim 4.96 \times 10^{-10} \text{ cm}^3 \text{ s}^{-1}$)	N/A	At 760 Torr: 200, 275, 298, 325 & 400 K

sCI 3 + TFA	N/A	N/A	sCI 3 + TFA TS ($k_{d-d} \sim 3.98 \times 10^{-10} \text{ cm}^3 \text{ s}^{-1}$)	N/A	At 760 Torr: 200, 275, 298, 325 & 400 K
sCI 4 + TFA	N/A	N/A	sCI 4 + TFA TS ($k_{d-d} \sim 4.35 \times 10^{-10} \text{ cm}^3 \text{ s}^{-1}$)	N/A	At 760 Torr: 200, 275, 298, 325 & 400 K
sCI 5 + TFA	N/A	N/A	sCI 5 + TFA TS ($k_{d-d} \sim 4.27 \times 10^{-10} \text{ cm}^3 \text{ s}^{-1}$)	N/A	At 760 Torr: 200, 275, 298, 325 & 400 K
sCI 1 + H₂O	grainsize 10	N/A	N/A	N/A	-At 293 K & 50 Torr; 297 K & 760 Torr -At 760 Torr: 200, 275, 298, 325 & 400 K
sCI 2 + H₂O	grainsize 10	N/A	N/A	N/A	At 760 Torr: 200, 275, 298, 325 & 400 K
sCI 3 + H₂O	grainsize 10	N/A	N/A	N/A	At 760 Torr: 200, 275, 298, 325 & 400 K
sCI 4 + H₂O	grainsize 10	N/A	N/A	N/A	At 760 Torr: 200, 275, 298, 325 & 400 K
sCI 5 + H₂O	grainsize 10	N/A	N/A	N/A	At 760 Torr: 200, 275, 298, 325 & 400 K
sCI 1 + (H₂O)₂	grainsize 25	N/A	N/A	N/A	At 760 Torr: 200, 275, 298, 325 & 400 K
sCI 2 + (H₂O)₂	grainsize 25	N/A	N/A	N/A	At 760 Torr: 200, 275, 298, 325 & 400 K
sCI 3 + (H₂O)₂	grainsize 25	N/A	N/A	N/A	At 760 Torr: 200, 275, 298, 325 & 400 K
sCI 4 + (H₂O)₂	grainsize 25	sCI 4 + (H₂O)₂ TS_{(H₂O)₂} 1, 2, 3 & 4	N/A	N/A	At 760 Torr: 200, 275, 298, 325 & 400 K
sCI 5 + (H₂O)₂	grainsize 25	N/A	N/A	N/A	At 760 Torr: 200, 275, 298, 325 & 400 K
sCI 1 + MeOH	grainsize 10	N/A	N/A	N/A	-At 295 K & 90 Torr; 262.1 K & 100 Torr -At 10 Torr: 254.5, 292.6 & 327.8 K -At 760 Torr: 200, 275, 298, 325 & 400 K
sCI 2 + MeOH	grainsize 15	N/A	N/A	N/A	At 760 Torr: 200, 275, 298, 325 & 400 K
sCI 3 + MeOH	grainsize 15	N/A	N/A	N/A	At 760 Torr: 200, 275, 298, 325 & 400 K
sCI 4 + MeOH	grainsize 25	N/A	sCI 4 + MeOH TS_{AAAH}1 ($k_{d-d} \sim 5.50 \times 10^{-10} \text{ cm}^3 \text{ s}^{-1}$)	N/A	At 760 Torr: 200, 275, 298, 325 & 400 K
sCI 5 + MeOH	grainsize 25	N/A	N/A	N/A	At 760 Torr: 200, 275, 298, 325 & 400 K
sCI 1 + H₂S	grainsize 10	N/A	N/A	N/A	-At 100 Torr: 278, 299 & 318 K -At 299 K & 250 Torr; 299 K & 500 Torr -At 760 Torr: 200, 275, 298, 325 & 400 K
sCI 2 + H₂S	grainsize 10	N/A	N/A	N/A	At 760 Torr: 200, 275, 298, 325 & 400 K
sCI 3 + H₂S	grainsize 10	N/A	N/A	N/A	At 760 Torr: 200, 275, 298, 325 & 400 K

sCI 4 + H₂S	grainsize 10	N/A	N/A	N/A	At 760 Torr: 200, 275, 298, 325 & 400 K
sCI 5 + H₂S	grainsize 10	N/A	N/A	N/A	At 760 Torr: 200, 275, 298, 325 & 400 K
sCI 1 + HCl	grainsize 25	N/A	N/A	N/A	-At 295 K with 27, 31 & 35 Torr -At 760 Torr: 200, 275, 298, 325 & 400 K
sCI 2 + HCl	grainsize 10	N/A	N/A	N/A	At 760 Torr: 200, 275, 298, 325 & 400 K
sCI 3 + HCl	grainsize 10	N/A	N/A	N/A	At 760 Torr: 200, 275, 298, 325 & 400 K
sCI 4 + HCl	grainsize 10	N/A	N/A	N/A	At 760 Torr: 200, 275, 298, 325 & 400 K
sCI 5 + HCl	grainsize 10	N/A	N/A	N/A	At 760 Torr: 200, 275, 298, 325 & 400 K
sCI 1 + HF	grainsize 10	N/A	N/A	N/A	At 760 Torr: 200, 275, 298, 325 & 400 K
sCI 2 + HF	grainsize 10	N/A	N/A	N/A	At 760 Torr: 200, 275, 298, 325 & 400 K
sCI 3 + HF	grainsize 10	N/A	N/A	N/A	At 760 Torr: 200, 275, 298, 325 & 400 K
sCI 4 + HF	grainsize 10	N/A	N/A	N/A	At 760 Torr: 200, 275, 298, 325 & 400 K
sCI 5 + HF	grainsize 10	N/A	N/A	N/A	At 760 Torr: 200, 275, 298, 325 & 400 K

S1.2 Master Equation rate constants (k_{ME}) Calculated using MESMER

Table S2: Master Equation (k_{ME}) rate constants for all non-barrierless reactions between 200 – 400 K. (For rate constants of all barrierless reactions see Section S3)

Co-reactant	sCl	Grain size	k_{ME} (cm ³ molec. ⁻¹ s ⁻¹)				
			200 K	275 K	298.15 K	325 K	400 K
HCHO	1	10	2.77E-11	3.48E-12	2.79E-12	2.25E-12	1.40E-12
	2	10	2.02E-12	3.69E-13	2.69E-13	1.98E-13	1.11E-13
	3	10	2.27E-11	3.32E-12	2.49E-12	1.82E-12	8.59E-13
	5	5	4.59E-11	2.12E-11	1.66E-11	1.25E-11	5.86E-12
CF ₃ CFO	1	10	2.27E-10	2.92E-12	1.98E-12	1.32E-12	5.11E-13
SO ₂	2	10	9.81E-12	2.52E-12	1.90E-12	1.41E-12	7.20E-13
HNO ₃	1	15	3.89E-04	5.48E-08	8.22E-09	1.21E-09	3.95E-11
	2	15	5.47E-09	5.11E-11	2.06E-11	8.69E-12	1.70E-12
	3	15	1.99E-07	3.03E-10	8.87E-11	3.29E-11	8.53E-12
	4	15	4.03E-05	2.85E-07	9.51E-08	3.07E-08	2.44E-09
	5	15	2.97E-07	8.58E-11	7.57E-11	6.64E-11	4.30E-11
HF	1	10	2.24E-12	4.31E-13	3.32E-13	2.62E-13	1.71E-13
	2	10	2.51E-19	3.63E-18	6.67E-18	1.24E-17	4.79E-17
	3	10	2.44E-15	2.59E-15	2.86E-15	3.23E-15	4.48E-15
	4	10	5.35E-10	2.05E-11	1.19E-11	7.20E-12	2.73E-12
	5	10	2.41E-12	4.90E-13	3.66E-13	2.78E-13	1.64E-13
HCl	1	5	7.60E-08	8.37E-10	4.70E-10	3.21E-10	1.99E-10
	2	5	1.46E-14	3.34E-14	4.04E-14	4.93E-14	7.76E-14
	3	5	4.26E-10	9.05E-11	7.28E-11	5.72E-11	3.23E-11
	4	5	4.18E-10	2.05E-10	1.64E-10	1.28E-10	6.67E-11
	5	5	3.92E-09	1.42E-10	1.13E-10	8.66E-11	4.51E-11
H ₂ S	1	10	4.04E-15	6.31E-15	7.06E-15	8.00E-15	1.09E-14
	2	10	2.73E-17	1.37E-16	1.96E-16	2.84E-16	6.49E-16
	3	10	4.34E-15	5.36E-15	5.72E-15	6.18E-15	7.67E-15
	4	10	3.02E-12	7.91E-13	6.08E-13	4.74E-13	3.00E-13
	5	10	1.61E-14	1.26E-14	1.23E-14	1.21E-14	1.26E-14
H ₂ O	1	10	1.35E-17	8.04E-17	1.18E-16	1.73E-16	4.00E-16
	2	10	1.63E-20	6.27E-19	1.35E-18	2.90E-18	1.47E-17
	3	10	1.61E-17	7.99E-17	1.12E-16	1.59E-16	3.37E-16
	4	10	8.71E-11	8.15E-12	5.13E-12	3.23E-12	1.24E-12
	5	10	1.43E-12	3.06E-13	2.27E-13	1.69E-13	9.66E-14
(H ₂ O) ₂	1	25	3.18E-09	9.88E-12	3.28E-12	1.22E-12	2.18E-13
	2	25	1.15E-09	7.04E-12	2.71E-12	1.09E-12	1.62E-13
	3	25	5.60E-09	7.50E-12	3.74E-12	1.96E-12	4.00E-13
	4	35	2.71E-06	2.21E-06	1.14E-06	4.91E-07	3.65E-08
	5	45	1.05E-07	8.51E-09	1.46E-09	1.30E-09	1.91E-10
MeOH	1	10	2.65E-14	1.34E-14	1.20E-14	1.09E-14	9.48E-15
	2	10	2.55E-17	7.34E-17	9.43E-17	1.23E-16	2.28E-16
	3	10	1.86E-13	4.32E-14	3.30E-14	2.57E-14	1.62E-14
	4	10	1.10E-10	6.08E-11	5.17E-11	4.27E-11	2.52E-11
	5	20	6.92E-09	7.91E-11	3.31E-11	1.04E-11	2.22E-12

S1.3 Canonical Rate constants (k_{CAN}) calculated by MESMER

Note: **sCI 4 + HCHO**; **sCI 1 + CF₃CHO**; **sCIs 1,3,4 & 5 with SO₂**; and **sCIs 1 – 5 with TFA** are barrierless reactions (see IRCS in section S7) and so are not included in this section

Table S3: Overall Canonical Rate Constants (k_{CAN}) of HCHO + sCIs 1–3, and 5 based on a steady state treatment of individual canonical rate coefficients of reaction (k_1 , k_{-1} , and k_2)

sCI#	T (K)	k_{CAN} (cm ³ molec. ⁻¹ s ⁻¹)	k_1 (cm ³ molec. ⁻¹ s ⁻¹)	k_{-1} (s ⁻¹)	k_2 (s ⁻¹)
1	200	1.82E-09	9.99E-11	7.83E+10	1.43E+12
	275	5.66E-11	1.00E-10	2.44E+12	1.38E+12
	298	2.83E-11	1.00E-10	4.81E+12	1.36E+12
	325	1.45E-11	1.00E-10	9.29E+12	1.34E+12
	400	3.81E-12	1.00E-10	3.40E+13	1.29E+12
2	200	2.65E-12	1.01E-10	4.56E+10	1.20E+09
	275	4.17E-13	1.01E-10	9.49E+11	3.93E+09
	298	2.94E-13	1.01E-10	1.70E+12	4.98E+09
	325	2.11E-13	1.00E-10	2.99E+12	6.26E+09
	400	1.13E-13	1.00E-10	8.81E+12	9.95E+09
3	200	2.52E-10	1.01E-10	2.19E+10	5.48E+10
	275	1.15E-11	1.01E-10	9.49E+11	1.08E+11
	298	6.24E-12	1.01E-10	2.00E+12	1.24E+11
	325	3.45E-12	1.01E-10	4.11E+12	1.41E+11
	400	1.08E-12	1.00E-10	1.70E+13	1.83E+11
5	200	5.61E-08	1.01E-10	1.02E+09	5.71E+11
	275	5.31E-10	1.00E-10	9.80E+10	5.20E+11
	298	2.10E-10	1.00E-10	2.42E+11	5.07E+11
	325	8.46E-11	1.00E-10	5.83E+11	4.93E+11
	400	1.39E-11	1.00E-10	3.35E+12	4.63E+11

Table S4: Overall Canonical Rate Constants (k_{CAN}) of CF₃CFO + sCI 1 based on a steady state treatment of individual canonical rate coefficients of reaction (k_1 , k_{-1} , k_2 , k_3 , k_{-3} , and k_4)

sCI#	T (K)	k_{CAN} (cm ³ molec. ⁻¹ s ⁻¹)	k_1 (cm ³ molec. ⁻¹ s ⁻¹)	k_{-1} (s ⁻¹)	k_2 (s ⁻¹)	k_3 (cm ³ molec. ⁻¹ s ⁻¹)	k_{-3} (s ⁻¹)	k_4 (s ⁻¹)
1	200	1.21E-09	1.01E-10	3.17E+09	3.50E+10	1.01E-10	9.36E+09	8.14E+09
	275	1.82E-11	1.01E-10	4.48E+11	7.03E+10	1.00E-10	1.13E+12	2.68E+10
	298	7.97E-12	1.01E-10	1.20E+12	8.08E+10	1.00E-10	2.92E+12	3.41E+10
	325	3.58E-12	1.01E-10	3.11E+12	9.26E+10	1.00E-10	7.37E+12	4.33E+10
	400	7.42E-13	1.01E-10	2.08E+13	1.22E+11	1.00E-10	4.63E+13	7.04E+10

Table S5: Overall Canonical Rate Constants (k_{CAN}) of $SO_2 + sCl\text{ 2}$ based on a steady state treatment of individual canonical rate coefficients of reaction (k_1 , k_{-1} , k_2 , k_3 , k_{-3} , and k_4)

sCl#	T (K)	k_{CAN} (cm ³ molec. ⁻¹ s ⁻¹)	k_1 (cm ³ molec. ⁻¹ s ⁻¹)	k_{-1} (s ⁻¹)	k_2 (s ⁻¹)	k_3 (cm ³ molec. ⁻¹ s ⁻¹)	k_{-3} (s ⁻¹)	k_4 (s ⁻¹)
2	200	7.06E-11	1.01E-10	1.78E+11	3.47E+10	1.01E-10	3.15E+11	1.59E+11
	275	5.20E-12	1.01E-10	3.95E+12	5.90E+10	1.01E-10	4.78E+12	1.76E+11
	298	3.16E-12	1.01E-10	7.18E+12	6.56E+10	1.01E-10	8.03E+12	1.79E+11
	325	1.97E-12	1.01E-10	1.28E+13	7.29E+10	1.00E-10	1.32E+13	1.82E+11
	400	7.94E-13	1.00E-10	3.87E+13	9.04E+10	1.00E-10	3.39E+13	1.89E+11

Table S6: Overall Canonical Rate Constants (k_{CAN}) of $HNO_3 + sCl\text{s 1-5}$ based on a steady state treatment of individual canonical rate coefficients of reaction (k_1 , k_{-1} , and k_2)

sCl#	T (K)	k_{CAN} (cm ³ molec. ⁻¹ s ⁻¹)	k_1 (cm ³ molec. ⁻¹ s ⁻¹)	k_{-1} (s ⁻¹)	k_2 (s ⁻¹)
1	200	6.58E-03	1.01E-10	6.61E+02	4.30E+10
	275	2.66E-06	1.01E-10	3.96E+06	1.04E+11
	298	5.47E-07	1.01E-10	2.30E+07	1.25E+11
	325	1.15E-07	1.01E-10	1.30E+08	1.49E+11
	400	4.87E-09	1.00E-10	4.44E+09	2.15E+11
2	200	7.22E-09	1.00E-10	1.21E+07	8.73E+08
	275	7.72E-11	1.00E-10	2.52E+09	1.94E+09
	298	3.13E-11	1.00E-10	7.32E+09	2.28E+09
	325	1.29E-11	1.00E-10	2.08E+10	2.68E+09
	400	2.20E-12	1.00E-10	1.68E+11	3.70E+09
3	200	8.68E-07	1.01E-10	9.60E+05	8.29E+09
	275	2.95E-09	1.00E-10	7.70E+08	2.26E+10
	298	9.42E-10	1.00E-10	2.95E+09	2.77E+10
	325	3.08E-10	1.00E-10	1.11E+10	3.40E+10
	400	3.23E-11	1.00E-10	1.59E+11	5.14E+10
4	200	1.27E-01	1.00E-10	2.72E+02	3.45E+11
	275	1.42E-05	1.00E-10	1.91E+06	2.70E+11
	298	2.24E-06	1.00E-10	1.14E+07	2.56E+11
	325	3.63E-07	1.00E-10	6.68E+07	2.42E+11
	400	8.88E-09	1.00E-10	2.42E+09	2.14E+11
5	200	9.93E-03	1.01E-10	5.48E+02	5.39E+10
	275	2.13E-06	1.01E-10	3.72E+06	7.89E+10
	298	3.88E-07	1.01E-10	2.22E+07	8.56E+10
	325	7.25E-08	1.01E-10	1.29E+08	9.28E+10
	400	2.41E-09	1.00E-10	4.59E+09	1.10E+11

Table S7: Overall Canonical Rate Constants (k_{CAN}) of HF + sCl^s 1 – 5 based on a steady state treatment of individual canonical rate coefficients of reaction (k_1 , k_{-1} , and k_2)

sCl#	T (K)	k_{CAN} (cm ³ molec. ⁻¹ s ⁻¹)	k_1 (cm ³ molec. ⁻¹ s ⁻¹)	k_{-1} (s ⁻¹)	k_2 (s ⁻¹)
1	200	2.26E-12	1.01E-10	2.60E+04	5.81E+02
	275	4.37E-13	1.01E-10	2.29E+07	9.92E+04
	298	3.36E-13	1.01E-10	9.19E+07	3.07E+05
	325	2.65E-13	1.01E-10	3.64E+08	9.57E+05
	400	1.72E-13	1.01E-10	6.15E+09	1.05E+07
2	200	2.51E-19	1.01E-10	6.81E+07	1.70E-01
	275	3.63E-18	1.00E-10	5.17E+09	1.87E+02
	298	6.67E-18	1.00E-10	1.25E+10	8.28E+02
	325	1.24E-17	1.00E-10	2.96E+10	3.67E+03
	400	4.79E-17	1.00E-10	1.72E+11	8.23E+04
3	200	2.44E-15	1.01E-10	1.10E+07	2.67E+02
	275	2.59E-15	1.00E-10	2.07E+09	5.35E+04
	298	2.86E-15	1.00E-10	6.05E+09	1.72E+05
	325	3.23E-15	1.00E-10	1.74E+10	5.62E+05
	400	4.48E-15	1.00E-10	1.52E+11	6.79E+06
4	200	1.05E-09	1.01E-10	3.99E+07	4.17E+08
	275	4.61E-11	1.01E-10	5.39E+09	2.47E+09
	298	2.44E-11	1.01E-10	1.47E+10	3.57E+09
	325	1.31E-11	1.00E-10	3.94E+10	5.13E+09
	400	3.66E-12	1.00E-10	2.96E+11	1.08E+10
5	200	2.42E-12	1.01E-10	1.67E+06	4.02E+04
	275	4.93E-13	1.00E-10	4.61E+08	2.26E+06
	298	3.68E-13	1.00E-10	1.45E+09	5.31E+06
	325	2.79E-13	1.00E-10	4.50E+09	1.25E+07
	400	1.64E-13	1.00E-10	4.53E+10	7.42E+07

Table S8: Overall Canonical Rate Constants (k_{CAN}) of HCl + sCl_i 1-5 based on a steady state treatment of individual canonical rate coefficients of reaction (k_1 , k_{-1} , and k_2)

sCl#	T (K)	k_{CAN} (cm ³ molec. ⁻¹ s ⁻¹)	k_1 (cm ³ molec. ⁻¹ s ⁻¹)	k_{-1} (s ⁻¹)	k_2 (s ⁻¹)
1	200	6.60E-07	1.00E-10	2.16E+08	1.43E+12
	275	8.46E-09	1.00E-10	1.75E+10	1.48E+12
	298	3.48E-09	1.00E-10	4.29E+10	1.49E+12
	325	1.45E-09	1.00E-10	1.04E+11	1.51E+12
	400	2.46E-10	1.00E-10	6.22E+11	1.53E+12
2	200	1.46E-15	1.01E-10	1.28E+10	1.85E+05
	275	3.35E-15	1.01E-10	2.02E+11	6.70E+06
	298	4.05E-15	1.01E-10	3.49E+11	1.40E+07
	325	4.93E-15	1.01E-10	5.96E+11	2.92E+07
	400	7.76E-15	1.00E-10	1.72E+12	1.33E+08
3	200	2.80E-10	1.01E-10	1.62E+10	4.52E+10
	275	2.56E-11	1.00E-10	3.93E+11	1.00E+11
	298	1.59E-11	1.00E-10	7.46E+11	1.18E+11
	325	1.01E-11	1.00E-10	1.40E+12	1.40E+11
	400	4.05E-12	1.00E-10	4.93E+12	1.99E+11
4	200	3.30E-09	1.00E-10	9.27E+08	3.05E+10
	275	1.39E-10	1.00E-10	2.91E+10	4.03E+10
	298	7.32E-11	1.00E-10	5.81E+10	4.24E+10
	325	3.90E-11	1.00E-10	1.14E+11	4.45E+10
	400	1.10E-11	1.00E-10	4.44E+11	4.87E+10
5	200	1.38E-09	1.01E-10	1.18E+09	1.63E+10
	275	6.62E-11	1.00E-10	5.00E+10	3.30E+10
	298	3.64E-11	1.00E-10	1.06E+11	3.85E+10
	325	2.03E-11	1.00E-10	2.22E+11	4.50E+10
	400	6.41E-12	1.00E-10	9.77E+11	6.25E+10

Table S9: Overall Canonical Rate Constants (k_{CAN}) of $H_2S + sCl\# 1-5$: based on a steady state treatment of individual canonical rate coefficients of reaction (k_1 , k_{-1} , k_2 , k_3 , k_{-3} and k_4)

sCl#	T (K)	k_{CAN} (cm ³ molec. ⁻¹ s ⁻¹)	k_1 (cm ³ molec. ⁻¹ s ⁻¹)	k_{-1} (s ⁻¹)	k_2 (s ⁻¹)	k_3 (cm ³ molec. ⁻¹ s ⁻¹)	k_{-3} (s ⁻¹)	k_4 (s ⁻¹)
1	200	4.04E-15	1.01E-10	4.10E+11	9.05E+06	1.00E-10	4.62E+11	8.34E+06
	275	6.31E-15	1.01E-10	3.66E+12	1.24E+08	1.00E-10	3.99E+12	1.16E+08
	298	7.06E-15	1.01E-10	5.55E+12	2.09E+08	1.00E-10	6.01E+12	1.96E+08
	325	8.00E-15	1.01E-10	8.26E+12	3.51E+08	1.00E-10	8.88E+12	3.30E+08
	400	1.09E-14	1.01E-10	1.75E+13	1.01E+09	1.00E-10	1.86E+13	9.50E+08
2	200	2.73E-17	1.01E-10	5.05E+11	1.01E+05	1.00E-10	8.12E+11	5.66E+04
	275	1.37E-16	1.01E-10	1.67E+12	1.54E+06	1.00E-10	1.94E+12	8.56E+05
	298	1.96E-16	1.01E-10	2.05E+12	2.64E+06	1.00E-10	2.23E+12	1.48E+06
	325	2.84E-16	1.01E-10	2.47E+12	4.50E+06	1.00E-10	2.52E+12	2.52E+06
	400	6.49E-16	1.00E-10	3.35E+12	1.33E+07	1.00E-10	3.00E+12	7.50E+06
3	200	3.01E-10	1.01E-10	5.11E+13	1.52E+14	1.01E-10	1.52E+14	2.04E+09
	275	5.36E-15	1.01E-10	1.09E+14	3.72E+09	1.00E-10	2.42E+14	4.59E+09
	298	5.72E-15	1.01E-10	1.22E+14	4.40E+09	1.00E-10	2.56E+14	5.35E+09
	325	6.18E-15	1.01E-10	1.35E+14	5.18E+09	1.00E-10	2.67E+14	6.17E+09
	400	7.67E-15	1.01E-10	1.54E+14	7.10E+09	1.00E-10	2.69E+14	8.15E+09
4	200	2.99E-12	1.01E-10	7.85E+10	1.84E+09	1.00E-10	7.97E+10	4.96E+08
	275	7.39E-13	1.01E-10	6.78E+11	3.61E+09	1.00E-10	6.36E+11	1.29E+09
	298	5.71E-13	1.00E-10	1.02E+12	4.10E+09	1.00E-10	9.38E+11	1.56E+09
	325	4.49E-13	1.00E-10	1.50E+12	4.63E+09	1.00E-10	1.36E+12	1.87E+09
	400	2.92E-13	1.00E-10	3.07E+12	5.90E+09	1.00E-10	2.69E+12	2.68E+09
5	200	1.60E-14	1.01E-10	3.85E+10	3.70E+06	1.01E-10	3.97E+10	2.49E+06
	275	1.26E-14	1.00E-10	5.76E+11	4.19E+07	1.01E-10	5.97E+11	3.12E+07
	298	1.23E-14	1.00E-10	9.69E+11	6.82E+07	1.01E-10	1.00E+12	5.19E+07
	325	1.21E-14	1.00E-10	1.59E+12	1.10E+08	1.00E-10	1.65E+12	8.55E+07
	400	1.26E-14	1.00E-10	4.13E+12	2.92E+08	1.00E-10	4.29E+12	2.36E+08

Table S10: Overall Canonical Rate Constants (k_{CAN}) of $H_2O + sCl\# 1-5$: based on a steady state treatment of individual canonical rate coefficients of reaction ($k_1, k_{-1}, k_2, k_3, k_{-3}$ and k_4)

sCl#	T (K)	k_{CAN} (cm ³ molec. ⁻¹ s ⁻¹)	k_1 (cm ³ molec. ⁻¹ s ⁻¹)	k_{-1} (s ⁻¹)	k_2 (s ⁻¹)	k_3 (cm ³ molec. ⁻¹ s ⁻¹)	k_{-3} (s ⁻¹)	k_4 (s ⁻¹)
1	200	1.35E-17	1.00E-10	6.00E+07	7.04E+00	1.00E-10	6.00E+07	1.02E+00
	275	8.04E-17	1.00E-10	3.68E+09	2.38E+03	1.00E-10	3.68E+09	5.69E+02
	298	1.18E-16	1.00E-10	8.39E+09	7.79E+03	1.00E-10	8.39E+09	2.07E+03
	325	1.73E-16	1.00E-10	1.88E+10	2.51E+04	1.00E-10	1.88E+10	7.38E+03
	400	4.00E-16	1.00E-10	9.34E+10	2.73E+05	1.00E-10	9.34E+10	9.98E+04
2	200	1.63E-20	1.00E-10	1.54E+10	6.74E-01	1.00E-10	1.54E+10	1.83E+00
	275	6.27E-19	1.00E-10	1.15E+11	2.24E+02	1.00E-10	1.15E+11	4.98E+02
	298	1.35E-18	1.00E-10	1.69E+11	7.28E+02	1.00E-10	1.69E+11	1.55E+03
	325	2.90E-18	1.00E-10	2.44E+11	2.32E+03	1.00E-10	2.44E+11	4.74E+03
	400	1.47E-17	1.00E-10	4.86E+11	2.49E+04	1.00E-10	4.86E+11	4.65E+04
3	200	1.61E-17	1.01E-10	6.88E+08	9.27E+01	1.01E-10	6.88E+08	1.72E+01
	275	7.99E-17	1.00E-10	2.41E+10	1.52E+04	1.00E-10	2.41E+10	4.05E+03
	298	1.12E-16	1.00E-10	4.89E+10	4.25E+04	1.00E-10	4.89E+10	1.22E+04
	325	1.59E-16	1.00E-10	9.71E+10	1.17E+05	1.00E-10	9.71E+10	3.64E+04
	400	3.37E-16	1.00E-10	3.78E+11	9.31E+05	1.00E-10	3.78E+11	3.37E+05
4	200	1.12E-10	1.01E-10	1.29E+09	5.84E+08	1.01E-10	5.61E+08	3.69E+08
	275	8.88E-12	1.00E-10	4.52E+10	1.63E+09	1.01E-10	2.69E+10	1.40E+09
	298	5.33E-12	1.00E-10	9.11E+10	1.98E+09	1.01E-10	5.80E+10	1.81E+09
	325	3.23E-12	1.00E-10	1.80E+11	2.38E+09	1.01E-10	1.22E+11	2.31E+09
	400	1.19E-12	1.00E-10	6.89E+11	3.38E+09	1.01E-10	5.35E+11	3.71E+09
5	200	1.45E-12	1.01E-10	2.69E+08	1.25E+06	1.01E-10	2.69E+08	2.64E+06
	275	3.10E-13	1.00E-10	2.09E+10	2.46E+07	1.00E-10	2.09E+10	3.98E+07
	298	2.28E-13	1.00E-10	4.96E+10	4.46E+07	1.00E-10	4.96E+10	6.84E+07
	325	1.70E-13	1.00E-10	1.15E+11	7.99E+07	1.00E-10	1.15E+11	1.16E+08
	400	9.68E-14	1.00E-10	6.20E+11	2.60E+08	1.00E-10	6.20E+11	3.39E+08

Table S11: Overall Canonical Rate Constants (k_{CAN}) of $(H_2O)_2 + sCl\# 1-5$ based on a steady state treatment of individual canonical rate coefficients of reaction (k_1 , k_{-1} , k_2 , k_3 , k_{-3} and k_4)

sCl#	T (K)	k_{CAN} (cm ³ mol. ⁻¹ s ⁻¹)	k_1 (cm ³ mol. ⁻¹ s ⁻¹)	k_{-1} (s ⁻¹)	k_2 (s ⁻¹)	k_3 (cm ³ mol. ⁻¹ s ⁻¹)	k_{-3} (s ⁻¹)	k_4 (s ⁻¹)
1	200	8.04E-09	1.01E-10	1.01E+07	1.99E+08	1.01E-10	7.06E+06	1.47E+07
			1.01E-10	5.45E+06	1.62E+08	1.01E-10	9.74E+06	2.71E+08
	275	4.28E-11	1.01E-10	8.14E+09	9.31E+08	1.01E-10	5.92E+09	1.25E+08
			1.01E-10	6.10E+09	8.84E+08	1.01E-10	9.04E+09	1.30E+09
	298	1.44E-11	1.00E-10	3.19E+10	1.24E+09	1.01E-10	2.33E+10	1.88E+08
			1.01E-10	2.56E+10	1.22E+09	1.01E-10	3.65E+10	1.75E+09
	325	4.85E-12	1.00E-10	1.23E+11	1.64E+09	1.01E-10	9.00E+10	2.79E+08
			1.01E-10	1.05E+11	1.66E+09	1.01E-10	1.45E+11	2.32E+09
	400	5.09E-13	1.00E-10	1.92E+12	2.75E+09	1.01E-10	1.41E+12	5.98E+08
			1.01E-10	1.89E+12	2.99E+09	1.00E-10	2.41E+12	3.96E+09
2	200	1.57E-09	1.01E-10	5.04E+07	2.28E+08	1.01E-10	4.92E+07	2.13E+07
			1.01E-10	1.52E+07	8.51E+07	1.01E-10	4.44E+07	2.24E+08
	275	1.01E-11	1.01E-10	3.58E+10	1.09E+09	1.00E-10	4.04E+10	2.22E+08
			1.01E-10	1.53E+10	5.13E+08	1.00E-10	3.58E+10	1.12E+09
	298	3.57E-12	1.00E-10	1.36E+11	1.46E+09	1.00E-10	1.59E+11	3.51E+08
			1.01E-10	6.24E+10	7.25E+08	1.00E-10	1.39E+11	1.52E+09
	325	1.26E-12	1.00E-10	5.05E+11	1.94E+09	1.00E-10	6.10E+11	5.48E+08
			1.00E-10	2.51E+11	1.01E+09	1.00E-10	5.33E+11	2.03E+09
	400	1.48E-13	1.00E-10	7.31E+12	3.30E+09	1.00E-10	9.49E+12	1.31E+09
			1.00E-10	4.27E+12	1.94E+09	1.00E-10	8.18E+12	3.55E+09
3	200	5.86E-08	1.01E-10	2.86E+08	7.70E+09	1.01E-10	2.59E+08	8.47E+10
			1.01E-10	1.69E+08	1.55E+10	1.01E-10	1.58E+08	2.14E+10
	275	1.33E-10	1.01E-10	1.32E+11	1.32E+10	1.01E-10	1.53E+11	9.42E+10
			1.01E-10	8.35E+10	2.19E+10	1.01E-10	8.85E+10	3.05E+10
	298	3.78E-11	1.01E-10	4.58E+11	1.44E+10	1.01E-10	5.63E+11	9.40E+10
			1.00E-10	2.96E+11	2.29E+10	1.01E-10	3.21E+11	3.21E+10
	325	1.08E-11	1.00E-10	1.57E+12	1.54E+10	1.01E-10	2.03E+12	9.29E+10
			1.00E-10	1.03E+12	2.36E+10	1.01E-10	1.15E+12	3.33E+10
	400	8.15E-13	1.00E-10	1.90E+13	1.70E+10	1.00E-10	2.79E+13	8.70E+10
			1.00E-10	1.30E+13	2.42E+10	1.00E-10	1.53E+13	3.44E+10
4	200	2.87E-01	1.00E-10	1.14E+04	2.95E+12	1.01E-10	5.96E+03	8.52E+12
			1.00E-10	2.78E+04	9.32E+12	1.00E-10	2.01E+04	1.66E+13
	275	1.04E-05	1.00E-10	1.26E+08	1.61E+12	1.00E-10	8.93E+07	3.99E+12
			1.00E-10	2.77E+08	4.25E+12	1.00E-10	2.31E+08	7.07E+12
	298	1.25E-06	1.00E-10	8.51E+08	1.39E+12	1.00E-10	6.41E+08	3.34E+12
			1.00E-10	1.83E+09	3.54E+12	1.00E-10	1.57E+09	5.81E+12
	325	1.53E-07	1.00E-10	5.62E+09	1.19E+12	1.00E-10	4.52E+09	2.78E+12
			1.00E-10	1.18E+10	2.93E+12	1.00E-10	1.05E+10	4.75E+12
	400	1.99E-09	1.00E-10	2.69E+11	8.39E+11	1.00E-10	2.49E+11	1.85E+12
			1.00E-10	5.47E+11	1.93E+12	1.00E-10	5.23E+11	3.05E+12
5	200	5.33E-03	1.01E-10	2.26E+04	1.59E+11	1.01E-10	1.04E+04	4.16E+11
			1.01E-10	2.17E+04	9.18E+10	1.01E-10	2.74E+04	4.16E+10
	275	4.80E-07	1.00E-10	1.69E+08	1.47E+11	1.01E-10	1.12E+08	3.39E+11
			1.01E-10	1.53E+08	9.17E+10	1.01E-10	1.60E+08	4.41E+10
	298	7.07E-08	1.00E-10	1.04E+09	1.41E+11	1.01E-10	7.48E+08	3.18E+11
			1.01E-10	9.33E+08	8.96E+10	1.01E-10	9.39E+08	4.35E+10
	325	1.06E-08	1.00E-10	6.31E+09	1.35E+11	1.01E-10	4.89E+09	2.96E+11
			1.01E-10	5.58E+09	8.66E+10	1.00E-10	5.40E+09	4.25E+10
	400	2.12E-10	1.00E-10	2.49E+11	1.17E+11	1.00E-10	2.29E+11	2.48E+11
			1.00E-10	2.16E+11	7.77E+10	1.00E-10	1.93E+11	3.88E+10

Table S12: Overall Canonical Rate Constants (k_{CAN}) of MeOH + **sCl**s 1-5 based on a steady state treatment of individual canonical rate coefficients of reaction (k_1 , k_{-1} , k_2 , k_3 , k_{-3} , and k_4).

Note: **sCl 4** + MeOH has one barrierless step and one with a barrier so the dipole-dipole capture rate constant is used.

sCl#	T (K)	k_{CAN} (cm ³ molec. ⁻¹ s ⁻¹)	k_1 (cm ³ molec. ⁻¹ s ⁻¹)	k_{-1} (s ⁻¹)	k_2 (s ⁻¹)	k_3 (cm ³ molec. ⁻¹ s ⁻¹)	k_{-3} (s ⁻¹)	k_4 (s ⁻¹)
1	200	2.65E-14	1.01E-10	2.89E+08	6.86E+04	1.01E-10	2.89E+08	7.70E+03
	275	1.34E-14	1.00E-10	2.21E+10	2.43E+06	1.00E-10	2.21E+10	5.23E+05
	298	1.20E-14	1.00E-10	5.21E+10	5.01E+06	1.00E-10	5.21E+10	1.24E+06
	325	1.09E-14	1.00E-10	1.21E+11	1.02E+07	1.00E-10	1.21E+11	2.89E+06
	400	9.48E-15	1.00E-10	6.41E+11	4.41E+07	1.00E-10	6.41E+11	1.64E+07
2	200	2.55E-17	1.01E-10	2.80E+10	6.90E+03	1.01E-10	2.80E+10	1.90E+02
	275	7.34E-17	1.01E-10	2.34E+11	1.59E+05	1.01E-10	2.34E+11	1.10E+04
	298	9.43E-17	1.01E-10	3.49E+11	3.02E+05	1.01E-10	3.49E+11	2.52E+04
	325	1.23E-16	1.01E-10	5.10E+11	5.67E+05	1.01E-10	5.10E+11	5.71E+04
	400	2.28E-16	1.00E-10	1.04E+12	2.06E+06	1.00E-10	1.04E+12	3.06E+05
3	200	1.87E-13	1.01E-10	2.70E+09	4.53E+06	1.01E-10	2.70E+09	4.97E+05
	275	4.34E-14	1.00E-10	1.30E+11	4.62E+07	1.00E-10	1.30E+11	1.00E+07
	298	3.31E-14	1.00E-10	2.80E+11	7.38E+07	1.00E-10	2.80E+11	1.85E+07
	325	2.57E-14	1.00E-10	5.88E+11	1.17E+08	1.00E-10	5.88E+11	3.37E+07
	400	1.62E-14	1.00E-10	2.55E+12	2.98E+08	1.00E-10	2.55E+12	1.15E+08
4	200	>> k_{d-d}	>> k_{d-d}	>> k_{d-d}	>> k_{d-d}	1.00E-10	2.33E+08	1.41E+10
	275	>> k_{d-d}	>> k_{d-d}	>> k_{d-d}	>> k_{d-d}	1.00E-10	2.32E+10	1.76E+10
	298	>> k_{d-d}	>> k_{d-d}	>> k_{d-d}	>> k_{d-d}	1.00E-10	5.79E+10	1.82E+10
	325	>> k_{d-d}	>> k_{d-d}	>> k_{d-d}	>> k_{d-d}	1.00E-10	1.41E+11	1.88E+10
	400	>> k_{d-d}	>> k_{d-d}	>> k_{d-d}	>> k_{d-d}	1.00E-10	8.27E+11	1.98E+10
5	200	6.33E-09	1.01E-10	2.09E+08	5.93E+09	1.01E-10	1.20E+08	4.14E+09
	275	9.09E-11	1.00E-10	2.25E+10	9.23E+09	1.01E-10	1.53E+10	7.52E+09
	298	4.65E-11	1.00E-10	3.99E+10	1.00E+10	1.01E-10	3.99E+10	8.43E+09
	325	1.71E-11	1.00E-10	1.40E+11	1.09E+10	1.01E-10	1.02E+11	9.43E+09
	400	3.29E-12	1.00E-10	8.47E+11	1.26E+10	1.00E-10	6.59E+11	1.17E+10

S1.4 sCl 1 + Co-reactant Under Experimental Conditions From the Literature

Table 13: sCl 1 + co-reactants under experimental conditions, atmospheric pressure and temperature applied to MESMER method. The sCl 4 + MeOH rate constant is only for TS2 as TS1 is barrierless.

Reaction	p (Torr)	T (K)	Grain size	k_{CAN} (cm ³ molec. ⁻¹ s ⁻¹)				k_{ME}
				k_1	k_{-1}	k_2	k_{TS}	
				cm ³ s ⁻¹	s ⁻¹	s ⁻¹	cm ³ s ⁻¹	
sCl 1 + HNO ₃	27	295	10	1.01E-10	1.86E+07	1.22E+11	6.62E-07	6.18E-11
	31	295		1.01E-10	1.86E+07	1.22E+11	6.62E-07	6.18E-11
	35	295		1.01E-10	1.86E+07	1.22E+11	6.62E-07	6.18E-11
	760	295		1.01E-10	1.86E+07	1.22E+11	6.62E-07	1.04E-08
sCl 1 + H ₂ O	50	293	10	1.00E-10	7.10E+09	6.12E+03	8.64E-17	1.09E-16
				1.00E-10	7.10E+09	1.59E+03	2.24E-17	
	760	297	10	1.00E-10	8.12E+09	7.43E+03	9.17E-17	1.16E-16
				1.00E-10	8.12E+09	1.96E+03	2.42E-17	
sCl 1 + H ₂ S	100	299	10	1.01E-10	5.64E+12	2.14E+08	3.81E-15	7.10E-15
				1.00E-10	6.11E+12	2.00E+08	3.29E-15	
	250	299	10	1.01E-10	5.64E+12	2.14E+08	3.81E-15	7.10E-15
				1.00E-10	6.11E+12	2.00E+08	3.29E-15	
	500	299	10	1.01E-10	5.64E+12	2.14E+08	3.81E-15	7.10E-15
				1.00E-10	6.11E+12	2.00E+08	3.29E-15	
	100	278	10	1.01E-10	3.88E+12	1.33E+08	3.45E-15	6.40E-15
				1.00E-10	4.23E+12	1.24E+08	2.95E-15	
sCl 1 + MeOH	100	318	10	1.01E-10	7.51E+12	3.10E+08	4.15E-15	7.75E-15
				1.00E-10	8.09E+12	2.91E+08	3.60E-15	
	90	295	10	1.00E-10	4.70E+10	4.59E+06	9.81E-15	1.03E-13
				1.00E-10	4.70E+10	1.11E+06	2.38E-15	
	10	254.5	10	1.00E-10	8.86E+09	1.14E+06	1.29E-14	1.15E-12
				1.00E-10	8.86E+09	2.12E+05	2.41E-15	
sCl 1 + HCl	100	262.1	10	1.00E-10	1.27E+10	1.53E+06	1.21E-14	1.10E-13
				1.00E-10	1.27E+10	3.02E+05	2.40E-15	
	10	292.6	10	1.00E-10	4.32E+10	4.27E+06	9.94E-15	9.29E-13
				1.00E-10	4.32E+10	1.02E+06	2.38E-15	
	10	327.8	10	1.00E-10	1.30E+11	1.10E+07	8.43E-15	8.20E-13
				1.00E-10	1.30E+11	3.13E+06	2.41E-15	
	27	295	5	1.00E-10	3.85E+10	1.49E+12	3.88E-09	5.04E-10
				1.00E-10	3.85E+10	1.49E+12	3.88E-09	5.04E-10
				1.00E-10	3.85E+10	1.49E+12	3.88E-09	5.04E-10
				1.00E-10	3.85E+10	1.49E+12	3.88E-09	4.98E-10

S1.5 Breakdown of k_{TST} Constants using Wigner Factor for Tunneling Constant

Table S14: Breakdown of rate constant (k_{TST}) of sCI + atmospheric co-reactant using: equilibrium rate constant (K_{eq}); Wigner tunneling constant (κ_{Wigner}); unimolecular rate constant (k_2); individual rate constant of transition state (k_{TS}); pathway breakdown (Γ); total rate constant (k_{total});

Note: sCI 4 + HCHO; sCI 1 + CF₃CFO; sCIs 1,3,4 and 5 with SO₂; and sCIs 1-5 with TFA are barrierless reactions (see IRCs in section S7) and so are not included. The sCI 4 + MeOH rate constant is only for TS2 as TS1 is barrierless.

Co-reactant	CI	#TS	K_{eq}	κ_{Wigner}	k_2	k_{TS}	Γ	k_{total}
			cm ³		s ⁻¹	cm ³ s ⁻¹		cm ³ s ⁻¹
HCHO	1	TS _C	2.08E-23	1.01	1.37E+12	2.88E-11	1.00	2.88E-11
	2	TS _C	5.86E-23	1.00	5.00E+9	2.93E-13	1.00	2.93E-13
	3	TS _C	5.20E-23	1.00	1.21E+11	6.28E-12	1.00	6.28E-12
	5	TS _C	4.03E-22	1.00	5.15E+11	2.08E-10	1.00	2.08E-10
CF ₃ CFO	1	TS _{CYC} 1	5.73E-23	1.01	1.15E+11	6.69E-12	1.00	7.86E-12
		TS _{CYC} 2	4.95E-23	1.03	2.29E+10	1.17E-12	1.00	
SO ₂	2	TS _{ENDO}	1.37E-23	1.00	6.73E+10	9.21E-13	1.00	3.21E-12
		TS _{EXO}	1.28E-23	1.00	1.79E+11	2.29E-12	1.00	
HNO ₃	1	TS	4.30E-18	1.06	1.19E+11	5.40E-07	1.00	5.40E-7
	2	TS	1.42E-20	1.04	2.13E+9	3.15E-11	1.00	3.15E-11
	3	TS	3.32E-20	1.07	2.60E+10	9.19E-10	1.00	9.19E-10
	4	TS	8.73E-18	1.00	2.56E+11	2.24E-6	1.00	2.24E-6
	5	TS	4.41E-18	1.11	7.72E+10	3.79E-7	1.00	3.79E-7
HCl	1	TS	2.21E-21	1.01	1.66E+12	3.69E-9	1.00	3.69E-9
	2	TS	2.21E-21	1.00	6.34E+11	4.19E-15	1.00	4.19E-15
	3	TS	1.80E-21	1.13	2.05E+6	1.63E-11	1.00	1.63E-11
	4	TS	1.34E-22	1.17	1.04E+11	7.37E-11	1.00	7.37E-11
	5	TS	1.70E-21	1.01	4.31E+10	3.59E-11	1.00	3.59E-11
HF	1	TS	1.12E-18	1.67	1.46E+5	5.47E-13	1.00	5.47E-13
	2	TS	8.44E-21	1.54	4.51E+2	1.17E-17	1.00	1.17E-17
	3	TS	1.74E-20	1.70	7.82E+4	4.63E-15	1.00	4.63E-15
	4	TS	2.18E-19	1.14	1.02E+8	5.09E-11	1.00	5.09E-11
	5	TS	6.99E-20	1.43	3.38E+6	6.76E-13	1.00	6.76E-13
MeOH	1	TS _{AAAH} 1	1.94E-21	1.09	4.51E+6	1.90E-14	0.80	1.19E-14
		TS _{AAAH} 2	1.94E-21	1.10	1.11E+6	4.72E-15	0.20	
	2	TS _{AAAH} 1	2.96E-22	1.06	2.79E+5	1.75E-16	0.92	9.47E-17
		TS _{AAAH} 2	2.96E-22	1.07	2.30E+4	1.47E-17	0.08	
	3	TS _{AAAH} 1	3.55E-22	1.06	7.11E+7	5.34E-14	0.80	3.34E-14
		TS _{AAAH} 2	3.55E-22	1.05	1.78E+7	1.33E-14	0.20	
	4	TS _{AAAH} 1	>> k_{d-d}	N/A	>> k_{d-d}	>> k_{d-d}	>> k_{d-d}	>> k_{d-d}
		TS _{AAAH} 2	1.79E-21	1.02	1.74E+10	6.33E-11	1.00	
	5	TS _{AAAH} 1	1.72E-21	1.04	9.86E+9	3.53E-11	0.46	3.88E-11
		TS _{AAAH} 2	2.50E-21	1.04	8.14E+9	4.22E-11	0.54	

Table S15: Breakdown of rate constant (k_{TST}) of sCI + atmospheric co-reactant using: equilibrium rate constant (K_{eq}); Wignar tunneling constant (K_{Wignar}); unimolecular rate constant (k_2); individual rate constant of transition state (k_{TS}); pathway breakdown (Γ); total rate constant (k_{total});

Co-reactant	CI	#TS	K_{eq}	K_{Wignar}	k_2	k_{TS}	Γ	k_{total}
			cm^3		cm^3	$\text{cm}^3 \text{s}^{-1}$		$\text{cm}^3 \text{s}^{-1}$
H_2S	1	TS1	1.81E-23	1.08	1.95E+8	3.80E-15	0.54	1.41E-15
		TS2	1.68E-23	1.08	1.79E+8	3.25E-15	0.46	
	2	TS1	5.14E-23	1.03	2.48E+6	1.32E-16	0.66	3.98E-16
		TS2	4.80E-23	1.04	1.35E+6	6.75E-17	0.34	
	3	TS1	8.63E-25	1.04	4.11E+9	3.70E-15	0.63	1.17E-14
		TS2	4.08E-25	1.05	5.01E+9	2.14E-15	0.37	
	4	TS1	1.01E-22	1.01	4.03E+9	4.09E-13	0.71	5.75E-13
		TS2	1.06E-22	1.01	1.55E+9	1.65E-13	0.29	
	5	TS1	1.03E-22	1.04	6.59E+7	7.11E-15	0.58	2.47E-14
		TS2	1.02E-22	1.04	4.91E+7	5.23E-15	0.42	
H_2O	1	TSH ₂ O 1	1.22E-20	1.18	6.40E+3	1.85E-16	0.79	1.17E-16
		TSH ₂ O 2	1.22E-20	1.20	1.67E+3	4.89E-17	0.21	
	2	TSH ₂ O 1	6.07E-22	1.11	6.55E+2	4.41E-19	0.32	1.38E-18
		TSH ₂ O 2	6.07E-22	1.12	1.38E+3	9.40E-19	0.68	
	3	TSH ₂ O 1	2.07E-21	1.11	3.82E+4	8.82E-17	0.78	1.14E-16
		TSH ₂ O 2	2.07E-21	1.13	1.09E+4	2.54E-17	0.22	
	4	TSH ₂ O 1	1.12E-21	1.03	1.94E+9	2.23E-12	0.42	5.37E-12
		TSH ₂ O 2	1.74E-21	1.05	1.72E+9	3.13E-12	0.58	
	5	TSH ₂ O 1	2.06E-21	1.08	4.09E+7	9.11E-14	0.40	2.30E-13
		TSH ₂ O 2	2.06E-21	1.08	6.24E+7	1.39E-13	0.60	
$(\text{H}_2\text{O})_2$	1	TS(H_2O) ₂ 1	3.13E-21	1.11	1.12E+9	3.87E-12	0.27	1.41E-11
		TS(H_2O) ₂ 2	4.28E-21	1.15	1.60E+8	7.87E-13	0.06	
		TS(H_2O) ₂ 3	3.86E-21	1.12	1.08E+9	4.68E-12	0.33	
		TS(H_2O) ₂ 4	2.71E-21	1.10	1.60E+9	4.78E-12	0.34	
	2	TS(H_2O) ₂ 1	1.21E-22	1.08	3.65E+9	4.79E-13	0.20	2.41E-12
		TS(H_2O) ₂ 2	1.17E-22	1.07	1.10E+10	1.38E-12	0.57	
		TS(H_2O) ₂ 3	4.99E-23	1.06	2.62E+9	1.39E-13	0.06	
		TS(H_2O) ₂ 4	5.39E-23	1.06	7.17E+9	4.09E-13	0.17	
	3	TS(H_2O) ₂ 1	2.20E-22	1.07	1.35E+10	3.17E-12	0.08	3.84E-11
		TS(H_2O) ₂ 2	1.83E-22	1.05	8.91E+10	1.72E-11	0.45	
		TS(H_2O) ₂ 3	3.47E-22	1.05	2.18E+10	7.95E-12	0.21	
		TS(H_2O) ₂ 4	3.17E-22	1.05	3.01E+10	1.00E-11	0.26	
	4	TS(H_2O) ₂ 1	1.16E-19	1.04	1.58E+12	1.90E-7	0.14	1.36E-06
		TS(H_2O) ₂ 2	1.53E-19	1.03	3.35E+12	5.28E-7	0.39	
		TS(H_2O) ₂ 3	5.45E-20	1.02	3.96E+12	2.21E-7	0.16	
		TS(H_2O) ₂ 4	6.28E-20	1.02	6.59E+12	4.21E-7	0.31	
	5	TS(H_2O) ₂ 1	9.31E-20	1.05	1.39E+11	1.36E-8	0.19	7.22E-08
		TS(H_2O) ₂ 2	1.32E-19	1.05	3.21E+11	4.44E-8	0.61	
		TS(H_2O) ₂ 3	1.06E-19	1.06	8.56E+10	9.60E-9	0.13	
		TS(H_2O) ₂ 4	1.00E-19	1.07	4.35E+10	4.68E-9	0.06	

S1.5 Conventional Transition State Theory Rate Constant (k_{TST}) Equations

$$k_{TST} = K_{eq} \cdot k_2 \quad \text{Equation S1}$$

$$K_{eq} = \frac{RT}{P_0} e^{-(G_{PRC} - (G_{CI} + G_{Co})) / RT} \quad \text{Equation S2}$$

$$k_2 = \kappa \frac{k_B T}{h} e^{-(G_{TS} - G_{PRC}) / RT} \quad \text{Equation S3}$$

$$k_{total} = \sum k_{TST} \quad \text{Equation S4}$$

S2. MEMSER derived product branching ratios in HCHO & SO₂ reactions

S2.1 Full potential energy surfaces for HCHO & SO₂ + sCl reactions

The complexities of the HCHO & SO₂ + sCl systems are not fully expressed in the diagrams in the main manuscript, which only exhibit a condensed and lowest energy path version. The visual representation of such potential energy surfaces for HCHO and SO₂ + sCl systems are found in this section. For all the energy values on the potential energy surface, please see Section S5.

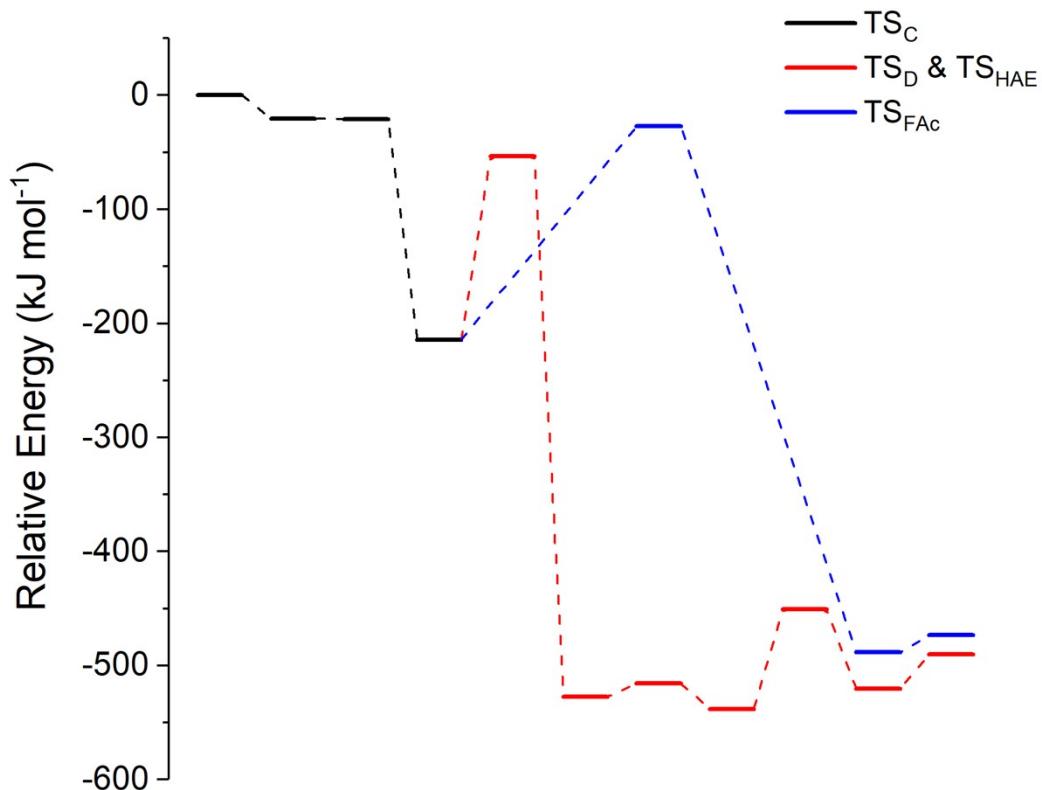


Figure S1: sCl1 + HCHO full potential energy surface with labels of channels in legend.

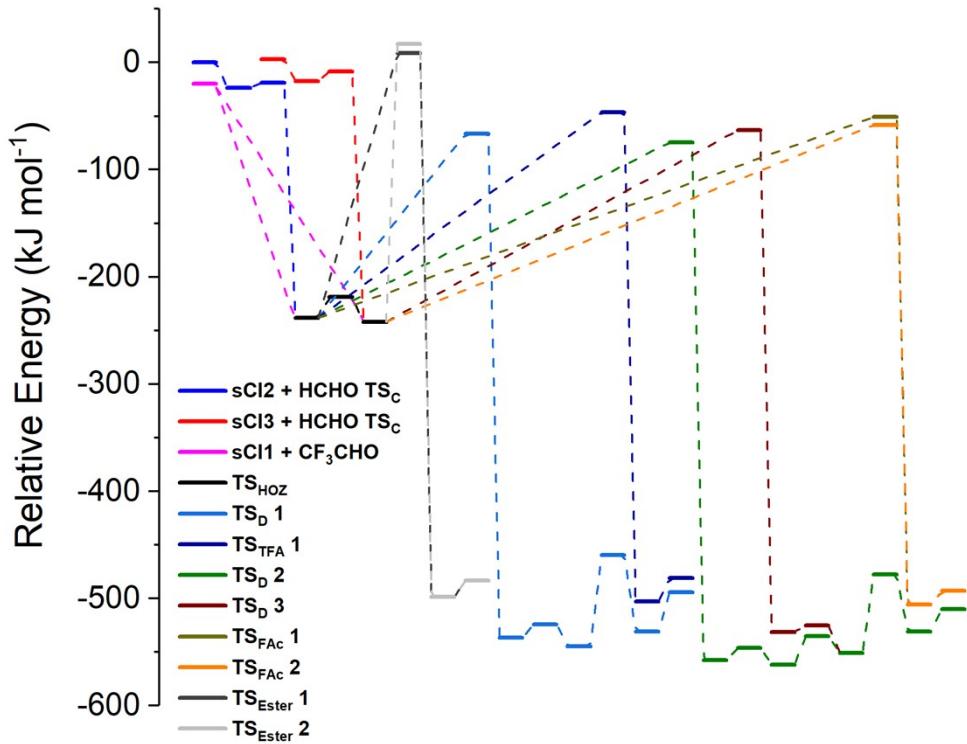


Figure S2: Potential Energy Surface including all reactions HCHO + sClIs 2 & 3 & sCl1 + CF₃CHO with legend attached.

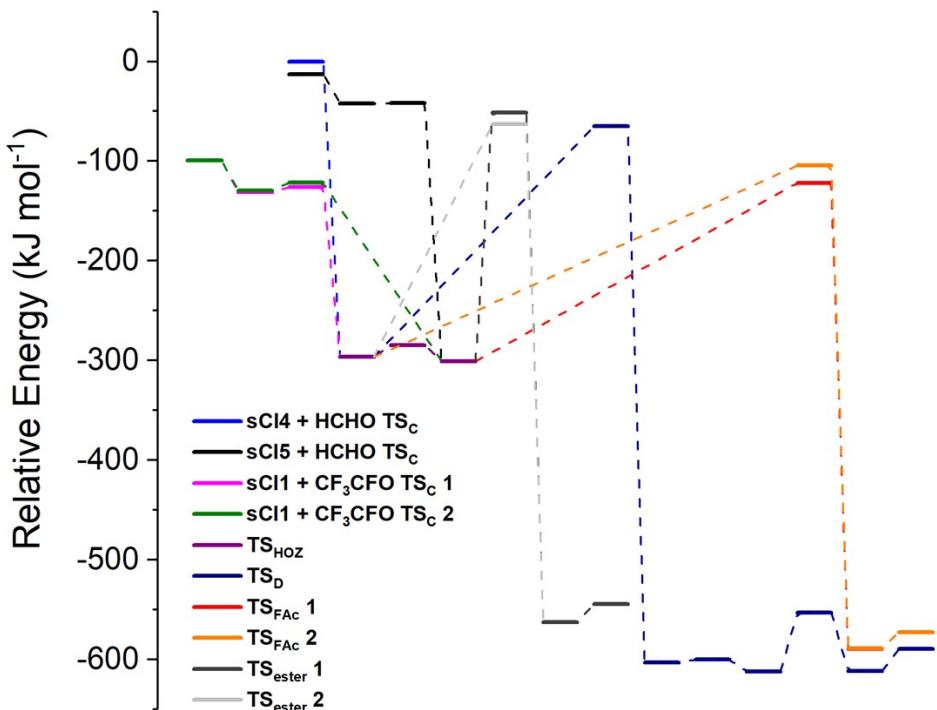


Figure S3: Potential Energy Surface including all reactions HCHO + sClIs 4 & 5 & sCl1 + CF₃CFO with legend attached

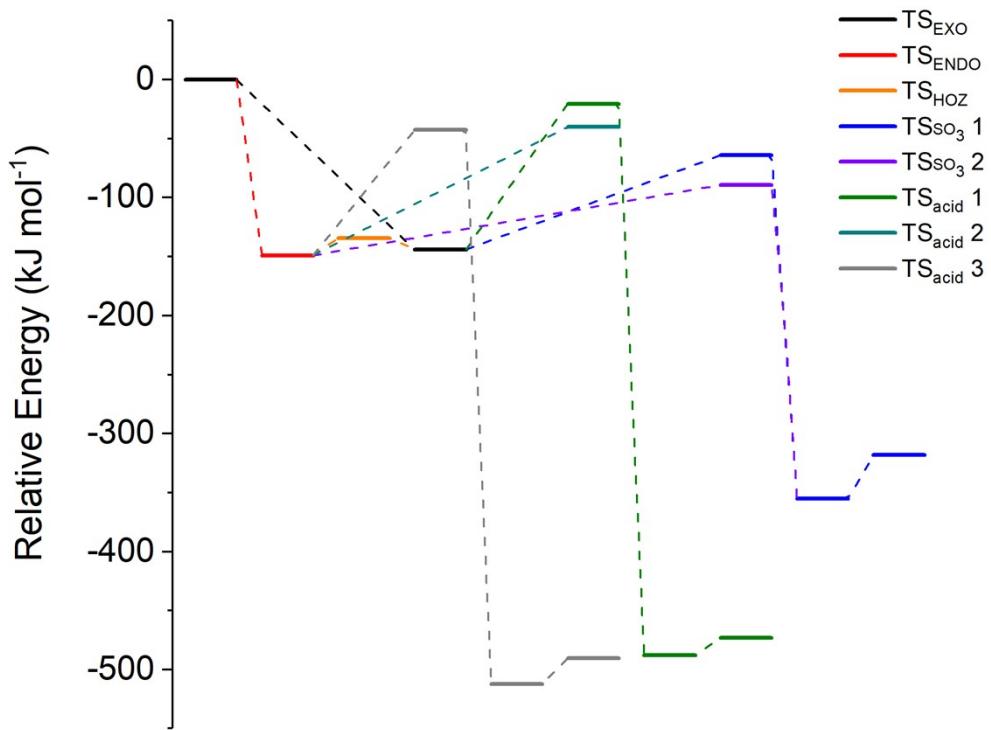


Figure S4: Full Potential Energy Surface of *sCl1* + SO_2 with legend attached.

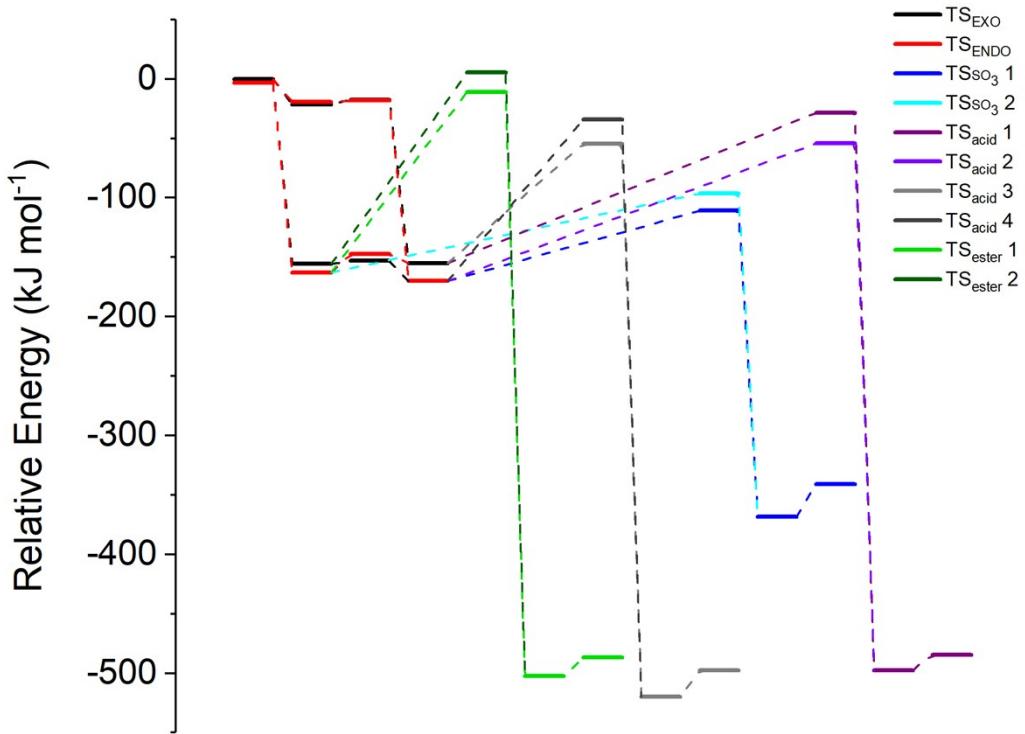
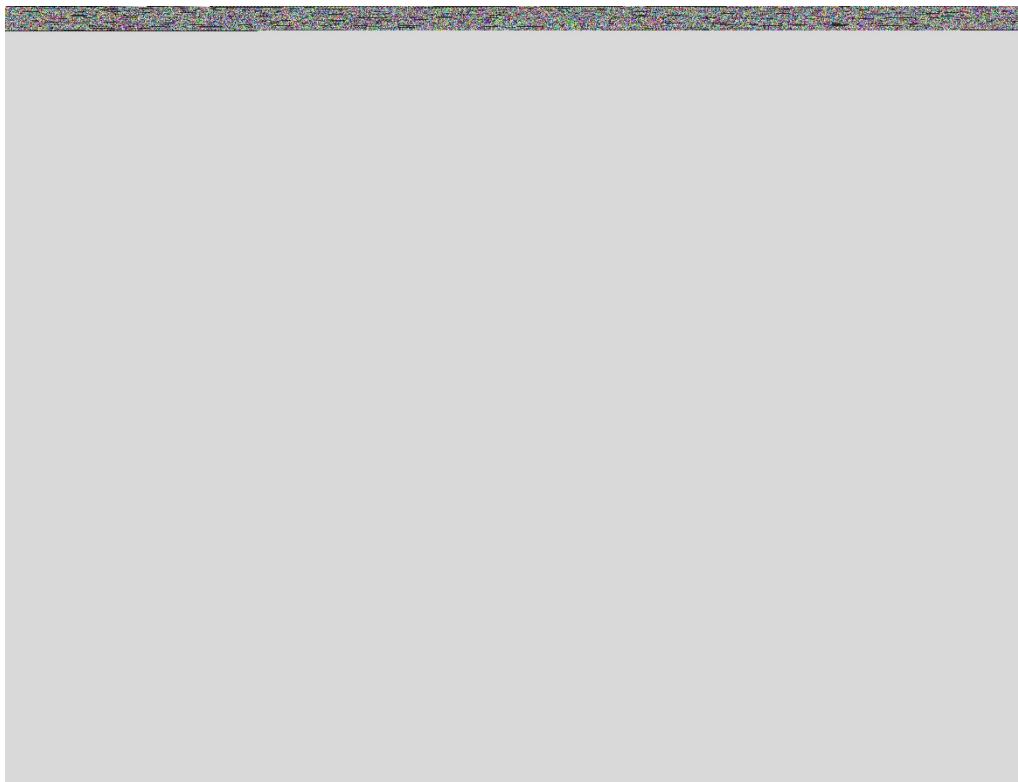


Figure S5: Potential Energy Surface including all *sCl2* & *sCl3* + SO_2 reactions with legend attached.



*Figure S6: Potential Energy Surface including all **sCI4** & **sCI5** + SO₂ reactions with legend attached*

S2.2 Product branching ratio of sCI 1 reactions HCHO, CF₃CHO and CF₃CFO

As seen in Figures S2, S3, S4 & S5, many of these reactions lead to barrierless POZ-to-final product pathways. To determine their product branching fraction, the MESMER software employs a reverse ILT method using a user provided pre-exponential. This pre-exponential can be either a previously measured literature k_{EXP} value for the forward reaction barrierless channels or a calculated rate-limiting value. Here the product branching ratios are calculated in the following tables excluding the barrierless channels entirely for the purpose of being comprehensive and using either the dipole-dipole capture limit (k_{d-d}) or the collision-limiting rate constant (k_{COLL}) as the pre-exponential. The product branching ratios seen in the main body of the text uses the k_{d-d} constant for the reverse ILT pre-exponential. For more details, the dipole-dipole capture limit (k_{d-d}) and the collision-limiting rate constant (k_{COLL}) see Section S3.

Table S16: The product branching ratio from the sCI 1 + HCHO reaction of: the heterozonide intermediate product (Γ_{HOZ}); direct transfer pathway ($\Gamma_{TS_{FAC}}$); hydroxyalkyl hydroperoxide pathway ($\Gamma_{TS_{HAE}}$) using MESMER grainsize of 10.

T (K)	Γ_{HOZ}	$\Gamma_{TS_{FAC}}$	$\Gamma_{TS_{HAE}}$
200	1.000	0.000	0.000
275	0.000	0.025	0.975
298.15	0.000	0.027	0.973
325	0.000	0.030	0.970
400	0.000	0.038	0.962

Table S17: The product branching ratio of the sCI 1 + CF₃CHO reaction with CI 2 + HCHO and CI 3 + HCHO as sinks (not barrierless) using MESMER grainsize of 40.

Product branching ratio (r)	Temperature (K)				
	200	275	298	325	400
CI 2 + HCHO (TS_{CYC} 1)	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001
CI 3 + HCHO (TS_{CYC} 2)	<0.0001	<0.0001	<0.0001	<0.0001	0.0001
HCOOH + CF₃CHO	0.8447	0.8282	0.8232	0.8176	0.8051
TS _{HAE} 2	0.6908	0.6479	0.6344	0.6195	0.5862
TS _{HAE} 3	0.0969	0.1084	0.1118	0.1155	0.1234
TS _{FAC} 1	0.0097	0.0142	0.0158	0.0177	0.0222
TS _{FAC} 2	0.0473	0.0578	0.0611	0.0648	0.0733
CF₃COOH + HCHO	0.1553	0.1717	0.1768	0.1824	0.1948
TS _{HAE} 1	0.1499	0.1624	0.1659	0.1696	0.1771
TS _{TFA} 1	0.0054	0.0094	0.0109	0.0128	0.0177
CF₃OCHO + HCHO	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001
TS _{ESTER} 1	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001
TS _{ESTER} 2	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001

Table S18: the product branching ratio of the sCl 1 + CF₃CFO reaction with Cl 4 + HCHO and Cl 5 + HCHO as sinks for barrierless exit channel using the k_{d-d} as pre-exponential factors (7.20 × 10⁻¹⁰ cm³ molec⁻¹ s⁻¹). MESMER grainsize = 40

Product branching ratio (<i>r</i>)	Temperature (K)				
	200	275	298	325	400
Cl 4 + HCHO (TS_{CYC} 1)	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001
Cl 5 + HCHO (TS_{CYC} 2)	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001
HCOOH + CF₃CFO	1.0000	1.0000	1.0000	1.0000	1.0000
TS _{HAE} 2	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001
TS _{FAC} 1	0.9940	0.9839	0.9784	0.9705	0.9419
TS _{FAC} 2	0.0060	0.0161	0.0216	0.0295	0.0581
CF₃OCFO + HCHO	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001
TS _{ESTER} 1	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001
TS _{ESTER} 2	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001

Table S19: The product branching ratio of the sCl 1 + CF₃CFO reaction with Cl 4 + HCHO and Cl 5 + HCHO as sinks using the k_{COLL} as pre-exponential factor (1.49 × 10⁻¹⁰ cm³ molec⁻¹ s⁻¹) for barrierless exit channels. MESMER grainsize = 40

Product branching ratio (<i>r</i>)	Temperature (K)				
	200	275	298	325	400
Cl 4 + HCHO (TS_{CYC} 1)	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001
Cl 5 + HCHO (TS_{CYC} 2)	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001
HCOOH + CF₃CFO	1.0000	1.0000	1.0000	1.0000	1.0000
TS _{HAE} 2	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001
TS _{FAC} 1	0.9940	0.9839	0.9784	0.9705	0.9419
TS _{FAC} 2	0.0060	0.0161	0.0216	0.0295	0.0581
CF₃OCFO + HCHO	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001
TS _{ESTER} 1	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001
TS _{ESTER} 2	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001

(Due to the very low yield of Cl 4 + HCHO & Cl 5 + HCHO, no models were ran excluding these pathways for sCl 1 + CF₃CFO reactions, as they would yield the same results)

S2.3 Product branching ratio of HCHO reactions with sClIs 2 & 3

Table S20: the product branching ratio of the **sCl 2 + HCHO** reaction with **Cl 1 + CF₃CHO** and **Cl 3 + HCHO** as sinks using the k_{d-d} as pre-exponential factor ($6.50 \times 10^{-10} \text{ cm}^3 \text{ molec}^{-1} \text{ s}^{-1}$) for barrierless exit channels – using a MESMER grainsize of 40.

Product branching ratio (r)	Temperature (K)				
	200	275	298	325	400
Cl 1 + CF₃CHO	0.9809	0.9915	0.9931	0.9944	0.9965
TS _{cyc} 3 (barrierless)	0.4914	0.4965	0.4973	0.4980	0.4998
TS _{cyc} 4 (barrierless)	0.4895	0.4950	0.4958	0.4963	0.4967
Cl 3 + HCHO (TS_{CYC} 2)	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001
HCOOH + CF₃CHO	0.0158	0.0069	0.0056	0.0046	0.0028
TS _{HAE} 2	0.0121	0.0052	0.0042	0.0033	0.0020
TS _{HAE} 3	0.0022	0.0010	0.0008	0.0007	0.0004
TS _{FAC} 1	0.0003	0.0002	0.0001	0.0001	0.0001
TS _{FAC} 2	0.0012	0.0006	0.0005	0.0004	0.0003
CF₃COOH + HCHO	0.0034	0.0016	0.0013	0.0011	0.0007
TS _{HAE} 1	0.0032	0.0015	0.0012	0.0010	0.0006
TS _{TFA} 1	0.0002	0.0001	0.0001	0.0001	0.0001
CF₃OCHO + HCHO	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001
TS _{ESTER} 1	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001
TS _{ESTER} 2	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001

Table S21: the product branching ratio of the **sCl 2 + HCHO** reaction with **Cl 1 + CF₃CHO** and **Cl 3 + HCHO** as sinks using the k_{COLL} as pre-exponential factor ($1.52 \times 10^{-10} \text{ cm}^3 \text{ molec}^{-1} \text{ s}^{-1}$) for barrierless exit channels – using a MESMER grainsize of 40.

Product branching ratio (r)	Temperature (K)				
	200	275	298	325	400
Cl 1 + CF₃CHO	0.9251	0.9695	0.9758	0.9809	0.9884
TS _{cyc} 3 (barrierless)	0.4634	0.4853	0.4885	0.4910	0.4949
TS _{cyc} 4 (barrierless)	0.4617	0.4841	0.4873	0.4899	0.4935
Cl 3 + HCHO (TS_{CYC} 2)	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001
HCOOH + CF₃CHO	0.0617	0.0248	0.0195	0.0153	0.0092
TS _{HAE} 2	0.0475	0.0184	0.0144	0.0111	0.0065
TS _{HAE} 3	0.0084	0.0037	0.0029	0.0024	0.0015
TS _{FAC} 1	0.0012	0.0006	0.0005	0.0004	0.0003
TS _{FAC} 2	0.0046	0.0021	0.0017	0.0014	0.0009
CF₃COOH + HCHO	0.0133	0.0058	0.0046	0.0037	0.0024
TS _{HAE} 1	0.0125	0.0053	0.0042	0.0034	0.0021
TS _{TFA} 1	0.0008	0.0004	0.0004	0.0003	0.0003
CF₃OCHO + HCHO	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001
TS _{ESTER} 1	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001
TS _{ESTER} 2	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001

Table S22: the product branching ratio of the **sCl 2 + HCHO** reaction with no reference to **Cl 1 + CF₃CHO** and **Cl 3 + HCHO** as sinks using a MESMER grainsize of 30.

Product branching ratio (<i>r</i>)	Temperature (K)				
	200	275	298	325	400
HCOOH + CF₃CHO	0.8227	0.8101	0.8054	0.7996	0.7836
TS _{HAE} 2	0.6331	0.5992	0.5867	0.5712	0.5284
TS _{HAE} 3	0.1125	0.1206	0.1235	0.1269	0.1356
TS _{FAC} 1	0.0157	0.0202	0.0220	0.0244	0.0315
TS _{FAC} 2	0.0614	0.0700	0.0732	0.0771	0.0881
CF₃COOH + HCHO	0.1773	0.1899	0.1946	0.2004	0.2164
TS _{HAE} 1	0.1667	0.1745	0.1771	0.1801	0.1870
TS _{TFA} 1	0.0106	0.0154	0.0175	0.0203	0.0294
CF₃OCHO + HCHO	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001
TS _{ESTER} 1	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001
TS _{ESTER} 2	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001

Table S23: the product branching ratio of **sCl 3 + HCHO** with **Cl 1 + CF₃CHO** and **Cl 2 + HCHO** as sinks using the k_{d-d} as pre-exponential factor ($6.50 \times 10^{-10} \text{ cm}^3 \text{ molec}^{-1} \text{ s}^{-1}$) for barrierless exit channels – using a MESMER grainsize of 40.

Product branching ratio (<i>r</i>)	Temperature (K)				
	200	275	298	325	400
Cl 1 + CF₃CHO	0.9749	0.9888	0.9910	0.9928	0.9956
TS _{cyc} 3	0.4873	0.4943	0.4953	0.4961	0.4970
TS _{cyc} 4	0.4876	0.4945	0.4956	0.4966	0.4985
Cl 2 + HCHO (TS_{CYC} 1)	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001
HCOOH + CF₃CHO	0.0207	0.0091	0.0074	0.0059	0.0036
TS _{HAE} 2	0.0161	0.0069	0.0055	0.0044	0.0026
TS _{HAE} 3	0.0028	0.0013	0.0011	0.0009	0.0005
TS _{FAC} 1	0.0004	0.0002	0.0002	0.0001	0.0001
TS _{FAC} 2	0.0015	0.0007	0.0006	0.0005	0.0003
CF₃COOH + HCHO	0.0044	0.0020	0.0017	0.0014	0.0009
TS _{HAE} 1	0.0041	0.0019	0.0016	0.0013	0.0008
TS _{TFA} 1	0.0002	0.0001	0.0001	0.0001	0.0001
CF₃OCHO + HCHO	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001
TS _{ESTER} 1	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001
TS _{ESTER} 2	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001

Table S24: the product branching ratio of **sCI 3 + HCHO** with **Cl 1 + CF₃CHO** and **Cl 2 + HCHO** as sinks using the k_{COLL} as pre-exponential factor ($1.52 \times 10^{-10} \text{ cm}^3 \text{ molec}^{-1} \text{ s}^{-1}$) for barrierless exit channels – using a MESMER grainsize of 40.

Product branching ratio (r)	Temperature (K)				
	200	275	298	325	400
Cl 1 + CF₃CHO	0.9027	0.9593	0.9683	0.9756	0.9858
TS _{cyc} 3 (barrierless)	0.4512	0.4796	0.4841	0.4877	0.4927
TS _{cyc} 4 (barrierless)	0.4515	0.4797	0.4842	0.4878	0.4931
Cl 2 + HCHO (TS_{CYC} 1)	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001
HCOOH + CF₃CHO	0.0803	0.0332	0.0257	0.0197	0.0113
TS _{HAE} 2	0.0624	0.0250	0.0192	0.0145	0.0081
TS _{HAE} 3	0.0107	0.0047	0.0038	0.0030	0.0018
TS _{FAC} 1	0.0014	0.0007	0.0006	0.0005	0.0003
TS _{FAC} 2	0.0058	0.0027	0.0022	0.0017	0.0011
CF₃COOH + HCHO	0.0170	0.0075	0.0059	0.0047	0.0028
TS _{HAE} 1	0.0160	0.0070	0.0055	0.0043	0.0026
TS _{TFA} 1	0.0009	0.0005	0.0005	0.0004	0.0003
CF₃OCHO + HCHO	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001
TS _{ESTER} 1	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001
TS _{ESTER} 2	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001

Table S25: the product branching ratio of **sCI 3 + HCHO** without using **Cl 1 + CF₃CHO** and **Cl 2 + HCHO** as sinks – using a MESMER grainsize of 40.

Product branching ratio (r)	Temperature (K)				
	200	275	298	325	400
HCOOH + CF₃CHO	0.8250	0.8154	0.8116	0.8065	0.7907
TS _{HAE} 2	0.6398	0.6142	0.6037	0.5901	0.5478
TS _{HAE} 3	0.1107	0.1170	0.1195	0.1226	0.1317
TS _{FAC} 1	0.0149	0.0182	0.0196	0.0216	0.0283
TS _{FAC} 2	0.0595	0.0660	0.0687	0.0722	0.0830
CF₃COOH + HCHO	0.1750	0.1846	0.1884	0.1935	0.2093
TS _{HAE} 1	0.1652	0.1714	0.1737	0.1765	0.1840
TS _{TFA} 1	0.0098	0.0132	0.0148	0.0170	0.0252
CF₃OCHO + HCHO	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001
TS _{ESTER} 1	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001
TS _{ESTER} 2	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001

S2.4 Product branching ratio of HCHO reactions with sClIs 4 & 5

Table S26: the product branching ratio of **sCl 4 + HCHO** with **Cl 1 + CF₃CFO** and **Cl 5 + HCHO** no barrierless exit channels.

Product branching ratio (<i>r</i>)	Temperature (K)				
	200	275	298	325	400
Cl 1 + CF₃CFO	0.8441	0.8437	0.8436	0.8435	0.8429
TS _{CYC} 3	0.5708	0.5701	0.5698	0.5695	0.5683
TS _{CYC} 4	0.2732	0.2737	0.2738	0.2740	0.2746
Cl 5 + HCHO (TS_{CYC} 2)	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001
HCOOH + CF₃CFO	0.1559	0.1562	0.1564	0.1565	0.1570
TS _{HAE} 2	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001
TS _{FAC} 1	0.1354	0.1353	0.1353	0.1353	0.1353
TS _{FAC} 2	0.0205	0.0209	0.0210	0.0212	0.0217
CF₃OCFO + HCHO	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001
TS _{ESTER} 1	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001
TS _{ESTER} 2	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001

Table S27: The product branching ratio of **sCl 4 + HCHO** with no **Cl 1 + CF₃CFO** and **Cl 4 + HCHO** exit channels using a MESMER grainsize of 20

Product branching ratio (<i>r</i>)	Temperature (K)				
	200	275	298	325	400
HCOOH + CF₃CFO	0.9996	0.9992	0.9990	0.9988	0.9978
TS _{HAE} 2	0.0004	0.0006	0.0007	0.0009	0.0016
TS _{FAC} 1	0.8450	0.8271	0.8209	0.8133	0.7910
TS _{FAC} 2	0.1542	0.1715	0.1774	0.1845	0.2053
CF₃OCFO + HCHO	0.0004	0.0008	0.0010	0.0012	0.0022
TS _{ESTER} 1	<0.0001	0.0001	0.0001	0.0002	0.0004
TS _{ESTER} 2	0.0004	0.0007	0.0008	0.0011	0.0019

Table S28: the product branching ratio of **sCl 5 + HCHO** with **Cl 1 + CF₃CFO** and **Cl 4 + HCHO** as sinks using the k_{d-d} as pre-exponential factor ($7.20 \times 10^{-10} \text{ cm}^3 \text{ molec}^{-1} \text{ s}^{-1}$) for barrierless exit channels using MESMER grainsize of 40

Product branching ratio (r)	Temperature (K)				
	200	275	298	325	400
Cl 1 + CF₃CFO	0.8456	0.8432	0.8421	0.8406	0.8355
TS _{CYC} 3	0.5750	0.5693	0.5669	0.5637	0.5541
TS _{CYC} 4	0.2705	0.2740	0.2753	0.2769	0.2813
Cl 4 + HCHO (TS_{CYC} 1)	<0.0001	<0.0001	<0.0001	<0.0001	0.0002
HCOOH + CF₃CFO	0.1544	0.1567	0.1578	0.1593	0.1641
TS _{HAE} 2	<0.0001	<0.0001	<0.0001	0.0001	0.0001
TS _{FAC} 1	0.1356	0.1354	0.1354	0.1354	0.1356
TS _{FAC} 2	0.0188	0.0213	0.0224	0.0238	0.0284
CF₃OCFO + HCHO	<0.0001	<0.0001	<0.0001	0.0001	0.0002
TS _{ESTER} 1	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001
TS _{ESTER} 2	<0.0001	<0.0001	<0.0001	0.0001	0.0001

Table S29: the product branching ratio of **sCl 5 + HCHO** with **Cl 1 + CF₃CFO** and **Cl 4 + HCHO** as sinks using the k_{COLL} as pre-exponential factor ($1.49 \times 10^{-10} \text{ cm}^3 \text{ molec}^{-1} \text{ s}^{-1}$) for barrierless exit channels using MESMER grainsize of 40.

Product branching ratio (r)	Temperature (K)				
	200	275	298	325	400
Cl 1 + CF₃CFO	0.8455	0.8432	0.8422	0.8407	0.8356
TS _{CYC} 3	0.5748	0.5693	0.5669	0.5637	0.5542
TS _{CYC} 4	0.2706	0.2740	0.2753	0.2769	0.2814
Cl 4 + HCHO (TS_{CYC} 1)	<0.0001	<0.0001	<0.0001	<0.0001	0.0001
HCOOH + CF₃CFO	0.1545	0.1567	0.1578	0.1593	0.1642
TS _{HAE} 2	<0.0001	<0.0001	<0.0001	0.0001	0.0001
TS _{FAC} 1	0.1357	0.1354	0.1354	0.1354	0.1356
TS _{FAC} 2	0.0188	0.0213	0.0224	0.0238	0.0285
CF₃OCFO + HCHO	<0.0001	<0.0001	<0.0001	0.0001	0.0002
TS _{ESTER} 1	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001
TS _{ESTER} 2	<0.0001	<0.0001	<0.0001	0.0001	0.0001

Table S30: product branching ratios of **sCl 5 + HCHO** with no **Cl + aldehyde** exit channels using MESMER grainsize of 30

Product branching ratio (r)	Temperature (K)				
	200	275	298	325	400
HCOOH + CF₃CFO	0.9999	0.9998	0.9997	0.9996	0.9992
TS _{HAE} 2	0.0001	0.0002	0.0003	0.0003	0.0006
TS _{FAC} 1	0.8735	0.8624	0.8581	0.8524	0.8324
TS _{FAC} 2	0.1263	0.1372	0.1414	0.1470	0.1662
CF₃OCFO + HCHO	0.0001	0.0002	0.0003	0.0004	0.0008
TS _{ESTER} 1	<0.0001	<0.0001	<0.0001	<0.0001	0.0001
TS _{ESTER} 2	0.0001	0.0002	0.0003	0.0003	0.0007

S2.5 sCl 1 + HCHO & SO₂ yields of the HOZ/SOZ intermediate products

Table S31: Yields from the first step of the sCl + CF₃CHO & SO₂ reactions.

Intermediate Product branching ratio (Γ)	Temperature (K)				
	200	275	298	325	400
sCl 1 + CF₃CFO					
TS _{cyc} 1	0.7486	0.7647	0.7663	0.7668	0.7609
TS _{cyc} 2	0.2514	0.2353	0.2337	0.2332	0.2391
sCl 2 + SO₂					
TS _{EXO}	0.3278	0.3102	0.3070	0.3042	0.3001
TS _{ENDO}	0.6722	0.6898	0.6930	0.6958	0.6999
sCl 3 + SO₂					
TS _{EXO}	0.4871	0.4991	0.4987	0.4975	0.4905
TS _{ENDO}	0.5129	0.5009	0.5013	0.5025	0.5095
sCl 4 + SO₂					
TS _{EXO}	0.0027	0.4981	0.4990	0.4994	0.4998
TS _{ENDO}	0.9973	0.5019	0.5010	0.5006	0.5002
sCl 5 + SO₂					
TS _{EXO}	0.4715	0.4905	0.4938	0.4962	0.4986
TS _{ENDO}	0.5285	0.5095	0.5062	0.5038	0.5014

(All other sCl + aldehyde/ketone and sCl + SO₂ reactions in this study produce one HOZ or SOZ conformer at 100% efficiency)

S2.6 Product branching ratio of SO₂ reactions with sCl 1

Table S32: SOZ fragmentation branching ratios (Γ) of sCl 1 + SO₂ reaction all identified pathways included (grainsize=20)

Product branching ratio (Γ)	Temperature (K)				
	200	275	298	325	400
SO ₃ + HCHO	0.9949	0.9936	0.9931	0.9926	0.9908
TS _{SO₃} 1	0.0649	0.0694	0.0709	0.0725	0.0771
TS _{SO₃} 2	0.9106	0.9001	0.8966	0.8925	0.8803
SO ₂ + HCOOH	0.0051	0.0064	0.0069	0.0074	0.0092
TS _{acid} 1	0.0001	0.0001	0.0002	0.0002	0.0003
TS _{acid} 2	0.0049	0.0061	0.0065	0.0070	0.0086
TS _{acid} 3	0.0196	0.0242	0.0258	0.0278	0.0337

Table S33: SOZ decomposition branching ratios (Γ) of sCl 1 + SO₂ reaction with some pathways included (grainsize=10)

Product branching ratio (Γ)	Temperature (K)				
	200	275	298	325	400
SO ₃ + HCHO	0.9947	0.9933	0.9928	0.9922	0.9903
TS _{SO₃} 1	0.0640	0.0686	0.0700	0.0717	0.0764
TS _{SO₃} 2	0.9307	0.9247	0.9227	0.9204	0.9138
SO ₂ + HCOOH	0.0053	0.0067	0.0072	0.0078	0.0097
TS _{acid} 1	0.0001	0.0001	0.0002	0.0002	0.0003
TS _{acid} 2	0.0052	0.0066	0.0071	0.0076	0.0094

S2.7 SO₂ reactions with sCl 2 & 3

Table S34: SOZ decomposition branching ratios (Γ) of sCl 2 + SO₂ reaction all identified pathways included (grainsize=50), using the k_{d-d} as pre-exponential factor ($4.08 \times 10^{-10} \text{ cm}^3 \text{ molec}^{-1} \text{ s}^{-1}$) for cyclo-reversion pathway.

Product branching ratio (Γ)	Temperature (K)				
	200	275	298	325	400
sCl 3 + SO ₂	0.0001	0.0012	0.0023	0.0046	0.0224
TS _{EXO}	0.0001	0.0009	0.0017	0.0032	0.0148
TS _{ENDO}	<0.0001	0.0003	0.0007	0.0013	0.0076
SO ₃ + CF ₃ CHO	0.9988	0.9971	0.9958	0.9931	0.9733
TS _{SO₃} 1	0.6907	0.6910	0.6909	0.6923	0.6883
TS _{SO₃} 2	0.3081	0.3061	0.3048	0.3008	0.2850
SO ₂ + CF ₃ COOH	0.0011	0.0016	0.0019	0.0024	0.0043
TS _{acid} 1	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001
TS _{acid} 2	0.0002	0.0003	0.0004	0.0004	0.0008
TS _{acid} 3	0.0008	0.0013	0.0015	0.0018	0.0032
TS _{acid} 4	<0.0001	<0.0001	0.0001	0.0001	0.0002
SO ₂ + CF ₃ OCHO	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001
TS _{ester} 1	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001
TS _{ester} 2	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001

Table S35: SOZ decomposition branching ratios (Γ) of sCl 2 + SO₂ reaction without cyclo-revision pathway to Cl 3 + SO₂ revision. Grainsize = 40.

Product branching ratio (Γ)	Temperature (K)				
	200	275	298	325	400
SO ₃ + CF ₃ CHO	0.9989	0.9984	0.9981	0.9976	0.9955
TS _{SO₃} 1	0.6910	0.6901	0.6914	0.6936	0.6958
TS _{SO₃} 2	0.3080	0.3082	0.3067	0.3040	0.2997
SO ₂ + CF ₃ COOH	0.0011	0.0016	0.0019	0.0024	0.0045
TS _{acid} 1	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001
TS _{acid} 2	0.0002	0.0003	0.0004	0.0005	0.0008
TS _{acid} 3	0.0008	0.0013	0.0015	0.0018	0.0034
TS _{acid} 4	<0.0001	<0.0001	0.0001	0.0001	0.0003
SO ₂ + CF ₃ OCHO	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001
TS _{ester} 1	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001
TS _{ester} 2	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001

Table S36: SOZ decomposition branching ratios (Γ) of sCl 2 + SO₂ reaction without sCl pathways and with only pathways that lead directly to SOZs included. Grainsize = 30.

Product branching ratio (Γ)	Temperature (K)				
	200	275	298	325	400
SO ₃ + CF ₃ CHO	0.9998	0.9997	0.9996	0.9995	0.9992
TS _{SO₃} 1	0.6921	0.6903	0.6928	0.6955	0.6992
TS _{SO₃} 2	0.3077	0.3094	0.3069	0.3041	0.3000
SO ₂ + CF ₃ COOH	0.0002	0.0003	0.0004	0.0005	0.0008
TS _{acid} 1	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001
TS _{acid} 2	0.0002	0.0003	0.0004	0.0005	0.0008

Table S37: *sCl 3 + SO₂* SOZ fragmentation branching ratios (Γ) with all identified pathways included. Grainsize = 50

Product branching ratio (r)	Temperature (K)				
	200	275	298	325	400
sCl 2 + SO₂	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001
TS _{EXO}	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001
TS _{ENDO}	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001
SO₃ + CF₃CHO	0.9987	0.9977	0.9973	0.9966	0.9941
TS _{SO₃} 1	0.5088	0.4989	0.4989	0.4995	0.5045
TS _{SO₃} 2	0.4899	0.4989	0.4984	0.4971	0.4896
SO₂ + CF₃COOH	0.0013	0.0023	0.0027	0.0033	0.0059
TS _{acid} 1	<0.0001	<0.0001	<0.0001	<0.0001	0.0001
TS _{acid} 2	0.0002	0.0004	0.0005	0.0006	0.0010
TS _{acid} 3	0.0010	0.0016	0.0019	0.0024	0.0040
TS _{acid} 4	0.0001	0.0002	0.0003	0.0004	0.0008
SO₂ + CF₃OCHO	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001
TS _{ester} 1	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001
TS _{ester} 2	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001

Table S38: SOZ fragmentation branching ratios (Γ) of the *sCl 3 + SO₂* reaction with all identified pathways except cyclo-reversion pathways *Cl 2 + SO₂*. Grainsize = 40

Product branching ratio (r)	Temperature (K)				
	200	275	298	325	400
SO₃ + CF₃CHO	0.9987	0.9977	0.9973	0.9967	0.9941
TS _{SO₃} 1	0.5117	0.4989	0.4989	0.4995	0.5045
TS _{SO₃} 2	0.4870	0.4989	0.4984	0.4971	0.4896
SO₂ + CF₃COOH	0.0013	0.0023	0.0027	0.0033	0.0059
TS _{acid} 1	<0.0001	<0.0001	<0.0001	<0.0001	0.0001
TS _{acid} 2	0.0002	0.0004	0.0005	0.0006	0.0010
TS _{acid} 3	0.0010	0.0016	0.0019	0.0024	0.0040
TS _{acid} 4	0.0001	0.0002	0.0003	0.0004	0.0008
SO₂ + CF₃OCHO	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001
TS _{ester} 1	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001
TS _{ester} 2	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001

Table S39: SOZ decomposition branching ratios (Γ) of *sCl 2 + SO₂* reaction without *sCl* pathways and with only pathways that lead directly to SOZs included. Grainsize = 30

Product branching ratio (r)	Temperature (K)				
	200	275	298	325	400
SO₃ + CF₃CHO	0.9997	0.9996	0.9995	0.9994	0.9990
TS _{SO₃} 1	0.5102	0.5005	0.5008	0.5019	0.5086
TS _{SO₃} 2	0.4895	0.4991	0.4987	0.4975	0.4904
SO₂ + CF₃COOH	0.0003	0.0004	0.0005	0.0006	0.0010
TS _{acid} 1	<0.0001	<0.0001	<0.0001	<0.0001	0.0001
TS _{acid} 2	0.0002	0.0004	0.0005	0.0006	0.0010

S2.8 SO₂ reactions with sClIs 4 & 5

Table S40: SOZ decomposition branching ratios (Γ) of sCl 4 + SO₂ reaction with all identified pathways included.
 Grainsize = 50. The k_{d-d} ($4.42 \times 10^{-10} \text{ cm}^3 \text{ molec}^{-1} \text{ s}^{-1}$) used as pre-exponential factor for cyclo-reversion pathway.

Product branching ratio (r)	Temperature (K)				
	200	275	298	325	400
sCl 5 + SO ₂	<0.0001	0.0010	0.0017	0.0029	0.0107
TS _{EXO}	<0.0001	0.0002	0.0003	0.0005	0.0020
TS _{ENDO}	<0.0001	0.0009	0.0014	0.0024	0.0087
SO ₃ + CF ₃ CHO	1.0000	0.9913	0.9892	0.9861	0.9716
TS _{SO₃} 1	<0.0001	0.1025	0.1052	0.1082	0.1147
TS _{SO₃} 2	1.0000	0.4995	0.4984	0.4972	0.4936
TS _{SO₃} 3	<0.0001	0.3892	0.3856	0.3807	0.3633
SO ₂ + CF ₃ OCHO	<0.0001	0.0077	0.0091	0.0110	0.0177
TS _{ester} 1	<0.0001	0.0021	0.0024	0.0029	0.0045
TS _{ester} 2	<0.0001	0.0056	0.0067	0.0081	0.0132

Table S41: SOZ decomposition branching ratios (Γ) of sCl 4 + SO₂ reaction without cyclo-revision to Cl 5 + SO₂ revision.
 Grainsize = 50

Product branching ratio (r)	Temperature (K)				
	200	275	298	325	400
SO ₃ + CF ₃ CHO	1.0000	0.9923	0.9909	0.9889	0.9818
TS _{SO₃} 1	0.0005	0.1027	0.1056	0.1087	0.1169
TS _{SO₃} 2	0.9973	0.4998	0.4986	0.4977	0.4956
TS _{SO₃} 3	0.0022	0.3897	0.3867	0.3825	0.3693
SO ₂ + CF ₃ OCHO	<0.0001	0.0077	0.0091	0.0111	0.0182
TS _{ester} 1	<0.0001	0.0021	0.0024	0.0029	0.0046
TS _{ester} 2	<0.0001	0.0057	0.0067	0.0082	0.0136

Table S42: SOZ decomposition branching ratios (Γ) of sCl 4 + SO₂ reaction without Cl 5 + SO₂ cyclo-revision pathways and with only pathways that lead directly to SOZs included. Grainsize = 30

Product branching ratio (r)	Temperature (K)				
	200	275	298	325	400
SO ₃ + CF ₃ CHO	1.000	1.000	1.000	1.000	1.000
TS _{SO₃} 1	0.0057	0.0886	0.0924	0.0965	0.1071
TS _{SO₃} 2	0.9643	0.5095	0.5062	0.5039	0.5014
TS _{SO₃} 3	0.0300	0.4019	0.4014	0.3996	0.3915

Table S43: SOZ decomposition branching ratios (Γ) of sCl 5 + SO₂ reaction with all identified pathways included.
 Grainsize = 50. The k_{d-d} ($4.51 \times 10^{-10} \text{ cm}^3 \text{ molec}^{-1} \text{ s}^{-1}$) used as pre-exponential factor for cyclo-reversion pathway.

Product branching ratio (Γ)	Temperature (K)				
	200	275	298	325	400
sCl 4 + SO ₂	<0.0001	<0.0001	<0.0001	<0.0001	0.0001
TS _{EXO}	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001
TS _{ENDO}	<0.0001	<0.0001	<0.0001	<0.0001	0.0001
SO ₃ + CF ₃ CHO	0.9998	0.9963	0.9955	0.9943	0.9895
TS _{SO₃} 1	0.0076	0.0881	0.0917	0.0957	0.1053
TS _{SO₃} 2	0.9523	0.5083	0.5049	0.5023	0.4987
TS _{SO₃} 3	0.0399	0.3999	0.3988	0.3963	0.3856
SO ₂ + CF ₃ OCHO	0.0002	0.0037	0.0045	0.0057	0.0103
TS _{ester} 1	0.0001	0.0011	0.0013	0.0016	0.0027
TS _{ester} 2	0.0001	0.0026	0.0032	0.0041	0.0077

Table S44: SOZ decomposition branching ratios (Γ) of sCl 5 + SO₂ reaction without cyclo-revision to Cl 4 + SO₂ revision.
 Grainsize = 50

Product branching ratio (Γ)	Temperature (K)				
	200	275	298	325	400
SO ₃ + CF ₃ CHO	0.9981	0.9963	0.9955	0.9943	0.9897
TS _{SO₃} 1	0.0753	0.0881	0.0917	0.0957	0.1053
TS _{SO₃} 2	0.5279	0.5084	0.5049	0.5023	0.4987
TS _{SO₃} 3	0.3950	0.3998	0.3988	0.3964	0.3856
SO ₂ + CF ₃ OCHO	0.0019	0.0037	0.0045	0.0057	0.0103
TS _{ester} 1	0.0007	0.0011	0.0013	0.0016	0.0027
TS _{ester} 2	0.0012	0.0026	0.0032	0.0041	0.0076

Table S45: SOZ decomposition branching ratios (Γ) of sCl 5 + SO₂ reaction without Cl 4 pathways and with only pathways that lead directly to SOZs included. Grainsize = 30

Product branching ratio (Γ)	Temperature (K)				
	200	275	298	325	400
SO ₃ + CF ₃ CHO	1.000	1.000	1.000	1.000	1.000
TS _{SO₃} 1	0.0042	0.0886	0.0924	0.0965	0.1071
TS _{SO₃} 2	0.9740	0.5096	0.5063	0.5039	0.5014
TS _{SO₃} 3	0.0219	0.4018	0.4014	0.3996	0.3915

Table S46: Pathway Distribution (Γ) of **sClIs 1-5** reactions with H_2S , H_2O and MeOH.

sCl	T (K)	H_2S		H_2O		MeOH	
		Γ_{TS1}	Γ_{TS2}	Γ_{TS1}	Γ_{TS2}	Γ_{TS1}	Γ_{TS2}
1	200	0.551	0.449	0.874	0.126	0.899	0.101
	275	0.539	0.461	0.807	0.193	0.823	0.177
	298.15	0.537	0.463	0.790	0.210	0.802	0.198
	325	0.534	0.466	0.773	0.227	0.780	0.220
	400	0.529	0.471	0.732	0.268	0.728	0.272
2	200	0.743	0.257	0.270	0.730	0.973	0.027
	275	0.677	0.323	0.311	0.689	0.935	0.065
	298.15	0.662	0.338	0.320	0.680	0.923	0.077
	325	0.647	0.353	0.329	0.671	0.908	0.092
	400	0.614	0.386	0.348	0.652	0.871	0.129
3	200	0.689	0.311	0.843	0.157	0.900	0.100
	275	0.644	0.356	0.789	0.211	0.821	0.179
	298.15	0.634	0.366	0.777	0.223	0.799	0.201
	325	0.624	0.376	0.763	0.237	0.776	0.224
	400	0.604	0.396	0.734	0.266	0.721	0.279
4	200	0.734	0.266	0.438	0.562	N/A	N/A
	275	0.710	0.290	0.424	0.576	N/A	N/A
	298.15	0.700	0.300	0.422	0.578	N/A	N/A
	325	0.688	0.312	0.420	0.580	N/A	N/A
	400	0.658	0.342	0.417	0.583	N/A	N/A
5	200	0.604	0.396	0.325	0.675	0.472	0.528
	275	0.581	0.419	0.383	0.617	0.465	0.535
	298.15	0.576	0.424	0.396	0.604	0.464	0.536
	325	0.571	0.429	0.408	0.592	0.462	0.538
	400	0.561	0.439	0.435	0.565	0.459	0.541

Table S47: Branching Ratios (Γ) for $(H_2O)_2$ reaction with *sCIs 1 – 5*

sCl	T (K)	Γ_{TS1}	Γ_{TS2}	Γ_{TS3}	Γ_{TS4}
1	200	0.273	0.171	0.275	0.281
	275	0.279	0.143	0.283	0.295
	298.15	0.281	0.137	0.284	0.298
	325	0.282	0.133	0.286	0.300
	400	0.284	0.126	0.286	0.304
2	200	0.273	0.154	0.314	0.259
	275	0.289	0.122	0.315	0.274
	298.15	0.292	0.117	0.314	0.277
	325	0.296	0.113	0.311	0.280
	400	0.302	0.111	0.303	0.284
3	200	0.161	0.316	0.249	0.274
	275	0.144	0.333	0.246	0.277
	298.15	0.140	0.336	0.245	0.278
	325	0.138	0.339	0.245	0.279
	400	0.134	0.341	0.245	0.280
4	200	0.241	0.259	0.244	0.255
	275	0.228	0.273	0.235	0.264
	298.15	0.225	0.277	0.232	0.266
	325	0.220	0.282	0.229	0.269
	400	0.211	0.292	0.222	0.275
5	200	0.254	0.287	0.243	0.216
	275	0.256	0.324	0.233	0.188
	298.15	0.256	0.333	0.230	0.181
	325	0.256	0.344	0.227	0.174
	400	0.256	0.366	0.219	0.159

S3. Collision Frequency Rate-Limiting Coefficients

S3.1. Equations

Often, when k_{ME} values $\geq 10^{-10} \text{ cm}^3 \text{ molec.}^{-1} \text{ s}^{-1}$ each collision leads to a reaction which means the main limit to reaction is the number of collisions. One way to model the collision frequency is using the collision-limited reaction rate coefficient (k_{COLL}), which is modelled using the assumption that molecules have no attractive or repulsive forces and the only interactions between molecules are the momentum from elastic collisions. Another model is the dipole–dipole capture limit (k_{d-d}), which incorporates the changes to collision frequency caused by significant permanent dipoles within the reactants. These approaches have been used in a variety of different studies that examine bimolecular reactions involving Criegee intermediates.⁴⁻⁶

Collision Limit Rate Equation (k_{COLL}):

$$k_{COLL} = \left(\frac{8k_B T}{\pi \mu} \right)^{1/2} \pi (r_1 + r_2)^2 \quad \text{Equation S5}$$

$$\mu = \frac{m_1 m_2}{m_1 + m_2} \quad \text{Equation S6}$$

k_B is the Boltzmann constant, T is the temperature, μ are the reduced mass of reactants, r_1 and r_2 are the covalent radii of reactants and m_1 and m_2 are the masses.

Dipole-Dipole Capture Limit (k_{d-d}):

$$k_{d-d} = C \sqrt{\pi/\mu} (\mu_{D1} \mu_{D2})^{(2/3)} (k_B T)^{-1/6} \quad \text{Equation S7}$$

μ_{D1} and μ_{D2} , the dipole moments of reactants, μ , reduced mass, k_B , the Boltzmann constant, T , temperature, and, C , isotropic, adiabatic or non-adiabatic anisotropic constant which depends on the anisotropy of the capture potential model.

S3.2. Properties of Criegee intermediates 1-5 and co-reactants

Table S48: Reactants Dipole moments (μ_D , Debye), mass (m , amu), radii (r , Å) and C=O-O bond ratios

Reactant	Dipole Moment	Mass	radii	C=O-O Bond Ratio
	(μ_D , Debye)	(m , amu)	(r , Å)	
HCHO	4.3104	46.005	3.209	1.078
Syn-CF ₃ CHOO	3.4822	113.993	4.082	1.064
Anti-CF ₃ CHOO	2.5055	113.993	4.418	1.071
Syn-CF ₃ CFOO	3.016	131.983	4.099	1.114
Anti-CF ₃ CFOO	2.9299	131.983	4.447	1.101
HCHO	2.3887	30.011	2.017	N/A
HNO ₃	2.2588	62.996	2.938	N/A
SO ₂	1.8065	63.962	2.491	N/A
CF ₃ COOH	2.2289	113.993	4.237	N/A
HCl	1.1157	35.977	1.284	N/A
H ₂ O	1.8472	18.011	1.536	N/A
(H ₂ O) ₂	2.631	36.021	3.911	N/A
H ₂ S	0.9902	33.988	1.944	N/A
HF	1.8124	20.006	0.924	N/A
MeOH	1.6556	32.026	2.829	N/A

S3.3 Gas-collision limit

Table S49: Gas-Collision Limit for Criegee intermediates reactions in this study.

Reaction	Collision Limit ($10^{-10} \text{ cm}^3 \text{ molec.}^{-1} \text{ s}^{-1}$)				
T (K)	200	275	298.15	325	400
sCl 1 + HCHO	1.04	1.21	1.26	1.32	1.46
sCl 1 + HNO ₃	1.18	1.39	1.45	1.51	1.67
sCl 1 + SO ₂	1.02	1.19	1.24	1.29	1.44
sCl 1 + CF ₃ COOH	1.57	1.84	1.91	2.00	2.21
sCl 1 + HCl	0.73	0.85	0.89	0.93	1.03
sCl 1 + H ₂ O	1.01	1.19	1.23	1.29	1.43
sCl 1 + (H ₂ O) ₂	1.82	2.14	2.23	2.32	2.58
sCl 1 + H ₂ S	0.97	1.14	1.19	1.24	1.37
sCl 1 + HF	0.74	0.87	0.90	0.94	1.05
sCl 1 + MeOH	1.36	1.59	1.66	1.73	1.92
sCl 1 + CF ₃ CHO	1.24	1.46	1.52	1.58	1.76
sCl 1 + CF ₃ CFO	1.27	1.49	1.55	1.62	1.80
sCl 2 + HCHO	1.23	1.45	1.51	1.57	1.74
sCl 2 + HNO ₃	1.25	1.47	1.53	1.59	1.77
sCl 2 + SO ₂	1.09	1.28	1.33	1.39	1.54
sCl 2 + CF ₃ COOH	1.48	1.74	1.81	1.89	2.10
sCl 2 + HCl	0.89	1.04	1.09	1.13	1.26
sCl 2 + H ₂ O	1.29	1.52	1.58	1.65	1.83
sCl 2 + (H ₂ O) ₂	1.97	2.31	2.41	2.52	2.79
sCl 2 + H ₂ S	1.15	1.34	1.40	1.46	1.62
sCl 2 + HF	0.98	1.15	1.20	1.25	1.39
sCl 2 + MeOH	1.54	1.81	1.88	1.97	2.18
sCl 3 + HCHO	1.37	1.61	1.68	1.75	1.94
sCl 3 + HNO ₃	1.37	1.61	1.68	1.75	1.94
sCl 3 + SO ₂	1.21	1.41	1.47	1.54	1.70
sCl 3 + CF ₃ COOH	1.60	1.88	1.96	2.04	2.27
sCl 3 + HCl	1.00	1.18	1.23	1.28	1.42
sCl 3 + H ₂ O	1.45	1.70	1.77	1.85	2.05
sCl 3 + (H ₂ O) ₂	2.14	2.51	2.62	2.73	3.03
sCl 3 + H ₂ S	1.28	1.50	1.56	1.63	1.81
sCl 3 + HF	1.12	1.31	1.36	1.43	1.58
sCl 3 + MeOH	1.70	1.99	2.07	2.16	2.40
sCl 4 + HCHO	1.22	1.43	1.49	1.56	1.73
sCl 4 + HNO ₃	1.23	1.44	1.50	1.56	1.73
sCl 4 + SO ₂	1.07	1.25	1.31	1.36	1.51
sCl 4 + CF ₃ COOH	1.44	1.68	1.75	1.83	2.03
sCl 4 + HCl	0.88	1.03	1.08	1.12	1.25
sCl 4 + H ₂ O	1.29	1.51	1.57	1.64	1.82
sCl 4 + (H ₂ O) ₂	1.95	2.29	2.38	2.48	2.76
sCl 4 + H ₂ S	1.14	1.33	1.39	1.45	1.61
sCl 4 + HF	0.98	1.15	1.19	1.25	1.38
sCl 4 + MeOH	1.53	1.79	1.87	1.95	2.16
sCl 5 + HCHO	1.37	1.60	1.67	1.74	1.93
sCl 5 + HNO ₃	1.35	1.58	1.65	1.72	1.91
sCl 5 + SO ₂	1.19	1.39	1.45	1.51	1.68
sCl 5 + CF ₃ COOH	1.56	1.83	1.90	1.99	2.20
sCl 5 + HCl	1.00	1.17	1.22	1.27	1.41
sCl 5 + H ₂ O	1.45	1.70	1.77	1.85	2.06
sCl 5 + (H ₂ O) ₂	2.12	2.49	2.59	2.71	3.00
sCl 5 + H ₂ S	1.27	1.49	1.55	1.62	1.80
sCl 5 + HF	1.12	1.31	1.37	1.43	1.58
sCl 5 + MeOH	1.69	1.98	2.06	2.15	2.38

S3.4 Isotropic dipole-dipole capture moment

Table S50: Isotropic Dipole-Dipole Capture Limit for Criegee intermediates reactions in this study

Reaction	Isotropic Dipole-Dipole Capture Limit ($10^{-10} \text{ cm}^3 \text{ molec.}^{-1} \text{ s}^{-1}$)				
T (K)	200	275	298.15	325	400
sCl 1 + HCHO	11.34	10.75	10.61	10.45	10.10
sCl 1 + HNO ₃	9.03	8.56	8.45	8.32	8.04
sCl 1 + SO ₂	7.75	7.35	7.25	7.15	6.91
sCl 1 + CF ₃ COOH	8.06	7.64	7.54	7.43	7.18
sCl 1 + HCl	6.47	6.14	6.06	5.97	5.77
sCl 1 + H ₂ O	11.31	10.73	10.58	10.43	10.08
sCl 1 + (H ₂ O) ₂	11.46	10.87	10.73	10.57	10.21
sCl 1 + H ₂ S	6.07	5.76	5.68	5.60	5.41
sCl 1 + HF	10.76	10.21	10.07	9.93	9.59
sCl 1 + MeOH	8.71	8.26	8.15	8.03	7.76
sCl 1 + CF ₃ CHO	6.95	6.59	6.50	6.41	6.19
sCl 1 + CF ₃ CFO	3.57	3.38	3.34	3.29	3.18
sCl 2 + HCHO	8.60	8.15	8.04	7.93	7.66
sCl 2 + HNO ₃	6.34	6.01	5.93	5.85	5.65
sCl 2 + SO ₂	5.43	5.15	5.08	5.01	4.84
sCl 2 + CF ₃ COOH	5.30	5.03	4.96	4.89	4.72
sCl 2 + HCl	4.82	4.57	4.51	4.45	4.30
sCl 2 + H ₂ O	8.95	8.49	8.38	8.26	7.98
sCl 2 + (H ₂ O) ₂	8.54	8.10	7.99	7.88	7.61
sCl 2 + H ₂ S	4.55	4.32	4.26	4.20	4.06
sCl 2 + HF	8.45	8.01	7.91	7.79	7.53
sCl 2 + MeOH	6.56	6.22	6.14	6.05	5.85
sCl 3 + HCHO	6.90	6.55	6.46	6.37	6.15
sCl 3 + HNO ₃	5.09	4.83	4.76	4.69	4.53
sCl 3 + SO ₂	4.36	4.14	4.08	4.02	3.89
sCl 3 + CF ₃ COOH	4.26	4.04	3.98	3.93	3.79
sCl 3 + HCl	3.87	3.67	3.62	3.57	3.45
sCl 3 + H ₂ O	7.19	6.82	6.73	6.63	6.40
sCl 3 + (H ₂ O) ₂	6.86	6.50	6.42	6.33	6.11
sCl 3 + H ₂ S	3.66	3.47	3.42	3.37	3.26
sCl 3 + HF	6.79	6.43	6.35	6.26	6.04
sCl 3 + MeOH	5.27	5.00	4.93	4.86	4.70
sCl 4 + HCHO	7.70	7.30	7.20	7.10	6.86
sCl 4 + HNO ₃	5.62	5.33	5.26	5.18	5.00
sCl 4 + SO ₂	4.82	4.57	4.51	4.44	4.29
sCl 4 + CF ₃ COOH	4.65	4.41	4.35	4.29	4.14
sCl 4 + HCl	4.31	4.09	4.03	3.98	3.84
sCl 4 + H ₂ O	8.06	7.64	7.54	7.43	7.18
sCl 4 + (H ₂ O) ₂	7.63	7.24	7.14	7.04	6.80
sCl 4 + H ₂ S	4.07	3.86	3.81	3.76	3.63
sCl 4 + HF	7.60	7.21	7.11	7.01	6.77
sCl 4 + MeOH	5.87	5.57	5.50	5.42	5.23
sCl 5 + HCHO	7.55	7.16	7.07	6.97	6.73
sCl 5 + HNO ₃	5.51	5.23	5.16	5.08	4.91
sCl 5 + SO ₂	4.72	4.48	4.42	4.36	4.21
sCl 5 + CF ₃ COOH	4.56	4.32	4.27	4.21	4.06
sCl 5 + HCl	4.23	4.01	3.96	3.90	3.77
sCl 5 + H ₂ O	7.90	7.50	7.40	7.29	7.04
sCl 5 + (H ₂ O) ₂	7.49	7.10	7.01	6.91	6.67
sCl 5 + H ₂ S	3.99	3.79	3.74	3.68	3.56
sCl 5 + HF	7.45	7.07	6.97	6.87	6.64
sCl 5 + MeOH	5.76	5.46	5.39	5.31	5.13

S3.5 Adiabatic anisotropic dipole-dipole capture moment

Table S51: Adiabatic anisotropic Dipole-Dipole Capture Limit for Criegee intermediates reactions in this study

Reaction	Adiabatic anisotropic Dipole-Dipole Capture Limit ($10^{-10} \text{ cm}^3 \text{ molec.}^{-1} \text{ s}^{-1}$)				
T (K)	200	275	298.15	325	400
sCl 1 + HCHO	7.45	7.06	6.97	6.87	6.63
sCl 1 + HNO ₃	5.93	5.62	5.55	5.47	5.28
sCl 1 + SO ₂	5.09	4.83	4.76	4.70	4.54
sCl 1 + CF ₃ COOH	5.29	5.02	4.95	4.88	4.72
sCl 1 + HCl	4.25	4.03	3.98	3.92	3.79
sCl 1 + H ₂ O	7.43	7.05	6.95	6.85	6.62
sCl 1 + (H ₂ O) ₂	7.53	7.14	7.05	6.94	6.71
sCl 1 + H ₂ S	3.99	3.78	3.73	3.68	3.56
sCl 1 + HF	7.07	6.70	6.61	6.52	6.30
sCl 1 + MeOH	5.72	5.42	5.35	5.27	5.10
sCl 1 + CF ₃ CHO	4.57	4.33	4.27	4.21	4.07
sCl 1 + CF ₃ CFO	2.34	2.22	2.19	2.16	2.09
sCl 2 + HCHO	5.65	5.36	5.28	5.21	5.03
sCl 2 + HNO ₃	4.16	3.95	3.90	3.84	3.71
sCl 2 + SO ₂	3.57	3.39	3.34	3.29	3.18
sCl 2 + CF ₃ COOH	3.48	3.30	3.26	3.21	3.10
sCl 2 + HCl	3.17	3.00	2.96	2.92	2.82
sCl 2 + H ₂ O	5.88	5.58	5.50	5.42	5.24
sCl 2 + (H ₂ O) ₂	5.61	5.32	5.25	5.18	5.00
sCl 2 + H ₂ S	2.99	2.84	2.80	2.76	2.66
sCl 2 + HF	5.55	5.26	5.19	5.12	4.94
sCl 2 + MeOH	4.31	4.09	4.03	3.98	3.84
sCl 3 + HCHO	4.53	4.30	4.24	4.18	4.04
sCl 3 + HNO ₃	3.34	3.17	3.13	3.08	2.98
sCl 3 + SO ₂	2.87	2.72	2.68	2.64	2.55
sCl 3 + CF ₃ COOH	2.80	2.65	2.62	2.58	2.49
sCl 3 + HCl	2.54	2.41	2.38	2.35	2.27
sCl 3 + H ₂ O	4.72	4.48	4.42	4.35	4.21
sCl 3 + (H ₂ O) ₂	4.51	4.27	4.22	4.16	4.01
sCl 3 + H ₂ S	2.40	2.28	2.25	2.21	2.14
sCl 3 + HF	4.46	4.23	4.17	4.11	3.97
sCl 3 + MeOH	3.46	3.28	3.24	3.19	3.08
sCl 4 + HCHO	5.06	4.80	4.73	4.66	4.51
sCl 4 + HNO ₃	3.69	3.50	3.45	3.40	3.29
sCl 4 + SO ₂	3.16	3.00	2.96	2.92	2.82
sCl 4 + CF ₃ COOH	3.05	2.90	2.86	2.82	2.72
sCl 4 + HCl	2.83	2.69	2.65	2.61	2.52
sCl 4 + H ₂ O	5.29	5.02	4.95	4.88	4.72
sCl 4 + (H ₂ O) ₂	5.01	4.76	4.69	4.62	4.47
sCl 4 + H ₂ S	2.67	2.54	2.50	2.47	2.38
sCl 4 + HF	4.99	4.73	4.67	4.60	4.45
sCl 4 + MeOH	3.86	3.66	3.61	3.56	3.44
sCl 5 + HCHO	4.96	4.70	4.64	4.58	4.42
sCl 5 + HNO ₃	3.62	3.43	3.39	3.34	3.22
sCl 5 + SO ₂	3.10	2.94	2.90	2.86	2.76
sCl 5 + CF ₃ COOH	3.00	2.84	2.80	2.76	2.67
sCl 5 + HCl	2.78	2.63	2.60	2.56	2.47
sCl 5 + H ₂ O	5.19	4.92	4.86	4.79	4.63
sCl 5 + (H ₂ O) ₂	4.92	4.66	4.60	4.54	4.38
sCl 5 + H ₂ S	2.62	2.49	2.45	2.42	2.34
sCl 5 + HF	4.90	4.64	4.58	4.52	4.36
sCl 5 + MeOH	3.78	3.59	3.54	3.49	3.37

S3.6 Non-adiabatic anisotropic dipole-dipole capture moment

Table S52: Non-adiabatic anisotropic Dipole-Dipole Capture Limit for Criegee intermediate reactions in this study

Reaction	Non-adiabatic anisotropic Dipole-Dipole Capture Limit ($10^{-10} \text{ cm}^3 \text{ molec.}^{-1} \text{ s}^{-1}$)				
T (K)	200	275	298.15	325	400
sCl 1 + HCHO	5.43	5.15	5.08	5.00	4.83
sCl 1 + HNO ₃	4.32	4.10	4.04	3.98	3.85
sCl 1 + SO ₂	3.71	3.52	3.47	3.42	3.31
sCl 1 + CF ₃ COOH	3.86	3.66	3.61	3.56	3.44
sCl 1 + HCl	3.10	2.94	2.90	2.86	2.76
sCl 1 + H ₂ O	5.42	5.14	5.07	4.99	4.82
sCl 1 + (H ₂ O) ₂	5.49	5.20	5.13	5.06	4.89
sCl 1 + H ₂ S	2.91	2.76	2.72	2.68	2.59
sCl 1 + HF	5.15	4.89	4.82	4.75	4.59
sCl 1 + MeOH	4.17	3.95	3.90	3.84	3.71
sCl 1 + CF ₃ CHO	3.33	3.16	3.11	3.07	2.96
sCl 1 + CF ₃ CFO	1.71	1.62	1.60	1.57	1.52
sCl 2 + HCHO	4.12	3.90	3.85	3.80	3.67
sCl 2 + HNO ₃	3.03	2.88	2.84	2.80	2.70
sCl 2 + SO ₂	2.60	2.47	2.43	2.40	2.32
sCl 2 + CF ₃ COOH	2.54	2.41	2.37	2.34	2.26
sCl 2 + HCl	2.31	2.19	2.16	2.13	2.06
sCl 2 + H ₂ O	4.29	4.06	4.01	3.95	3.82
sCl 2 + (H ₂ O) ₂	4.09	3.88	3.83	3.77	3.64
sCl 2 + H ₂ S	2.18	2.07	2.04	2.01	1.94
sCl 2 + HF	4.04	3.84	3.78	3.73	3.60
sCl 2 + MeOH	3.14	2.98	2.94	2.90	2.80
sCl 3 + HCHO	3.30	3.13	3.09	3.05	2.94
sCl 3 + HNO ₃	2.44	2.31	2.28	2.25	2.17
sCl 3 + SO ₂	2.09	1.98	1.95	1.93	1.86
sCl 3 + CF ₃ COOH	2.04	1.93	1.91	1.88	1.81
sCl 3 + HCl	1.85	1.76	1.73	1.71	1.65
sCl 3 + H ₂ O	3.44	3.26	3.22	3.17	3.07
sCl 3 + (H ₂ O) ₂	3.28	3.11	3.07	3.03	2.93
sCl 3 + H ₂ S	1.75	1.66	1.64	1.61	1.56
sCl 3 + HF	3.25	3.08	3.04	3.00	2.89
sCl 3 + MeOH	2.52	2.39	2.36	2.33	2.25
sCl 4 + HCHO	3.69	3.50	3.45	3.40	3.28
sCl 4 + HNO ₃	2.69	2.55	2.52	2.48	2.40
sCl 4 + SO ₂	2.30	2.19	2.16	2.13	2.05
sCl 4 + CF ₃ COOH	2.23	2.11	2.08	2.05	1.98
sCl 4 + HCl	2.06	1.96	1.93	1.90	1.84
sCl 4 + H ₂ O	3.86	3.66	3.61	3.56	3.44
sCl 4 + (H ₂ O) ₂	3.65	3.47	3.42	3.37	3.26
sCl 4 + H ₂ S	1.95	1.85	1.82	1.80	1.74
sCl 4 + HF	3.64	3.45	3.40	3.35	3.24
sCl 4 + MeOH	2.81	2.67	2.63	2.59	2.51
sCl 5 + HCHO	3.62	3.43	3.38	3.33	3.22
sCl 5 + HNO ₃	2.64	2.50	2.47	2.43	2.35
sCl 5 + SO ₂	2.26	2.14	2.12	2.09	2.01
sCl 5 + CF ₃ COOH	2.18	2.07	2.04	2.01	1.94
sCl 5 + HCl	2.02	1.92	1.89	1.87	1.80
sCl 5 + H ₂ O	3.78	3.59	3.54	3.49	3.37
sCl 5 + (H ₂ O) ₂	3.58	3.40	3.35	3.31	3.19
sCl 5 + H ₂ S	1.91	1.81	1.79	1.76	1.70
sCl 5 + HF	3.57	3.38	3.34	3.29	3.18
sCl 5 + MeOH	2.76	2.62	2.58	2.54	2.46

S4. Effective Rate Constants (k_{EFF})

S4.1 The formula for the Effective Rate Constants (k_{EFF})

The main approach in this study to determine the atmospheric impact of these sCI + co-reactant reactions is the effective rate constant (k_{EFF}), which is seen in Equation S8.

$$k_{\text{EFF}} = k_{\text{ME}} [\text{co-reactant}]$$

Equation 8

By incorporating the local co-reactant abundance, [co-reactant], the k_{EFF} constant produces a single measure of sCI loss path contribution. This k_{EFF} constant can be used to compare the tropospheric impact that reaction with each trace gas has on the local sCI population. This approach has been adopted in various other sCI studies as a modest measure of the atmospheric impact of each reaction.^{6–8} The significant variety of local co-reactant abundances in many different environments and the range of effective rate constants produced from these values is covered in this section.

S4.2 Literature Abundances of Atmospheric Co-reactants Across Environments

Table S53: Representative abundances of atmospheric co-reactants across different environments in the literature.

Co-reactant Environment	Abundance (Original units)	Study	Ref
Formaldehyde (HCHO)			
Mean of European homes	23.1 $\mu\text{g}/\text{m}^3$	Salthammer <i>et al.</i>	9
Range of European homes	20–30 $\mu\text{g}/\text{m}^3$	Salthammer <i>et al.</i>	9
Mean of Rural European sites	3.49 ppb	Salthammer <i>et al.</i>	9
Range of Rural European sites	0.4–5.5 ppb	Salthammer <i>et al.</i>	9
Homes with particle floorboards	32.0 $\mu\text{g}/\text{m}^3$	Raw <i>et al.</i>	10
Homes without particle floorboards	20.3 $\mu\text{g}/\text{m}^3$	Raw <i>et al.</i>	10
Homes with particle floorboards in main bedroom England	37.7 $\mu\text{g}/\text{m}^3$	Raw <i>et al.</i>	10
Homes without particle floorboards in main bedroom England	23.2 $\mu\text{g}/\text{m}^3$	Raw <i>et al.</i>	10
New homes	128 & 135 $\mu\text{g}/\text{m}^3$	Raw <i>et al.</i>	10
Average home	22.2 $\mu\text{g}/\text{m}^3$	Raw <i>et al.</i>	10
Minimum	1 $\mu\text{g}/\text{m}^3$	Raw <i>et al.</i>	10
Maximum	171 $\mu\text{g}/\text{m}^3$	Raw <i>et al.</i>	10
In buildings in the absent of ozone	<30 $\mu\text{g}/\text{m}^3$	Uhde <i>et al.</i>	11
Highest found with ozone present	148 $\mu\text{g}/\text{m}^3$	Uhde <i>et al.</i>	11
General	0.5 $\mu\text{g}/\text{m}^3$	WHO Regional Publications	12
Urban general	1–20 $\mu\text{g}/\text{m}^3$	WHO Regional Publications	12
Heavy traffic	100 $\mu\text{g}/\text{m}^3$	WHO Regional Publications	12
Mobile Home	490 $\mu\text{g}/\text{m}^3$	WHO Regional Publications	12
São Paulo, Brazil	5.0±2.8 ppbv	Nguyen <i>et al.</i>	13
Osaka, Japan	1.9±0.9 ppbv	Nguyen <i>et al.</i>	13
Acetaldehyde (CH₃CHO)			
São Paulo, Brazil	5.4±2.8 ppbv	Nguyen <i>et al.</i>	13
Osaka, Japan	1.5±0.8 ppbv	Nguyen <i>et al.</i>	13

General Aldehydes and Ketones (from Vereecken <i>et al.</i>)			
R ₁ R ₂ C=O – Boreal Forest	1.2 × 10 ¹¹	Vereecken <i>et al.</i>	8
R ₁ R ₂ C=O – Tropical Forest	1.9 × 10 ¹⁰	Vereecken <i>et al.</i>	8
R ₁ R ₂ C=O – Mega city	5.1 × 10 ¹¹	Vereecken <i>et al.</i>	8
R ₁ R ₂ C=O – Rural Europe	6.6 × 10 ¹⁰	Vereecken <i>et al.</i>	8
Sulphur Dioxide (SO ₂)			
Beijing urban area, China	16.8 ± 13.1 ppb	Lin <i>et al.</i>	14
Gucheng rural area, China	14.8 ± 9.4 ppb	Lin <i>et al.</i>	14
Shangdianzi, China	7.5 ± 4.0 ppb	Lin <i>et al.</i>	14
London Bloomsbury, UK	2 µg/m ³	defra	15
London Marylebone Road, UK	7 µg/m ³	defra	15
London N. Kensington, UK	14 µg/m ³	defra	15
Grangemouth, Falkirk, UK	18 µg/m ³	defra	15
Londonderry Rosemount, NI, UK	18 µg/m ³	defra	15
0.1 km from Volcanic vents (Mount Etna)	~10,000 µg/m ³	Aiuppa <i>et al.</i>	16
10 km from Volcanic vents (Mount Etna)	7 µg/m ³	Aiuppa <i>et al.</i>	16
Sao Paulo citywide concentrations	4.94+2.99 ppb	Bravo <i>et al.</i>	17
Typical indoor	22.9 (<50) ppb	Salthammer <i>et al.</i>	18
Typical urban concentrations	15 ppbv	Hunter <i>et al.</i>	19
SO ₂ – Boreal Forest	1.7 × 10 ¹⁰	Vereecken <i>et al.</i>	8
SO ₂ – Tropical Forest	~0	Vereecken <i>et al.</i>	8
SO ₂ – Mega city	9 × 10 ¹⁰	Vereecken <i>et al.</i>	8
SO ₂ – Rural Europe	6.6 × 10 ⁹	Vereecken <i>et al.</i>	8
Nitric Acid (HNO ₃) 1.1 × 10 ¹¹			
Houston, TX	~4.5 ppb	Leong <i>et al.</i>	20
Los Angeles Southern California	20 ppbv	Foreman <i>et al.</i>	21
Southern California rural	0.1-8.0 ppb	Crisp <i>et al.</i>	22
Houston, TX	0.4-4.6 ppb	Leong <i>et al.</i>	20
HNO ₃	~10 ¹¹	Kumar and Fransisco	7
Trifluoroacetic acid (CF ₃ COOH)			
Beijing (04/14-04/15)	1459 ± 223 pg/m ³	Zhang <i>et al.</i>	23
Beijing (04/14-04/15)	1162 ± 173 pg/m ³	Zhang <i>et al.</i>	23
lowest abundance of local TFA	283 pg/m ³	Zhang <i>et al.</i>	23
highest abundance of local TFA	6317 pg/m ³	Zhang <i>et al.</i>	23
Average Beijing (05/12-04/13)	1580 ± 558 pg/m ³	Wu <i>et al.</i>	24
Europe projected future abundance	0.06 — 0.94 ppt	Henne <i>et al.</i>	25
Carboxylic acid (RCOOH)			
RCOOH – Boreal Forest	1 × 10 ¹¹	Vereecken <i>et al.</i>	8
RCOOH – Tropical Forest	5 × 10 ¹⁰	Vereecken <i>et al.</i>	8
RCOOH – Mega city	N/A	Vereecken <i>et al.</i>	8
RCOOH – Rural Europe	N/A	Vereecken <i>et al.</i>	8
Hydrofluoric acid (HF)			
Stratosphere	0.1-0.5 ppb	Mankin <i>et al.</i>	26
vicinity of volcano by ambient T (K)	up to 15 ppbv	Cheng <i>et al.</i>	27
At source of volcano	900 ug m ⁻³	Cheng <i>et al.</i>	27

Hydrochloric acid (HCl)			
Southern California (overall)	1.3 ppb	Crisp <i>et al.</i>	22
Southern California (halfway range)	8 ppb	Crisp <i>et al.</i>	22
Southern California (Mean)	2.2 ± 2.3 ppb	Crisp <i>et al.</i>	22
Southern California (lowest)	~0 ppb	Crisp <i>et al.</i>	22
Southern California (highest)	16 ppb	Crisp <i>et al.</i>	22
Los Angeles (Daytime)	5 ppb	Crisp <i>et al.</i>	22
Los Angeles (Nighttime)	3 ppb	Crisp <i>et al.</i>	22
San Francisco Bay	3.9 ppb	Crisp <i>et al.</i>	22
Standard tropospheric range	100 - 300 pptv	Sanhueza <i>et al.</i>	28
Industrial or marine environments	1.5 - 3 ppb	Graedel and Keene	29
Hydrogen sulphide (H₂S)			
Volcanoes	500 ppb	Kumar <i>et al.</i>	30
construction and demolition waste emissions - average	7 — 100 ppm	Kumar <i>et al.</i>	30
construction and demolition waste emissions- highest end	5,000–12,000 ppm	Kumar <i>et al.</i>	30
ambient air range	0.11–0.33 ppb	Abdollahi and Hosseini	31
industrial landfill, Terre Haute, Indiana, USA	2.0 – 300 ppb	Levels <i>et al.</i>	32
Dakota City 1999	≥ 90 ppb	Inserra <i>et al.</i>	33
Methanol (CH₃OH)			
Troposphere over Remote oceans	1 ppbv	Heikes <i>et al.</i>	34
Average Urban areas and forest	20 ppbv	Heikes <i>et al.</i>	34
São Paulo, Brazil	34.1±9.2 ppb	Nguyen <i>et al.</i>	13
Osaka, Japan	5.8±3.8	Nguyen <i>et al.</i>	13
wooded North Carolina industrial area	78.5—297 ppbv	Kelly <i>et al.</i>	34,35

S4.3 Effective Rate Constants Used in Manuscript

Table S54: Co-reactant and their chosen atmospheric concentration ([co-reactant]); Criegee Intermediate number [sCI]; computational master equation rate constant (k_{ME})*; dipole-dipole capture limit (k_{d-d}); literature experimental rate constant (k_{EXP}); and effective rate constants (k_{EFF} = rate constant \times [co-reactant]).

[Co-reactant] (molec./cm ³)	sCI	k_{ME} (cm ³ s ⁻¹)	k_{d-d} (cm ³ s ⁻¹)	k_{EXP} (cm ³ s ⁻¹)	k_{EFF} (s ⁻¹)
HCHO	1	2.79 × 10⁻¹²	1.06×10^{-9}	-	5.60
	23	2.69 × 10⁻¹³	8.04×10^{-10}	-	0.54
	24	2.49 × 10⁻¹²	6.46×10^{-10}	-	5.0
(Urban Heavy traffic) ¹²	25	» k_{d-d}	7.20 × 10⁻¹⁰	-	1447
	26	1.66 × 10⁻¹¹	7.07×10^{-10}	-	33.4
CF ₃ CHO	1	» k_{d-d}	6.50×10^{-10}	-	-
CF ₃ CFO	1	1.98 × 10⁻¹²	3.34×10^{-10}	-	-
SO ₂	1	» k_{d-d}	7.25 × 10⁻¹⁰	3.80×10^{-11}	300
	23	1.90 × 10⁻¹²	5.08×10^{-10}	-	0.79
	24	» k_{d-d}	4.08 × 10⁻¹⁰	-	169
(Beijing urban area) ¹⁴	25	» k_{d-d}	4.51 × 10⁻¹⁰	-	186
	26	» k_{d-d}	4.42 × 10⁻¹⁰	-	183
HNO ₃	1	8.22×10^{-9}	8.45 × 10⁻¹⁰	5.1×10^{-10}	94
	23	2.06 × 10⁻¹¹	5.93×10^{-10}	-	2.3
	24	8.87 × 10⁻¹¹	4.76×10^{-10}	-	9.8
(Houston, Texas) ²⁰	25	9.25×10^{-8}	5.26 × 10⁻¹⁰	-	58
	26	7.53 × 10⁻¹¹	5.16×10^{-10}	-	8.4
TFA	1	» k_{d-d}	7.54 × 10⁻¹⁰	3.50×10^{-10}	2.5×10^{-2}
	23	» k_{d-d}	4.96 × 10⁻¹⁰	-	1.7×10^{-2}
	24	» k_{d-d}	3.98 × 10⁻¹⁰	-	1.3×10^{-2}
(Beijing higher emission areas)	25	» k_{d-d}	4.35 × 10⁻¹⁰	-	1.5×10^{-2}
	26	» k_{d-d}	4.27 × 10⁻¹⁰	-	1.4×10^{-2}
H ₂ O	1	1.18 × 10⁻¹⁶	1.06×10^{-9}	2.4×10^{-16}	73
	23	1.35 × 10⁻¹⁸	8.38×10^{-10}	-	0.83
	24	1.12 × 10⁻¹⁶	6.73×10^{-10}	-	70
(~80% humidity Sao Paulo) ^{17,36}	25	5.13 × 10⁻¹²	7.54×10^{-10}	-	3.2×10^6
	26	2.27 × 10⁻¹³	7.40×10^{-10}	-	1.4×10^5
(H ₂ O) ₂	1	3.28 × 10⁻¹²	1.07×10^{-9}	7.50×10^{-12}	2.9×10^3
	23	2.71 × 10⁻¹²	7.99×10^{-10}	-	2.4×10^3
	24	3.74 × 10⁻¹²	6.42×10^{-10}	-	3.3×10^3
(~80% humidity Sao Paulo) ^{17,36}	25	1.14×10^{-6}	7.14 × 10⁻¹⁰	-	6.2×10^5
	26	1.46×10^{-9}	7.01 × 10⁻¹⁰	-	6.1×10^5
MeOH	1	1.20 × 10⁻¹⁴	8.15×10^{-10}	1.1×10^{-13}	0.010
	23	9.43 × 10⁻¹⁷	6.14×10^{-10}	-	7.9×10^{-5}
	24	3.30 × 10⁻¹⁴	4.93×10^{-10}	-	0.028
(Sao Paulo, Brazil) ¹³	25	» k_{d-d}	5.50 × 10⁻¹⁰	-	462
	26	3.31 × 10⁻¹¹	5.39×10^{-10}	-	28
H ₂ S	1	7.06 × 10⁻¹⁵	5.68×10^{-10}	1.7×10^{-13}	0.052
	23	1.96 × 10⁻¹⁶	4.26×10^{-10}	-	1.4×10^{-3}
	24	5.72 × 10⁻¹⁵	3.42×10^{-10}	-	0.042
(Industrial landfill) ³²	25	6.08 × 10⁻¹³	3.81×10^{-10}	-	4.5
	26	1.23 × 10⁻¹⁴	3.74×10^{-10}	-	0.091
HCl	1	4.70 × 10⁻¹⁰	6.06×10^{-10}	4.1×10^{-11}	58
	23	4.04 × 10⁻¹⁵	4.51×10^{-10}	-	5.0×10^{-3}
	24	7.28 × 10⁻¹¹	3.62×10^{-10}	-	8.96
(Daytime Los Angeles) ²²	25	1.64 × 10⁻¹⁰	4.03×10^{-10}	-	20
	26	1.13 × 10⁻¹⁰	3.96×10^{-10}	-	14
HF	1	3.32 × 10⁻¹³	8.15×10^{-10}	-	0.12
	23	6.67 × 10⁻¹⁸	6.14×10^{-10}	-	2.5×10^{-6}
(near Volcano vicinity ambient T) ²⁷	24	2.86 × 10⁻¹⁵	4.93×10^{-10}	-	1.1×10^{-3}
	25	1.19 × 10⁻¹¹	5.50×10^{-10}	-	4.4
	26	3.66 × 10⁻¹³	5.39×10^{-10}	-	0.14

* "» k_{d-d} " refers to barrierless reactions that yield no k_{ME} and that the k_{d-d} should be used instead.

S4.4 Lower Range of Effective Rate Constants

Table S55: Co-reactant and their chosen atmospheric concentration ([co-reactant]); Criegee Intermediate number [sCI]; computational master equation rate constant (k_{ME})*; dipole-dipole capture limit (k_{d-d}); literature experimental rate constant (k_{EXP}); and effective rate constants ($k_{EFF} = k_{ME}$ or $k_{d-d} \times [co\text{-reactant}]$).

[Co-reactant] (molec./cm ³)	sCI	k_{ME} (cm ³ s ⁻¹)	k_{d-d} (cm ³ s ⁻¹)	k_{EXP} (cm ³ s ⁻¹)	k_{EFF} (s ⁻¹)
HCHO	1	2.79×10^{-12}	1.06×10^{-9}	-	1.3
	23	2.69×10^{-13}	8.04×10^{-10}	-	0.12
4.6×10^{11}	24	2.49×10^{-12}	6.46×10^{-10}	-	1.2
(European Home average) ⁹	25	» k_{d-d}	7.20×10^{-10}	-	334
	26	1.66×10^{-11}	7.07×10^{-10}	-	7.7
CF ₃ CHO	1	» k_{d-d}	6.50×10^{-10}	-	-
CF ₃ CFO	1	1.98×10^{-12}	3.34×10^{-10}	-	-
SO ₂	1	» k_{d-d}	7.25×10^{-10}	3.80×10^{-11}	88.3
	23	1.90×10^{-12}	5.08×10^{-10}	-	0.23
1.2×10^{11}	24	» k_{d-d}	4.08×10^{-10}	-	49.7
(Sao Paulo, Brazil) ¹⁷	25	» k_{d-d}	4.51×10^{-10}	-	54.8
	26	» k_{d-d}	4.42×10^{-10}	-	53.8
HNO ₃	1	8.22×10^{-9}	8.45×10^{-10}	5.1×10^{-10}	60
9.9×10^9	23	2.06×10^{-11}	5.93×10^{-10}	-	9.8
(Houston, Texas, lower boundary) ²⁰	24	8.87×10^{-11}	4.76×10^{-10}	-	53
	25	9.25×10^{-8}	5.26×10^{-10}	-	58
	26	7.53×10^{-11}	5.16×10^{-10}	-	57
TFA	1	» k_{d-d}	7.54×10^{-10}	3.50×10^{-10}	1.7×10^{-3}
	23	» k_{d-d}	4.96×10^{-10}	-	1.1×10^{-3}
2.3×10^6	24	» k_{d-d}	3.98×10^{-10}	-	9.2×10^{-4}
(Europe urban projected) ²⁵	25	» k_{d-d}	4.35×10^{-10}	-	1.0×10^{-3}
	26	» k_{d-d}	4.27×10^{-10}	-	9.9×10^{-4}
H ₂ O	1	1.18×10^{-16}	1.06×10^{-9}	2.4×10^{-16}	28
	23	1.35×10^{-18}	8.38×10^{-10}	-	0.32
2.4×10^{17}	24	1.12×10^{-16}	6.73×10^{-10}	-	27
(Mega city) ⁸	25	5.13×10^{-12}	7.54×10^{-10}	-	1.2×10^6
	26	2.27×10^{-13}	7.40×10^{-10}	-	5.44×10^4
(H ₂ O) ₂	1	3.28×10^{-12}	1.07×10^{-9}	7.50×10^{-12}	279
	23	2.71×10^{-12}	7.99×10^{-10}	-	231
8.5×10^{13}	24	3.74×10^{-12}	6.42×10^{-10}	-	318
(Mega city) ⁸	25	1.14×10^{-6}	7.14×10^{-10}	-	6.1×10^4
	26	1.46×10^{-9}	7.01×10^{-10}	-	6.0×10^4
MeOH	1	1.20×10^{-14}	8.15×10^{-10}	1.1×10^{-13}	1.7×10^{-3}
	23	9.43×10^{-17}	6.14×10^{-10}	-	1.3×10^{-5}
1.4×10^{11}	24	3.30×10^{-14}	4.93×10^{-10}	-	4.7×10^{-3}
(Osaka, Japan) ¹³	25	» k_{d-d}	5.50×10^{-10}	-	79
	26	3.31×10^{-11}	5.39×10^{-10}	-	4.7
H ₂ S	1	7.06×10^{-15}	5.68×10^{-10}	1.7×10^{-13}	5.7×10^{-5}
	23	1.96×10^{-16}	4.26×10^{-10}	-	1.6×10^{-6}
1.2×10^9	24	5.72×10^{-15}	3.42×10^{-10}	-	4.6×10^{-5}
(ambient air range) ³¹	25	6.08×10^{-13}	3.81×10^{-10}	-	4.9×10^{-3}
	26	1.23×10^{-14}	3.74×10^{-10}	-	1.0×10^{-4}
HCl	1	4.70×10^{-10}	6.06×10^{-10}	4.1×10^{-11}	3.5
	23	4.04×10^{-15}	4.51×10^{-10}	-	3.0×10^{-4}
7.4×10^9	24	7.28×10^{-11}	3.62×10^{-10}	-	0.54
(Standard troposphere) ²⁸	25	1.64×10^{-10}	4.03×10^{-10}	-	1.2
	26	1.13×10^{-10}	3.96×10^{-10}	-	0.83
HF	1	3.32×10^{-13}	8.15×10^{-10}	-	0.12
3.7×10^{11}	23	6.67×10^{-18}	6.14×10^{-10}	-	2.5×10^{-6}
(near Volcano vicinity)	24	2.86×10^{-15}	4.93×10^{-10}	-	1.1×10^{-3}
ambient T) ²⁷	25	1.19×10^{-11}	5.50×10^{-10}	-	4.4
	26	3.66×10^{-13}	5.39×10^{-10}	-	0.14

* "» k_{d-d} " refers to barrierless reactions that yield no k_{ME} and that the k_{d-d} should be used instead.

S4.5 Medium Range of Effective Rate Constants

Table S56: Co-reactant and their chosen atmospheric concentration ([co-reactant]); Criegee Intermediate number [sCI]; computational master equation rate constant (k_{ME})*; dipole-dipole capture limit (k_{d-d}); literature experimental rate constant (k_{EXP}); and effective rate constants ($k_{EFF} = k_{ME}$ or $k_{d-d} \times [co\text{-reactant}]$).

[Co-reactant] (molec./cm ³)	sCI	k_{ME} (cm ³ s ⁻¹)	k_{d-d} (cm ³ s ⁻¹)	k_{EXP} (cm ³ s ⁻¹)	k_{EFF} (s ⁻¹)
HCHO	1	2.79 × 10⁻¹²	1.06×10^{-9}	-	5.60
	23	2.69 × 10⁻¹³	8.04×10^{-10}	-	0.54
	2.0×10^{12}	2.49 × 10⁻¹²	6.46×10^{-10}	-	5.0
(Heavy traffic) ¹²	24	$\gg k_{d-d}$	7.20 × 10⁻¹⁰	-	1447
	25	$\gg k_{d-d}$	1.66 × 10⁻¹¹	7.07×10^{-10}	33.4
CF ₃ CHO	1	$\gg k_{d-d}$	6.50×10^{-10}	-	-
CF ₃ CFO	1	1.98 × 10⁻¹²	3.34×10^{-10}	-	-
SO ₂	1	$\gg k_{d-d}$	7.25 × 10⁻¹⁰	3.80×10^{-11}	300
	23	1.90 × 10⁻¹²	5.08×10^{-10}	-	0.79
	4.1×10^{11}	$\gg k_{d-d}$	4.08 × 10⁻¹⁰	-	169
(Beijing urban area) ¹⁴	24	$\gg k_{d-d}$	4.51 × 10⁻¹⁰	-	186
	25	$\gg k_{d-d}$	4.42 × 10⁻¹⁰	-	183
HNO ₃	1	8.22×10^{-9}	8.45 × 10⁻¹⁰	5.1×10^{-10}	60
	23	2.06 × 10⁻¹¹	5.93×10^{-10}	-	9.8
	1.1×10^{11}	8.87 × 10⁻¹¹	4.76×10^{-10}	-	53
(Houston, Texas) ²⁰	24	9.25×10^{-8}	5.26 × 10⁻¹⁰	-	58
	25	7.53 × 10⁻¹¹	5.16×10^{-10}	-	8.4
TFA	1	$\gg k_{d-d}$	7.54 × 10⁻¹⁰	3.50×10^{-10}	4.6×10^{-3}
	23	$\gg k_{d-d}$	4.96 × 10⁻¹⁰	-	3.0×10^{-3}
	6.1×10^6	$\gg k_{d-d}$	3.98 × 10⁻¹⁰	-	2.4×10^{-3}
(Beijing average emission areas) ²³	24	$\gg k_{d-d}$	4.35 × 10⁻¹⁰	-	2.7×10^{-3}
	25	$\gg k_{d-d}$	4.27 × 10⁻¹⁰	-	2.6×10^{-3}
H ₂ O	1	1.18 × 10⁻¹⁶	1.06×10^{-9}	2.4×10^{-16}	73
	6.2×10^{17}	1.35 × 10⁻¹⁸	8.38×10^{-10}	-	0.83
(~80% humidity, Sao Paulo) ^{17,36}	24	1.12 × 10⁻¹⁶	6.73×10^{-10}	-	70
	25	5.13 × 10⁻¹²	7.54×10^{-10}	-	3.2×10^6
	26	2.27 × 10⁻¹³	7.40×10^{-10}	-	1.4×10^5
(H ₂ O) ₂	1	3.28 × 10⁻¹²	1.07×10^{-9}	7.50×10^{-12}	2.9×10^3
	8.7×10^{14}	2.71 × 10⁻¹²	7.99×10^{-10}	-	2.4×10^3
(~80% humidity, Sao Paulo) ^{17,36}	24	3.74 × 10⁻¹²	6.42×10^{-10}	-	3.3×10^3
	25	1.14×10^{-6}	7.14 × 10⁻¹⁰	-	6.2×10^5
	26	1.46×10^{-9}	7.01 × 10⁻¹⁰	-	6.1×10^5
MeOH	1	1.20 × 10⁻¹⁴	8.15×10^{-10}	1.1×10^{-13}	0.010
	23	9.43 × 10⁻¹⁷	6.14×10^{-10}	-	7.9×10^{-5}
	8.4×10^{11}	3.30 × 10⁻¹⁴	4.93×10^{-10}	-	0.028
(Sao Paulo, Brazil) ¹³	24	$\gg k_{d-d}$	5.50 × 10⁻¹⁰	-	462
	25	3.31 × 10⁻¹¹	5.39×10^{-10}	-	28
H ₂ S	1	7.06 × 10⁻¹⁵	5.68×10^{-10}	1.7×10^{-13}	0.052
	23	1.96 × 10⁻¹⁶	4.26×10^{-10}	-	1.4×10^{-3}
	1.2×10^{13}	5.72 × 10⁻¹⁵	3.42×10^{-10}	-	0.042
(Industrial landfill) ³²	24	6.08 × 10⁻¹³	3.81×10^{-10}	-	4.5
	25	1.23 × 10⁻¹⁴	3.74×10^{-10}	-	0.091
HCl	1	4.70 × 10⁻¹⁰	6.06×10^{-10}	4.1×10^{-11}	19
	3.2×10^{10}	4.04 × 10⁻¹⁵	4.51×10^{-10}	-	1.3×10^{-4}
(Southern California Overall) ²²	24	7.28 × 10⁻¹¹	3.62×10^{-10}	-	0.24
	25	1.64 × 10⁻¹⁰	4.03×10^{-10}	-	0.53
	26	1.13 × 10⁻¹⁰	3.96×10^{-10}	-	0.36
HF	1	3.32 × 10⁻¹³	8.15×10^{-10}	-	9.0
	23	6.67 × 10⁻¹⁸	6.14×10^{-10}	-	1.8×10^{-4}
	2.7×10^{13}	2.86 × 10⁻¹⁵	4.93×10^{-10}	-	0.076
(At Volcanic source) ²⁷	24	1.19 × 10⁻¹¹	5.50×10^{-10}	-	320
	25	3.66 × 10⁻¹³	5.39×10^{-10}	-	9.9

* “ $\gg k_{d-d}$ ” refers to barrierless reactions that yield no k_{ME} and that the k_{d-d} should be used instead.

S4.6 Higher Range of Effective Rate Constants

Table S57: Co-reactant and their chosen atmospheric concentration ([co-reactant]); Criegee Intermediate number [sCI]; computational master equation rate constant (k_{ME})*; dipole-dipole capture limit (k_{d-d}); literature experimental rate constant (k_{EXP}); and effective rate constants ($k_{EFF} = k_{ME}$ or $k_{d-d} \times [co\text{-reactant}]$).

[Co-reactant] (molec./cm ³)	sCI	k_{ME} (cm ³ s ⁻¹)	k_{d-d} (cm ³ s ⁻¹)	k_{EXP} (cm ³ s ⁻¹)	k_{EFF} (s ⁻¹)
HCHO	1	2.79×10^{-12}	1.06×10^{-9}	-	27.4
	23	2.69×10^{-13}	8.04×10^{-10}	-	2.64
9.8×10^{12}	24	2.49×10^{-12}	6.46×10^{-10}	-	24.5
(Caravan/ mobile home) ¹²	25	» k_{d-d}	7.20×10^{-10}	-	7088
	26	1.66×10^{-11}	7.07×10^{-10}	-	164
CF ₃ CHO	1	» k_{d-d}	6.50×10^{-10}	-	-
CF ₃ CFO	1	1.98×10^{-12}	3.34×10^{-10}	-	-
SO ₂	1	» k_{d-d}	7.25×10^{-10}	3.80×10^{-11}	68217
	23	1.90×10^{-12}	5.08×10^{-10}	-	178
9.4×10^{13}	24	» k_{d-d}	4.08×10^{-10}	-	38398
(0.1 km from Mt. Etna) ¹⁶	25	» k_{d-d}	4.51×10^{-10}	-	42373
	26	» k_{d-d}	4.42×10^{-10}	-	41563
HNO ₃	1	8.22×10^{-9}	8.45×10^{-10}	5.1×10^{-10}	416
	23	2.06×10^{-11}	5.93×10^{-10}	-	10.1
4.9×10^{11}	24	8.87×10^{-11}	4.76×10^{-10}	-	43.7
(Southern California rural) ²²	25	9.25×10^{-8}	5.26×10^{-10}	-	259
	26	7.53×10^{-11}	5.16×10^{-10}	-	37.3
TFA	1	» k_{d-d}	7.54×10^{-10}	3.50×10^{-10}	2.5×10^{-2}
	23	» k_{d-d}	4.96×10^{-10}	-	1.7×10^{-2}
3.3×10^7	24	» k_{d-d}	3.98×10^{-10}	-	1.3×10^{-2}
(Beijing higher emission areas) ¹⁴	25	» k_{d-d}	4.35×10^{-10}	-	1.5×10^{-2}
	26	» k_{d-d}	4.27×10^{-10}	-	1.4×10^{-2}
H ₂ O	1	1.18×10^{-16}	1.06×10^{-9}	2.4×10^{-16}	91
7.7×10^{17}	23	1.35×10^{-18}	8.38×10^{-10}	-	1.0
(~100% Sao Paulo outside) ^{17,36}	24	1.12×10^{-16}	6.73×10^{-10}	-	87
	25	5.13×10^{-12}	7.54×10^{-10}	-	4.0×10^6
	26	2.27×10^{-13}	7.40×10^{-10}	-	1.8×10^5
(H ₂ O) ₂	1	3.28×10^{-12}	1.07×10^{-9}	7.50×10^{-12}	4.5×10^3
1.4×10^{15}	23	2.71×10^{-12}	7.99×10^{-10}	-	3.7×10^3
(~100% Sao Paulo outside) ^{17,36}	24	3.74×10^{-12}	6.42×10^{-10}	-	5.1×10^3
	25	1.14×10^{-6}	7.14×10^{-10}	-	9.7×10^5
	26	1.46×10^{-9}	7.01×10^{-10}	-	9.5×10^5
MeOH	1	1.20×10^{-14}	8.15×10^{-10}	1.1×10^{-13}	0.088
	23	9.43×10^{-17}	6.14×10^{-10}	-	6.9×10^{-4}
7.3×10^{12}	24	3.30×10^{-14}	4.93×10^{-10}	-	0.24
(North Carolina Industrial zone) ³⁵	25	» k_{d-d}	5.50×10^{-10}	-	4.0×10^3
	26	3.31×10^{-11}	5.39×10^{-10}	-	242
H ₂ S	1	7.06×10^{-15}	5.68×10^{-10}	1.7×10^{-13}	2100
	23	1.96×10^{-16}	4.26×10^{-10}	-	58
3.0×10^{17}	24	5.72×10^{-15}	3.42×10^{-10}	-	1700
(construction Waste- peak) ³⁰	25	6.08×10^{-13}	3.81×10^{-10}	-	1.8×10^5
	26	1.23×10^{-14}	3.74×10^{-10}	-	1500
HCl	1	4.70×10^{-10}	6.06×10^{-10}	4.1×10^{-11}	190
3.9×10^{11}	23	4.04×10^{-15}	4.51×10^{-10}	-	0.016
(Southern California Highest) ²²	24	7.28×10^{-11}	3.62×10^{-10}	-	29
	25	1.64×10^{-10}	4.03×10^{-10}	-	65
	26	1.13×10^{-10}	3.96×10^{-10}	-	44
HF	1	3.32×10^{-13}	8.15×10^{-10}	-	9.0
	23	6.67×10^{-18}	6.14×10^{-10}	-	1.8×10^{-4}
2.7×10^{13}	24	2.86×10^{-15}	4.93×10^{-10}	-	0.076
(At Volcanic source) ²⁷	25	1.19×10^{-11}	5.50×10^{-10}	-	320
	26	3.66×10^{-13}	5.39×10^{-10}	-	9.9

* » k_{d-d} refers to barrierless reactions that yield no k_{ME} and that the k_{d-d} should be used instead.

S4.7 Effective Rate Constants Using General Abundances in Forest/City/Rural Environments

Table S58: Co-reactant and their chosen atmospheric concentration ([co-reactant]); Criegee Intermediate number [sCI]; computational master equation rate constant (k_{ME})*; dipole-dipole capture limit (k_{d-d}); literature experimental rate constant (k_{EXP}); and effective rate constants ($k_{EFF} = k_{ME}$ or $k_{d-d} \times [co\text{-reactant}]$). Abundance obtained from study by Vereecken et al.⁸

[Co-reactant] (molec./cm ³)	sCI	k_{ME} (cm ³ s ⁻¹)	k_{d-d} (cm ³ s ⁻¹)	k_{EXP} (cm ³ s ⁻¹)	k_{EFF} (s ⁻¹)
H_2O 3.9×10^{17} (Boreal Forest)	1	1.18×10^{-16}	1.06×10^{-9}	2.4×10^{-16}	46
	23	1.35×10^{-18}	8.38×10^{-10}	-	0.53
	24	1.12×10^{-16}	6.73×10^{-10}	-	44
	25	5.13×10^{-12}	7.54×10^{-10}	-	2.0×10^6
	26	2.27×10^{-13}	7.40×10^{-10}	-	8.8×10^4
H_2O 6.1×10^{17} (Tropical Forest)	1	1.18×10^{-16}	1.06×10^{-9}	2.4×10^{-16}	72
	23	1.35×10^{-18}	8.38×10^{-10}	-	0.82
	24	1.12×10^{-16}	6.73×10^{-10}	-	69
	25	5.13×10^{-12}	7.54×10^{-10}	-	3.1×10^6
	26	2.27×10^{-13}	7.40×10^{-10}	-	1.4×10^5
H_2O 2.4×10^{17} (Mega city)	1	1.18×10^{-16}	1.06×10^{-9}	2.4×10^{-16}	28
	23	1.35×10^{-18}	8.38×10^{-10}	-	0.32
	24	1.12×10^{-16}	6.73×10^{-10}	-	27
	25	5.13×10^{-12}	7.54×10^{-10}	-	1.2×10^6
	26	2.27×10^{-13}	7.40×10^{-10}	-	5.44×10^4
H_2O 3.8×10^{17} (Rural Europe)	1	1.18×10^{-16}	1.06×10^{-9}	2.4×10^{-16}	45
	23	1.35×10^{-18}	8.38×10^{-10}	-	0.51
	24	1.12×10^{-16}	6.73×10^{-10}	-	43
	25	5.13×10^{-12}	7.54×10^{-10}	-	1.9×10^6
	26	2.27×10^{-13}	7.40×10^{-10}	-	8.6×10^4
$(\text{H}_2\text{O})_2$ 2.3×10^{14} (Boreal Forest)	1	3.28×10^{-12}	1.07×10^{-9}	7.50×10^{-12}	754
	23	2.71×10^{-12}	7.99×10^{-10}	-	624
	24	3.74×10^{-12}	6.42×10^{-10}	-	861
	25	1.14×10^{-6}	7.14×10^{-10}	-	1.6×10^5
	26	1.46×10^{-9}	7.01×10^{-10}	-	1.6×10^5
$(\text{H}_2\text{O})_2$ 5.5×10^{14} (Tropical Forest)	1	3.28×10^{-12}	1.07×10^{-9}	7.50×10^{-12}	1.8×10^3
	23	2.71×10^{-12}	7.99×10^{-10}	-	1.5×10^3
	24	3.74×10^{-12}	6.42×10^{-10}	-	2.1×10^3
	25	1.14×10^{-6}	7.14×10^{-10}	-	3.9×10^5
	26	1.46×10^{-9}	7.01×10^{-10}	-	3.9×10^5
$(\text{H}_2\text{O})_2$ 8.5×10^{13} (Mega city)	1	3.28×10^{-12}	1.07×10^{-9}	7.50×10^{-12}	279
	23	2.71×10^{-12}	7.99×10^{-10}	-	231
	24	3.74×10^{-12}	6.42×10^{-10}	-	318
	25	1.14×10^{-6}	7.14×10^{-10}	-	6.1×10^4
	26	1.46×10^{-9}	7.01×10^{-10}	-	6.0×10^4
$(\text{H}_2\text{O})_2$ 2.1×10^{14} (Rural Europe)	1	3.11×10^{-12}	1.07×10^{-9}	7.50×10^{-12}	688
	23	2.25×10^{-12}	7.99×10^{-10}	-	570
	24	6.64×10^{-10}	6.42×10^{-10}	-	786
	25	1.14×10^{-6}	7.14×10^{-10}	-	1.5×10^5
	26	1.46×10^{-9}	7.01×10^{-10}	-	1.5×10^5

* "» k_{d-d} " refers to barrierless reactions that yield no k_{ME} and that the k_{d-d} should be used instead.

Table S59: Co-reactant and their chosen atmospheric concentration ([co-reactant]); Criegee Intermediate number [sCI]; computational master equation rate constant (k_{ME})*; dipole-dipole capture limit (k_{d-d}); literature experimental rate constant (k_{EXP}); and effective rate constants ($k_{EFF} = k_{ME}$ or $k_{d-d} \times [\text{co-reactant}]$). Abundance obtained from study by Vereecken et al..⁸

[Co-reactant] (molec./cm ³)	sCI	k_{ME} (cm ³ s ⁻¹)	k_{d-d} (cm ³ s ⁻¹)	k_{EXP} (cm ³ s ⁻¹)	k_{EFF} (s ⁻¹)
SO₂ 1.7 × 10 ¹⁰ (Boreal Forest)	1	» k_{d-d}	7.25 × 10⁻¹⁰	3.80×10^{-11}	12
	23	1.90 × 10⁻¹²	5.08×10^{-10}	-	0.032
	24	» k_{d-d}	4.08 × 10⁻¹⁰	-	6.9
	25	» k_{d-d}	4.51 × 10⁻¹⁰	-	7.7
	26	» k_{d-d}	4.42 × 10⁻¹⁰	-	7.5
SO₂ 9.0 × 10 ¹⁰ (Mega city)	1	» k_{d-d}	7.25 × 10⁻¹⁰	3.80×10^{-11}	65
	23	1.90 × 10⁻¹²	5.08×10^{-10}	-	0.17
	24	» k_{d-d}	4.08 × 10⁻¹⁰	-	37
	25	» k_{d-d}	4.51 × 10⁻¹⁰	-	41
	26	» k_{d-d}	4.42 × 10⁻¹⁰	-	40
SO₂ 6.6 × 10 ⁹ (Rural Europe)	1	» k_{d-d}	7.25 × 10⁻¹⁰	3.80×10^{-11}	4.8
	23	1.90 × 10⁻¹²	5.08×10^{-10}	-	0.013
	24	» k_{d-d}	4.08 × 10⁻¹⁰	-	2.7
	25	» k_{d-d}	4.51 × 10⁻¹⁰	-	3.0
	26	» k_{d-d}	4.42 × 10⁻¹⁰	-	2.9
R₁R₂C=O 1.2 × 10 ¹¹ (R₁R₂CO: Boreal Forest)	1	2.79 × 10⁻¹²	1.06×10^{-9}	-	0.33
	23	2.69 × 10⁻¹³	8.04×10^{-10}	-	0.032
	24	2.49 × 10⁻¹²	6.46×10^{-10}	-	0.30
	25	» k_{d-d}	7.20 × 10⁻¹⁰	-	86
	26	1.66 × 10⁻¹¹	7.07×10^{-10}	-	2.0
R₁R₂C=O 1.9 × 10 ¹⁰ (R₁R₂CO: Tropical Forest)	1	2.79 × 10⁻¹²	1.06×10^{-9}	-	0.053
	23	2.69 × 10⁻¹³	8.04×10^{-10}	-	5.1×10^{-3}
	24	2.49 × 10⁻¹²	6.46×10^{-10}	-	0.047
	25	» k_{d-d}	7.20 × 10⁻¹⁰	-	14
	26	1.66 × 10⁻¹¹	7.07×10^{-10}	-	0.32
R₁R₂C=O 5.1 × 10 ¹¹ (R₁R₂CO: Mega city)	1	2.79 × 10⁻¹²	1.06×10^{-9}	-	1.4
	23	2.69 × 10⁻¹³	8.04×10^{-10}	-	0.14
	24	2.49 × 10⁻¹²	6.46×10^{-10}	-	1.3
	25	» k_{d-d}	7.20 × 10⁻¹⁰	-	367
	26	1.66 × 10⁻¹¹	7.07×10^{-10}	-	8.5
R₁R₂C=O 6.6 × 10 ¹⁰ (R₁R₂CO: Rural Europe)	1	2.79 × 10⁻¹²	1.06×10^{-9}	-	0.18
	23	2.69 × 10⁻¹³	8.04×10^{-10}	-	0.018
	24	2.49 × 10⁻¹²	6.46×10^{-10}	-	0.16
	25	» k_{d-d}	7.20 × 10⁻¹⁰	-	48
	26	1.66 × 10⁻¹¹	7.07×10^{-10}	-	1.1
RCOOH 1.0 × 10 ¹¹ (RCOOH: Boreal Forest)	1	» k_{d-d}	7.54 × 10⁻¹⁰	3.50×10^{-10}	75
	23	» k_{d-d}	4.96 × 10⁻¹⁰	-	50
	24	» k_{d-d}	3.98 × 10⁻¹⁰	-	40
	25	» k_{d-d}	4.35 × 10⁻¹⁰	-	43
	26	» k_{d-d}	4.27 × 10⁻¹⁰	-	43
RCOOH 5.0 × 10 ¹⁰ (RCOOH: Tropical Forest)	1	» k_{d-d}	7.54 × 10⁻¹⁰	3.50×10^{-10}	38
	23	» k_{d-d}	4.96 × 10⁻¹⁰	-	25
	24	» k_{d-d}	3.98 × 10⁻¹⁰	-	20
	25	» k_{d-d}	4.35 × 10⁻¹⁰	-	22
	26	» k_{d-d}	4.27 × 10⁻¹⁰	-	21

* "» k_{d-d} " refers to barrierless reactions that yield no k_{ME} and that the k_{d-d} should be used instead.

S5. Relative Energies, Enthalpies and Gibbs Free Energies

Table S60: **sCl 1 + HCHO** relative energies (ΔE), zero-point corrected energy (ΔZPE), enthalpies ($\Delta H_{298.15}$) and Gibbs free energies ($\Delta G_{298.15}$) [units=kJ mol⁻¹]

#sCl	Stationary Point	ΔE	ΔZPE	$\Delta H_{298.15}$	$\Delta G_{298.15}$
sCl 1 + HCHO Cycloaddition					
	PRC	-30.01	-20.697	-22.37	18.78
	TS _C	-31.24	-20.771	-25.26	22.54
	HOZ	-240.60	-214.398	-221.72	-167.46
HOZ to FAc pathway instant					
	HOZ	-240.60	-214.398	-221.72	-167.46
	TS _{FAc} 1	-31.27	-27.221	-32.64	18.22
	C _{FAc} 2	-499.21	-488.439	-487.37	-455.44
	HCHO + HCOOH con 2	-479.10	-473.302	-473.41	-473.31
HOZ to FAc pathway via HAE					
	HOZ	-240.60	-214.398	-221.72	-167.46
	TS _d	-55.03	-53.629	-60.17	-7.28
	HAE con 1	-552.41	-527.558	-532.15	-484.81
	TS _{iso}	-538.07	-515.748	-521.06	-473.82
	HAE con 2	-564.24	-537.972	-543.27	-493.49
	TS _{HAE}	-461.21	-450.820	-456.96	-405.34
	C _{FAc} 1	-534.87	-520.548	-521.48	-483.02
	HCHO + HCOOH con 1	-496.94	-490.251	-490.54	-490.25

Table S61: **sCl 2 & 3 + HCHO** relative energies (ΔE), zero-point corrected energy (ΔZPE), enthalpies ($\Delta H_{298.15}$) and Gibbs free energies ($\Delta G_{298.15}$) [units=kJ mol⁻¹]

#sCl	Stationary Point	ΔE	ΔZPE	$\Delta H_{298.15}$	$\Delta G_{298.15}$
sCl 2 + HCHO Cycloaddition					
	sCl 2+ HCHO PRC	-25.46	-20.762	-19.54	16.22
	sCl 2+ HCHO TS _C 1	-19.91	-11.620	-15.07	33.88
	HOZ 1	-269.20	-245.226	-251.81	-196.12
sCl 3 + HCHO Cycloaddition					
	sCl 3+ HCHO	-2.54	-3.384	-2.84	-5.13
	sCl 3+ HCHO PRC2	-33.92	-26.983	-27.12	11.38
	sCl 3+ HCHO TS _C 2	-30.86	-22.337	-25.61	21.15
	HOZ 2	-265.64	-241.572	-248.03	-192.96
sCl 1 + CF₃CHO Cycloaddition (barrierless to both HOZ 1 and HOZ 2)					
	sCl 1 + CF ₃ CHO	-25.58	-23.031	-23.88	-28.29
	HOZ 1	-269.20	-245.226	-251.81	-196.12
	HOZ 2	-265.64	-241.572	-248.03	-192.96
HOZ interconversion					
	HOZ 1	-269.20	-245.226	-251.81	-196.12
	TS _{HOZ}	-244.72	-222.160	-229.91	-172.06
	HOZ 2	-265.64	-241.572	-248.03	-192.96
HOZ to TFA pathway via HAE					
	HOZ 2	-69.21	-69.80	-75.28	-21.27
	TS _d 1	-563.17	-540.11	-543.46	-497.71

	HAE1	-563.17	-540.11	-543.46	-497.71
	TS _{iso} 1	-548.18	-527.55	-531.87	-484.43
	HAE2	-572.00	-547.90	-551.77	-504.07
	TS _{HAE} 2	-470.55	-462.94	-467.47	-419.42
	C _{TFA} 1	-546.45	-534.10	-534.02	-497.41
	HCHO + TFA con 1	-503.26	-497.58	-497.56	-499.29
	HOZ to TFA pathway instant				
	HOZ 1	-269.20	-245.226	-251.81	-196.12
	TS _{acid} 1	-51.64	-49.669	-53.97	-2.98
	C _{TFA} 2	-515.99	-506.127	-504.67	-470.12
	HCHO + TFA con 2	-490.02	-484.350	-484.35	-486.57
	HOZ to FAc pathway via HAE 1				
	HOZ 2	-265.64	-241.572	-248.03	-192.96
	TS _d 2	-78.75	-77.785	-83.39	-29.16
	HAE3	-583.13	-561.064	-564.90	-516.69
	TS _{iso} 2	-569.16	-549.560	-554.18	-505.61
	HAE5	-587.81	-565.054	-569.35	-518.55
	TS _{iso} 3	-559.56	-538.677	-544.96	-489.08
	HAE6	-577.33	-554.290	-558.81	-507.64
	TS _{HAE} 2	-490.71	-481.145	-486.71	-433.25
	C _{FAc} 1	-547.63	-534.425	-533.65	-501.21
	CF ₃ CHO + HCOOH con 1	-522.52	-513.282	-514.42	-518.54
	HOZ to FAc pathway via HAE 2				
	HOZ 1	-269.20	-245.226	-251.81	-196.12
	TS _d 3	-66.82	-66.612	-72.08	-18.28
	HAE4	-556.17	-534.737	-538.48	-488.78
	TS _{iso} 4	-547.52	-528.525	-533.16	-482.92
	HAE6	-577.33	-554.290	-558.81	-507.64
	TS _{HAE} 2	-490.71	-481.145	-486.71	-433.25
	C _{FAc} 1	-547.63	-534.425	-533.65	-501.21
	CF ₃ CHO + HCOOH con 1	-522.52	-513.282	-514.42	-518.54
	HOZ to FAc pathway instant 1				
	HOZ 2	-265.64	-241.572	-248.03	-192.96
	TS _{FAc} 2	-57.38	-54.01	-58.84	-6.25
	C _{FAc} 2	-519.30	-508.906	-506.27	-489.77
	CF ₃ CHO + HCOOH con 2	-504.68	-496.333	-497.29	-501.59
	HOZ to FAc pathway instant 2				
	HOZ 1	-269.20	-245.226	-251.81	-196.12
	TS _{FAc} 3	-64.58	-61.516	-66.49	-13.52
	C _{FAc} 1	-519.30	-508.906	-506.27	-489.77
	CF ₃ CHO + HCOOH con 2	-504.68	-496.333	-497.29	-501.59
	HOZ to Ester pathway 1				
	HOZ 1	-269.20	-245.226	-251.81	-196.12
	TSester 2	8.61	14.232	10.93	59.39
	Cester 1	-509.42	-501.720	-499.59	-471.59
	SO ₂ + CF ₃ OCHO	-491.39	-486.849	-487.23	-486.48
	HOZ to Ester pathway 2				
	HOZ 2	-265.64	-241.572	-248.03	-192.96
	TSester 1	-1.75	5.375	2.10	49.25
	Cester 1	-509.42	-501.720	-499.59	-471.59

Table S62: **sClis 4 & 5 + HCHO** relative energies (ΔE), zero-point corrected energy (ΔZPE), enthalpies ($\Delta H_{298.15}$) and Gibbs free energies ($\Delta G_{298.15}$) [units=kJ mol⁻¹]

#sCl	Stationary Point	ΔE	ΔZPE	$\Delta H_{298.15}$	$\Delta G_{298.15}$
4 & 5					
sCl 4 + HCHO Cycloaddition					
	sCl 4 + HCHO	0.00	0.00	0.00	0.00
	HOZ 1	-319.80	-296.57	-303.49	-246.27
sCl 5 + HCHO Cycloaddition					
	sCl 5 + HCHO	-13.71	-13.00	-12.94	-14.17
	sCl 5 + HCHO PRC2	-49.26	-42.03	-41.98	-2.73
	sCl 5 + HCHO TSc2	-50.58	-41.87	-45.21	3.44
	HOZ 2	-324.11	-300.82	-307.87	-250.07
sCl 1 + CF₃CFO Cycloaddition (barrierless) and interconversion					
	sCl 1 + CF₃CFO	-104.69	-99.204	-100.88	-104.39
	sCl 1 + CF₃CFO PRC 1	-142.02	-130.881	-132.56	-88.12
	sCl 1 + CF₃CFO TSc 3	-137.70	-125.889	-130.53	-78.24
	HOZ 1	-319.80	-296.57	-303.49	-246.27
	sCl 1 + CF₃CFO PRC 2	-140.32	-129.582	-130.88	-87.76
	sCl 1 + CF₃CFO TSc 4	-133.79	-121.601	-126.45	-73.87
	HOZ 2	-324.11	-300.82	-307.87	-250.07
HOZ interconversion					
	HOZ 1	-319.80	-296.57	-303.49	-246.27
	TS _{HOZ}	-306.38	-284.60	-292.77	-233.16
	HOZ 2	-324.11	-300.82	-307.87	-250.07
HOZ to FAc pathway via HAE					
	HOZ 1	-319.80	-296.57	-303.49	-246.27
	TS _d 3	-62.48	-64.96	-69.63	-17.20
	HAE4	-623.96	-603.09	-608.14	-554.75
	TS _{iso} 4	-620.09	-600.04	-607.14	-548.57
	HAE6	-633.12	-611.96	-616.94	-563.57
	TS _{HAE} 2	-562.48	-552.91	-558.75	-505.22
	C _{FAc} 1	-627.18	-611.53	-611.33	-578.03
	CF ₃ CFO + HCOOH con 1	-601.63	-589.46	-591.42	-594.65
HOZ to FAc pathway instant 1					
	HOZ 2	-324.11	-300.82	-307.87	-250.07
	TS _{FAc} 1	-125.35	-121.86	-127.32	-72.20
	C _{FAc} 2	-602.46	-589.07	-587.71	-562.25
	CF ₃ CFO + HCOOH con 2	-583.79	-572.51	-574.29	-577.70
HOZ to FAc pathway instant 2					
	HOZ 1	-319.80	-296.57	-303.49	-246.27
	TS _{FAc} 2	-107.18	-104.33	-109.25	-56.05
	C _{FAc} 3	-602.30	-588.97	-587.54	-563.31
HOZ to Ester pathway 1					
	HOZ 2	-324.11	-300.82	-307.87	-250.07
	TSester 1	-58.73	-51.48	-55.19	-5.01
	Cester 1	-572.76	-562.29	-561.57	-527.35
	SO ₂ + CF ₃ OCHO	-550.95	-544.10	-545.28	-543.71
HOZ to Ester pathway 2					
	HOZ 1	-319.80	-296.57	-303.49	-246.27
	TSester 1	-70.99	-62.91	-66.83	-16.40
	Cester 1	-572.76	-562.29	-561.57	-527.35

Table S63:sCl + SO₂ relative energies (ΔE), zero-point corrected energy (ΔZPE), enthalpies ($\Delta H_{298.15}$) and Gibbs free energies ($\Delta G_{298.15}$) [units=kJ mol⁻¹]

#sCl	Stationary Point	ΔE	ΔZPE	$\Delta H_{298.15}$	$\Delta G_{298.15}$
1					
SOZ interconversion					
	SOZ 1	-159.88	-143.95	-149.04	-94.15
	TS _{SOZ}	-148.79	-134.22	-140.76	-83.16
	SOZ 2	-164.58	-148.85	-154.01	-98.88
SOZ to C_{so₃} pathway 1					
	SOZ 1	-159.88	-143.95	-149.04	-94.15
	TS _{SO₃ 1}	-71.81	-63.96	-68.44	-15.15
	C _{so₃}	-364.76	-355.00	-355.80	-311.56
	HCHO + SO ₃	-319.41	-317.93	-317.75	-315.70
SOZ to C_{so₃} pathway 2					
	SOZ 2	-164.58	-148.85	-154.01	-98.88
	TS _{SO₃ 2}	-96.60	-89.19	-93.94	-39.79
	C _{so₃}	-364.76	-355.00	-355.80	-311.56
SOZ to C_{acid} pathway 1					
	SOZ 2	-164.58	-148.85	-154.01	-98.88
	TS _{acid 1}	-18.91	-20.42	-23.58	27.15
	C _{acid 2}	-496.42	-488.05	-485.39	-459.81
	SO ₂ + FA con 2	-479.10	-473.30	-473.41	-473.31
SOZ to C_{acid} pathway 2					
	SOZ 1	-164.58	-148.85	-154.01	-98.88
	TS _{acid 2}	-38.71	-39.84	-43.17	8.09
	C _{acid 2}	-496.42	-488.05	-485.39	-459.81
SOZ to C_{acid} pathway 3					
	SOZ 1	-159.88	-143.95	-149.04	-94.15
	TS _{acid 3}	-38.10	-42.54	-44.69	3.51
	C _{acid 1}	-523.27	-512.07	-511.27	-474.88
	SO ₂ + FA con 1	-447.56	-444.39	-448.56	-397.19
sCl 2 + SO₂ Cycloaddition 1					
2 & 3	PRC 1	-25.32	-21.41	-19.94	19.83
	TS _{cyc1}	-22.24	-17.56	-19.25	31.05
	SOZ 1	-167.41	-154.97	-158.68	-103.21
sCl 2 + SO₂ Cycloaddition 2					
	PRC 2	-22.93	-19.35	-17.58	19.99
	TS _{cyc 2}	-22.44	-18.26	-19.50	28.79
	SOZ 2	-182.52	-169.97	-173.88	-117.93
SOZ interconversion 1					
	SOZ 1	-167.41	-154.97	-158.68	-103.21
	TS _{SOZ 1}	-158.48	-147.23	-152.55	-93.32
	SOZ3	-175.58	-162.78	-166.61	-110.95
SOZ interconversion 2					
	SOZ 2	-182.52	-169.97	-173.88	-117.93
	TS _{SOZ 2}	-164.88	-153.08	-158.54	-99.10
	SOZ4	-168.13	-155.49	-159.08	-103.90
SOZ to C_{so₃} pathway 1					
	SOZ3	-175.58	-162.78	-166.61	-110.95
	TS _{SO₃ 1}	-102.70	-96.16	-99.88	-44.28
	C _{so₃} 1	-376.65	-368.24	-367.24	-328.61

	$\text{CF}_3\text{CHO} + \text{SO}_3$	-345.00	-340.98	-341.64	-343.99
SOZ to C_{so_3} pathway 2					
	SOZ 2	-182.52	-169.97	-173.88	-117.93
	TS_{SO_3} 2	-102.70	-96.16	-99.88	-44.28
	C_{so_3} 1	-376.65	-368.24	-367.24	-328.61
SOZ to C_{acid} pathway 1					
	SOZ 1	-167.41	-154.97	-158.68	-103.21
	TS_{acid} 1	-24.22	-28.27	-29.89	20.72
	C_{acid} 2	-503.48	-496.08	-492.63	-472.03
	$\text{SO}_2 + \text{TFA}$ con 2	-490.02	-484.35	-484.35	-486.57
SOZ to C_{acid} pathway 2					
	SOZ 2	-182.52	-169.97	-173.88	-117.93
	TS_{acid} 2	-51.12	-54.28	-56.40	-4.50
	C_{acid} 3	-504.21	-496.76	-493.37	-472.07
SOZ to C_{acid} pathway 3					
	SOZ 2	-182.52	-169.97	-173.88	-117.93
	TS_{acid} 3	-48.54	-54.58	-55.50	-7.46
	C_{acid} 3	-504.21	-496.76	-493.37	-472.07
SOZ to C_{acid} pathway 4					
	SOZ 1	-167.41	-154.97	-158.68	-103.21
	TS_{acid} 4	-27.66	-33.91	-34.58	12.63
	C_{acid} 1	-528.65	-520.00	-517.52	-488.76
	$\text{SO}_2 + \text{TFA}$ con 1	-503.26	-497.58	-497.56	-499.29
SOZ to C_{ester} pathway 1					
	SOZ3	-175.58	-162.78	-166.61	-110.95
	TS_{ester} 1	-11.45	-11.06	-11.57	35.51
	Cester 1	-509.50	-502.18	-499.90	-467.54
	$\text{SO}_2 + \text{CF}_3\text{OCFO}$	-491.39	-486.85	-487.23	-486.48
SOZ to C_{ester} pathway 2					
	SOZ4	-168.13	-155.49	-159.08	-103.90
	TS_{ester} 2	5.96	5.63	5.37	48.87
	Cester 1	-509.50	-502.18	-499.90	-467.54
4&5	SOZ interconversion 1				
	SOZ 1	-203.93	-193.02	-196.79	-140.53
	TS_{SOZ} 1	-196.71	-187.24	-192.42	-134.96
	SOZ3	-226.47	-215.04	-219.25	-161.44
	SOZ interconversion 2				
	SOZ 2	-215.60	-204.39	-208.40	-151.49
	TS_{SOZ} 2	-210.99	-200.86	-206.57	-145.35
	SOZ4	-211.13	-200.27	-203.99	-147.93
	SOZ to C_{so_3} pathway 1				
	SOZ 2	-215.60	-204.39	-208.40	-151.49
	TS_{SO_3} 1	-117.49	-112.23	-115.98	-60.87
	C_{so_3} 1	-443.42	-434.24	-432.59	-402.54
	$\text{CF}_3\text{CHO} + \text{SO}_3$	-424.10	-417.15	-418.64	-420.10
	SOZ to C_{so_3} pathway 2				
	SOZ3	-226.47	-215.04	-219.25	-161.44
	TS_{SO_3} 2	-157.94	-152.16	-156.41	-98.42
	C_{so_3} 2	-443.42	-434.24	-432.59	-402.54

SOZ to C _{so₃} pathway 3				
SOZ4	-211.13	-200.27	-203.99	-147.93
TS _{SO₃} 3	-134.70	-129.54	-133.57	-76.21
C _{so₃} 3	-443.42	-434.24	-432.59	-402.54
SOZ to C _{ester} pathway 1				
SOZ 2	-215.60	-204.39	-208.40	-151.49
TS _{ester} 1	-85.28	-83.91	-85.02	-35.78
Cester 1	-561.50	-553.32	-550.70	-531.54
SO ₂ + CF ₃ OCHO	-550.95	-544.10	-545.28	-543.71
SOZ to C _{ester} pathway 2				
SOZ 1	-203.93	-193.02	-196.79	-140.53
TS _{ester} 2	-69.69	-69.29	-69.92	-24.46
Cester 2	-561.72	-553.43	-550.90	-529.31

Table S64: sCl + HNO₃ relative energies (ΔE), zero-point corrected energy (ΔZPE), enthalpies ($\Delta H_{298.15}$) and Gibbs free energies ($\Delta G_{298.15}$) [units=kJ mol⁻¹]

#sCl	Stationary Point	ΔE	ΔZPE	$\Delta H_{298.15}$	$\Delta G_{298.15}$
1	PRC	-58.97	-54.38	-54.44	-11.56
	TS	-48.34	-48.32	-51.29	-1.74
	Pr	-192.76	-179.86	-181.93	-132.29
2	PRC	-39.46	-35.47	-33.82	2.60
	TS	-28.94	-29.05	-31.33	22.38
	Pr	-205.61	-196.26	-196.60	-147.26
3	PRC	-47.09	-42.96	-42.10	0.50
	TS	-34.54	-35.92	-38.21	14.08
	Pr	-200.32	-190.26	-191.08	-139.84
4	PRC	-60.66	-56.31	-55.35	-13.31
	TS	-59.30	-57.16	-59.45	-5.41
	Pr	-220.67	-213.94	-214.08	-164.68
5	PRC	-60.27	-56.03	-55.16	-11.62
	TS	-49.84	-52.49	-54.68	-0.74
	Pr	-209.83	-203.44	-203.90	-152.16

Table S65: sCl + TFA relative energies (ΔE), zero-point corrected energy (ΔZPE), enthalpies ($\Delta H_{298.15}$) and Gibbs free energies ($\Delta G_{298.15}$) [units=kJ mol⁻¹]

#sCl	Stationary Point	ΔE	ΔZPE	$\Delta H_{298.15}$	$\Delta G_{298.15}$
1	PRC	-45.19	-41.91	-40.16	-5.73
	Pr	-194.67	-180.21	-182.57	-130.73
2	PRC	-33.09	-30.30	-27.81	5.36
	Pr	-195.66	-184.09	-185.55	-130.37
3	PRC	-34.83	-31.94	-29.55	173.58
	Pr	-204.27	-192.57	-193.93	-138.83
4	PRC	-41.65	-38.73	-36.27	-2.79
	Pr	-224.07	-214.23	-215.92	-158.97
5	PRC	-40.48	-37.60	-35.19	0.04
	Pr	-216.16	-207.43	-208.82	-153.23

Table S66:sCl + HF relative energies (ΔE), zero-point corrected energy (ΔZPE), enthalpies ($\Delta H_{298.15}$) and Gibbs free energies ($\Delta G_{298.15}$) [units=kJ mol⁻¹]

#sCl	Stationary Point	ΔE	ΔZPE	$\Delta H_{298.15}$	$\Delta G_{298.15}$
1	PRC	-49.75	-39.94	-43.70	-8.23
	TS	-5.74	-1.52	-8.35	35.32
	Pr	-198.52	-178.46	-183.89	-142.94
2	PRC	-34.78	-26.13	-28.54	3.90
	TS	20.19	22.62	16.53	61.77
	Pr	-209.51	-192.00	-196.73	-154.41
3	PRC	-40.17	-30.95	-34.22	2.10
	TS	5.75	8.86	2.69	47.20
	Pr	-211.14	-192.83	-197.92	-153.99
4	PRC	-37.88	-29.10	-32.14	4.34
	TS	-21.37	-16.75	-23.08	23.14
	Pr	-251.93	-235.33	-240.46	-197.01
5	PRC	-42.30	-33.68	-36.44	-1.35
	TS	-7.91	-4.96	-11.10	34.41
	Pr	-244.40	-228.59	-233.72	-188.66

Table S67:sCl + HCl relative energies (ΔE), zero-point corrected energy (ΔZPE), enthalpies ($\Delta H_{298.15}$) and Gibbs free energies ($\Delta G_{298.15}$) [units=kJ mol⁻¹]

#sCl	Stationary Point	ΔE	ΔZPE	$\Delta H_{298.15}$	$\Delta G_{298.15}$
1	PRC	-31.56	-25.70	-28.93	7.08
	TS	-30.45	-25.83	-31.26	10.50
	Pr	-202.76	-180.51	-185.15	-144.07
2	PRC	-23.31	-17.26	-18.64	12.23
	TS	3.34	6.03	1.41	44.79
	Pr	-213.79	-194.13	-197.86	-154.89
3	PRC	-25.24	-19.12	-21.40	14.04
	TS	-15.75	-13.43	-18.13	24.28
	Pr	-213.84	-193.33	-197.47	-152.83
4	PRC	-27.68	-21.39	-22.89	7.87
	TS	-24.08	-19.06	-23.84	20.19
	Pr	-231.78	-213.42	-217.41	-172.92
5	PRC	-28.93	-22.96	-24.70	9.23
	TS	-18.59	-16.50	-21.07	22.66
	Pr	-225.22	-207.62	-211.63	-165.72

Table S68:*sCl + H₂S* relative energies (ΔE), zero-point corrected energy (ΔZPE), enthalpies ($\Delta H_{298.15}$) and Gibbs free energies ($\Delta G_{298.15}$) [units=kJ mol⁻¹]

#sCl	Stationary Point	ΔE	ΔZPE	$\Delta H_{298.15}$	$\Delta G_{298.15}$
1	PRC1	-20.13	-13.92	-14.65	19.13
	TS1	-2.57	4.26	-1.18	44.84
	Pr 1	-210.81	-188.36	-192.97	-148.26
	PRC2	-19.88	-13.71	-14.42	19.32
	TS2	-2.20	4.61	-0.87	45.24
	Pr 2	-209.59	-187.39	-191.75	-147.66
2	PRC1	-13.67	-9.89	-8.34	16.54
	TS1	2.81	9.95	5.34	53.07
	Pr 1	-226.11	-205.92	-209.55	-163.12
	PRC2	-11.88	-7.98	-6.42	16.71
	TS2	5.08	11.86	7.37	54.74
	Pr 2	-229.42	-209.23	-212.81	-166.63
3	PRC1	-11.12	-6.04	-5.77	26.67
	TS1	-5.20	1.73	-3.10	44.82
	Pr 1	-228.05	-207.02	-211.14	-162.81
	PRC2	-9.33	-4.25	-3.99	28.53
	TS2	-3.89	3.00	-1.86	46.19
	Pr 2	-222.85	-202.41	-206.02	-159.23
4	PRC1	-19.79	-15.27	-14.27	14.87
	TS1	-15.76	-9.16	-12.98	33.07
	Pr1	-240.02	-221.25	-225.09	-177.10
	PRC2	-19.28	-14.83	-13.77	14.75
	TS2	-13.31	-6.89	-10.74	35.32
	Pr2	-238.68	-220.07	-223.56	-177.03
5	PRC1	-23.07	-18.27	-17.64	14.81
	TS1	-7.06	-1.01	-5.53	43.20
	Pr1	-236.10	-218.41	-222.02	-173.29
	PRC2	-23.07	-18.31	-17.65	14.84
	TS2	-6.51	-0.41	-4.94	43.97
	Pr2	-235.98	-218.50	-221.90	-174.09

Table S69:*sCl + H₂O relative energies (ΔE), zero-point corrected energy (ΔZPE), enthalpies ($\Delta H_{298.15}$) and Gibbs free energies ($\Delta G_{298.15}$) [units=kJ mol⁻¹]*

#sCl	Stationary Point	ΔE	ΔZPE	$\Delta H_{298.15}$	$\Delta G_{298.15}$
1	PRC	-33.45	-25.23	-26.71	2.98
	TS1	3.66	14.28	6.75	54.28
	Pr 1	-198.40	-177.58	-183.33	-139.24
	TS2	7.30	17.54	9.95	57.61
	Pr 2	-197.78	-177.21	-182.69	-139.24
2	PRC	-20.02	-13.83	-13.66	10.42
	TS1	16.17	25.70	19.17	67.37
	Pr 1	-216.28	-198.27	-203.24	-157.68
	TS2	14.94	24.60	18.00	65.52
	Pr 2	-210.94	-194.29	-198.52	-155.45
3	PRC	-29.63	-22.29	-23.22	7.38
	TS1	1.92	12.20	5.33	54.25
	Pr 1	-220.98	-201.72	-207.21	-159.98
	TS2	4.56	14.62	7.62	57.37
	Pr 2	-216.01	-197.85	-202.59	-157.32
4	PRC1	-29.54	-22.47	-23.08	8.91
	TS1	-23.47	-13.40	-19.82	28.92
	Pr1	-262.27	-245.71	-250.98	-204.25
	PRC2	-31.47	-24.20	-25.02	7.81
	TS2	-23.25	-13.34	-19.86	28.12
	Pr2	-262.26	-245.48	-250.81	-204.05
5	PRC	-34.39	-26.80	-27.95	7.40
	TS1	-15.50	-5.89	-12.48	36.97
	Pr1	-248.73	-233.05	-238.13	-190.76
	TS2	-17.18	-7.47	-14.07	35.93
	Pr2	-257.20	-241.56	-246.67	-198.96

Table S70:*sCl + (H₂O)₂* relative energies (ΔE), zero-point corrected energy (ΔZPE), enthalpies ($\Delta H_{298.15}$) and Gibbs free energies ($\Delta G_{298.15}$) [units=kJ mol⁻¹]

#sCI	Stationary Point	ΔE	ΔZPE	$\Delta H_{298.15}$	$\Delta G_{298.15}$
1	PRC1.1	-49.94	-39.05	-42.86	6.35
	TS1.1	-37.08	-26.33	-36.42	27.73
	Pr1.1	-209.63	-187.13	-193.79	-136.65
	AAAH- con 1 + H ₂ O	-178.18	-166.33	-170.00	-151.72
	PRC1.2	-49.71	-39.63	-42.97	5.58
	TS1.2	-31.89	-22.56	-32.71	31.77
	Pr1.2	-212.35	-190.79	-196.74	-140.82
	AAAH- con 2 + H ₂ O	-177.56	-165.96	-169.36	-151.71
	PRC1.3	-52.02	-40.84	-44.84	5.83
	TS1.3	-37.47	-27.17	-37.34	27.30
	Pr1.3	-215.73	-193.17	-199.71	-142.41
	AAAH- con 2 + H ₂ O	-177.56	-165.96	-169.36	-151.71
	PRC1.4	-50.89	-39.70	-43.73	6.71
	TS1.4	-38.06	-27.03	-37.10	27.20
	Pr1.4	-211.93	-189.14	-195.99	-138.25
	AAAH- con 1 + H ₂ O	-178.18	-166.33	-170.00	-151.72
2	PRC1.1	-44.07	-34.62	-37.44	14.41
	TS1.1	-33.71	-24.33	-33.38	32.85
	Pr1.1	-225.49	-207.53	-212.46	-155.75
	AAAH- con 2 + H ₂ O	-190.73	-183.04	-185.19	-167.93
	PRC1.2	-45.59	-35.42	-38.62	14.50
	TS1.2	-37.16	-27.10	-36.10	30.21
	Pr1.2	-227.04	-208.90	-213.81	-157.22
	AAAH- con 2 + H ₂ O	-190.73	-183.04	-185.19	-167.93
	PRC1.3	-41.52	-32.45	-34.99	16.62
	TS1.3	-32.06	-21.91	-30.89	35.88
	Pr1.3	-225.79	-207.51	-212.36	-155.73
	AAAH- con 1 + H ₂ O	-196.06	-187.02	-189.91	-170.15
	PRC1.4	-42.99	-33.27	-36.24	16.42
	TS1.4	-35.14	-24.60	-33.56	33.19
	Pr1.4	-230.34	-211.42	-216.72	-158.84
	AAAH- con 1 + H ₂ O	-196.06	-187.02	-189.91	-170.15
3	PRC1.1	-45.80	-36.24	-39.29	12.94
	TS1.1	-39.81	-29.91	-39.16	28.14
	Pr1.1	-230.20	-211.29	-216.65	-157.80
	AAAH con 2 + H ₂ O	-195.79	-186.60	-189.25	-169.79
	PRC1.2	-47.90	-37.15	-40.95	13.39
	TS1.2	-44.92	-34.20	-43.41	23.91
	Pr1.2	-234.30	-214.87	-220.32	-162.09
	AAAH con 2 + H ₂ O	-195.79	-186.60	-189.25	-169.79
	PRC1.3	-46.58	-36.49	-39.81	11.81
	TS1.3	-42.62	-31.73	-40.88	25.82
	Pr1.3	-232.92	-212.74	-218.65	-158.95
	AAAH con 2 + H ₂ O	-200.76	-190.47	-193.88	-172.45
	PRC1.4	-47.44	-37.11	-40.58	12.03
	TS1.4	-43.44	-32.44	-41.56	25.24
	Pr1.4	-234.79	-214.44	-220.45	-160.33
	AAAH con 2 + H ₂ O	-200.76	-190.47	-193.88	-172.45

4	PRC1.1	-65.71	-54.79	-58.92	-2.60
	TS1.1	-67.51	-57.30	-66.16	0.80
	Pr1.1	-278.44	-261.06	-266.52	-208.19
	AAAH- con 2 + H ₂ O	-242.05	-234.23	-237.48	-216.53
	PRC1.2	-67.94	-56.28	-60.84	-3.29
	TS1.2	-70.81	-59.93	-68.72	-1.75
	Pr1.2	-278.44	-261.06	-266.52	-208.19
	AAAH- con 2 + H ₂ O	-242.05	-234.23	-237.48	-216.53
	PRC1.3	-65.12	-53.83	-58.33	-0.73
	TS1.3	-69.10	-58.00	-66.66	0.38
	Pr1.3	-283.44	-266.55	-271.62	-214.61
	AAAH- con 1 + H ₂ O	-242.05	-234.46	-237.64	-216.72
	PRC1.4	-66.14	-54.49	-59.28	-1.08
	TS1.4	-70.78	-59.46	-68.03	-1.22
	Pr1.4	-283.44	-266.55	-271.62	-214.61
	AAAH- con 1 + H ₂ O	-242.05	-234.46	-237.64	-216.72
5	PRC1.1	-62.87	-53.00	-56.31	-2.06
	TS1.1	-60.67	-51.06	-59.89	7.37
	Pr1.1	-271.27	-255.42	-260.48	-201.38
	AAAH con 2 + H ₂ O	-236.98	-230.31	-233.34	-211.43
	PRC1.2	-65.55	-54.87	-58.65	-2.92
	TS1.2	-64.48	-54.26	-63.06	4.42
	Pr1.2	-273.87	-257.64	-262.87	-203.30
	AAAH con 2 + H ₂ O	-236.98	-230.31	-233.34	-211.43
	PRC1.3	-62.79	-52.60	-55.99	-2.37
	TS1.3	-59.52	-50.09	-59.06	8.25
	Pr1.3	-271.91	-255.94	-260.97	-202.54
	AAAH con 1 + H ₂ O	-228.51	-221.80	-224.80	-203.23
	PRC1.4	-61.51	-51.64	-54.83	-2.24
	TS1.4	-57.37	-48.45	-57.53	10.05
	Pr1.3	-271.91	-255.94	-260.97	-202.54
	AAAH con 1 + H ₂ O	-228.51	-221.80	-224.80	-203.23

Table 71: sCI + MeOH relative energies (ΔE), zero-point corrected energy (ΔZPE), enthalpies ($\Delta H_{298.15}$) and Gibbs free energies ($\Delta G_{298.15}$) [units=kJ mol⁻¹]

#sCl	Stationary Point	ΔE	ΔZPE	$\Delta H_{298.15}$	$\Delta G_{298.15}$
1	PRC	-34.19	-28.02	-27.75	7.54
	TS1	-12.01	-3.82	-8.37	42.59
	Con 1	-214.74	-199.05	-202.39	-153.88
	TS2	-7.16	0.10	-4.34	46.06
	Con 2	-205.71	-190.64	-193.87	-146.25
2	PRC 1	-20.14	-16.11	-14.10	12.20
	TS1	-0.87	5.80	2.56	54.13
	Pr1	-234.32	-221.49	-223.71	-173.78
	TS2	6.00	11.67	8.27	60.32
3	PRC	-31.11	-25.71	-24.89	11.75
	TS1	-16.83	-9.41	-13.27	39.95
	Pr1	-229.68	-215.37	-218.26	-165.47
	TS2	-12.45	-5.49	-9.13	43.38
	Pr2	-230.78	-216.98	-219.65	-168.36
4	PRC1	-30.26	-25.74	-24.07	8.00
	TS1	N/A	N/A	N/A	N/A
	Pr1	-278.79	-266.92	-269.31	-218.19
	PRC2	-35.73	-30.11	-29.29	7.75
	TS2	-33.98	-27.30	-30.75	22.32
	Pr2	-274.66	-262.01	-264.94	-212.13
5	PRC	-35.99	-30.77	-29.72	7.83
	TS1	-32.60	-26.18	-29.59	23.81
	Pr1	-258.18	-247.21	-249.63	-197.29
	PRC2	-36.93	-31.82	-30.73	6.91
	TS2	-32.73	-26.39	-29.66	23.37
	Pr2	-267.52	-256.12	-258.66	-206.58

S6. Partially Barrierless sCl 4 + MeOH Reaction

As noted in the main body of the text, many of the bimolecular HFO-sCl reactions, such as **sCl 4** + HCHO and **sCl 3** + TFA, are barrierless and other, such as **sCl 1** + HNO₃ and **sCl 4** + (H₂O)₂, have large rate constants ($k_{ME} \geq 10^{-10}$ cm³ s⁻¹). In these conditions nearly every collision would lead to a reaction meaning that the only constraint to reaction is collision frequency. This leads to employing a rate-limiting collision model, such as: the collision-limited rate coefficient (k_{COLL}), calculated using the ideal gas laws; or the often-larger dipole-dipole capture limit (k_{d-d}), which also includes the impact of dipole-dipole interactions.

However, the **sCl 4** (*syn*-CF₃CFOO) reaction with MeOH shows an unusual phenomenon because there is a verified TS_{AAAH} 2 structure (seen in Figure S7), but the current *B3LYP/aug-cc-pVTZ* method is unable to produce a fixed TS_{AAAH} 1 structure. This is unusual because nearly all other sCl + MeOH reactions in both this study and the Watson *et al.* study produce two TS_{AAAH} channels.^{6,37} To identify other reaction channel TS_{AAAH} 1, like a structure shown in Figure S7, a minimum energy pathway (MEP) between products and reactants using the Gaussian09 *modredundant* operation. This profile is seen in Figure S7:

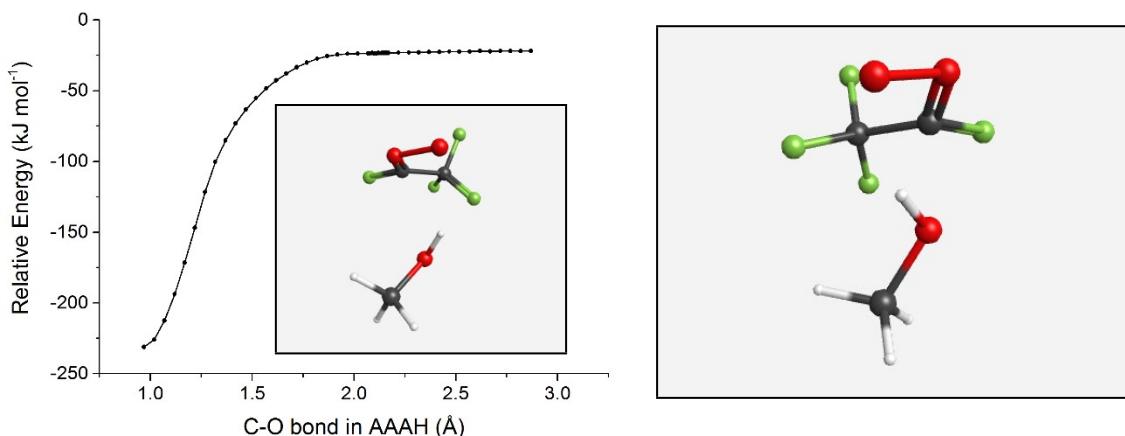


Figure S7: The minimum energy profile (MEP) of the barrierless sCl 4 + MeOH TS_{AAAH} 1 channel (left) with a graphic of an expected TS_{AAAH} 1; and the sCl 4 + MeOH TS_{AAAH} 2 structure (right). Graphics of sCl 4 + MeOH TS_{AAAH} 1 & 2 are generated using WebMO software.³⁸

This MEP generated for the **sCl 4** + MeOH TS_{AAAH} 1 channel, shown in Figure S7, has no TS inversion point meaning that it is barrierless. The TS_{AAAH} 2 barrier does have a low energy (-27.3 kJ mol⁻¹) which if it were the only barrier would produce a rate constant of $\sim 5.17 \times 10^{-11}$ cm³ s⁻¹. However, as the TS_{AAAH} 1 channel is barrierless, the k_{d-d} constant (5.50×10^{-10} cm³ s⁻¹) is employed as the rate constant. By the inclusion of the barrierless channel, this increases the rate constant by an order of magnitude proving that identifying TS_{AAAH} 1 channel as barrierless is important.

Only one other sCl + MeOH reaction has no identified TS_{AAAH} 1 barrier and one TS_{AAAH} 2 barrier, which is the CF₂OO + MeOH reaction in the Watson *et al.* study.^{6,37} This CF₂OO + MeOH significantly with the **sCl 4** + MeOH reaction in one way because, even if there were another TS_{AAAH} 1 barrier for the CF₂OO + MeOH reaction, the overall k_{ME} at 298 K constant from the TS_{AAAH} 2 barrier is so high (1.96×10^{-8} cm³ s⁻¹) that it exceeds any likely k_{d-d} or k_{COLL} values.^{6,37}

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S8. Catersian coordinates for all structures and IRCS:

S8.1 Criegee intermediates

Compound: sCI 1 (HCHOO)	Energy (kJ mol⁻¹): -189.400847568329
Reaction Coordinates: 6 -1.068412 0.202194 -0.000000 8 -0.003231 -0.459294 -0.000000 8 1.179333 0.194958 0.000000 1 -1.021445 1.284352 -0.000000 1 -1.976903 -0.382830 0.000002	Frequencies (cm⁻¹): 527.7609, 673.7578, 912.8476, 951.0664, 1242.0554, 1402.0013, 1543.5428, 3118.1651, 3267.9600

Compound: sCI 2 (<i>Syn</i> -CF ₃ CHOO)	Energy (kJ mol⁻¹): -526.2050097
Reaction Coordinates: 6 0.544326 0.408563 0.000000 6 -0.965669 0.489858 0.000000 8 -1.720767 -0.517005 -0.000000 8 -1.205470 -1.752491 -0.000000 1 -1.466000 1.447260 0.000000 9 1.014961 -0.207422 1.087387 9 1.014961 -0.207422 -1.087387 9 1.014961 1.672420 0.000000	Frequencies (cm⁻¹): 83.4264, 179.3385, 247.1426, .2827, 478.5065, 507.3023, ,536.2006, 591.5612, 758.0400, 772.8798, 885.1813, 947.2031, 1151.7894, 1176.1918, 1243.1084, 1364.7774, 1539.7613, ,3220.6741

Compound: sCI 3 (<i>Anti</i> -CF ₃ CHOO)	Energy (kJ mol⁻¹): -526.2059784
Reaction Coordinates: 6 -0.814124 -0.010987 0.000000 6 0.599627 -0.529934 0.000001 8 1.532385 0.307937 -0.000000 8 2.803565 -0.122750 -0.000000 1 0.814702 -1.591587 0.000002 9 -0.871266 1.317793 -0.000001 9 -1.465219 -0.472475 -1.084562 9 -1.465219 -0.472472 1.084563	Frequencies (cm⁻¹): 53.4225, 186.5103, 191.8874, 388.9843, 394.1412, 416.4183, 552.7580, 560.7356, 699.8777, 883.4653, 889.1822, 988.9465, 1131.2630, 1171.7198, 1270.0033, 1355.9541, 1546.1599, 3188.0668

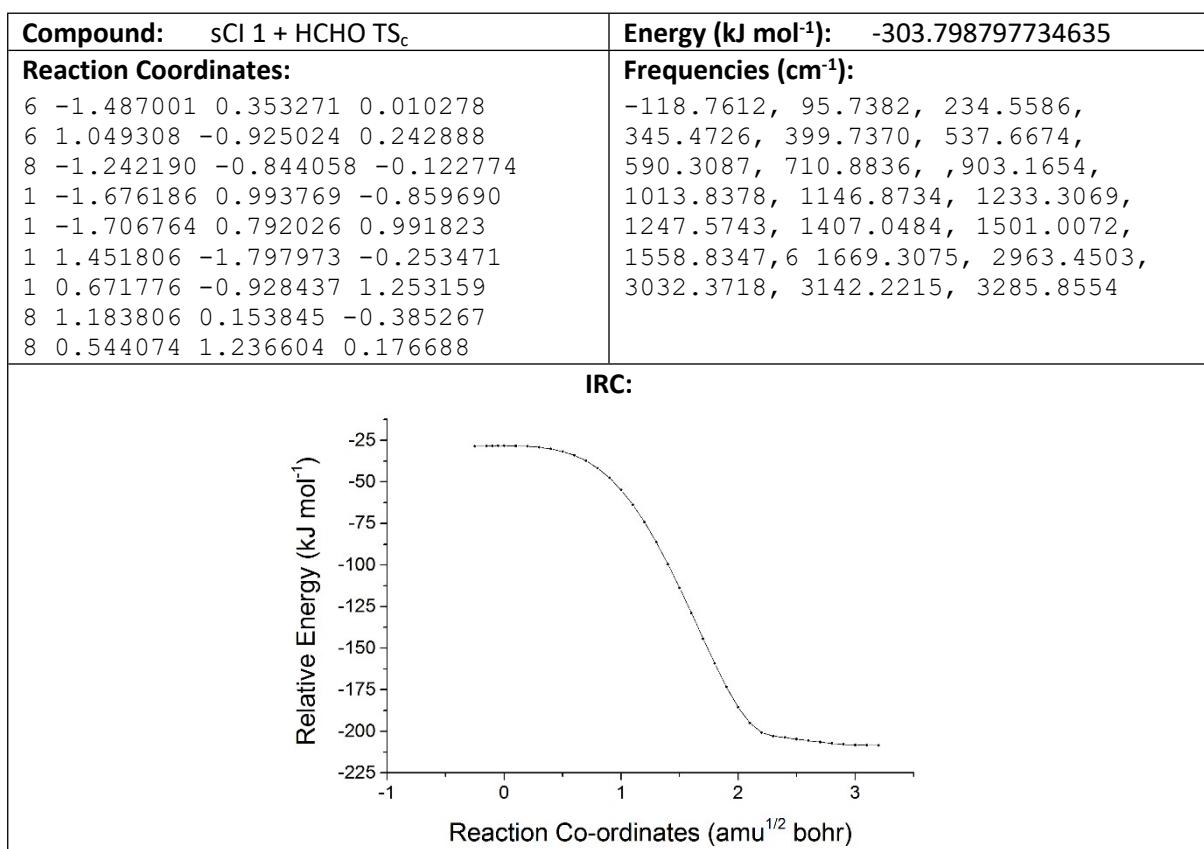
Compound: sCI 4 (<i>Syn</i> -CF ₃ CFOO)	Energy (kJ mol⁻¹): -625.3731138
Reaction Coordinates: 6 0.267347 0.734443 0.000000 6 -0.668190 -0.480912 0.000000 8 -0.317071 -1.670141 0.000000 8 1.031885 -1.966693 0.000000 9 -1.951916 -0.250006 0.000000 9 1.031885 0.735046 1.087938 9 1.031885 0.735046 -1.087938 9 -0.480015 1.843635 0.000000	Frequencies (cm⁻¹): 58.8820, 180.2110, 218.4479, 269.2791, 292.9260, ,363.8515, 484.2548, 493.0742, 582.0033, 631.0249, 653.9584, 780.2000, 927.2456, 1158.8972, 1195.1079, 1198.3764, 1398.0465, 1580.9318

Compound: sCI 5 (<i>Anti</i> -CF ₃ CFOO)	Energy (kJ mol⁻¹): -625.3783361
Reaction Coordinates: 6 -0.277404 -0.919423 0.000000 6 0.018963 0.571292 0.000000 8 1.190937 0.987253 0.000000 8 1.417147 2.337638 0.000000 9 -0.997920 1.369496 0.000000 9 0.847756 -1.621862 0.000000 9 -0.997920 -1.235503 1.084149 9 -0.997920 -1.235503 -1.084149	Frequencies (cm⁻¹): 41.8080, 154.2443, 183.9192, 293.0541, 346.5294, 372.3393, 406.8758, 515.4784, 576.4030, 659.0393, 728.8220, .5490, 906.9387, 1155.2541, 1155.7283, 1229.1869, 1400.7892, 1601.0242

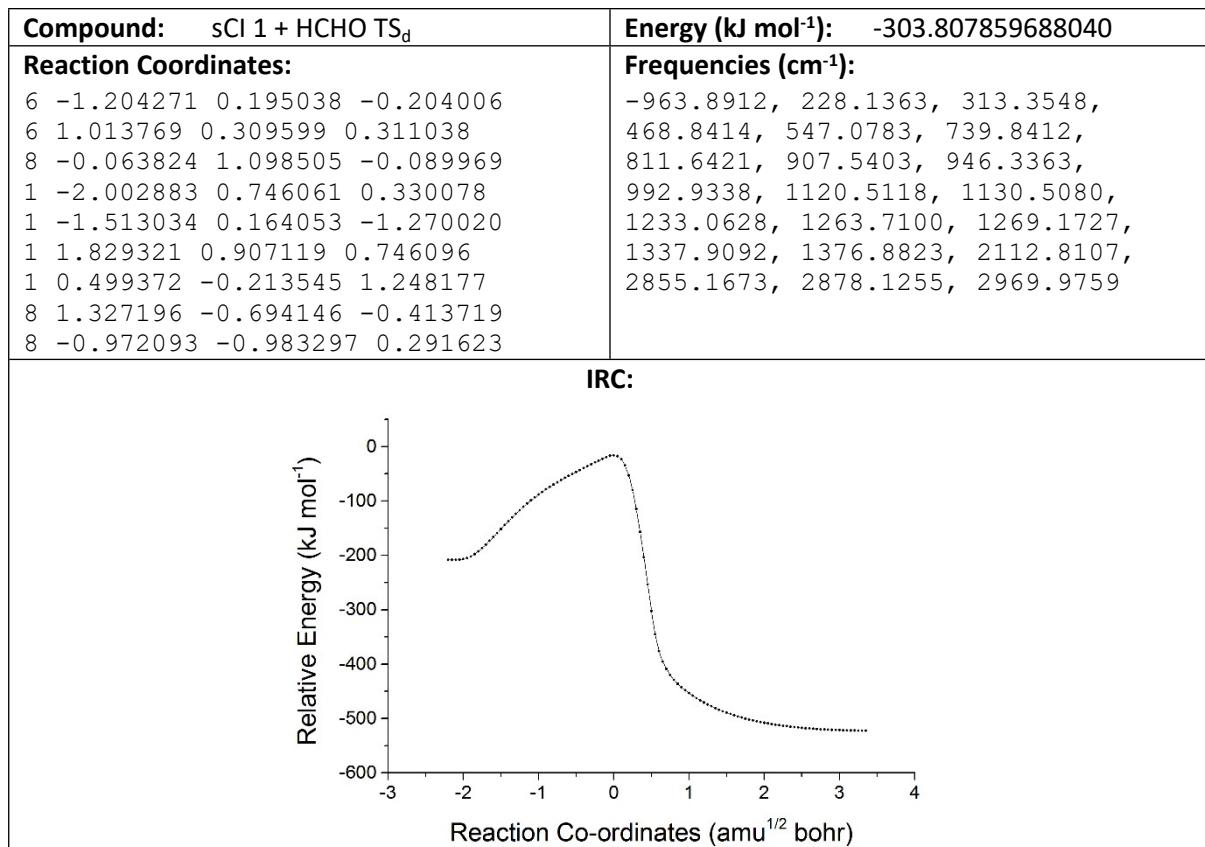
S8.2 Reactions with HCHO

Compound: HCHO	Energy (kJ mol⁻¹): -114.386051813463
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -0.000000 -0.526970 -0.000000	1198.1780, 1262.9587, 1530.2218,
8 0.000000 0.673271 0.000000	1813.2454, 2885.3911, 2940.0951
1 -0.000000 -1.112173 0.938294	
1 -0.000000 -1.112173 -0.938294	

Compound: sCI 1 + HCHO PRC	Energy (kJ mol⁻¹): -303.798329359066
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -1.568793 0.372025 -0.025055	74.8509, 103.5754, 200.3540,
8 -1.363160 -0.828477 -0.072239	278.7803, 321.9694, 483.9337,
1 -1.657461 0.978246 -0.937211	530.6353, 685.8280, 894.8752,
1 -1.783433 0.887713 0.921406	1002.3931, 1157.3881, 1239.1392,
6 1.125843 -0.947653 0.216022	1254.7518, 1413.4948, 1517.9212,
1 1.548975 -1.796415 -0.304241	1564.5284, 1716.1748, 2949.5550,
1 0.693612 -0.991078 1.204393	3015.1791, 3137.8091, 3282.9945
8 1.223576 0.146118 -0.385058	
8 0.621585 1.229271 0.203528	



Compound: sCl 1 + HCHO HOZ	Energy (kJ mol⁻¹): -303.878537225920
Reaction Coordinates:	Frequencies (cm⁻¹):
6 1.118320 0.321848 -0.133769 8 0.000001 1.185457 -0.000005 1 -1.921112 0.617737 -0.541368 1 -1.444573 0.296995 1.177335 6 -1.118317 0.321850 0.133776 1 1.921105 0.617728 0.541390 1 1.444593 0.296997 -1.177322 8 0.669522 -0.948459 0.290933 8 -0.669527 -0.948454 -0.290938	163.2068, 371.3640, 708.1220, 750.0747, 861.2831, 925.9772, 954.1919, 1041.1937, 1086.5608, 1142.2906, 1145.0755, 1218.3662, 1222.1489, 1370.6921, 1414.6761, 1519.1635, 1529.2393, 3014.9743, 3016.9237, 3096.0501, 3097.3415



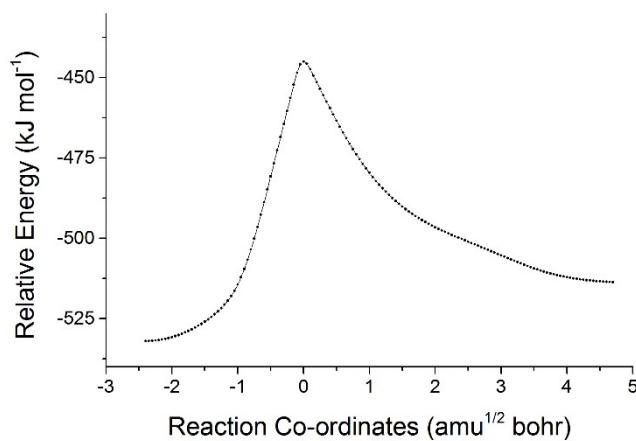
Compound: sCl 1 + HCHO HAE con 1	Energy (kJ mol⁻¹): -303.997300763222
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -1.305850 0.289288 -0.231552 6 0.940582 0.108875 0.562488 8 -0.113961 0.892842 -0.041694 1 2.217403 0.120005 -0.894036 1 -1.978441 0.989618 -0.741641 1 0.504100 -0.659409 1.192085 1 1.510830 0.833055 1.143035 8 1.709527 -0.532334 -0.400318 8 -1.603351 -0.819539 0.106380	70.4857, 248.4112, 299.0854, 381.9429, 547.6482, 755.4857, 887.0359, 1041.6181, 1059.1008, 1083.6745, 1160.7621, 1288.9470, 1396.6575, 1400.1716, 1435.6149, 1511.3613, 1800.1743, 3033.9528, 3065.1750, 3152.7357, 3803.4239

Compound: sCl 1 + HCHO TS _{ISO}	Energy (kJ mol⁻¹): -303.991840303773
Reaction Coordinates: 6 1.312004 0.313671 -0.206358 6 -0.930064 0.023397 0.582096 8 0.087792 0.853126 -0.000989 1 -1.526009 -0.131808 -1.262605 1 1.954963 1.077031 -0.660097 1 -0.464130 -0.834440 1.062866 1 -1.444117 0.655822 1.304341 8 -1.842940 -0.400862 -0.395533 8 1.653606 -0.800891 0.059156	Frequencies (cm⁻¹): -422.7115, 38.5836, 260.2890, 315.7944, , 510.8464, 752.0342, 886.7029, 1041.3061, 1062.2173, 1136.1380, 1174.7639, 1275.0176, 1301.7533, 1400.7928, 1454.4948, 1531.3924, 1804.1671, 3042.0345, 3062.5202, 3115.8914, 3834.6789

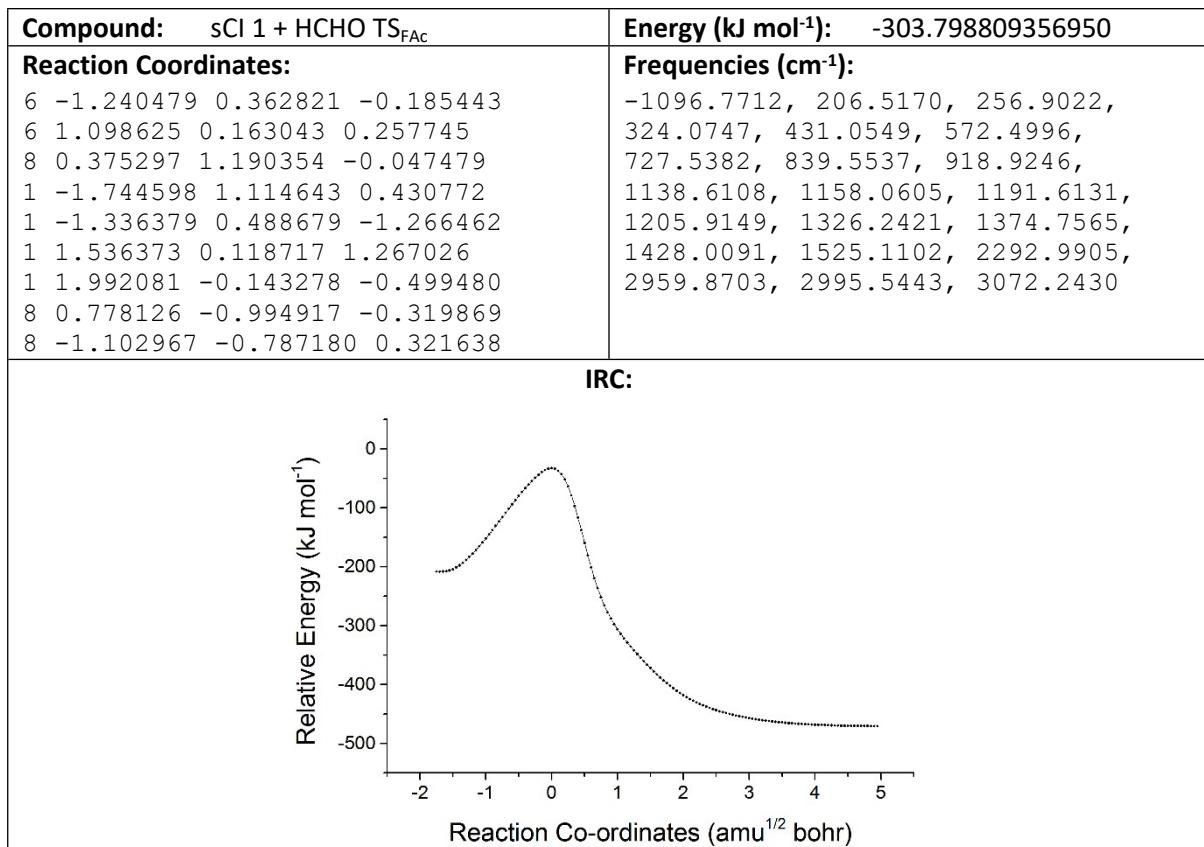
Compound: sCl 1 + HCHO HAE con 2	Energy (kJ mol⁻¹): -304.001807069015
Reaction Coordinates: 6 -1.260600 -0.237487 -0.167386 6 1.023485 -0.390383 0.400595 8 -0.168614 -1.005790 -0.171215 1 0.907472 1.414099 -0.227820 1 -2.108826 -0.788602 -0.589538 1 1.764187 -1.182273 0.386199 1 0.774725 -0.093172 1.420070 8 1.496495 0.660749 -0.360576 8 -1.317239 0.897188 0.233271	Frequencies (cm⁻¹): 134.0932, 255.9643, 311.9088, 498.3134, 561.1357, 795.3260, 864.0522, 1045.5382, 1057.3010, 1110.4985, 1185.2944, 1293.0581, 1394.8699, 1405.6822, 1444.0186, 1510.2254, 1767.9081, 3044.7503, 3055.0210, 3155.2797, 3770.5265

Compound: sCl 1 + HCHO TS _{HAE}	Energy (kJ mol⁻¹): -303.962565700012
Reaction Coordinates: 6 1.259724 0.228825 -0.065861 6 -1.300930 0.415269 0.217050 8 0.390711 1.135692 -0.108964 1 -0.140029 -1.130261 0.047319 1 2.302132 0.521079 -0.236182 1 -1.881506 1.181054 -0.297803 1 -1.172603 0.572406 1.291527 8 -1.283919 -0.759522 -0.264454 8 1.035612 -1.002275 0.159419	Frequencies (cm⁻¹): -1115.5862, 186.3394, 274.1524, 418.1722, 517.4046, 579.7763, 825.7630, 933.1747, 1065.4507, 1221.2134, 1269.0745, 1326.4903, 1369.5579, 1384.3739, 1426.3933, 1614.5353, 1664.1975, 1769.6394, 3013.1948, 3043.6088, 3103.6379

IRC:



Compound: sCl 1 + HCHO C _{FA} 1	Energy (kJ mol⁻¹): -303.990619638538
Reaction Coordinates: 6 1.657241 0.078650 -0.000000 8 1.043083 -1.098158 -0.000000 1 0.062756 -0.961278 0.000000 1 2.745848 -0.055528 0.000001 8 1.117502 1.158015 -0.000000 6 -2.029026 0.516294 -0.000000 1 -3.098216 0.781136 -0.000000 1 -1.289414 1.329652 0.000001 8 -1.684368 -0.642813 0.000000	Frequencies (cm⁻¹): 79.6236, 129.7900, 169.6478, 170.9258, 222.4930, 278.0783, 683.5053, 929.9644, 1069.2322, 1203.2888, 1228.4469, 1278.6536, 1380.1043, 1433.9700, 1520.6897, 1759.5075, 1791.2246, 2935.6401, 3032.6033, 3035.6839, 3329.9795

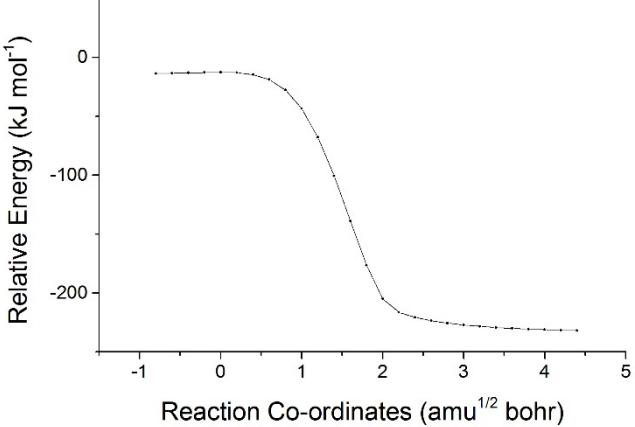


Compound: sCl 1 + HCHO C _{FA} 2	Energy (kJ mol⁻¹): -303.977039930179
Reaction Coordinates: 6 -1.065592 -0.087673 -0.000001 8 -0.608544 1.017095 -0.000001 1 -0.449781 -0.999539 -0.000001 8 -2.394409 -0.311317 0.000001 1 -2.569904 -1.260625 0.000001 6 2.137059 0.348295 0.000001 1 2.195113 0.930061 0.935342 1 2.195122 0.930065 -0.935336 8 2.028034 -0.851239 -0.000001	Frequencies (cm⁻¹): 62.0233, 85.0729, 96.4283, 104.1922, 151.3635, 200.7762, 548.5041, 664.7080, 1053.7673, 1121.0790, 1195.0673, 1262.7951, 1275.3896, 1425.0166, 1531.7929, 1794.9661, 1835.5258, 2912.6865, 2973.8170, 2995.6876, 3779.8452

Compound:	HCOOH conformer 1	Energy (kJ mol⁻¹):	-189.590120303627
Reaction Coordinates:		Frequencies (cm⁻¹):	
6 0.000000 0.421050 -0.000000 8 1.158545 0.117329 -0.000000 8 -1.028178 -0.446131 0.000000 1 -0.659713 -1.343779 0.000000 1 -0.383224 1.447894 -0.000000			

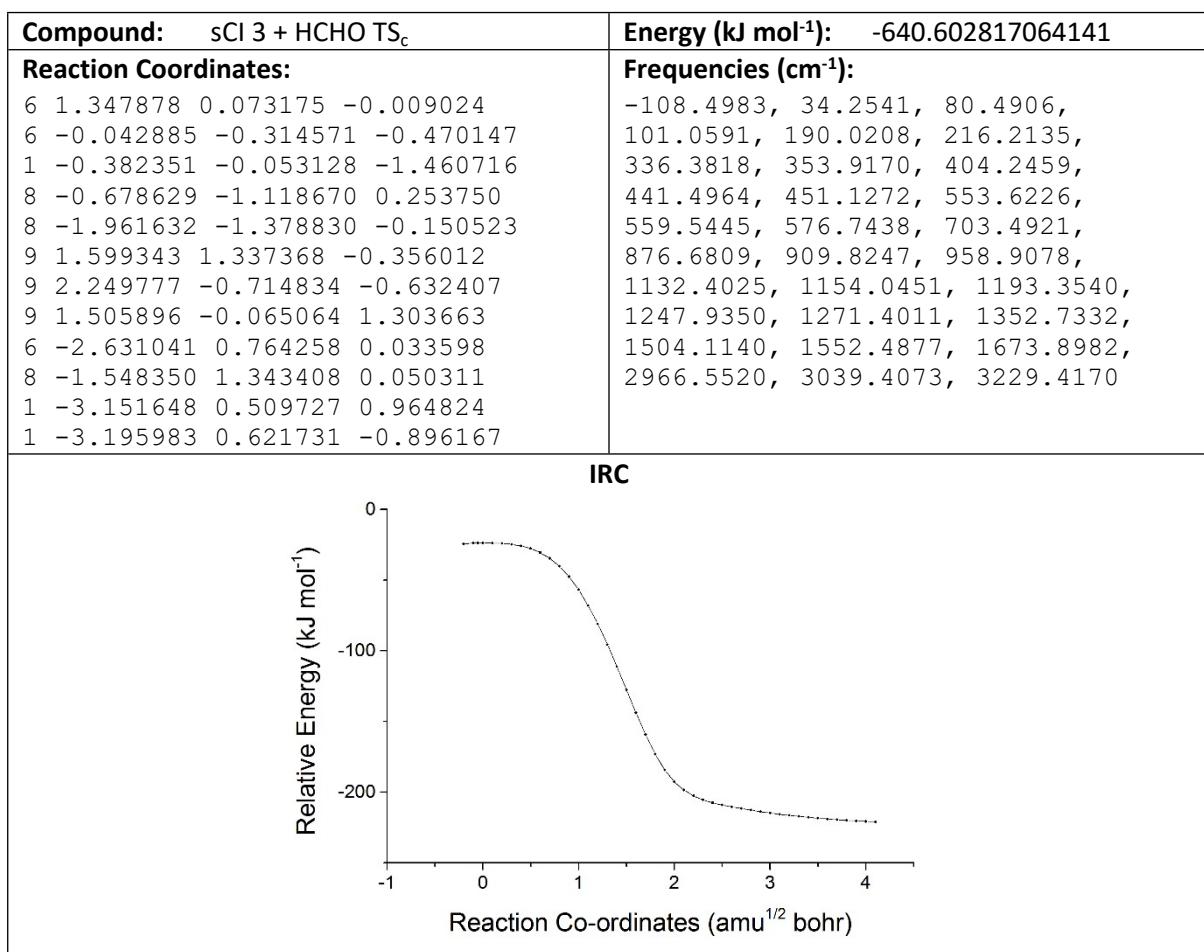
Compound:	HCOOH conformer 2	Energy (kJ mol⁻¹):	-189.583325855473
Reaction Coordinates:		Frequencies (cm⁻¹):	
6 0.133997 0.361764 -0.000000 8 1.173188 -0.220201 0.000000 8 -1.056592 -0.278816 -0.000000 1 -1.779073 0.361263 0.000001 1 0.042323 1.460290 -0.000000			

Compound:	sCI 2 + HCHO PRC	Energy (kJ mol⁻¹):	-640.600757425560
Reaction Coordinates:		Frequencies (cm⁻¹):	
6 -1.180344 -0.230641 -0.046857 6 -0.299183 0.435311 0.995105 1 -0.446023 0.217189 2.043332 8 0.555870 1.315583 0.742164 8 0.850811 1.621339 -0.538900 9 -1.964927 -1.112890 0.600682 9 -0.513822 -0.882972 -0.993276 9 -1.975264 0.679816 -0.627729 8 2.104802 -1.100010 0.536169 6 2.603324 -0.596540 -0.440591 1 2.258798 -0.835449 -1.459457 1 3.438696 0.118587 -0.362384			

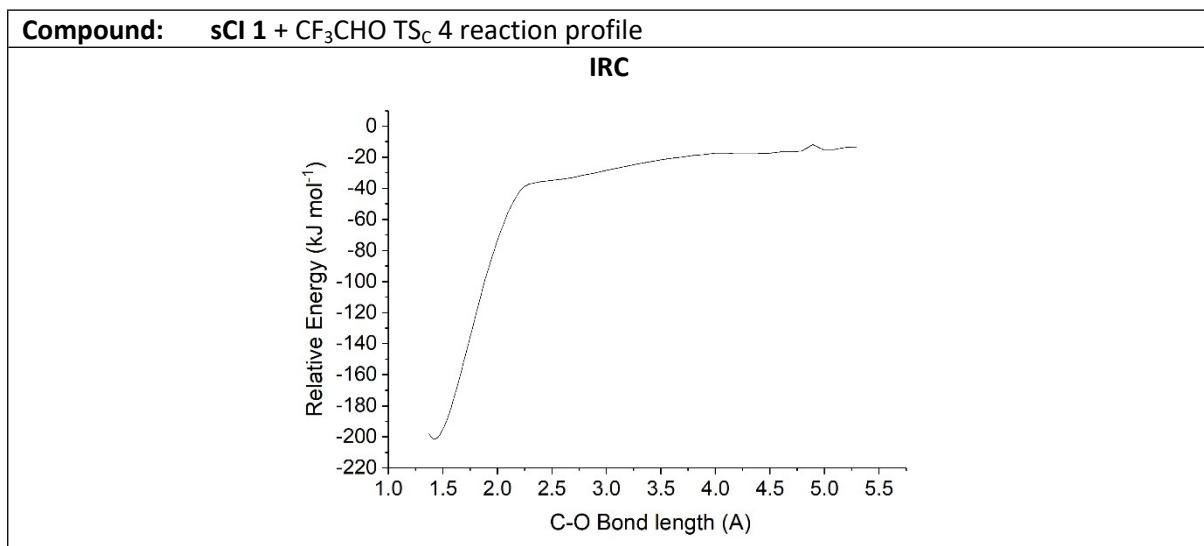
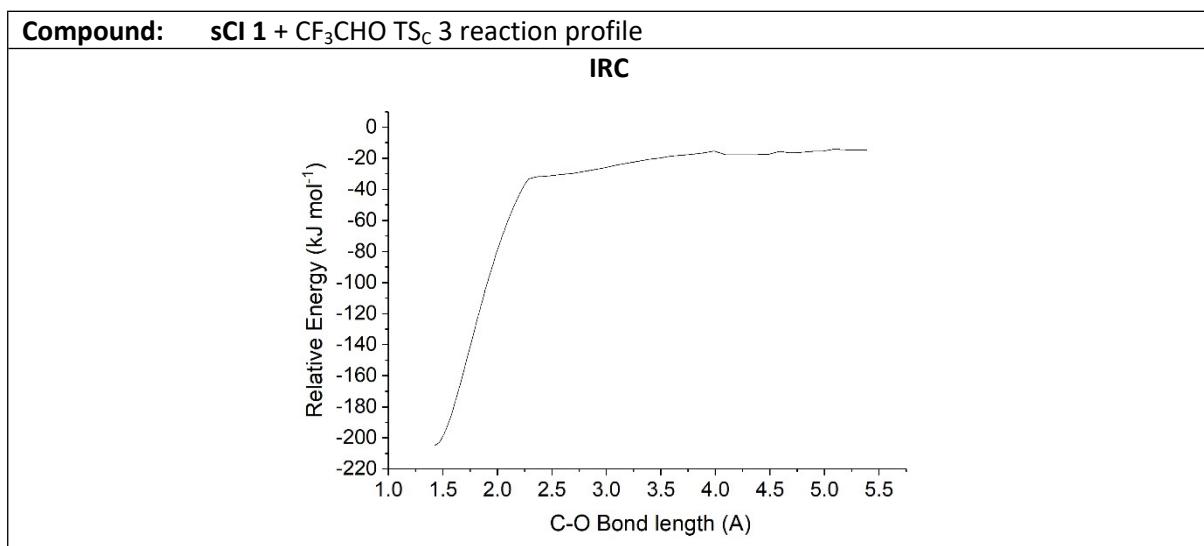
Compound: sCl 2 + HCHO TS _c	Energy (kJ mol⁻¹): -640.598646446407
Reaction Coordinates: 6 1.092215 -0.129997 0.058184 6 0.100620 0.466841 -0.934798 1 0.343043 0.361471 -1.984193 8 -0.797561 1.308568 -0.687912 8 -1.272647 1.384786 0.588980 9 1.764815 -1.099970 -0.580750 9 0.611298 -0.619326 1.188676 9 1.973002 0.848592 0.363137 8 -1.513961 -1.198018 -0.587087 6 -2.163773 -0.843793 0.385501 1 -1.799910 -0.986918 1.408525 1 -3.186182 -0.459212 0.270934	Frequencies (cm⁻¹): -101.5617, 38.4107, 137.2510, 151.3563, 177.4183, 217.1369, 308.7619, 321.1385, 348.7483, 452.6528, 476.9346, 513.6888, 538.4700, 575.6568, 755.0955, 846.8308, 874.2992, 926.7008, 1124.3981, 1158.5445, 1168.3125, 1255.2191, 1280.9832, 1368.3635, 1515.2006, 1562.4061, .6061, 2970.6576, 3052.7393, 3203.3552
IRC 	

Compound: sCl 2 + HCHO HOZ 1	Energy (kJ mol⁻¹): -640.693594205454
Reaction Coordinates: 6 -1.139558 0.021002 0.054719 6 0.170641 0.051083 -0.760444 1 -0.090393 0.101412 -1.820168 8 0.948084 -1.114764 -0.588504 8 1.785506 -0.749925 0.551876 6 2.154623 0.552911 0.174737 8 0.968223 1.133091 -0.357516 1 2.456979 1.080978 1.077043 1 2.936113 0.542263 -0.589467 9 -1.845283 1.141421 -0.185469 9 -0.938327 -0.072135 1.371621 9 -1.886550 -1.027268 -0.333743	Frequencies (cm⁻¹): 59.1063, 109.3494, 182.6206, 246.9681, 316.5757, 361.0581, 425.8088, 520.4651, 571.5285, 674.7369, 736.3570, 797.4111, 848.6080, 871.4982, 946.6412, 997.2191, 1056.8708, 1107.0675, 1149.2593, 1160.3118, 1162.8976, 1228.4602, 1290.9247, 1296.5657, 1398.3701, 1404.9809, 1520.9829, 3026.1795, 3063.2864, 3121.7528

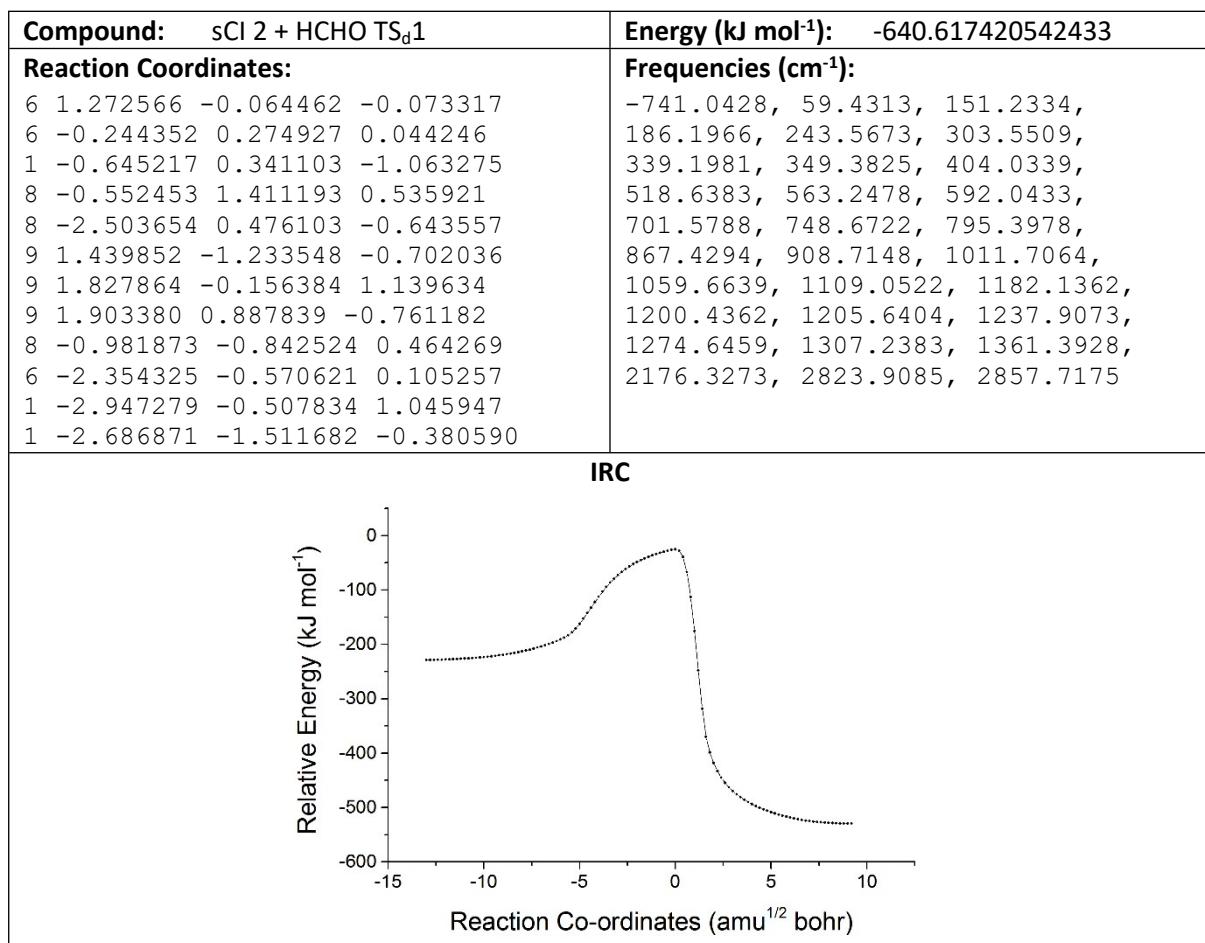
Compound: sCl 3 + HCHO PRC	Energy (kJ mol⁻¹): -640.603980608439
Reaction Coordinates: 6 1.395048 0.105593 -0.002788 6 0.022720 -0.321612 -0.476169 1 -0.363411 0.002589 -1.431854 8 -0.599263 -1.140283 0.236215 8 -1.841748 -1.515337 -0.175639 9 1.542932 1.421931 -0.179678 9 2.330250 -0.518267 -0.749247 9 1.608632 -0.193064 1.275855 6 -2.809371 0.706607 0.105628 8 -1.810139 1.383829 -0.036680 1 -3.152938 0.385323 1.099579 1 -3.461164 0.447480 -0.741295	Frequencies (cm⁻¹): 26.0215, 58.6615, 71.4215, 113.1524, 179.8264, 198.8805, 262.8157, 292.8331, 389.0975, 416.4647, 420.4978, 428.0257, 552.7682, 565.6375, 701.5734, 881.9017, 894.2454, 960.3389, 1133.9182, 1171.0645, 1189.3448, 1257.3784, 1269.2401, 1357.9965, 1522.7347, 1562.6462, 1734.2093, 2944.3504, 3012.1390, 3219.1546



Compound: sCl 2 + HCHO HOZ 2	Energy (kJ mol⁻¹): -640.692237225941
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -1.199474 -0.006887 0.025742 6 0.188982 0.075516 -0.632497 1 0.066771 0.219589 -1.709669 8 0.886315 1.153776 -0.023764 8 2.263774 0.688372 -0.073467 9 -1.912002 -0.986844 -0.552341 9 -1.865454 1.147074 -0.144863 9 -1.122709 -0.261077 1.336176 6 2.082671 -0.644790 0.338694 8 0.942805 -1.084525 -0.386997 1 1.899865 -0.705867 1.413986 1 2.958623 -1.210111 0.027110	69.3077, 77.4879, 179.4407, 253.9583, 338.8035, 359.3502, 394.3628, 520.8056, 560.9398, 693.6963, 725.0348, 762.4881, 855.7256, 889.6512, 943.1106, 1019.7862, 1053.9849, 1110.1529, 1156.8169, 1164.3491, 1169.6625, 1229.0507, 1279.8213, 1324.1151, 1395.3592, 1417.9899, 1520.2746, 3034.5928, 3042.4603, 3125.6720



Compound: sCl 2 + HCHO TS _{HOZ}	Energy (kJ mol⁻¹): -640.684269884806
Reaction Coordinates: 6 -1.234099 -0.016283 0.006992 6 0.217213 -0.006609 -0.495282 1 0.218277 -0.049061 -1.592572 8 0.855493 1.160309 -0.027043 8 2.258508 0.764235 0.099653 9 -1.857504 -1.113969 -0.452291 9 -1.882837 1.059394 -0.467529 9 -1.314504 -0.008957 1.336003 8 0.921671 -1.073107 0.066267 6 2.262564 -0.669437 -0.024369 1 2.695489 -0.937726 -0.993919 1 2.820401 -1.098946 0.805786	Frequencies (cm⁻¹): -165.8267, 67.5700, 127.7216, 175.6983, 283.1192, 342.4707, 385.5527, 523.0921, 560.2367, 704.2322, 737.7067, 743.3313, 872.9650, 886.7348, 969.5435, 1045.4194, 1080.9912, 1104.8589, 1149.3572, 1169.1195, 1171.0625, 1213.1812, 1288.3676, 1324.0129, 1403.4090, 1440.7109, 1542.0138, 2991.1447, 3001.7444, 3106.4815



Compound: sCl 2 + HCHO HAE1	Energy (kJ mol⁻¹): -640.805561786150
Reaction Coordinates: 6 -1.351795 -0.125467 0.048198 6 0.065137 0.490331 -0.114630 8 0.286253 1.656338 0.015812 8 0.932103 -0.477171 -0.423884 6 2.335472 -0.103488 -0.537474 8 2.973678 -0.179972 0.685937 1 3.074254 -1.101793 0.946812 1 2.399971 0.923885 -0.880158 1 2.725373 -0.805410 -1.272305 9 -1.357699 -1.041318 1.031295 9 -1.738702 -0.730048 -1.087855 9 -2.240127 0.818198 0.346133	Frequencies (cm⁻¹): 27.9049, 64.9955, 82.2236, 179.2410, 252.1594, 310.6391, 345.4473, 394.3004, 414.0244, 511.8310, 545.7231, 597.1920, 729.1634, 784.3867, 829.9422, 904.5459, 1062.5305, 1093.5179, 1149.6677, 1155.4669, 1212.3045, 1289.3237, 1330.8387, 1400.0656, 1440.7955, 1512.2298, 1841.2052, 3075.9799, 3157.7365, 3800.2458

Compound: sCl 2 + HCHO TS _{ISO} 1	Energy (kJ mol⁻¹): -640.799853308352
Reaction Coordinates: 6 -2.306299 -0.187066 -0.541446 1 -2.717596 -0.973662 -1.173786 1 -2.373790 0.790343 -1.020955 8 -2.873212 -0.183469 0.739477 1 -3.832746 -0.200502 0.667780 8 -0.929146 -0.531110 -0.379707 6 -0.075965 0.469745 -0.131854 6 1.350154 -0.115599 0.056431 9 2.220727 0.857467 0.311018 9 1.376565 -0.988847 1.073888 9 1.743814 -0.756991 -1.057586 8 -0.316787 1.636664 -0.074482	Frequencies (cm⁻¹): -238.4497, 25.7190, 72.3081, 88.5166, 183.0110, 252.5813, 326.5163, 348.8901, 412.3888, 509.5526, 542.9971, 591.4858, 730.3949, 782.7179, 829.1229, 963.6820, 1080.2566, 1134.6460, 1154.5610, 1155.6226, 1212.5796, 1256.2745, 1300.2406, 1335.7964, 1469.5455, 1523.2246, 1846.9871, 3041.7344, 3095.0383, 3824.0172

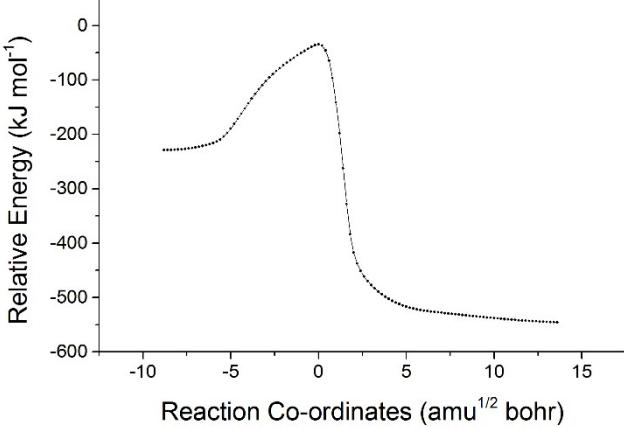
Compound: sCl 2 + HCHO HAE2	Energy (kJ mol⁻¹): -640.808924257914
Reaction Coordinates: 6 -2.318124 -0.533342 -0.393417 1 -2.737399 -1.532535 -0.416846 1 -2.393672 -0.038330 -1.361679 8 -2.916957 0.172141 0.624254 1 -2.679712 1.103745 0.538181 8 -0.888763 -0.765465 -0.148492 6 -0.130277 0.319304 -0.178029 6 1.358012 -0.057160 0.056207 9 2.122069 1.031756 0.040807 9 1.510068 -0.668874 1.240348 9 1.786173 -0.891063 -0.906048 8 -0.495740 1.449316 -0.356285	Frequencies (cm⁻¹): 27.6747, 80.3699, 121.4909, 183.6917, 250.5901, 295.7323, 345.1689, 415.1064, 509.5916, 513.2878, 551.7353, 590.8439, 739.6659, 792.6490, 838.9950, 890.4196, 1055.1402, 1117.6863, 1154.7276, 1163.8789, 1213.1588, 1290.4440, 1345.6009, 1406.7222, 1448.6631, 1511.0362, 1811.0926, 3062.8440, 3165.5278, 3776.0480

Compound: sCl 2 + HCHO TS _{HAE} 1	Energy (kJ mol⁻¹): -640.770283337633
Reaction Coordinates: 6 -1.390799 0.035220 0.021329 6 0.155415 -0.046897 -0.116251 1 1.833446 -1.112235 -0.029863 8 0.626879 -1.209964 -0.281780 8 2.801207 -0.543215 0.375049 9 -1.988022 -0.775715 -0.858875 9 -1.746400 -0.348928 1.259495 9 -1.836261 1.276977 -0.177237 8 0.797982 1.019752 -0.026542 6 2.702880 0.589884 -0.173390 1 3.096326 1.459636 0.350032 1 2.592859 0.679766 -1.254562	Frequencies (cm⁻¹): -1089.5421, 16.5601, 76.9026, 165.4483, 216.2266, 266.7990, 281.1986, 388.8310, 462.6379, 475.7872, 517.5003, 574.2078, 636.4659, 747.7625, 804.0394, 858.0309, 881.8098, 1156.5163, 1189.1043, 1208.1041, 1220.3110, 1246.2111, 1347.3573, 1438.7791, 1465.4423, 1624.4152, 1668.6753, 1790.7038, 3045.2640, 3141.5807
<p>IRC</p> <p>Relative Energy (kJ mol⁻¹)</p> <p>Reaction Co-ordinates (amu^{1/2} bohr)</p>	

Compound: sCl 2 + HCHO C _{TFA} 1	Energy (kJ mol⁻¹): -640.799194296748
Reaction Coordinates: 6 -1.584811 0.005077 -0.000001 6 -0.039654 0.156627 0.000008 1 1.529652 -0.913874 0.000007 8 0.540782 -1.026051 0.000005 8 3.222675 -0.659518 0.000004 9 -1.992953 -0.669392 -1.087888 9 -1.992964 -0.669470 1.087833 9 -2.176496 1.197977 0.000039 8 0.498638 1.231972 0.000014 6 3.647165 0.473450 -0.000009 1 4.731211 0.662243 -0.000012 1 2.967885 1.337445 -0.000019	Frequencies (cm⁻¹): 25.6546, 47.4526, 85.6725, 89.5655, 138.7878, 212.8202, 239.2894, 272.2450, 275.9329, 398.0737, 430.5026, 516.2915, 588.3812, 699.8772, 781.4965, 807.0049, 948.7741, 1152.7095, 1171.4562, 1205.8043, 1228.0063, 1276.9303, 1322.5883, 1468.8858, 1522.6090, 1770.4123, 1819.6128, 2945.5541, 3040.2391, 3222.9658

Compound: sCl 2 + HCHO TS _{acid} 1	Energy (kJ mol⁻¹): -640.610730401605
Reaction Coordinates: 6 1.170040 -0.019333 0.102506 6 -0.168113 -0.057013 -0.702440 1 0.095200 0.410473 -1.787063 8 -0.833645 1.094628 -0.827126 8 -1.972031 0.668468 0.724614 9 1.933853 -1.065833 -0.239568 9 0.934323 -0.078480 1.416071 9 1.854340 1.102900 -0.157453 8 -0.842555 -1.144501 -0.647246 6 -2.345900 -0.451047 0.296177 1 -2.344359 -1.284700 1.006253 1 -3.003801 -0.537460 -0.570039	Frequencies (cm⁻¹): -1053.5436, 42.5106, 105.4209, 173.6081, 261.0965, 263.0094, 289.2795, 345.4982, 404.0330, 428.2578, 517.2237, 537.4321, 592.1376, 704.0861, 797.1483, 818.3344, 854.2981, 933.9543, 1132.9398, 1155.0775, 1183.1777, 1194.8084, 1196.7304, 1273.4821, 1394.2535, 1446.8064, 1542.9920, 2285.7600, 3003.4676, 3093.2029
<p style="text-align: center;">IRC</p> <p style="text-align: center;">Relative Energy (kJ mol⁻¹)</p> <p style="text-align: center;">Reaction Co-ordinates (amu^{1/2} bohr)</p>	

Compound: sCl 2 + HCHO C _{TFA} 2	Energy (kJ mol⁻¹): -640.787589964388
Reaction Coordinates: 6 1.019299 -0.360811 -0.053058 6 0.148131 0.922496 -0.173713 8 0.046381 1.641340 0.947305 1 0.368706 1.141827 1.710470 8 -0.304737 1.263154 -1.223148 9 2.273904 -0.073288 -0.444518 9 0.557232 -1.344920 -0.812178 9 1.087654 -0.796104 1.225627 8 -2.080165 -0.612552 0.614197 6 -2.894107 -0.391330 -0.244657 1 -2.637248 0.185901 -1.148649 1 -3.932339 -0.756588 -0.160453	Frequencies (cm⁻¹): 43.6805, 46.0995, 76.9727, 88.5261, 90.8361, 152.1740, 182.0909, 250.5153, 254.9155, 383.7001, 426.9844, 506.3784, 531.5898, 586.9922, 689.0061, 762.0313, 796.3404, 1125.2518, 1139.2762, 1210.1226, 1212.7643, 1246.2164, 1270.1461, 1372.0470, 1527.1554, 1799.1525, 1869.7527, 2909.7687, 2982.7885, 3755.1592

Compound: sCl 2 + HCHO TS _d 2	Energy (kJ mol⁻¹): -640.621055369406
Reaction Coordinates: 6 -0.175640 0.329999 -0.502393 1 -0.058554 0.493890 -1.591676 6 1.274700 -0.031924 0.005777 9 1.738207 -1.102156 -0.647013 9 2.091775 0.995729 -0.230496 9 1.276257 -0.294315 1.313791 8 -0.688512 1.334729 0.131328 8 -0.918574 -0.908852 -0.312853 6 -2.105447 -0.552757 0.321201 1 -2.608572 -1.415811 0.787748 8 -2.842231 0.328488 -0.233668 1 -1.656171 0.021768 1.251429	Frequencies (cm⁻¹): -858.3901, 83.4451, 100.7020, 167.0216, 224.4556, 310.1539, 342.4178, 408.5183, 472.7327, 531.3920, 566.5172, 584.1015, 720.7180, 750.4830, 808.8318, 848.8824, 940.9329, 988.9430, 1105.4595, 1136.6271, 1147.7638, 1181.9117, 1219.2248, 1243.8147, 1252.5114, 1271.6928, 1375.0411, 2178.8550, 2883.0314, 2954.8922
IRC 	

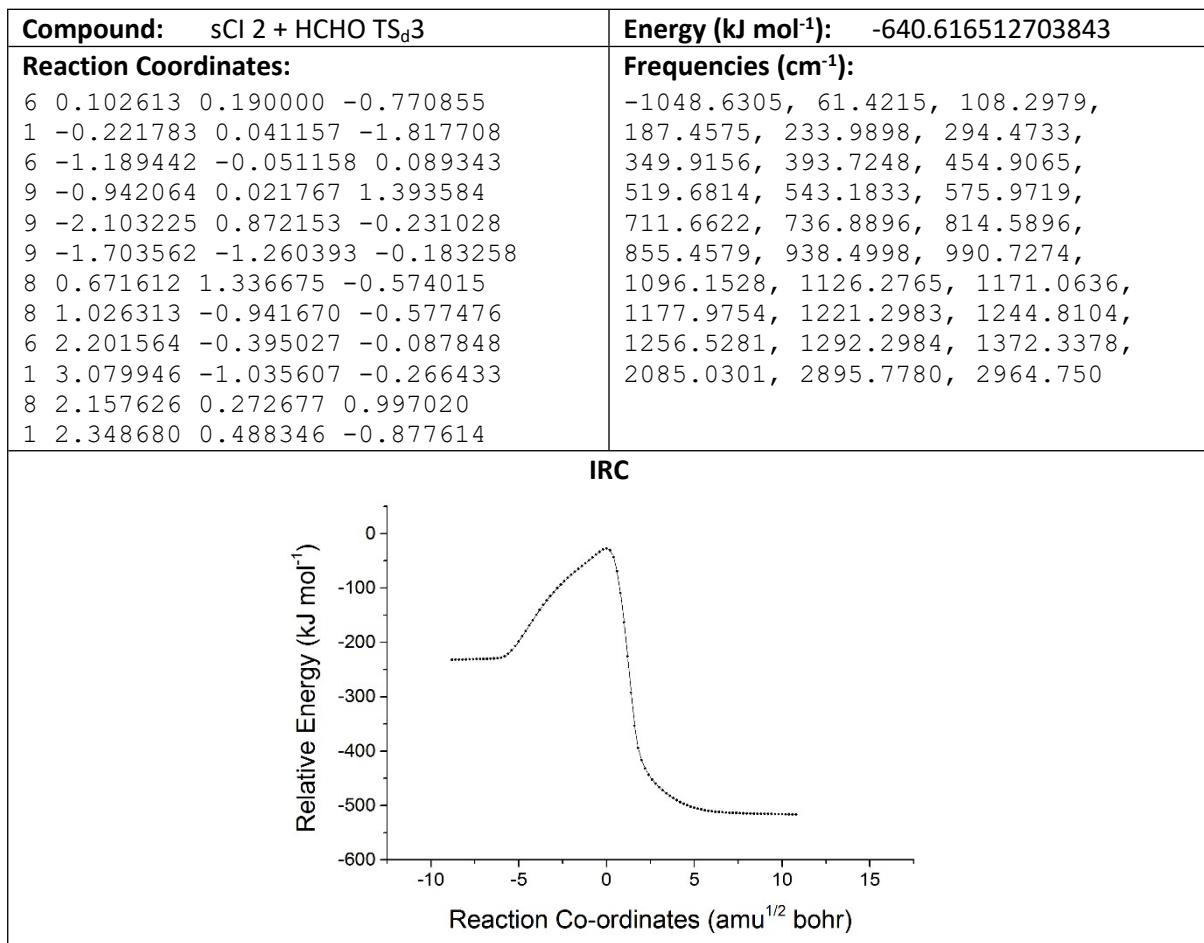
Compound: sCl 2 + HCHO HAE3	Energy (kJ mol⁻¹): -640.813164616868
Reaction Coordinates: 6 -0.189522 0.476776 -0.240249 1 -0.577133 0.286665 -1.237829 6 1.154475 -0.250359 -0.066579 9 1.008237 -1.579445 -0.146258 9 2.013485 0.132641 -1.021670 9 1.710486 0.040828 1.127511 8 -0.061070 1.841415 -0.086936 1 0.325339 2.036387 0.776578 8 -1.067326 -0.079665 0.744616 6 -2.334230 -0.377776 0.357102 1 -2.884504 -0.748554 1.228277 8 -2.776342 -0.263320 -0.745795	Frequencies (cm⁻¹): 32.0458, 66.2195, 177.5694, 205.0507, 255.5542, 262.8436, 365.1985, 385.3922, 425.7036, 513.8521, 562.9730, 589.7024, 680.8692, 764.7864, 881.3630, , 959.4360, 1038.3529, 1099.5497, 1146.6269, 1158.8844, 1191.1342, 1265.9866, 1318.5849, 1370.9634, 1396.6574, 1449.4686, 1809.0497, 3057.9653, 3129.6457, 3768.2098

Compound: sCl 2 + HCHO TS _{ISO} 2	Energy (kJ mol⁻¹): -640.807843829511
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -0.158204 0.531709 -0.275737 1 -0.562091 0.254605 -1.249995 6 1.125250 -0.280012 -0.029274 9 0.857357 -1.595448 -0.127519 9 2.045912 0.026063 -0.954538 9 1.649418 -0.050709 1.179213 8 0.151353 1.885728 -0.204465 1 -0.525151 2.338407 0.308813 8 -1.086252 0.178143 0.746633 6 -2.303982 -0.305716 0.373145 1 -2.874678 -0.526092 1.281221 8 -2.688933 -0.459116 -0.745074	-339.0386, 21.8885, 66.0356, 174.9537, 209.7879, 258.4835, 271.7906, 375.9960, 408.9945, 504.7021, 546.2406, 581.1835, 686.3237, 766.7458, 880.0458, 961.8218, 1037.9249, 1108.4232, 1140.8063, 1167.3180, 1179.8049, 1245.5598, 1311.0747, 1381.8595, 1398.2379, 1447.5863, 1812.9194, 3061.1220, 3082.6758, 3827.1680

Compound: sCl 2 + HCHO HAE5	Energy (kJ mol⁻¹): -640.814946031998
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -0.210661 0.309903 -0.247637 1 -0.430875 0.333422 -1.316440 6 1.255880 -0.101420 -0.059852 9 1.476777 -1.299065 -0.627801 9 2.055084 0.793281 -0.660280 9 1.600789 -0.177441 1.228334 8 -0.407314 1.523418 0.368428 1 -1.252566 1.879199 0.066418 8 -0.991056 -0.750088 0.345904 6 -2.314706 -0.652497 0.145899 1 -2.822176 -1.505794 0.607070 8 -2.860544 0.239956 -0.445556	67.4596, 75.6599, 159.7185, 229.0168, 251.3286, 282.3992, 355.7306, 403.4892, 457.7356, 527.4247, 560.9981, 577.8060, 696.8742, 787.0476, 887.3511, 953.7355, 1043.7123, 1129.5545, 1144.6425, 1176.4830, 1194.2838, 1253.9800, 1297.3140, 1377.7165, 1392.9043, 1480.8655, 1775.7436, 3064.0576, 3066.2607, 3773.3065

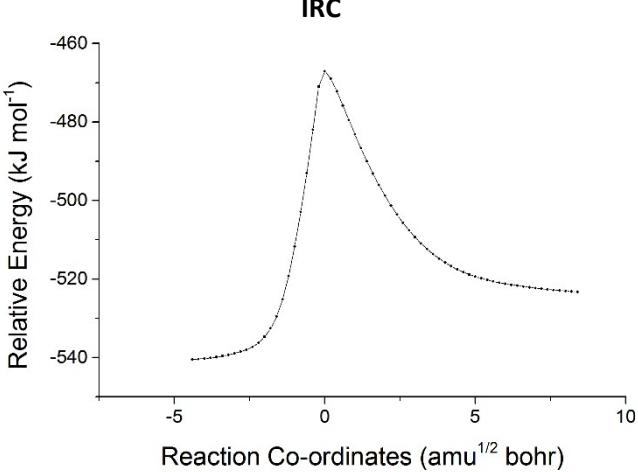
Compound: sCl 2 + HCHO TS _{ISO} 3	Energy (kJ mol⁻¹): -640.804184792221
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -0.073278 0.304727 -0.670768 1 0.145021 0.312637 -1.742577 6 1.228980 -0.082193 0.065831 9 1.719098 -1.237879 -0.414811 9 2.152502 0.869287 -0.121552 9 1.029377 -0.227967 1.382204 8 -0.506133 1.534381 -0.234289 1 -1.419321 1.457247 0.096643 8 -0.969506 -0.814841 -0.472175 6 -2.201526 -0.704448 0.034236 1 -2.648110 -1.702810 0.087808 8 -2.763291 0.304641 0.377683	-120.2227, 72.1141, 163.1147, 205.5213, 279.6381, 290.2422, 338.0677, 391.7004, 464.5542, 508.5861, 544.2558, 572.1653, 707.8013, 809.6028, 847.6336, 904.6825, 1041.7221, 1151.3840, 1158.4978, 1180.9676, 1190.4618, 1269.3497, 1350.4665, 1397.6865, 1413.3643, 1458.0097, 1760.0678, 3036.0700, 3060.7974, 3562.0744

Compound: sCl 2 + HCHO HAE6	Energy (kJ mol⁻¹): -640.810954254638
Reaction Coordinates: 6 -0.023422 0.519662 -0.759457 1 0.400592 0.711002 -1.742809 6 1.089680 -0.149560 0.081852 9 1.464574 -1.316150 -0.468800 9 2.162028 0.656306 0.113136 9 0.712418 -0.388332 1.346300 8 -0.451637 1.712395 -0.239187 8 -1.072358 -0.448844 -1.004296 6 -2.066917 -0.604329 -0.113028 1 -2.765359 -1.357094 -0.494272 8 -2.195222 -0.019867 0.928208 1 -0.928721 1.535550 0.587345	Frequencies (cm⁻¹): 41.6872, 104.5853, 181.4384, 230.9965, 246.0566, 278.7974, 363.6505, 415.0763, 499.4583, 530.9290, 571.2800, 580.2686, 703.3542, 796.4718, 852.6964, 921.4691, 1041.5112, 1135.0504, 1154.7981, 1157.9586, 1177.1062, 1277.2267, 1350.7516, 1389.9819, 1406.1744, 1453.7478, 1778.4832, 3055.5168, 3113.8394, 3681.8234



Compound: sCl 2 + HCHO HAE4	Energy (kJ mol⁻¹): -640.802893876772
Reaction Coordinates: 6 0.003320 0.869526 -0.446611 1 -0.448004 1.351622 -1.313680 6 -0.990751 -0.244810 -0.023382 9 -0.907543 -0.608257 1.249326 9 -2.241268 0.226824 -0.231581 9 -0.836126 -1.331722 -0.797407 8 0.156092 1.730646 0.621894 1 0.442813 2.592697 0.303624 8 1.246046 0.388794 -0.977262 6 2.085126 -0.380418 -0.232513 1 2.989831 -0.555784 -0.827072 8 1.884565 -0.798931 0.864007	Frequencies (cm⁻¹): 45.2715, 100.4461, 159.7963, 195.2888, 256.9147, 262.5438, 317.4227, 378.9325, 414.6407, 510.0905, 550.3725, 586.8127, 671.2922, 776.2965, 868.9183, 962.7327, 1037.1402, 1097.2875, 1122.7443, 1153.0702, 1173.1465, 1241.1347, 1311.4807, 1397.6599, 1415.5469, 1461.4654, 1825.1669, 3037.6121, 3078.2371, 3813.5831

Compound: sCl 2 + HCHO TS _{ISO} 4	Energy (kJ mol⁻¹): -640.799602135995
Reaction Coordinates: 6 -0.008981 0.703216 -0.618432 1 0.454897 1.055477 -1.544508 6 1.044954 -0.226075 0.029517 9 1.450361 -1.148558 -0.855586 9 2.120492 0.537396 0.353339 9 0.648222 -0.849616 1.135229 8 -0.386453 1.757944 0.201981 1 0.360995 2.058173 0.730998 8 -1.152087 0.002061 -1.068513 6 -2.125534 -0.449922 -0.219502 1 -2.923603 -0.860758 -0.849254 8 -2.127285 -0.418657 0.968585	Frequencies (cm⁻¹): -205.8349, 29.6258, 111.6347, 163.0098, 198.8716, 233.8819, 261.6935, 365.4356, 414.8042, 501.7893, 544.4509, 579.2414, 686.6904, 777.6938, 869.5887, 961.1872, 1032.4184, 1105.0513, 1120.2350, 1153.8295, 1180.0968, 1249.2081, 1329.0742, 1390.0135, 1408.1359, 1439.4732, 1830.9429, 3032.2077, 3040.2313, 3805.9390

Compound: sCl 2 + HCHO TS _{HAE} 2	Energy (kJ mol⁻¹): -640.777964091842
Reaction Coordinates: 6 -1.152459 0.080578 0.088526 6 -0.033283 -0.462620 -0.831746 1 -0.299276 -0.321810 -1.883881 8 0.573899 -1.523208 -0.483750 8 2.298411 -0.380018 0.678641 9 -1.608559 1.271322 -0.327484 9 -0.751636 0.201253 1.362196 9 -2.178307 -0.787312 0.053382 8 1.065071 0.900315 -0.699501 6 2.074660 0.682226 0.028917 1 2.814064 1.484922 0.095223 1 1.499183 -1.108300 0.318517	Frequencies (cm⁻¹): -968.8390, 66.1525, 81.9851, 177.5996, 246.6260, 280.8294, 330.3906, 389.7532, 484.2083, 522.6232, 557.7775, 587.2552, 662.2032, 745.8993, 846.9394, 864.7260, 1062.9738, 1129.9689, 1171.0954, 1194.3591, 1240.7771, 1274.2240, 1359.3244, 1379.3200, 1386.7151, 1574.1782, 1699.5035, 1801.5823, 3041.8683, 3078.8532
IRC  Relative Energy (kJ mol ⁻¹) Reaction Co-ordinates (amu ^{1/2} bohr)	

Compound: sCl 2 + HCHO C _{FAc} 1	Energy (kJ mol⁻¹): -640.799644233095
Reaction Coordinates: 6 1.829574 -0.075226 0.000000 6 0.291807 0.035526 -0.000003 8 -0.263422 1.100263 -0.000005 1 -0.249851 -0.921563 -0.000003 9 2.218400 -0.762929 -1.089059 9 2.432490 1.107775 -0.000003 9 2.218396 -0.762921 1.089067 1 -2.123558 1.071582 -0.000002 8 -3.099430 0.950319 0.000002 6 -3.378418 -0.349903 0.000002 1 -4.462694 -0.509946 0.000007 8 -2.567803 -1.243054 -0.000003	Frequencies (cm⁻¹): 28.8611, 44.0666, 71.4606, 106.0397, 119.5735, 145.4401, 190.5235, 279.0593, 330.0197, 433.6041, 524.6309, 528.5168, 671.6616, 704.5942, 839.6375, 867.8719, 1012.6684, 1066.0493, 1162.4944, 1180.2903, 1189.8676, 1287.0851, 1365.2464, 1402.9839, 1421.8376, 1773.3347, 1819.2331, 3016.2653, 3046.6481, 3453.4462

Compound: sCl 2 + HCHO TS _{F_{Ac}} 1	Energy (kJ mol⁻¹): -640.612916191139
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -1.240405 -0.026036 0.011886 6 0.135836 0.407482 -0.547764 1 0.137618 0.408999 -1.642472 8 0.718010 1.333292 0.105212 8 2.531811 0.535366 -0.129761 9 -1.702755 -1.106895 -0.631309 9 -2.115923 0.974405 -0.195571 9 -1.203698 -0.285322 1.319510 8 0.991060 -1.056635 -0.286017 6 2.064870 -0.626086 0.301543 1 3.149347 -0.774754 -0.253715 1 2.225567 -0.902282 1.353046	-1193.8039, 72.1271, 92.5885, 172.7720, 246.7119, 259.0645, 304.4696, 372.7167, 439.4271, 457.7900, 523.2859, 566.3051, 662.7034, 705.8703, 756.4539, 845.5288, 868.5412, 1116.5453, 1148.5230, 1163.0003, 1187.5554, 1221.3952, 1249.2538, 1315.7469, 1334.6176, 1406.7356, 1461.4554, 2235.1404, 2988.7376, 3028.7966
IRC	
<p>The plot displays the reaction coordinate (IRC) for the reaction. The y-axis represents the relative energy in kJ mol⁻¹, ranging from 0 to -500. The x-axis represents the reaction coordinates in amu^{1/2} bohr, ranging from -5 to 10. A single peak is observed, starting at approximately -250 kJ mol⁻¹ at -5 amu^{1/2} bohr, rising to a maximum of about -120 kJ mol⁻¹ at 0 amu^{1/2} bohr, and then gradually decreasing towards -500 kJ mol⁻¹ at 7 amu^{1/2} bohr.</p>	

Compound: sCl 2 + HCHO C _{F_{Ac}} 2	Energy (kJ mol⁻¹): -640.788851617549
Reaction Coordinates:	Frequencies (cm⁻¹):
6 1.761049 -0.285003 0.062007 6 0.767135 0.875943 -0.175926 1 -0.257265 0.540397 -0.401807 8 1.105685 2.020870 -0.112233 9 1.319382 -1.051507 1.080043 9 2.990040 0.134105 0.349675 9 1.818020 -1.060276 -1.039329 8 -2.419881 -0.169238 -0.879313 6 -3.495306 -0.228361 -0.363768 8 -3.693042 0.103780 0.926100 1 -4.622004 -0.005565 1.165569 1 -4.407068 -0.554496 -0.887567	7.7036, 15.0143, 16.8036, 26.8905, 62.3992, 79.8049, 94.8453, 252.0524, 314.0173, 421.4039, 518.3137, 525.3090, 541.7266, 664.0237, 700.2285, 829.9758, 995.0095, 1037.5429, 1118.4110, 1152.2179, 1173.2951, 1271.6813, 1282.9539, 1414.6775, 1423.2612, 1831.9042, 1845.8547, 2982.1286, 2998.3238, 3779.6741

Compound: sCl 2 + HCHO TS _{FAC} 2	Energy (kJ mol⁻¹): -640.615657587046
Reaction Coordinates:	Frequencies (cm⁻¹):
6 1.147981 -0.047437 0.071635 6 -0.076256 0.257681 -0.819982 1 0.131984 -0.008703 -1.863370 8 -0.731916 1.334815 -0.635515 8 -1.928481 0.546508 0.772565 9 1.667176 -1.252307 -0.222732 9 0.876371 -0.014318 1.378858 9 2.094288 0.876690 -0.181581 8 -1.080636 -1.067320 -0.500936 6 -2.141677 -0.496699 -0.009589 1 -2.458690 -0.736594 1.153285 1 -3.065839 -0.538569 -0.602108	-1223.3404, 55.2869, 120.1231, 182.4610, 242.8627, 279.4588, 308.6175, 372.9038, 441.0465, 453.2497, 518.5714, 571.7535, 649.2592, 714.2083, 781.7622, 841.9288, 847.1269, 1080.7267, 1144.2891, 1156.3786, 1162.0579, 1222.3706, 1266.8167, 1291.5830, 1322.0916, 1407.5412, 1473.0139, 2243.0956, 2991.9693, 3009.3838
	IRC 

Compound: SCI 2 + HCHO TS _{ester1}	Energy (kJ mol⁻¹): -640.591726379101
Reaction Coordinates:	Frequencies (cm⁻¹):
6 1.366331 -0.026686 0.061260 6 -0.368759 0.156879 -0.654823 1 -0.095006 0.135080 -1.722017 8 -0.528243 -0.990651 -0.039830 8 -2.562658 -0.772364 -0.186019 9 1.910254 1.083396 -0.410810 9 1.342257 -0.001711 1.364600 9 2.005600 -1.065334 -0.417362 6 -2.540376 0.329368 0.410312 8 -0.975599 1.192560 -0.211818 1 -2.239606 0.402513 1.457551 1 -3.199567 1.121523 0.037456	-560.5388, 50.0694, 60.3894, 102.9234, 178.5244, 242.8272, 256.1040, 306.1896, 352.1444, 491.3535, 533.1564, 542.4147, 573.0182, 711.7424, 738.2562, 847.6028, 946.0872, 1055.5919, 1141.1780, 1188.2539, 1222.2839, 1262.1485, 1301.4676, 1344.3947, 1391.0679, 1441.0526, 1534.6863, 2949.1530, 2992.2289, 3074.1389

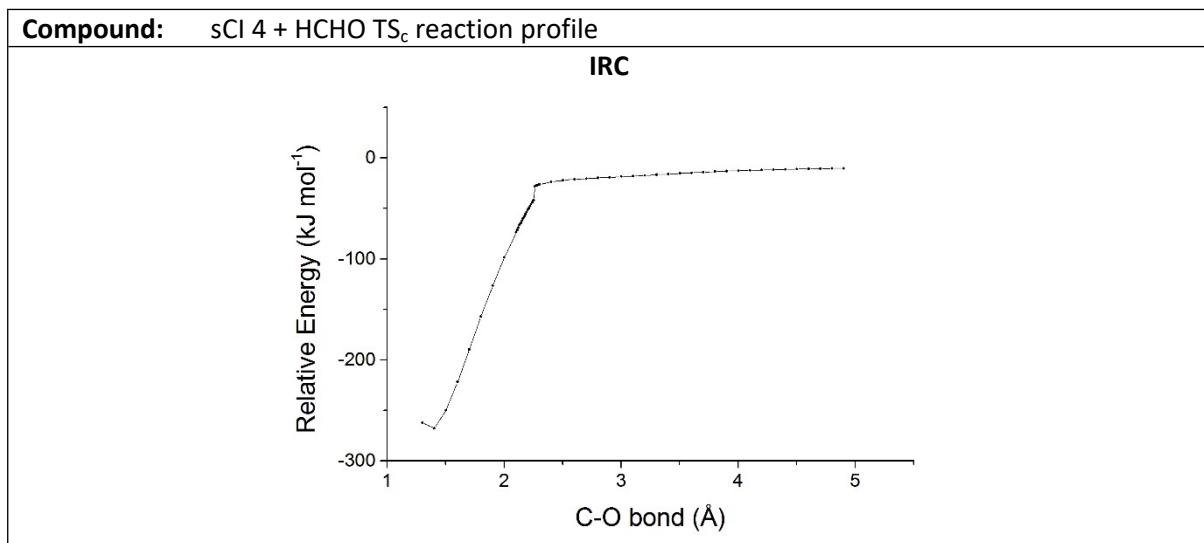
Compound: sCI 2 + HCHO C _{ester} 1	Energy (kJ mol⁻¹): -640.785087532247
Reaction Coordinates:	Frequencies (cm⁻¹):
6 1.791858 0.065785 -0.000001	17.6764, 39.3605, 53.9006, 79.8186,
8 0.650662 0.840112 -0.000017	93.9123, 111.7895, 112.6025,
6 -0.593997 0.236702 -0.000007	188.4592, 231.1980, 373.5513,
1 -1.352802 1.022550 -0.000021	441.8426, 547.8735, 581.2838,
8 -0.791968 -0.935019 0.000015	619.1427, 818.3698, 840.7698,
9 1.871240 -0.706996 -1.084234	1056.6360, 1094.8676, 1160.5880,
9 2.825281 0.904836 -0.000020	1205.7308, 1212.1275, 1237.8546,
9 1.871241 -0.706947 1.084267	1270.3234, 1396.9949, 1529.3276,
6 -4.178424 -0.338090 -0.000012	1796.6426, 1829.5309, 2903.8149,
8 -3.745980 0.785560 0.000010	2979.3096, 3101.5598
1 -3.510320 -1.215360 -0.000071	
1 -5.265067 -0.536821 0.000027	

Compound: sCl 2 + HCHO TS _{ester} 2	Energy (kJ mol⁻¹): -640.587782961932
Reaction Coordinates:	Frequencies (cm⁻¹):
6 1.378359 0.010001 0.025154 6 -0.332347 -0.141342 -0.462417 1 -0.200721 -0.278102 -1.556026 8 -0.824975 -1.139154 0.189235 8 -2.605340 0.761834 0.178537 9 1.471795 0.194651 1.310528 9 1.966170 0.961699 -0.661715 9 1.884549 -1.158456 -0.319820 6 -2.709063 -0.441339 -0.112929 8 -0.558469 1.084594 -0.065633 1 -3.097189 -1.143606 0.633883 1 -2.716139 -0.781448 -1.154760	-462.6014, 72.7718, 82.3414, 102.8373, 171.3199, 249.8559, 274.1585, 312.0720, 324.4221, 443.0121, 521.3093, 547.7848, 576.1725, 709.5628, 721.9533, 744.0114, 933.7447, 1091.5664, 1150.9995, 1197.9039, 1221.0053, 1231.3293, 1303.8615, 1330.7661, 1377.2627, 1441.3485, 1573.5282, 2855.4590, 2970.3282, 3047.9176
IRC	
<p>The plot displays the relative energy of the system as it progresses along the reaction coordinate. The y-axis represents the relative energy in kJ mol⁻¹, ranging from -500 to 0. The x-axis represents the reaction coordinate in amu^{1/2} bohr, ranging from -5 to 5. The curve starts at approximately -200 kJ mol⁻¹ at x = -5, remains relatively flat until x = 0, where it reaches a sharp peak of about 0 kJ mol⁻¹, and then rapidly decreases towards -500 kJ mol⁻¹ as the reaction coordinate increases.</p>	

Compound: CF ₃ COOH conformer 2	Energy (kJ mol⁻¹): -526.391647939292
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -0.581404 -0.010401 -0.000000 6 0.970727 -0.156509 -0.000004 8 1.477110 -1.232961 0.000000 8 1.654341 0.993148 -0.000001 1 1.057725 1.754791 -0.000017 9 -1.102989 -0.583142 -1.084628 9 -1.102975 -0.583095 1.084656 9 -0.954622 1.295701 -0.000023	21.6602, 251.5689, 256.7023, 383.9419, 424.0892, 504.2317, 558.0545, 581.8488, 686.6362, 774.0185, 789.5230, 1111.2671, 1173.1962, 1190.7699, 1238.1250, 1367.2329, ,1882.0972, 3753.0385

Compound: CF ₃ CHO	Energy (kJ mol⁻¹): -451.199957828211
Reaction Coordinates:	Frequencies (cm⁻¹):
6 0.015009 0.360282 0.000000 6 0.501814 -1.106406 -0.000000 8 -0.245475 -2.035870 -0.000000 1 1.601864 -1.198064 -0.000000 9 0.501814 0.987869 1.088491 9 0.501814 0.987869 -1.088491 9 -1.307962 0.464459 0.000000	73.5798, 250.7650, 307.2813, 422.6928, 518.9491, 524.0490, 701.8523, 834.5377, 974.2730, 1150.4161, 1174.8562, 1288.1325, 1403.3576, 1849.8517, 2955.8187

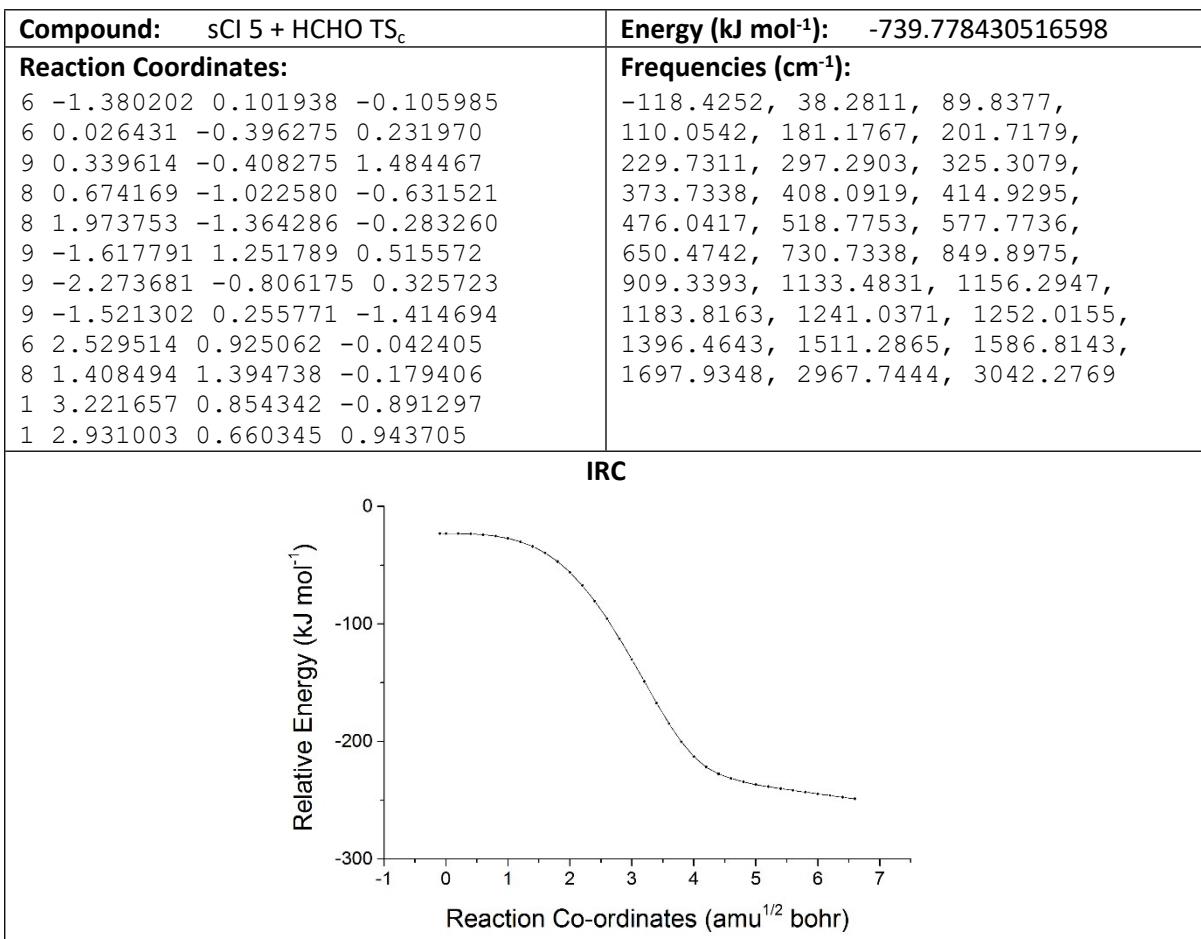
Compound: CF ₃ OCHO	Energy (kJ mol⁻¹): -526.392169685219
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -0.663849 -0.000032 0.000000	88.2784, 182.4643, 213.8832,
8 0.336896 0.952890 -0.000000	372.9073, 440.5306, 548.4446,
6 1.661196 0.558397 0.000000	580.4343, 618.6761, 819.4841,
8 2.058027 -0.558385 0.000000	841.6604, 1029.7431, , 1099.7328,
1 2.265245 1.470480 0.000000	1158.7538, 1210.5947, 1240.5018,
9 -0.613300 -0.773286 1.084140	1402.5579, 1846.9478, 3073.3668
9 -0.613300 -0.773286 -1.084140	
9 -1.818813 0.660272 0.000000	



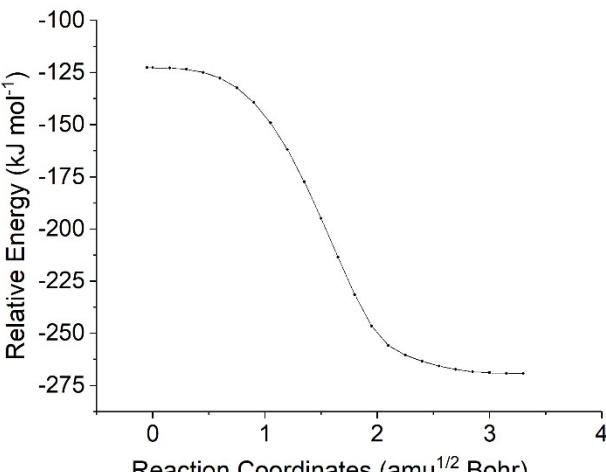
Compound: sCl 4 + HCHO HOZ 1	Energy (kJ mol⁻¹): -739.880971561098
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -1.203229 -0.230964 0.058463	62.9168, 82.3412, 181.5539,
6 0.191425 0.440783 -0.055823	225.3051, 306.3116, 333.8341,
9 -0.018442 1.768072 -0.284796	372.3751, 383.1249, 472.7730,
8 0.923264 -0.051022 -1.159256	537.6227, 578.2342, 616.4528,
8 2.011888 -0.814233 -0.568390	733.4968, 748.0603, 840.5352,
9 -1.873473 0.255806 1.109021	900.2599, 961.3323, 1051.7447,
9 -1.065362 -1.550484 0.213301	1075.2302, 1084.3749, 1160.0607,
9 -1.918007 -0.002321 -1.047544	1174.0468, 1194.6284, 1204.5660,
8 0.930952 0.217053 1.072410	1240.5871, 1353.3296, 1400.3061,
6 2.246328 -0.074884 0.601161	1519.3524, 3039.2896, 3147.1681
1 2.755836 -0.715948 1.315224	
1 2.785743 0.852302 0.394021	

Compound: sCl 5 + HCHO PRC	Energy (kJ mol⁻¹): -739.777926184174
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -1.414853 0.141074 -0.112497	34.8048, 60.5055, 70.4194,
6 -0.050978 -0.450704 0.236511	101.2900, 143.9768, 186.7900,
9 0.273257 -0.457980 1.482950	200.3901, 237.6640, 298.5269,
8 0.616211 -1.031007 -0.636363	339.1001, 362.6188, 384.0722,
8 1.855522 -1.524854 -0.274925	406.9206, 514.5633, 577.3863,
9 -1.545297 1.342529 0.441524	663.2692, 730.6738, 857.5480,
9 -2.373565 -0.660000 0.384471	900.9980, 1138.3690, 1178.3830,
9 -1.564546 0.229935 -1.426569	1181.7442, 1238.6573, 1260.7686,

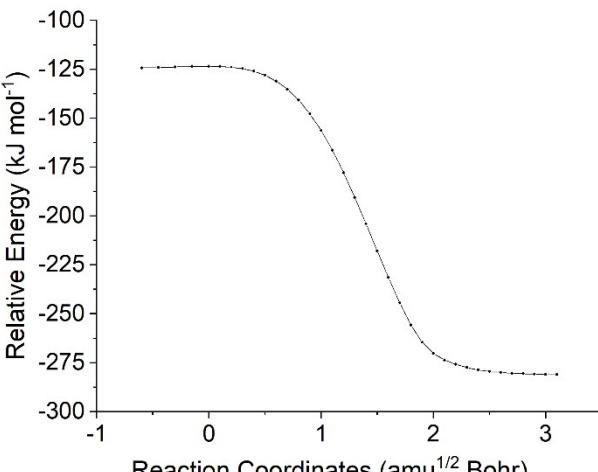
6 2.716861 0.890240 -0.070701 8 1.640783 1.447690 -0.108533 1 3.315311 0.728201 -0.979729 1 3.169732 0.563154 0.877031	1402.6054, 1526.8058, 1612.0198, 1752.6652, 2940.4081, 3008.9838
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Compound:	sCl 1 + CF ₃ CFO PRC 3	Energy (kJ mol ⁻¹):	-739.813259682201
Reaction Coordinates:		Frequencies (cm⁻¹):	
6 1.276276 -0.317441 0.042768 6 0.199237 0.765686 0.278781 9 0.332320 1.754487 -0.609756 8 -0.466774 0.863601 1.257424 9 1.452943 -0.610239 -1.240654 9 2.442962 0.159259 0.528183 9 0.975427 -1.429876 0.713343 6 -2.711550 -0.451865 0.655072 1 -2.030034 -1.117147 1.167434 1 -3.638928 -0.080588 1.070988 8 -2.468023 -0.081048 -0.512669 8 -1.283662 -0.487955 -1.072029	42.13, 63.96, 88.06, 113.40, 161.90, 192.84, 220.11, 240.50, 279.50, 379.83, 428.50, 513.34, 528.59, 586.64, 680.60, 693.20, 716.22, 802.10, 885.19, 1024.72, 1084.45, 1147.22, 1223.53, 1243.79, 1311.45, 1417.81, 1571.35, 1851.85, 3127.16, 3272.45		

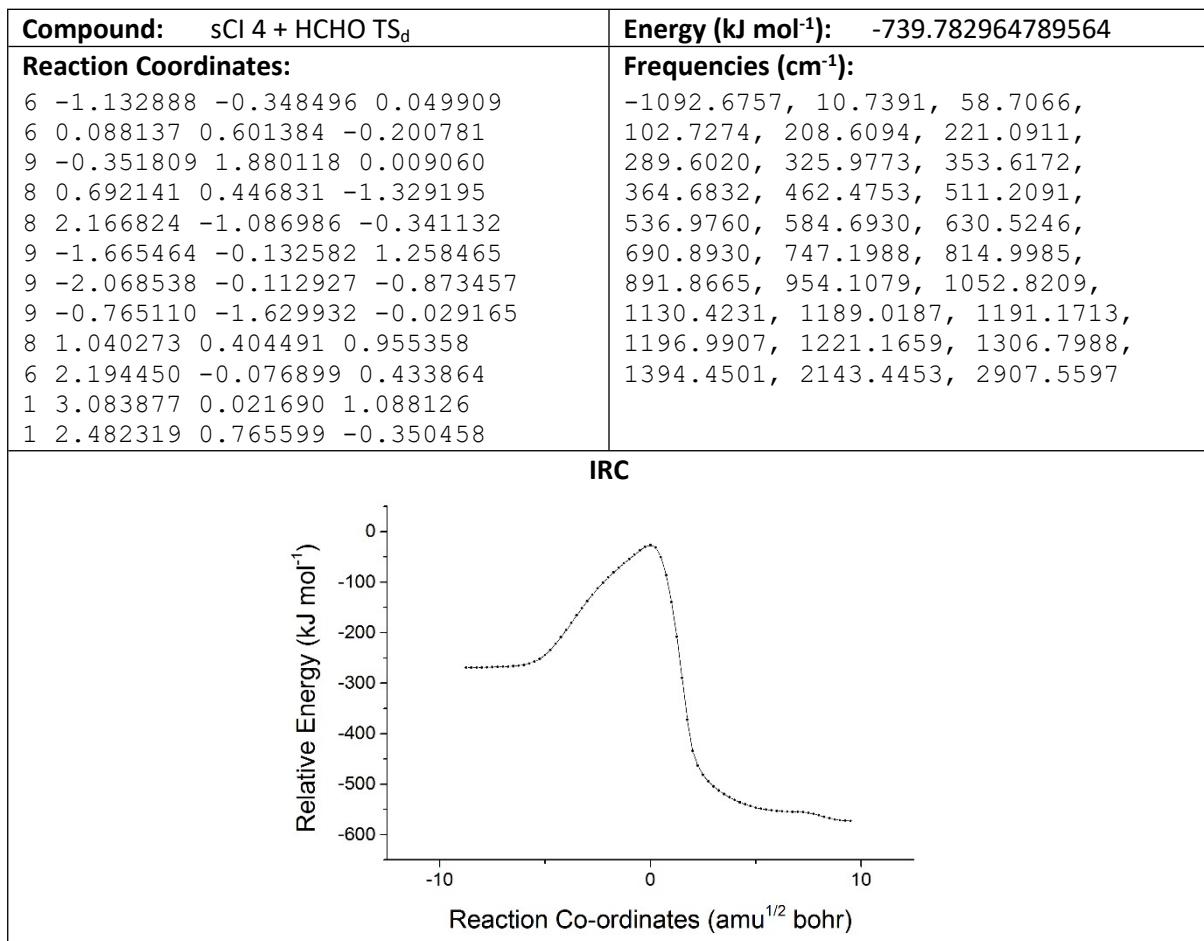
Compound: sCl 1 + HCHO TS _c 3	Energy (kJ mol⁻¹): -739.811613027458
Reaction Coordinates: 6 1.245988 -0.265523 -0.092261 6 0.041332 0.700989 -0.145855 9 0.189726 1.675069 0.774721 8 -0.632687 0.863682 -1.129472 9 0.969226 -1.399460 -0.736156 9 2.281569 0.329362 -0.721769 9 1.631765 -0.547889 1.150351 6 -2.741525 0.035486 -0.376479 1 -2.703817 1.033485 0.033747 1 -3.404323 -0.266334 -1.177649 8 -2.081522 -0.894047 0.143083 8 -1.137941 -0.482961 1.064781	Frequencies (cm⁻¹): -109.92, 56.94, 80.14, 142.31, 211.44, 220.36, 269.82, 308.99, 347.33, 383.64, 438.27, 518.42, 539.95, 587.75, 681.43, 700.45, 719.16, 802.82, 898.27, 1048.01, 1067.15, 1146.40, 1218.82, 1236.65, 1296.61, 1415.35, 1570.32, 1749.53, 3135.01, 3279.84
IRC  <p>Relative Energy (kJ mol⁻¹)</p> <p>Reaction Coordinates (amu^{1/2} Bohr)</p>	

Compound: sCl 1 + CF ₃ CFO PRC 4	Energy (kJ mol⁻¹): -739.812611750193
Reaction Coordinates: 6 1.260836 -0.292596 -0.100143 6 0.149504 0.780864 -0.149444 9 0.297955 1.652770 0.857863 8 -0.528324 1.011186 -1.101148 9 0.919770 -1.358554 -0.820669 9 2.367461 0.250369 -0.655394 9 1.568501 -0.670948 1.136250 6 -2.843813 0.077420 -0.381769 1 -2.757537 1.070839 0.035688 1 -3.523078 -0.196672 -1.178448 8 -2.165570 -0.862809 0.084248 8 -1.243822 -0.539755 1.050455	Frequencies (cm⁻¹): 47.51, 65.26, 85.39, 118.69, 174.40, 200.92, 245.37, 260.90, 295.89, 379.05, 431.01, 513.18, 531.84, 586.90, 683.81, 689.53, 708.55, 800.11, 888.72, 1029.39, 1075.49, 1142.75, 1227.75, 1241.45, 1305.29, 1418.19, 1572.94, 1830.56, 3129.42, 3274.44

Compound: sCl 5 + HCHO TS _c 4	Energy (kJ mol⁻¹): -739.810121917775
Reaction Coordinates: 6 -1.248348 -0.270044 0.006638 6 -0.008169 0.583650 -0.339239 9 -0.144618 1.819566 0.181718 8 0.652411 0.405323 -1.339539 9 -1.571793 -0.215886 1.298125 9 -2.295289 0.186594 -0.706355 9 -1.043079 -1.547385 -0.334192 6 2.510833 -0.639402 -0.503616 1 1.860205 -1.459838 -0.765152 1 3.419149 -0.395132 -1.040673 8 2.340293 0.000196 0.562465 8 1.093266 -0.202550 1.135756	Frequencies (cm⁻¹): -183.41, 56.54, 69.23, 156.11, 217.15, 232.18, 267.46, 317.85, 359.89, 391.03, 438.72, 523.41, 544.08, 588.04, 680.37, 717.37, 735.62, 813.89, 897.93, 1063.92, 1073.84, 1155.76, 1205.38, 1234.21, 1299.76, 1411.54, 1567.26, 1700.60, 3135.05, 3279.77
IRC  Relative Energy (kJ mol ⁻¹) -100 -125 -150 -175 -200 -225 -250 -275 -300 -1 0 1 2 3 Reaction Coordinates (amu ^{1/2} Bohr)	

Compound: sCl 4 + HCHO HOZ 2	Energy (kJ mol⁻¹): -739.882612577417
Reaction Coordinates: 6 1.226910 -0.202398 -0.060082 6 -0.201622 0.381734 0.127609 9 -0.062041 1.673083 0.536046 8 -0.874081 0.296507 -1.089563 8 -2.266214 0.162553 -0.664667 9 1.914395 -0.099267 1.079534 9 1.884341 0.455130 -1.019776 9 1.155217 -1.494768 -0.399219 6 -2.112229 -0.756332 0.387029 8 -0.929811 -0.323497 1.059019 1 -1.972248 -1.775555 0.020385 1 -2.972454 -0.654571 1.044696	Frequencies (cm⁻¹): 64.4536, 102.3338, 183.0668, 231.2002, 294.0750, 331.7999, 368.8521, 380.9341, 498.1395, 541.8807, 578.3703, 614.4270, 736.4144, 747.4253, 853.1992, 888.9747, 962.5108, 1062.4176, 1072.9408, 1103.2970, 1140.8250, 1158.2995, 1199.1282, 1201.9356, 1230.5616, 1346.8978, 1405.6107, 1520.8092, 3040.2533, 3134.6001

Compound: sCl 4 + HCHO TS _{HOZ}	Energy (kJ mol⁻¹): -739.875860313659
Reaction Coordinates:	Frequencies (cm⁻¹):
6 1.265369 -0.175561 0.037602 6 -0.208538 0.310764 -0.045038 9 -0.184141 1.676299 -0.198393 8 -0.829394 -0.290458 -1.120223 8 -2.230537 -0.415601 -0.725606 9 1.884369 0.424204 1.057845 9 1.910458 0.124204 -1.093056 9 1.313647 -1.495448 0.223619 8 -0.898614 -0.037440 1.082003 6 -2.251377 -0.184335 0.692050 1 -2.668140 -1.050180 1.204313 1 -2.815225 0.729644 0.888491	-189.6719, 62.0184, 109.2801, 182.8716, 230.0115, 331.1898, 348.7338, 374.9499, 499.8881, 542.7314, 582.2789, 610.0967, 746.4942, 754.1945, 869.7451, 901.9045, 968.8900, 1048.1702, 1086.4867, 1121.6619, 1139.7391, 1146.8914, 1196.4040, 1206.6516, 1208.5915, 1362.0792, 1415.8068, 1546.9949, 3037.1675, 3106.9218



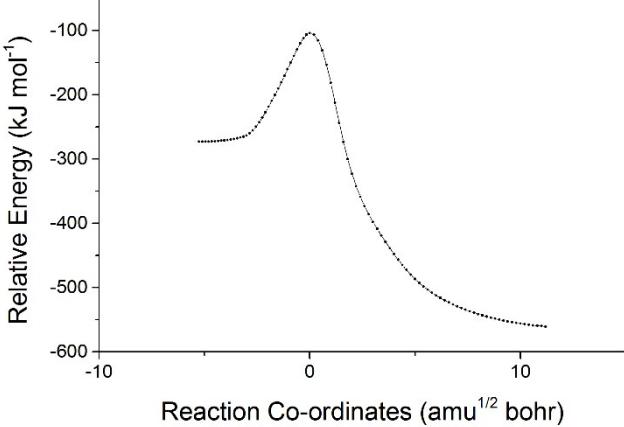
Compound: sCl 4 + HCHO HAE Con 1	Energy (kJ mol⁻¹): -739.996820249036
Reaction Coordinates: 6 -0.027344 0.687582 0.061851 9 0.541121 1.825651 -0.379093 6 1.069431 -0.424089 -0.008015 9 1.545648 -0.560058 -1.249390 9 0.546169 -1.601185 0.374737 9 2.079687 -0.122821 0.807936 8 -0.492323 0.894840 1.311132 1 -1.196959 0.244012 1.495796 8 -1.029442 0.369800 -0.917077 6 -2.150050 -0.303772 -0.587385 1 -2.754555 -0.439892 -1.488583 8 -2.455027 -0.694230 0.506741	Frequencies (cm⁻¹): 57.1808, 70.9230, 175.4045, 217.9644, 261.4777, 273.1843, 344.6606, 348.2922, 389.4253, 465.3067, 496.8327, 576.1035, 582.0878, 591.6309, 692.0873, 767.3125, 831.8341, 957.2695, 1041.7683, 1089.3072, 1164.7213, 1187.9694, 1210.3953, 1227.3088, 1340.8463, 1404.0025, 1408.7922, 1775.9420, 3078.0550, 3558.3955

Compound: sCl 4 + HCHO TS _{ISO}	Energy (kJ mol⁻¹): -739.995345880234
Reaction Coordinates: 6 -1.180838 -0.342869 -0.047973 6 0.073704 0.584064 0.009736 9 -0.311472 1.726564 0.635169 8 0.533091 0.864102 -1.223076 8 2.708270 -0.408426 -0.546306 9 -0.834243 -1.531697 -0.567266 9 -2.119928 0.203241 -0.820694 9 -1.692200 -0.548962 1.169557 8 0.994616 -0.042609 0.911524 6 2.210140 -0.487251 0.545545 1 2.693131 -0.941839 1.415039 1 1.421598 0.471337 -1.346912	Frequencies (cm⁻¹): -65.3287, 65.3295, 166.1356, 213.5199, 249.6288, 294.2800, 334.9333, 360.6191, 367.4474, 473.8145, 495.1819, 570.1397, 578.3899, 600.4688, 687.7175, 766.2511, 829.2022, 951.9805, 1042.0443, 1086.6002, 1174.1938, 1187.2994, 1210.9436, 1233.8137, 1329.3611, 1409.5539, 1441.6080, 1769.3460, 3080.5580, 3478.084

Compound: sCl 4 + HCHO HAE Con 2	Energy (kJ mol⁻¹): -740.000308948885
Reaction Coordinates: 6 1.321852 -0.149789 -0.076490 6 -0.155453 0.309802 0.082835 9 -0.271021 0.919507 1.304423 8 -0.467139 1.141710 -0.926994 8 -2.894484 0.204685 -0.202411 9 1.670431 -0.952740 0.933231 9 2.129042 0.912548 -0.076403 9 1.484130 -0.813728 -1.225184 8 -0.932051 -0.884704 0.150989 6 -2.272728 -0.797419 0.017660 1 -2.706482 -1.794969 0.133869 1 -1.419383 1.335588 -0.875184	Frequencies (cm⁻¹): 62.9145, 78.9746, 159.2883, 218.4675, 240.9289, 261.5800, 319.6150, 363.1625, 387.2717, 494.0228, 520.8418, 570.2369, 584.1607, 606.8505, 657.1655, 769.9403, 843.4823, 972.8608, 1041.0624, 1049.4737, 1157.4872, 1199.4818, 1205.8249, 1227.3689, 1264.5355, 1400.2948, 1476.3678, 1784.9247, 3074.5473, 3642.2040

Compound: SCI 4 + HCHO TS _{HAE} 1	Energy (kJ mol⁻¹): -739.973401978791
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -1.270961 0.225027 -0.120025 6 0.067462 -0.548936 0.046984 9 -0.045608 -1.375286 1.118974 8 0.616065 -1.007043 -0.985121 8 2.721721 -0.011133 -0.566773 9 -1.706195 0.736593 1.034153 9 -2.198727 -0.617761 -0.589049 9 -1.119164 1.223240 -0.997924 8 1.021045 0.649180 0.745659 6 2.215773 0.724389 0.322488 1 2.840018 1.495011 0.779358 1 1.842939 -0.646990 -0.921553	-949.5685, 39.2733, 64.5013, 169.6876, 218.9110, 259.7737, 315.1775, 356.2053, 360.5612, 421.2820, 485.2187, 541.0347, 583.6573, 625.6704, 695.3956, 773.3250, 878.5740, 885.8521, 1061.8765, 1064.7041, 1178.3447, 1184.4422, 1203.2622, 1306.9569, 1376.1017, 1392.3137, 1618.0324, 1702.9903, 1832.9846, 3099.3367

Compound:	SCI 4 + HCHO C _{FAc} 3	Energy (kJ mol⁻¹):	-739.998047248532	
Reaction Coordinates:		Frequencies (cm⁻¹):		
6	-1.303574	-0.423757	-0.134810	19.1597, 41.5247, 46.2241, 60.3739,
6	-0.542531	0.923451	-0.061129	103.4226, 115.2644, 137.3165,
9	-0.880671	1.585261	1.039484	232.0005, 242.2931, 378.9404,
8	0.177014	1.361022	-0.887458	421.5547, 511.4081, 586.5512,
9	-1.346035	-1.037976	1.041452	650.9862, 693.5929, 730.3301,
9	-0.738419	-1.221730	-1.034678	759.1088, 805.5547, 1060.7112,
9	-2.564929	-0.160186	-0.526656	1102.4056, 1154.7082, 1162.0222,
8	1.707875	-0.538008	0.986448	1234.4325, 1313.3144, 1332.7441,
6	2.746001	-0.483644	0.380256	1412.3901, 1787.5158, 1911.7446,
8	2.938640	0.151744	-0.778891	3053.1846, 3652.1981
1	2.105506	0.574602	-1.052806	
1	3.677370	-0.957283	0.709707	

Compound: sCl 4 + HCHO TS _{FAC} 1	Energy (kJ mol⁻¹): -739.806907643778
Reaction Coordinates: 6 1.249403 -0.207960 -0.071445 6 -0.136173 0.515847 -0.043915 9 -0.036599 1.614349 0.743981 8 -0.722589 0.624809 -1.149411 8 -2.550474 0.172171 -0.451244 9 1.736422 -0.392010 1.156046 9 2.116045 0.545375 -0.758823 9 1.144722 -1.394462 -0.677279 8 -0.973738 -0.526973 0.935336 6 -2.022844 -0.827505 0.223626 1 -3.122417 -0.555871 0.681826 1 -2.090814 -1.845743 -0.184181	Frequencies (cm⁻¹): -1180.7629, 71.5676, 94.3298, 177.0207, 221.5961, 286.7658, 296.2595, 310.7880, 364.2419, 418.3103, 467.9715, 471.7714, 541.7660, 587.4465, 665.9389, 704.6313, 756.7313, 833.3443, 888.0859, 1062.5610, 1150.5311, 1182.8067, 1202.5667, 1226.2364, 1259.7834, 1305.1679, 1394.7117, 1473.7019, 2271.2614, 2992.2183
IRC  <p>Relative Energy (kJ mol⁻¹)</p> <p>Reaction Co-ordinates (amu^{1/2} bohr)</p>	

Compound: sCl 4 + HCHO C _{FAC} 1	Energy (kJ mol⁻¹): -739.988628614865
Reaction Coordinates: 6 -1.483755 -0.338849 -0.052076 6 -0.609282 0.901099 0.255281 9 -0.273207 1.517624 -0.882979 8 -0.344831 1.310813 1.324133 9 -1.077788 -0.992867 -1.137908 9 -2.745012 0.090180 -0.267307 9 -1.493370 -1.169297 0.984494 8 1.652661 -0.858738 0.114745 6 2.838184 -0.717960 0.124003 8 3.429826 0.431611 -0.250386 1 4.389305 0.358859 -0.171998 1 3.542958 -1.504848 0.434112	Frequencies (cm⁻¹): 12.2529, 20.2627, 35.6667, 44.4359, 61.7051, 76.2215, 106.1653, 231.3434, 242.0643, 378.2535, 420.8184, 511.7076, 538.9062, 585.0890, 663.4870, 689.6856, 755.8780, 800.1773, 1037.2738, 1083.9196, 1119.7911, 1149.5036, 1229.3486, 1271.2585, 1311.0063, 1415.9550, 1846.2374, 1934.9597, 2981.0252, 3779.8872

Compound: sCl 2 + HCHO TS _{FAC} 2	Energy (kJ mol⁻¹): -739.799988971889
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -1.263159 -0.231854 0.043601 6 0.102664 0.427301 -0.293785 9 0.021464 1.768105 -0.050184 8 0.710354 0.068060 -1.326239 8 2.392624 -0.674104 -0.554776 9 -1.708869 0.127474 1.248179 9 -1.156407 -1.558841 -0.000709 9 -2.163586 0.156430 -0.871899 8 1.020614 -0.082642 1.079098 6 2.216518 -0.013896 0.573236 1 2.952503 -0.932840 0.826383 1 2.789200 0.914505 0.722162	-1104.2800, 62.1223, 68.0935, 171.2913, 216.9350, 272.6382, 279.8117, 307.6771, 350.9695, 387.9892, 427.3649, 484.5099, 540.8308, 585.7682, 660.9810, 703.0031, 750.5784, 812.8046, 884.6414, 1047.5704, 1160.2951, 1176.9495, 1212.6657, 1233.5197, 1260.8937, 1309.0540, 1400.2571, 1491.3314, 2343.7280, 2968.5175
IRC	
<p>The plot shows a single curve representing the reaction coordinate. The y-axis is labeled 'Relative Energy (kJ mol⁻¹)' and ranges from 0 to -600. The x-axis is labeled 'Reaction Co-ordinates (amu^{1/2} bohr)' and ranges from 0 to 10. The curve starts at a relative energy of approximately -280 at a reaction coordinate of 0, rises to a peak of about -110 at a reaction coordinate of 0, and then falls to a minimum of about -580 before leveling off.</p>	

Compound: sCl 4 + HCHO C _{FAC} 2	Energy (kJ mol⁻¹): -739.988569845236
Reaction Coordinates:	Frequencies (cm⁻¹):
6 1.210441 -0.586106 -0.060428 6 0.951717 0.878761 0.367628 9 1.170879 1.701187 -0.662919 8 0.687562 1.243968 1.452601 9 0.766535 -0.844231 -1.288894 9 0.643623 -1.427640 0.797277 9 2.543536 -0.792621 -0.043619 8 -3.160120 -0.849352 0.341988 6 -2.883208 0.325688 -0.253731 8 -1.765750 0.680545 -0.478184 1 -3.774505 0.918577 -0.511101 1 -4.113886 -0.960173 0.442444	9.2098, 19.1505, 33.4684, 43.5209, 59.6264, 74.8819, 104.5050, 231.1540, 242.2426, 378.5483, 420.8136, 511.9706, 538.9994, 585.0978, 662.8509, 689.9521, 757.8900, 800.6257, 1037.2046, 1085.4595, 1119.5570, 1150.2551, 1227.3961, 1271.4277, 1311.4866, 1416.1844, 1846.9631, 1934.9215, 2980.6069, 3779.7123

Compound: sCl 4 + HCHO TS _{ester1}	Energy (kJ mol⁻¹): -739.781535601567
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -1.422805 -0.213581 0.086132 6 0.359595 0.380720 -0.081407 9 0.191519 1.620006 -0.668748 8 0.538537 -0.583618 -0.907970 8 2.469527 -0.856868 -0.466226 9 -1.926600 0.722605 0.853416 9 -1.399451 -1.367685 0.681583 9 -2.034024 -0.242012 -1.065347 8 0.871900 0.395574 1.076719 6 2.664197 0.007717 0.427705 1 3.007473 -0.336057 1.408659 1 2.863899 1.049987 0.168430	-588.2564, 60.7916, 99.0798, 108.1427, 164.7614, 193.2517, 230.6891, 255.7045, 306.6388, 317.2887, 487.2591, 534.9642, 554.2207, 567.5308, 621.5868, 676.4830, 756.5990, 807.2190, 898.5517, 976.0160, 1150.8830, 1197.0207, 1251.1035, 1261.5896, 1330.9552, 1395.3279, 1473.9622, 1554.1570, 2995.2314, 3082.5091
IRC	
<p>The plot displays the reaction coordinate (IRC) for the reaction. The y-axis represents the relative energy in kJ mol⁻¹, ranging from 0 to -600. The x-axis represents the reaction coordinate in amu^{1/2} bohr, ranging from -5 to 10. A single curve starts at a relative energy of approximately -320 kJ mol⁻¹ at -4 amu^{1/2} bohr, rises to a peak of about -110 kJ mol⁻¹ at 0.2 amu^{1/2} bohr, and then gradually decreases towards -550 kJ mol⁻¹ at 8 amu^{1/2} bohr.</p>	

Compound: sCl 4 + HCHO C _{ester1}	Energy (kJ mol⁻¹): -739.977318615772
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -1.438091 -0.328538 -0.053297 9 -2.123748 0.239571 0.936736 9 -2.280767 -0.693219 -1.009719 9 -0.817177 -1.406218 0.414875 8 -0.556511 0.567248 -0.640931 6 0.433924 1.093616 0.117396 9 1.086785 1.935488 -0.663881 8 0.673618 0.909472 1.257292 8 2.281535 -0.867481 -0.682601 6 2.928435 -1.225038 0.266485 1 3.752697 -1.952181 0.162579 1 2.726723 -0.842568 1.281745	31.7264, 43.5110, 60.0890, 78.1354, 102.3299, 117.0928, 149.9059, 165.4967, 179.4153, 378.1240, 403.6030, 427.4693, 550.9837, 609.0345, 669.9364, 737.6815, 757.5186, 879.5778, 1023.6817, 1149.5186, 1207.6212, 1215.8080, 1240.9116, 1267.7803, 1278.8360, 1527.3657, 1803.0161, 1911.4485, 2906.0910, 2974.0356

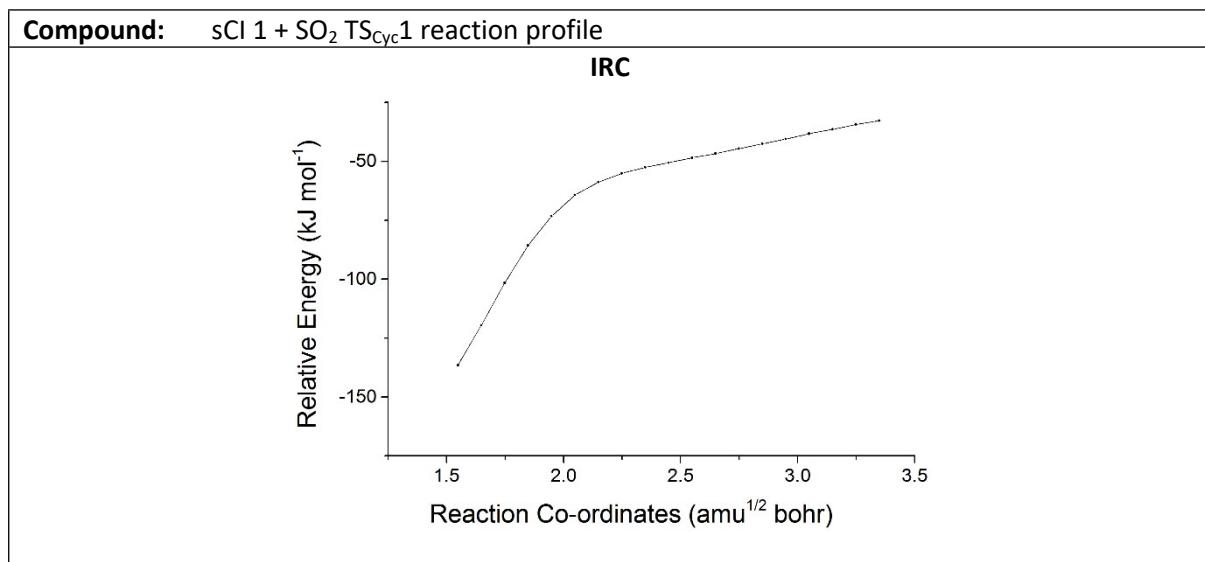
Compound: sCl 4 + HCHO TS _{ester} 2	Energy (kJ mol⁻¹): -739.786202678286
Reaction Coordinates:	Frequencies (cm⁻¹):
6 1.396175 -0.267274 0.002936 6 -0.388780 0.495260 0.047108 9 -0.090167 1.814698 -0.114078 8 -0.540319 -0.195561 -1.022155 8 -2.553430 -0.185580 -0.752387 9 1.934085 0.381624 1.010036 9 1.284892 -1.539686 0.259211 9 2.044179 -0.022649 -1.101464 6 -2.469377 -0.623178 0.426331 8 -0.970239 0.192826 1.133098 1 -2.121480 -1.639456 0.624488 1 -3.151636 -0.188761 1.165464	-594.8467, 54.8487, 91.2012, 121.2305, 171.7936, 201.1541, 241.4115, 246.9698, 297.8808, 328.8755, 519.5294, 529.9244, 550.6988, 593.3187, 627.0561, 672.7171, 743.0208, 879.0350, 937.2457, 970.3107, 1143.5766, 1189.1849, 1256.0499, 1280.3139, 1328.2734, 1378.7216, 1490.3402, 1533.3773, 2992.9535, 3075.6329
IRC	
<p>The plot shows a single reaction coordinate curve. The y-axis is labeled "Relative Energy (kJ mol⁻¹)" and ranges from 0 to -600. The x-axis is labeled "Reaction Co-ordinates (amu^{1/2} bohr)" and ranges from -2 to 8. The curve starts at approximately -280 at x = -1.5, rises to a peak of about -100 at x = 0, and then gradually declines to approximately -520 at x = 7.</p>	

Compound: CF ₃ CFO	Energy (kJ mol⁻¹): -451.199957828211
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -0.590884 0.016760 -0.000000 6 0.950258 0.161191 -0.000000 8 1.560686 1.164510 0.000000 9 1.511346 -1.053957 -0.000000 9 -0.986585 -0.656718 1.087722 9 -1.165035 1.213645 -0.000003 9 -0.986585 -0.656723 -1.087719	43.7845, 226.0109, 238.7949, 378.7825, 419.3861, 512.0808, 584.7772, 688.3344, 765.2522, 800.0823, 1086.2930, 1167.3575, 1222.6180, 1305.3726, 1935.9581

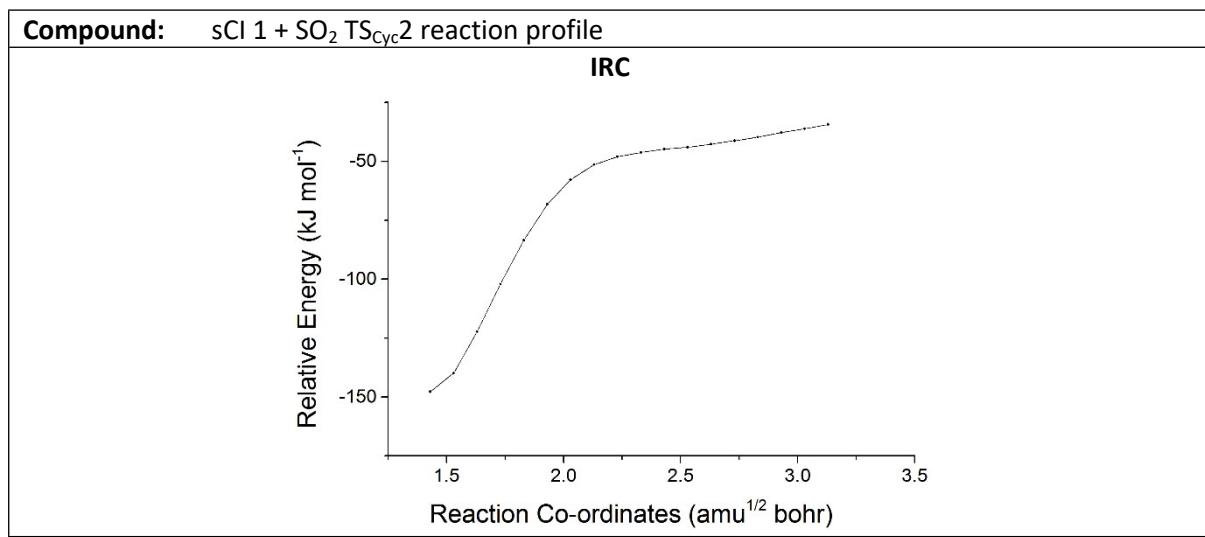
Compound: CF ₃ OCFO	Energy (kJ mol⁻¹): -526.392169685219
Reaction Coordinates:	Frequencies (cm⁻¹):
6 0.985021 -0.064022 0.000000 9 1.142265 0.688671 1.083247 9 1.895578 -1.026215 -0.000000 9 1.142264 0.688671 -1.083247 8 -0.249812 -0.707922 0.000001 6 -1.372203 0.051286 0.000000 9 -2.386217 -0.800803 -0.000000 8 -1.477929 1.223361 -0.000000	75.3080, 177.0143, 376.9655, 401.4174, 427.2777, 550.7688, 610.1072, 668.4452, 737.8881, 774.4134, 880.1638, 1021.3954, 1141.4172, 1225.8672, 1237.8952, 1275.4731, 1923.5619

S8.3 Reactions with SO₂

Compound: SO ₂	Energy (kJ mol⁻¹): -548.08584429715
Reaction Coordinates: 16 0.000000 0.000000 0.372669 8 0.000000 1.245290 -0.372669 8 -0.000000 -1.245290 -0.372669	Frequencies (cm⁻¹): 513.3417, 1156.5416, 1337.5576



Compound: sCl 1 + SO ₂ SOZ 1	Energy (kJ mol⁻¹): -737.54758505440
Reaction Coordinates: 6 1.382069 -0.646247 0.414243 1 1.138310 -0.656463 1.477779 8 1.589137 0.652527 -0.058367 8 0.269739 1.234086 0.029932 16 -0.903903 0.016549 -0.430359 8 0.274519 -1.149994 -0.346833 8 -1.787516 -0.049041 0.719351 1 2.264694 -1.231458 0.169849	Frequencies (cm⁻¹): 122.2352, 300.3218, 365.2906, 429.7004, 512.9059, 611.8820, 691.2164, 714.2195, 868.9158, 923.9394, 1046.7403, 1148.9215, 1243.2803, 1247.7933, 1377.9775, 1505.9329, 3052.1770, 3146.1962

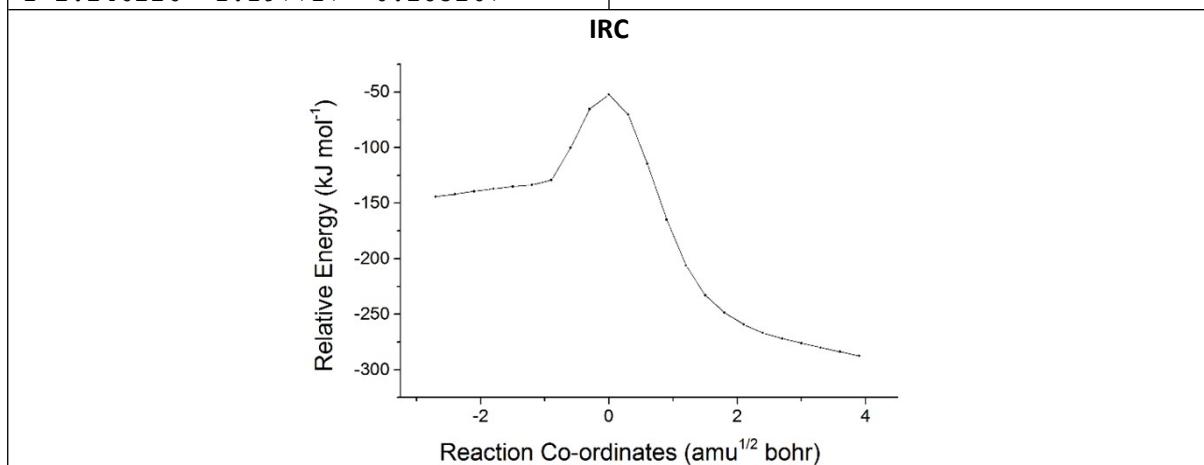




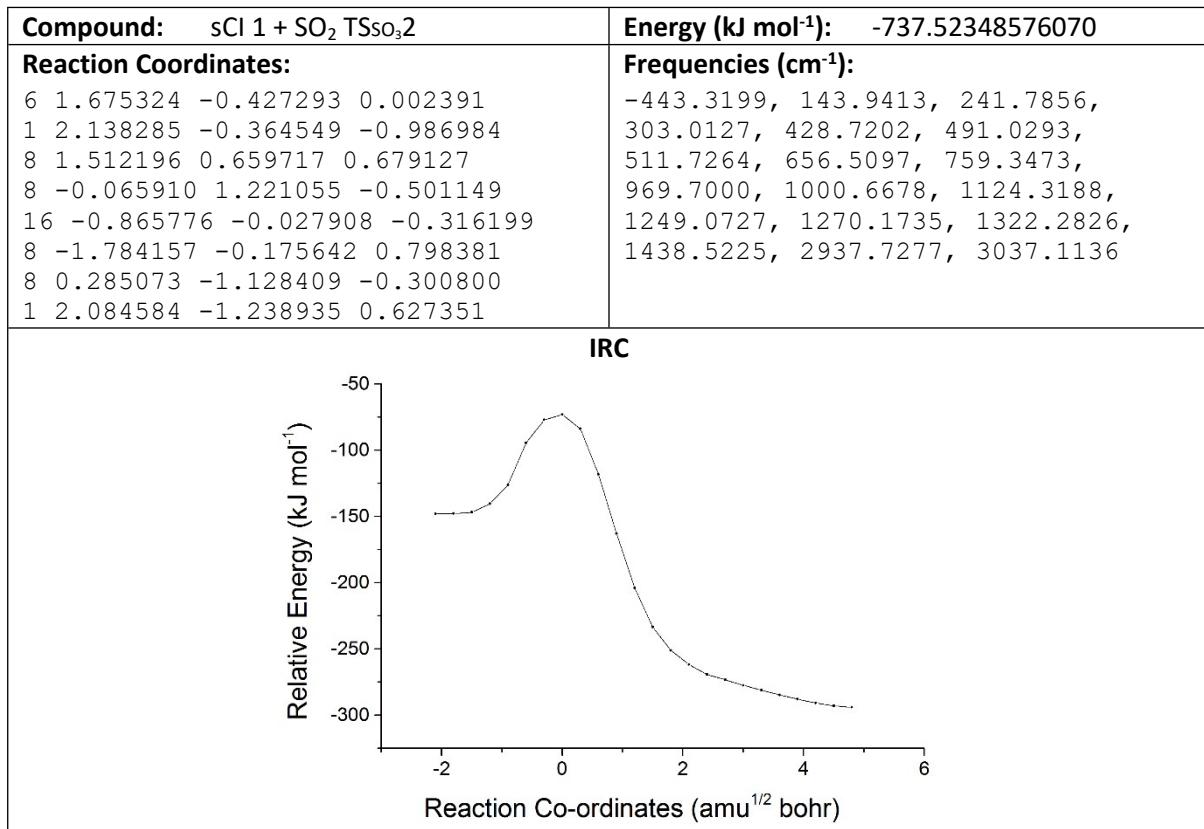
Compound: sCl 1 + SO ₂ SOZ 2	Energy (kJ mol⁻¹): -737.54937458407
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -1.556878 -0.547909 0.071274 1 -2.103692 -0.542148 -0.874056 8 -1.293568 0.748906 0.523130 8 -0.267816 1.181102 -0.421646 16 0.909912 -0.063250 -0.409928 8 -0.292974 -1.188917 -0.119714 8 1.727057 -0.003975 0.785128 1 -2.095215 -1.055326 0.870085	132.0686, 287.7793, 374.3085, 447.0565, 519.4221, 630.6105, 667.5639, 714.3830, 835.9980, 947.7173, 1040.2338, 1133.8437, 1234.1709, 1258.4403, 1386.7702, 1504.2540, 3036.7579, 3124.0647

Compound: sCl 1 + SO ₂ TS _{hoz}	Energy (kJ mol⁻¹): -737.54937458407
Reaction Coordinates:	Frequencies (cm⁻¹):
6 1.386780 -0.692255 0.278605 1 1.274693 -1.092237 1.287451 8 1.397845 0.745999 0.369330 8 0.212787 1.242656 -0.331163 16 -0.931332 0.018255 -0.381583 8 0.303256 -1.082345 -0.546129 8 -1.538487 -0.158847 0.927095 1 2.302725 -1.026016 -0.206819	-179.8160, 159.1954, 353.4794, 451.0161, 572.8258, 671.8872, 697.3124, 723.2182, 833.2018, 938.3023, 1040.7095, 1126.7846, 1197.2006, 1241.8372, 1393.0334, 1531.9412, 3041.7383, 3108.8411

Compound: sCl 1 + SO ₂ TS _{so₃1}	Energy (kJ mol⁻¹): -737.51404208200
Reaction Coordinates:	Frequencies (cm⁻¹):
6 1.501312 -0.650356 0.305770 1 1.174843 -1.145000 1.224048 8 1.700162 0.635117 0.346011 8 -0.052180 1.317998 -0.225692 16 -0.881951 0.066084 -0.304427 8 0.227311 -0.967082 -0.711171 8 -1.665008 -0.337595 0.852781 1 2.246226 -1.197717 -0.283267	-559.6034, 124.6928, 185.5359, 320.2554, 418.0848, 482.2966, 535.8077, 616.2304, 804.3402, 960.9313, 1009.8710, 1160.8091, 1224.0481, 1249.8633, 1310.5436, 1506.3998, 2983.3701, 3065.2443

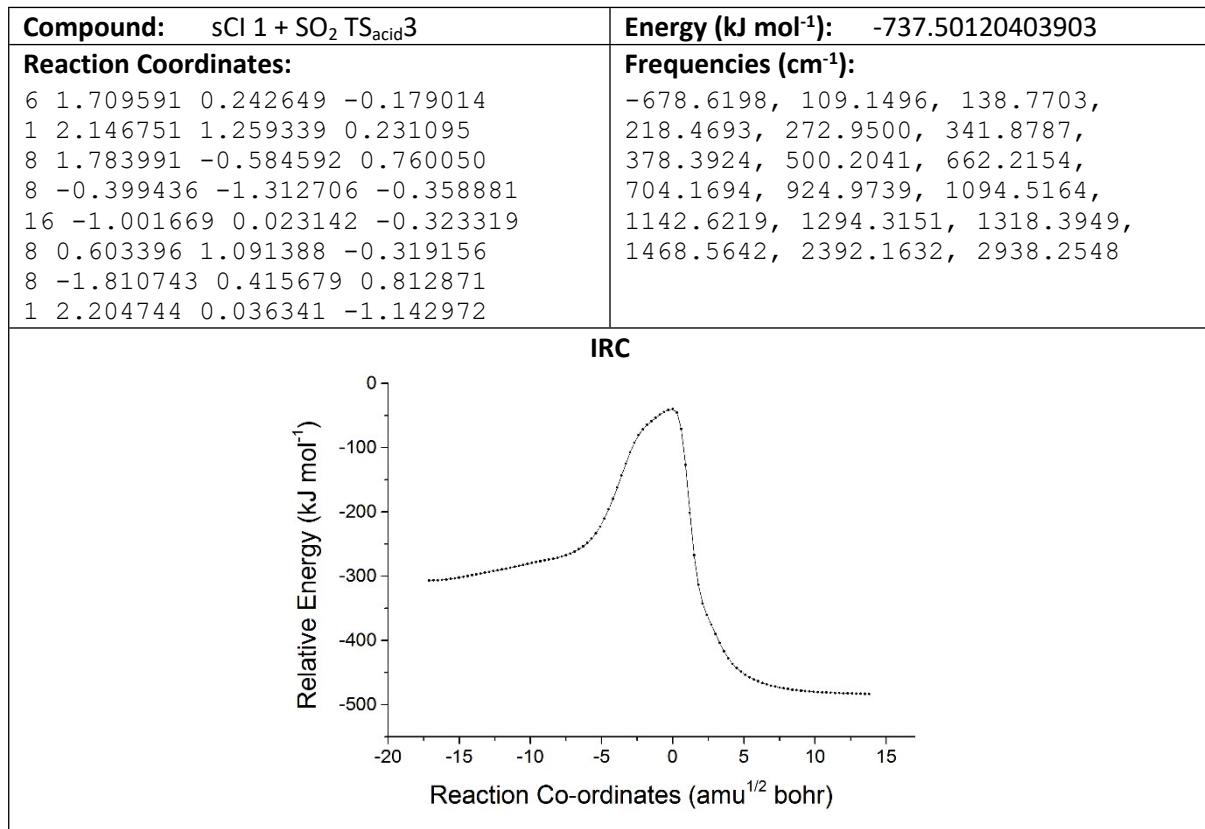


Compound: sCl 1 + SO ₂ Cso ₃ 1	Energy (kJ mol⁻¹): -737.62562027792
Reaction Coordinates: 6 2.341652 0.110229 -0.000001 1 3.391734 -0.210515 0.000002 1 2.117654 1.185058 -0.000005 8 1.456406 -0.714452 0.000001 8 -1.068999 -0.553788 1.247214 16 -0.728472 0.078182 0.000000 8 -0.306356 1.461227 -0.000032 8 -1.069019 -0.553840 -1.247182	Frequencies (cm⁻¹): 57.7179, 151.3388, 177.0498, 181.1805, 298.8735, 366.7314, 474.7464, 514.5355, 515.9279, 1039.7032, 1215.7655, 1262.8801, 1342.6023, 1367.2268, 1508.6567, 1769.9047, 2969.2137, 3063.1764



Compound: sCl 1 + SO ₂ TS _{acid} 1	Energy (kJ mol⁻¹): -737.49389238926
Reaction Coordinates: 6 1.629372 -0.345072 -0.136515 1 2.240069 -0.218982 -1.044043 8 1.512066 0.699746 0.661067 8 -0.200944 1.243956 -0.386806 16 -0.992011 -0.032259 -0.373489 8 0.632019 -1.195266 -0.210455 8 -1.755655 -0.317004 0.823119 1 2.355981 -0.645892 0.743562	Frequencies (cm⁻¹): -832.8318, 133.5888, 195.8912, 247.3174, 358.5430, 385.4304, 432.5792, 505.1556, 704.0409, 802.4629, 986.7936, 1136.3870, 1204.3920, 1276.1307, 1298.1726, 1376.3094, 2452.6451, 2962.2799
IRC	
Compound: sCl 1 + SO ₂ TS _{acid} 2	Energy (kJ mol⁻¹): -737.50143644919
Reaction Coordinates: 6 1.405536 -0.555411 0.316090 1 1.245825 -0.987224 1.317976 8 1.705258 0.735066 0.276800 8 -0.214092 1.312346 -0.104809 16 -1.027584 0.071542 -0.362876 8 0.694356 -1.055207 -0.664845 8 -1.658572 -0.514248 0.807833 1 2.546705 -0.648638 0.071658	Frequencies (cm⁻¹): -731.4789, 123.8317, 184.0541, 234.6166, 338.4323, 409.5670, 426.8130, 509.2331, 691.9101, 790.8737, 971.3345, 1128.3870, 1220.4001, 1255.4461, 1285.1249, 1375.6071, 2501.8042, 2945.3882
IRC	

Compound: sCl 1 + SO ₂ C _{acid} 1	Energy (kJ mol⁻¹): -737.67576753941
Reaction Coordinates: 6 -2.324179 -0.010590 -0.355769 1 -3.146885 0.039791 1.368678 8 -1.344757 -0.030178 -1.041439 8 1.571462 -1.228815 0.473015 16 1.508648 -0.009039 -0.313940 8 -2.260866 0.029061 0.985361 8 1.571843 1.253882 0.401687 1 -3.347860 -0.023224 -0.760015	Frequencies (cm⁻¹): 21.5880, 28.3617, 29.0618, 78.5591, 96.3446, 133.8663, 517.6701, 540.0397, 665.5994, 1037.9677, 1126.1208, 1156.9469, 1270.4993, 1328.5625, 1414.4276, 1835.1316, 2984.2698, 3780.5005



Compound: sCl 1 + SO ₂ C _{acid} 2	Energy (kJ mol⁻¹): -737.68599404256
Reaction Coordinates: 6 2.153108 0.466614 -0.030781 8 1.189385 1.145996 0.230713 8 2.145949 -0.843591 -0.256551 1 1.232434 -1.183415 -0.184780 1 3.169916 0.865614 -0.112439 8 -2.084303 0.668245 -0.679832 16 -1.347910 -0.002589 0.371589 8 -0.720335 -1.275707 0.022731	Frequencies (cm⁻¹): 31.4878, 68.9111, 113.5839, 137.3046, 162.6219, 182.6245, 519.2643, 661.9957, 774.6861, 1064.2063, 1148.5395, 1178.5391, 1313.9712, 1351.0209, 1416.0962, 1763.6747, 3056.0992, 3574.5704

Compound: SO ₃	Energy (kJ mol⁻¹): -623.22229804288
Reaction Coordinates: 16 0.000000 -0.000000 0.000000 8 0.246828 -1.418352 0.000000 8 -1.351743 0.495417 0.000000	Frequencies (cm⁻¹): 475.6389, 507.9530, 507.9566, 1044.8229, 1363.4683, 1363.4939

8 1.104915 0.922936 -0.000000	
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Compound: sCl 2 + SO ₂ PRC1	Energy (kJ mol⁻¹): -1074.30049656118
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -1.680849 -0.330936 0.008724	29.2239, 48.4725, 62.4815, 91.7570,
6 -1.137494 0.933439 0.662474	100.6754, 174.7347, 180.0268,
8 -0.286665 1.695181 0.151214	226.5884, 268.6855, 328.0287,
8 0.351463 1.317116 -0.987851	471.8117, 505.2088, 516.6429,
1 -1.567548 1.282222 1.592022	535.2394, 580.3347, 756.1669,
9 -2.292584 -0.014933 -1.139885	830.0680, 877.0832, 910.6656,
9 -0.764980 -1.257811 -0.222516	1135.9560, 1154.7572, 1181.4849,
9 -2.603039 -0.837675 0.848126	1253.8791, 1308.2990, 1367.8966,
16 2.325088 0.008750 0.066354	1568.4158, 3206.0587
8 1.581156 -0.072555 1.323138	
8 2.381748 -1.195175 -0.743052	

Compound: sCl 2 + SO ₂ TS _{Cyc} 1	Energy (kJ mol⁻¹): -1074.29932649603
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -1.567535 -0.331819 -0.021801	-44.3238, 45.4533, 84.8741,
6 -0.988288 0.893200 0.685292	109.5867, 127.1715, 192.3251,
8 -0.211136 1.738966 0.185226	213.1513, 285.4045, 305.8816,
8 0.497696 1.392847 -0.932081	334.5709, 462.8314, 505.3981,
1 -1.437201 1.192710 1.623913	515.4080, 535.7839, 572.3408,
9 -2.387958 0.121866 -0.988788	754.0951, 843.8360, 874.8355,
9 -0.705693 -1.173798 -0.552521	916.4321, 1106.9795, 1146.0291,
9 -2.314631 -0.984971 0.884328	1161.5810, 1276.1099, 1295.7925,
16 2.173682 -0.031753 0.130690	1367.9148, 1566.9924, 3200.2906
8 1.283290 -0.181356 1.294872	
8 2.263619 -1.165560 -0.770900	

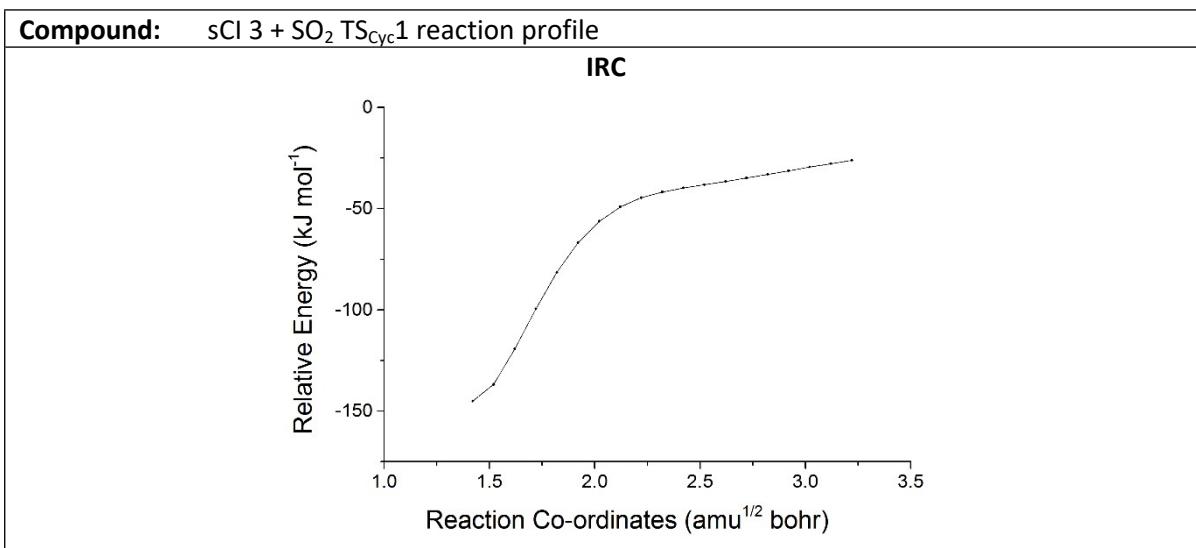
Compound: sCl 2 + SO ₂ SOZ 1	Energy (kJ mol⁻¹): -1074.35461788075
Reaction Coordinates:	Frequencies (cm⁻¹):
6 1.509783 -0.211551 0.070085	44.2404, 70.6265, 157.2900,
6 0.472691 0.607230 -0.732729	209.8630, 255.8569, 303.1416,
8 -0.097190 1.667245 -0.006489	343.8475, 400.9612, 422.9259,
8 -1.133556 1.051570 0.784769	510.4862, 527.3151, 561.9128,
1 0.991158 1.063821 -1.576545	628.6370, 670.6377, 673.0996,
9 2.487991 0.611495 0.483098	794.4516, 871.0570, 895.9295,
9 1.000662 -0.831871 1.132375	1002.9074, 1065.5800, 1165.6786,
9 2.057394 -1.134082 -0.739936	1168.0381, 1268.0762, 1290.8391,
16 -1.965027 -0.084757 -0.232373	1306.0599, 1396.4551, 3091.8167
8 -0.582831 -0.198334 -1.205133	
8 -2.106424 -1.256938 0.600670	

Compound: sCl 2 + SO ₂ PRC2	Energy (kJ mol⁻¹): -1074.29958900339
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -1.884175 -0.194669 -0.056099	24.6678, 6.0920, 55.9841, 81.8352,
6 -1.045646 0.530654 0.987311	103.1086, 146.9604, 190.9617,
8 -0.092184 1.297137 0.725886	214.3366, 264.9684, 327.1424,
8 0.348291 1.408367 -0.552527	474.3552, 505.0058, 518.9497,
1 -1.314868 0.467760 2.033230	534.9524, 582.8485, 756.8278,

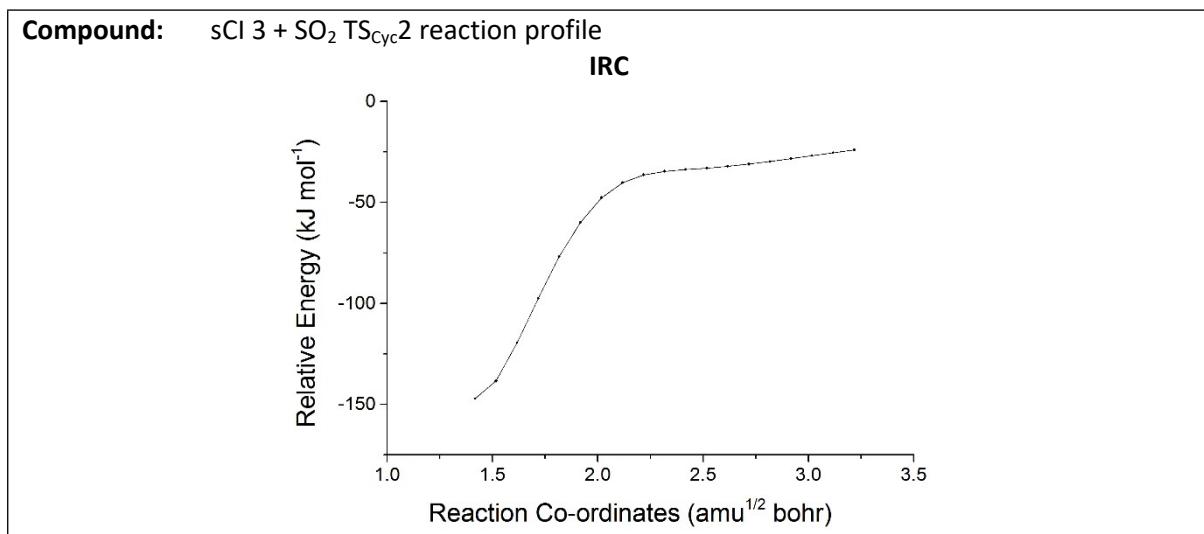
9 -2.513437 0.695230 -0.833304	825.7794, 877.5626, 913.2445,
9 -1.191282 -1.028633 -0.819448	1139.9864, 1156.8720, 1179.9236,
9 -2.815482 -0.900882 0.610603	1250.4563, 1309.9265, 1368.0173,
16 2.275655 -0.411951 -0.361018	1568.8053, 3207.4250
8 1.470431 -1.130483 0.624267	
8 3.419101 0.326993 0.144266	

Compound: sCl 2 + SO ₂ TS _{Cyc2}	Energy (kJ mol⁻¹): -1074.29940211685
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -1.767970 -0.200627 -0.050972 6 -0.892727 0.535501 0.961344 8 -0.010909 1.378492 0.685315 8 0.437785 1.465943 -0.599539 1 -1.147977 0.467376 2.011078 9 -2.588927 0.704888 -0.612708 9 -1.135920 -0.854988 -1.007771 9 -2.520183 -1.071524 0.642652 16 2.118402 -0.439758 -0.360278 8 1.208848 -1.095346 0.588120 8 3.292152 0.195178 0.212553	-31.0837, 46.6099, 55.9795, 106.4244, 127.9574, 150.6068, 212.6447, 267.3147, 290.7125, 327.6061, 469.3283, 504.8931, 519.0325, 534.8190, 577.3396, 755.5527, 838.7640, 874.0912, 912.2329, 1120.7541, 1149.9620, 1164.0053, 1271.7624, 1294.7358, 1367.7513, 1573.1131, 3202.7905

Compound: sCl 2 + SO ₂ SOZ 2	Energy (kJ mol⁻¹): -1074.36037274421
Reaction Coordinates:	Frequencies (cm⁻¹):
6 1.660626 -0.102587 0.034820 6 0.359614 0.028112 -0.791388 8 -0.261150 1.276678 -0.666093 8 -1.040876 1.153037 0.563436 1 0.609532 -0.077818 -1.848446 9 2.505897 0.881224 -0.304427 9 1.452955 -0.048177 1.353799 9 2.248145 -1.276966 -0.250026 16 -1.869280 -0.327151 0.446264 8 -0.560881 -0.969187 -0.405574 8 -2.972776 -0.241233 -0.485078	48.0706, 78.6188, 141.0420, 202.3361, 257.8851, 294.1922, 383.5957, 399.3418, 467.0707, 518.5585, 545.0363, 578.2684, 638.3053, 663.9512, 690.5619, 788.0789, 833.9486, 863.3242, 1009.0424, 1064.5929, 1159.6992, 1175.8721, 1265.5730, 1282.5502, 1289.6916, 1391.5748, 3085.6763



Compound: sCl 2 + SO ₂ SOZ3	Energy (kJ mol⁻¹): -1074.35773058988
Reaction Coordinates:	Frequencies (cm⁻¹):
6 1.702452 -0.118431 0.047577 6 0.387703 0.424608 -0.543796 8 -0.198834 1.268729 0.409836 8 -1.564676 1.367343 -0.075039 16 -2.078769 -0.251684 -0.317001 8 -0.511096 -0.644797 -0.784013 8 -2.332760 -0.913456 0.942857 1 0.594851 0.949559 -1.480180 9 2.537742 0.903356 0.291605 9 2.283775 -0.929409 -0.847908 9 1.509979 -0.794193 1.176345	51.6628, 66.2206, 165.4099, 188.0028, 271.1719, 303.6911, 367.9778, 380.8930, 444.9106, 520.1523, 532.6612, 559.3600, 636.8891, 685.6619, 705.9521, 715.0671, 872.6268, 881.7316, , 1019.6375, 1089.4928, 1170.7546, 1177.1922, 1265.7594, 1283.3818, 1342.3435, 1410.2857, 3050.3375



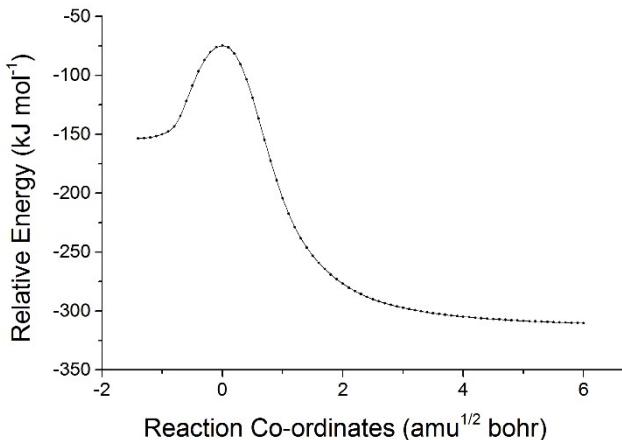
Compound: sCl 2 + SO ₂ SOZ4	Energy (kJ mol⁻¹): -1074.35489019767
Reaction Coordinates:	Frequencies (cm⁻¹):
6 1.779444 -0.130676 -0.041084 6 0.306102 0.102417 -0.412299 8 -0.074918 1.371532 0.041590 8 -1.506844 1.335170 -0.093337 16 -2.072519 -0.245121 0.411857 8 -0.508107 -0.825571 0.290510 8 -2.873972 -0.681963 -0.712054 1 0.169485 -0.002257 -1.491093 9 1.987951 -0.045424 1.272100 9 2.154784 -1.347066 -0.458012 9 2.544850 0.781422 -0.657643	51.7553, 81.6765, 142.5523, 164.0122, 266.7253, 320.0954, 323.3741, 385.3553, 427.9052, 524.1165, 526.5158, 558.0106, 610.8184, 685.2486, 717.4421, 729.0240, 887.8522, 932.4643, 996.8673, 1088.8626, 1172.7260, 1183.7142, 1259.5769, 1278.4975, 1332.3607, 1420.1472, 3063.9397

Compound: sCl 2 + SO ₂ TS _{SO₂1}	Energy (kJ mol⁻¹): -1074.35121769837
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -1.525315 -0.164536 -0.056157	-144.3866, 50.6273, 105.2007,
6 -0.430735 0.665276 0.657307	181.7474, 205.0582, 309.4378,
8 0.145384 1.597292 -0.246252	330.6954, 397.4601, 417.8668,
8 1.535307 1.215057 -0.487627	517.0891, 558.6261, 567.8528,
1 -0.910361 1.222606 1.465011	652.0541, 679.3330, 685.5999,
9 -2.054881 -1.045162 0.807896	785.9350, 842.6908, 854.7400,
9 -2.503007 0.669459 -0.452593	1010.1829, 1080.4706, 1162.5741,
9 -1.076416 -0.825045 -1.120134	1171.6860, 1266.5673, 1295.1893,
16 1.940963 -0.260886 0.233548	1303.7839, 1404.6539, 3062.0777
8 0.581371 -0.131710 1.211230	
8 1.775436 -1.336408 -0.718004	

Compound: sCl 2 + SO ₂ TS _{SO₂2}	Energy (kJ mol⁻¹): -1074.35365432423
Reaction Coordinates:	Frequencies (cm⁻¹):
6 1.748937 -0.114434 -0.024259	-101.5726, 62.6211, 99.9642,
6 0.274018 -0.031580 -0.460387	155.2252, 220.0687, 319.4128,
8 -0.153842 1.321783 -0.344886	339.1073, 379.8420, 449.1546,
8 -1.482972 1.313361 0.269186	518.6561, 554.8601, 588.8998,
1 0.176216 -0.373303 -1.494543	669.4501, 686.7225, 717.7112,
9 2.486196 0.699763 -0.793583	738.3950, 859.8401, 869.9778,
9 1.915846 0.233503 1.250191	1010.8214, 1073.4618, 1171.8732,
9 2.194913 -1.368014 -0.190779	1176.4586, 1252.2470, 1280.7008,
16 -2.072430 -0.261113 0.390273	1331.7885, 1416.1239, 3048.0420
8 -0.498174 -0.817146 0.413770	
8 -2.680969 -0.650508 -0.867371	

Compound: sCl 2 + SO ₂ TS _{SO₃1}	Energy (kJ mol⁻¹): -1074.33512300027
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -1.681073 0.106352 0.052263	-441.1456, 46.9288, 90.8543,
6 -0.432132 -0.144608 -0.832003	126.8517, 191.0219, 253.8848,
8 0.079204 -1.329812 -0.848233	286.2551, 330.2239, 420.2077,
8 1.230573 -1.027724 0.793877	443.1656, 502.5533, 517.3251,
1 -0.614523 0.241506 -1.849452	538.3314, 574.7456, 694.5466,
9 -2.117836 1.366947 -0.108056	749.8890, 789.9627, 844.1961,
9 -1.449414 -0.096900 1.350768	994.5220, 1091.1145, 1149.2041,
9 -2.655367 -0.725443 -0.342431	1185.0667, 1223.4300, 1266.6729,
16 1.847120 0.256751 0.351536	1269.7930, 1382.1165, 2946.7385
8 0.609315 0.938962 -0.387725	
8 3.048831 0.290896 -0.457819	

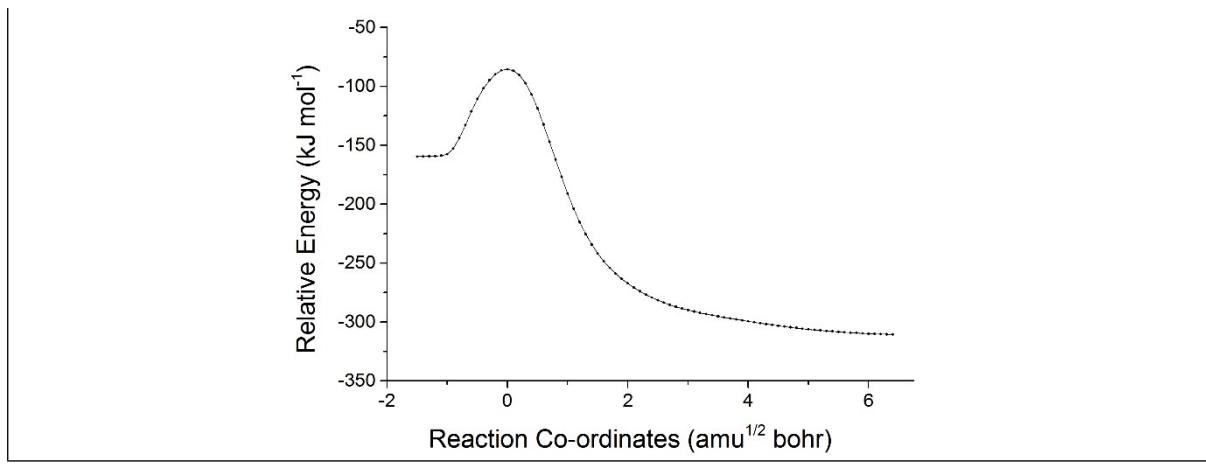
IRC



Compound: sCl 2 + SO ₂ C _{so} ₃ 1	Energy (kJ mol⁻¹): -1074.43431110038
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -2.359197 0.002975 -0.000000 6 -0.844225 -0.282288 -0.000001 8 -0.032595 0.601469 -0.000001 8 2.641537 0.592391 -1.247684 1 -0.574830 -1.348409 -0.000002 9 -2.904523 -0.565307 1.089032 9 -2.636321 1.298767 -0.000005 9 -2.904526 -0.565316 -1.089025 16 2.370290 -0.068647 0.000000 8 1.984410 -1.460059 -0.000014 8 2.641529 0.592367 1.247698	20.1907, 42.5260, 69.5071, 81.9062, 83.0338, 166.6485, 171.1706, 294.3903, 329.2156, 434.3420, 463.1476, 509.8777, 511.3396, 527.5328, 529.7630, 706.9608, 846.3766, 999.3410, 1043.9802, 1163.3722, 1181.5146, 1289.2590, 1351.4924, 1373.8588, 1399.6574, 1819.7549, 3013.7213

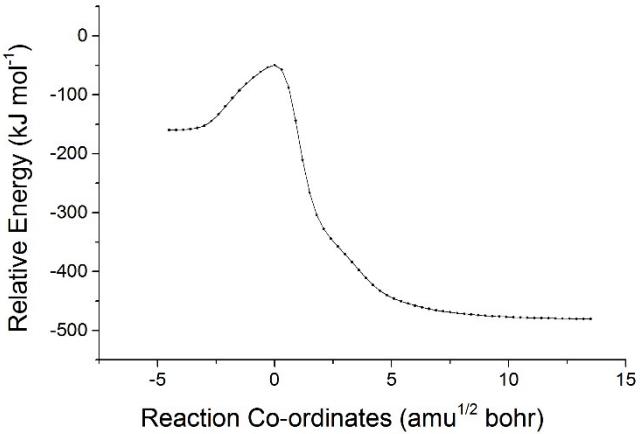
Compound: sCl 2 + SO ₂ TS _{so} ₃ 2	Energy (kJ mol⁻¹): -1074.32997089072
Reaction Coordinates:	Frequencies (cm⁻¹):
6 1.728009 -0.125183 0.034650 6 0.441912 0.660406 -0.396289 8 -0.050099 1.358879 0.576907 8 -1.915066 1.267761 -0.144913 1 0.611404 1.151115 -1.360112 9 2.138415 -0.915789 -0.963504 9 1.529476 -0.858648 1.121655 9 2.695958 0.767152 0.278919 16 -1.992157 -0.223075 -0.273980 8 -0.517223 -0.532291 -0.772050 8 -2.396494 -1.060311 0.837555	-441.5083, 56.4485, 72.5008, 161.8351, 186.0835, 266.6349, 292.1361, 317.1228, 403.2586, 441.0609, 507.0897, 515.8940, 530.1940, 567.2088, 679.8117, 737.2216, 794.7079, 840.8249, 975.2500, 1151.3726, 1175.2529, 1210.0695, 1257.0745, 1277.7910, 1303.9932, 1360.3222, 3030.9695

IRC

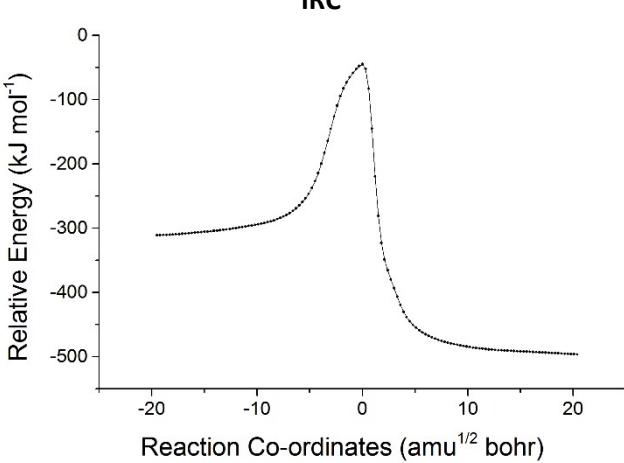


Compound: sCl 2 + SO ₂ TS _{acid} 1	Energy (kJ mol⁻¹): -1074.30007800359
Reaction Coordinates: 6 -1.505222 -0.239235 -0.072830 6 -0.468416 0.779170 0.532436 8 -0.117868 1.770499 -0.267234 8 1.675590 0.954167 -0.770812 1 -1.112337 1.616668 1.033243 9 -2.586837 0.431945 -0.501419 9 -0.984302 -0.900871 -1.104109 9 -1.892435 -1.114031 0.857215 16 2.020959 -0.176837 0.163869 8 0.396458 0.320788 1.397348 8 1.769694 -1.517986 -0.319049	Frequencies (cm⁻¹): -688.6455, 34.5698, 102.7656, 106.7723, 153.6091, 184.0275, 240.9328, 282.9228, 334.7932, 384.6055, 428.8683, 438.7417, 490.1544, 522.3457, 589.2997, 694.3193, 792.1899, 813.5240, 883.2912, 961.6328, 1139.4703, 1187.9310, 1206.3680, 1248.5025, 1281.7435, 1340.6893, 2496.1816
IRC	
<p>Relative Energy (kJ mol⁻¹)</p> <p>Reaction Co-ordinates (amu^{1/2} bohr)</p>	

Compound: sCl 2 + SO ₂ C _{acid} 1	Energy (kJ mol⁻¹): -1074.48261883124
Reaction Coordinates: 6 2.244544 -0.262546 -0.114768 6 0.894474 0.357660 0.356378 8 0.884820 1.682848 0.496233 1 1.738515 2.069558 0.254068 8 -0.046837 -0.341178 0.576054 9 2.072883 -0.954956 -1.237657 9 3.167254 0.703181 -0.353907 9 2.733065 -1.071123 0.825914 8 -2.852061 0.915641 -0.944483 16 -3.036633 0.092495 0.238408 8 -3.454086 -1.284071 0.043772	Frequencies (cm⁻¹): 10.4592, 11.1773, 14.8872, 26.3993, 43.1879, 67.6075, 108.1088, 252.6177, 259.5524, 385.2853, 426.5345, 505.0798, 516.9555, 571.0955, 584.3263, 688.8721, 776.4952, 793.7122, 1118.9072, 1157.7687, 1179.0938, 1197.1657, 1239.8601, 1332.5658, 1372.3901, 1868.7735, 3746.5623

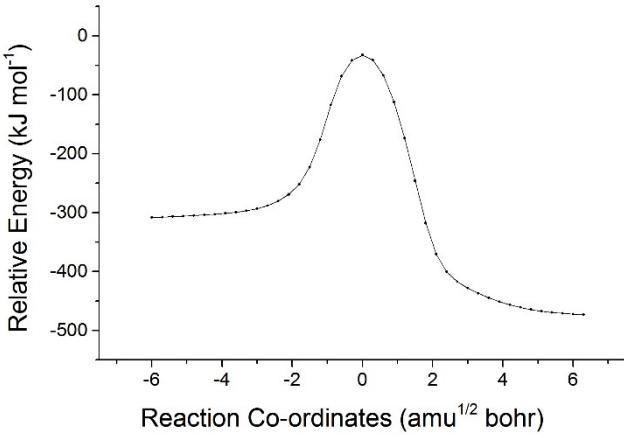
Compound: sCl 2 + SO ₂ TS _{acid} 2	Energy (kJ mol⁻¹): -1074.31032548775
Reaction Coordinates: 6 1.719219 -0.101523 0.094679 6 0.407056 0.052457 -0.744778 8 -0.078855 1.275075 -0.892810 8 -1.295967 1.040241 0.709241 1 0.733771 0.348845 -1.848093 9 2.545062 0.924515 -0.130028 9 1.435752 -0.149389 1.398708 9 2.343233 -1.234436 -0.248524 16 -1.963491 -0.284174 0.456166 8 -0.410283 -0.954650 -0.706844 8 -3.088894 -0.282400 -0.451010	Frequencies (cm⁻¹): -887.2658, 41.0240, 81.3727, 118.9355, 184.5431, 234.3204, 260.7966, 266.7216, 358.3049, 388.3995, 428.5897, 449.4796, 503.2875, 521.6048, 590.2512, 700.7906, 797.3061, 818.5908, 886.7907, 979.8302, 1149.4563, 1190.2700, 1204.9755, 1265.7669, 1286.7673, 1378.7296, 2403.5005
IRC 	

Compound: sCl 2 + SO ₂ C _{acid} 2	Energy (kJ mol⁻¹): -1074.48289929713
Reaction Coordinates: 6 -2.331526 -0.166516 0.000018 6 -0.825440 0.234423 -0.000032 8 -0.574098 1.541614 -0.000171 1 -1.391387 2.060194 -0.000217 8 0.021437 -0.606313 0.000049 9 -3.128764 0.930676 -0.000075 9 -2.620147 -0.883912 -1.084761 9 -2.620128 -0.883715 1.084931 8 3.209723 0.291203 -1.242572 16 3.045037 -0.442475 0.000064 8 3.209681 0.291561 1.242495	Frequencies (cm⁻¹): 7.9403, 13.7672, 18.1848, 27.8477, 46.3445, 67.3022, 109.4827, 252.8838, 259.5677, 385.3358, 427.0890, 505.0127, 517.1221, 570.0799, 584.6192, 689.1895, 776.5047, 793.8786, 1121.8089, 1157.6692, 1181.6738, 1193.8914, 1238.5964, 1331.6294, 1373.0344, 1866.6247, 3747.6477

Compound: sCl 2 + SO ₂ TS _{acid} 3	Energy (kJ mol⁻¹): -1074.30934048683
Reaction Coordinates: 6 1.728380 -0.129799 0.105668 6 0.489885 0.281918 -0.767563 8 0.192326 1.481797 -0.897550 8 -1.440532 0.892288 0.995355 1 0.630922 -0.217479 -1.837251 9 2.700575 0.766552 -0.040366 9 1.387182 -0.198364 1.394931 9 2.192927 -1.331188 -0.272209 16 -1.958439 -0.308316 0.343069 8 -0.484910 -0.738650 -0.812091 8 -3.158338 -0.247332 -0.463424	Frequencies (cm⁻¹): -761.9918, 33.0059, 71.7498, 91.8382, 131.5788, 187.0189, 244.1248, 277.6319, 321.4680, 337.8402, 400.8382, 428.6618, 502.3980, 517.5784, 579.7048, 670.5444, 751.5279, 779.8846, 845.4066, 977.9521, 1103.1089, 1170.8089, 1210.1752, 1223.9501, 1306.7573, 1516.0627, 2325.4651
IRC 	

Compound: sCl 2 + SO ₂ C _{acid} 3	Energy (kJ mol⁻¹): -1074.49220772753
Reaction Coordinates: 6 -2.428819 -0.098941 0.012849 6 -0.878113 -0.026969 0.000391 8 -0.182944 -1.004912 0.003846 8 -0.475453 1.235062 -0.015518 1 0.504113 1.254826 -0.019547 9 -2.838189 -1.364577 0.029163 9 -2.919259 0.521800 1.097360 9 -2.934592 0.498452 -1.078159 8 4.423124 -0.352641 0.234115 16 3.030917 -0.280874 -0.155708 8 2.369168 1.009183 0.027076	Frequencies (cm⁻¹): 10.4595, 19.9218, 36.7816, 47.8233, 71.4015, 92.8783, 126.1916, 252.5346, 254.4247, 391.1867, 425.0695, 513.8513, 528.8470, 586.4652, 685.4860, 746.0666, 799.0625, 820.6465, 1146.8988, 1155.9481, 1162.3236, 1204.1614, 1278.1810, 1321.4712, 1436.3382, 1826.7385, 3524.3309

Compound: sCl 2 + SO ₂ TS _{acid} 4	Energy (kJ mol⁻¹): -1074.30138997618
Reaction Coordinates: 6 -1.497821 -0.271097 -0.082599 6 -0.562574 0.861364 0.490426 8 -0.446143 1.902397 -0.203525 8 1.828179 -1.522474 -0.221845 1 -1.022648 1.154923 1.518209 9 -2.676354 0.257539 -0.419533 9 -0.945041 -0.819783 -1.166104 9 -1.704821 -1.230321 0.826601 16 1.970706 -0.132616 0.164348 8 0.481865 0.391824 1.297929 8 1.859808 0.923056 -0.842985	Frequencies (cm⁻¹): -441.0022, 40.1146, 58.8253, 78.2172, 101.2594, 185.7786, 238.6275, 279.3428, 340.8934, 369.1801, 377.0582, 423.5238, 486.6397, 521.1562, 576.1011, 669.4742, 749.4600, 776.2415, 832.2168, 969.7756, 1090.0897, 1173.6587, 1199.9138, 1221.0713, 1289.3918, 1448.7891, 2476.7645
IRC	
<p>The plot shows a single reaction coordinate curve represented by black dots. The x-axis is labeled "Reaction Co-ordinates (amu^{1/2} bohr)" and ranges from -15 to 15. The y-axis is labeled "Relative Energy (kJ mol⁻¹)" and ranges from 0 down to -500. The curve starts at a relative energy of about -300 at a reaction coordinate of -10, remains relatively flat until -5, then rises sharply to a peak of about -80 at 0, and finally falls back to a minimum of about -480 at 15.</p>	

Compound: sCl 2 + SO ₂ TS _{ester} 1	Energy (kJ mol⁻¹): -1074.29521365977
Reaction Coordinates: 6 1.907634 -0.201494 0.010597 6 0.319341 0.835769 -0.322223 8 0.148875 1.494348 0.703135 8 -2.502454 -1.154144 0.790211 1 0.629705 1.306123 -1.270803 9 2.172713 -0.929108 -1.038725 9 1.743607 -0.892227 1.092480 9 2.793668 0.748959 0.157964 16 -2.141148 -0.275051 -0.307714 8 -0.302258 -0.389701 -0.530099 8 -2.359547 1.167052 -0.193433	Frequencies (cm⁻¹): -376.2534, 55.8878, 58.7232, 88.2166, 111.0868, 182.5976, 209.2988, 250.7035, 260.5697, 343.0394, 358.6689, 446.6899, 496.5903, 539.0026, 555.7381, 669.7221, 704.3805, 880.2197, 1005.9261, 1070.1386, 1094.6664, 1279.1533, 1284.1591, 1334.0598, 1347.3691, 1524.3376, 2933.3175
IRC 	

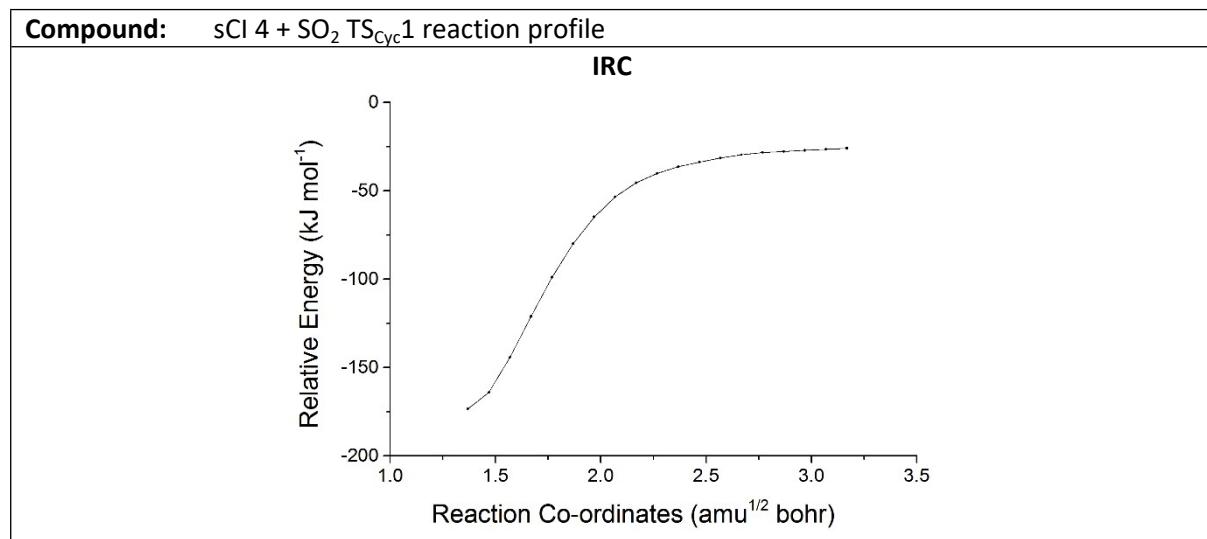
Compound: sCl 2 + SO ₂ C _{ester} 1	Energy (kJ mol⁻¹): -1074.48491076640
Reaction Coordinates: 6 2.489750 0.054945 0.021225 9 3.664742 -0.422508 0.415892 9 2.317397 1.265565 0.546732 9 2.488990 0.153804 -1.306848 8 1.530686 -0.840754 0.468921 6 0.204279 -0.610540 0.208814 8 -0.236492 0.320211 -0.389198 1 -0.374555 -1.429350 0.644969 8 -3.528716 1.114411 0.585437 16 -3.176166 0.069126 -0.356488 8 -2.916869 -1.258222 0.181918	Frequencies (cm⁻¹): 22.5722, 27.3868, 43.6691, 55.6613, 78.4418, 103.8095, 131.6737, 206.5524, 230.6056, 377.5828, 442.6376, 518.2029, 551.5669, 581.7440, 619.2136, 823.7137, 847.0023, 1047.8936, 1116.4729, 1154.9553, 1161.2201, 1215.7607, 1246.9872, 1327.6292, 1404.2327, 1818.7131, 3085.7404

Compound: sCl 2 + SO ₂ TS _{ester} 2	Energy (kJ mol⁻¹): -1074.28858393056
Reaction Coordinates:	Frequencies (cm⁻¹):
6 1.927643 -0.162049 0.052578 6 0.254333 0.308269 -0.576751 8 0.244085 1.551032 -0.533290 8 -2.324640 1.192076 0.341498 1 0.269447 -0.211321 -1.550195 9 2.037732 -1.464759 0.024301 9 2.091976 0.333440 1.240500 9 2.728281 0.388857 -0.827102 16 -2.169287 -0.261877 0.335354 8 -0.324575 -0.456447 0.437230 8 -2.641696 -1.010886 -0.821653	-370.1730, 16.1198, 59.0942, 67.3353, 109.0338, 167.2670, 226.5308, 246.5961, 279.0198, 325.4507, 365.6545, 468.5518, 508.7152, 543.8417, 559.1264, 664.1420, 698.8355, 866.9331, 984.6277, 1087.8560, 1112.9125, 1263.7941, 1268.3799, 1320.8891, 1351.4615, 1470.4502, 2929.3623

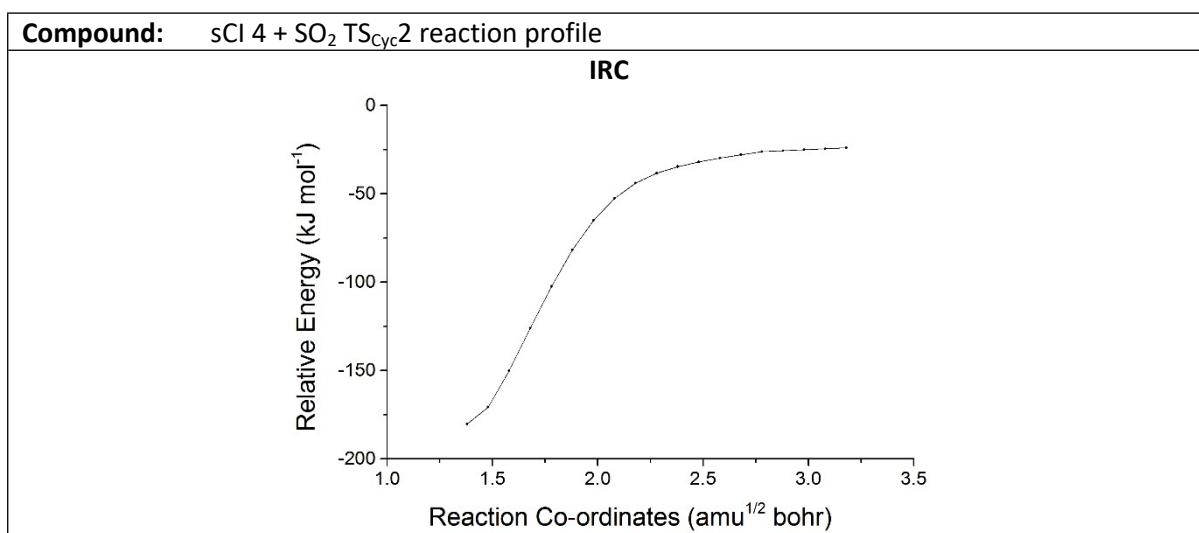
IRC

Relative Energy (kJ mol⁻¹)

Reaction Co-ordinates (amu^{1/2} bohr)



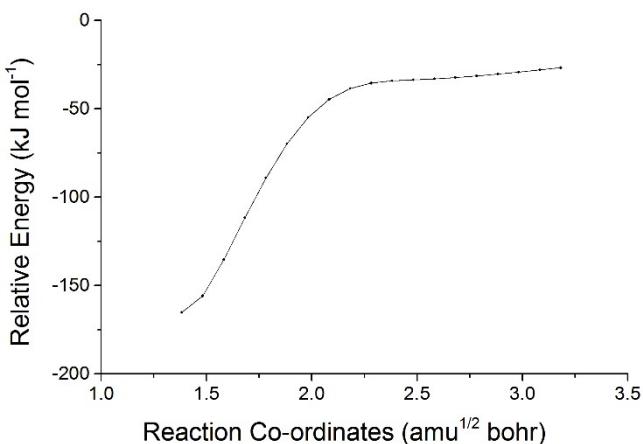
Compound: sCl 4 + SO ₂ SOZ 1	Energy (kJ mol⁻¹): -1173.53663043677
Reaction Coordinates:	Frequencies (cm⁻¹):
6 1.554889 -0.444082 0.043971 6 0.418853 0.592260 -0.202881 8 -0.143415 1.061194 1.019003 8 -1.305516 0.258857 1.272495 9 0.975223 1.683375 -0.767063 9 2.520008 0.104416 0.786549 9 1.071205 -1.509809 0.684478 9 2.079728 -0.839043 -1.119604 16 -2.122305 0.153740 -0.233675 8 -0.568098 0.060591 -0.987590 8 -2.695603 -1.168061 -0.249779	45.0068, 58.5952, 129.6858, 213.5561, 224.3170, 262.4941, 314.9923, 344.4554, 355.5024, 424.7648, 458.8257, 504.6762, 546.0177, 566.3235, 595.6943, 678.9116, 700.9905, 741.4844, 823.3340, 959.3827, 1049.1246, 1089.6695, 1186.4706, 1198.4167, 1209.5305, 1289.7024, 1326.0809



Compound: sCl 4 + SO ₂ SOZ 2	Energy (kJ mol⁻¹): -1173.54107480780
Reaction Coordinates:	Frequencies (cm⁻¹):
6 1.700331 -0.154031 -0.148664 6 0.290126 0.421615 0.178222 8 -0.275947 -0.212825 1.306963 8 -1.295685 -1.117905 0.824562 16 -2.064144 -0.354469 -0.480549 8 -0.540673 0.263838 -0.911824 8 -2.887610 0.744374 -0.032400 9 0.443800 1.719711 0.510316 9 2.513062 0.005565 0.897738 9 1.602657 -1.457920 -0.424287 9 2.227468 0.471104 -1.204541	52.0720, 62.2597, 115.6779, 186.1203, 220.2380, 287.1115, 337.0379, 366.2857, 370.7109, 444.9691, 462.2021, 513.7914, 556.2475, 589.4587, 603.7358, 673.9915, 707.4480, 739.9283, 823.3987, 922.4626, 1057.9091, 1076.2351, 1166.6656, 1200.6382, 1206.7401, 1278.0201, 1326.4546

Compound: sCl 5 + SO₂ TS_{Cyc1} reaction profile

IRC

**Compound:** sCl 4 + SO₂ SOZ3**Energy (kJ mol⁻¹):** -1173.53937137970**Reaction Coordinates:**

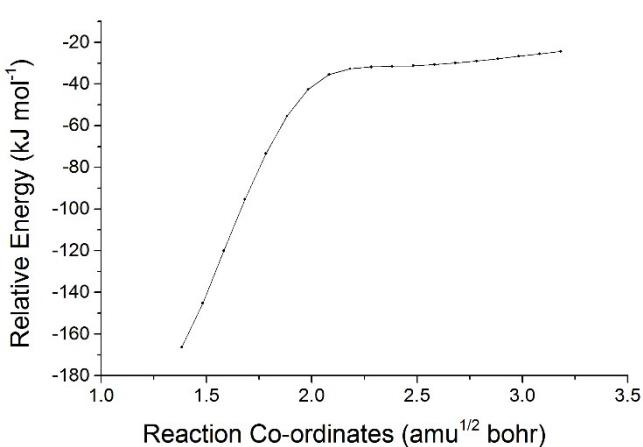
6 1.624021 -0.369006 -0.066746
6 0.353772 0.513907 0.141124
8 -0.226344 0.719616 -1.101301
8 -1.613861 1.041606 -0.800585
9 0.743711 1.684853 0.707724
9 2.468636 0.238292 -0.904194
9 2.233981 -0.549345 1.105701
9 1.290437 -1.554870 -0.569920
16 -2.139784 -0.085078 0.360085
8 -0.552474 -0.117628 0.973985
8 -2.389958 -1.378411 -0.229779

Frequencies (cm⁻¹):

53.3878, 68.0440, 151.6147,
187.7091, 228.6704, 288.5831,
322.0043, 349.3383, 381.0432,
412.5424, 495.2303, 532.1745,
559.7927, 593.1859, 611.7228,
650.3890, 708.1821, 748.7677,
862.8946, 872.8553, 1070.9764,
1093.4081, 1121.3912, 1203.8397,
1217.8460, 1280.2051, 1321.1927

Compound: sCl 5 + SO₂ TS_{Cyc2} reaction profile

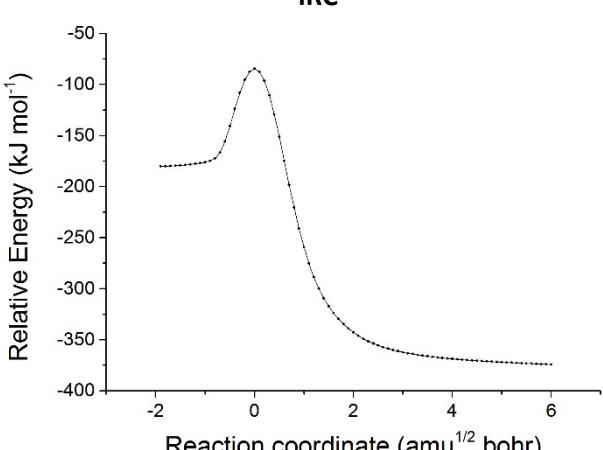
IRC



Compound: sCl 4 + SO ₂ SOZ4	Energy (kJ mol⁻¹): -1173.54521761536
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -1.778298 -0.133786 0.148059 6 -0.307417 0.102434 -0.310864 8 0.091734 1.343827 0.168329 8 1.533488 1.319434 0.113336 9 -0.271368 0.056396 -1.665292 9 -2.562575 0.837897 -0.323002 9 -2.209507 -1.304759 -0.320778 9 -1.855159 -0.145465 1.479357 16 2.090638 -0.266878 0.539013 8 0.523530 -0.863798 0.224799 8 2.995195 -0.616770 -0.528957	46.1589, 65.1210, 128.8650, 147.8182, 225.6994, 278.1023, 308.3286, 360.6921, 370.7065, 423.8983, 479.7653, 525.5699, 543.1417, 567.8888, 613.8405, 640.0293, 691.9122, 750.1205, 872.2915, 924.5181, 1078.2136, 1091.8311, 1117.2803, 1207.6709, 1212.2055, 1282.2160, 1335.4386

Compound: sCl 4 + SO ₂ TS _{H0Z1}	Energy (kJ mol⁻¹): -1173.53388111950
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -1.605497 0.411392 -0.090610 6 -0.406323 -0.535798 0.221133 8 0.108691 -1.057032 -0.980044 8 1.495851 -0.681514 -1.126061 9 -0.891300 -1.580363 0.940705 9 -2.141387 0.867559 1.044041 9 -1.185318 1.446912 -0.818228 9 -2.543014 -0.248470 -0.774978 16 2.149123 -0.051263 0.324664 8 0.557708 0.130605 0.922345 8 2.654514 1.257428 -0.003943	-159.4294, 20.7987, 61.5444, 186.1793, 208.8163, 252.6619, 309.6699, 337.1985, 356.3283, 412.3298, 461.4061, 513.1769, 554.4058, 572.6469, 597.5426, 650.9426, 700.4405, 744.3853, 849.4978, 939.8015, 1061.2802, 1087.6116, 1158.9090, 1201.5767, 1212.2239, 1288.0152, 1318.8162

Compound: sCl 4 + SO ₂ TS _{H0Z2}	Energy (kJ mol⁻¹): -1173.53931994965
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -1.762539 -0.012924 -0.172343 6 -0.279093 0.160238 0.279076 8 0.134175 -1.014808 0.898621 8 1.502083 -1.304033 0.481358 9 -0.233441 1.196656 1.158317 9 -2.208773 1.122235 -0.711780 9 -1.857571 -0.987962 -1.076651 9 -2.522903 -0.320903 0.880330 16 2.110800 -0.109799 -0.563686 8 0.517626 0.412641 -0.816737 8 2.831265 0.879033 0.202589	-73.1102, 60.9569, 89.3910, 152.7468, 210.2652, 280.6427, 305.0765, 360.0640, 369.2964, 423.3978, 478.9291, 527.1686, 549.7716, 587.5226, 604.3232, 669.3209, 698.7581, 748.6942, 858.8646, 895.0221, 1069.7381, 1085.3498, 1119.4200, 1206.0798, 1212.6063, 1280.9879, 1324.0728

Compound: sCl 4 + SO ₂ TS ₂ O ₃ 1	Energy (kJ mol⁻¹): -1173.50370754569
Reaction Coordinates: 6 -1.668664 -0.237000 -0.168337 6 -0.413921 0.543417 0.338601 8 0.072233 0.141382 1.476579 8 3.097138 0.560455 -0.262500 9 -0.662681 1.866634 0.289683 9 -1.413638 -1.547102 -0.141910 9 -2.696874 0.019967 0.647230 9 -2.010279 0.093547 -1.420071 16 1.984108 -0.352264 -0.343062 8 0.640213 0.337422 -0.821845 8 1.415545 -1.051721 0.869394	Frequencies (cm⁻¹): -607.2684, 31.0897, 62.5197, 107.9680, 179.7528, 222.9153, 286.2726, 331.5185, 356.5920, 359.3231, 416.2636, 459.0275, 516.0829, 524.9626, 555.8419, 589.5866, 657.4417, 730.7599, 799.7264, 891.7564, 939.3861, 1099.1259, 1187.0108, 1199.5459, 1221.4319, 1289.6165, 1338.8248
IRC 	

Compound: sC4 + SO ₂ CS ₂ O ₃ 1	Energy (kJ mol⁻¹): -1173.62784782028
Reaction Coordinates: 6 2.480382 -0.197438 -0.000000 6 0.990324 0.225637 0.000003 8 0.063167 -0.503181 0.000008 8 -2.590732 0.669827 -1.245757 9 0.890156 1.550175 0.000000 9 3.086529 0.290825 1.087804 9 2.581707 -1.519403 0.000001 9 3.086524 0.290823 -1.087808 16 -2.691450 -0.044144 -0.000001 8 -2.590737 0.669888 1.245720 8 -2.952357 -1.458369 0.000033	Frequencies (cm⁻¹): 10.6507, 20.4972, 25.4670, 42.7300, 56.9356, 89.3577, 107.8055, 231.8421, 243.1069, 383.0292, 423.8821, 467.4804, 509.3791, 509.7174, 513.3808, 586.9497, 694.4524, 769.1260, 806.4467, 1047.9805, 1106.0690, 1173.2322, 1229.7700, 1312.0304, 1364.8420, 1369.6582, 1912.9001

Compound:	sCl 4 + SO ₂ TS ₂ O ₃ 2	Energy (kJ mol⁻¹):	-1173.51911565224
Reaction Coordinates:	Frequencies (cm⁻¹):		
6 1.641991 -0.377135 0.066413 6 0.404346 0.615056 0.020625 8 -0.092071 0.826587 1.179483 8 -1.946170 1.079105 0.567999 9 0.765621 1.706244 -0.698967 9 2.081261 -0.628174 -1.164642 9 2.621593 0.200698 0.760119 9 1.298166 -1.516645 0.654253 16 -2.052978 -0.132169 -0.311558 8 -0.570151 -0.129146 -0.905302 8 -2.432874 -1.423038 0.221049	-454.1605, 58.6418, 74.9140, 139.1367, 193.4154, 225.2846, 294.4132, 309.7366, 334.7464, 376.4847, 434.5312, 460.9770, 527.9330, 545.8351, 554.8277, 580.4535, 666.8636, 739.8676, 800.0289, 902.6551, 974.8452, 1057.3650, 1180.4144, 1203.0018, 1225.8099, 1268.3720, 1310.7709		

IRC

The plot shows a single curve representing the reaction coordinate. The y-axis is labeled "Relative Energy (kJ mol⁻¹)" and ranges from -400 to -100. The x-axis is labeled "Reaction coordinate (amu^{1/2} bohr)" and ranges from -2 to 6. The curve starts at a relative energy of about -200 at a reaction coordinate of -1, rises to a peak of about -100 at a reaction coordinate of 0, and then gradually decreases to approximately -380 at a reaction coordinate of 6.

Compound: sCl 4 + SO ₂ TS ₂ O ₃	Energy (kJ mol⁻¹): -1173.51026243837
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -1.780248 -0.026481 -0.181348 6 -0.330964 0.078455 0.433961 8 0.018791 -0.951516 1.110358 8 1.770743 -1.335123 0.367013 9 -0.228904 1.275837 1.075993 9 -2.059986 1.043211 -0.922984 9 -1.873485 -1.118473 -0.933310 9 -2.663251 -0.100650 0.814656 16 2.057744 -0.116000 -0.470885 8 0.547504 0.247348 -0.835633 8 2.809712 0.994895 0.071922	-517.8014, 60.7850, 75.0296, 117.7628, 168.3618, 223.2707, 288.8933, 321.0568, 357.7665, 401.9041, 413.9117, 439.8401, 510.8812, 541.3957, 552.9248, 588.5194, 646.5623, 741.3055, 796.7976, 905.0981, 950.8611, 1049.9909, 1201.1346, 1203.7136, 1221.8266, 1267.7164, 1289.7390

IRC

The plot shows a single curve representing the reaction coordinate. The y-axis is labeled "Relative Energy (kJ mol⁻¹)" and ranges from -400 to -100. The x-axis is labeled "Reaction coordinate (amu^{1/2} bohr)" and ranges from -2 to 6. The curve starts at a relative energy of about -180 at a reaction coordinate of -1, remains relatively flat until x=0, then rises to a peak of about -95 at x=0, and then gradually decreases, leveling off around -380 at x=6.

Compound: sCl 4 + SO ₂ TS _{ester} 1	Energy (kJ mol⁻¹): -1173.49144000029
Reaction Coordinates:	Frequencies (cm⁻¹):
6 1.813541 -0.425632 0.042717 6 0.274173 0.730620 0.067894 8 0.066864 0.917325 1.254961 8 -2.497059 -1.481374 0.252421 9 0.702671 1.753524 -0.724793 9 2.118193 -0.700091 -1.194231 9 1.518967 -1.476391 0.744034 9 2.726344 0.316466 0.605889 16 -2.171217 -0.197071 -0.337911 8 -0.408018 -0.192280 -0.697802 8 -2.334586 1.041535 0.423522	-432.3408, 52.1672, 65.8122, 74.4737, 101.0197, 173.4216, 204.0668, 237.4051, 246.1582, 261.4374, 324.8358, 399.0983, 480.5209, 522.8040, 543.1451, 550.5618, 617.6896, 642.3351, 750.8306, 865.2657, 965.8175, 1056.6910, 1092.7159, 1283.7314, 1295.9325, 1334.5393, 1561.5094

IRC

Relative Energy (kJ mol⁻¹)

Reaction coordinate (amu^{1/2} bohr)

Compound: sCl 4 + SO ₂ C _{ester} 1	Energy (kJ mol⁻¹): -1173.67282156150
Reaction Coordinates:	Frequencies (cm⁻¹):
6 2.246873 -0.570136 0.101837 8 1.940832 0.791280 0.167279 6 0.721378 1.198058 -0.243863 8 -0.171463 0.557311 -0.672656 9 0.693301 2.509378 -0.080159 9 1.426723 -1.287616 0.858143 9 2.198158 -1.015942 -1.147950 9 3.482804 -0.683183 0.560926 16 -3.222589 -0.092447 -0.196467 8 -3.110889 0.436899 1.151248 8 -3.215600 -1.534503 -0.361245	4.0162, 9.3991, 10.7021, 33.6407, 46.7280, 80.6640, 87.7644, 108.0431, 178.9699, 377.6690, 401.8068, 427.2020, 516.1418, 551.9297, 609.9868, 669.3564, 739.3163, 777.5055, 882.0199, 1027.4147, 1136.8106, 1158.2408, 1231.0917, 1248.0268, 1282.9061, 1335.0611, 1909.4847

Compound: sCl 4 + SO ₂ TS _{ester} 2	Energy (kJ mol⁻¹): -1173.48550327144
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -1.928820 -0.043969 -0.274499 9 -2.756202 0.164973 0.713137 9 -2.017816 0.864366 -1.198659 9 -2.036458 -1.257296 -0.737470 6 -0.230013 0.172164 0.564537 8 -0.217642 1.305626 1.000133 9 -0.229429 -0.888169 1.431347 8 0.377679 -0.215843 -0.633797 16 2.166005 -0.028066 -0.515698 8 2.322683 1.308562 0.047720 8 2.724287 -1.182719 0.165411	-484.6874, 18.2260, 52.9006, 73.6833, 96.9497, 157.9094, 184.7185, 237.6844, 242.4637, 268.2822, 319.2501, 401.4810, 492.4707, 513.6895, 538.3604, 550.0115, 599.3580, 656.1937, 759.4240, 827.2184, 963.3271, 1011.3589, 1097.8469, 1282.8564, 1293.6220, 1328.5142, 1574.4308
IRC	
<p>Relative Energy (kJ mol⁻¹)</p> <p>Reaction coordinate (amu^{1/2} bohr)</p>	

Compound: sCl 4 + SO ₂ C _{ester} 2	Energy (kJ mol⁻¹): -1173.67290696018
Reaction Coordinates:	Frequencies (cm⁻¹):
6 2.575150 -0.328946 0.000001 9 2.302730 -1.047093 -1.083133 9 3.857326 -0.002343 -0.000075 9 2.302824 -1.046954 1.083252 8 1.874870 0.878202 -0.000047 6 0.524808 0.841279 0.000013 9 0.119153 2.097405 -0.000051 8 -0.191560 -0.096475 0.000103 8 -3.554650 0.056304 -1.243433 16 -3.276802 -0.639925 0.000012 8 -3.554811 0.056426 1.243351	5.3901, 13.1636, 13.2084, 35.5690, 48.4244, 83.3192, 87.0781, 108.4973, 181.3048, 378.5363, 402.3076, 427.2285, 516.3645, 552.1696, 609.9813, 670.0603, 739.8784, 777.3801, 881.6767, 1029.0054, 1139.0554, 1158.2309, 1228.1209, 1248.4840, 1284.6130, 1334.6173, 1907.1417

S8.4 Reactions with HNO₃

Compound: HNO ₃	Energy (kJ mol⁻¹): -280.649387120256
Reaction Coordinates: 7 0.152578 0.031983 0.000001 8 -1.152277 -0.512972 -0.000001 1 -1.719886 0.275568 0.000002 8 0.214973 1.237996 -0.000000 8 1.018784 -0.787456 -0.000000	Frequencies (cm⁻¹): 479.4853, 584.8104, 646.7663, 783.1012, 896.0178, 1315.3030, 1341.5904, 1744.4127, 3711.8876

Compound: sCl 1 + HNO ₃ PRC	Energy (kJ mol⁻¹): -470.072694506353
Reaction Coordinates: 7 1.496892 -0.134866 -0.004943 8 0.667020 -1.033235 0.097948 8 2.688425 -0.260005 -0.068599 8 1.033607 1.141055 -0.058826 1 0.017702 1.095134 0.053631 6 -2.072567 -1.012782 0.120554 1 -2.498385 -1.822306 -0.459681 1 -1.658928 -1.120610 1.114625 8 -2.091056 0.101544 -0.429116 8 -1.535900 1.159208 0.283930	Frequencies (cm⁻¹): 52.2249, 78.8218, 102.2949, 138.9453, 196.7530, 265.7980, 511.6154, 649.6811, 676.6982, 698.8865, 799.8732, 841.0081, 972.2831, 981.5672, 1060.8488, 1235.4845, 1308.1266, 1425.6956, 1481.8275, 1597.1874, 1708.0574, 2730.1492, 3121.7345, 3266.4971

Compound: sCl 1 + HNO ₃ TS	Energy (kJ mol⁻¹): -470.068648204036
Reaction Coordinates: 7 1.382578 -0.063861 -0.008349 8 0.628369 -1.058966 0.034369 8 2.580436 -0.135427 -0.065503 8 0.846896 1.149168 0.004906 1 -0.251112 1.114260 0.091169 6 -1.671765 -1.057011 0.209054 1 -1.838608 -2.006498 -0.286142 1 -1.395054 -0.967906 1.249688 8 -1.962928 -0.044666 -0.454559 8 -1.613108 1.171046 0.199461	Frequencies (cm⁻¹): -248.9265, 39.6720, 92.4916, 198.5650, 324.0410, 328.8518, 500.7085, 660.3195, 702.1595, 736.2226, 801.4920, 819.3851, 1019.9294, 1047.0691, 1133.6644, 1160.2011, 1235.8102, 1425.0247, 1489.9202, 1598.7784, 1614.2350, 1818.4260, 3129.3696, 3269.8948
IRC	
<p>The plot shows a smooth curve representing the potential energy surface. The y-axis is labeled "Relative Energy (kJ mol⁻¹)" and ranges from -200 to -50. The x-axis is labeled "Reaction coordinate (amu^{1/2} bohr)" and ranges from -2 to 4. The curve starts at a high energy point around -470 kJ mol⁻¹ at a reaction coordinate of -2, remains relatively flat until a coordinate of 0, then drops sharply to around -180 kJ mol⁻¹ at a coordinate of 4.</p>	

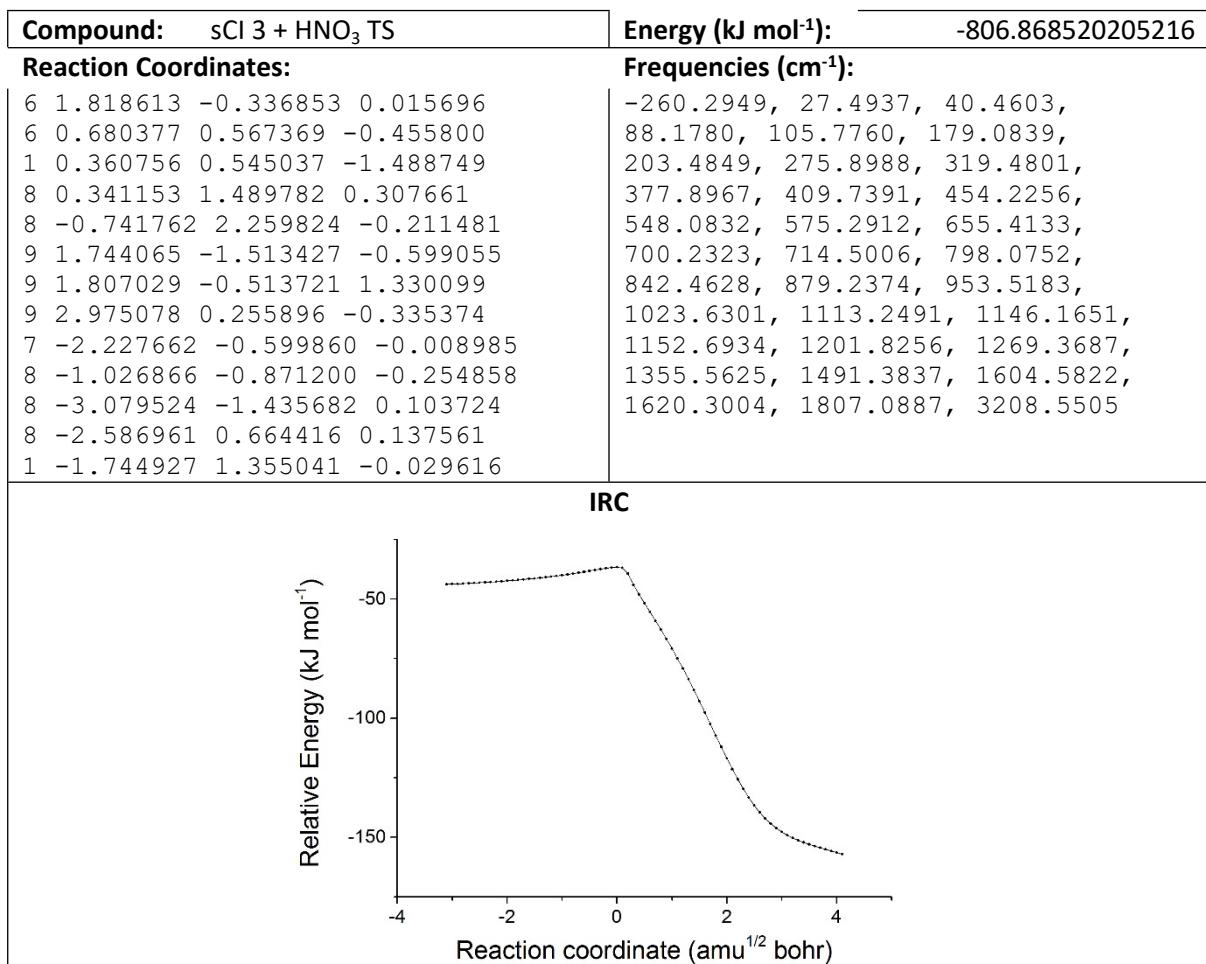
Compound: sCl 1 + HNO ₃ Pr	Energy (kJ mol⁻¹): -470.123652687780
Reaction Coordinates:	Frequencies (cm⁻¹):
7 1.319520 0.088216 0.023524 8 0.339308 -0.963575 -0.181566 8 2.203213 0.022791 -0.773304 8 1.153464 0.838501 0.944072 1 -2.557702 1.413878 -0.088671 6 -0.881066 -0.827107 0.545024 1 -1.176279 -1.846124 0.792990 1 -0.722852 -0.212452 1.427306 8 -1.915046 -0.352282 -0.253432 8 -1.717616 1.078293 -0.431574	59.4191, 85.4098, 119.9476, 176.7737, 319.2844, 409.9451, 490.4761, 609.3846, 649.9965, 762.5552, 827.8298, 879.8808, 967.8881, 1053.3829, 1314.9395, 1324.3246, 1376.4783, 1413.3481, 1450.9690, 1729.9552, 3075.3309, 3143.0899, 3755.3317

Compound: sCl 2 + HNO ₃ PRC	Energy (kJ mol⁻¹): -806.869427684078
Reaction Coordinates:	Frequencies (cm⁻¹):
6 2.515030 -0.052204 -0.107496 6 1.389569 -0.745025 0.644316 1 1.560507 -1.689924 1.142933 8 0.227784 -0.294611 0.738235 8 -0.080258 0.887326 0.149683 9 2.227109 0.088467 -1.399867 9 3.605581 -0.830507 0.003357 9 2.793296 1.139411 0.420540 7 -3.324819 -0.165770 -0.101840 8 -2.563876 -1.114006 -0.129467 8 -4.520237 -0.167655 -0.204149 8 -2.760562 1.079774 0.072769 1 -1.771052 0.930738 0.116199	10.6140, 29.7830, 41.8008, 71.7241, 87.0513, 112.6532, 156.6378, 230.4114, 250.4582, 346.9186, 480.9087, 511.7626, 534.9130, 590.1757, 637.4624, 683.9769, 759.8603, 795.8386, 829.4514, 880.5639, 898.0185, 915.0071, 944.1144, 1161.7325, 1192.1409, 1241.8657, 1328.2504, 1371.8255, 1456.0238, 1578.6816, 1718.6016, 3124.6857, 3206.9984

Compound: sCl 2 + HNO ₃ TS	Energy (kJ mol⁻¹): -806.865419736050
Reaction Coordinates:	Frequencies (cm⁻¹):
6 1.457042 -0.475653 -0.141690 6 0.852838 0.357702 1.001659 1 1.011599 -0.015041 2.007643 8 0.497497 1.549201 0.927013 8 0.160563 1.992571 -0.376021 9 2.631826 0.100273 -0.447165 9 0.735781 -0.596154 -1.240535 9 1.705108 -1.697626 0.349632 7 -2.096267 -0.333890 0.053274 8 -1.146941 -0.657051 0.801488 8 -3.113098 -0.960538 -0.038094 8 -1.997430 0.768090 -0.684379 1 -1.056180 1.263359 -0.537811	-203.8987, 49.4921, 52.7535, 87.1304, 156.0214, 182.5010, 225.3687, 278.4959, 313.8332, 331.5867, 435.8589, 494.9747, 533.3277, 566.1095, 672.6409, 713.0831, 749.8274, 795.0142, 821.6187, 871.5104, 928.1138, 1012.3118, 1072.1094, 1156.5460, 1157.5824, 1203.4557, 1278.5401, 1364.2600, 1517.8599, 1612.8964, 1678.5251, 1999.6916, 3181.7398
IRC	
<p>The plot shows a reaction coordinate (IRC) with the y-axis labeled "Relative Energy (kJ mol⁻¹)" ranging from 0 to -150 and the x-axis labeled "Reaction coordinate (amu^{1/2} bohr)" ranging from -4 to 4. The energy curve starts at approximately -45 kJ mol⁻¹ at x = -4, remains relatively flat until x = 0, then drops sharply to about -150 kJ mol⁻¹ at x = 2.5.</p>	

Compound: sCl 2 + HNO ₃ Pr	Energy (kJ mol⁻¹): -806.932710531538
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -1.234029 -0.565793 0.129147 6 -0.349089 0.145478 -0.926576 1 -0.712055 -0.170001 -1.906167 8 -0.403111 1.520637 -0.953015 8 -0.576378 2.097828 0.361211 1 0.337481 2.085119 0.692904 8 0.970650 -0.458847 -0.960221 7 1.946953 -0.186099 0.064084 8 2.831375 -0.979569 0.047569 8 1.800977 0.790017 0.754362 9 -2.448500 -0.017571 0.164558 9 -0.709971 -0.557213 1.359726 9 -1.368583 -1.853111 -0.240951	44.3668, 69.3809, 93.2717, 165.4610, 193.4546, 225.3272, 240.7962, 311.0701, 340.0379, 373.9625, 427.2332, 497.8110, 521.5399, 571.7072, 598.0567, 692.9346, 751.7581, 791.1962, 835.0732, 879.9702, 908.8883, 950.6032, 1108.0501, 1146.4447, 1199.1320, 1261.5324, 1325.5211, 1370.8742, 1383.7320, 1426.0054, 1726.0957, 3084.8214, 3675.8920

Compound: sCl 3 + HNO ₃ PRC	Energy (kJ mol⁻¹): -806.873300576158
Reaction Coordinates:	Frequencies (cm⁻¹):
6 2.112628 -0.328139 0.028361 6 0.908402 0.441619 -0.483405 1 0.465875 0.214030 -1.445228 8 0.470057 1.364109 0.223449 8 -0.606103 2.078684 -0.261243 9 1.843444 -1.636496 -0.000813 9 2.452909 0.025482 1.261816 9 3.148127 -0.092228 -0.796140 7 -2.653927 -0.536448 -0.005582 8 -1.512598 -0.885133 -0.284588 8 -3.621877 -1.233995 0.095295 8 -2.859805 0.794106 0.223427 1 -1.972286 1.247236 0.060069	21.4603, 39.8574, 42.1601, 82.6932, 121.6740, 130.1110, 185.7289, 193.1965, 279.0018, 384.2481, 408.0808, 414.2012, 549.8226, 560.8570, 642.0486, 692.4597, 699.3590, 796.8093, 883.5112, 893.9237, 921.1985, 934.1911, 964.5752, 1152.1060, 1186.0423, 1272.6374, 1309.2140, 1361.4585, 1467.2626, 1597.6705, 1714.9508, 2962.3969, 3199.0004



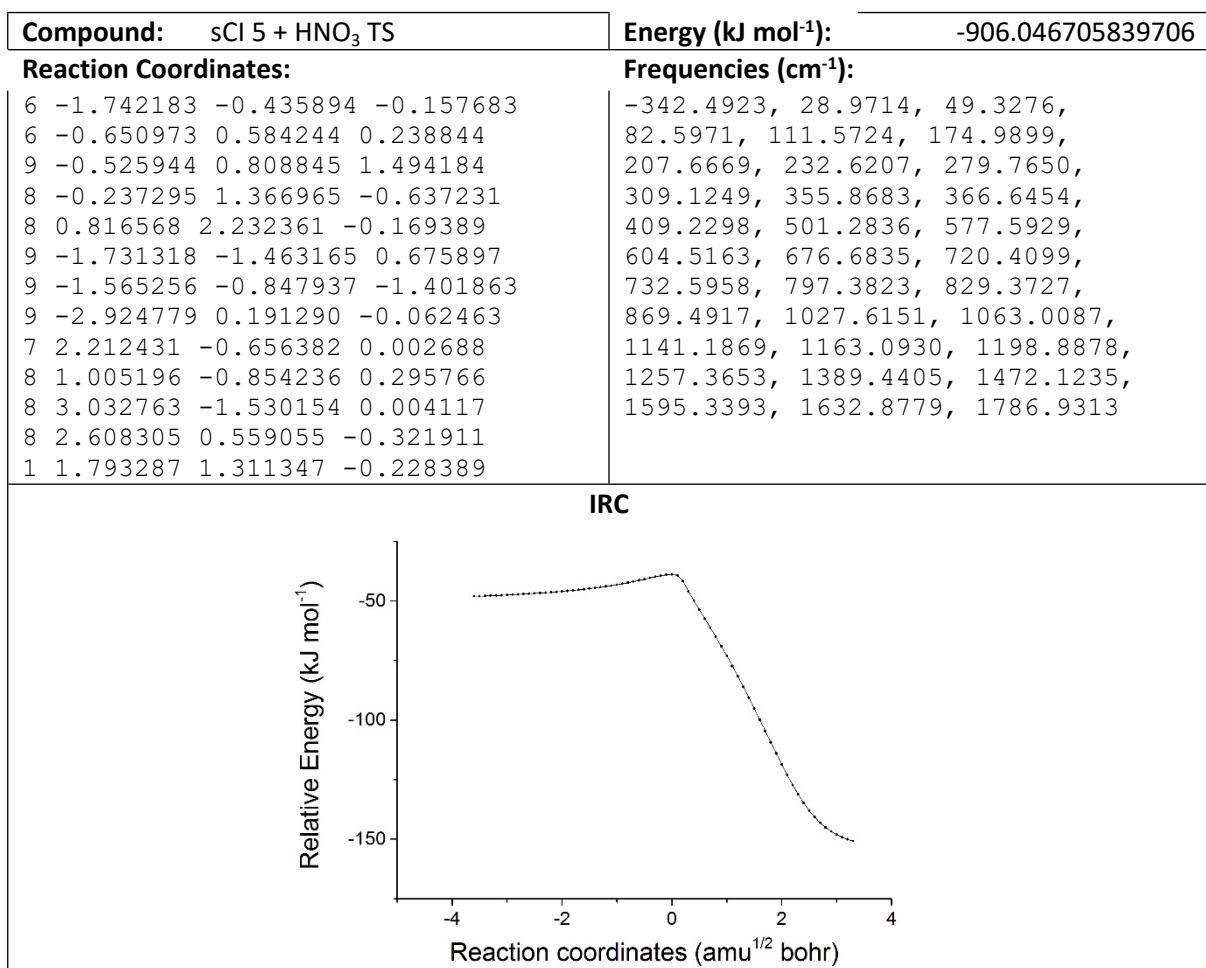
Compound: sCl 3 + HNO ₃ Pr	Energy (kJ mol⁻¹): -806.931663308635
Reaction Coordinates:	Frequencies (cm⁻¹):
6 1.554013 -0.423933 -0.077892 6 0.164526 0.246695 -0.151113 1 -0.026945 0.562928 -1.173348 8 0.130797 1.346934 0.710919 8 -0.158323 2.541173 -0.049835 1 -1.116845 2.619476 0.077256 8 -0.738620 -0.763913 0.274806 7 -2.123510 -0.506101 -0.059810 8 -2.841991 -1.374969 0.312222 8 -2.360662 0.517636 -0.645814 9 1.603211 -1.485321 -0.897005 9 1.854320 -0.826693 1.159300 9 2.481083 0.455219 -0.476695	49.4586, 57.3126, 79.0791, 84.2792, 143.7444, 200.6062, 254.1429, 293.8911, 327.8966, 346.1094, 396.5055, 460.3064, 534.3509, 564.3218, 632.4380, 686.2828, 735.0960, 768.6991, 831.3235, 896.1398, 956.4319, 1034.0722, 1083.2847, 1165.6245, 1186.5185, 1265.2167, 1315.8773, 1363.6233, 1367.2948, 1402.3144, 1735.4845, 3121.9660, 3715.7873

Compound: sCl 4 + HNO ₃ PRC	Energy (kJ mol⁻¹): -906.045603878535
Reaction Coordinates:	Frequencies (cm⁻¹):
6 1.658334 0.372214 0.473504 6 1.097655 -0.333428 -0.787865 9 1.505255 0.115087 -1.928174 8 0.375536 -1.332887 -0.814800 8 -0.098135 -1.807086 0.428713 9 0.703171 0.787599 1.281294 9 2.381676 1.417913 0.071727 9 2.455978 -0.487057 1.107298 7 -2.601910 0.376304 -0.044351 8 -1.580861 0.753597 -0.602810 8 -3.656535 0.942884 0.004730 8 -2.555480 -0.829958 0.598602 1 -1.613484 -1.181122 0.491835	16.2163, 36.9939, 57.1225, 63.5689, 97.7905, 148.4175, 177.8060, 206.6922, 232.4838, 295.4329, 326.5404, 364.4840, 479.4276, 504.9650, 581.2690, 625.4322, 646.2131, 689.5643, 693.3045, 778.4011, 796.2974, 908.9884, 915.6068, 960.0898, 1154.7636, 1195.1805, 1236.8306, 1317.4677, 1410.4083, 1478.1360, 1633.9975, 1716.9084, 2949.9386

Compound: sCl 4 + HNO ₃ TS	Energy (kJ mol⁻¹): -906.045088871693
Reaction Coordinates:	Frequencies (cm⁻¹):
6 1.398496 -0.436428 -0.468817 6 0.867915 0.352803 0.765201 9 1.204294 -0.149419 1.903834 8 0.443344 1.516117 0.791631 8 -0.032299 1.992907 -0.477231 9 1.702773 -1.669054 -0.072328 9 2.524212 0.191170 -0.837436 9 0.572753 -0.496022 -1.489867 7 -2.249696 -0.361885 0.046000 8 -1.207449 -0.703022 0.633327 8 -3.263342 -0.998166 0.032105 8 -2.260915 0.801357 -0.617246 1 -1.321586 1.271330 -0.538819	-59.3438, 41.5877, 45.9090, 81.3198, 146.6454, 182.1514, 202.5032, 222.6189, 271.4277, 289.0449, 340.0328, 368.7527, 476.7036, 511.9915, 584.5135, 618.8684, 662.2343, 671.8021, 717.1168, 778.5606, 796.8572, 913.2326, 1001.6289, 1052.8716, 1137.9920, 1188.2205, 1246.8738, 1258.8145, 1399.1859, 1525.1097, 1626.9631, 1707.7546, 2259.7416
IRC <p>Relative Energy (kJ mol⁻¹)</p> <p>O-H bond length (Å)</p>	

Compound: sCl 4 + HNO ₃ Pr	Energy (kJ mol⁻¹): -906.106548786604
Reaction Coordinates:	Frequencies (cm⁻¹):
6 1.227307 -0.388855 -0.480905 6 0.286623 -0.045093 0.725794 9 0.807811 -0.639351 1.815327 8 0.136881 1.282818 1.051278 8 0.468229 2.163071 -0.038453 1 -0.399261 2.247293 -0.469648 8 -0.957084 -0.760446 0.587626 7 -2.061352 -0.183353 -0.162301 8 -3.001400 -0.904766 -0.146699 8 -1.908607 0.888922 -0.681185 9 2.401415 0.221046 -0.353700 9 0.674689 -0.061869 -1.653575 9 1.431750 -1.710595 -0.479619	44.8465, 68.6884, 90.0800, 170.5517, 192.8878, 215.5921, 220.9380, 292.2829, 303.6735, 329.7248, 378.8194, 385.4820, 439.3492, 486.2138, 534.3046, 589.1877, 605.8283, 733.3051, 735.8503, 759.7463, 794.2091, 817.1514, 973.8141, 1002.2183, 1112.1058, 1174.8338, 1190.1717, 1216.3470, 1324.8542, 1344.1334, 1428.7677, 1744.5606, 3684.7511

Compound: sCl 5 + HNO ₃ PRC	Energy (kJ mol⁻¹): -906.050679111630
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -2.024036 -0.408554 -0.156972 6 -0.898435 0.547176 0.259645 9 -0.631948 0.608199 1.510031 8 -0.341672 1.256494 -0.586459 8 0.674143 2.112512 -0.136549 9 -1.770253 -1.623626 0.317061 9 -2.134048 -0.448742 -1.475439 9 -3.171585 0.032730 0.373663 7 2.638426 -0.592217 -0.022548 8 1.462692 -0.892302 0.143819 8 3.584626 -1.326169 -0.000491 8 2.919907 0.722010 -0.265209 1 2.038779 1.216373 -0.226934	-342.4923, 28.9714, 49.3276, 82.5971, 111.5724, 174.9899, 207.6669, 232.6207, 279.7650, 309.1249, 355.8683, 366.6454, 409.2298, 501.2836, 577.5929, 604.5163, 676.6835, 720.4099, 732.5958, 797.3823, 829.3727, 869.4917, 1027.6151, 1063.0087, 1141.1869, 1163.0930, 1198.8878, 1257.3653, 1389.4405, 1472.1235, 1595.3393, 1632.8779, 1786.9313

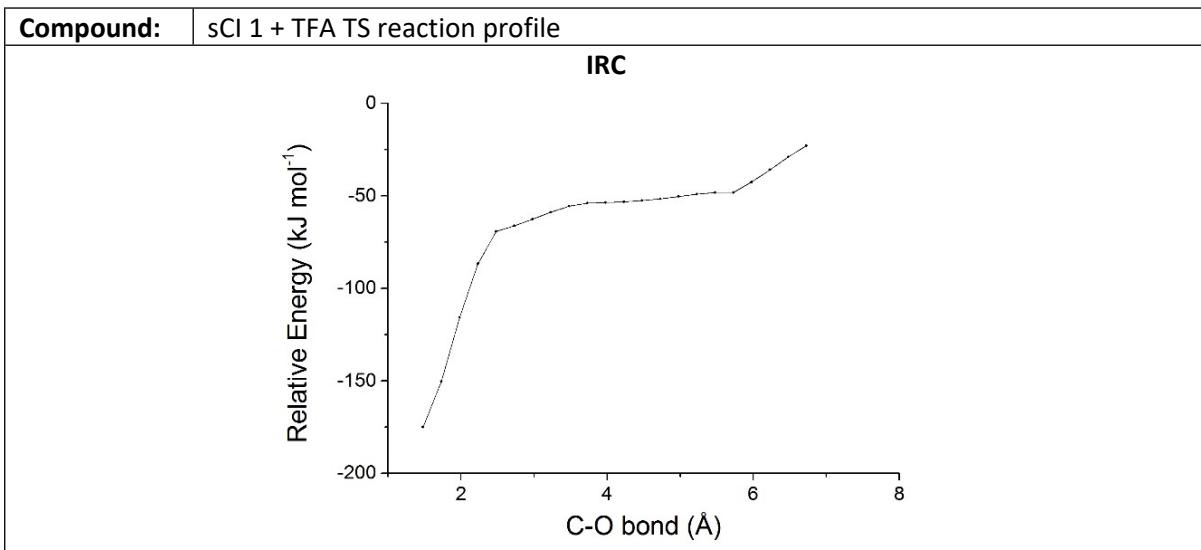


Compound: sCl 5 + HNO ₃ Pr	Energy (kJ mol⁻¹): -906.107641820181
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -1.465714 -0.616502 -0.018853 6 -0.115197 0.174139 0.129909 9 -0.052168 0.626231 1.411907 8 0.029375 1.215056 -0.779855 8 -0.915161 2.264455 -0.484275 1 -0.365195 2.867713 0.039005 8 0.871098 -0.747902 -0.156537 7 2.324097 -0.279474 -0.042045 8 3.035235 -1.130435 -0.445880 8 2.511379 0.791426 0.427935 9 -1.437120 -1.713019 0.746717 9 -1.643942 -0.981473 -1.289204 9 -2.497153 0.135148 0.363674	35.8609, 48.3223, 97.7733, 106.6229, 136.6438, 216.4055, 218.5739, 253.9864, 277.4184, 303.7563, 362.2100, 388.0438, 420.9834, 486.3317, 532.0771, 574.0953, 591.9207, 684.5924, 740.6256, 752.6333, 787.4242, 825.9496, 976.3362, 1070.5744, 1118.2801, 1131.5996, 1199.1168, 1211.7936, 1299.2723, 1364.2598, 1411.7186, 1805.9040, 3725.8754

S8.5 Reactions with TFA

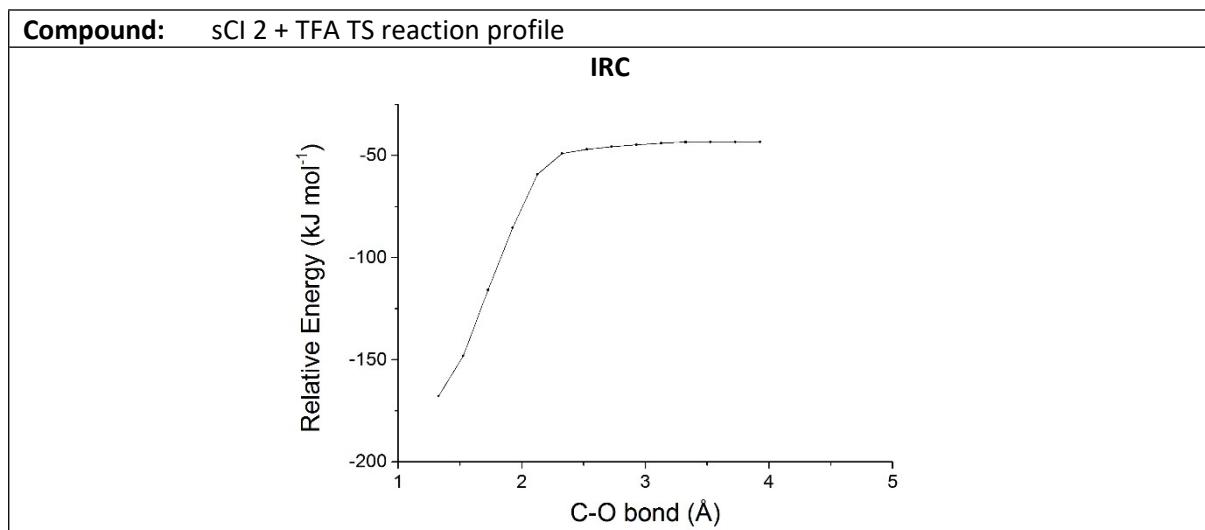
Compound: TFA (CF_3COOH) conformer 1	Energy (kJ mol⁻¹): -526.396690747069
Reaction Coordinates:	Frequencies (cm⁻¹):
6 0.600232 -0.000700 0.000001	30.3904, 237.0884, 242.9524,
6 -0.943150 0.162408 0.000002	384.3951, , 420.8035, 503.5971,
8 -1.495890 1.220466 0.000000	582.5530, 595.2069, 666.0224,
8 -1.526983 -1.040018 0.000002	787.6331, 791.1373, 1135.0381,
1 -2.487822 -0.910329 -0.000003	1156.4050, 1192.2434, 1246.2417,
9 0.999822 -0.678016 -1.087812	1400.8344, 1853.8568, 3732.4833
9 1.192393 1.189052 0.000043	
9 0.999820 -0.678091 1.087766	

Compound: sCI 1 + TFA PRC	Energy (kJ mol⁻¹): -715.814752163838
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -2.117654 -0.201348 -0.000023	15.3101, 29.2792, 36.3834, 50.1211,
6 -0.740521 0.519046 0.000004	77.4504, 101.5880, 169.1057,
8 -0.649473 1.715143 -0.000182	257.8198, 279.2418, 395.5788,
8 0.232966 -0.368608 0.000253	433.3071, 515.8541, 550.4847,
1 1.134260 0.077798 0.000265	591.1333, 681.1988, 706.1226,
9 -2.248565 -0.983360 1.087847	780.6440, 807.7600, 880.7623,
9 -3.118080 0.678690 -0.000299	972.0749, 1018.4284, 1141.8677,
9 -2.248314 -0.983755 -1.087640	1169.1472, 1202.8182, 1242.1331,
6 4.708687 -0.314834 -0.000181	1335.0572, 1418.3576, 1486.5195,
1 5.296134 -1.223034 -0.000511	1576.3894, 1824.8056, 3021.8298,
1 5.113259 0.690859 0.000074	3114.8857, 3264.5288
8 3.472618 -0.479587 -0.000102	
8 2.679877 0.637179 0.000306	



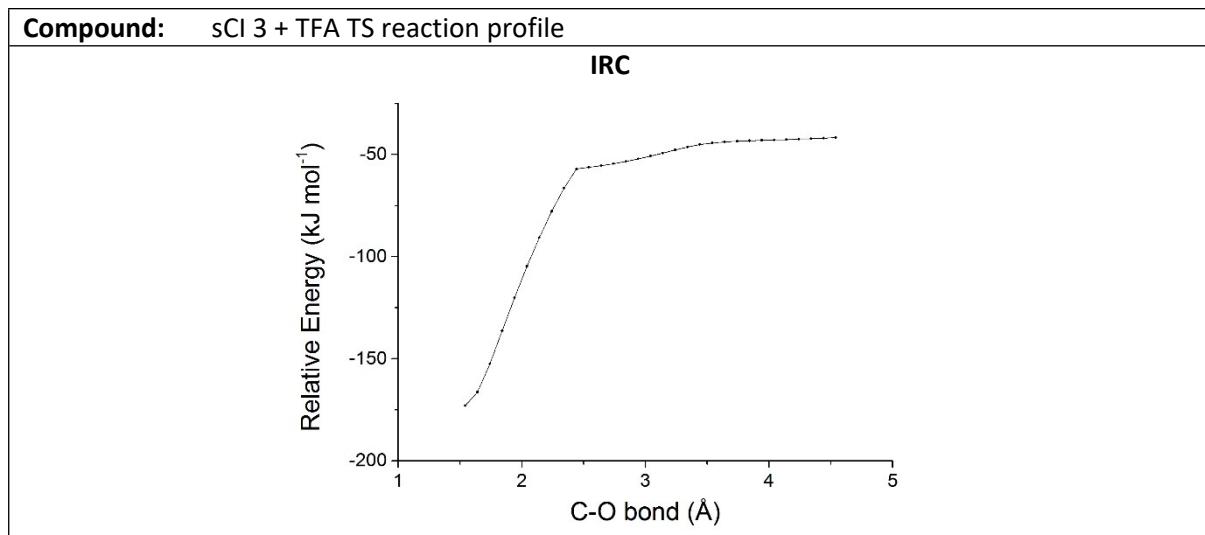
Compound: sCl 1 + TFA Pr	Energy (kJ mol⁻¹): -715.871685168671
Reaction Coordinates:	Frequencies (cm⁻¹):
6 1.697881 0.022687 0.042552 6 0.163468 0.215542 -0.116231 8 -0.437279 -0.958119 -0.095178 8 -0.330901 1.305557 -0.235475 1 -2.196182 1.415419 -0.011856 9 2.182883 -0.686901 -0.989582 9 2.314212 1.200094 0.072752 9 1.976930 -0.636459 1.175780 6 -1.892893 -1.058709 -0.328960 1 -2.076150 -2.112990 -0.142555 1 -2.090489 -0.767998 -1.358185 8 -2.652048 -0.337966 0.557765 8 -3.044039 0.927759 -0.027376	26.1591, 61.2351, 109.7878, 146.1666, 202.8784, 235.0052, 284.5678, 340.3942, 425.7682, 460.6063, 515.6973, 550.8166, 556.4842, 587.4142, 741.1113, 785.3393, 817.4884, 856.4786, 912.6225, 1084.6461, 1146.4313, 1157.7037, 1164.2388, 1217.5341, 1304.1561, 1338.5859, 1429.5640, 1470.8223, 1479.9155, 1804.5397, 3083.1395, 3158.9283, 3560.1146

Compound: sCl 2 + TFA PRC	Energy (kJ mol⁻¹): -1052.601700422370
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -3.799697 -0.091409 -0.000002 6 -2.347221 0.459861 -0.000005 8 -2.108307 1.633985 -0.000084 8 -1.484502 -0.540871 0.000090 1 -0.547214 -0.201783 0.000087 9 -4.020862 -0.851857 1.087808 9 -4.685788 0.902500 -0.000097 9 -4.020806 -0.852015 -1.087714 6 2.986750 -1.120937 -0.000139 1 3.407058 -2.117903 -0.000289 6 3.892144 0.100749 0.000001 9 5.160273 -0.346119 -0.000099 9 3.707438 0.844929 1.088662 9 3.707372 0.845228 -1.088443 8 1.736864 -1.066737 -0.000094 8 1.120900 0.141888 0.000090	7.4879, 14.3532, 20.6102, 31.1970, 55.4650, 76.5639, 94.0482, 96.9370, 212.7209, 246.4799, 256.8046, 272.6323, 344.7727, 397.8334, 434.7728, 480.6449, 513.6752, 515.4363, 534.8743, 588.9227, 589.4324, 700.5560, 761.7026, 777.5234, 804.9535, 826.1389, 882.6705, 910.8651, 921.5577, 1144.3144, 1159.2873, 1167.2045, 1194.3575, 1203.5754, 1241.3283, 1319.8137, 1369.8098, 1464.7495, 1574.1753, 1831.1291, 3185.8483, 3207.3049



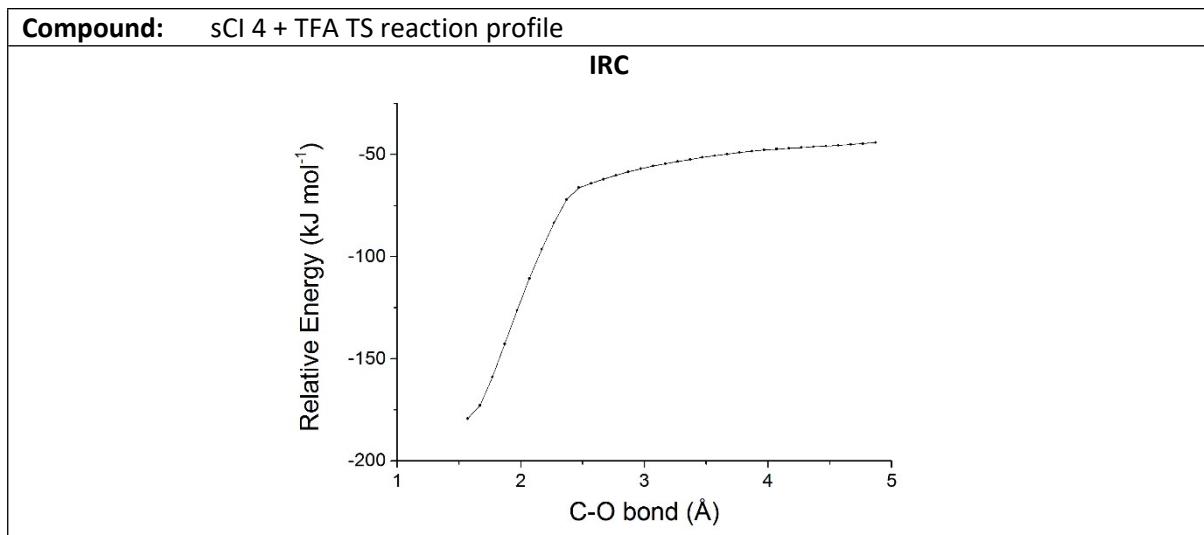
Compound: sCl 2 + TFA Pr	Energy (kJ mol⁻¹): -1052.676223405220
Reaction Coordinates:	Frequencies (cm⁻¹):
6 2.466755 -0.218822 -0.010253 6 1.077154 0.480250 -0.048466 8 0.223035 -0.229661 0.676838 8 0.898034 1.503630 -0.647334 1 -0.795315 2.179440 -0.631200 9 2.920504 -0.276317 1.250781 9 2.376110 -1.467759 -0.486261 9 3.341071 0.457006 -0.747324 6 -1.184517 0.121186 0.891958 1 -1.378609 -0.275710 1.886916 6 -2.012724 -0.706613 -0.124124 9 -1.906052 -2.007793 0.205177 9 -1.584011 -0.572463 -1.382367 9 -3.299166 -0.365633 -0.056782 8 -1.472670 1.453093 0.975477 8 -1.716172 2.040048 -0.324908	20.9216, 46.4107, 59.5233, 66.9627, 140.9286, 180.6824, 208.0201, 222.3190, 241.0757, 274.9078, 308.1438, 334.1048, 362.8241, 433.7150, 463.3602, 517.3656, 523.8811, 555.0554, 587.9584, 609.8561, 615.5457, 729.8126, 772.5619, 776.2809, 802.3635, 872.2091, 915.3374, 953.1669, 1107.5102, 1140.5180, 1162.3852, 1165.8355, 1202.4495, 1221.0011, 1267.8230, 1322.9942, 1382.6886, 1408.0581, 1477.9667, 1816.0211, 3111.6009, 3521.6190

Compound: sCl 3 + TFA PRC	Energy (kJ mol⁻¹): -1052.615934886960
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -3.826946 -0.368024 -0.000004 6 -2.584060 0.564096 -0.000003 8 -2.680549 1.758804 -0.000038 8 -1.477753 -0.157249 0.000039 1 -0.672574 0.431164 0.000037 9 -3.828923 -1.159749 1.087783 9 -4.953816 0.341783 -0.000036 9 -3.828888 -1.159792 -1.087759 6 2.969286 0.773171 0.000006 1 3.182798 1.836550 0.000002 6 4.078973 -0.256742 -0.000008 9 4.845283 -0.070606 1.085930 9 4.845246 -0.070616 -1.085974 9 3.615989 -1.499861 0.000006 8 1.792081 0.363588 0.000023 8 0.792750 1.288215 0.000034	8.0185, 13.9625, 17.7010, 31.4483, 52.0818, 71.8914, 86.6937, 120.1090, 190.2986, 210.7532, 256.8293, 275.0837, 386.6084, 394.2236, 396.0285, 421.2855, 451.0603, 515.4830, 554.8464, 559.4263, 589.1000, 698.3424, 705.4141, 777.8235, 805.1851, 889.5173, 920.3806, 930.2730, 947.7599, 1144.8606, 1151.5780, 1165.9538, 1177.1230, 1203.4499, 1274.2055, 1321.6529, 1361.6588, 1469.0207, 1581.1019, 1829.5164, 3173.9935, 3180.2493



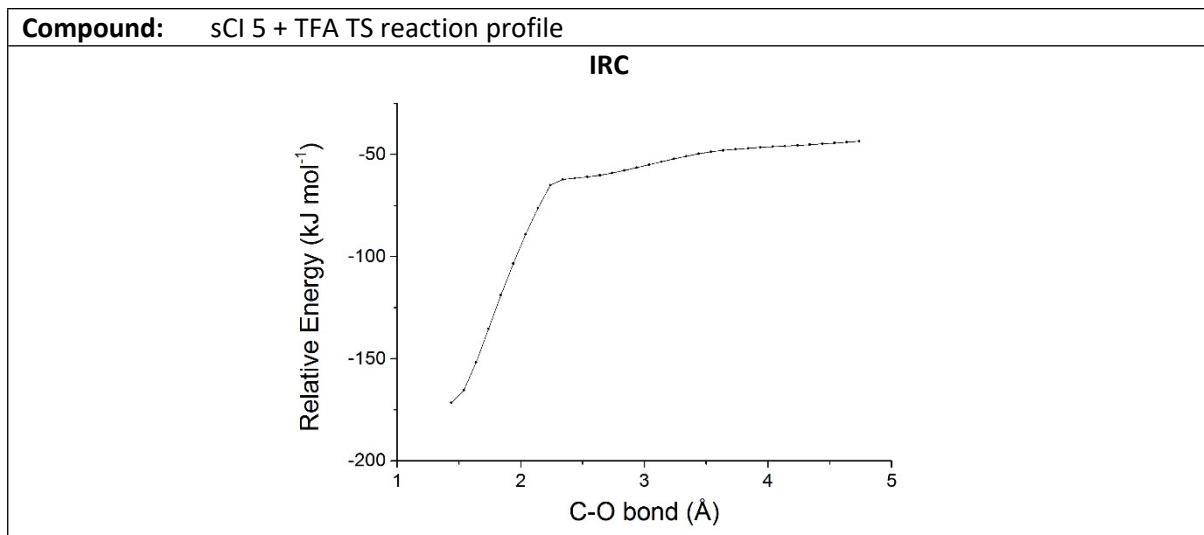
Compound: sCl 3 + TFA Pr	Energy (kJ mol⁻¹): -1052.680470535266
Reaction Coordinates:	Frequencies (cm⁻¹):
6 2.483742 -0.372757 0.043442 6 1.223571 0.514359 -0.157311 8 0.130443 -0.232202 -0.054291 8 1.287584 1.692255 -0.374094 1 -0.321270 2.775251 -0.088311 9 3.581760 0.373543 -0.007121 9 2.436718 -0.993124 1.228581 9 2.548991 -1.301029 -0.922371 6 -1.164747 0.389585 -0.277037 1 -1.228744 0.736793 -1.306612 6 -2.179111 -0.747309 -0.045264 9 -2.159505 -1.183022 1.216602 9 -1.911159 -1.779682 -0.857902 9 -3.407912 -0.300273 -0.326609 8 -1.428419 1.408876 0.610965 8 -1.293451 2.685692 -0.053664	17.8491, 45.2659, 57.3567, 85.9303, 91.0158, 151.4031, 198.4283, 211.4683, 224.0042, 267.5022, 321.0179, 347.0376, 363.7984, 417.9770, 437.2892, 506.6700, 519.4156, 526.9220, 556.0132, 568.8105, 623.4046, 705.7929, 743.5205, 783.8159, 844.5009, 924.0822, 935.2929, 966.8157, 1099.5354, 1155.7595, 1167.5156, 1170.7616, 1193.8652, 1221.3136, 1272.0266, 1319.1611, 1369.9295, 1392.6546, 1461.7710, 1815.8484, , 3107.3684, 3594.7285

Compound: sCl 4 + TFA PRC	Energy (kJ mol⁻¹): -1151.785667585767
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -4.026188 0.115087 -0.000003 6 -2.590419 -0.478626 0.000010 8 -2.387789 -1.660124 0.000042 8 -1.698604 0.494542 -0.000019 1 -0.769777 0.125201 -0.000013 9 -4.225296 0.881788 -1.087785 9 -4.941674 -0.852194 0.000027 9 -4.225289 0.881850 1.087737 6 2.783742 0.841986 0.000010 9 3.411589 1.974591 0.000026 6 3.639321 -0.443219 -0.000005 9 4.924581 -0.087026 0.000003 9 3.385389 -1.156257 -1.088876 9 3.385383 -1.156287 1.088843 8 1.548119 0.894821 0.000007 8 0.850635 -0.318584 -0.000011	7.5674, 12.4297, 19.3225, 31.2593, 50.4704, 72.6003, 74.4628, 94.2524, 216.1131, 223.6194, 256.7145, 267.3417, 268.5318, 320.3421, 365.0210, 398.2187, 434.9504, 492.4184, 502.1436, 515.5134, 583.1883, 588.9652, 634.5484, 681.7848, 703.3383, 778.2821, 779.5247, 806.1318, 916.0536, 931.0692, 1144.0082, 1159.5167, 1167.1897, 1203.5568, 1210.5858, 1215.0487, 1326.4277, 1409.5507, 1475.7611, 1612.9563, 1828.5912, 3134.7645



Compound: sCl 4 + TFA Pr	Energy (kJ mol⁻¹): -1052.676223405220
Reaction Coordinates:	Frequencies (cm⁻¹):
6 2.570852 -0.220706 -0.048599 6 1.190871 0.497624 -0.075921 8 0.283900 -0.301589 0.489465 8 1.044352 1.588401 -0.545206 1 -0.664295 2.301576 -0.536636 9 3.492197 0.545148 -0.621273 9 2.937107 -0.467545 1.215532 9 2.502004 -1.383092 -0.709593 6 -1.101065 0.050357 0.709143 9 -1.402883 -0.578986 1.851964 6 -1.936484 -0.600185 -0.449206 9 -1.876419 -1.929874 -0.319117 9 -1.443070 -0.281041 -1.649312 9 -3.207065 -0.219797 -0.371946 8 -1.331892 1.376630 0.958426 8 -1.583557 2.108129 -0.257954	20.2362, 43.7482, 55.1268, 71.0110, 130.8120, 182.2072, 200.8834, 216.0355, 230.5157, 259.6638, 289.3327, 312.5332, 347.1908, 356.0338, 399.2152, 434.5129, 490.6210, 515.8851, 523.0897, 560.4187, 596.1424, 614.6958, 622.2812, 679.8581, 747.6963, 769.5610, 778.4174, 874.1228, 952.1876, 996.2569, 1112.7478, 1154.2816, 1169.0466, 1189.6280, 1209.4820, 1220.9972, 1226.1166, 1321.2883, 1351.3150, 1478.4904, 1827.5146, 3538.5897

Compound: sCl 5 + TFA PRC	Energy (kJ mol⁻¹): -1151.790445383220
Reaction Coordinates:	Frequencies (cm⁻¹):
6 4.041057 -0.371401 -0.006528 6 2.784865 0.542457 -0.033261 8 2.860138 1.727432 -0.196890 8 1.695334 -0.181267 0.147064 1 0.879129 0.394538 0.130234 9 3.966641 -1.314052 -0.963965 9 5.150320 0.337787 -0.206780 9 4.150014 -0.995914 1.180402 6 -2.746033 0.517700 0.030689 9 -3.121296 1.746140 0.046299 6 -3.846116 -0.543715 -0.034636 9 -4.564366 -0.357332 -1.144769 9 -4.648539 -0.404246 1.023560 9 -3.319784 -1.757664 -0.040486 8 -1.552125 0.186312 0.064841 8 -0.603184 1.197866 0.120465	10.8869, 16.0276, 18.2924, 33.1220, 39.6677, 68.5474, 72.3076, 109.6280, 158.5230, 203.3723, 257.2177, 263.1301, 323.1912, 339.7670, 376.5955, 402.9848, 407.9736, 441.1388, 515.5822, 519.6200, 576.9508, 589.1252, 690.4387, 700.7299, 732.6585, 778.2141, 805.1062, 866.9637, 878.4136, 931.1100, 1143.9517, 1159.9433, 1166.6482, 1175.0592, 1203.4005, 1242.0594, 1324.6173, 1404.8743, 1470.9930, 1632.4361, 1829.4762, 3145.9245



Compound: sCl 5 + TFA Pr	Energy (kJ mol⁻¹): -1151.857358431300
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -2.478273 -0.365041 0.052667 6 -1.231413 0.534401 -0.189079 8 -0.195732 0.002265 0.469856 8 -1.274726 1.527937 -0.851783 1 0.139962 2.699595 -0.432420 9 -3.536735 0.153442 -0.559265 9 -2.258822 -1.595347 -0.428923 9 -2.738447 -0.458172 1.362722 6 1.185449 0.380691 0.339739 9 1.599361 0.605801 1.597192 6 1.928243 -0.878655 -0.221274 9 1.470194 -1.176197 -1.441770 9 1.716059 -1.919260 0.583947 9 3.236846 -0.646728 -0.291123 8 1.475822 1.397572 -0.509050 8 0.999621 2.647248 0.032861	16.5182, 35.7526, 42.6654, 58.4852, 128.4627, 176.2977, 193.2411, 204.6366, 231.9707, 261.3122, 277.9253, 325.7491, 342.7255, 366.9216, 381.0447, 427.6308, 485.9296, 514.6093, 529.4620, 555.3729, 562.0244, 587.2120, 642.9921, 722.2777, 750.2701, 769.7615, 816.3052, 860.7169, 949.7770, 1006.6998, 1096.7814, 1147.6885, 1168.4412, 1201.0525, 1210.1647, 1225.0795, 1225.6274, 1317.1342, 1334.0822, 1463.8412, 1831.1421, 3559.6296

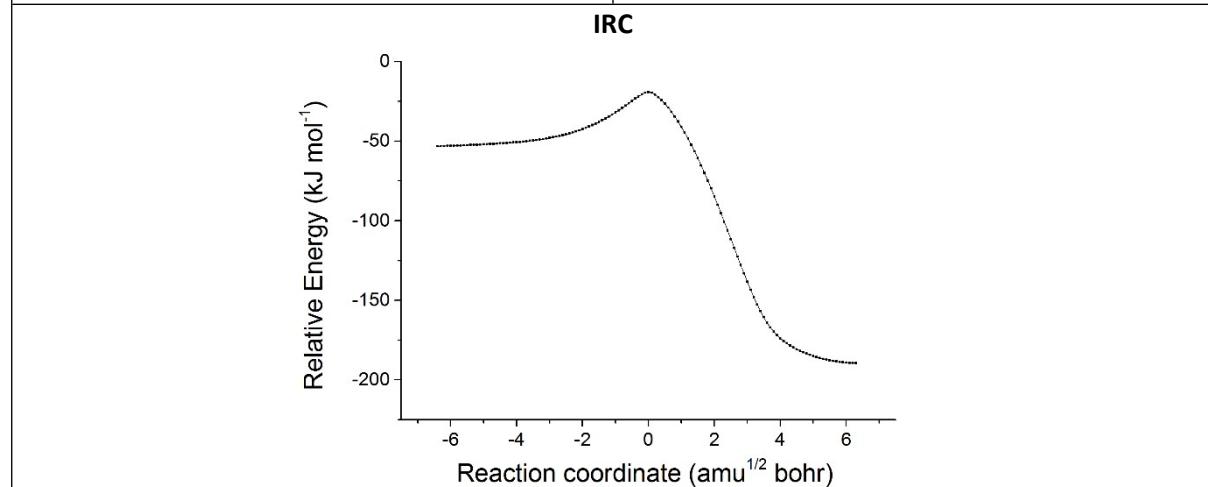
S8.6 Reactions with HF

sCl 1 + HF

Compound: HF	Energy (kJ mol⁻¹): -100.388736810254
Reaction Coordinates:	Frequencies (cm⁻¹):
9 0.000000 0.000000 0.092410 1 0.000000 0.000000 -0.831689	4072.1142

Compound: sCl 1 + HF PRC	Energy (kJ mol⁻¹): -289.808532547218
Reaction Coordinates:	Frequencies (cm⁻¹):
6 0.968789 -0.977363 -0.000007 1 1.860439 -1.590782 0.000013 1 -0.050820 -1.352498 -0.000039 8 1.192344 0.252290 0.000012 8 0.149073 1.146514 -0.000009 1 -1.194974 0.358284 -0.000001 9 -1.906524 -0.304584 0.000005	79.1803, 152.0807, 307.7401, 557.6608, 691.8667, 864.2225, 887.5397, 1046.3924, 1095.0887, 1278.6763, 1433.0717, 1562.9479, 3045.8930, 3102.0931, 3246.6880

Compound: sCl 1 + HF TS	Energy (kJ mol⁻¹): -289.791769474177
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -0.304787 1.051637 0.223711 1 -0.360805 2.020957 -0.260655 1 0.124674 0.914501 1.204044 8 -0.967644 0.141252 -0.333197 8 -0.540291 -1.132555 0.180239 1 0.624123 -0.809496 0.047753 9 1.500468 -0.056152 -0.123305	-829.2644, 336.8971, 483.6272, 530.5988, 696.5045, 765.8268, 835.9935, 1117.4420, 1197.3750, 1234.8638, 1420.0552, 1580.7735, 1822.9523, 3125.6008, 3268.4885

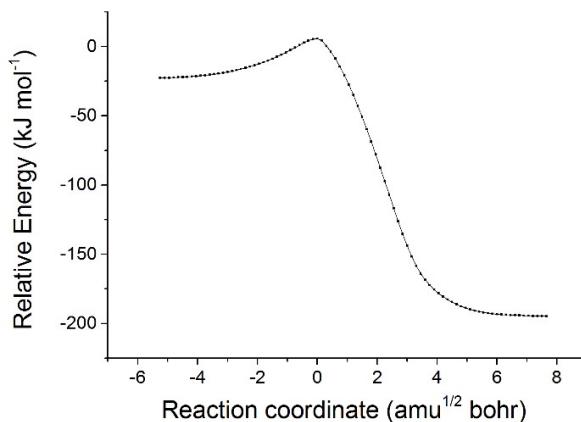


Compound: sCl 1 + HF Pr	Energy (kJ mol⁻¹): -289.865197002865
Reaction Coordinates:	Frequencies (cm⁻¹):
6 0.631649 0.519902 0.269119 1 1.175202 1.409728 -0.049661 1 0.499281 0.459144 1.348441 8 -0.574863 0.581176 -0.391641 8 -1.469572 -0.389753 0.211664 1 -1.362095 -1.142101 -0.388781 9 1.361467 -0.597507 -0.120544	168.4645, 277.9778, 421.6778, 595.9358, 882.7828, 995.6691, 1077.0262, 1160.2288, 1287.4215, 1383.4539, 1415.0926, 1480.2425, 3056.8428, 3125.4267, 3737.5366

Compound: sCl 2 + HF PRC	Energy (kJ mol⁻¹): -626.606991675634
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -1.371755 0.052827 0.000001 6 -0.311811 -1.035991 -0.000000 1 -0.594705 -2.080061 0.000002 8 0.919805 -0.815171 -0.000003 8 1.370679 0.461411 -0.000005 9 -2.569662 -0.558544 0.000006 9 -1.289552 0.816787 -1.088349 9 -1.289546 0.816792 1.088347 1 3.031578 0.273012 -0.000001 9 3.964389 0.095644 0.000003	26.9508, 54.9483, 84.1009, 132.1838, 246.9818, 247.4661, 354.7131, 480.7731, 510.2538, 534.0738, 590.3146, 749.8774, 756.3514, 821.6079, 866.9841, 896.0442, 918.1111, 1161.0342, 1191.3784, 1242.0397, 1370.7088, 1572.2553, 3208.6780, 3509.336

Compound: sCl 2 + HF TS	Energy (kJ mol⁻¹): -626.586057924863
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -0.889086 -0.029145 0.056776 6 0.302863 -0.341538 -0.860655 1 0.039340 -0.360371 -1.915298 8 1.376077 -0.931380 -0.588680 8 1.887292 -0.545928 0.692138 9 -1.656484 -1.145380 -0.046981 9 -1.575119 0.987143 -0.465474 9 -0.652471 0.193255 1.329209 1 1.652604 0.682188 0.400681 9 1.186123 1.489509 -0.204506	-744.6703, 64.3780, 155.9795, 218.8689, 290.9263, 305.9879, 388.0168, 478.4261, 514.3063, 536.6919, 562.2071, 731.4878, 809.2587, 838.2783, 880.0394, 1017.9591, 1097.0831, 1182.4137, 1196.0533, 1309.7201, 1372.5351, 1594.9569, 1789.9811, 3154.0510

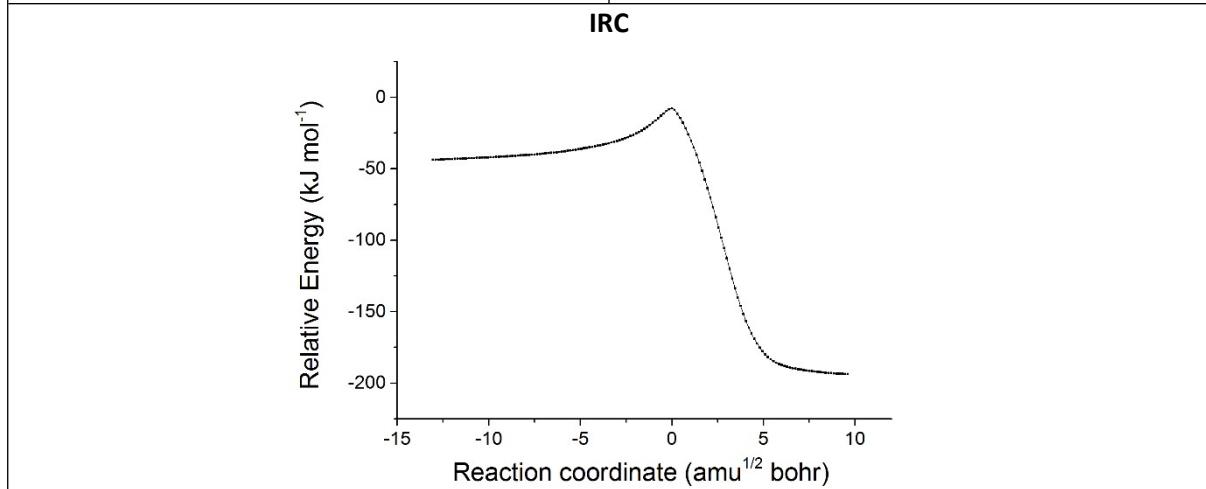
IRC



Compound: sCl 2 + HF Pr	Energy (kJ mol⁻¹): -626.673546284392
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -0.834570 -0.148326 0.006261 6 0.403604 0.587437 -0.563339 1 0.154218 0.966869 -1.556157 8 1.519250 -0.186749 -0.771280 8 1.871523 -0.876548 0.454931 9 -1.921667 0.615688 -0.223556 9 -0.766233 -0.375379 1.316841 9 -1.007864 -1.313996 -0.628918 1 2.545035 -0.281518 0.817457 9 0.669137 1.649949 0.270295	56.8425, 135.9833, 203.5157, 232.7615, 265.9833, 305.1939, 368.6668, 433.8158, 523.9364, 580.0369, 619.4572, 762.1754, 860.2378, 911.8567, 1057.9257, 1104.4746, 1150.6938, 1188.4176, 1283.4743, 1344.2761, 1394.2047, 1408.6765, 3084.3933, 3733.0817

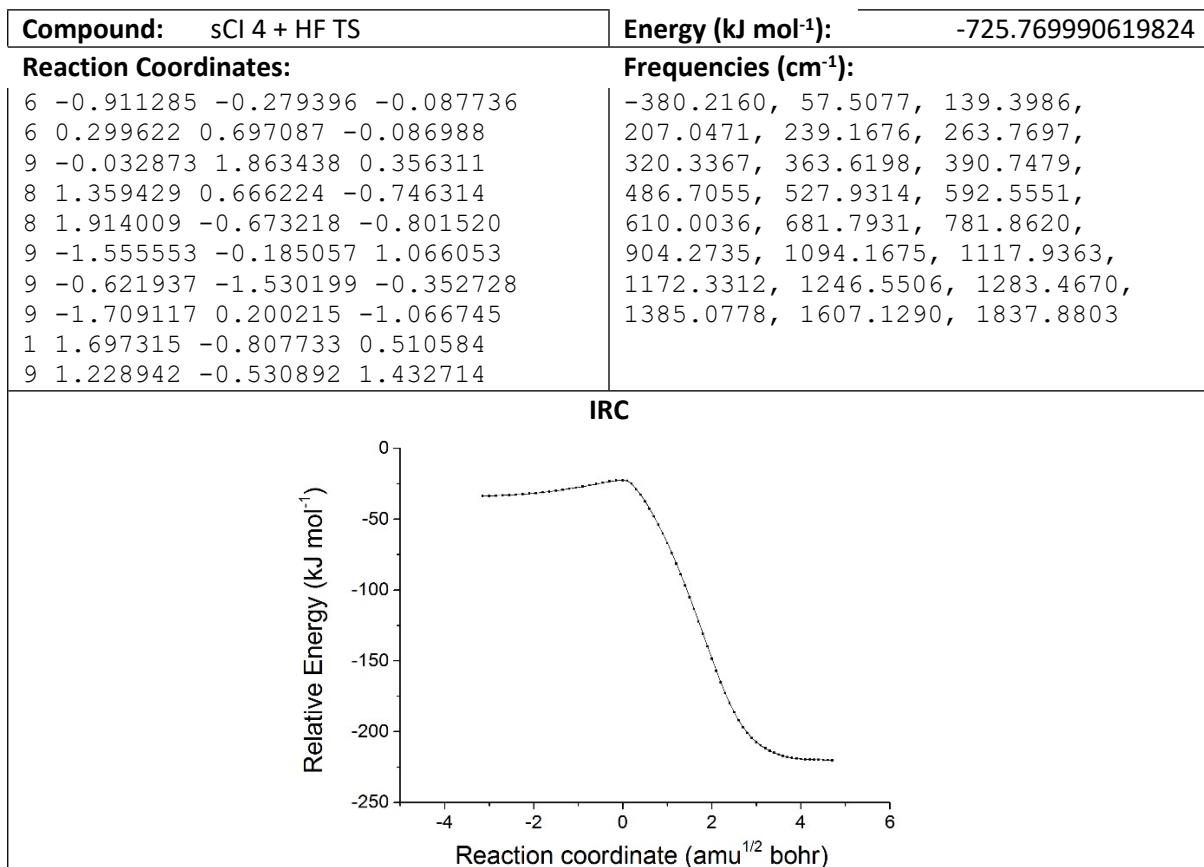
Compound: sCl 3 + HF PRC	Energy (kJ mol⁻¹): -626.610016871772
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -1.276051 0.122225 0.000000 6 0.238154 0.075338 0.000006 1 0.838781 0.982707 0.000012 8 0.768267 -1.055007 -0.000001 8 2.128693 -1.184191 0.000001 9 -1.691044 0.792038 -1.086257 9 -1.691051 0.792061 1.086241 9 -1.825166 -1.087203 0.000011 1 2.778804 0.267521 -0.000000 9 2.922163 1.222879 -0.000001	25.8279, 77.5369, 107.2193, 170.3821, 207.5782, 288.9583, 383.9496, 395.5151, 441.0878, 557.4098, 560.1105, 699.9805, 843.3993, 888.2642, 940.8348, 962.1647, 1046.2065, 1153.2153, 1187.3539, 1277.1579, 1388.8897, 1567.4454, 3116.9891, 3196.4029

Compound: sCl 3 + HF TS	Energy (kJ mol⁻¹): -626.592526379491
Reaction Coordinates:	Frequencies (cm⁻¹):
6 1.096361 -0.003094 -0.004462 6 -0.359680 -0.155275 -0.438103 1 -0.663416 0.236291 -1.398539 8 -1.066073 -0.963183 0.210641 8 -2.446944 -0.742017 -0.123188 9 1.292227 -0.377960 1.252167 9 1.472246 1.263687 -0.159861 9 1.844629 -0.776019 -0.813585 1 -2.284641 0.482190 0.001683 9 -1.649980 1.431772 0.093791	-850.5241, 34.8760, 114.4545, 182.7985, 287.3622, 336.1011, 376.2820, 453.9894, 513.0961, 549.5780, 621.8010, 675.8334, 751.4147, 869.5333, 909.5337, 990.2752, 1150.4901, 1190.6701, 1203.7874, 1284.1834, 1363.3160, 1575.9778, 1814.8849, 3212.0268



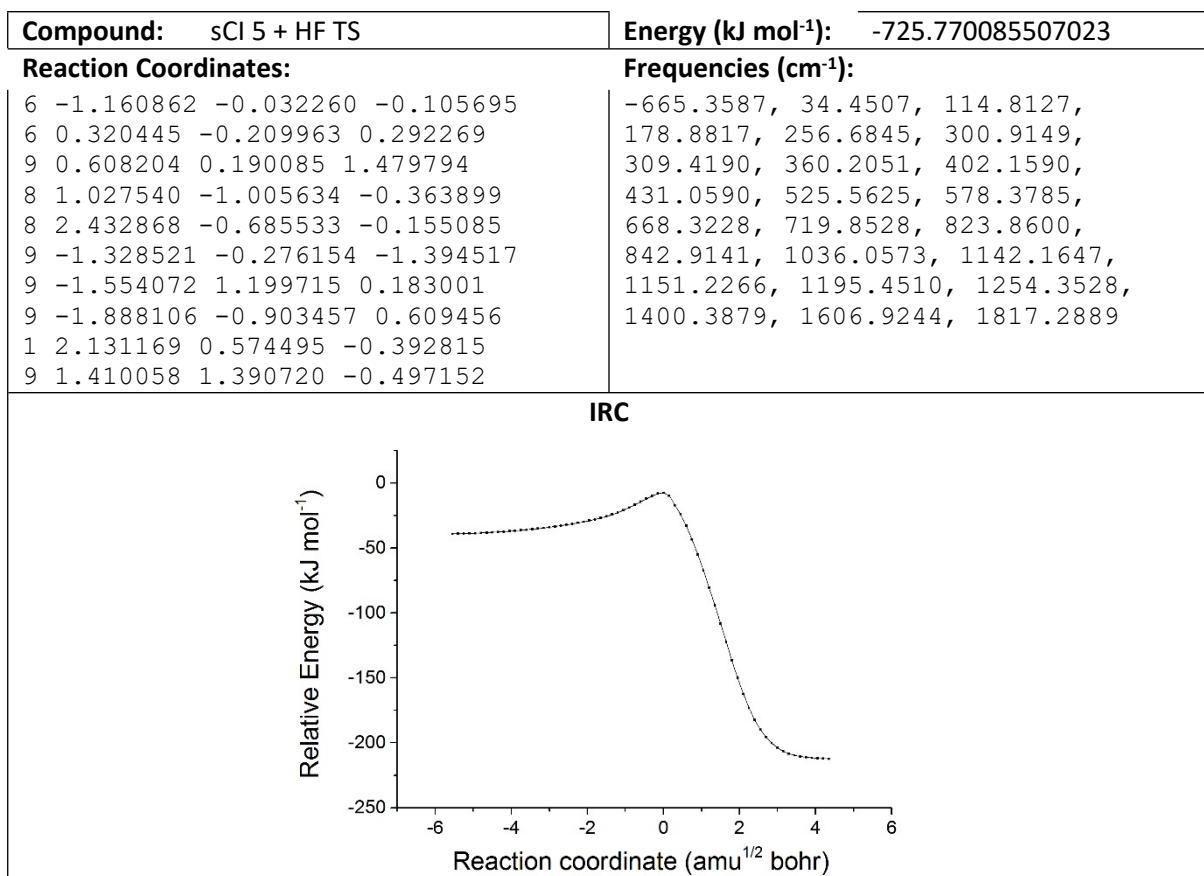
Compound: sCl 3 + HF Pr	Energy (kJ mol⁻¹): -626.675135107360
Reaction Coordinates:	Frequencies (cm⁻¹):
6 0.969274 -0.137135 -0.021241 6 -0.481096 0.279164 -0.340046 1 -0.659604 0.233344 -1.414282 8 -1.307314 -0.572338 0.355007 8 -2.626661 -0.476319 -0.234610 9 1.816862 0.714657 -0.615447 9 1.209603 -1.366423 -0.497401 9 1.215729 -0.131085 1.291727 1 -3.077769 0.085638 0.413486 9 -0.655516 1.584862 0.066159	65.4287, 108.2183, 184.1493, 231.2871, 271.0036, 350.3917, 357.5479, 394.4965, 522.2061, 565.1454, 617.2866, 716.1753, 879.1933, 954.1929, 1050.5288, 1128.3522, 1171.5778, 1185.5722, 1272.5277, 1359.8608, 1390.0408, 1400.2829, 3095.2578, 3732.6363

Compound: sCl 4 + HF PRC	Energy (kJ mol⁻¹): -725.776277323618
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -0.869815 -0.421998 -0.153846 6 -0.057258 0.866122 0.122023 9 -0.705444 1.716305 0.859512 8 1.041347 1.252070 -0.287829 8 1.874968 0.393522 -1.032891 9 -1.160303 -0.995607 1.009684 9 -0.271405 -1.272555 -0.951345 9 -2.015111 -0.010512 -0.727888 1 2.114898 -0.665378 0.158952 9 1.943042 -1.122532 0.987564	39.0244, 59.7387, 76.7172, 200.0033, 227.3232, 252.6997, 259.1565, 334.3436, 367.7980, 483.1061, 505.5070, 571.4816, 615.5282, 690.0793, 772.8026, 790.5163, 904.4336, 1008.4787, 1127.7364, 1176.1445, 1275.3385, 1391.3018, 1613.3843, 3263.4439



Compound: sCl 4 + HF Pr	Energy (kJ mol⁻¹): -725.857806332430
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -0.889167 -0.292627 -0.009267 6 0.415104 0.571256 0.005386 9 0.164758 1.722671 -0.629602 8 1.490616 0.038298 -0.679680 8 1.861000 -1.213187 -0.064101 9 -1.914062 0.459599 0.411129 9 -0.792497 -1.353455 0.790426 9 -1.142632 -0.705041 -1.253049 1 2.609374 -0.935161 0.485985 9 0.731328 0.838726 1.290825	39.7363, 126.2403, 202.0897, 225.5792, 249.4652, 288.7103, 343.6012, 376.3539, 399.9664, 489.8662, 531.6127, 590.5430, 653.5497, 687.2684, 782.6821, 980.2959, 1090.2120, 1155.7713, 1192.3813, 1201.6568, 1214.8916, 1350.7749, 1413.8516, 3727.6606

Compound: sCl 5 + HF PRC	Energy (kJ mol⁻¹): -725.783185514069
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -1.346110 0.061678 -0.133186 6 0.098792 -0.185894 0.312407 9 0.521788 0.510871 1.301754 8 0.788018 -1.043836 -0.255240 8 2.103466 -1.228631 0.181675 9 -1.485883 1.341631 -0.475855 9 -2.163912 -0.201940 0.891456 9 -1.655856 -0.715041 -1.159465 1 2.712299 0.234128 -0.283211 9 2.743833 1.141245 -0.580512	30.4167, 43.0085, 72.9994, 156.8884, 180.1979, 258.9762, 302.7274, 353.6017, 371.6271, 405.6737, 519.5743, 576.0599, 697.7263, 726.7195, 729.3323, 852.1026, 862.2567, 943.3627, 1160.6948, 1179.4724, 1241.3002, 1404.9745, 1636.1444, 3393.8484



Compound: sCl 5 + HF Pr	Energy (kJ mol⁻¹): -725.860158374757
Reaction Coordinates:	Frequencies (cm⁻¹):
6 1.071132 -0.164947 -0.001594 6 -0.426713 0.274719 -0.004529 9 -0.640177 1.108402 -1.036237 8 -1.174458 -0.867810 -0.087232 8 -2.571414 -0.496277 -0.117625 9 1.845035 0.918898 0.084820 9 1.369026 -0.819623 -1.124342 9 1.322689 -0.957224 1.043687 1 -2.829982 -0.688425 0.797048 9 -0.682080 0.965379 1.129688	60.1460, 121.4786, 168.5796, 220.2847, 255.1560, 295.0352, 336.5640, 374.0731, 376.8953, 523.2280, 529.6868, 594.5260, 606.2032, 738.2555, 833.3664, 943.2474, 1094.1752, 1144.8046, 1184.4607, 1206.5191, 1211.8461, 1345.0714, 1411.1882, 3726.8452

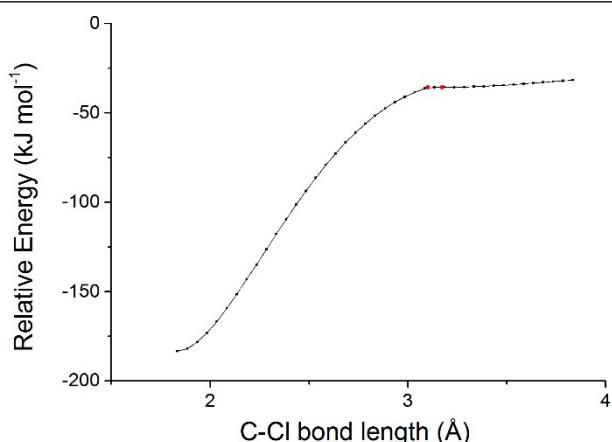
S8.7 Reactions with HCl

Compound: HCl	Energy (kJ mol⁻¹): -460.373228449128
Reaction Coordinates: 17 0.000000 0.000000 0.071311 1 0.000000 0.000000 -1.212290	Frequencies (cm⁻¹): 2935.7855

Compound: sCl 1 + HCL PRC	Energy (kJ mol⁻¹): -649.786096409260
Reaction Coordinates: 6 -1.336144 1.008146 0.177187 1 -1.868220 1.826217 -0.292080 1 -0.680692 1.112214 1.032231 8 -1.535125 -0.111981 -0.335374 8 -0.828331 -1.180579 0.194099 1 0.554522 -0.654641 0.096022 17 1.701112 0.118107 -0.045241	Frequencies (cm⁻¹): 60.4557, 176.4811, 230.4019, 509.8882, 663.3721, 754.8100, 850.5504, 1022.5705, 1035.4059, 1244.4996, 1423.1906, 1578.1814, 11.5544:13, 1619.2991, 1528.6350, 3120.4571, 3265.3577

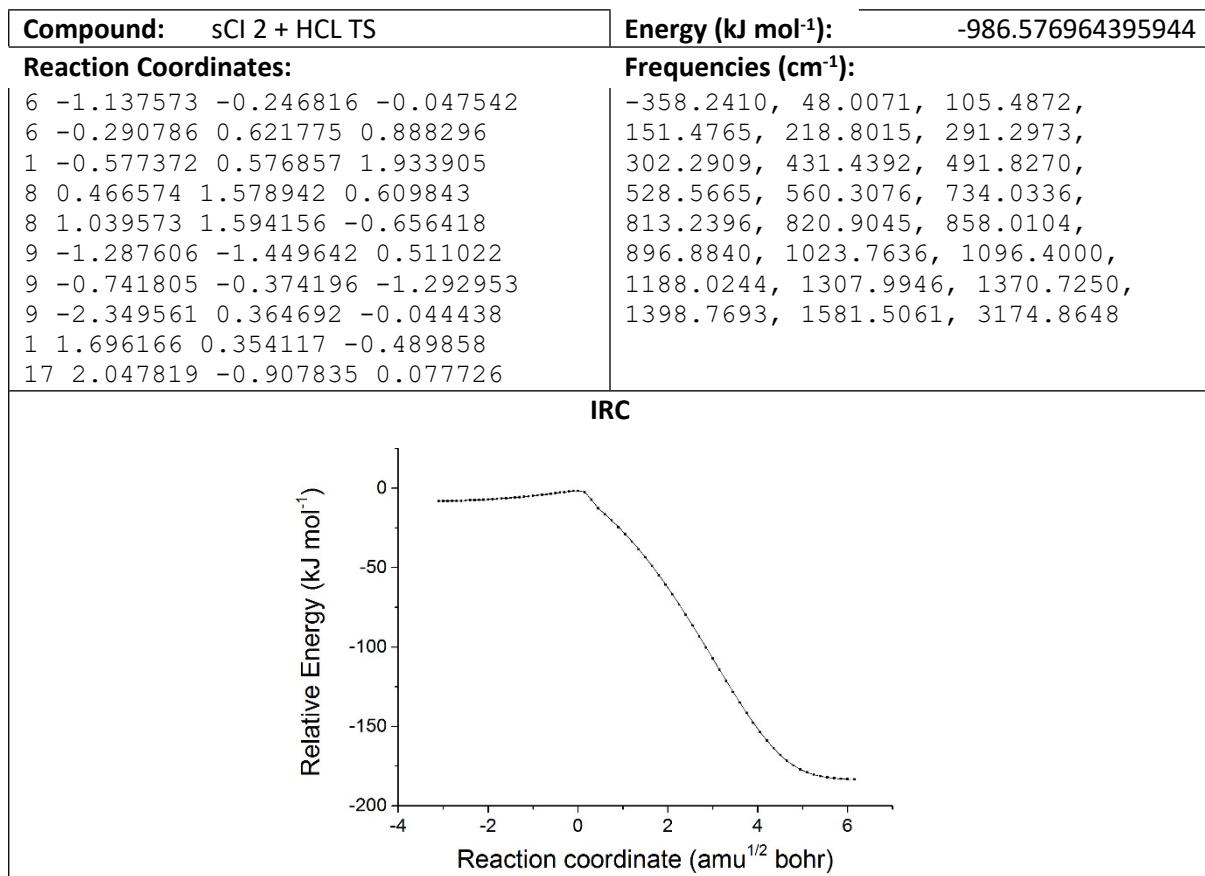
Compound: sCl 1 + HCL TS	Energy (kJ mol⁻¹): -649.785673633254
Reaction Coordinates: 6 -1.290274 1.012234 0.181602 1 -1.794768 1.846515 -0.290202 1 -0.647019 1.095745 1.047673 8 -1.515281 -0.099860 -0.338189 8 -0.822644 -1.182523 0.193323 1 0.504896 -0.678863 0.099185 17 1.669526 0.113074 -0.046314	Frequencies (cm⁻¹): -93.7733, 179.6707, 237.6760, 498.5803, 659.3270, 803.9659, 842.1572, 1033.4816, 1060.7591, 1247.6658, 1385.9919, 1428.7903, 1579.5775, 3122.0712, 3266.8750

IRC



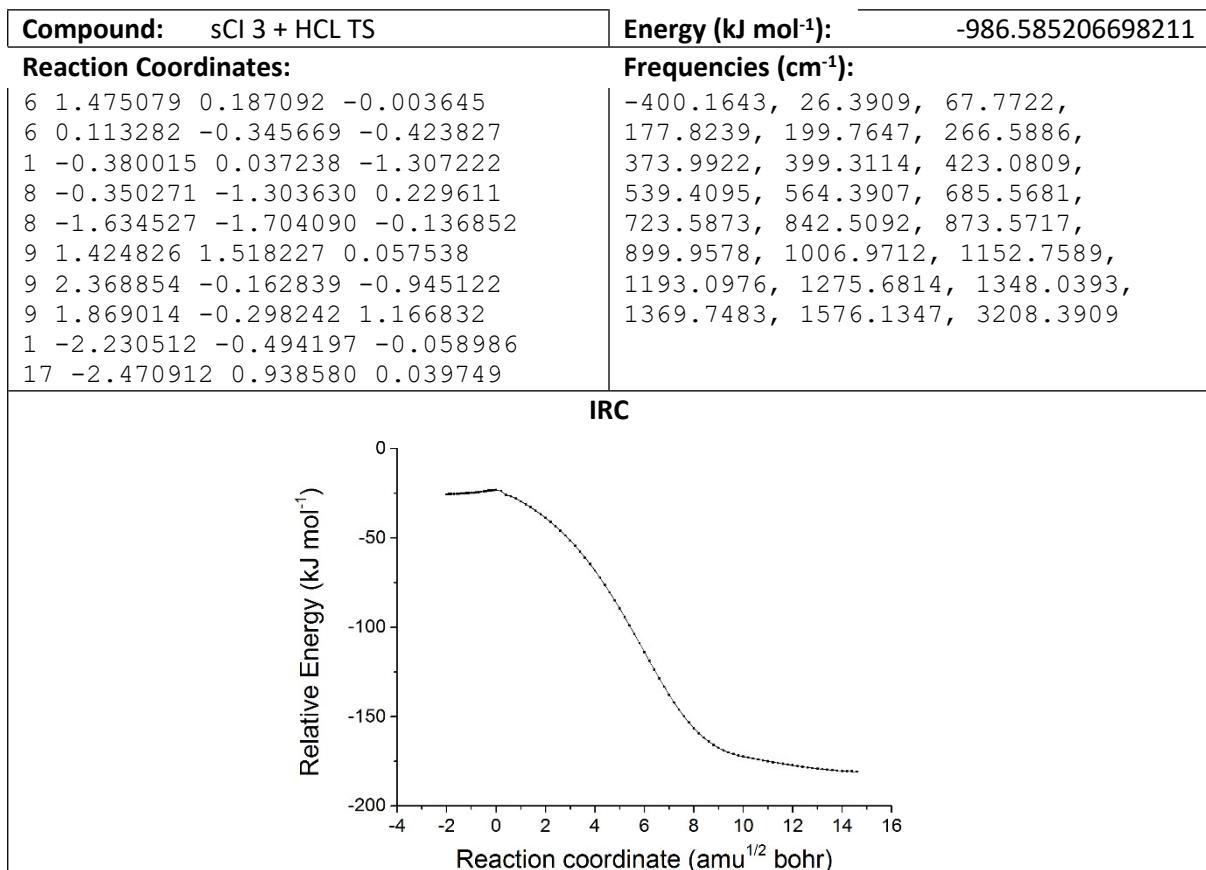
Compound: sCl 1 + HCL Pr	Energy (kJ mol⁻¹): -649.851301605594
Reaction Coordinates: 6 0.003850 0.838496 0.263509 1 0.297621 1.812033 -0.116982 1 -0.155121 0.828835 1.337016 8 -1.125194 0.483448 -0.438558 8 -1.737316 -0.651057 0.217450 1 -1.350257 -1.388632 -0.279390 17 1.416749 -0.290726 -0.044285	Frequencies (cm⁻¹): 143.1167, 260.0739, 365.5883, 494.0492, 664.4717, 877.6148, 996.3526, 1078.6954, 1267.3400, 1322.2118, 1390.8227, 1456.2616, 3091.0927, 3169.7044, 3717.0894

Compound: sCl 2 + HCl PRC	Energy (kJ mol⁻¹): -986.587116795699
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -1.914503 -0.050458 0.000001 6 -0.847721 1.030051 -0.000000 1 -1.123123 2.075925 0.000001 8 0.383224 0.799262 -0.000004 8 0.828024 -0.476783 -0.000006 9 -3.108809 0.568984 0.000006 9 -1.839190 -0.816240 1.088202 9 -1.839196 -0.816235 -1.088204 1 2.617586 -0.277000 -0.000004 17 3.910215 -0.040288 0.000002	18.7102, 37.3389, 83.0612, 92.6607, 213.5202, 246.5783, 337.0294, 479.6058, 504.7996, 533.0463, 534.5479, 590.3170, 642.5689, 762.8747, 814.3955, 883.6309, 920.2579, 1159.4939, 1188.9489, 1241.8058, 1369.4349, 1565.8153, 2527.4977, 3209.9424



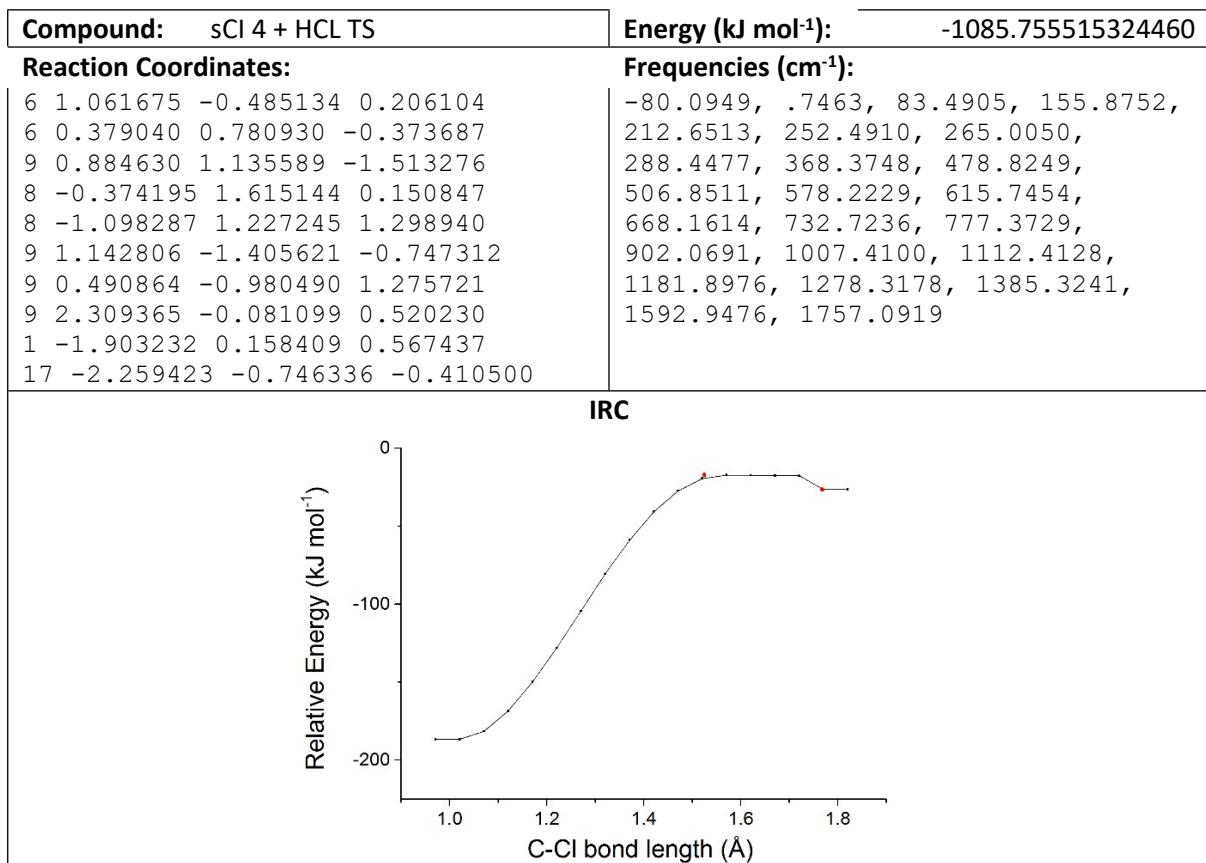
Compound: sCl 2 + HCl Pr	Energy (kJ mol⁻¹): -986.659667626854
Reaction Coordinates:	Frequencies (cm⁻¹):
6 1.016093 -0.086265 0.021645 6 -0.362052 -0.043383 -0.679606 1 -0.215083 -0.351869 -1.712506 8 -0.930566 1.207428 -0.779110 8 -1.082098 1.795492 0.533299 9 1.623303 -1.247388 -0.288003 9 0.963234 0.009638 1.348314 9 1.778337 0.913227 -0.447178 1 -1.999280 1.557641 0.741066 17 -1.464261 -1.266502 0.080441	59.1331, 127.2325, 176.6045, 208.4503, 234.4071, 305.8910, 328.7563, 382.3320, 505.5717, 524.4360, 577.9461, 715.5376, 806.8403, 845.2028, 905.7538, 1083.8250, 1135.3758, 1167.9676, 1262.5845, 1268.2043, 1380.2082, 1398.1375, 3116.5427, 3717.2765

Compound: sCl 3 + HCl PRC	Energy (kJ mol⁻¹): -986.588821076460
Reaction Coordinates:	Frequencies (cm⁻¹):
6 1.582120 -0.230508 0.004148 6 0.216761 0.291935 0.403795 1 -0.350700 -0.175656 1.199747 8 -0.219477 1.289563 -0.205145 8 -1.454922 1.771203 0.136133 9 1.479664 -1.523676 -0.327477 9 2.406751 -0.134788 1.061239 9 2.105067 0.442975 -1.013836 1 -2.403940 0.424704 0.044861 17 -2.856870 -0.833197 -0.036442	21.7466, 41.5599, 106.4836, 157.8438, 195.5869, 236.1469, 381.4594, 399.6774, 414.5187, 550.3914, 559.1233, 605.1213, 699.8698, 797.7048, 886.7464, 909.3514, 939.2583, 1151.0214, 1182.2896, 1273.7220, 1365.0410, 1572.7375, 2192.8173, 3188.7321



Compound: sCl 3 + HCl Pr	Energy (kJ mol⁻¹): -986.660655427993
Reaction Coordinates:	Frequencies (cm⁻¹):
6 1.117713 -0.163739 -0.020203 6 -0.382956 -0.078604 -0.356523 1 -0.522871 -0.220769 -1.424751 8 -1.013282 -1.059037 0.380371 8 -2.293145 -1.322743 -0.231825 9 1.796699 0.773587 -0.693662 9 1.589405 -1.363303 -0.395803 9 1.356619 -0.008933 1.284400 1 -2.885091 -0.816900 0.346833 17 -1.013745 1.584342 0.023264	62.6223, 105.0812, 183.4353, 187.3409, 251.3766, 322.8491, 339.7521, 364.9799, 461.5709, 534.4284, 562.4427, 700.6468, 762.1871, 878.4267, 954.3954, 1100.9976, 1141.3416, 1180.7016, 1259.3337, 1285.8863, 1358.3179, 1399.9080, 3119.8932, 3717.5641

Compound: sCl 4 + HCl PRC	Energy (kJ mol⁻¹): -1085.756886776330
Reaction Coordinates:	Frequencies (cm⁻¹):
6 1.788499 -0.376521 -0.000005 6 0.756492 0.768406 0.000006 9 1.213983 1.981969 -0.000002 8 -0.473795 0.640091 0.000021 8 -0.994569 -0.653408 0.000031 9 3.009876 0.160989 -0.000023 9 1.643236 -1.121713 -1.088590 9 1.643265 -1.121708 1.088587 1 -2.736778 -0.349696 0.000010 17 -4.022323 -0.058289 -0.000011	14.4742, 35.2968, 60.1189, 94.2784, 219.7336, 222.6743, 266.1698, 304.8624, 364.6273, 483.8170, 500.3309, 549.8405, 578.5807, 631.2857, 675.8252, 683.1820, 780.3017, 920.0740, 1160.6457, 1207.0769, 1210.1306, 1406.4075, 1607.6839, 2477.0829

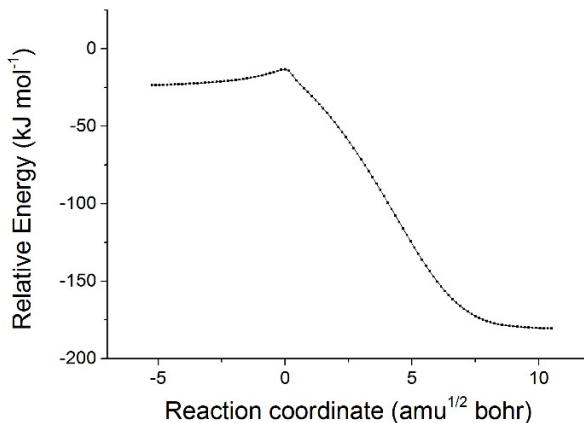


Compound: sCl 4 + HCl Pr	Energy (kJ mol⁻¹): -1085.834620970350
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -1.060856 -0.008297 -0.189123 6 0.355338 -0.108230 0.475200 9 0.197689 -0.699204 1.669813 8 0.927380 1.111898 0.791717 8 1.101114 1.895565 -0.401700 9 -1.647677 -1.209971 -0.157015 9 -1.008596 0.402535 -1.453358 9 -1.808957 0.848469 0.512907 1 2.027791 1.705099 -0.618622 17 1.434426 -1.126006 -0.551124	50.2033, 121.4143, 171.8280, 217.7511, 223.4230, 278.6540, 312.2242, 345.4722, 366.0803, 405.1581, 506.9394, 538.8667, 590.2412, 624.2460, 743.7466, 909.3164, 980.1865, 1102.6952, 1157.4600, 1191.4081, 1211.5291, 1295.0756, 1416.7299, 3710.0443

Compound: sCl 5 + HCl PRC	Energy (kJ mol⁻¹): -1085.762584407700
Reaction Coordinates: 6 1.652214 -0.266535 -0.132727 6 0.351559 0.434522 0.266554 9 -0.039641 0.277222 1.479483 8 -0.252232 1.146365 -0.549289 8 -1.429760 1.763605 -0.136867 9 1.529614 -1.576698 0.079388 9 2.646860 0.200300 0.630556 9 1.929256 -0.041822 -1.407761 1 -2.492164 0.382844 -0.129858 17 -2.980549 -0.847149 -0.130520	Frequencies (cm⁻¹): 26.6352, 29.6193, 55.4301, 131.9833, 181.2789, 204.0811, 294.9425, 344.6381, 364.5925, 405.7232, 517.2215, 526.6510, 576.2596, 686.3177, 706.9846, 728.9803, 860.5920, 873.3529, 1157.9005, 1176.9629, 1240.1319, 1399.6990, 1626.7741, 2402.4687

Compound: sCl 5 + HCl TS	Energy (kJ mol⁻¹): -1085.758646350020
Reaction Coordinates: 6 1.444993 -0.202352 -0.117201 6 0.113833 0.485166 0.242495 9 -0.193524 0.488128 1.486587 8 -0.439482 1.208925 -0.609714 8 -1.744542 1.629281 -0.252622 9 1.461858 -1.426052 0.392147 9 2.440338 0.511295 0.427362 9 1.603459 -0.248916 -1.429169 1 -2.214678 0.368181 -0.166095 17 -2.204428 -1.099459 -0.092902	Frequencies (cm⁻¹): -581.3091, 32.0565, 70.2436, 158.4864, 182.2178, 211.7313, 297.6019, 335.8299, 359.1080, 405.1791, 496.9126, 577.5197, 610.7858, 727.0141, 760.3198, 856.8177, 858.9326, 929.6950, 1145.8207, 1192.6900, 1252.5531, 1387.6973, 1411.2767, 1610.8001

IRC



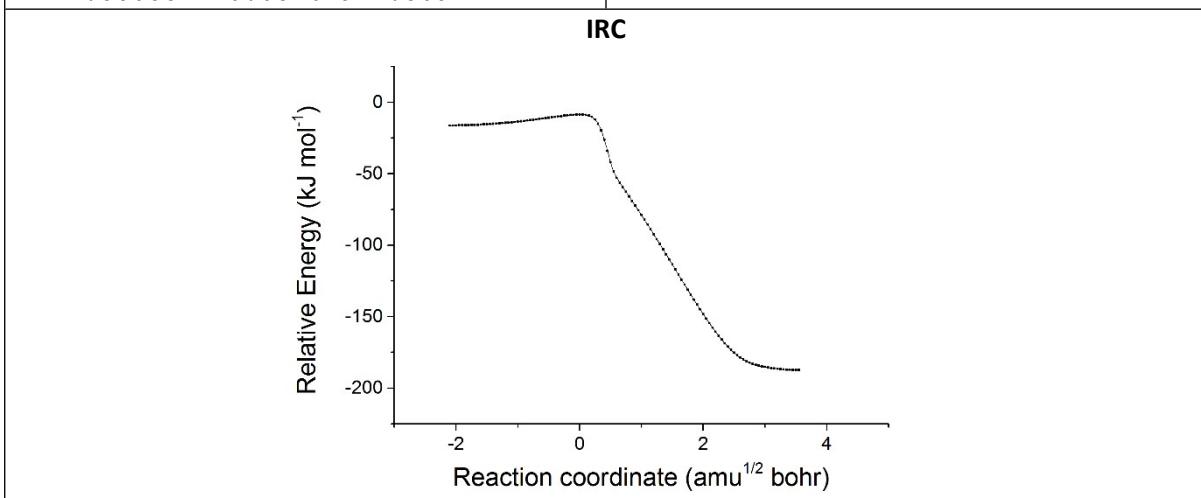
Compound: sCl 5 + HCl Pr	Energy (kJ mol⁻¹): -1085.837345614150
Reaction Coordinates: 6 -1.194899 -0.120815 -0.111015 6 0.331124 -0.083210 0.229355 9 0.474675 -0.305365 1.551533 8 0.899285 -1.074209 -0.524802 8 2.303812 -1.172668 -0.216102 9 -1.841384 0.800655 0.603411 9 -1.693257 -1.321710 0.195276 9 -1.401636 0.111274 -1.408265 1 2.690449 -0.706358 -0.974330 17 1.001285 1.549519 -0.134476	Frequencies (cm⁻¹): 64.4258, 128.1095, 167.6467, 186.4425, 223.1601, 270.7713, 308.4509, 349.4830, 369.6766, 425.8388, 452.6172, 546.5202, 575.5520, 706.3830, 742.8602, 932.2223, 962.3080, 1078.0033, 1146.1876, 1198.7855, 1211.6191, 1288.9764, 1413.0046, 3714.6669

S8.8 Reactions with H₂S

Compound: H ₂ S	Energy (kJ mol⁻¹): -398.964896610182
Reaction Coordinates:	Frequencies (cm⁻¹):
16 0.000000 -0.000000 0.103242 1 -0.000000 0.971861 -0.825934 1 -0.000000 -0.971861 -0.825934	1206.0101, 2682.8315, 2695.9979

Compound: sCl 1 + H ₂ S PRC1	Energy (kJ mol⁻¹): -588.373411489393
Reaction Coordinates:	Frequencies (cm⁻¹):
6 1.383147 1.052216 -0.190803 1 0.857402 1.062232 -1.135927 1 1.750328 1.932456 0.319036 8 1.620589 -0.035140 0.381047 8 1.154027 -1.183091 -0.196698 16 -1.894674 0.073705 -0.044649 1 -0.816788 -0.757434 -0.095069 1 -1.971973 0.016017 1.296363	63.4000, 131.8291, 149.4863, 162.3626, 233.3899, 476.4641, 530.3435, 672.4471, 883.9619, 975.7551, 1211.1926, 1238.8171, 1410.6771, 1558.3845, 2475.3799, 2689.0854, 3126.9057, 3271.963

Compound: sCl 1 + H ₂ S TS1	Energy (kJ mol⁻¹): -588.366724078241
Reaction Coordinates:	Frequencies (cm⁻¹):
6 0.850344 1.071250 -0.227889 1 0.603134 0.952593 -1.271614 1 0.880621 2.037578 0.259668 8 1.403926 0.127025 0.404241 8 1.163723 -1.125721 -0.183869 16 -1.571014 -0.031723 -0.035438 1 -0.410264 -0.882903 -0.126424 1 -1.580531 -0.037629 1.309724	-287.4348, 192.8922, 266.6416, 346.5469, 470.9489, 503.0721, 727.6948, 853.5293, 903.8597, 1013.4027, 1212.6890, 1229.2910, 1385.7411, 1535.5675, 1640.5057, 2683.0310, 3132.3039, 3269.0933

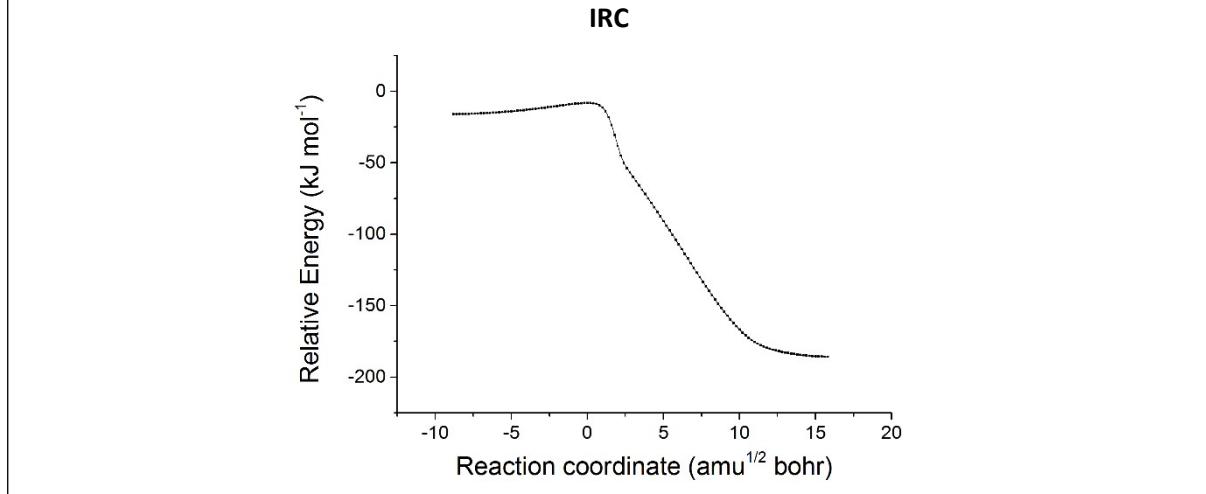


Compound: sCl 1 + H ₂ S Pr1	Energy (kJ mol⁻¹): -588.446035792738
Reaction Coordinates:	Frequencies (cm⁻¹):
6 0.001168 0.835726 0.290620 1 -0.181880 0.839497 1.364179 1 0.259165 1.833670 -0.056535 8 -1.165429 0.503394 -0.417719 8 -1.733621 -0.684680 0.175653 16 1.415014 -0.309297 0.024241 1 -1.234849 -1.384014 -0.276466	141.7941, 250.1772, 306.7244, 361.7041, 504.1382, 667.7742, 797.3210, 884.0190, 1004.4493, 1070.2357, 1270.4039, 1327.6749, 1388.0540, 1442.1300, 2670.1919, 3063.0127, 3131.2502, 3694.8508

1 1.702732 0.095537 -1.226224	
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Compound: sCl 1 + H ₂ S PRC2	Energy (kJ mol⁻¹): -588.373317449885
Reaction Coordinates:	Frequencies (cm⁻¹):
6 1.374106 1.051493 0.211173 1 0.855371 1.043508 1.160152 1 1.727983 1.941423 -0.291192 8 1.615258 -0.026559 -0.376434 8 1.163584 -1.185240 0.192467 16 -1.875235 0.076608 -0.116169 1 -0.812930 -0.764087 0.024453 1 -2.242037 -0.061138 1.169987	63.0680, 130.1110, 145.4168, 162.6574, 245.2900, 464.0536, 527.5536, 671.0392, 883.0327, 974.9178, 1215.9064, 1239.3125, 1410.1846, 1558.0192, 2476.9496, 2688.8034, 3126.9908, 3272.3093

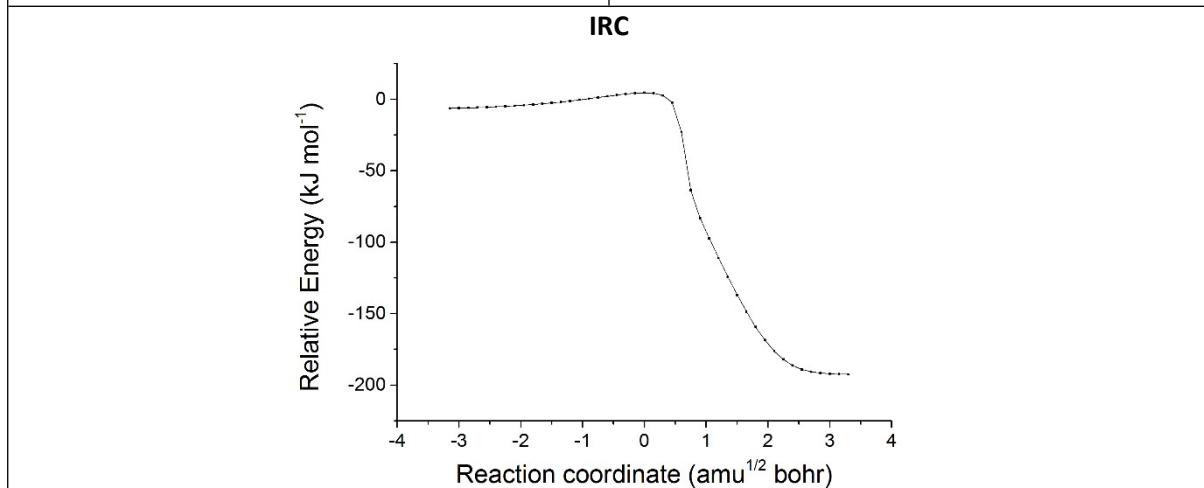
Compound: sCl 1 + H ₂ S TS2	Energy (kJ mol⁻¹): -588.366581083392
Reaction Coordinates:	Frequencies (cm⁻¹):
6 0.849619 1.069515 0.237062 1 0.578535 0.941856 1.273377 1 0.890091 2.038022 -0.245086 8 1.413656 0.127668 -0.389692 8 1.159069 -1.127293 0.188155 16 -1.550820 -0.025704 -0.128931 1 -0.404959 -0.883882 0.065341 1 -1.930063 -0.104826 1.159185	-295.7329, 192.3838, 268.3680, 346.2326, 491.6914, 513.1666, 715.6320, 853.2832, 888.9700, 1003.2864, 1205.4724, 1243.2746, 1385.4237, 1534.7258, 1629.1815, 2683.1069, 3135.1386, 3273.1921



Compound: sCl 1 + H ₂ S Pr2	Energy (kJ mol⁻¹): -588.445571445584
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -0.012289 0.836673 0.292041 1 0.137899 0.843930 1.369473 1 -0.261437 1.835409 -0.062595 8 1.182629 0.525484 -0.380415 8 1.707413 -0.699298 0.182137 16 -1.445103 -0.237773 -0.114264 1 1.423595 -1.344209 -0.482939 1 -1.225017 -1.160284 0.838262	138.0723, 205.9238, 297.5899, 356.7590, 494.4525, 686.5323, 780.5804, 886.2457, 1009.5045, 1062.2832, 1278.9894, 1319.8557, 1377.8454, 1439.6650, 2680.0174, 3066.0637, 3129.7951, 3725.2061

Compound: sCl 2 + H ₂ S PRC1	Energy (kJ mol⁻¹): -925.175112004368
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -1.484229 -0.344913 -0.006771 6 -0.855916 0.661362 0.935858 8 0.009345 1.502272 0.588972 8 0.449903 1.536780 -0.681884 1 -1.150956 0.698319 1.974948 9 -2.154044 0.255184 -0.994064 9 -0.584013 -1.177125 -0.531100 9 -2.361052 -1.072857 0.711674 16 3.190953 -0.600746 0.031241 1 4.010342 0.349054 0.514893 1 2.344237 0.306641 -0.499502	9.9024, 23.7127, 61.6911, 106.0087, 107.9417, 135.5994, 181.9940, 245.6847, 314.0242, 327.5125, 477.9915, 506.4081, 535.5159, 589.4703, 757.9055, 792.8041, 883.2286, 931.3426, 1155.0768, 1180.1698, 1209.2232, 1242.6978, 1366.4837, 1550.9099, , 2626.7917, 2690.7441, 3215.4330

Compound: sCl 2 + H ₂ S TS1	Energy (kJ mol⁻¹): -925.168836880090
Reaction Coordinates:	Frequencies (cm⁻¹):
6 1.079377 -0.272562 0.017459 6 0.062519 0.407790 -0.903023 8 -0.458449 1.540497 -0.680542 8 -0.719497 1.804417 0.660310 1 0.145936 0.148250 -1.951564 9 2.137961 0.548220 0.135183 9 0.674673 -0.606080 1.237017 9 1.493402 -1.393073 -0.605044 16 -1.985813 -0.912407 0.044771 1 -2.799656 -0.366896 -0.878246 1 -1.755408 0.304882 0.684309	-186.5290, 50.4707, 126.9609, 154.6013, 210.6029, 314.1125, 322.1871, 331.7323, 435.1666, 465.2846, 510.9986, 542.6879, 571.0339, 731.1105, 763.7064, 845.8696, 865.8312, 913.6169, 1145.9236, 1154.6731, 1216.9637, 1258.5030, 1344.6574, 1524.9039, 2125.4049, 2679.8543, 3182.1468

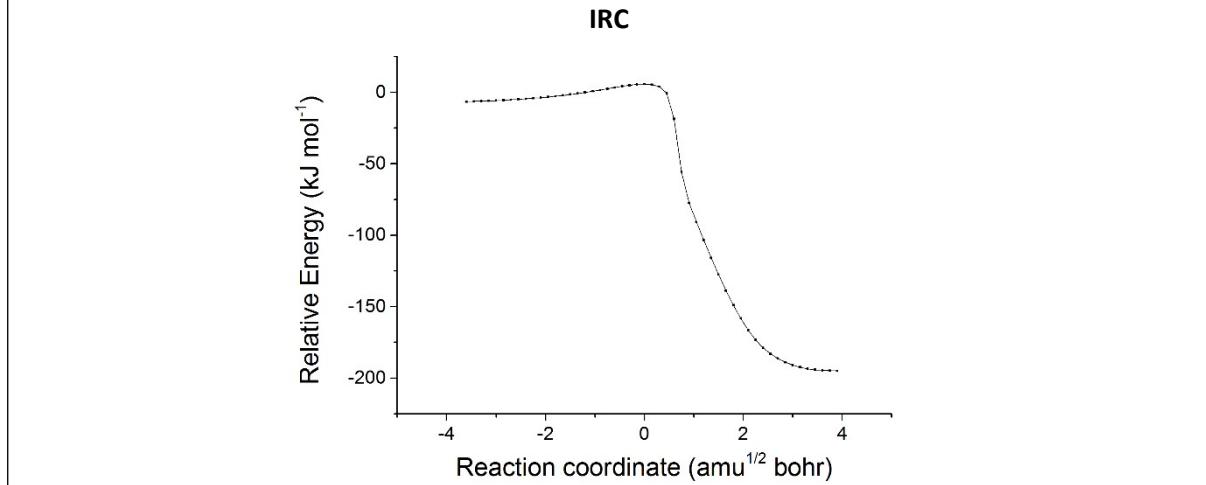


Compound: sCl 2 + H ₂ S Pr1	Energy (kJ mol⁻¹): -925.256025297439
Reaction Coordinates:	Frequencies (cm⁻¹):
6 1.023936 -0.130202 0.007316 6 -0.359856 -0.030549 -0.670201 8 -0.837859 1.281713 -0.779296 8 -1.004095 1.851805 0.537383 1 -0.222963 -0.334569 -1.706347 9 1.801492 0.896816 -0.354922 9 0.959825 -0.154651 1.346215 9 1.632480 -1.266220 -0.389625 16 -1.524866 -1.200782 0.148899	60.8407, 124.4968, 176.2385, 192.0415, 221.4286, 285.4577, 327.4123, 332.3845, 379.4655, 502.9899, 524.0442, 568.9290, 712.1710, 780.1388, 832.6499, 895.8157, 966.2179, 1039.8374, 1122.2301, 1175.2785, 1249.6693, 1260.7064, 1364.3021, 1394.9375, 2680.3929, 3106.7373, 3692.8576

1 -2.278442 -1.380984 -0.949920	
1 -1.893773 1.540930 0.771481	

Compound: sCl 2 + H ₂ S PRC2	Energy (kJ mol⁻¹): -925.174429805702
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -2.021718 -0.038630 0.002520 6 -0.932518 1.013408 0.041877 8 0.295322 0.750975 0.020220 8 0.714242 -0.524723 -0.042152 1 -1.182713 2.063680 0.091864 9 -1.978894 -0.759165 -1.119635 9 -1.960687 -0.858685 1.053557 9 -3.203548 0.606804 0.042095 16 4.235132 0.005906 -0.070083 1 4.435857 -0.406631 1.193404 1 2.921818 -0.310813 -0.039009	9.7858, 20.5426, 55.3474, 63.8132, 84.8668, 177.3682, 191.4593, 247.0686, 328.2667, 354.2698, 479.2267, 508.8930, 536.1159, 591.1425, 758.8193, 791.7027, 884.2323, 934.7058, 1155.4684, 1181.4413, 1216.3689, 1242.5905, 1366.8974, 1550.6684, 2608.8873, 2691.3503, 3215.2254

Compound: sCl 2 + H ₂ S TS2	Energy (kJ mol⁻¹): -925.167972073153
Reaction Coordinates:	Frequencies (cm⁻¹):
6 1.084640 -0.267962 0.020490 6 0.061333 0.409724 -0.893056 8 -0.466565 1.540319 -0.671974 8 -0.744439 1.796104 0.668712 1 0.150330 0.153688 -1.942151 9 2.171907 0.524589 0.062317 9 0.721390 -0.535782 1.268061 9 1.442690 -1.426479 -0.567284 16 -2.059113 -0.840429 -0.076063 1 -1.625042 -1.640531 0.913255 1 -1.791177 0.330804 0.639548	-200.4295, 46.0578, 134.0311, 158.5604, 213.1732, 290.9336, 312.7965, 319.5703, 408.7537, 472.4336, 517.6245, 550.6113, 573.3513, 731.3071, 761.6964, 844.7599, 865.9906, 916.4786, 1140.3623, 1152.3552, 1224.1910, 1264.0710, 1345.3041, 1524.3551, 2085.2288, 2694.0451, 3181.0752

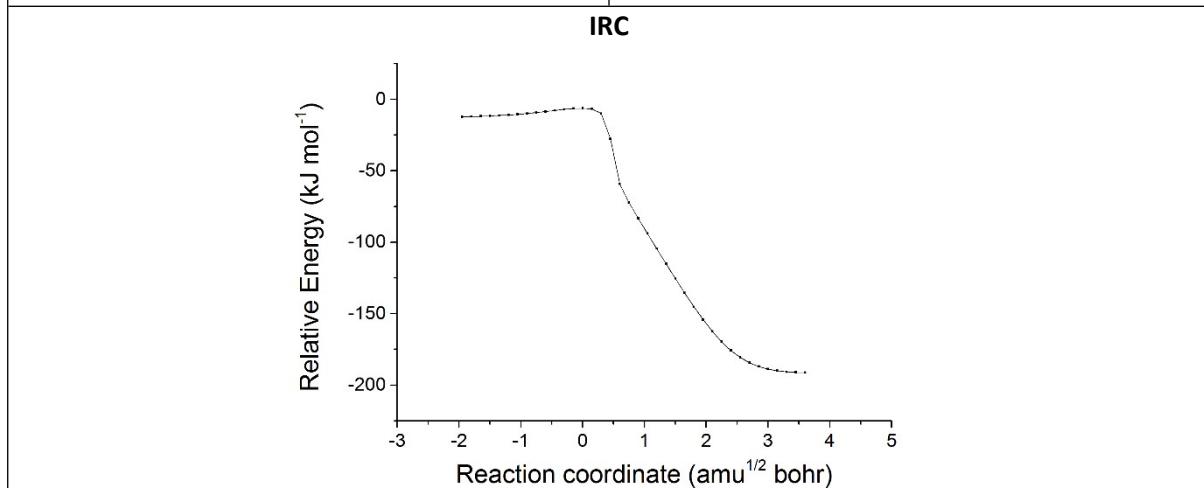


Compound: sCl 2 + H ₂ S Pr2	Energy (kJ mol⁻¹): -925.257286575136
Reaction Coordinates:	Frequencies (cm⁻¹):
6 1.038101 -0.073681 0.017437 6 -0.345508 -0.051413 -0.667474 8 -0.874309 1.246164 -0.774009 8 -1.218307 1.740515 0.541365 1 -0.172466 -0.327747 -1.708250 9 1.795556 0.938754 -0.429649 9 0.976078 0.020668 1.354219 9 1.674062 -1.222024 -0.282036 16 -1.485535 -1.318144 -0.007489	57.7190, 119.0022, 183.5200, 191.0234, 220.5122, 285.8439, 322.5651, 331.8899, 379.7520, 507.0559, 523.3513, 575.9561, 710.5756, 730.3250, 850.4188, 893.4750, 976.3478, 1031.4681, 1131.8965, 1166.8457, 1247.6794, 1285.3044, 1356.0743, 1388.4082, 2684.6621, 3088.2825, 3730.1768

1	-1.301491	-1.020792	1.290740
1	-2.183383	1.659403	0.515897

Compound: sCl 3 + H ₂ S PRC1	Energy (kJ mol⁻¹): -925.175112004368
Reaction Coordinates:	Frequencies (cm⁻¹):
6 1.505791 -0.283359 -0.011193 6 0.260325 0.445841 -0.455155 1 -0.241765 0.189576 -1.378153 8 -0.136738 1.398076 0.254026 8 -1.246501 2.074357 -0.133452 9 2.508066 -0.023426 -0.873639 9 1.895522 0.071654 1.210826 9 1.286545 -1.606577 -0.032310 16 -2.822560 -0.892838 -0.052930 1 -2.655982 0.447927 0.043415 1 -2.683283 -1.081307 1.271215	16.7560, 35.6648, 94.9104, 129.5199, 144.0315, 187.5120, 195.2086, 226.6565, 356.3090, 388.4219, 406.5327, 422.0334, 552.0897, 561.1364, 699.6543, 868.5870, 888.1502, 960.6589, 1135.9140, 1176.5214, 1209.5645, 1267.4663, 1356.5936, 1553.5525, 2574.0746, 2688.7827, 3207.5983

Compound: sCl 3 + H ₂ S TS1	Energy (kJ mol⁻¹): -925.172857190441
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -1.310051 -0.131310 -0.019387 6 0.013684 0.470188 -0.452503 8 0.483121 1.387908 0.280974 8 1.747748 1.813111 -0.134919 1 0.324899 0.372530 -1.482418 9 -1.370451 -0.327937 1.299070 9 -1.499155 -1.299640 -0.643653 9 -2.318385 0.691074 -0.367375 16 2.010557 -1.162230 -0.043231 1 2.319981 0.217673 -0.042042 1 1.809374 -1.207415 1.286670	-211.4560, 36.1197, 92.0602, 175.9207, 196.7871, 283.0048, 313.1884, 378.3885, 411.0245, 418.9613, 486.3471, 552.6049, 586.6452, 698.7495, 805.2558, 876.8659, 919.3305, 933.5828, 1139.9428, 1182.4716, 1216.9751, 1254.8214, 1338.8958, 1515.4506, 1912.7856, 2681.5459, 3207.0263

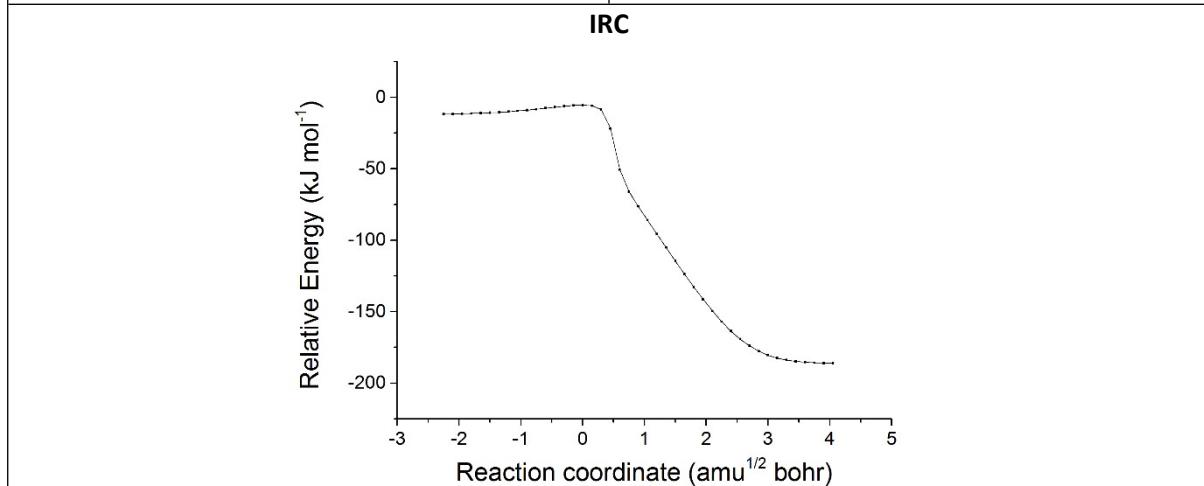


Compound: sCl 3 + H ₂ S Pr1	Energy (kJ mol⁻¹): -925.257735936140
Reaction Coordinates:	Frequencies (cm⁻¹):
6 1.126290 -0.140551 -0.031638 6 -0.374715 -0.078433 -0.357479 8 -0.970615 -1.118009 0.379968 8 -2.255017 -1.394028 -0.209024 1 -0.488446 -0.255832 -1.427000 9 1.372263 0.107002 1.266689 9 1.623239 -1.352756 -0.318668 9 1.801010 0.759696 -0.761611 16 -1.116036 1.571429 -0.048806	63.8128, 102.0609, 186.4836, 187.1548, 245.7465, 301.4951, 321.9712, 351.3219, 365.2552, 454.8641, 534.7094, 561.6093, 685.2206, 715.5061, 873.8865, 932.6839, 970.9561, 1059.7406, 1138.6731, 1177.0163, 1243.1934, 1311.3835, 1339.9383, 1399.0168, 2671.5249, 3082.4983, 3694.2213

1	-0.702156	1.668348	1.228465
1	-2.825833	-0.770645	0.268896

Compound: sCl 3 + H ₂ S PRC2	Energy (kJ mol⁻¹): -925.174429805702
Reaction Coordinates:	Frequencies (cm⁻¹):
6 1.503567 -0.284485 -0.011828 6 0.262393 0.449685 -0.460027 1 -0.232363 0.202682 -1.389467 8 -0.140290 1.396906 0.252344 8 -1.244397 2.078583 -0.143226 9 2.521171 0.007486 -0.846631 9 1.866725 0.037533 1.226876 9 1.293556 -1.607184 -0.077163 16 -2.780476 -0.902092 0.105562 1 -2.645184 0.445483 0.070505 1 -3.286178 -0.950318 -1.139579	16.8693, 36.2707, 93.7280, 127.5679, 142.8341, 192.0942, 193.9604, 235.7731, 350.9350, 388.0237, 405.4637, 420.8305, 552.1704, 560.7231, 699.7189, 868.7005, 887.7995, 959.2261, 1134.4405, 1177.7167, 1212.8077, 1267.9318, 1356.3589, 1554.0258, 2572.1811, 2688.1979, 3207.6933

Compound: sCl 3 + H ₂ S TS2	Energy (kJ mol⁻¹): -925.172354744837
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -1.313290 -0.136608 -0.016959 6 0.012714 0.461428 -0.447671 8 0.472921 1.396341 0.269616 8 1.739945 1.815318 -0.146902 1 0.338165 0.341324 -1.470376 9 -1.399575 -0.286104 1.303791 9 -1.479678 -1.327746 -0.605359 9 -2.320475 0.664428 -0.417352 16 1.993002 -1.140977 0.118183 1 2.311340 0.234651 0.001975 1 2.360541 -1.417753 -1.146189	-217.3443, 37.4007, 92.6508, 175.1327, 197.5688, 276.0734, 313.8069, 377.8487, 409.8519, 428.7319, 504.0821, 553.3237, 586.0863, 698.2452, 797.9012, 875.7437, 913.3066, 924.9756, 1137.9528, 1181.1166, 1221.2218, 1260.3439, 1339.4354, 1516.3591, 1892.3908, 2682.5789, 3211.9961

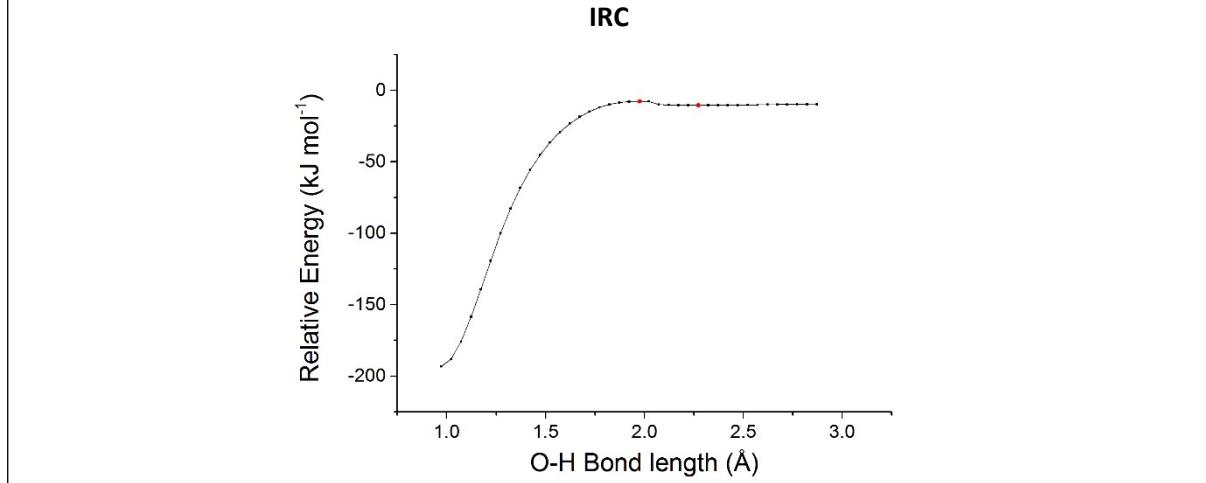


Compound: sCl 3 + H ₂ S Pr2	Energy (kJ mol⁻¹): -925.255754732320
Reaction Coordinates:	Frequencies (cm⁻¹):
6 1.134988 -0.125424 -0.029628 6 -0.364644 -0.085999 -0.364057 8 -0.934831 -1.167480 0.329847 8 -2.255123 -1.380183 -0.204046 1 -0.472188 -0.217740 -1.438933 9 1.367666 0.136530 1.266332 9 1.653715 -1.330722 -0.307283 9 1.796972 0.781311 -0.764700 16 -1.089668 1.545198 0.104755	60.8375, 103.4306, 134.5855, 184.9709, 193.2162, 286.4429, 316.1939, 345.9497, 365.6998, 459.7549, 532.6234, 561.6597, 689.6998, 743.5890, 858.3025, 935.5761, 969.8389, 1060.0747, 1133.8875, 1176.0136, 1248.1780, 1284.2039, 1341.6047, 1396.1450, 2689.6046, 3097.4785, 3703.1385

1 -1.766818 1.718580 -1.042276	
1 -2.793900 -0.858226 0.411706	

Compound: sCl 4 + H ₂ S PRC1	Energy (kJ mol⁻¹): -1024.345549005980
Reaction Coordinates:	Frequencies (cm⁻¹):
6 1.255536 -0.535628 0.094499 6 0.626941 0.785770 -0.380958 8 -0.126808 1.515607 0.276098 8 -0.522536 1.094310 1.539803 9 0.962643 1.208923 -1.564048 9 2.026957 -0.306980 1.155540 9 0.326616 -1.431341 0.398410 9 2.015402 -1.018278 -0.892853 16 -3.000013 -0.564130 -0.301112 1 -3.719085 0.534199 -0.591370 1 -2.365378 0.040780 0.727263	13.8664, 36.9070, 53.5079, 92.9337, 136.2438, 162.2992, 185.3213, 221.2522, 268.6021, 293.8398, 361.4009, 365.2395, 483.2266, 495.6568, 581.6967, 629.1721, 660.3300, 779.6001, 920.3828, 1158.5904, 1193.0932, 1207.0404, 1213.5979, 1399.6569, 1594.5146, 2610.1312, 2690.0168

Compound: sCl 4 + H ₂ S TS1	Energy (kJ mol⁻¹): -1024.344013062570
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -1.068099 -0.371050 -0.242756 6 -0.149806 0.520690 0.627818 8 0.372002 1.604912 0.291859 8 0.697364 1.725914 -1.073496 9 -0.348033 0.376851 1.912128 9 -2.176887 0.344798 -0.489359 9 -0.545925 -0.760632 -1.390633 9 -1.409574 -1.449026 0.466501 16 2.142449 -0.885776 -0.036405 1 2.851505 -0.192450 0.873466 1 1.945591 0.212507 -0.835992	-91.8475, 37.6949, 103.0268, 109.1560, 196.2418, 251.7585, 272.1509, 290.8062, 306.5540, 368.1350, 388.3813, 476.4108, 503.3478, 581.3188, 599.5715, 622.8849, 636.5919, 779.7094, 918.7570, 1129.6648, 1176.7367, 1214.0730, 1236.0744, 1354.9605, 1549.5736, 2369.6784, 2681.9410

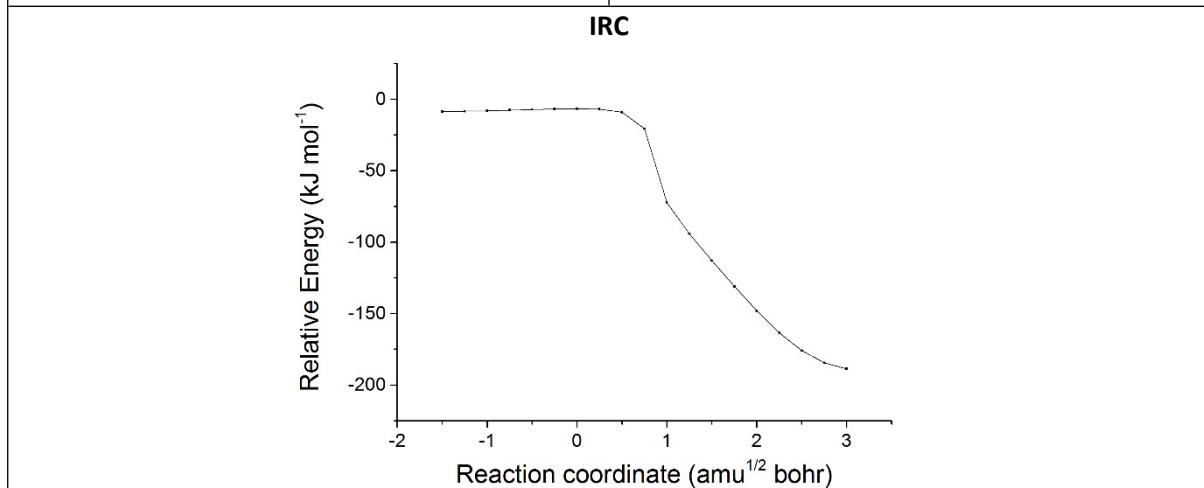


Compound: sCl 4 + H ₂ S Pr1	Energy (kJ mol⁻¹): -1024.429431015390
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -1.063006 -0.096610 -0.176165 6 0.364858 -0.080465 0.466586 8 0.807099 1.210387 0.789287 8 0.864917 2.002775 -0.408568 9 0.265422 -0.664548 1.686821 9 -1.843260 0.815874 0.402587 9 -1.015127 0.146435 -1.491994 9 -1.618803 -1.302127 -0.000355 16 1.534813 -1.013646 -0.610945	54.3934, 121.0861, 167.5154, 207.0012, 213.8160, 266.5950, 294.2510, 326.0987, 341.1989, 365.7182, 402.0517, 504.0054, 533.5151, 570.6240, 619.2433, 741.0527, 848.8025, 965.6506, 986.2968, 1073.9629, 1140.1976, 1183.4904, 1203.7794, 1296.0267, 1415.5088, 2674.3831, 3673.9102

1 2.431715 -1.133411 0.385610	
1 1.729943 1.748193 -0.772298	

Compound: sCl 4 + H ₂ S PRC2	Energy (kJ mol⁻¹): -1024.345353229650
Reaction Coordinates:	Frequencies (cm⁻¹):
6 1.290490 -0.529706 0.062238 6 0.625134 0.799092 -0.334777 8 -0.151494 1.465079 0.362419 8 -0.534378 0.958574 1.598056 9 0.948948 1.301501 -1.489789 9 2.048350 -0.347356 1.141503 9 0.386033 -1.469738 0.301393 9 2.070049 -0.927379 -0.947657 16 -3.108159 -0.396842 -0.360503 1 -2.936307 -1.694105 -0.052990 1 -2.420334 0.024770 0.723423	12.1242, 35.9569, 52.7329, 92.3689, 131.0919, 163.6602, 185.7154, 220.6224, 268.9694, 293.8148, 352.6309, 364.3246, 483.3067, 495.7094, 581.7737, 629.2548, 660.4071, 779.5938, 920.6439, 1159.0171, 1193.5348, 1206.2207, 1213.6770, 1399.8534, 1594.0341, 2612.4339, 2691.7932

Compound: sCl 4 + H ₂ S TS2	Energy (kJ mol⁻¹): -1024.343079346060
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -1.071211 -0.363698 -0.245937 6 -0.143560 0.530370 0.609849 8 0.395490 1.605527 0.265975 8 0.765781 1.693599 -1.091961 9 -0.369535 0.414344 1.892212 9 -2.212221 0.326010 -0.415284 9 -0.596972 -0.699277 -1.430353 9 -1.348295 -1.474452 0.441689 16 2.182199 -0.810150 0.100843 1 1.863614 -1.755094 -0.800983 1 1.962861 0.224832 -0.782456	-106.4507, 34.4170, 112.2563, 116.6783, 198.6133, 249.1393, 270.2184, 276.6378, 290.3873, 369.0565, 394.2326, 476.6437, 508.8478, 581.8044, 606.0935, 631.4851, 641.8063, 780.1505, 918.8704, 1125.8058, 1174.7524, 1218.2103, 1239.2021, 1353.1718, 1543.8164, 2316.0817, 2695.1504

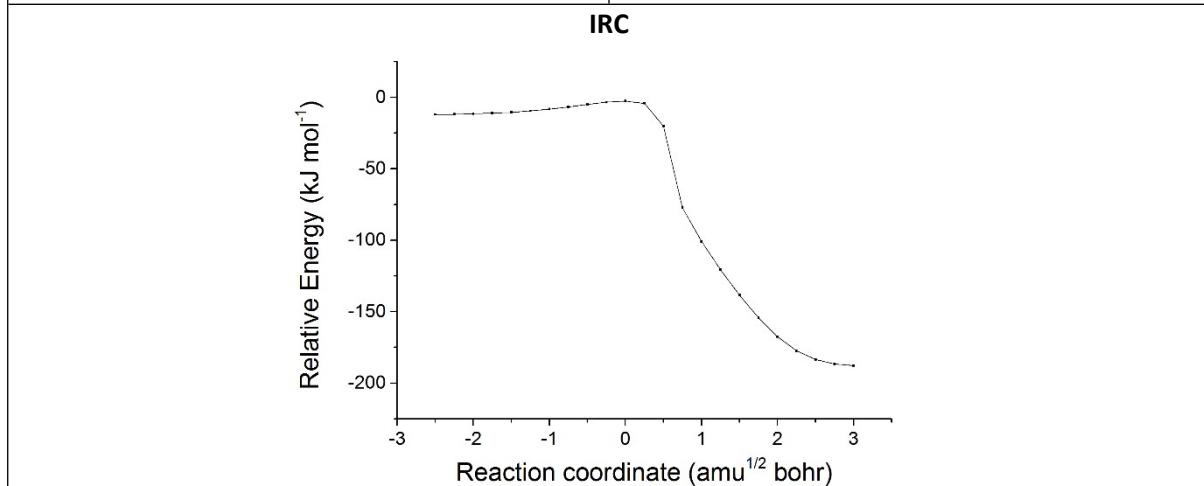


Compound: sCl 4 + H ₂ S Pr2	Energy (kJ mol⁻¹): -1024.428919132390
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -1.106304 0.072990 -0.218224 6 0.285982 -0.089087 0.463772 8 0.862223 1.170956 0.686912 8 1.642238 1.563219 -0.463465 1 2.518795 1.636667 -0.056694 9 0.033549 -0.561191 1.715421 16 1.413265 -1.206650 -0.455943 1 0.747460 -2.301924 -0.046329 9 -1.901340 0.848771 0.527851	57.2635, 70.9939, 175.7844, 201.2701, 214.0455, 240.9941, 300.2199, 308.2382, 339.6347, 369.4160, 400.5555, 502.7278, 530.9276, 585.2028, 602.8116, 736.3936, 886.0484, 946.5138, 983.6877, 1100.7233, 1134.3376, 1177.3711, 1198.8231, 1281.0401, 1405.6513, 2679.1684, 3732.9729

9	-0.982485	0.632823	-1.423831
9	-1.704417	-1.120976	-0.359747

Compound: sCl 5 + H ₂ S PRC1	Energy (kJ mol⁻¹): -1024.352019623190
Reaction Coordinates:	Frequencies (cm⁻¹):
6 1.497901 -0.354035 0.113257 6 0.326787 0.557329 -0.235223 9 -0.019078 0.609081 -1.476740 8 -0.193080 1.273361 0.636115 8 -1.239250 2.096406 0.275803 9 2.611061 0.116300 -0.469978 9 1.680445 -0.398850 1.426610 9 1.270769 -1.583402 -0.350508 16 -2.739843 -0.999585 -0.011305 1 -2.696080 0.342916 0.141142 1 -2.844680 -1.215609 1.311733	24.1450, 33.1121, 65.0647, 98.5150, 140.2987, 166.9719, 185.3141, 197.8389, 294.3611, 326.7809, 360.0382, 382.8149, 406.7516, 511.8452, 576.5956, 655.5405, 728.7778, 856.9027, 892.9799, 1146.7193, 1172.6328, 1209.1424, 1230.9405, 1399.6934, 1611.3284, 2609.0297, 2689.4607

Compound: sCl 5 + H ₂ S TS1	Energy (kJ mol⁻¹): -1024.345922619310
Reaction Coordinates:	Frequencies (cm⁻¹):
6 1.336779 -0.162720 -0.101569 6 -0.019006 0.479715 0.216224 8 -0.515172 1.245306 -0.649683 8 -1.802057 1.722392 -0.318134 9 -0.263186 0.631311 1.489973 9 1.395870 -0.518762 -1.382031 9 1.531605 -1.232089 0.666795 9 2.316371 0.720581 0.146452 16 -1.882243 -1.286634 0.014244 1 -2.329438 0.029629 -0.155333 1 -1.749424 -1.496399 -1.308676	-212.8655, 41.9183, 92.7584, 169.9711, 177.8020, 214.1780, 286.9091, 319.6417, 355.8809, 364.4907, 404.7532, 463.5875, 507.2484, 576.2040, 629.5656, 721.3812, 762.2249, 829.4224, 881.0242, 1129.7304, 1178.1249, 1207.3719, 1225.3137, 1358.6645, 1530.2690, 2071.9114, 2682.1682

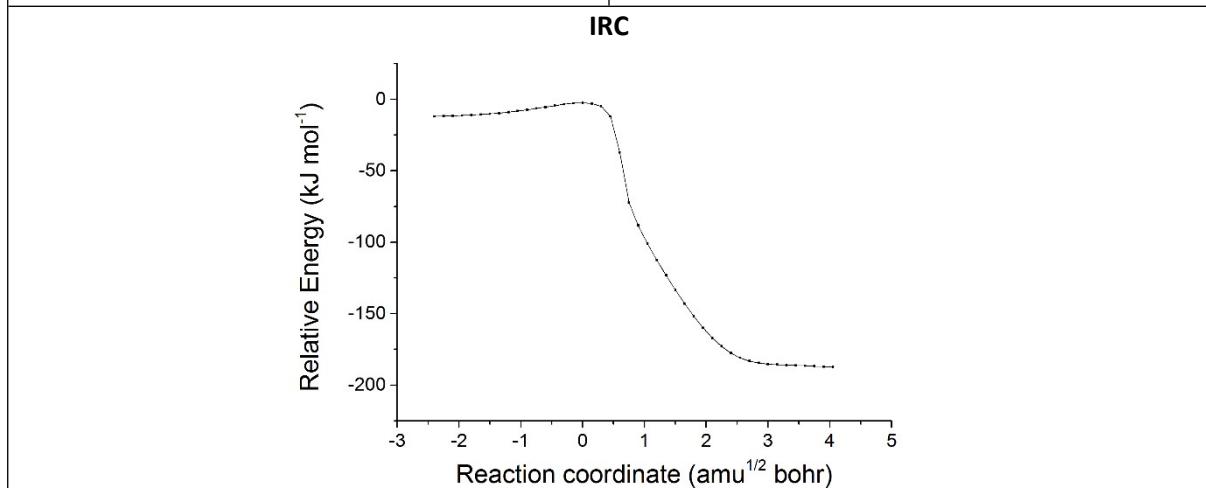


Compound: sCl 5 + H ₂ S Pr1	Energy (kJ mol⁻¹): -1024.433159728430
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -1.200903 -0.095266 -0.099958 6 0.326675 -0.089049 0.223901 8 0.843690 -1.117581 -0.540345 8 2.245704 -1.284167 -0.257708 9 0.464397 -0.342426 1.555653 9 -1.413293 0.221313 -1.387446 9 -1.725040 -1.300860 0.129524 9 -1.837661 0.792828 0.664830 16 1.114990 1.556337 -0.055220	66.2156, 137.0343, 156.8050, 176.9816, 200.0302, 245.6916, 278.2893, 295.7701, 347.7871, 367.4199, 411.0122, 439.2158, 545.3320, 571.0266, 694.6552, 738.6398, 887.3473, 938.7696, 993.2027, 1069.6573, 1123.6025, 1182.6165, 1203.6161, 1282.2079, 1414.9629, 2679.6380, 3680.8781

1 0.651037 1.684950 -1.311783	
1 2.643713 -0.604157 -0.826976	

Compound: sCl 3 + H ₂ S PRC2	Energy (kJ mol⁻¹): -1024.352020666000
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -1.489163 -0.357160 -0.112118 6 -0.325648 0.564108 0.235950 9 0.005099 0.635842 1.480987 8 0.200272 1.270865 -0.639145 8 1.238752 2.103600 -0.278495 9 -2.612149 0.122593 0.444942 9 -1.651870 -0.429464 -1.426202 9 -1.266596 -1.575910 0.381370 16 2.710350 -1.019175 -0.143759 1 2.696623 0.331667 -0.190098 1 3.044108 -1.039826 1.158502	24.7913, 33.7033, 65.0276, 97.7672, 141.4668, 166.3063, 186.3025, 196.5192, 293.6860, 323.5083, 358.0252, 379.9072, 406.7070, 511.7565, 576.6243, 655.0017, 728.6802, 856.4586, 892.6362, 1145.3146, 1172.1473, 1210.5034, 1232.6576, 1399.1867, 1611.2523, 2611.1316, 2690.4468

Compound: sCl 5 + H ₂ S TS2	Energy (kJ mol⁻¹): -1024.345712139260
Reaction Coordinates:	Frequencies (cm⁻¹):
6 1.337645 -0.163068 -0.103782 6 -0.018713 0.475773 0.220019 8 -0.507990 1.263483 -0.629415 8 -1.797439 1.732071 -0.294004 9 -0.268582 0.597141 1.496683 9 1.413666 -0.473263 -1.392902 9 1.515718 -1.259729 0.631286 9 2.317632 0.705926 0.191884 16 -1.859989 -1.285882 -0.160405 1 -2.322831 0.034760 -0.220607 1 -2.193410 -1.431987 1.134461	-214.3248, 43.9396, 94.7246, 172.0993, 177.1296, 218.7101, 287.8239, 316.2546, 346.1578, 363.8654, 405.7866, 461.7908, 513.6250, 576.1385, 630.2736, 720.9692, 758.5877, 828.0010, 879.3764, 1128.0337, 1176.9447, 1213.5395, 1228.9550, 1357.9335, 1531.4213, 2070.7635, 2687.8493



Compound: sCl 5 + H ₂ S Pr2	Energy (kJ mol⁻¹): -1024.433112228060
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -1.202362 -0.089113 -0.109268 6 0.318048 -0.089101 0.243524 8 0.827351 -1.177761 -0.442171 8 2.255224 -1.241345 -0.260519 9 0.431335 -0.281065 1.592046 9 -1.379704 0.255548 -1.393450 9 -1.735591 -1.297017 0.084936 9 -1.853762 0.786093 0.658312 16 1.058272 1.538292 -0.208423	60.9853, 112.5120, 142.6837, 171.9181, 186.3778, 257.8669, 279.9189, 301.2905, 349.5684, 369.4286, 417.1116, 449.7881, 543.9532, 571.0201, 686.0786, 737.2994, 855.3090, 942.0929, 999.5997, 1067.2588, 1100.2573, 1187.4977, 1204.3373, 1292.5349, 1408.4410, 2689.6565, 3708.1167

1 1.989300 1.468211 0.758434	
1 2.563125 -0.830796 -1.084279	

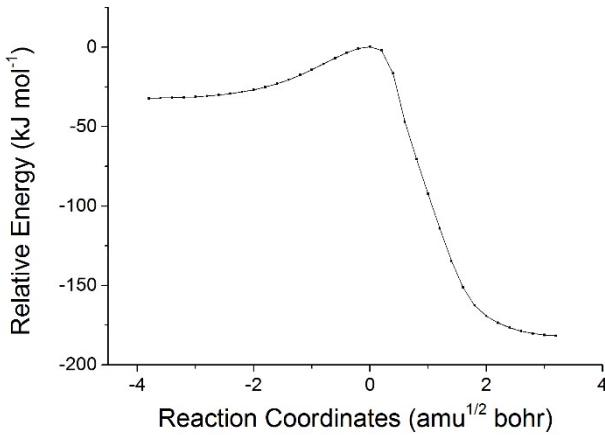
S8.9 Reactions with H₂O

Compound: H ₂ O	Energy (kJ mol⁻¹): -76.372669648641
Reaction Coordinates:	Frequencies (cm⁻¹):
8 -0.000000 -0.000000 0.116992	1626.9293, 3796.3639, 3899.0961
1 -0.000000 0.763520 -0.467967	
1 -0.000000 -0.763520 -0.467967	

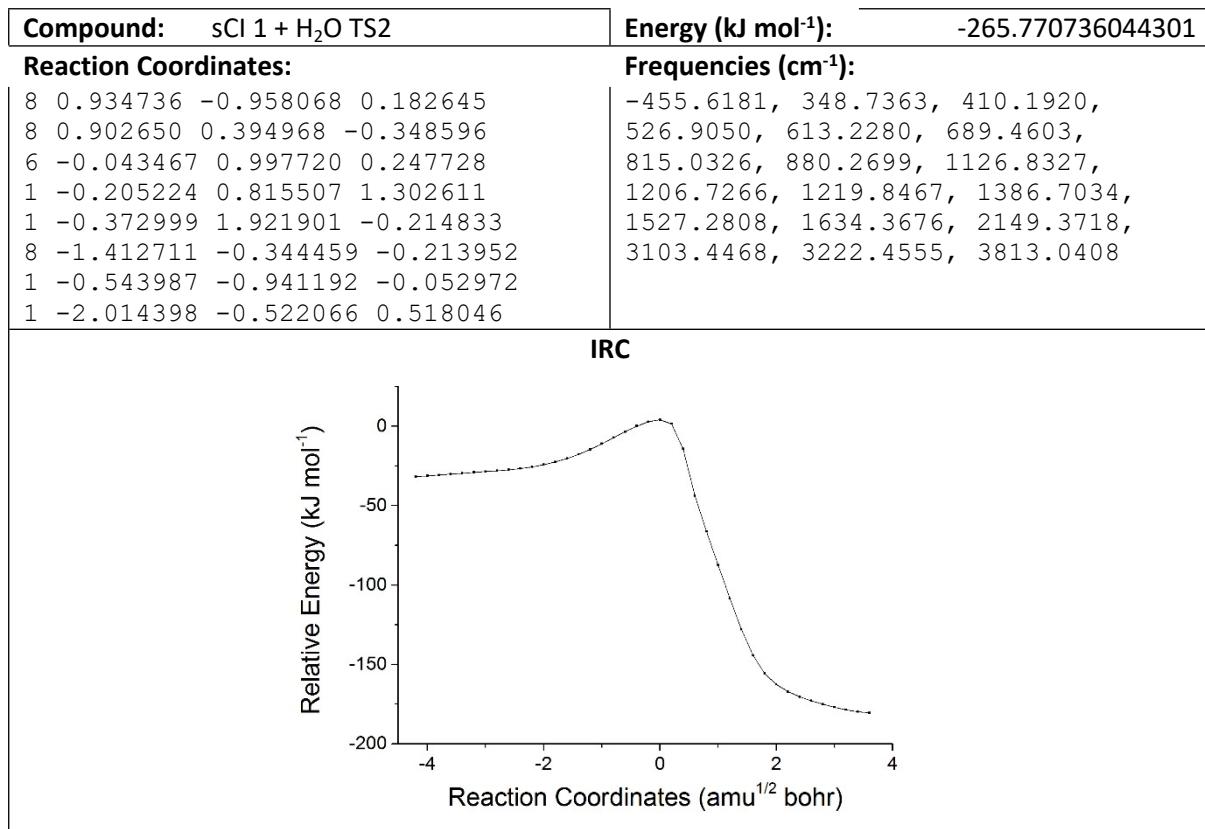
Compound: sCl 1 + H ₂ O PRC	Energy (kJ mol⁻¹): -265.786257491251
Reaction Coordinates:	Frequencies (cm⁻¹):
8 0.357047 -1.171861 -0.000014	21.8780, 75.7289, 148.6369,
8 1.289920 -0.173564 0.000019	209.9306, 493.6039, 549.1761,
6 0.924903 1.024191 -0.000004	629.7151, 691.8102, 881.1881,
1 -0.129754 1.281859 -0.000044	1017.5999, 1272.0176, 1429.7599,
1 1.739280 1.736480 0.000025	1558.0952, 1645.6134, 3091.7849,
8 -2.012469 0.273086 -0.000021	3248.5013, 3499.8200, 3871.1216
1 -1.353128 -0.449947 -0.000007	
1 -2.881795 -0.134832 0.000186	

Compound: sCl 1 + H ₂ O TS1	Energy (kJ mol⁻¹): -265.772122072372
Reaction Coordinates:	Frequencies (cm⁻¹):
8 -0.950605 -0.946963 0.167769	-433.3116, 326.5769, 426.8495,
8 -0.886084 0.399108 -0.376202	531.9388, 597.5414, 656.1524,
6 0.046206 0.999114 0.246331	809.2086, 899.0486, 1157.5178,
1 0.186984 0.820730 1.304039	1211.4121, 1251.2657, 1392.4448,
1 0.394195 1.921099 -0.209200	1546.5882, 1605.7472, 2196.2030,
8 1.446354 -0.374579 -0.005867	3103.8784, 3225.5749, 3798.4004
1 0.557419 -0.948603 0.076115	
1 1.706835 -0.408441 -0.934541	

IRC



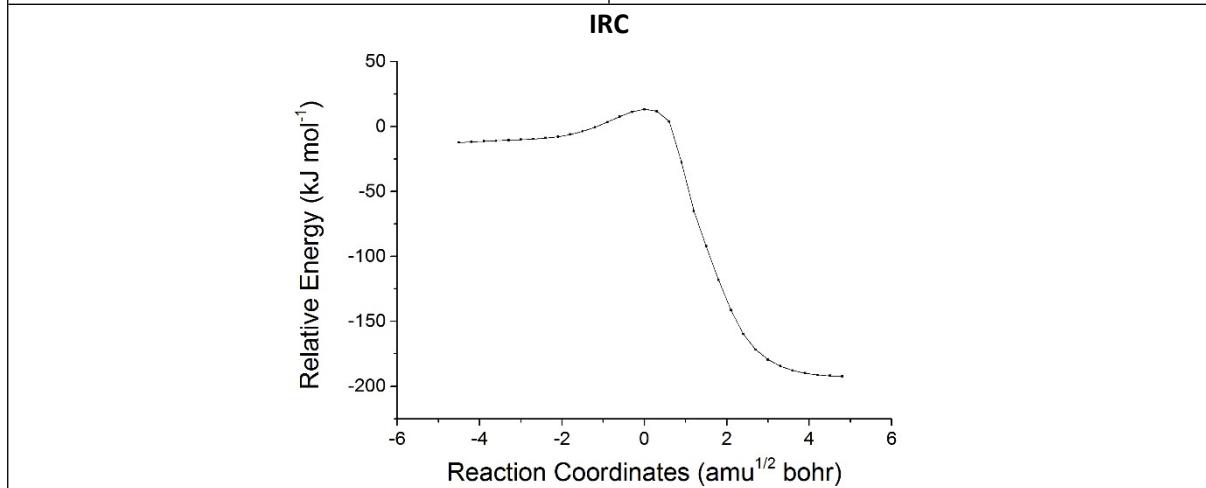
Compound:	sCl 1 + H ₂ O Pr1	Energy (kJ mol⁻¹):	-265.849083661745
Reaction Coordinates:		Frequencies (cm⁻¹):	
6 0.613216 0.542122 0.285606 8 1.381072 -0.594495 -0.014848 8 -0.614068 0.578422 -0.401400 8 -1.485958 -0.421022 0.182400 1 0.443979 0.526639 1.360537 1 1.100746 1.469391 -0.028131 1 -1.183095 -1.228893 -0.258159 1 1.710703 -0.523108 -0.917100		159.0267, 283.2184, 374.4906, 424.9952, 616.1269, 877.8835, 1010.2947, 1032.0533, 1068.5203, 1276.8372, 1369.8581, 1382.2124, 1414.7043, 1484.3437, 3030.0864, 3111.6326, 3728.0279, 3798.4265	



Compound:	sCl 1 + H ₂ O Pr2	Energy (kJ mol⁻¹):	-265.848846609638
Reaction Coordinates:		Frequencies (cm⁻¹):	
6 -0.647196 0.538648 0.261567 8 0.625293 0.629426 -0.338740 8 1.425464 -0.454794 0.213290 1 1.601714 -0.966752 -0.587818 1 -0.529616 0.517625 1.347966 1 -1.138554 1.456034 -0.061035 8 -1.414034 -0.537971 -0.192946 1 -1.144148 -1.332081 0.278647		167.4320, 205.9488, 363.0291, 433.1822, 606.0638, 871.9321, 998.8296, 1044.5819, 1074.2590, 1272.8836, 1372.2326, 1384.1981, 1412.5021, 1484.6978, 3025.6587, 3107.6286, 3758.6628, 3815.8683	

Compound: sCl 2 + H ₂ O PRC	Energy (kJ mol⁻¹): -602.585302654489
Reaction Coordinates:	Frequencies (cm⁻¹):
6 1.418122 -0.035776 -0.000003 6 0.319563 1.008948 -0.000001 8 -0.904792 0.737430 0.000011 8 -1.312365 -0.546032 0.000023 1 0.561195 2.062640 -0.000010 9 1.370649 -0.806322 1.087881 9 1.370633 -0.806333 -1.087879 9 2.594145 0.620050 -0.000015 8 -4.162630 0.061529 -0.000005 1 -4.718098 -0.722268 -0.000085 1 -3.249743 -0.269386 0.000006	12.9685, 28.0322, 34.6398, 84.0523, 109.5999, 211.6467, 247.2163, 334.3561, 375.5031, 479.5982, 510.8715, 534.4531, 536.6024, 591.4602, 759.7032, 800.6129, 884.1656, 931.3447, 1157.2991, 1183.4831, 1242.5624, 1368.6752, 1557.3767, 1658.4346, 3213.8618, 3648.1189, 3869.8220

Compound: sCl 2 + H ₂ O TS1	Energy (kJ mol⁻¹): -602.571519549324
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -0.880942 -0.040291 0.035680 6 0.374737 -0.063934 -0.855948 8 1.370730 -0.828340 -0.705354 8 1.782336 -0.820106 0.676516 1 0.161124 0.207677 -1.886552 9 -1.469403 -1.243069 -0.117446 9 -0.724983 0.193086 1.324767 9 -1.712041 0.890310 -0.469775 8 1.168700 1.545355 0.045488 1 1.846465 1.896020 -0.545448 1 1.613361 0.783438 0.582497	-335.6905, 48.6609, 176.1151, 213.7137, 280.4627, 322.3603, 348.6340, 448.8651, 514.6696, 537.3357, 550.1125, 572.5144, 667.5975, 744.2436, 808.1946, 872.6150, 993.2852, 1139.3767, 1148.0772, 1195.9793, 1292.0680, 1346.8227, 1554.7138, 1616.4176, 2586.7975, 3144.4219, 3801.2117



Compound: sCl 2 + H ₂ O Pr1	Energy (kJ mol⁻¹): -602.660056059595
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -0.840616 -0.134696 0.006007 6 0.411407 0.581582 -0.551944 8 1.505490 -0.266093 -0.775674 8 1.864918 -0.931170 0.458894 1 0.162315 0.911744 -1.564471 9 -1.020367 -1.310094 -0.610249 9 -0.798843 -0.344692 1.322507 9 -1.922869 0.632033 -0.251339 8 0.694373 1.640349 0.310457	64.6652, 135.2115, 205.5945, 222.1310, 272.2757, 294.3622, 343.4660, 378.4008, 436.4043, 525.6920, 583.1678, 617.3587, 763.2320, 842.4769, 902.3706, 1043.7056, 1097.6907, 1131.6755, 1170.8830, 1236.4595, 1291.7286, 1366.8442, 1389.1393, 1442.1396, 3049.9342, 3723.7563, 3812.8925

1	1.185863	2.316989	-0.166310
1	2.387533	-0.249960	0.908728

Compound: sCl 2 + H ₂ O TS2	Energy (kJ mol⁻¹): -602.571987979725
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -0.872000 -0.055866 0.026501 6 0.388125 -0.059725 -0.862867 8 1.376054 -0.829330 -0.693386 8 1.752043 -0.797944 0.701867 1 0.188509 0.212024 -1.895001 9 -1.361524 -1.300237 0.049360 9 -0.744726 0.376239 1.278093 9 -1.777175 0.740324 -0.583722 8 1.278925 1.534698 -0.102153 1 0.727023 2.081282 0.470334 1 1.682365 0.793917 0.498653	-353.5599, 32.7498, 178.7265, 213.7614, 289.5076, 323.6146, 366.4348, 447.5482, 507.7217, 533.8073, 567.4192, 588.3382, 709.5027, 739.4606, 811.9352, 864.6585, 1000.0724, 1111.6344, 1160.0164, 1190.1058, 1257.9738, 1345.9494, 1542.6598, 1655.3576, 2544.8281, 3157.6649, 3805.1193
IRC	
<p>Relative Energy (kJ mol⁻¹)</p> <p>Reaction Coordinates (amu^{1/2} bohr)</p>	

Compound: sCl 2 + H ₂ O Pr2	Energy (kJ mol⁻¹): -602.658023397310
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -0.841050 -0.157441 0.010445 6 0.405739 0.532657 -0.593230 8 1.494897 -0.309691 -0.777127 8 1.946041 -0.824612 0.499873 1 0.139028 0.801180 -1.622219 9 -1.170039 -1.245554 -0.691500 9 -0.722055 -0.493929 1.295152 9 -1.878208 0.721576 -0.076130 8 0.792725 1.643641 0.164788 1 0.020802 2.136425 0.465070 1 2.515462 -0.102443 0.805901	45.8073, 137.2188, 149.7556, 211.4607, 231.3373, 267.9918, 313.5215, 368.2515, 437.2812, 520.0905, 583.4169, 618.8499, 755.5434, 854.2256, 905.1872, 1039.2504, 1097.6375, 1127.1441, 1192.6499, 1236.5467, 1296.0157, 1351.6378, 1398.2769, 1425.9067, 3020.2585, 3727.7188, 3803.4861

Compound: sCl 3 + H ₂ O PRC	Energy (kJ mol⁻¹): -602.589934294451
Reaction Coordinates:	Frequencies (cm⁻¹):
6 1.265134 0.155609 -0.000000 6 -0.238467 0.008675 0.000003 8 -0.691074 -1.158101 -0.000001 8 -2.031785 -1.370658 0.000001 1 -0.897012 0.874551 0.000007 9 1.644005 0.852596 1.085581 9 1.897770 -1.014649 -0.000008 9 1.644000 0.852607 -1.085576 8 -2.920854 1.280203 0.000006	18.9310, 29.3495, 78.1991, 106.2923, 155.7969, 203.6063, 225.9814, 381.2152, 394.8102, 416.6995, 467.3076, 556.2957, 557.9028, 579.8286, 700.0094, 892.9234, 942.1189, 959.7400, 1141.7667, 1183.2773, 1280.4930, 1388.8948, 1556.7441, 1626.8972, 3124.7371, 3577.8101, 3872.5923

1 -3.823997	1.606901	-0.000046
1 -2.961273	0.306298	-0.000000

Compound: sCl 3 + H ₂ O TS1	Energy (kJ mol⁻¹): -602.577917489654
Reaction Coordinates:	Frequencies (cm⁻¹):
6 1.039070 -0.051694 -0.022089 6 -0.419427 -0.182932 -0.448222 8 -1.146757 -0.949377 0.249124 8 -2.521237 -0.668508 -0.120692 1 -0.665573 0.023250 -1.482782 9 1.617879 0.945434 -0.698388 9 1.153163 0.194419 1.287084 9 1.694390 -1.188659 -0.300101 8 -1.326421 1.520273 -0.024545 1 -1.163554 1.758943 0.897664 1 -2.122302 0.845713 -0.021481	-344.2658, 39.3846, 140.2218, 175.5019, 268.4806, 340.7036, 381.7182, 433.9197, 491.8150, 551.9982, 564.2127, 603.5697, 674.9171, 717.4654, 848.3778, 884.9556, 1079.4615, 1171.3551, 1184.0130, 1221.8795, 1258.0253, 1343.7326, 1536.7055, 1608.3451, 2435.5100, 3172.2175, 3783.6617
IRC	
<p>Relative Energy (kJ mol⁻¹)</p> <p>Reaction Coordinates (amu^{1/2} bohr)</p>	

Compound: sCl 3 + H ₂ O Pr1	Energy (kJ mol⁻¹): -602.662813970698
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -0.963295 -0.146555 -0.034177 6 0.480869 0.288837 -0.359532 8 1.306945 -0.586625 0.368263 8 2.626042 -0.522137 -0.216993 1 0.654164 0.172304 -1.428349 9 -1.168282 -1.440472 -0.294445 9 -1.249598 0.074900 1.266542 9 -1.829589 0.568705 -0.767756 8 0.691587 1.629400 -0.052602 1 0.435506 1.791110 0.864764 1 3.035510 0.189595 0.297415	66.6299, 105.1886, 185.4932, 233.3104, 264.2291, 352.9867, 355.6744, 385.5909, 428.9955, 519.5011, 563.7230, 626.6447, 712.4712, 857.8522, 949.6848, 1040.3403, 1088.3219, 1151.0923, 1194.1080, 1250.6670, 1293.5017, 1361.2445, 1390.1415, 1437.9011, 3102.6609, 3728.9304, 3764.5581

Compound: sCl 3 + H ₂ O TS2	Energy (kJ mol⁻¹): -602.576912007240
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -1.039856 -0.049677 -0.010843 6 0.419539 -0.170430 -0.434149 8 1.150764 -0.956828 0.233990 8 2.523335 -0.666412 -0.143361 1 0.657816 0.049055 -1.468419 9 -1.552721 1.085282 -0.508187 9 -1.729827 -1.080080 -0.533333 9 -1.194383 -0.057248 1.306915 8 1.306977 1.471994 0.180893 1 1.388018 2.132416 -0.517541 1 2.119841 0.817564 0.095175	-364.8804, 52.2464, 143.0267, 175.9772, 274.1325, 343.0331, 381.2444, 435.7241, 478.2902, 553.8646, 583.4874, 590.5433, 694.4015, 732.1931, 850.8644, 881.0498, 1063.1370, 1153.9872, 1161.0462, 1190.5992, 1280.3717, 1337.8593, 1529.0559, 1631.0497, 2387.3409, 3161.7624, 3808.2251
IRC	
<p>Relative Energy (kJ mol⁻¹)</p> <p>Reaction Coordinates (amu^{1/2} bohr)</p>	

Compound: sCl 3 + H ₂ O Pr2	Energy (kJ mol⁻¹): -602.660922289812
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -0.977653 -0.122020 -0.020540 6 0.472388 0.292316 -0.339887 8 1.280676 -0.678073 0.280541 8 2.623938 -0.461809 -0.218113 1 0.609833 0.242347 -1.422859 9 -1.229129 -1.358117 -0.477948 9 -1.232482 -0.098538 1.291950 9 -1.828074 0.720918 -0.628725 8 0.676183 1.576208 0.153493 1 1.316213 2.032740 -0.399749 1 3.066331 -0.165837 0.590302	67.1172, 105.0116, 130.0950, 184.4267, 243.6539, 324.8792, 345.4993, 359.5577, 404.8713, 522.4187, 565.5794, 632.5000, 706.2299, 854.7571, 950.9452, 1050.2587, 1103.0362, 1142.7893, 1177.1310, 1242.9134, 1283.6311, 1354.4268, 1388.8170, 1458.0452, 3049.1963, 3747.1703, 3833.6975

Compound: sCl 4 + H ₂ O PRC1	Energy (kJ mol⁻¹): -701.757032802282
Reaction Coordinates:	Frequencies (cm⁻¹):
6 0.967467 -0.448045 -0.009361 6 0.154124 0.861668 -0.039692 8 -0.853476 1.115458 0.628635 8 -1.371797 0.108216 1.448886 9 0.598175 1.811820 -0.805522 9 1.460108 -0.630382 1.215244 9 0.228312 -1.484773 -0.360912 9 1.985779 -0.325431 -0.866525 8 -2.659392 -0.685120 -0.979841	8.9142, 50.9829, 56.8709, 125.1006, 160.5746, 181.1844, 225.8585, 279.1395, 292.2263, 336.4398, 363.7598, 481.9951, 500.3346, 573.4040, 583.7145, 626.9831, 674.6696, 778.6339, 914.2553, 1156.3613, 1190.7739, 1224.5363, 1401.3945, 1608.7930, 1641.3966, 3642.7446, 3869.6071

1	-3.601914	-0.543870	-1.101342
1	-2.501686	-0.587409	-0.026351

Compound: sCl 4 + H ₂ O TS1	Energy (kJ mol⁻¹): -701.754722973199
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -0.922876 -0.252529 0.014628 6 0.362255 0.565884 -0.294872 8 1.349650 0.211899 -0.976788 8 1.789010 -1.121052 -0.588485 9 0.114206 1.848404 -0.318084 9 -1.590196 -0.321598 -1.153003 9 -0.722079 -1.470293 0.466790 9 -1.669098 0.426961 0.885625 8 1.193811 0.106194 1.564075 1 1.851343 0.781472 1.770174 1 1.657121 -0.589197 0.988920	-182.9477, 43.9809, 165.4231, 198.8283, 243.5936, 262.9655, 302.4476, 351.3823, 371.3221, 477.0843, 499.7850, 525.3643, 565.6404, 605.5389, 625.5149, 679.1056, 780.3852, 887.6970, 1089.4681, 1123.5063, 1179.4890, 1256.3551, 1357.2587, 1570.7234, 1615.7012, 2887.0314, 3807.7211
IRC	
<p>Relative Energy (kJ mol⁻¹)</p> <p>Reaction Coordinates (amu^{1/2} bohr)</p>	

Compound: sCl 4 + H ₂ O Pr1	Energy (kJ mol⁻¹): -701.845677031827
Reaction Coordinates:	Frequencies (cm⁻¹):
6 0.889789 -0.270159 -0.010129 6 -0.431897 0.565724 -0.011041 8 -1.498264 -0.049484 0.670262 8 -1.771437 -1.321015 0.048761 9 -0.220327 1.671207 0.757854 9 1.087753 -0.810191 1.194456 9 0.872008 -1.241878 -0.923082 9 1.921381 0.540493 -0.285449 8 -0.714328 0.904974 -1.300334 1 -1.390579 1.595805 -1.297851 1 -2.431879 -1.081688 -0.618650	51.9068, 129.7086, 196.6813, 208.7458, 233.2065, 285.1346, 323.0522, 346.4562, 381.7148, 419.7601, 492.7066, 538.4071, 588.2672, 653.1779, 672.0489, 781.1123, 972.7409, 1033.4888, 1094.4151, 1182.9092, 1202.5091, 1207.0919, 1232.2767, 1402.9368, 1432.4007, 3728.7382, 3765.6990

Compound: sCl 4 + H ₂ O PRC2	Energy (kJ mol⁻¹): -701.757770997730
Reaction Coordinates:	Frequencies (cm⁻¹):
6 0.977575 -0.375535 -0.047959 6 0.014302 0.825274 0.048720 8 -0.967734 0.922922 0.791449 8 -1.354925 -0.209247 1.516763 9 0.320286 1.874211 -0.652500 9 1.562304 -0.551250 1.138425 9 0.366719 -1.487941 -0.415148 9 1.920185 -0.080620 -0.948960 8 -2.458473 -0.520975 -1.109929	22.4289, 55.7258, 71.6779, 148.8436, 161.0054, 182.5709, 232.8751, 282.2735, 294.0666, 364.2100, 373.9421, 478.9771, 500.5261, 510.9374, 581.3517, 626.6840, , 671.8741, 778.5441, 913.6810, 1155.7015, 1186.0906, 1223.3890, 1399.8811, 1611.1781, 1643.5507, 3661.4245, 3872.2454

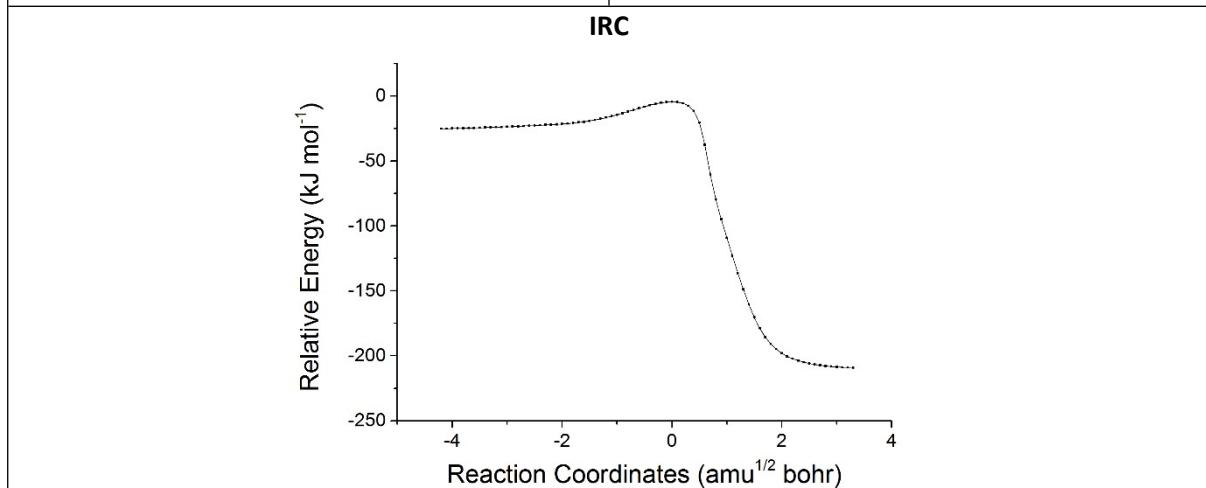
1 -2.795570 -1.317291 -1.528523	
1 -2.432075 -0.712349 -0.158653	

Compound: sCl 4 + H ₂ O TS2	Energy (kJ mol⁻¹): -701.754639690128
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -0.917298 -0.252424 -0.015764 6 0.371649 0.577981 -0.264789 8 1.364908 0.247768 -0.952222 8 1.799098 -1.098137 -0.590989 9 0.113971 1.855245 -0.249161 9 -1.583242 -0.275303 -1.178912 9 -0.722470 -1.491868 0.394882 9 -1.667441 0.384204 0.892193 8 1.278018 0.176684 1.517952 1 0.761571 -0.259947 2.205324 1 1.708783 -0.554405 0.949053	-224.4680, 28.2180, 165.0634, 199.2546, 254.2594, 262.0571, 307.3955, 357.3279, 373.8725, 477.9454, 515.0983, 551.0126, 567.1130, 602.8137, 611.6785, 727.2799, 786.6461, 895.3756, 1061.2898, 1139.5150, 1176.2030, 1240.2368, 1354.8576, 1556.1705, 1650.3191, 2770.0033, 3814.6039
IRC	
<p>The plot shows a reaction coordinate path starting at approximately -10 kJ mol⁻¹ and remaining flat until about 0.5 amu¹/² bohr. It then drops sharply, reaching a minimum of about -220 kJ mol⁻¹ at 3.5 amu¹/² bohr.</p>	

Compound: sCl 4 + H ₂ O Pr2	Energy (kJ mol⁻¹): -701.845674474457
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -0.888438 -0.308532 -0.010458 6 0.410990 0.559492 0.002560 8 1.463801 0.006090 -0.725260 8 1.980769 -1.149354 -0.036166 9 0.109977 1.683447 -0.712380 9 -1.317669 -0.516289 -1.252096 9 -0.715606 -1.485877 0.590294 9 -1.851494 0.359527 0.669975 8 0.819587 0.863153 1.266749 1 0.065762 1.113776 1.815604 1 2.658803 -0.745908 0.527062	46.9670, 131.6299, 205.9108, 226.8588, 245.4856, 290.8668, 306.3950, 347.6743, 377.9855, 414.5608, 484.6199, 529.2277, 590.4181, 642.2163, 692.5148, 784.1847, 968.1183, 1044.0527, 1092.7799, 1158.1546, 1195.9693, 1221.8558, 1289.7655, 1378.5484, 1419.4410, 3725.9129, 3782.9088

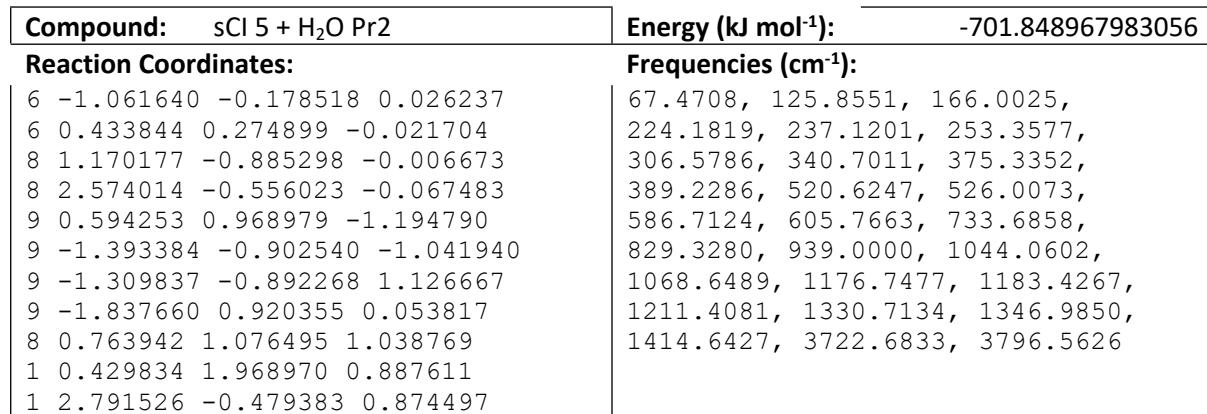
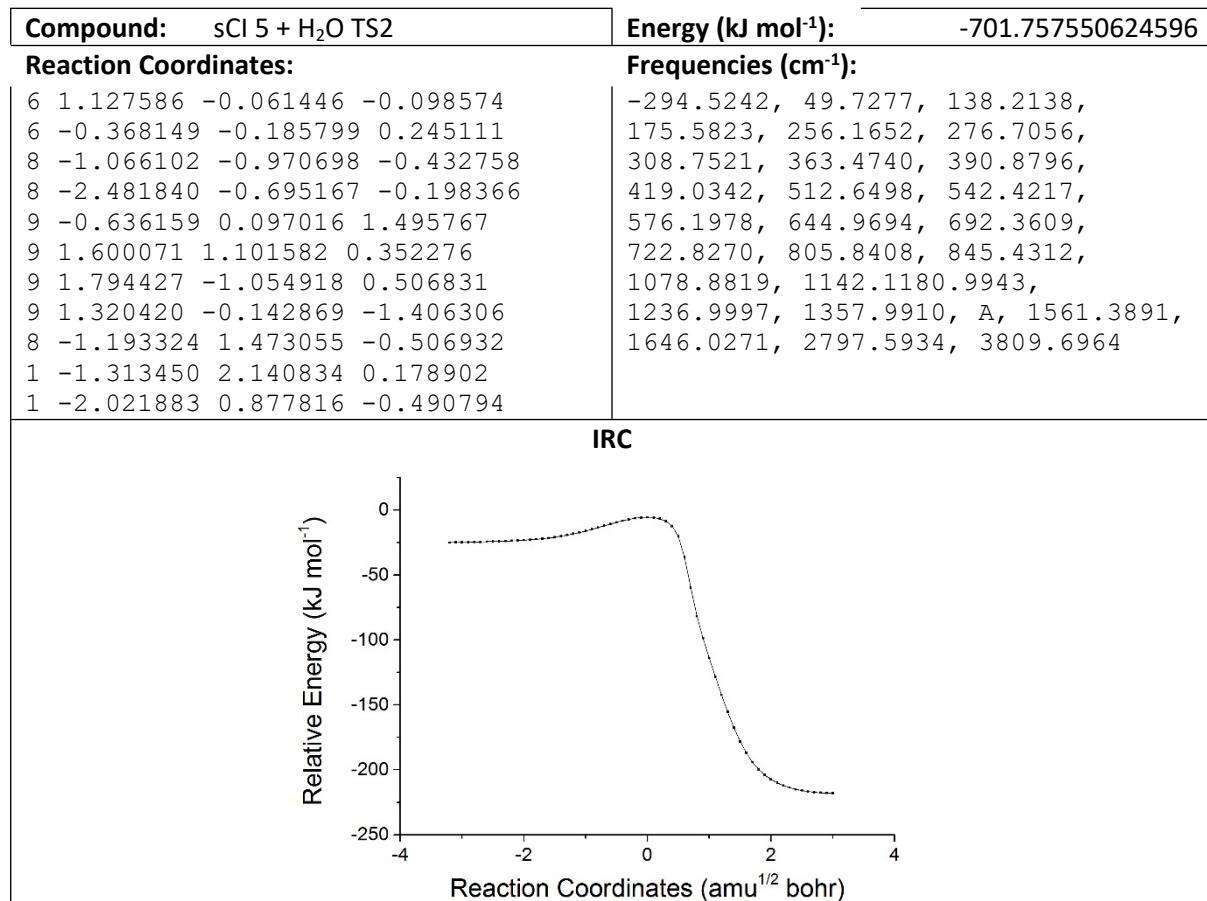
Compound: sCl 5 + H ₂ O PRC1	Energy (kJ mol⁻¹): -701.764103446066
Reaction Coordinates:	Frequencies (cm⁻¹):
6 1.253199 0.089592 0.115211 6 -0.153286 -0.368008 -0.261044 8 -0.872970 -0.931109 0.575069 8 -2.156873 -1.298958 0.195285 9 -0.507050 -0.188673 -1.485279 9 1.402006 1.379556 -0.186073 9 1.477454 -0.097338 1.407847 9 2.147090 -0.615696 -0.594844 8 -2.203577 1.547561 0.297113 1 -2.753188 2.142143 -0.220012 1 -2.654432 0.687754 0.280419	32.7360, 46.8598, 92.4712, 157.7812, 179.9572, 181.7624, 189.5062, 294.9945, 332.2552, 361.8550, 371.8605, 406.9222, 515.8260, 532.4507, 576.6471, 676.3573, 728.5135, 858.4915, 876.1783, 1149.8613, 1173.9584, 1235.9757, 1403.0899, 1628.8278, 1641.2467, 3658.9141, 3871.8932

Compound: sCl 5 + H ₂ O TS1	Energy (kJ mol⁻¹): -701.756908629529
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -1.126733 -0.047615 -0.093143 6 0.367795 -0.221556 0.235940 8 1.063441 -0.897170 -0.559318 8 2.478562 -0.642997 -0.295352 9 0.643596 -0.130985 1.504069 9 -1.657265 0.907431 0.663875 9 -1.279619 0.274686 -1.378264 9 -1.769314 -1.197000 0.143966 8 1.217696 1.555346 -0.143389 1 1.005306 1.861075 -1.034714 1 2.034135 0.945334 -0.240420	-292.2117, 38.7233, 137.0646, 175.9347, 252.9241, 290.7045, 310.8719, 365.5904, 388.1022, 418.6521, 507.5719, 527.4436, 574.7313, 656.1128, 696.2299, 726.4874, 809.6219, 845.9113, 1123.2579, 1160.3658, 1191.9307, 1222.6289, 1369.5290, 1563.9859, 1622.8314, 2744.9152, 3792.4031



Compound: sCl 5 + H ₂ O Pr1	Energy (kJ mol⁻¹): -701.845741873404
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -1.062154 -0.164450 0.008092 6 0.430105 0.300164 -0.048891 8 1.168471 -0.873619 0.086297 8 2.569482 -0.553850 -0.039440 9 0.649193 0.876050 -1.249055 9 -1.314131 -1.157639 -0.839044 9 -1.351761 -0.581518 1.259980 9 -1.864993 0.861751 -0.279192 8 0.705758 1.255653 0.899749	62.5858, 126.1553, 156.3520, 187.5452, 228.7942, 295.2365, 316.7880, 342.0580, 377.0116, 393.9381, 521.8175, 530.1352, 587.2667, 609.4954, 733.9792, 831.8169, 943.4958, 1051.7784, 1091.0788, 1157.7878, 1177.4707, 1219.1982, 1284.5333, 1384.6587, 1403.4261, 3729.2527, 3786.2786

1 0.356738 0.972335 1.754092	
1 2.821091 -0.399880 0.883653	

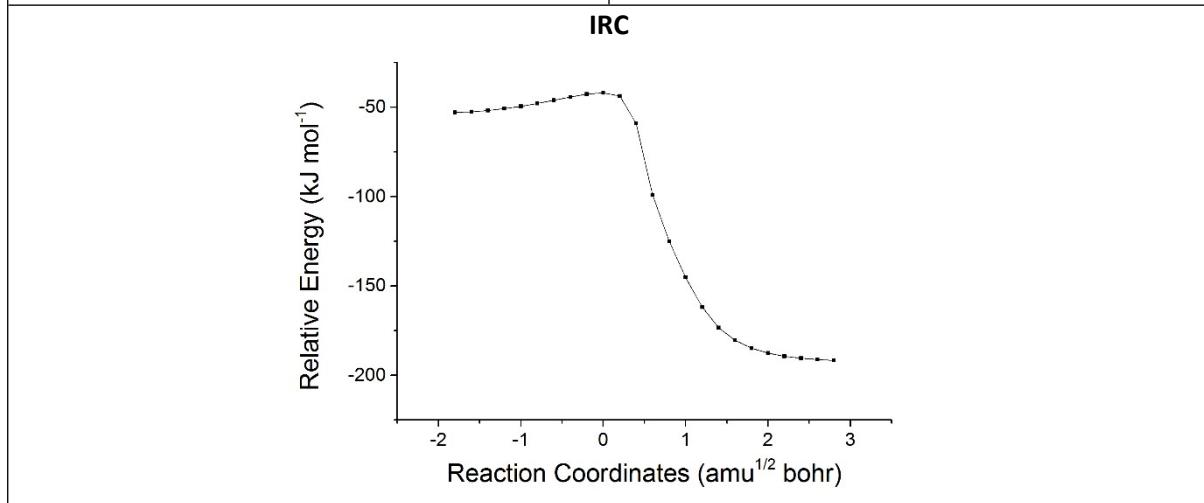


S8.10 Reactions with (H₂O)₂

Compound: (H ₂ O) ₂	Energy (kJ mol⁻¹): -152.753039538317
Reaction Coordinates: 8 -1.516947 0.000002 -0.121143 1 -1.926876 -0.000012 0.747807 1 -0.561489 -0.000002 0.044554 8 1.391185 -0.000002 0.109911 1 1.747231 -0.766159 -0.351264 1 1.747228 0.766171 -0.351242	Frequencies (cm⁻¹): 136.9386, 159.6379, 161.0494, 190.5114, 364.4335, 626.4275, 1628.1114, , 1647.2164, 3675.9138, 3791.8236, 3871.0307, 3890.8064

Compound: sCl 1 + (H ₂ O) ₂ PRC1	Energy (kJ mol⁻¹): -342.172906791257
Reaction Coordinates: 6 -0.941512 -1.103855 0.343798 8 -1.411547 -0.220247 -0.397069 8 -1.297519 1.104921 0.041057 1 -0.528363 -0.842561 1.308425 1 -1.044879 -2.119710 -0.010485 8 1.374142 1.413463 0.128117 1 0.383318 1.436018 0.084927 1 1.688347 2.087966 -0.479194 8 1.534375 -1.298054 -0.030337 1 1.664081 -0.321597 -0.036516 1 1.890960 -1.617648 -0.864079	Frequencies (cm⁻¹): 34.7620, 92.3912, 128.1685, 229.3716, 245.7725, 251.4917, 264.5672, 292.4871, 435.7695, 509.2517, 527.9532, 708.4598, 757.5050, 820.9549, 928.0187, 1050.4965, 1233.0677, 1420.8772, 1585.0220, 1657.2334, 1667.4210, 3135.5422, 3229.4431, 3276.7943, 3402.4775, 3852.3723, 3864.9663

Compound: sCl 1 + (H ₂ O) ₂ TS1	Energy (kJ mol⁻¹): -342.168010859089
Reaction Coordinates: 6 1.192422 0.529632 0.316311 8 1.160920 -0.499868 -0.434508 8 0.318766 -1.553240 0.115411 1 0.932868 0.410061 1.360688 1 1.931337 1.267746 0.027052 8 -1.845919 -0.326514 0.101626 1 -0.993520 -0.964941 0.078703 1 -2.433201 -0.575827 -0.616514 8 -0.296944 1.611904 -0.010087 1 -1.070723 0.887825 -0.013058 1 -0.215877 1.939076 -0.914276	Frequencies (cm⁻¹): -331.9075, 71.9222, 194.6061, 351.7547, 414.3802, 422.4307, 475.2839, 500.9594, 542.9366, 635.6221, 785.6676, 811.7532, 997.0156, 1174.5653, 1210.7296, 1231.3393, 1340.8124, 1379.1202, 1510.8264, 1574.8904, 1701.3286, 1933.9475, 2324.7193, 3111.8165, 3225.2142, 3799.8888, 3856.3891

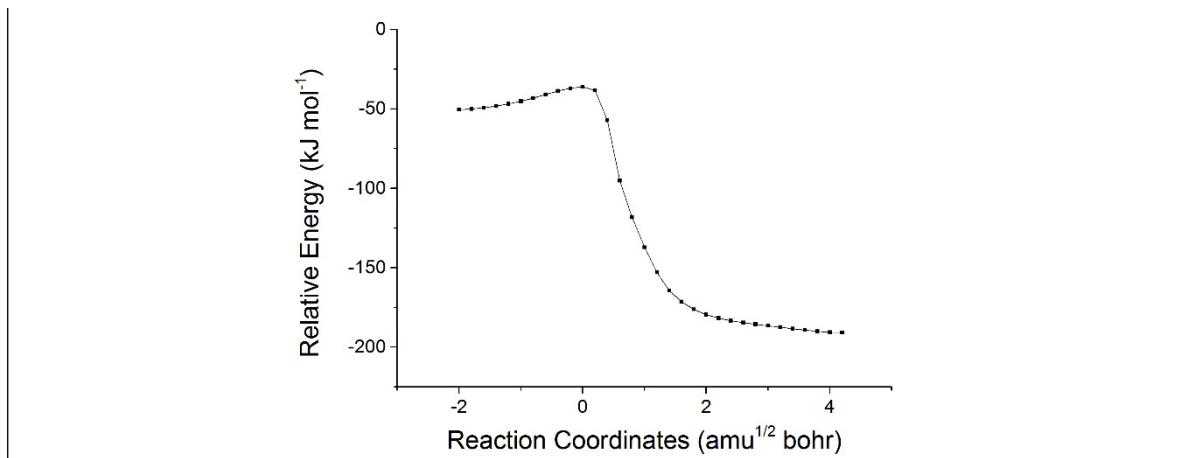


Compound: sCl 1 + (H ₂ O) ₂ Pr1	Energy (kJ mol⁻¹): -342.233729892064
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -1.310155 -0.387971 0.211196 8 -0.951903 0.746125 -0.518256 8 -0.007548 1.526793 0.257320 1 0.849294 1.100333 0.040465 1 -1.377092 -0.153293 1.270671 1 -2.278477 -0.684059 -0.197767 8 -0.375966 -1.458639 0.095441 1 -0.493571 -1.871459 -0.766643 8 2.199355 -0.167618 0.041687 1 2.736653 -0.384393 -0.725326 1 1.512619 -0.852597 0.101887	80.2143, 135.1051, 177.1882, 218.1612, , 245.4740, 311.2889, 418.3506, 430.3680, , 548.4011, 606.7308, 762.0375, 878.9051, 985.8147, 1044.7150, 1057.6103, 1281.4671, 1376.6100, 1409.4124, 1477.9090, 1487.4796, 1643.2095, 3043.3769, 3125.9909, 3485.3233, 3647.4380, 3800.8669, 3864.6927

Compound: sCl 1 + (H ₂ O) ₂ PRC2	Energy (kJ mol⁻¹): -342.172820831981
Reaction Coordinates:	Frequencies (cm⁻¹):
6 1.183018 1.034344 0.165699 8 1.329699 -0.039021 -0.445103 8 1.120434 -1.214004 0.287038 1 0.963648 1.023652 1.224482 1 1.336764 1.932257 -0.416177 8 -1.553924 -1.256559 -0.144544 1 -0.591428 -1.418379 0.018269 1 -2.032764 -1.857041 0.431979 8 -1.306680 1.470287 -0.041135 1 -1.570090 0.522258 -0.059751 1 -1.920468 1.905555 0.556951	44.7709, 92.2913, 121.6969, 190.2055, 221.7461, 229.2992, 252.4455, 269.8093, 454.4412, 508.4079, 558.2405, 679.0943, 700.7851, 820.4844, 849.8782, 1042.4600, 1234.4812, 1421.5825, 1590.4005, 1666.4577, 1681.3650, 3134.9513, 3276.2042, 3280.1314, 3422.0270, 3856.0097, 3867.7459

Compound: sCl 1 + (H ₂ O) ₂ TS2	Energy (kJ mol⁻¹): -342.166034138439
Reaction Coordinates:	Frequencies (cm⁻¹):
6 1.320571 0.408091 0.121837 8 0.985248 -0.661014 -0.484936 8 0.151751 -1.508354 0.360187 1 1.343904 0.381680 1.205155 1 2.027564 1.018985 -0.425759 8 -1.846984 -0.152850 -0.195992 1 -1.096225 -0.874018 0.055889 1 -2.570960 -0.233348 0.430105 8 -0.101441 1.596433 -0.018511 1 -0.949800 0.944191 -0.069664 1 -0.186498 2.120247 0.787262	-390.4521, 77.4610, 199.6630, 350.7358, 398.6994, 428.1677, 472.3849, 496.8026, 547.9891, 695.6734, 814.1253, 819.6207, 974.9830, 1065.5883, 1190.1939, 1237.5317, 1311.3755, 1373.6134, 1500.2665, 1605.5800, 1703.6058, 1871.0365, 2225.0288, 3105.5587, 3220.4414, 3799.7280, 3857.9859

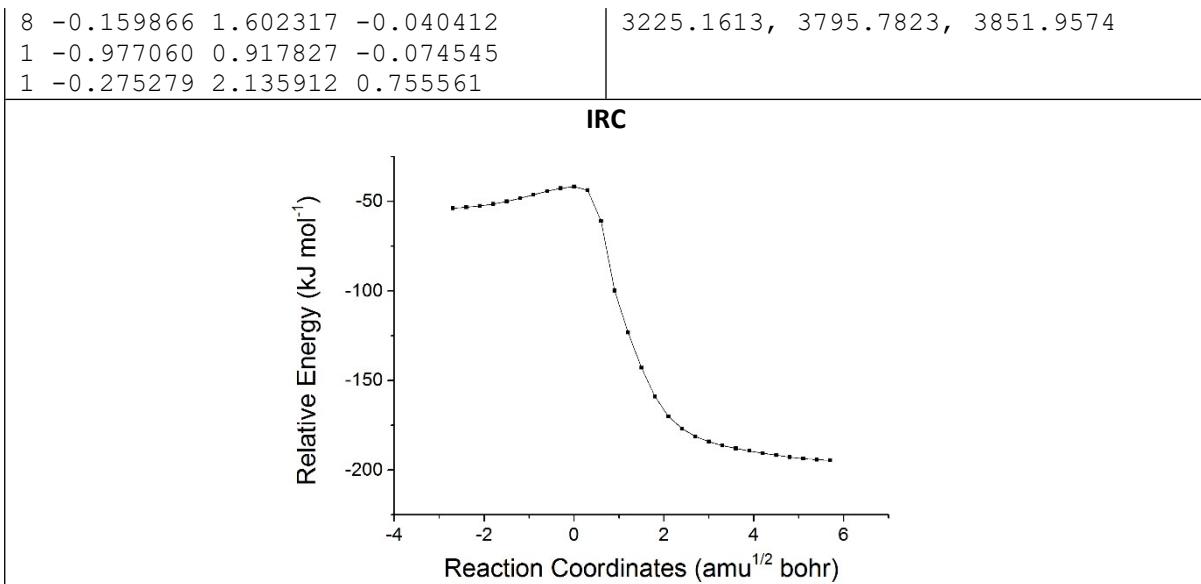
IRC



Compound: sCl 1 + (H ₂ O) ₂ Pr2	Energy (kJ mol⁻¹): -342.234765929815
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -1.379368 -0.331531 -0.143936 8 -0.679769 0.790306 -0.597330 8 -0.021638 1.388099 0.551088 1 0.893352 1.074027 0.403685 1 -2.043645 -0.057252 0.678174 1 -1.943257 -0.656645 -1.017525 8 -0.533159 -1.399710 0.242642 1 -0.346092 -1.309180 1.182284 8 2.153792 -0.217292 -0.222703 1 2.906674 -0.673391 0.162826 1 1.455373 -0.879593 -0.335403	109.1770, 119.5232, 156.5756, 175.0228, 220.1963, 279.2470, 378.5575, 410.0549, 490.8357, 612.7953, 691.0145, 881.9182, 1015.8576, 1037.4611, 1059.6915, 1284.5684, 1380.9111, 1410.5284, 1482.8420, 1487.0272, 1630.3323, 3034.4784, 3109.8974, 3543.4798, 3698.6881, 3813.3170, 3873.2790

Compound: sCl 1 + (H ₂ O) ₂ PRC3	Energy (kJ mol⁻¹): -342.173701402131
Reaction Coordinates:	Frequencies (cm⁻¹):
6 1.112307 1.068481 0.203569 8 1.344169 0.032511 -0.444929 8 1.171224 -1.184469 0.230492 1 0.857189 0.999313 1.252239 1 1.242551 1.998643 -0.331254 8 -1.514347 -1.304856 0.053226 1 -0.531564 -1.416493 0.119399 1 -1.799220 -1.866540 -0.672457 8 -1.363138 1.408770 -0.093024 1 -1.572574 0.445034 -0.085667 1 -1.973493 1.813512 0.530199	53.1327, 101.9900, 132.7817, 216.3090, 243.7112, 262.3092, 281.8423, 301.6424, 438.3076, 477.5225, 510.9797, , 696.7678, 816.7400, 821.8222, 950.4880, 1046.3538, 1235.4963, 1422.2397, 1589.6123, 1640.0256, 1668.4226, , 3134.9327, 3239.5935, 3277.1102, 3384.2768, 3849.1270, 3859.6874

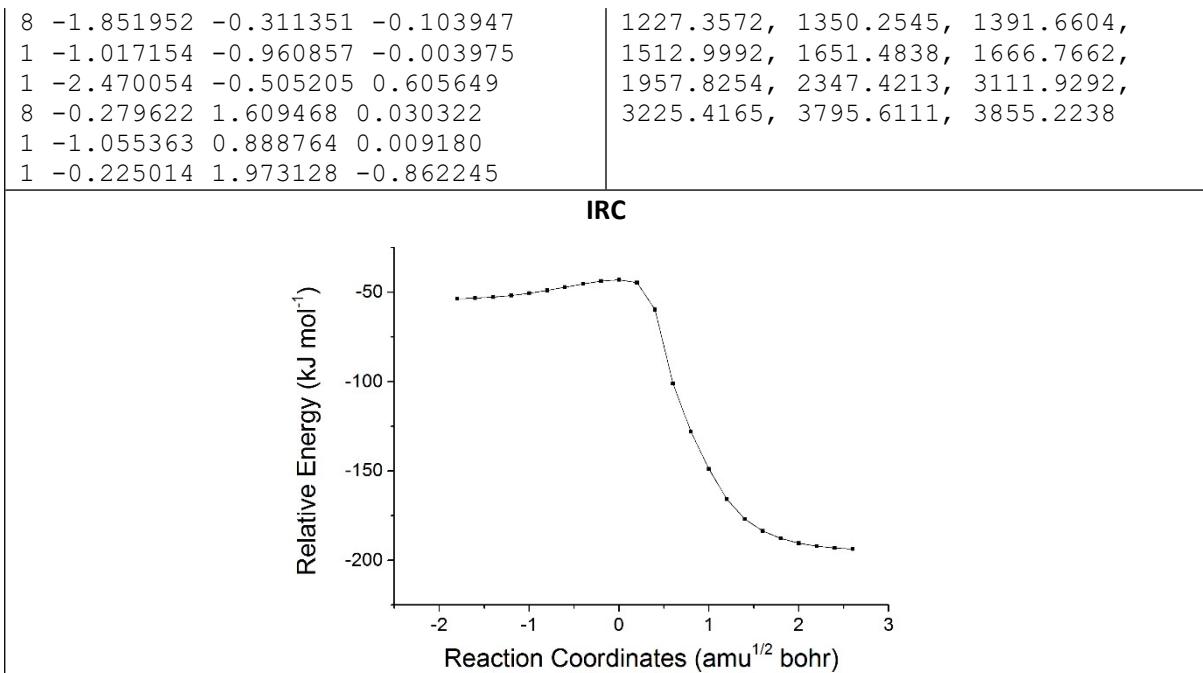
Compound: sCl 1 + (H ₂ O) ₂ TS3	Energy (kJ mol⁻¹): -342.168160410100
Reaction Coordinates:	Frequencies (cm⁻¹):
6 1.307862 0.444852 0.124825 8 1.010389 -0.627475 -0.492431 8 0.206552 -1.515403 0.339739 1 1.306652 0.417002 1.208064 1 1.999163 1.084525 -0.409325 8 -1.868295 -0.197275 -0.001163 1 -1.074048 -0.895604 0.133812 1 -2.336835 -0.426090 -0.808385	-357.3390, 83.2044, 206.4492, 353.9082, 392.4595, 425.2889, 475.5641, 507.1320, 550.1419, 625.3160, 800.6382, 818.4120, 964.7382, 1192.1193, 1215.3899, 1240.6843, 1362.3744, 1421.4184, 1509.2727, 1574.1521, 1652.4906, 1867.5595, 2284.4761, 3108.9557,



Compound: sCl 1 + (H ₂ O) ₂ Pr3	Energy (kJ mol⁻¹): -342.236055056007
Reaction Coordinates:	Frequencies (cm⁻¹):
6 1.379185 -0.321230 0.130494 8 0.687283 0.800439 0.597518 8 -0.001863 1.393427 -0.534429 1 -0.902001 1.029461 -0.396438 1 2.012555 -0.049944 -0.716190 1 1.974993 -0.634252 0.987427 8 0.525348 -1.397117 -0.215878 1 0.333934 -1.347236 -1.157888 8 -2.159058 -0.297409 0.068324 1 -2.681414 -0.178741 0.866979 1 -1.426865 -0.886630 0.308862	102.8232, 138.8594, 161.2587, 217.9648, 234.9476, 270.5088, 401.6741, 440.9790, 540.4841, 624.9479, 753.4956, 882.5366, 1015.2832, 1038.1739, 1058.5376, 1284.3506, 1377.0341, 1410.5592, 1482.3422, 1486.2977, 1632.4035, 3037.2242, 3108.2353, 3498.1506, 3681.9869, 3811.6239, 3862.6764

Compound: sCl 1 + (H ₂ O) ₂ PRC4	Energy (kJ mol⁻¹): -342.173268681123
Reaction Coordinates:	Frequencies (cm⁻¹):
6 1.035941 1.088804 0.263467 8 1.393543 0.111275 -0.419126 8 1.227174 -1.156101 0.155147 1 0.679362 0.944771 1.273830 1 1.175061 2.058545 -0.193582 8 -1.453888 -1.351848 -0.097799 1 -0.469779 -1.435667 -0.012495 1 -1.834915 -1.938947 0.560283 8 -1.452168 1.361236 0.051517 1 -1.617832 0.390095 0.007285 1 -1.864840 1.731884 -0.734030	50.1715, 95.4727, 125.9329, 230.6779, 244.3856, 258.3228, 278.3355, 306.0719, 455.3526, 497.6357, 520.4366, 705.8264, 768.2027, 821.0998, 923.8951, 1049.5097, 1230.3444, 1419.6329, 1586.9402, 1653.3753, 1680.2783, 3136.0331, 3239.6046, 3277.0653, 3385.2974, 3850.4666, 3861.9778

Compound: sCl 1 + (H ₂ O) ₂ TS4	Energy (kJ mol⁻¹): -342.168384007199
Reaction Coordinates:	Frequencies (cm⁻¹):
6 1.218398 0.517405 0.297770 8 1.143425 -0.519292 -0.438134 8 0.301105 -1.548507 0.154723 1 1.002418 0.412531 1.353308 1 1.951131 1.244668 -0.032246	-323.4815, 80.5480, 190.8851, 346.9591, 381.6466, 419.4027, 479.1782, 508.9306, 558.1013, 663.7923, 772.5704, 815.4304, 1002.2355, 1105.5810, 1208.6211,



Compound: sCl 1 + (H ₂ O) ₂ Pr4	Energy (kJ mol⁻¹): -342.234605099278
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -1.304886 -0.428786 0.217856 8 -1.000200 0.731096 -0.495025 8 -0.047719 1.514145 0.267961 1 0.809870 1.123777 -0.003827 1 -1.389838 -0.214913 1.280822 1 -2.256525 -0.761623 -0.202357 8 -0.320406 -1.449584 0.094944 1 -0.369645 -1.814832 -0.795559 8 2.196659 -0.145098 -0.146881 1 2.878349 -0.161457 0.530961 1 1.530433 -0.802702 0.114842	90.9279, 136.5893, 170.6335, 225.5069, 258.9680, 329.8411, 426.2758, 439.7446, 528.9552, 608.3053, 766.9180, 880.2249, 992.4407, 1045.1451, 1060.6803, 1281.0766, 1375.5973, 1412.5202, 1479.2496, 1496.9437, 1626.5333, 3042.1788, 3121.4472, 3499.4066, 3641.3869, 3792.6196, 3861.7379

Compound: sCl 2 + (H ₂ O) ₂ PRC1	Energy (kJ mol⁻¹): -678.974835113330
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -1.226886 0.069658 -0.173860 6 -0.327629 -0.153120 1.036917 8 0.548333 -1.036039 1.112313 8 0.877813 -1.749719 -0.030010 1 -0.543624 0.376619 1.953269 9 -2.014559 1.127672 0.099043 9 -0.591222 0.307883 -1.313380 9 -2.019818 -1.000758 -0.332959 8 2.879487 0.074156 -0.588399 1 2.302773 -0.717172 -0.511804 1 3.224136 0.073448 -1.485108 8 1.106164 1.829012 0.588222 1 1.819692 1.348332 0.111352 1 0.860129 2.567097 0.022605	37.4045, 57.1535, 91.4813, 107.9226, 134.2558, 193.0122, 218.4290, 225.2269, 231.9446, 276.5813, 295.5504, 320.5879, 430.4866, 480.3114, 500.0589, 526.4522, 539.5277, 591.2657, 652.9329, 753.0271, 816.7426, 846.1514, 855.7267, 885.9751, 1150.1725, 1170.1522, 1262.1541, 1371.4928, 1593.0624, 1662.6313, 1668.8617, 3222.1025, 3389.2633, 3463.1346, 3847.8267, 3865.5964

Compound: sCl 2 + (H ₂ O) ₂ TS1	Energy (kJ mol⁻¹): -678.970889225415
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -1.173357 -0.023374 0.164242 6 -0.151008 0.080429 -0.982054 8 0.689286 1.025230 -1.064898 8 1.119634 1.538355 0.223773 1 -0.541650 -0.236571 -1.942191 9 -2.015736 -1.041562 -0.149138 9 -0.676530 -0.288170 1.368967 9 -1.889874 1.098391 0.215695 8 2.710861 -0.401175 0.533202 1 2.185805 0.495286 0.505416 1 2.922854 -0.588301 1.451650 8 0.810773 -1.490775 -0.664878 1 1.644758 -1.159895 -0.104261 1 0.329264 -2.143876 -0.141050	-290.0059, 46.3837, 64.5511, 168.5922, 182.7072, 230.3468, 301.4527, 314.3332, 372.6685, 395.7978, 408.1420, 473.8506, 507.6275, 518.0961, 554.1721, 598.6566, 675.9497, 744.7321, 805.4973, 824.7409, 879.3618, 1008.2971, 1075.2479, 1122.4730, 1206.1978, 1262.7319, 1277.8725, 1351.6705, 1530.8641, 1673.7946, 1713.3045, 2078.5849, 2543.4807, 3168.9662, 3787.2140, 3854.6140
IRC	
<p>The plot shows the reaction coordinate (IRC) for the reaction of sCl 2 + (H₂O)₂ at the TS1 level. The y-axis represents Relative Energy in kJ mol⁻¹, ranging from 0 down to -200. The x-axis represents the Reaction Coordinates in amu^{1/2} bohr, ranging from -2 to 5. The energy curve starts at approximately -52 kJ mol⁻¹ at x = -2, remains relatively flat until x = 0, then drops sharply to about -195 kJ mol⁻¹ by x = 2, and levels off at this minimum value.</p>	

Compound: sCl 2 + (H ₂ O) ₂ Pr1	Energy (kJ mol⁻¹): -679.043932127222
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -1.319171 -0.025832 0.129707 6 -0.099793 -0.222029 -0.803922 8 0.688475 0.894559 -0.978112 8 1.172711 1.381058 0.299920 1 2.046883 0.932451 0.348848 1 -0.505125 -0.398751 -1.805744 8 0.677895 -1.319844 -0.364164 1 0.115687 -2.093579 -0.242853 9 -2.120810 -1.120416 -0.019587 9 -1.024117 0.063500 1.425112 9 -2.028327 1.047038 -0.229328 8 3.336058 -0.357324 0.286854 1 3.755770 -0.629253 1.108307 1 2.658744 -1.022384 0.094960	36.4715, 56.0671, 125.3481, 135.0123, 185.8482, 211.5810, 235.6883, 243.2222, 254.3634, 298.8896, 330.1348, 376.5204, 437.9258, 469.2699, 520.0383, 582.8195, 611.9422, 744.2850, 754.9017, 852.2004, 908.7220, 1033.2727, 1092.9097, 1112.6719, 1195.4946, 1247.0912, 1300.8185, 1359.0143, 1425.1312, 1528.9667, 1632.6548, 3033.4111, 3443.2887, 3715.3179, 3798.3722, 3865.9905

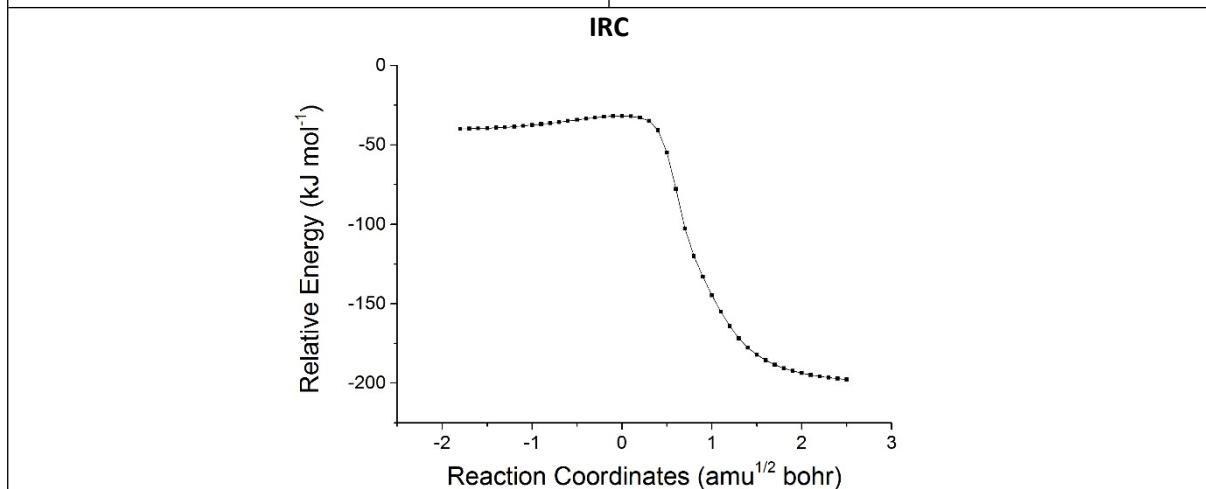
Compound: sCl 2 + (H ₂ O) ₂ PRC2	Energy (kJ mol⁻¹):	-678.975412531634
Reaction Coordinates:	Frequencies (cm⁻¹):	
6 -1.214882 0.071350 -0.175366 6 -0.325793 -0.161017 1.042058 8 0.545657 -1.048070 1.121063 8 0.887838 -1.752056 -0.026755 1 -0.554234 0.358827 1.961138 9 -2.001088 1.130774 0.097863 9 -0.570514 0.311513 -1.307792 9 -2.010442 -0.996175 -0.342068 8 2.774490 0.122967 -0.749245 1 2.224618 -0.679219 -0.605514 1 3.666755 -0.111412 -0.480022 8 1.103865 1.811852 0.619107 1 1.806175 1.330267 0.123785 1 0.844351 2.546984 0.055074	41.8023, 58.0072, 98.6309, 113.1135, 134.6924, 193.3712, 224.3711, 238.4057, 254.7038, 282.1835, 319.6521, 335.3226, 404.7950, 467.5258, 481.0277, 507.0066, 536.9264, 591.7953, 747.8361, 759.8966, 829.6064, 858.2697, 886.5529, 908.8787, 1149.1735, 1170.7208, 1265.3439, 1371.2946, 1593.6215, 1637.7621, 1663.6177, 3221.0408, 3359.0331, 3445.1478, 3842.8330, 3860.0165	

Compound: sCl 2 + (H ₂ O) ₂ TS2	Energy (kJ mol⁻¹): -678.972203381997
Reaction Coordinates:	Frequencies (cm⁻¹):
6 1.178354 0.022029 0.158851 6 0.147978 -0.071725 -0.980800 8 -0.697411 -1.010264 -1.065886 8 -1.114160 -1.543374 0.218888 1 0.525561 0.262066 -1.940232 9 2.018163 1.042347 -0.152675 9 0.690659 0.271855 1.367738 9 1.895185 -1.101577 0.189445 8 -2.649322 0.416120 0.673704 1 -2.152135 -0.487113 0.573358 1 -3.524230 0.315367 0.288595 8 -0.818695 1.511442 -0.634519 1 -1.653999 1.160100 -0.097130 1 -0.352551 2.142745 -0.070962	-269.7946, 50.1764, 63.9495, 173.4678, 183.5984, 229.1705, 302.2534, 313.6691, 372.7622, 379.2671, 408.0386, 477.1009, 511.5748, 517.6791, 549.2021, 595.0579, 610.6639, 742.1387, 800.8068, 814.9207, 879.3384, 1064.6348, 1124.1278, 1139.8038, 1205.8730, 1270.9158, 1347.7219, 1368.6629, 1535.3773, 1606.6560, 1661.9980, 2136.0068, 2593.0390, 3171.4988, 3783.4700, 3851.0261
IRC	
<p>The plot shows the potential energy surface along the reaction coordinate. The y-axis represents the relative energy in kJ mol⁻¹, ranging from 0 down to -200. The x-axis represents the reaction coordinate in amu^{1/2} bohr, ranging from -2 to 4. The curve starts at approximately -52 kJ mol⁻¹ at x = -1.5, remains relatively flat until x = 0, then drops sharply to about -195 kJ mol⁻¹ at x = 2.5, and levels off at this value for larger reaction coordinates.</p>	

Compound: sCl 2 + (H ₂ O) ₂ Pr2	Energy (kJ mol⁻¹): -679.044522555901
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -1.322906 -0.020520 0.112406 6 -0.084043 -0.241045 -0.789019 8 0.716228 0.869099 -0.963624 8 1.167558 1.378038 0.317840 1 2.031053 0.914065 0.408348 1 -0.467995 -0.429444 -1.797286 8 0.679699 -1.335631 -0.321104 1 0.102371 -2.044888 -0.015769 9 -2.130590 -1.110357 -0.037283 9 -1.055727 0.085110 1.412809 9 -2.012187 1.052759 -0.279522 8 3.301497 -0.378920 0.431258 1 3.995376 -0.356724 -0.234385 1 2.647576 -1.021918 0.119768	38.5690, 54.0929, 129.2809, 136.0169, 157.2516, 214.1037, 237.6501, 255.5837, 260.4528, 282.1900, 301.5859, 369.7651, 436.3490, 518.4205, 527.5382, 584.3273, 621.5032, 752.7175, 793.5276, 854.1331, 908.7737, 1041.4784, 1090.5830, 1113.8186, 1199.1434, 1244.6945, 1302.0778, 1353.1822, 1431.0382, 1512.5730, 1626.9675, 3031.5325, 3431.1048, 3713.0756, 3796.4734, 3864.3227

Compound: sCl 2 + (H ₂ O) ₂ PRC3	Energy (kJ mol⁻¹): -678.973863351660
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -1.190559 0.094535 -0.196108 6 -0.369816 -0.172653 1.062611 8 0.480011 -1.077431 1.176344 8 0.862903 -1.784611 0.048000 1 -0.634506 0.341912 1.974995 9 -1.984786 1.148371 0.055349 9 -0.480784 0.343171 -1.283526 9 -1.981592 -0.969140 -0.421792 8 2.702079 0.091542 -0.818995 1 2.159514 -0.701130 -0.614895 1 3.616997 -0.164226 -0.676270 8 1.073424 1.846982 0.551227 1 1.771615 1.367837 0.052656 1 1.525750 2.310843 1.261611	43.5999, 60.5434, 98.9220, 102.2304, 122.5014, 186.2915, 219.4830, 223.4351, 231.6112, 247.9051, 277.4087, 319.6655, 395.1912, 476.6845, 496.7676, 505.7365, 536.0064, 588.8097, 688.7703, 754.5842, 820.2495, 852.0429, 862.4511, 887.8182, 1150.9682, 1170.0254, 1268.7329, 1368.4085, 1591.2971, 1644.9660, 1665.7561, 3219.0326, 3402.1416, 3471.6488, 3853.1968, 3865.0665

Compound: sCl 2 + (H ₂ O) ₂ TS3	Energy (kJ mol⁻¹): -678.970261942381
Reaction Coordinates:	Frequencies (cm⁻¹):
6 1.151495 -0.045174 -0.184872 6 0.170938 0.072243 0.998151 8 -0.653792 1.026446 1.122183 8 -1.115610 1.577872 -0.136639 1 0.591290 -0.262956 1.940555 9 2.030092 -1.024962 0.114831 9 0.607304 -0.324768 -1.358357 9 1.837879 1.100638 -0.273055 8 -2.529554 -0.422462 -0.791602 1 -2.072740 0.486607 -0.603370 1 -3.451191 -0.349556 -0.529791 8 -0.772075 -1.527413 0.614835 1 -1.582115 -1.188548 0.037894 1 -1.129066 -1.841687 1.454050	-255.0687, 55.2659, 77.0544, 168.0979, 178.5946, 222.8289, 303.1951, 315.0054, 370.2693, 390.8977, 403.1160, 468.3366, 482.4316, 519.7345, 552.4007, 603.8288, 621.6264, 742.8942, 784.0733, 809.4791, 898.8488, 1058.3895, 1108.0793, 1143.1440, 1188.6024, 1271.7654, 1305.2514, 1345.2191, 1534.5795, 1613.5212, 1707.5809, 2181.4832, 2610.1198, 3155.3420, 3805.6386, 3855.2006



Compound: sCl 2 + (H ₂ O) ₂ Pr3	Energy (kJ mol⁻¹): -679.044046393108
Reaction Coordinates:	Frequencies (cm⁻¹):
6 1.235036 -0.074225 -0.201158 6 0.180498 -0.131186 0.928825 8 -0.576023 1.025870 1.082329 8 -1.115760 1.449222 -0.193629 1 -1.945106 0.927595 -0.260668 1 0.729962 -0.193591 1.871195 8 -0.585846 -1.296230 0.707444 1 -0.893072 -1.622913 1.558947 9 2.181647 -1.006601 0.055252 9 0.739031 -0.332562 -1.412406 9 1.839106 1.118160 -0.225203 8 -3.084291 -0.441560 -0.675776 1 -3.903200 -0.686352 -0.235492 1 -2.424486 -1.101660 -0.421706	38.0835, 64.6328, 96.1394, 135.5457, 197.8564, 201.4051, 234.9116, 243.3833, 261.0024, 303.1512, 331.9013, 371.1792, 436.1274, 449.1955, 525.3000, 582.6400, 609.0144, 750.8165, 767.7665, 844.1162, 907.7243, 1051.8707, 1076.2109, 1126.7896, 1177.7901, 1246.2906, 1291.8476, 1373.1354, 1437.7390, 1483.7264, 1632.1168, 3059.6863, 3479.9112, 3732.3887, 3815.2581, 3872.7493

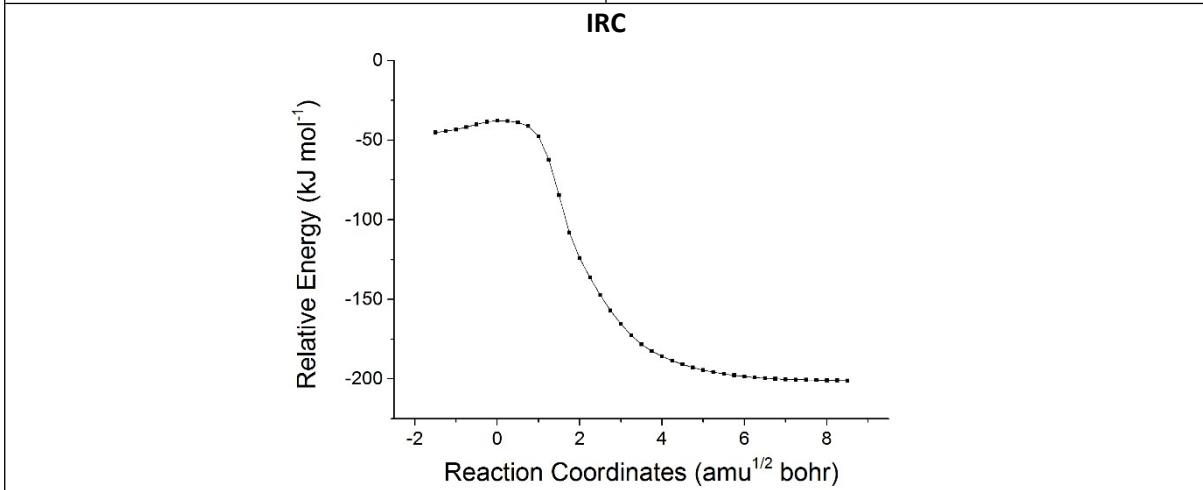
Compound: sCl 2 + (H ₂ O) ₂ PRC4	Energy (kJ mol⁻¹): -678.974423926339
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -1.195272 0.096888 -0.192917 6 -0.363456 -0.174980 1.057263 8 0.485322 -1.082150 1.161247 8 0.865860 -1.777624 0.024545 1 -0.627595 0.330073 1.975107 9 -1.987364 1.149471 0.068122 9 -0.497726 0.350479 -1.289404 9 -1.987331 -0.966137 -0.416690 8 2.796504 0.076841 -0.680249 1 2.233024 -0.714988 -0.535574 1 2.968365 0.113539 -1.625230 8 1.069922 1.825415 0.546226 1 1.765743 1.329712 0.056432 1 1.523766 2.266039 1.270788	44.5943, 61.0444, 104.8336, 125.3744, 189.9980, 221.1578, 232.1002, 241.0143, 280.2124, 303.1346, 320.0261, 427.4818, 469.8666, 496.1996, 500.6470, 536.0040, 590.0928, 694.7367, 754.2804, 834.5125, 857.8793, 885.0821, 896.8702, 1151.3690, 1169.9219, 1265.2790, 1368.1246, 1590.6157, 1649.7949, 1669.9483, 3218.8672, 3367.2346, 3453.9721, 3847.7674, 3858.5145

Compound: sCl 2 + (H ₂ O) ₂ TS4	Energy (kJ mol⁻¹): -678.971433697789
Reaction Coordinates:	Frequencies (cm⁻¹):
6 1.148516 -0.052598 -0.185936 6 0.180719 0.099865 1.002687 8 -0.638403 1.059164 1.109814 8 -1.110604 1.582452 -0.154346 1 0.603794 -0.220212 1.948898 9 2.026834 -1.026730 0.127879 9 0.590992 -0.363862 -1.349187 9 1.836346 1.087250 -0.317239 8 -2.611693 -0.414783 -0.662662 1 -2.124068 0.486864 -0.549865 1 -2.687895 -0.585007 -1.605794 8 -0.778960 -1.510779 0.657020 1 -1.572380 -1.177308 0.057052 1 -1.165135 -1.786280 1.497513	-242.4381, 56.9560, 77.5768, 163.9943, 178.3123, 224.0854, 302.7484, 314.3336, 362.0054, 378.2726, 389.8786, 472.0494, 518.6113, 520.7586, 551.5244, 606.7181, 632.0949, 734.4464, 779.3153, 813.3456, 899.3494, 1014.0479, 1074.2958, 1146.1441, 1188.0078, 1269.7607, 1330.6011, 1355.1327, 1539.8493, 1652.5416, 1695.9196, 2208.5205, 2657.1975, 3158.3164, 3799.5431, 3850.0937
IRC	
<p>Relative Energy (kJ mol⁻¹)</p> <p>Reaction Coordinates (amu^{1/2} bohr)</p>	

Compound: sCl 2 + (H ₂ O) ₂ Pr4	Energy (kJ mol⁻¹): -679.045781275488
Reaction Coordinates:	Frequencies (cm⁻¹):
6 1.229414 -0.104076 -0.197877 6 0.186925 -0.032536 0.942221 8 -0.541302 1.152571 0.996142 8 -1.125871 1.444229 -0.296751 1 -1.949458 0.909925 -0.286233 1 0.744590 -0.018329 1.882382 8 -0.605649 -1.191117 0.830119 1 -1.027534 -1.354289 1.680771 9 2.160525 -1.027778 0.129379 9 0.710813 -0.462019 -1.376728 9 1.854483 1.068510 -0.341170 8 -3.100430 -0.541605 -0.489931 1 -3.472311 -0.705733 -1.361849 1 -2.339708 -1.132937 -0.401095	42.2630, 62.8676, 110.5466, 138.7273, 184.2194, 196.6160, 232.1943, 261.7272, 299.5725, 324.8941, 355.9719, 403.2916, 436.8545, 455.0373, 525.8731, 582.1368, 611.4955, 738.4935, 765.3435, 848.2237, 906.3519, 1060.1520, 1075.5174, 1128.1962, 1175.7514, 1244.6938, 1288.3378, 1370.2384, 1443.8454, 1497.8364, 1615.5190, 3054.8039, 3485.5656, 3726.2780, 3802.1332, 3865.2728

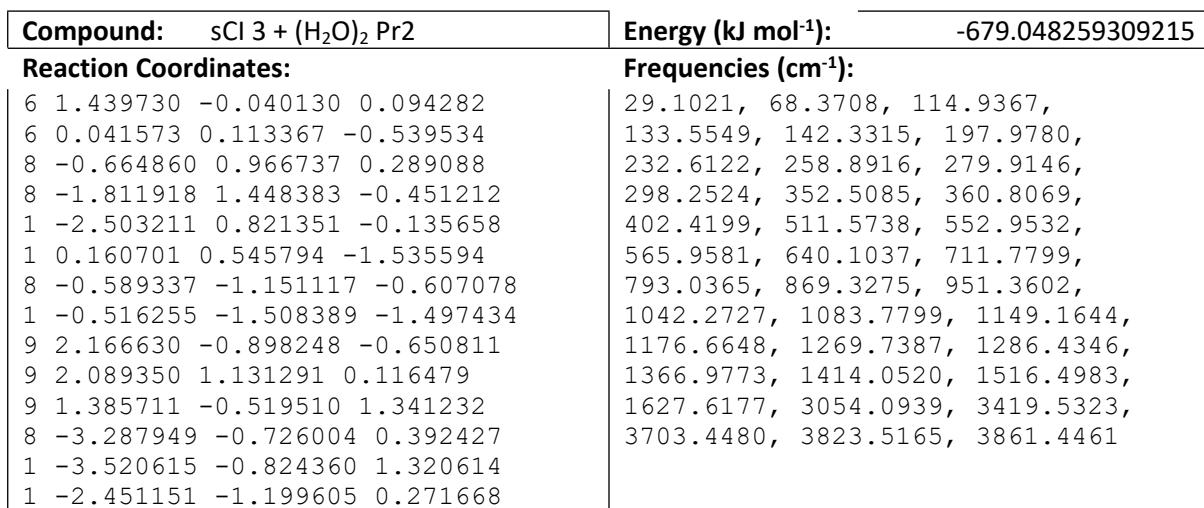
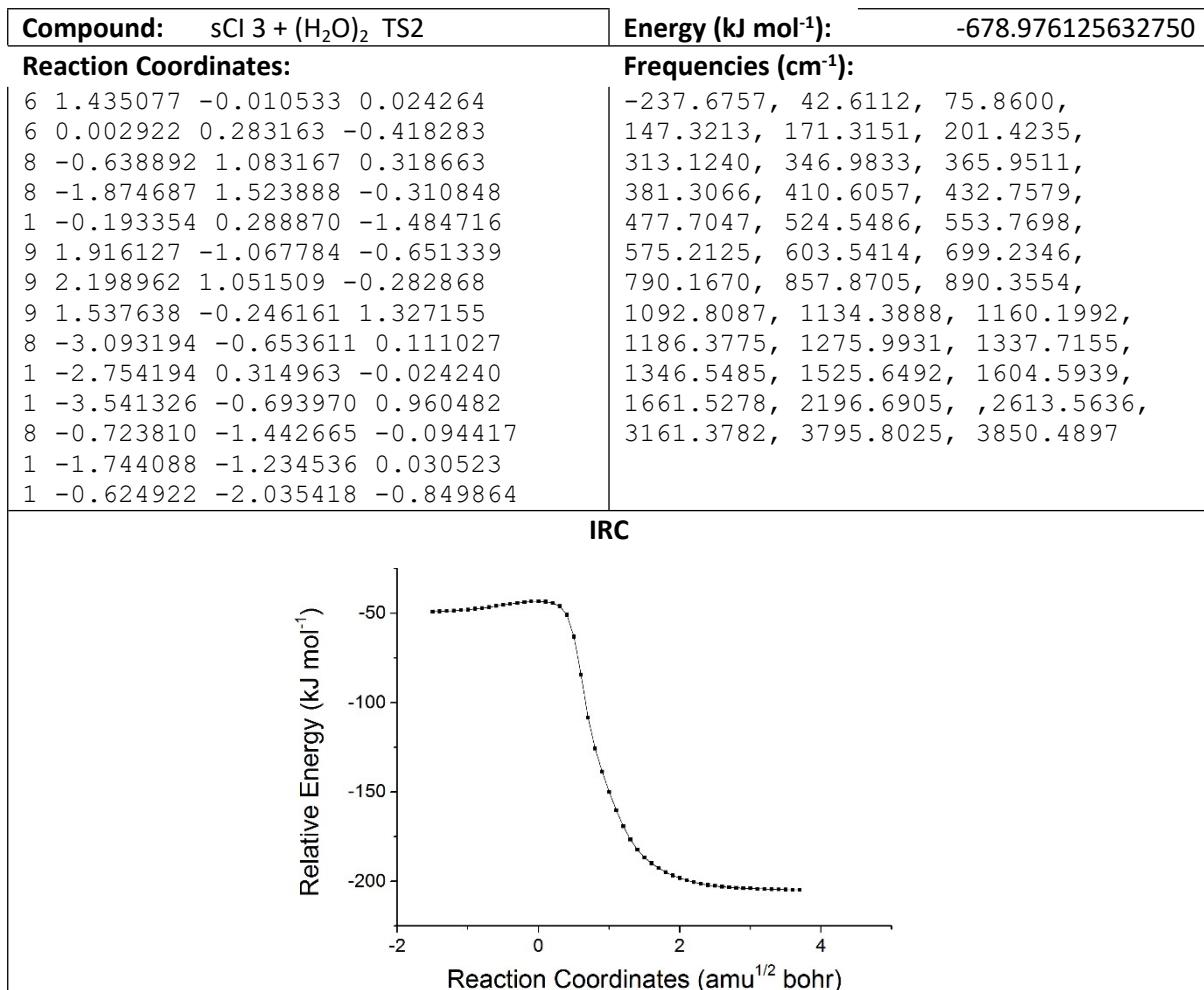
Compound: sCl 3 + (H ₂ O) ₂ PRC1	Energy (kJ mol⁻¹): -678.976463310368
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -1.499637 0.034189 0.029921 9 -2.421756 -0.917529 -0.234744 9 -1.481601 0.246835 1.339166 9 -1.879268 1.149648 -0.601056 6 -0.161123 -0.437563 -0.509120 8 0.528005 -1.160758 0.229694 8 1.703726 -1.674615 -0.315970 1 0.107263 -0.285593 -1.545557 8 3.323108 0.443097 0.301092 1 2.903426 -0.428271 0.114598 1 4.143665 0.463451 -0.197932 8 0.947209 1.699816 -0.129811 1 1.885003 1.425164 -0.007395 1 0.952445 2.474579 -0.698853	30.2311, 48.3526, 78.3380, 121.2136, 129.1577, 183.8853, 210.7690, 233.2283, 237.6079, 250.3556, 291.0299, 378.0381, 406.7864, 419.5053, 447.6724, 548.6580, 556.0472, 567.2963, 661.5093, 697.1129, 828.2620, 881.3964, 901.0656, 932.5150, 1129.1888, 1190.4575, 1271.2460, 1360.2601, , 1593.5489, 1654.6893, 1672.7892, 3208.8953, 3342.6622, 3427.3235, 3855.6257, 3865.3133

Compound: sCl 3 + (H ₂ O) ₂ TS1	Energy (kJ mol⁻¹): -678.974181399714
Reaction Coordinates:	Frequencies (cm⁻¹):
6 1.433893 -0.010821 0.025018 6 0.000721 0.275629 -0.421110 8 -0.642633 1.080587 0.312812 8 -1.879165 1.511302 -0.325547 1 -0.181859 0.294659 -1.490112 9 1.919606 -1.067259 -0.651454 9 2.194354 1.053299 -0.281898 9 1.536228 -0.247488 1.326860 8 -3.054122 -0.636397 0.296387 1 -2.742309 0.326894 0.061902 1 -3.810052 -0.849239 -0.256998 8 -0.720537 -1.425674 -0.112831 1 -1.747372 -1.231075 0.021585 1 -0.606141 -2.015594 -0.867951	-263.1970, 41.2328, 75.5833, 147.1956, 171.5294, 195.8320, 313.2451, 347.0769, 363.3666, 390.1600, 411.0510, 428.8565, 468.8022, 522.5049, 557.4381, 588.2506, 669.5502, 698.9099, 812.9743, 859.6684, 893.5243, 996.8146, 1104.8172, 1159.1604, 1187.8663, 1254.9663, 1279.0990, 1335.4104, 1518.4628, 1670.5092, 1710.7505, 2130.1417, 2552.0395, 3156.0023, 3798.9059, 3858.0062



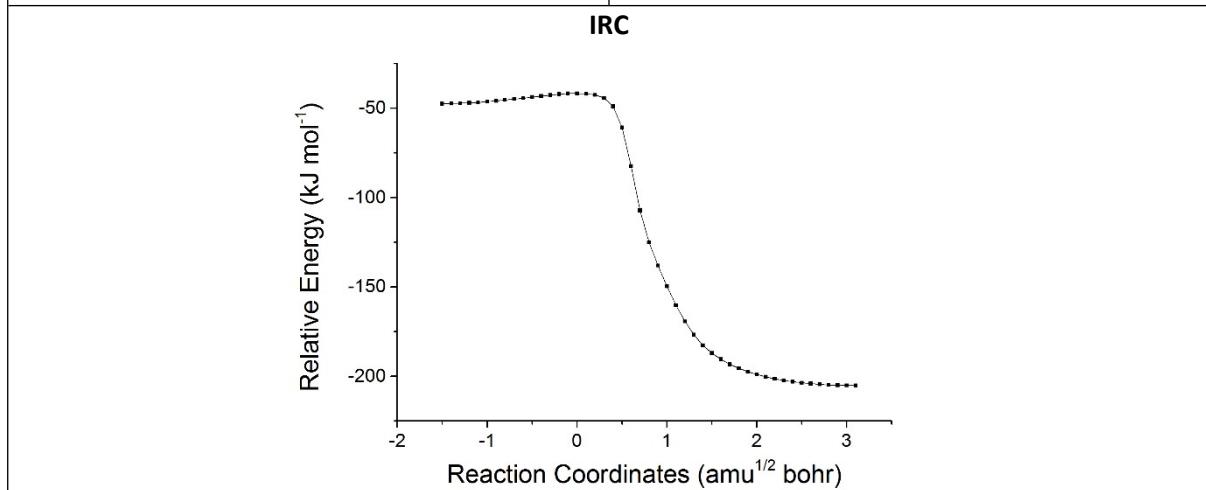
Compound: sCl 3 + (H ₂ O) ₂ Pr1	Energy (kJ mol⁻¹): -679.046695526630
Reaction Coordinates:	Frequencies (cm⁻¹):
6 1.449107 -0.036898 0.087914 6 0.037661 0.090957 -0.522040 8 -0.658996 0.963041 0.289197 8 -1.798886 1.451237 -0.458160 1 -2.502768 0.852493 -0.119323 1 0.137230 0.491542 -1.534270 8 -0.601361 -1.173366 -0.536544 1 -0.323123 -1.670114 -1.313281 9 2.147732 -0.948405 -0.627237 9 2.110176 1.125660 0.018976 9 1.428296 -0.441198 1.359734 8 -3.294813 -0.659336 0.509863 1 -4.052965 -1.078680 0.093172 1 -2.522371 -1.196702 0.280354	39.4378, 67.9434, 112.9031, 127.4279, 182.7777, 199.1615, 225.0892, 240.7713, 252.2604, 330.8000, 355.9574, 366.0785, 403.1514, 467.2147, 520.1325, 565.5773, 620.0285, 711.2871, 732.5789, 872.7445, 951.3378, 1036.5657, 1089.3071, 1145.6563, 1177.6616, 1276.8225, 1278.9342, 1379.7424, 1401.6927, 1531.8141, 1634.1229, 3045.2327, 3438.3881, 3713.2347, 3811.0091, 3869.9623

Compound: sCl 3 + (H ₂ O) ₂ PRC2	Energy (kJ mol⁻¹): -678.977263064497
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -1.489581 0.032561 0.017263 6 -0.130363 -0.424195 -0.483405 8 0.528723 -1.168690 0.262489 8 1.721301 -1.679972 -0.259762 1 0.165924 -0.264696 -1.511268 9 -1.863219 1.146325 -0.620301 9 -2.392537 -0.928016 -0.277285 9 -1.512613 0.240448 1.327654 8 3.328823 0.482446 0.065681 1 2.893099 -0.399289 -0.020606 1 3.863618 0.443308 0.863291 8 0.915829 1.691448 -0.041072 1 1.859539 1.407848 0.030886 1 0.895401 2.431960 -0.654738	30.6497, 52.9885, 88.3887, 126.1346, 131.7672, 188.4279, ,227.6577, 257.3088, 270.5375, 298.3777, 326.5543, 377.5462, 406.6573, 419.9501, 426.6031, 504.1159, 551.6708, 565.5651, 696.6782, 801.3020, 868.7779, 896.7671, 932.7880, 956.1836, 1130.7885, 1190.6128, 1270.0021, 1360.0737, 1591.7576, 1632.3812, 1658.9014, 3206.3576, 3280.1552, 3382.2637, 3846.1059, 3857.7617



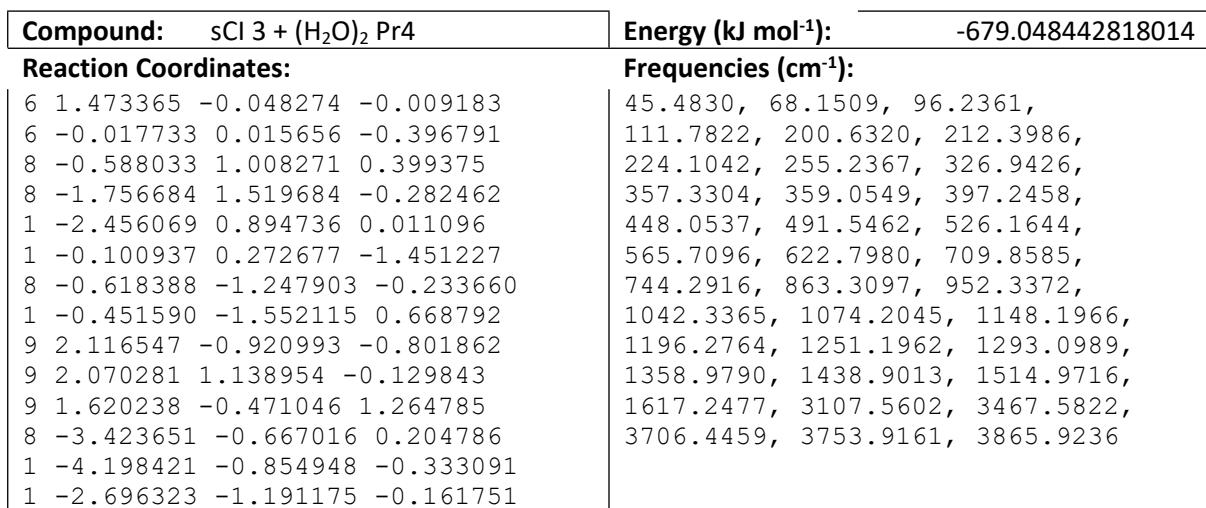
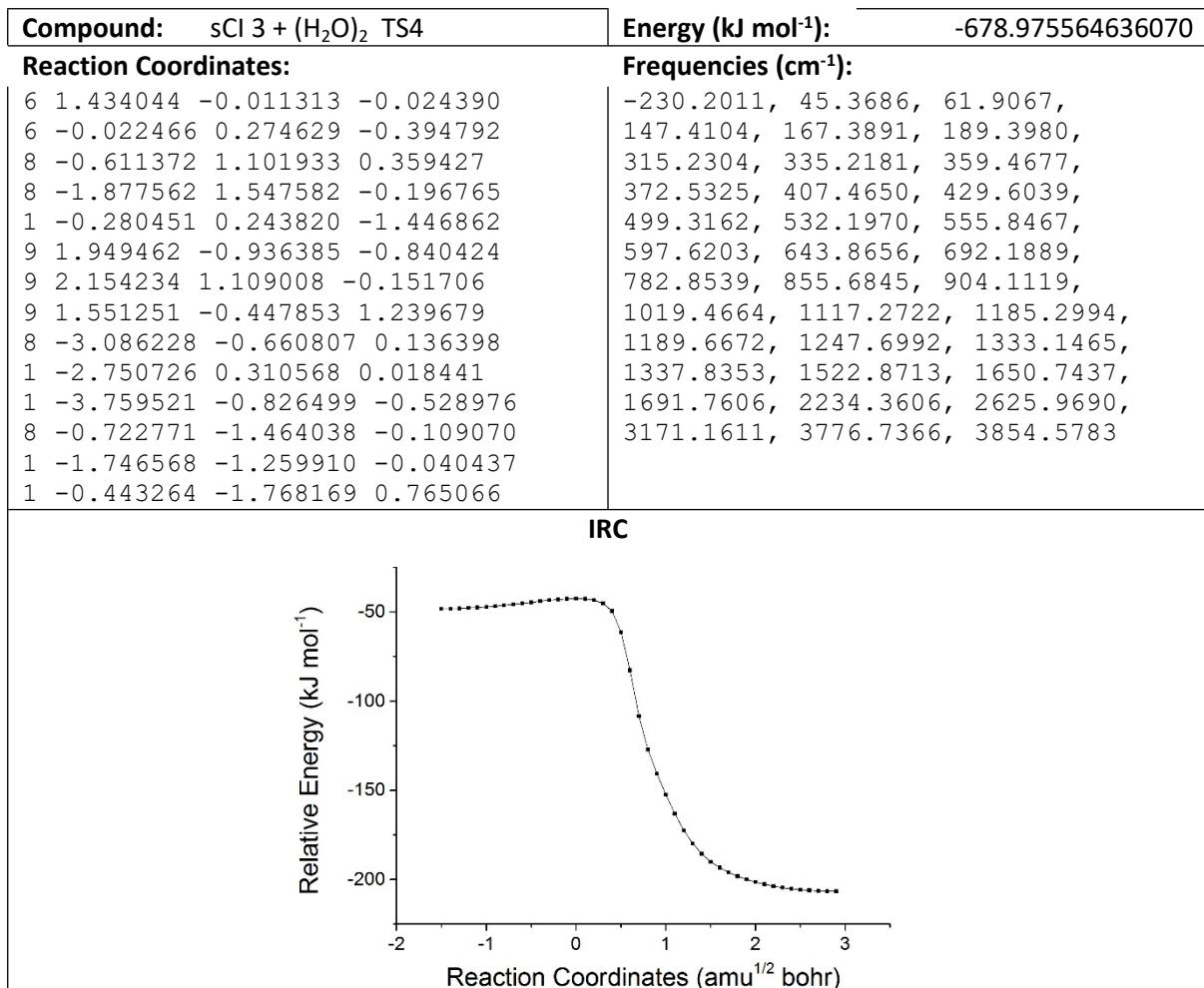
Compound: sCl 3 + (H ₂ O) ₂ PRC3	Energy (kJ mol⁻¹): -678.976758543801
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -1.485381 0.029360 0.008310 6 -0.118211 -0.417021 -0.481879 8 0.508388 -1.199324 0.254757 8 1.716295 -1.697852 -0.234009 1 0.213743 -0.202963 -1.488591 9 -1.903593 1.082738 -0.695441 9 -2.367865 -0.971427 -0.177248 9 -1.480214 0.342995 1.305648 8 3.317663 0.478742 0.072711 1 2.881421 -0.402415 -0.003996 1 3.910307 0.425006 0.826937 8 0.936741 1.748319 -0.204451 1 1.862031 1.456501 -0.027708 1 0.686411 2.322011 0.526067	22.7981, 33.1705, 84.6555, 116.8194, 135.7398, 183.9906, 226.2528, 250.2507, 260.1566, 277.5607, 297.0635, 378.3309, 404.9927, 413.8918, 425.4943, 540.8087, 552.1992, 568.2039, 696.2535, 731.1253, 865.8081, 891.1302, 921.7357, 942.0771, 1143.3441, 1195.7403, 1250.2764, 1360.4244, 1586.1495, 1639.6708, 1665.4309, 3208.9233, 3314.2435, 3405.6402, 3846.2551, 3863.2852

Compound: sCl 3 + (H ₂ O) ₂ TS3	Energy (kJ mol⁻¹): -678.975252603565
Reaction Coordinates:	Frequencies (cm⁻¹):
6 1.431618 -0.011054 -0.034022 6 -0.034263 0.263996 -0.375191 8 -0.606445 1.103863 0.379333 8 -1.881019 1.551034 -0.156779 1 -0.317238 0.214681 -1.420450 9 1.934015 -0.944984 -0.846552 9 2.143596 1.110552 -0.191312 9 1.577245 -0.428714 1.234683 8 -3.089048 -0.671805 -0.096550 1 -2.739105 0.303517 -0.088996 1 -3.714809 -0.772452 0.625598 8 -0.713585 -1.470124 -0.036222 1 -1.741506 -1.277314 -0.007735 1 -0.444395 -1.721504 0.857244	-234.0953, 40.5246, 63.1378, 148.1200, 169.9852, 188.6379, 317.5559, 337.1536, 365.4847, 395.3353, 407.4014, 434.1385, 485.7447, 516.9718, 555.8605, 592.6492, 628.4843, 696.8288, 787.2052, 853.9549, 900.9870, 1084.2440, 1123.9700, 1185.4779, 1191.7999, 1247.1847, 1307.6331, 1338.5639, 1518.0518, 1617.4855, 1696.5678, 2227.8831, 2604.6378, 3168.8170, 3781.6508, 3855.6553



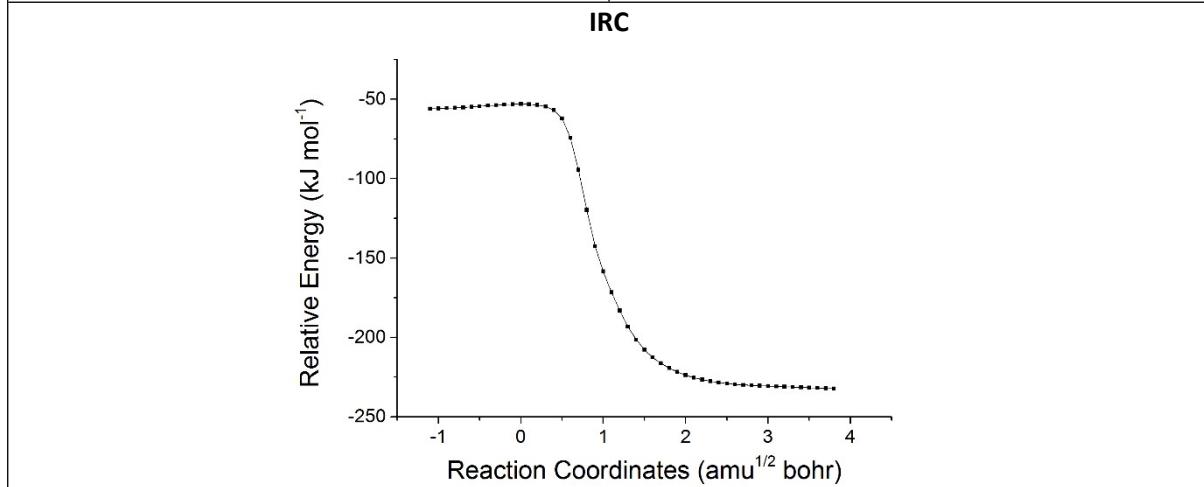
Compound: sCl 3 + (H ₂ O) ₂ Pr3	Energy (kJ mol⁻¹): -679.047730845463
Reaction Coordinates: 6 1.471334 -0.037790 -0.013291 6 -0.025816 -0.002631 -0.378566 8 -0.609598 0.972687 0.429755 8 -1.751106 1.522386 -0.268795 1 -2.465622 0.891842 -0.025845 1 -0.131809 0.259331 -1.429368 8 -0.595147 -1.281875 -0.211757 1 -0.359250 -1.609005 0.666398 9 2.123837 -0.889429 -0.820830 9 2.040916 1.162355 -0.128133 9 1.642956 -0.470542 1.256125 8 -3.462714 -0.631195 0.014372 1 -3.945142 -0.971691 0.773244 1 -2.692139 -1.205440 -0.106349	Frequencies (cm⁻¹): 42.3450, 68.1372, 93.4777, 108.7143, 196.3265, 225.6205, 232.9114, 251.4983, 300.4113, 355.5126, 363.0538, 395.7893, 417.7649, 498.8204, 529.5849, 566.0577, 617.7195, 712.3798, 766.2066, 860.6738, 952.0996, 1039.1784, 1072.2515, 1145.9665, 1198.6877, 1249.1969, 1289.4075, 1358.5433, 1439.6431, 1502.5135, 1634.4066, 3112.1802, 3454.4926, 3707.6218, 3759.0631, 3867.6327

Compound: sCl 3 + (H ₂ O) ₂ PRC4	Energy (kJ mol⁻¹): -678.977086514110
Reaction Coordinates: 6 -1.488498 0.031281 0.024393 6 -0.139236 -0.422659 -0.505341 8 0.517339 -1.188274 0.222487 8 1.710186 -1.685984 -0.301775 1 0.155884 -0.226073 -1.526722 9 -1.911345 1.099923 -0.654924 9 -2.386673 -0.956767 -0.156662 9 -1.449487 0.325814 1.324340 8 3.279493 0.457583 0.323859 1 2.878271 -0.422784 0.134575 1 4.085206 0.508678 -0.197337 8 0.967077 1.729600 -0.302147 1 1.874419 1.418739 -0.068966 1 0.707409 2.345586 0.389960	Frequencies (cm⁻¹): 29.7774, 35.1066, , 82.1077, 119.2790, 134.9192, 186.3092, 224.0826, 246.1701, 255.0849, 291.8496, 316.9709, 378.1117, 405.2150, 423.9155, 457.8222, 503.7150, 551.7795, 567.5064, 697.2139, 721.0081, 871.9403, 897.5611, 925.5407, 938.4995, 1142.2460, 1193.3356, 1253.5596, 1359.8805, 1587.0451, 1645.5366, 1672.1439, 3211.2763, 3307.8078, 3403.5872, 3843.0633, 3860.4701



Compound: sCl 4 + (H ₂ O) ₂ PRC1	Energy (kJ mol⁻¹): -778.151182680016
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -1.111046 -0.242350 -0.451022 6 -0.319261 0.204891 0.799848 8 0.575408 -0.431904 1.369785 8 1.120026 -1.545860 0.653615 9 -0.884438 1.141948 1.494932 9 -1.929498 0.753446 -0.810346 9 -0.360874 -0.568707 -1.485809 9 -1.858617 -1.293633 -0.091832 8 3.018649 -0.020536 -0.530220 1 2.469097 -0.728816 -0.111137 1 3.419388 -0.408470 -1.312330 8 0.971763 1.712560 -0.356346 1 1.791707 1.193541 -0.556293 1 0.715716 2.156936 -1.170372	40.0825, 56.3618, 104.6947, 124.7390, 139.3793, 203.2420, 215.3984, 255.4641, 258.4817, 270.6400, 290.8050, 315.0485, 329.4540, 367.7848, 474.7718, 484.1919, 516.7047, 573.4577, 595.0829, 616.8202, 678.6174, 729.7337, 775.5858, 897.1599, 919.6383, 1148.1800, 1174.6125, 1237.1290, 1396.7823, 1618.5332, 1663.0453, 1680.6735, 3236.0138, 3341.1214, 3843.7417, 81.5127:36,A, 3864.2392

Compound: sCl 4 + (H ₂ O) ₂ TS1	Energy (kJ mol⁻¹): -778.151865721165
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -1.071336 -0.436984 -0.252730 6 -0.195585 0.603494 0.504754 8 0.644406 0.301050 1.388751 8 1.223716 -1.028274 1.197705 9 -0.852254 1.710937 0.754235 9 -1.930356 0.240110 -1.038544 9 -0.390157 -1.267841 -1.028562 9 -1.773384 -1.127291 0.642186 8 2.865226 -0.188927 -0.587322 1 2.360559 -0.647889 0.173793 1 3.166402 -0.874471 -1.189704 8 0.827583 1.247575 -0.958306 1 1.697315 0.678420 -0.962383 1 0.405161 1.190257 -1.824314	-198.8784, 46.7101, 59.4403, 154.1929, 180.3588, 225.9582, 267.3994, 284.3814, 305.0015, 352.9306, 366.8464, 375.1442, 413.2466, 466.3781, 489.4181, 524.0339, 590.5112, 604.9577, 620.5837, 681.7673, 768.9943, 815.7543, 922.5824, 949.7419, 1141.7483, 1172.0635, 1196.5290, 1236.0527, 1347.4144, 1540.3012, 1669.2159, 1713.3148, 2389.5730, 2791.1720, 3797.1962, 3855.6133



Compound: sCl 4 + (H ₂ O) ₂ Pr1	Energy (kJ mol⁻¹): -778.232205653240
Reaction Coordinates: 6 -1.331732 -0.259160 -0.190820 6 -0.039214 0.512746 0.233494 8 0.764673 -0.162432 1.133093 8 1.312123 -1.347541 0.514380 1 2.182508 -1.003043 0.204879 9 -0.474965 1.589770 0.953448 8 0.685330 0.923230 -0.860574 1 0.094400 1.270164 -1.541213 9 -2.075203 0.566046 -0.970916 9 -1.068394 -1.350754 -0.905759 9 -2.056443 -0.598624 0.871114 8 3.488053 -0.006943 -0.504972 1 4.126455 0.403140 0.086804 1 2.895931 0.699771 -0.792911	Frequencies (cm⁻¹): 32.3377, 52.9666, 94.2196, 128.8150, 195.0532, 218.4669, 228.6420, 249.4912, 273.3105, 290.7010, 313.4979, 346.1309, 373.3548, 414.6536, 455.3092, 489.2980, 530.6004, 591.5891, 641.9204, 688.4057, 780.1559, 801.9113, 973.3600, 1055.2043, 1096.2389, 1155.3190, 1174.3793, 1225.3692, 1284.0680, 1391.8185, 1531.8581, 1625.7848, 3408.2526, 3757.6239, 3776.9067, 3869.8559

Compound: sCl 4 + (H ₂ O) ₂ PRC2	Energy (kJ mol⁻¹): -778.152030386117
Reaction Coordinates: 6 -1.102056 -0.246402 -0.448061 6 -0.309488 0.215351 0.798177 8 0.581044 -0.419529 1.378511 8 1.132067 -1.536629 0.665104 9 -0.883620 1.150068 1.490200 9 -1.922500 0.745436 -0.815800 9 -0.352326 -0.581202 -1.478943 9 -1.847989 -1.294069 -0.075748 8 2.932636 -0.065549 -0.689559 1 2.410849 -0.746220 -0.192153 1 3.788217 -0.003687 -0.256363 8 0.961724 1.705387 -0.353570 1 1.779495 1.176094 -0.549424 1 0.688847 2.108570 -1.184028	Frequencies (cm⁻¹): 44.2525, 56.9310, 108.9406, 127.2046, 144.0984, 205.2413, 220.0133, 259.9402, 278.6197, 289.1779, 291.1197, 327.8403, 367.6214, 373.9608, 453.8346, 477.7278, 511.0120, 544.8312, 583.1474, 616.4050, 680.1004, 775.2032, 854.8557, 905.2343, 1010.7203, 1147.5637, 1173.6930, 1240.8031, 1394.5360, 1614.3180, 1634.0857, 1665.3915, 3182.2936, 3305.4328, 3836.1632, 3858.2832

Compound: sCl 4 + (H ₂ O) ₂ TS2	Energy (kJ mol⁻¹): -778.153124563473
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -1.084712 -0.399692 -0.287877 6 -0.191557 0.546852 0.565262 8 0.658260 0.151351 1.399623 8 1.215031 -1.162107 1.079200 9 -0.822179 1.637356 0.924134 9 -1.940502 0.364194 -0.992171 9 -0.419363 -1.156160 -1.144925 9 -1.789585 -1.166016 0.542485 8 2.818351 -0.243774 -0.704543 1 2.332416 -0.728241 0.049174 1 3.660685 0.063565 -0.357981 8 0.841237 1.328127 -0.860702 1 1.709802 0.766160 -0.892860 1 0.436348 1.312415 -1.736981	-177.7862, 49.0472, 59.1808, 160.8012, 178.1110, 225.5547, 265.3601, 284.9991, 302.9429, 350.7331, 365.3687, 371.9797, 411.7477, 478.0180, 488.3767, 524.4662, 554.8934, 590.9387, 610.9689, 658.8232, 762.7452, 799.4225, 918.3369, 1066.2885, 1143.3405, 1176.7433, 1228.7899, 1283.7771, 1353.1035, 1546.1413, 1614.5921, 1666.4543, 2456.5473, 2835.4191, 3795.4097, 3851.4800
IRC	
<p>Relative Energy (kJ mol⁻¹)</p> <p>Reaction Coordinates (amu^{1/2} bohr)</p>	

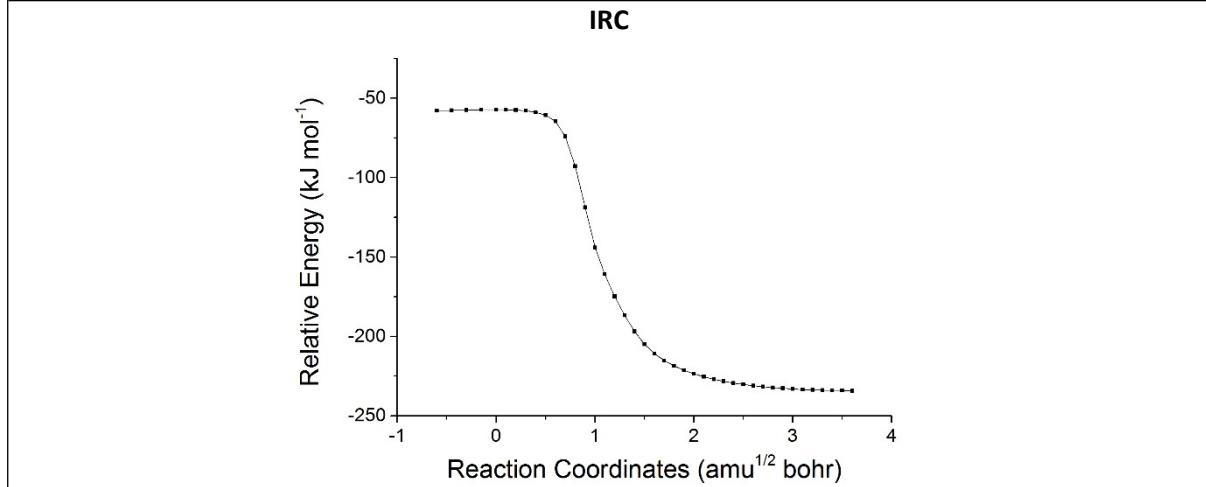
Compound: sCl 4 + (H ₂ O) ₂ PRC3	Energy (kJ mol⁻¹): -778.150954513382
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -1.080992 -0.293421 -0.439963 6 -0.309977 0.296432 0.768099 8 0.558811 -0.280279 1.437111 8 1.142219 -1.451989 0.846038 9 -0.903730 1.292366 1.361635 9 -1.932577 0.628095 -0.892785 9 -0.309351 -0.698009 -1.425552 9 -1.793038 -1.329614 0.028997 8 2.895030 -0.122247 -0.698902 1 2.389433 -0.744109 -0.114572 1 3.785256 -0.065695 -0.342575 8 0.878281 1.626475 -0.616822 1 1.728462 1.123576 -0.720068 1 1.096196 2.456950 -0.181662	44.4745, 66.9528, 96.0815, 125.4516, 153.5415, 200.0214, 219.7523, 260.3854, 275.8252, 281.7944, 289.4425, 326.5408, 354.4326, 368.4141, 467.1720, 478.9558, 512.2091, 570.8623, 586.5908, 615.7412, 664.1888, 775.7265, 811.7557, 899.0429, 990.4887, 1138.0675, 1179.2284, 1248.4219, 1385.5339, 1609.8626, 1639.6982, 1667.4722, 3188.2346, 3302.8218, 3837.7311, 3863.9650

Compound: sCl 4 + (H ₂ O) ₂ TS3	Energy (kJ mol⁻¹): -778.152472459163																								
Reaction Coordinates:	Frequencies (cm⁻¹):																								
6 -1.082084 -0.407518 -0.299218 6 -0.198345 0.551416 0.551826 8 0.634511 0.174899 1.410128 8 1.208951 -1.137901 1.126727 9 -0.818533 1.666468 0.875847 9 -1.970688 0.327254 -0.978632 9 -0.403180 -1.149068 -1.152950 9 -1.751044 -1.192463 0.551658 8 2.804368 -0.280419 -0.710283 1 2.322994 -0.734257 0.060595 1 3.676598 -0.023686 -0.399645 8 0.791809 1.255863 -0.967272 1 1.683872 0.744202 -0.952463 1 0.973014 2.181102 -0.761853	-146.3251, 49.3026, 72.0586, 154.6811, 178.9292, 219.8028, 260.3543, 284.4433, 301.4294, 346.7877, 360.6724, 366.8654, 398.8429, 455.9246, 480.5366, 524.7640, 572.1872, 592.4314, 615.8183, 681.7242, 751.1917, 778.9990, 909.6363, 1006.4112, 1144.1584, 1178.5495, 1208.3181, 1237.4715, 1352.2406, 1550.9475, 1624.7314, 1689.1680, 2567.1970, 2888.7471, 3804.3213, 3857.4399																								
IRC																									
<p>The plot shows the potential energy surface along the reaction coordinate. The y-axis represents the relative energy in kJ mol⁻¹, ranging from -250 to -50. The x-axis represents the reaction coordinate in amu^{1/2} bohr, ranging from -1 to 4. The curve is flat at approximately -48 kJ mol⁻¹ from x = -0.5 to 0.5, then drops sharply to about -225 kJ mol⁻¹ at x = 4.0.</p> <table border="1"> <caption>Data points estimated from the IRC plot</caption> <thead> <tr> <th>Reaction Coordinate (amu^{1/2} bohr)</th> <th>Relative Energy (kJ mol⁻¹)</th> </tr> </thead> <tbody> <tr><td>-0.5</td><td>-48</td></tr> <tr><td>0.0</td><td>-48</td></tr> <tr><td>0.5</td><td>-48</td></tr> <tr><td>0.8</td><td>-80</td></tr> <tr><td>1.0</td><td>-120</td></tr> <tr><td>1.2</td><td>-150</td></tr> <tr><td>1.5</td><td>-180</td></tr> <tr><td>2.0</td><td>-210</td></tr> <tr><td>2.5</td><td>-220</td></tr> <tr><td>3.0</td><td>-225</td></tr> <tr><td>4.0</td><td>-225</td></tr> </tbody> </table>		Reaction Coordinate (amu ^{1/2} bohr)	Relative Energy (kJ mol ⁻¹)	-0.5	-48	0.0	-48	0.5	-48	0.8	-80	1.0	-120	1.2	-150	1.5	-180	2.0	-210	2.5	-220	3.0	-225	4.0	-225
Reaction Coordinate (amu ^{1/2} bohr)	Relative Energy (kJ mol ⁻¹)																								
-0.5	-48																								
0.0	-48																								
0.5	-48																								
0.8	-80																								
1.0	-120																								
1.2	-150																								
1.5	-180																								
2.0	-210																								
2.5	-220																								
3.0	-225																								
4.0	-225																								

Compound: sCl 4 + (H ₂ O) ₂ Pr3	Energy (kJ mol⁻¹): -778.234111047535
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -1.404526 -0.102729 -0.158678 6 0.034254 0.408306 0.152328 8 0.766080 -0.465395 0.988045 8 1.165242 -1.636982 0.246539 1 2.053615 -1.369750 -0.045347 9 -0.113531 1.494110 0.982194 8 0.644104 0.732125 -1.004260 1 1.605443 0.791571 -0.855854 9 -2.113371 0.863791 -0.760818 9 -1.386305 -1.169082 -0.962133 9 -2.029860 -0.437116 0.975584 8 3.370981 0.253816 -0.332380 1 3.626690 0.591064 0.533536 1 4.152231 0.319812 -0.891232	33.9279, 48.3442, 76.1769, 116.7370, 164.3800, 207.5971, 232.5643, 262.9793, 278.8395, 285.3071, 314.5749, 352.1083, 377.7050, 397.3941, 479.3980, 491.5629, 531.8580, 585.8175, 631.6467, 656.0214, 687.7963, 785.5537, 965.8965, 1034.0498, 1076.4672, 1180.9327, 1200.5000, 1231.8247, 1279.5941, 1450.8232, 1474.8038, 1626.3847, 3592.4584, 3661.4694, 3782.2367, 3879.5888

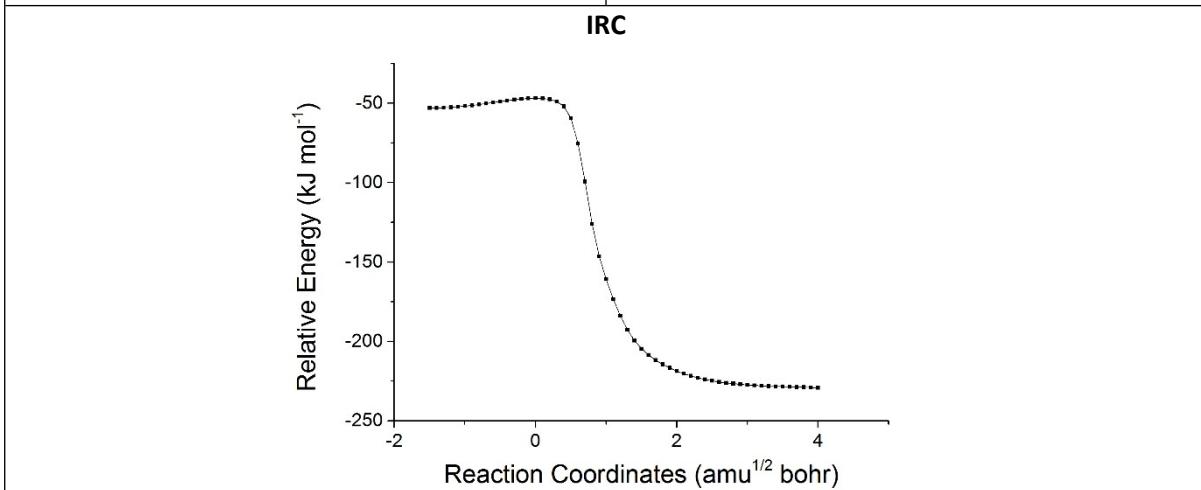
Compound: sCl 4 + (H ₂ O) ₂ PRC4	Energy (kJ mol⁻¹): -778.151343097919
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -1.087243 -0.309769 -0.423180 6 -0.300324 0.336446 0.745069 8 0.562981 -0.217347 1.441526 8 1.141377 -1.414420 0.896658 9 -0.890442 1.359501 1.296098 9 -1.936840 0.592432 -0.916534 9 -0.328843 -0.770648 -1.397152 9 -1.800836 -1.316349 0.102686 8 2.962826 -0.076017 -0.560233 1 2.434468 -0.715318 -0.014219 1 3.301488 -0.567780 -1.313216 8 0.878739 1.568034 -0.692410 1 1.724342 1.046292 -0.767222 1 1.110375 2.410320 -0.286886	46.2907, 67.0974, 91.8090, 125.6174, 157.8162, 203.5801, 221.3492, 264.0061, 277.3274, 288.9341, 289.9045, 328.7980, 367.2176, 400.7399, 478.6531, 506.1075, 512.5963, 568.4710, 590.0375, 619.6265, 663.8341, 776.0588, 796.5534, 902.5942, 1034.2265, 1140.0628, 1180.1310, 1242.4727, 1383.1677, 1606.5946, 1647.5553, 1678.8281, 3134.9621, 3275.3006, 3830.4288, 3858.6479

Compound: sCl 4 + (H ₂ O) ₂ TS4	Energy (kJ mol⁻¹): -778.153113870118
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -1.077057 -0.423552 -0.289992 6 -0.213471 0.573107 0.536270 8 0.614289 0.237747 1.414317 8 1.202688 -1.077550 1.192543 9 -0.841564 1.697007 0.803171 9 -1.964096 0.275233 -1.006670 9 -0.380214 -1.189805 -1.110522 9 -1.747493 -1.187118 0.578433 8 2.873625 -0.201596 -0.590203 1 2.368533 -0.672630 0.151143 1 3.159081 -0.873792 -1.215226 8 0.800188 1.222752 -1.014849 1 1.679394 0.693746 -0.977376 1 1.010146 2.146665 -0.830371	-131.8994, 49.7399, 71.5302, 148.2910, 178.8931, 220.1116, 257.5315, 284.2300, 299.6407, 338.1935, 345.6464, 363.6927, 394.7051, 479.8384, 488.8758, 522.9158, 585.5910, 593.2244, 608.6505, 679.4158, 739.0556, 778.5179, 910.2331, 935.9471, 1146.0830, 1180.7008, 1227.9638, 1243.4577, 1355.2067, 1557.3275, 1644.1015, 1693.0530, 2597.1393, 2930.6112, 3800.2592, 3853.4549



Compound: sCl 5 + (H ₂ O) ₂ PRC1	Energy (kJ mol⁻¹): -778.155321450549
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -1.504971 0.099406 0.193399 6 -0.160033 -0.476306 -0.265725 8 0.586751 -1.004163 0.567732 8 1.752981 -1.640438 0.080858 9 -0.001255 -0.619282 -1.534660 9 -2.417557 -0.889230 0.123292 9 -1.438786 0.532881 1.440451 9 -1.895849 1.079132 -0.615166 8 3.287219 0.603792 0.401354 1 2.917263 -0.305750 0.320743 1 4.114254 0.608406 -0.087241 8 0.878010 1.700375 -0.234065 1 1.818974 1.471403 -0.050230 1 0.880874 2.279305 -1.001596	29.4729, 57.0746, 90.7417, 110.8940, 132.9723, 165.3457, 188.1994, 242.3211, 245.6618, 251.7649, 285.3981, 302.2243, 350.6313, 361.6697, 408.5250, 456.8954, 521.4813, 556.2997, 576.4187, 668.3638, 675.6645, 726.7701, 836.6869, 853.3457, 875.4031, 1128.9044, 1186.8392, 1249.0461, 1403.4828, 1632.9929, 1660.5967, 1674.8863, 3334.9174, 3419.1514, 3854.0781, 3865.4630

Compound: sCl 5 + (H ₂ O) ₂ TS1	Energy (kJ mol⁻¹): -778.154483796209
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -1.452057 -0.094944 -0.202984 6 -0.035845 0.328312 0.244444 8 0.643489 0.940770 -0.616381 8 1.899689 1.481311 -0.106517 9 0.026824 0.653674 1.514325 9 -1.969470 -0.968997 0.665102 9 -2.234068 0.992813 -0.218975 9 -1.432751 -0.632623 -1.414317 8 3.030580 -0.790958 -0.439314 1 2.771664 0.195011 -0.357470 1 3.841343 -0.926600 0.058152 8 0.723371 -1.395286 0.379901 1 1.717345 -1.268679 0.095196 1 0.705207 -1.690439 1.298636	-231.7521, 36.0051, 70.1748, 146.3315, 159.4260, 185.6379, 275.1852, 297.2816, 324.5751, 352.2152, 376.7297, 388.5598, 399.7561, 415.8618, 503.1695, 530.5402, 577.9666, 635.6021, 680.1667, 720.5649, 786.4159, 823.5539, 862.7035, 937.3476, 1137.6409, 1181.8605, 1230.4427, 1238.2808, 1358.6367, 1535.1706, 1670.4773, 1707.3520, 2352.8141, 2779.8576, 3800.1789, 3857.8733



Compound: sCl 5 + (H ₂ O) ₂ Pr1	Energy (kJ mol⁻¹): -778.234695133756
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -1.391502 -0.324803 -0.183500 6 -0.060577 0.360168 0.265844 8 0.650426 0.577335 -0.877070 8 1.877358 1.273273 -0.560158 1 2.503994 0.515564 -0.497802 9 -0.415662 1.527793 0.894584 8 0.634214 -0.433231 1.158979 1 0.212620 -0.376145 2.025158 9 -2.119291 -0.605465 0.915204 9 -2.113200 0.467005 -0.974187 9 -1.148361 -1.471151 -0.823439 8 3.269025 -1.063971 -0.100132 1 4.117224 -1.122038 0.349666 1 2.599077 -1.320459 0.546508	41.1071, 60.7539, 94.4583, 129.5898, 168.0954, 191.6362, 221.2917, 237.0399, 249.2516, 304.8284, 316.2066, 339.9198, 370.2838, 386.2925, 400.1450, 519.9118, 527.7303, 588.4687, 604.3415, 728.9488, 739.0615, 825.2194, 938.9418, 1052.5637, 1076.7670, 1161.1253, 1179.2310, 1212.3881, 1326.7694, 1362.8227, 1545.5041, 1626.3435, 3424.4355, 3759.2232, 3792.0834, 3875.4461

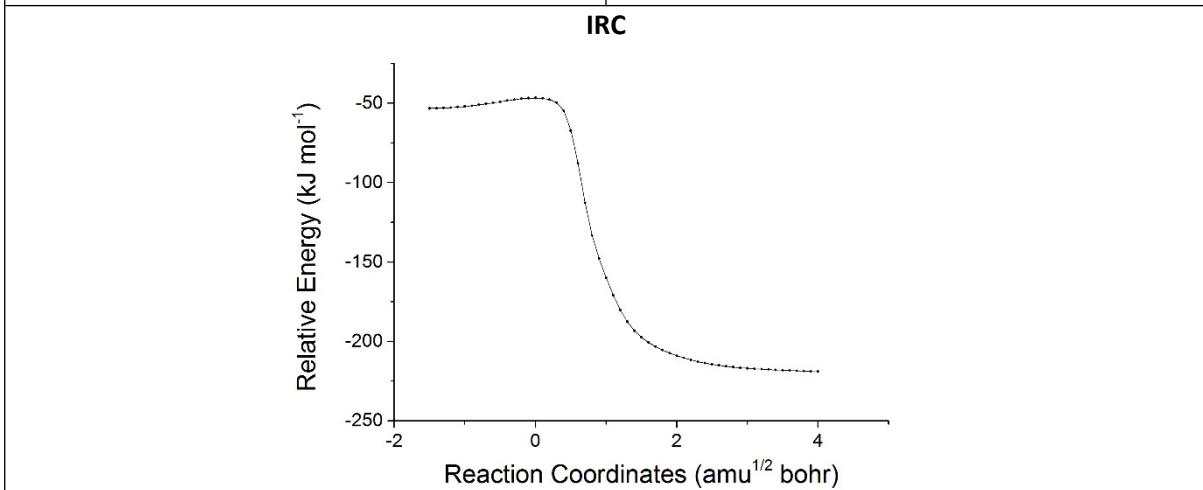
Compound: sCl 5 + (H ₂ O) ₂ PRC2	Energy (kJ mol⁻¹): -778.156341672640
Reaction Coordinates:	Frequencies (cm⁻¹):
6 1.505495 0.091967 -0.182594 6 0.145724 -0.463043 0.259704 8 -0.580325 -1.013433 -0.578212 8 -1.759359 -1.639526 -0.100613 9 -0.039231 -0.580917 1.527508 9 1.890669 1.082516 0.615014 9 2.406149 -0.904017 -0.077432 9 1.467986 0.503414 -1.438943 8 -3.304652 0.601187 -0.208372 1 -2.912148 -0.303888 -0.208056 1 -3.817922 0.675253 -1.017456 8 -0.861293 1.702999 0.161215 1 -1.805294 1.458593 0.002872 1 -0.857080 2.277717 0.932501	32.9450, 59.2730, 96.3828, 114.5765, 135.1450, 169.7028, 194.0592, 250.8541, 258.7338, 273.8438, 300.7887, 335.5543, 356.6988, 361.4179, 408.9676, 419.6888, 511.1819, 524.4003, 576.1856, 673.9396, 726.6771, 793.2353, 838.3362, 858.1026, 953.6168, 1129.8197, 1187.6602, 1247.7782, 1403.7621, 1627.8877, 1637.1693, 1664.8016, 3293.5011, 3392.0249, 3847.2631, 3859.1687

Compound: sCl 5 + (H ₂ O) ₂ TS2	Energy (kJ mol⁻¹): -778.155935062267
Reaction Coordinates:	Frequencies (cm⁻¹):
6 1.456606 0.088286 -0.198395 6 0.036941 -0.328311 0.242080 8 -0.639985 -0.946054 -0.615417 8 -1.893626 -1.493088 -0.108280 9 -0.046269 -0.623819 1.514582 9 1.965682 0.978723 0.656214 9 2.238800 -0.999666 -0.186658 9 1.448326 0.601553 -1.421252 8 -3.088705 0.769786 -0.261179 1 -2.789827 -0.205953 -0.262939 1 -3.460061 0.964938 -1.126053 8 -0.721292 1.424044 0.330129 1 -1.713908 1.278347 0.061623 1 -0.707471 1.754189 1.237248	-216.7111, 40.3464, 71.3754, 145.7554, 159.8329, 190.0347, 275.2563, 296.7082, 321.7700, 352.9757, 365.4023, 383.5055, 400.9371, 423.6081, 502.0856, 530.9374, 569.4630, 584.9973, 673.1975, 718.6764, 776.5858, 821.7894, 865.7069, 1058.9036, 1140.9912, 1183.5082, 1229.8557, 1306.4008, 1365.6796, 1541.4011, 1613.3738, 1666.1703, 2396.9997, 2814.5238, 3796.9692, 3852.1032
IRC	
<p>Relative Energy (kJ mol⁻¹)</p> <p>Reaction Coordinates (amu^{1/2} bohr)</p>	

Compound: sCl 5 + (H ₂ O) ₂ Pr2	Energy (kJ mol⁻¹): -778.235686023368
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -1.378531 -0.337083 -0.171891 6 -0.058616 0.384232 0.252293 8 0.667177 0.532385 -0.893775 8 1.888164 1.250847 -0.604039 1 2.512185 0.498809 -0.471100 9 -0.426818 1.585788 0.801457 8 0.627112 -0.349190 1.201680 1 0.251341 -0.172849 2.072726 9 -2.128952 -0.542988 0.925578 9 -2.085203 0.390208 -1.035037 9 -1.114776 -1.526471 -0.723517 8 3.280512 -1.025707 0.066205 1 3.427652 -1.733253 -0.569030 1 2.529728 -1.301115 0.608085	41.8459, 61.8886, 97.8940, 133.9444, 167.1742, 202.7544, 221.7193, 235.2008, 248.8184, 282.3243, 309.0498, 340.7138, 373.9820, 385.3118, 448.2935, 521.3360, 531.7991, 588.4637, 605.8464, 732.5280, 796.7416, 827.2153, 939.3215, 1050.6581, 1079.8613, 1161.1824, 1181.5318, 1208.9991, 1327.8406, 1357.9245, 1528.8648, 1624.7530, 3406.4916, 3756.1722, 3794.7283, 3870.0375

Compound: sCl 5 + (H ₂ O) ₂ PRC3	Energy (kJ mol⁻¹): -778.155292325029
Reaction Coordinates:	Frequencies (cm⁻¹):
6 1.520553 0.077431 -0.133529 6 0.135438 -0.454670 0.253545 8 -0.530569 -1.060240 -0.597416 8 -1.735393 -1.666024 -0.170800 9 -0.147230 -0.479485 1.505653 9 1.895351 1.053734 0.681438 9 2.401742 -0.932279 -0.020323 9 1.526387 0.507772 -1.390230 8 -3.298053 0.579879 -0.278807 1 -2.896847 -0.320345 -0.267278 1 -4.002248 0.571976 0.375149 8 -0.886250 1.748320 0.139381 1 -1.826386 1.476250 0.011460 1 -0.694596 2.372397 -0.567128	25.7960, 41.9021, 91.7538, 116.0609, 131.6452, 166.6911, 189.2848, 241.5052, 249.2321, 261.3399, 299.9707, 321.0976, 349.3877, 361.8052, 406.4638, 446.2832, 509.0366, 524.6695, 575.3292, 678.4179, 726.8862, 730.8695, 844.2676, 858.9722, 933.6184, 1136.4634, 1190.4264, 1239.2175, 1407.6330, 1630.3118, 1650.7443, 1673.1293, 3312.5611, 3406.4417, 3846.2994, 3859.1966

Compound: sCl 5 + (H ₂ O) ₂ TS3	Energy (kJ mol⁻¹): -778.154047110573
Reaction Coordinates:	Frequencies (cm⁻¹):
6 1.483665 0.044071 -0.083684 6 -0.005044 -0.255421 0.204566 8 -0.553413 -1.049839 -0.607125 8 -1.871996 -1.524924 -0.202453 9 -0.296320 -0.294613 1.477390 9 1.953513 0.959049 0.758604 9 2.193963 -1.077157 0.059407 9 1.630507 0.489838 -1.337807 8 -3.080035 0.726521 -0.184941 1 -2.743969 -0.245753 -0.204133 1 -3.659142 0.821790 0.576606 8 -0.694641 1.469487 -0.086641 1 -1.730137 1.287986 -0.079072 1 -0.472765 1.740066 -0.987767	-250.6799, 39.0414, 60.9470, 136.9924, 161.6331, 177.9353, 287.5360, 301.9577, 318.5255, 354.5256, 369.5178, 392.4984, 414.7727, 425.6464, 528.2966, 539.3143, 574.3108, 635.7136, 679.6143, 720.4778, 805.9932, 841.1296, 876.5957, 1023.1045, 1148.2582, 1189.5948, 1216.7132, 1332.4466, 1385.5696, 1517.5508, 1651.9311, 1680.8426, 2199.0066, 2688.3510, 3780.5289, 3849.5355



Compound: sCl 5 + (H ₂ O) ₂ Pr3	Energy (kJ mol⁻¹): -778.234941477938
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -1.281619 -0.396869 -0.141349 6 -0.073615 0.502297 0.285331 8 0.679689 0.627928 -0.880580 8 1.863158 1.410381 -0.608218 1 2.514029 0.701364 -0.464738 9 -0.587670 1.698848 0.673107 8 0.589165 -0.024461 1.352062 1 1.313173 -0.592262 1.036499 9 -2.127289 -0.534924 0.882116 9 -1.949836 0.114088 -1.178433 9 -0.840746 -1.623148 -0.487456 8 2.827771 -1.274646 0.090297 1 2.584162 -1.895382 -0.605510 1 3.591648 -1.653679 0.537359	30.8261, 58.0992, 91.0942, 125.3530, 168.5796, 188.2988, 222.2970, 239.3348, 272.1691, 295.7873, 313.8342, 340.1174, 378.3839, 379.9574, 466.4912, 514.4522, 531.0048, 583.5516, 603.2174, 630.9793, 732.1940, 829.7446, 939.0117, 1048.7712, 1086.3898, 1165.0852, 1200.8282, 1208.3361, 1331.2638, 1405.5919, 1457.5082, 1627.6146, 3614.9543, 3656.6230, 3783.3352, 3879.9193

Compound: sCl 5 + (H ₂ O) ₂ PRC4	Energy (kJ mol⁻¹): -778.154802564593
Reaction Coordinates:	Frequencies (cm⁻¹):
6 1.523631 0.070204 -0.117940 6 0.126545 -0.448086 0.244963 8 -0.517977 -1.072454 -0.609207 8 -1.734406 -1.668085 -0.200888 9 -0.192591 -0.438451 1.487670 9 1.887208 1.054233 0.691694 9 2.395072 -0.943889 0.026528 9 1.559768 0.483838 -1.380501 8 -3.302494 0.564233 -0.063406 1 -2.884949 -0.325637 -0.137135 1 -3.976892 0.606944 -0.746286 8 -0.867860 1.766934 0.050317 1 -1.815934 1.508925 -0.029492 1 -0.686495 2.350457 -0.692268	19.4727, 41.9568, 92.4435, 115.0151, 131.1127, 166.1585, 185.4852, 246.9869, 255.5549, 255.9820, 280.3702, 300.0568, 350.8270, 362.2080, 405.1394, 421.3097, 521.6741, 535.5956, 575.1269, 678.9162, 718.8823, 730.9504, 840.7204, 858.5765, 910.2685, 1136.8887, 1190.2595, 1239.3736, 1409.2583, 1630.3548, 1644.9207, 1672.4348, 3327.8774, 3413.8679, 3850.3844, 3863.5024

Compound: sCl 5 + (H ₂ O) ₂ TS4	Energy (kJ mol⁻¹): -778.153225068607
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -1.482042 -0.041610 -0.082930 6 0.010444 0.247204 0.200130 8 0.556018 1.044842 -0.614002 8 1.877025 1.518238 -0.210042 9 0.304275 0.288980 1.471268 9 -1.956400 -0.952454 0.760697 9 -2.184337 1.084144 0.058772 9 -1.634812 -0.489053 -1.337317 8 3.062989 -0.717905 0.000049 1 2.725841 0.253057 -0.108109 1 3.703589 -0.893273 -0.694674 8 0.686397 -1.465941 -0.099859 1 1.728012 -1.293563 -0.087031 1 0.454186 -1.718206 -1.003343	-269.4438, 38.7369, 61.8805, 137.6329, 165.4533, 177.4985, 287.8044, 300.9685, 321.0749, 355.1282, 391.5197, 406.9766, 418.2621, 429.1966, 509.3777, 536.6606, 573.7250, 630.1877, 682.9775, 721.9018, 812.5196, 845.7557, 873.0997, 1071.2964, 1151.0672, 1187.7224, 1217.4694, 1320.0152, 1376.8772, 1507.2047, 1621.9928, 1698.8260, 2132.0505, 2622.0161, 3783.7703, 3852.4844

IRC

Reaction Coordinates (amu ^{1/2} bohr)	Relative Energy (kJ mol ⁻¹)
-1.5	-48
-1.0	-48
0.0	-48
0.5	-50
1.0	-120
1.5	-160
2.0	-180
2.2	-190

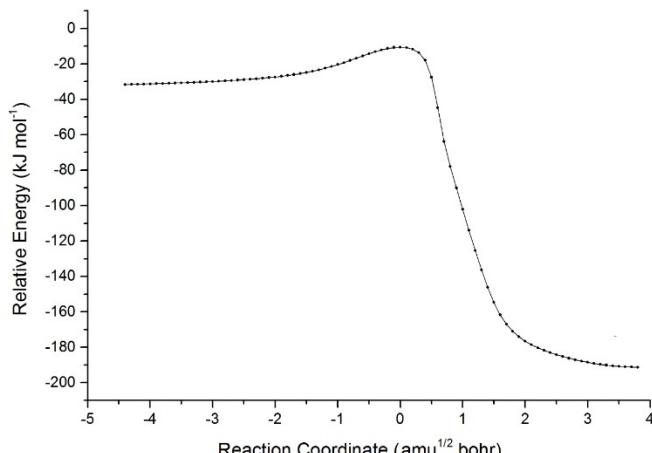
S8.11 Reactions with MeOH

Compound: MeOH	Energy (kJ mol⁻¹): -115.607115778238
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -0.667408 -0.020462 0.000000	290.6856, 1039.0192, 1076.1201,
8 0.748822 0.122098 0.000000	1170.5975, 1365.7526, 1477.3714,
1 1.151097 -0.750378 -0.000000	1498.2111, 1508.8548, 2994.1556,
1 -1.026778 -0.544214 -0.890795	3039.5029, 3108.4416, 3828.6843
1 -1.026779 -0.544207 0.890799	
1 -1.083670 0.984791 -0.000004	

Compound: sCl 1 + MeOH PRC	Energy (kJ mol⁻¹): -305.020987250893
Reaction Coordinates:	Frequencies (cm⁻¹):
6 2.662111 -0.046788 0.000044	24.8880, 82.1748, 110.6320,
8 1.315288 0.399727 -0.000067	111.6029, 135.5092, 259.5276,
1 0.698704 -0.357599 -0.000052	549.7940, 665.8058, 696.6622,
1 2.893216 -0.640865 -0.889276	879.6470, 1019.2042, 1061.0776,
1 3.300954 0.835093 0.000016	1138.6918, 1172.9739, 1273.0125,
1 2.893113 -0.640734 0.889478	1420.9590, 1449.4821, 1477.4975,
6 -1.635466 0.998502 0.000005	1497.7786, 1512.0957, 1560.2724,
8 -1.950142 -0.213639 0.000033	2992.6031, 3037.4403, 3082.1700,
8 -0.974471 -1.171188 -0.000023	3095.0315, 3243.6738, 3517.8399
1 -0.591507 1.299365 -0.000058	
1 -2.479754 1.675256 0.000055	

Compound: sCl 1 + MeOH TS1	Energy (kJ mol⁻¹): -305.012539571568
Reaction Coordinates:	Frequencies (cm⁻¹):
6 1.863745 -0.172056 0.389771	-303.3420, 135.4580, 151.7227,
8 0.934811 0.041282 -0.675937	188.5069, 331.0013, 453.7042,
1 0.218702 -0.701416 -0.634231	543.7435, 799.9595, 865.9194,
1 2.400052 -1.106997 0.222573	1006.5269, 1112.0219, 1165.7576,
1 2.571438 0.654242 0.391134	1181.7513, 1211.1356, 1251.4969,
1 1.353139 -0.227500 1.353711	,1381.3223, 1464.5139, 1490.1175,
6 -0.701900 1.051055 -0.161035	1494.0108, 1518.0904, 1546.2798,
8 -1.199505 0.166901 0.601952	2568.4657, 3021.4453, 3082.0592,
8 -1.260499 -1.070358 -0.153254	3100.9807, 3116.4553, 3224.8375
1 -0.936660 1.041876 -1.217190	
1 -0.376203 1.963201 0.329508	

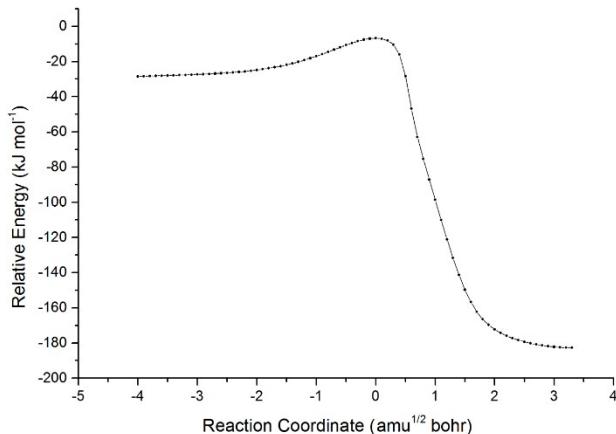
IRC



Compound: sCl 1 + MeOH Pr1	Energy (kJ mol⁻¹): -305.089754793660
Reaction Coordinates:	Frequencies (cm⁻¹):
6 1.866681 -0.450981 0.277009 8 0.921620 0.046899 -0.666119 1 -1.363992 -1.344923 -0.391532 1 1.387002 -1.064896 1.042174 1 2.576523 -1.057064 -0.280043 1 2.399568 0.372679 0.762678 6 -0.055601 0.877798 -0.117095 8 -1.003092 0.190929 0.670639 8 -1.879360 -0.542945 -0.220829 1 -0.552119 1.373600 -0.949241 1 0.373190 1.600643 0.586949	106.8116, , 132.8786, 193.1857, 247.0967, 336.7449, 481.1437, 625.3832, 875.9852, 931.0861, 997.2289, 1089.2781, 1166.6214, 1175.0153, 1227.8307, 1319.6211, 1377.1010, 1404.1314, 1465.0513, 1483.3431, 1491.0236, 1509.4422, 2998.4938, 3004.6735, 3060.3119, 3105.6338, 3124.8533, 3730.9399

Compound: sCl 1 + MeOH TS2	Energy (kJ mol⁻¹): -305.010689091126
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -2.062272 -0.126803 0.198882 8 -0.873223 0.024855 -0.569815 1 -0.214440 -0.747176 -0.352060 1 -1.847455 -0.183951 1.270177 1 -2.707715 0.727406 0.003845 1 -2.578012 -1.040668 -0.100269 6 0.640245 1.001204 0.262268 8 1.540533 0.241828 -0.206971 8 1.213700 -1.115998 0.192501 1 0.218716 0.788854 1.237825 1 0.612990 2.003651 -0.152141	-314.5301, 101.6876, 161.3604, 201.5254, 339.3872, 430.7475, 527.8844, 805.4684, 867.5779, 1023.6465, 1084.8305, 1162.9890, 1178.4686, 1211.8856, 1228.6814, 1382.1058, 1466.4546, 1494.9678, 1496.1083, 1518.4641, 1588.9879, 2483.6260, 3006.1591, 3068.6532, 3092.7445, 3110.5584, 3214.1642

IRC

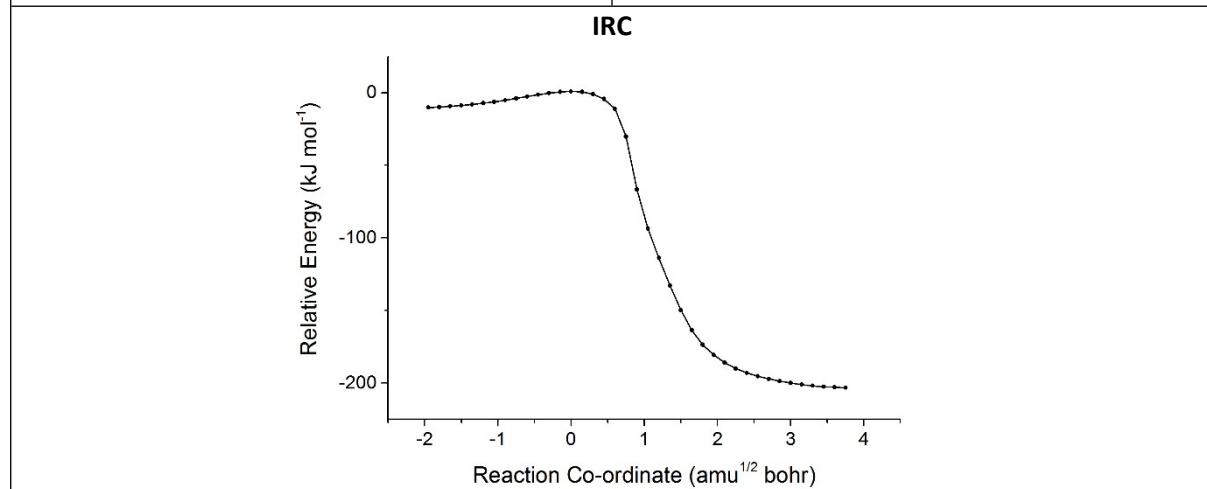


Compound: sCl 1 + MeOH Pr2	Energy (kJ mol⁻¹): -305.086315844733
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -2.116714 -0.273591 0.130007 8 -0.830169 -0.175072 -0.456215 1 1.492242 -1.320331 -0.436822 1 -2.675006 -1.004118 -0.449884 1 -2.647746 0.684065 0.098946 1 -2.055814 -0.610559 1.170591 6 0.028141 0.707631 0.225843 8 1.324359 0.556662 -0.254477 8 1.836333 -0.710029 0.232219	60.9145, 147.5248, 209.1549, 293.1227, 338.1013, 403.9749, 592.9671, 884.4403, 980.1253, 1063.0713, 1112.7765, 1149.6046, 1181.9207, 1223.0636, 1284.8284, 1380.7824, 1419.0086, 1472.8164, 1492.5940, 1494.2584, 1510.0321, 2973.2882, 2985.8560, 3021.2134, 3028.7648, 3121.4429, 3730.7851

1 -0.011022 0.520149 1.305723	
1 -0.215404 1.754062 0.004136	

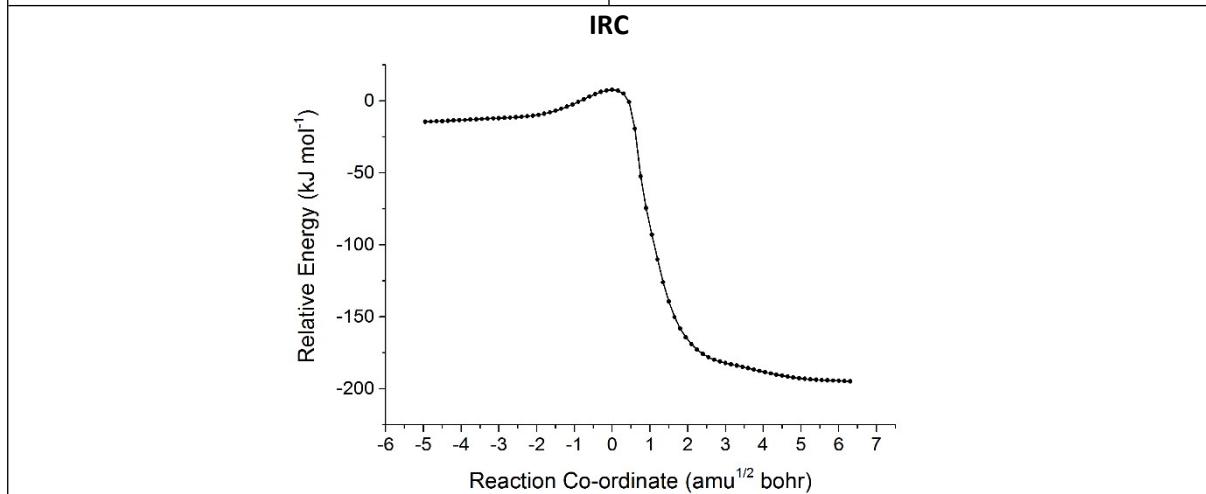
Compound: sCl 2 + MeOH PRC	Energy (kJ mol⁻¹): -641.819794756621
Reaction Coordinates:	Frequencies (cm⁻¹):
6 1.976708 -0.101676 -0.000009 6 0.996614 1.054995 -0.000301 8 -0.249822 0.916410 -0.000190 8 -0.792449 -0.316205 0.000213 1 1.349980 2.076669 -0.000628 9 1.846753 -0.862233 1.088123 9 1.846623 -0.862894 -1.087664 9 3.216292 0.424173 -0.000243 8 -3.584966 0.541108 0.000387 6 -4.538112 -0.507407 -0.000285 1 -4.454459 -1.141356 -0.889827 1 -5.526454 -0.049380 -0.000250 1 -4.454787 -1.142196 0.888688 1 -2.694655 0.158864 0.000358	7.8778, 13.7557, 25.1694, 81.2882, 84.6164, 111.2334, 128.0235, 216.1099, 246.9664, 334.8528, 479.5373, 509.5491, 535.8498, 587.6098, 593.6590, 759.4478, 800.7172, 884.0324, 930.6324, 1060.9119, 1120.5796, 1157.2830, 1171.8072, 1183.6161, 1242.3480, 1368.3585, 1435.6089, 1476.2118, 1497.0762, 1511.4617, 1558.0757, 2979.1708, 3018.1652, 3089.2369, 3213.6535, 3666.0389

Compound: sCl 2 + MeOH TS1	Energy (kJ mol⁻¹): -641.812455653684
Reaction Coordinates:	Frequencies (cm⁻¹):
6 1.150706 -0.248514 0.005650 6 -0.071605 0.205229 -0.809726 8 -0.611723 1.344246 -0.747619 8 -0.733833 1.770494 0.616132 1 -0.154234 -0.288556 -1.773672 9 2.171590 0.555970 -0.346428 9 1.053301 -0.254276 1.323237 9 1.454370 -1.497428 -0.400749 6 -2.690554 -0.860532 -0.159831 8 -1.391399 -0.737195 0.430712 1 -1.334405 0.156953 0.889988 1 -2.897122 -0.031950 -0.840073 1 -2.719147 -1.804051 -0.699014 1 -3.444072 -0.868249 0.627885	-238.8203, 43.8051, 115.7554, 142.3150, 155.5798, 189.5276, 207.0024, ,276.0069, 319.3805, 350.2625, 450.7578, 518.3362, 556.3651, 567.6475, 740.0036, 808.7216, 866.5284, 964.4575, 1019.1294, 1049.3368, 1140.5097, 1153.8883, 1176.3217, 1185.9025, 1284.6473, 1345.3465, 1461.2570, 1472.4469, 1494.5943, 1512.0143, 1557.6297, 2991.1165, 3032.5103, 3091.3535, 3130.1032, 3152.3093



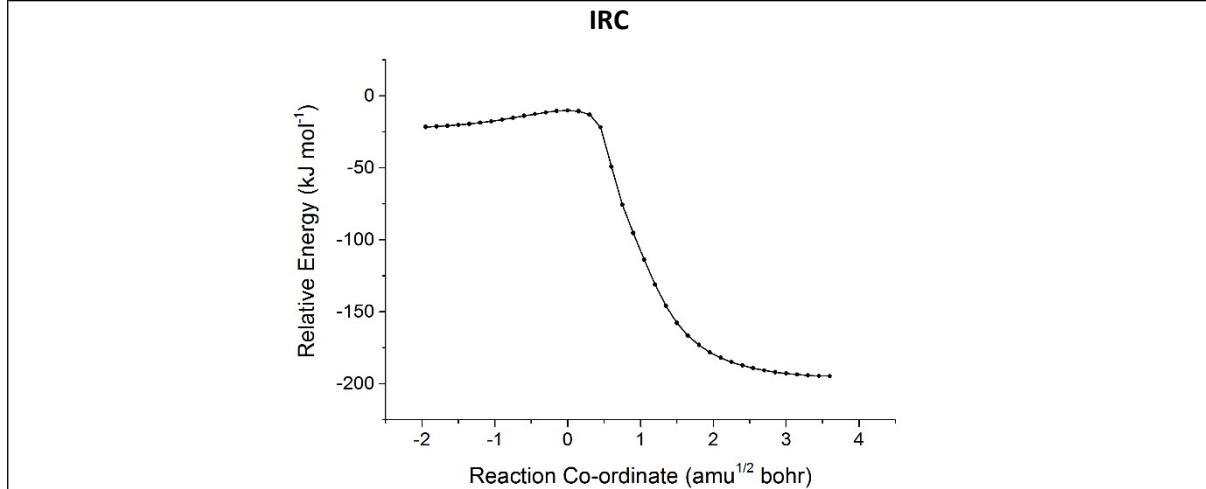
Compound: sCl 2 + MeOH Pr	Energy (kJ mol⁻¹): -641.901373422778
Reaction Coordinates:	Frequencies (cm⁻¹):
6 1.056705 -0.301665 -0.003500 6 -0.397832 -0.082695 -0.480375 8 -0.714080 1.258866 -0.757921 8 -0.472266 2.067993 0.418501 1 -0.482679 -0.547118 -1.470054 9 1.903483 0.451576 -0.718120 9 1.246603 -0.043594 1.291534 9 1.385379 -1.595895 -0.213342 6 -2.521234 -1.032375 -0.044004 8 -1.230634 -0.673220 0.454765 1 -1.272492 1.909798 0.941578 1 -2.430630 -1.768178 -0.847228 1 -3.059988 -1.473098 0.789745 1 -3.063378 -0.158881 -0.410168	60.3433, 90.3684, 124.3320, 151.2099, 188.2054, 212.0435, 237.2391, 305.6409, 340.8455, 363.2585, 422.9918, 528.8540, 570.8971, 655.1884, 754.2538, 868.8478, 912.9361, 1005.4468, 1038.2480, 1119.0917, 1138.1004, 1176.0669, 1184.7505, 1222.8966, 1280.1943, 1359.9493, 1384.1052, 1403.7418, 1478.3985, 1493.7604, 1507.6477, 3013.0098, 3019.4411, 3075.1426, 3138.0241, 3726.2609

Compound: sCl 2 + MeOH TS2	Energy (kJ mol⁻¹): -641.809840895326
Reaction Coordinates:	Frequencies (cm⁻¹):
6 0.960039 -0.367008 0.033781 6 0.122443 0.567493 -0.851126 8 -0.143691 1.775553 -0.591170 8 -0.619507 1.888320 0.773109 1 0.227665 0.343139 -1.908782 9 2.222760 0.096404 -0.002129 9 0.603569 -0.516218 1.299182 9 0.954865 -1.586276 -0.548416 6 -2.133213 -1.321958 0.136887 8 -1.682887 -0.076672 -0.411040 1 -1.606554 0.652054 0.300384 1 -2.003882 -2.088661 -0.622590 1 -1.581375 -1.594839 1.034505 1 -3.193516 -1.225657 0.374309	-277.7025, 52.5164, 87.4164, 135.7958, 147.4743, 219.8415, 244.7700, 290.8680, 320.4487, 370.4519, 440.5767, 514.9018, 547.2127, 570.6901, 736.4417, 796.0657, 859.8538, 956.7795, 1013.0329, 1054.6223, 1134.6948, 1157.8251, 1176.4975, 1200.3395, 1281.7528, 1349.0475, 1456.1694, 1477.7089, 1495.4363, ,1506.8177, 1574.5134, 2755.9507, 3037.8079, 3105.9950, 3134.6294, 3150.9265



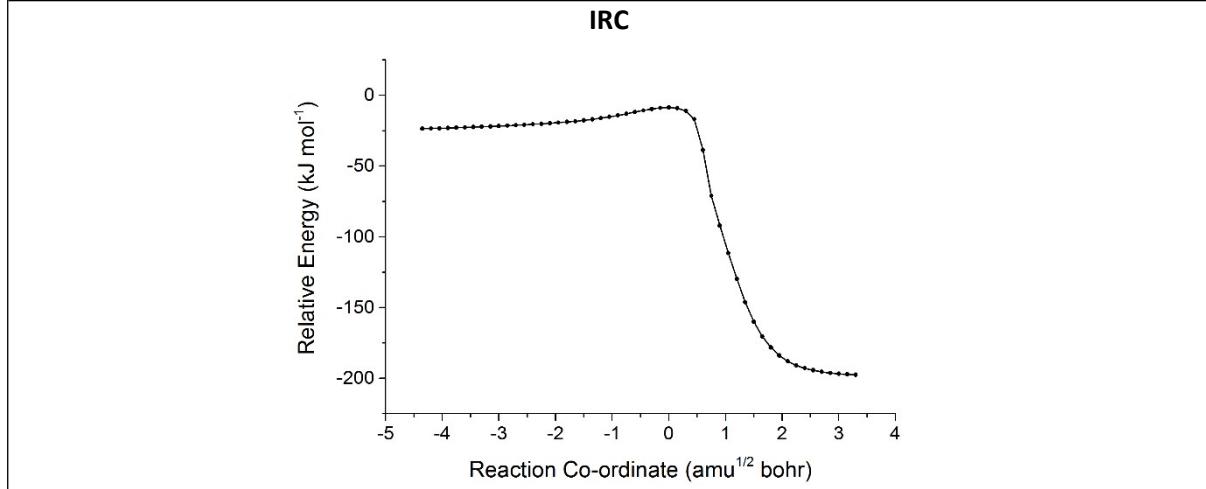
Compound: sCl 3 + MeOH PRC1	Energy (kJ mol⁻¹): -641.824942996462
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -1.702464 -0.317385 0.000001 6 -0.267270 0.153707 -0.000011 8 -0.081319 1.391538 0.000007 8 1.182001 1.890171 -0.000002 1 0.567221 -0.546139 -0.000033 9 -1.920794 -1.080498 -1.085601 9 -2.575951 0.686259 0.000039 9 -1.920758 -1.080548 1.085575 6 3.987682 -0.820759 0.000021 8 2.596940 -0.526234 -0.000035 1 2.443182 0.434526 -0.000022 1 4.483773 -0.424164 0.890025 1 4.483852 -0.424129 -0.889923 1 4.090831 -1.904185 0.000005	18.1908, 34.4140, 66.9668, 84.2489, 109.7187, 124.1555, 174.7514, 207.8882, 243.9959, 388.4638, 395.5289, 429.3178, 556.4368, 560.2631, 620.7028, 699.2395, 891.0590, 945.0049, 957.2409, 1056.3515, 1128.5630, 1140.7914, 1173.1779, 1183.4579, 1280.8269, 1379.4922, 1421.3731, 1477.8260, 1498.3830, 1510.6159, 1557.8223, 3000.7618, 3049.0279, 3103.5151, 3105.0225, 3595.1865

Compound: sCl 3 + MeOH TS1	Energy (kJ mol⁻¹): -641.819503461060
Reaction Coordinates:	Frequencies (cm⁻¹):
6 1.228989 0.036844 -0.016691 6 -0.081692 -0.623835 -0.423002 8 -0.677064 -1.303157 0.463605 8 -1.989265 -1.630824 -0.052192 1 -0.219375 -0.847004 -1.473984 9 2.226195 -0.857648 -0.132697 9 1.206699 0.483418 1.240251 9 1.494590 1.062726 -0.834186 6 -1.732897 1.659385 0.402170 8 -1.479661 0.761487 -0.688621 1 -2.046041 -0.080869 -0.548528 1 -1.570998 1.168913 1.362818 1 -2.766982 1.999803 0.343730 1 -1.062445 2.508280 0.298454	-243.0166, 36.9354, 105.3890, 136.5967, 150.0916, 183.1191, 202.5561, 273.7948, 356.5155, 382.0647, 434.5178, 510.9263, 556.1704, 626.3829, 699.0313, 842.7075, 874.2598, 972.1449, 1093.6682, 1098.0531, 1156.7695, 1179.4347, 1180.7510, 1231.7325, 1265.7809, 1339.2162, 1464.9060, 1477.8580, 1495.2058, 1509.6940, 1543.6809, 2693.1227, 3032.9744, 3096.4649, 3134.2897, 3170.8923



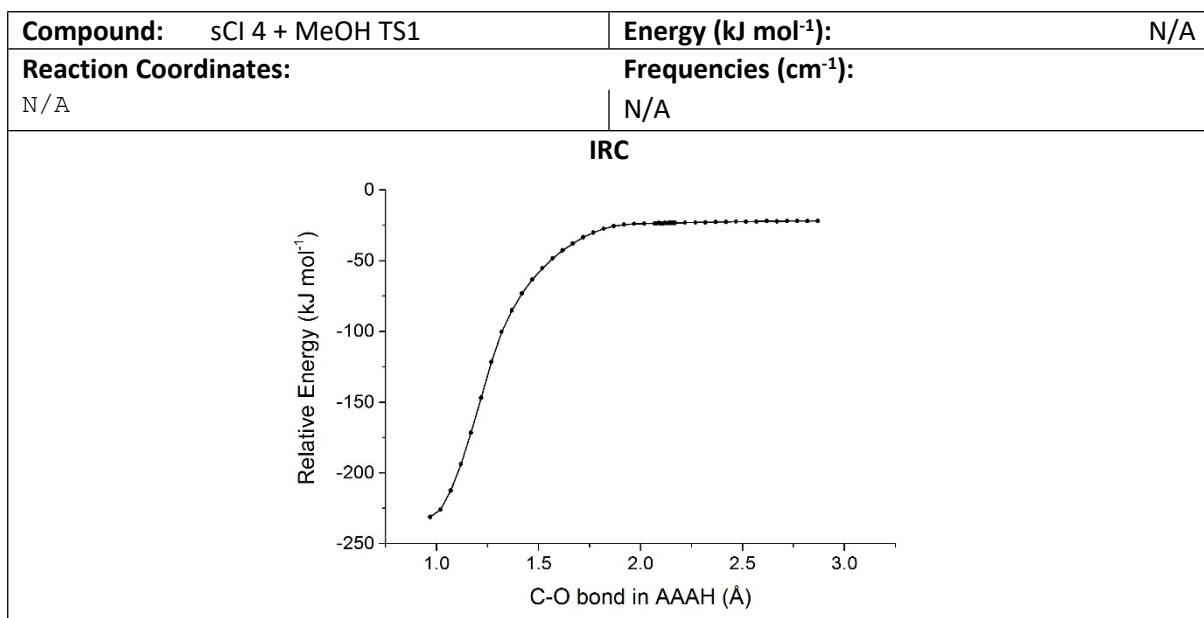
Compound: sCl 3 + MeOH Pr1	Energy (kJ mol⁻¹): -641.900575600613
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -1.077346 -0.234150 -0.016420 6 0.397023 -0.161881 -0.487208 8 1.120044 -0.864566 0.498306 8 2.346263 -1.329349 -0.110191 1 0.458794 -0.690367 -1.437924 9 -1.840623 0.563469 -0.781733 9 -1.541369 -1.486609 -0.135968 9 -1.241862 0.145906 1.262837 6 0.843658 2.057974 0.355278 8 0.867172 1.118444 -0.729322 1 2.958803 -0.610424 0.104143 1 -0.166490 2.428275 0.531427 1 1.236148 1.617870 1.271989 1 1.479588 2.881851 0.043896	61.1974, 99.9853, 120.8446, 175.3253, 192.0609, 226.6944, 232.3815, 305.6868, 353.7993, 357.0616, 405.5817, 529.1777, 562.7402, 651.7204, 731.2299, 869.2845, 929.3750, 956.3906, 1042.2889, 1126.8783, 1165.2849, 1170.3066, 1178.5134, 1221.2817, 1262.6593, 1355.1356, 1386.0573, 1406.5670, 1476.3858, 1500.9000, 1509.1502, 3034.6652, 3094.0915, 3100.0399, 3134.7282, 3734.0248

Compound: sCl 3 + MeOH TS2	Energy (kJ mol⁻¹): -641.817837970547
Reaction Coordinates:	Frequencies (cm⁻¹):
6 1.246729 -0.152999 -0.014850 6 -0.087939 0.463953 -0.409750 8 -0.388112 1.568790 0.119735 8 -1.780107 1.835183 -0.174680 1 -0.512401 0.177832 -1.365617 9 1.218922 -1.471246 -0.265562 9 2.222144 0.388616 -0.768270 9 1.542659 0.038787 1.265197 6 -2.160149 -1.627810 -0.199932 8 -1.501347 -0.588883 0.526825 1 -1.989149 0.295222 0.387213 1 -1.571607 -2.535336 -0.094325 1 -3.153390 -1.786818 0.221191 1 -2.262298 -1.375899 -1.258594	-225.5585, 40.1970, 100.8561, 103.6552, 163.1262, 184.2740, 200.7893, 274.4572, 346.5777, 380.2987, 428.4281, 494.4856, 553.9362, 609.1866, 696.1118, 844.6004, 876.1550, 1000.2079, 1049.8813, 1073.6859, 1149.3181, 1173.9607, 1179.1282, 1204.5798, 1277.1445, 1335.9007, 1468.1314, 1485.2395, 1496.1987, 1511.5663, 1570.4721, 2775.1052, 3017.4396, 3080.9557, 3130.7253, 3154.6612



Compound: sCl 3 + MeOH Pr2	Energy (kJ mol⁻¹): -641.900995173475
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -1.140921 -0.070875 -0.025442 6 0.390369 -0.091534 -0.230218 8 0.795574 -1.359801 0.194497 8 2.159133 -1.540207 -0.248752 1 0.593059 0.037779 -1.298686 9 -1.746540 -0.979668 -0.804939 9 -1.491807 -0.301256 1.242750 9 -1.613723 1.144187 -0.373417 6 1.430951 2.061296 -0.178662 8 1.007046 0.896056 0.530933 1 2.647360 -1.258947 0.539414 1 2.015439 2.648362 0.524422 1 2.056613 1.789365 -1.032606 1 0.579747 2.652358 -0.519588	57.8629, 68.4967, 113.4587, 145.8866, 174.8856, 236.2027, .0564, 318.4633, 344.6182, 369.3135, 416.0334, 523.7378, 563.2944, 600.2046, 710.5586, 874.5574, 953.2717, 1005.4349, 1096.5989, 1118.0767, 1151.8004, 1169.1534, 1177.0464, 1226.9317, 1272.1301, 1359.1766, 1388.8027, 1401.1096, 1480.1900, 1494.4337, 1510.4290, 3013.4252, 3022.8458, 3078.2457, 3132.4287, 3731.7653

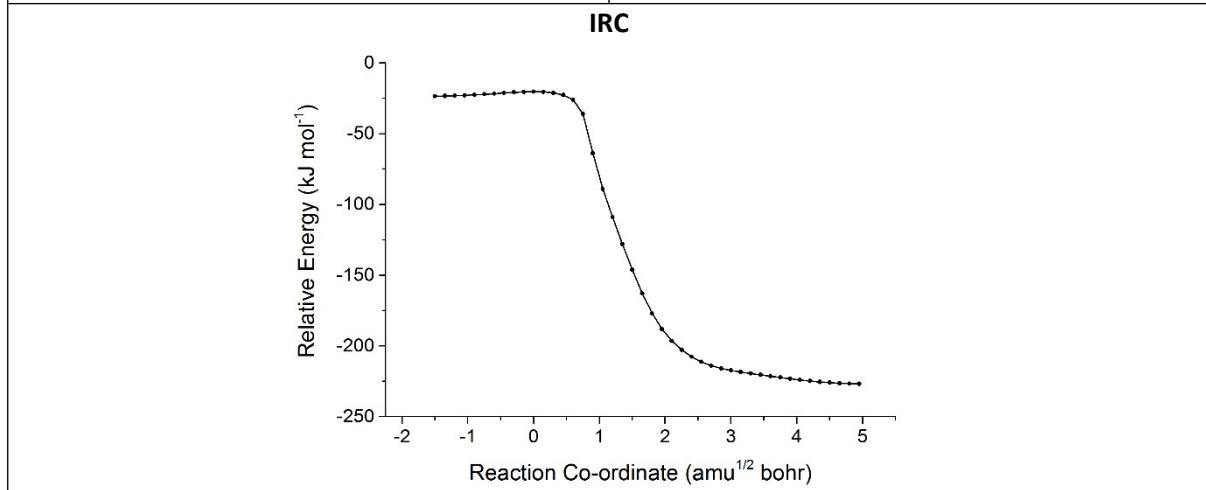
Compound: sCl 4 + MeOH PRC1	Energy (kJ mol⁻¹): -740.991754000359
Reaction Coordinates:	Frequencies (cm⁻¹):
6 1.345909 -0.507442 -0.044013 6 0.625157 0.851684 -0.146677 8 -0.223135 1.293586 0.635300 8 -0.649103 0.469357 1.681353 9 0.980931 1.633422 -1.120889 9 2.043850 -0.548257 1.090717 9 0.492606 -1.514647 -0.088400 9 2.194596 -0.607178 -1.072000 8 -2.380149 -0.513216 -0.386899 6 -3.780639 -0.285373 -0.442830 1 -4.331255 -0.968623 0.211071 1 -4.097218 -0.463621 -1.469291 1 -4.040672 0.743369 -0.173975 1 -2.062164 -0.332229 0.510432	13.4464, 24.6617, 45.5949, 55.0213, 91.4055, 120.5258, 153.4675, 186.3622, 223.7360, 286.8126, 294.2847, 363.6197, 481.8579, 499.8008, 579.4303, 601.6396, 627.1066, 673.1296, 778.5503, 913.7607, 1056.2770, 1114.3121, 1156.3328, 1172.9210, 1189.8948, 1224.7012, 1395.9169, 1416.2135, 1477.3762, 1497.2475, 1511.3971, 1608.5151, 2989.7309, 3033.1589, 3096.4651, 3665.396



Compound: sCl 4 + MeOH Pr1	Energy (kJ mol⁻¹):	-741.086414234267
Reaction Coordinates:	Frequencies (cm⁻¹):	
6 -1.143419 -0.203351 -0.157986 6 0.355218 -0.121118 0.283497 8 0.716882 1.129692 0.829131 8 0.469736 2.151148 -0.157813 9 0.498577 -0.935162 1.372468 9 -1.922991 0.388266 0.751186 9 -1.356176 0.367116 -1.344599 9 -1.504495 -1.492389 -0.247342 6 2.503829 -0.784249 -0.483154 8 1.114227 -0.524834 -0.762691 1 1.325266 2.209555 -0.608262 1 2.606413 -1.691707 0.108214 1 2.972764 -0.916680 -1.452615 1 2.960785 0.052604 0.044084	50.7280, 91.7135, 115.4661, 143.9077, 175.5203, 189.7637, 228.0648, 279.0368, 301.0774, 343.7678, 377.1871, 404.5267, 490.8650, 534.0094, 578.3152, 651.8784, 719.7240, 780.8109, 976.3218, 985.8803, 1069.1732, 1120.5552, 1156.4130, 1178.4296, 1186.2852, 1208.0363, 1225.0568, 1353.0381, 1398.3361, 1478.0025, 1493.4508, 1511.1965, 3047.7620, 3118.8728, 3154.7933, 3731.0958	

Compound: sCl 4 + MeOH PRC2	Energy (kJ mol⁻¹): -740.993837410730
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -0.929081 -0.620812 -0.234927 6 -0.496165 0.618695 0.577499 8 -0.182377 1.733354 0.147822 8 0.153174 1.822995 -1.219691 9 -0.703493 0.531270 1.857480 9 -2.153690 -0.360273 -0.719156 9 -0.130295 -0.942005 -1.232427 9 -1.017064 -1.660041 0.600950 8 1.994666 0.171217 0.315203 6 2.917818 -0.871055 0.009075 1 1.882285 0.743176 -0.459932 1 2.565636 -1.494846 -0.815822 1 3.014127 -1.486422 0.900852 1 3.899690 -0.463959 -0.243262	8.7072, 46.5863, 71.4762, 97.4421, 106.1329, 151.1997, 166.6269, 198.8084, 241.5571, 290.2502, 301.1464, 367.8671, 479.2797, 504.8591, 573.9102, 587.2462, 621.5557, 661.9547, 776.3573, 904.1488, 1047.7881, 1118.3302, 1138.6682, 1173.3138, 1174.2899, 1238.6741, 1384.8004, 1408.7191, 1477.5542, 1498.1583, 1510.5444, 1615.5861, 3010.9504, 3063.2416, 3114.2077, 3671.1396

Compound: sCl 4 + MeOH TS2	Energy (kJ mol⁻¹): -740.993170960665
Reaction Coordinates:	Frequencies (cm⁻¹):
6 0.912032 -0.498276 -0.251751 6 0.224113 0.640651 0.542940 8 -0.042847 1.794029 0.143502 8 -0.657774 1.759785 -1.178840 9 0.482013 0.543914 1.818389 9 0.836580 -1.628139 0.462242 9 0.429997 -0.723657 -1.457506 9 2.206117 -0.152187 -0.357672 6 -2.312308 -1.258884 0.017758 8 -1.725805 0.002627 0.354061 1 -2.151755 -1.932718 0.855253 1 -3.384317 -1.121322 -0.131474 1 -1.872395 -1.683438 -0.884327 1 -1.715513 0.625624 -0.431989	-139.5049, 45.0230, 110.9201, 124.5856, 127.9697, 211.7240, 234.6986, 259.1681, 269.3948, 299.0881, 353.7192, 373.6648, 477.4372, 520.0065, 591.1868, 608.4627, 656.8282, 774.8334, 887.3562, 919.3007, 1037.4115, 1130.0942, 1166.9577, 1176.9158, 1178.2235, 1243.7823, 1350.6568, 1454.6869, 1476.6927, 1497.6563, 1508.5865, 1591.4098, 3027.0633, 3066.5636, 3095.9547, 3132.7609



Compound: sCl 4 + MeOH Pr2	Energy (kJ mol⁻¹): -741.084841781096
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -0.344807 -0.941573 -0.121459 6 0.112414 0.462115 0.425148 8 -0.961502 1.339061 0.593550 8 -1.542744 1.619546 -0.695240 9 0.530358 0.245318 1.708747 9 -1.503957 -1.307551 0.423963 9 -0.478617 -0.930906 -1.452747 9 0.574092 -1.877788 0.186432 6 2.314643 0.392960 -0.558851 8 1.067317 1.074188 -0.322164 1 -0.984830 2.351446 -1.000773 1 2.185348 -0.422482 -1.268434 1 2.731827 0.020531 0.374593 1 2.972709 1.145487 -0.981134	53.5293, 94.1894, 117.9877, 172.1974, 184.4551, 227.5704, 251.8725, 293.5800, 311.8578, 346.8714, 385.0132, 404.2358, 495.8009, 535.1331, 590.8703, 649.2973, 667.2216, 763.0140, 973.6865, 1021.0925, 1090.3159, 1116.6744, 1163.8665, 1176.1525, 1190.3010, 1201.4369, 1229.9396, 1320.1716, 1407.5733, 1485.8850, 1500.9230, 1505.6670, 3051.1521, 3125.5750, 3152.1045, 3720.6253

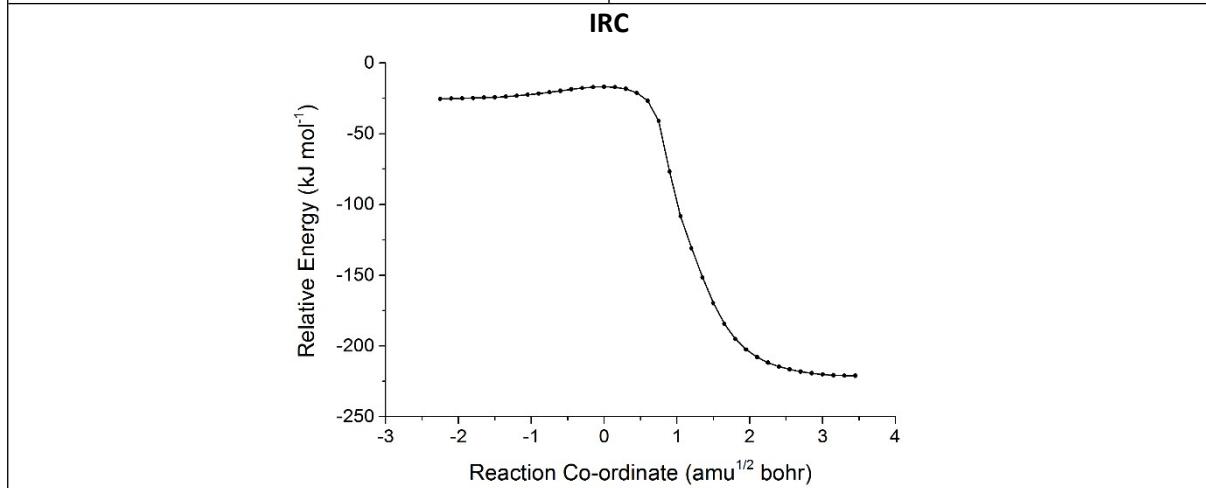
Compound: sCl 5 + MeOH PRC1	Energy (kJ mol⁻¹): -740.999159807260
Reaction Coordinates:	Frequencies (cm⁻¹):
6 1.490530 -0.385650 -0.110808 6 0.352908 0.588183 0.182951 8 -0.223060 1.173232 -0.746175 8 -1.277295 2.020177 -0.431860 9 0.090667 0.808213 1.421679 9 2.631497 0.105665 0.394281 9 1.631711 -0.559880 -1.418454 9 1.242669 -1.558110 0.473031 6 -3.154391 -1.296185 -0.266739 8 -2.245899 -0.501111 0.485391 1 -2.363558 0.431126 0.246002 1 -3.008124 -1.176368 -1.344319 1 -4.193753 -1.058374 -0.023763 1 -2.967710 -2.335846 -0.004026	23.4150, 30.2598, 55.7144, 93.4399, 96.9298, 128.3982, 163.7270, 180.7896, 193.8496, 298.6344, 348.4901, 371.8411, 406.8468, 513.5737, 576.3078, 606.4957, 668.9614, 728.7800, 858.6425, 879.7724, 1052.8820, 1112.2481, 1153.0375, 1174.2597, 1175.2901, 1231.5187, 1405.4956, 1405.7647, 1477.6171, 1497.5943, 1511.8428, 1627.0351, 2996.8581, 3043.5378, 3102.7127, 3663.0055

Compound: sCl 5 + MeOH TS1	Energy (kJ mol⁻¹): -740.997867488158
Reaction Coordinates:	Frequencies (cm⁻¹):
6 -1.259490 0.148992 -0.102836 6 0.059196 -0.610304 0.115493 8 0.678664 -1.019101 -0.894231 8 2.007574 -1.470926 -0.516689 9 0.126921 -1.208161 1.270269 9 -2.261522 -0.733491 -0.215233 9 -1.202728 0.879234 -1.214396 9 -1.509496 0.945964 0.934208 6 1.711322 1.877478 -0.137315 8 1.459590 0.746031 0.708515 1 2.061477 -0.003075 0.413809 1 1.599694 1.616308 -1.190411 1 2.726097 2.234749 0.039112 1 1.001370 2.655080 0.131037	-205.1661, 38.2984, 102.6548, 121.0210, 144.8702, 183.0251, 193.0694, 249.3440, 283.4880, 313.9283, 365.6897, 394.5012, 438.0071, 518.7592, 577.9844, 655.4661, 716.1101, 804.1347, 841.3071, 1005.0264, 1037.8971, 1137.6091, 1170.9206, 1179.4643, 1192.0744, 1221.4657, 1363.3945, 1464.4646, 1474.2037, 1496.0565, 1509.8960, 1567.7328, 3017.9730, 3045.0308, 3096.6112, 3135.6940
IRC	
<p>Relative Energy (kJ mol⁻¹)</p> <p>Reaction Co-ordinate (amu^{1/2} bohr)</p>	

Compound: sCl 5 + MeOH Pr1	Energy (kJ mol⁻¹): -741.083786682520
Reaction Coordinates:	Frequencies (cm⁻¹):
6 1.162766 -0.094593 -0.145678 6 -0.336932 -0.199013 0.310960 8 -0.989709 -0.731512 -0.802437 8 -2.349799 -1.046730 -0.429102 9 -0.363652 -1.069577 1.347633 9 1.874669 0.510363 0.811497 9 1.684982 -1.300826 -0.366239 9 1.282160 0.628147 -1.273515 6 -0.966001 2.093138 -0.078952 8 -0.862075 0.962079 0.798518 1 -2.836987 -0.340649 -0.878001 1 -0.005667 2.597393 -0.179178 1 -1.327022 1.802088 -1.065158 1 -1.680096 2.760303 0.394142	49.3012, 91.7056, 126.2046, 156.2736, 168.1687, 187.3095, 219.5661, 279.4858, 298.1538, 340.6490, 375.8051, 386.4420, 506.4326, 535.9658, 569.5700, 631.1461, 731.8482, 792.7966, 939.1143, 1040.9147, 1064.8271, 1121.6918, 1172.9430, 1174.9616, 1188.9573, 1201.2160, 1226.1309, 1319.3342, 1396.8534, 1481.1796, 1499.2622, 1511.3160, 3040.3468, 3106.9472, 3143.4978, 3740.4493

Compound: sCl 5 + MeOH PRC2	Energy (kJ mol⁻¹): -740.999519026160
Reaction Coordinates:	Frequencies (cm⁻¹):
6 1.478552 -0.363963 -0.107978 6 0.302930 0.526805 0.282388 8 -0.096395 1.409091 -0.489724 8 -1.199076 2.162088 -0.104616 9 -0.174435 0.358706 1.466674 9 1.147628 -1.644957 0.055155 9 1.832907 -0.149597 -1.366851 9 2.519460 -0.092485 0.695326 8 -2.108488 -0.523532 -0.567881 6 -3.143139 -1.265147 0.067307 1 -3.232368 -1.014743 1.128483 1 -4.109511 -1.101912 -0.417281 1 -2.885668 -2.318824 -0.021184 1 -2.300880 0.423135 -0.495280	27.2164, 29.1571, 44.1111, 98.3915, 112.5842, 130.8381, 155.6303, 186.9520, 189.4560, 296.7684, 352.8434, 370.3431, 406.8896, 513.4863, 563.0249, 576.8378, 667.3351, 727.7428, 855.7292, 873.3712, 1053.6440, 1116.0326, 1147.1925, 1173.5514, 1174.1936, 1235.5851, 1392.7941, 1416.8320, 1478.0135, 1498.0017, 1511.2876, 1628.5936, 2998.5418, 3046.4055, 3105.3834, 3683.5289

Compound: sCl 5 + MeOH TS2	Energy (kJ mol⁻¹): -740.997917500017
Reaction Coordinates:	Frequencies (cm⁻¹):
6 1.307922 0.174914 -0.088889 6 -0.030787 -0.497853 0.252216 8 -0.350067 -1.530668 -0.371599 8 -1.745683 -1.865783 -0.135037 9 -0.452090 -0.240559 1.461252 9 1.258492 1.470852 0.227443 9 2.282346 -0.398716 0.633781 9 1.590182 0.043528 -1.376615 8 -1.427798 0.665981 -0.686305 6 -2.123589 1.697954 0.021513 1 -1.953859 -0.183444 -0.649350 1 -3.062137 1.912886 -0.490910 1 -2.331973 1.403900 1.050571 1 -1.495299 2.584383 0.011431	-197.6026, 42.9807, 90.8632, 104.7350, 158.1694, 177.4376, 189.3850, 254.1702, 291.5018, 322.0428, 363.0570, 384.2663, 427.2196, 511.8214, 577.6791, 650.3933, 714.1091, 800.7403, 839.3112, 931.2749, 1039.6013, 1135.8987, 1168.9848, 1175.3687, 1181.7913, 1234.0324, 1363.7070, 1458.9705, 1475.8815, 1496.5038, 1509.9165, 1590.7369, 3030.0270, 3094.6498, 3118.1393, 3138.0592



Compound: sCl 5 + MeOH Pr2	Energy (kJ mol⁻¹): -741.087346378048
Reaction Coordinates:	Frequencies (cm⁻¹):
6 1.193125 -0.130966 -0.075831 6 -0.360430 -0.102593 0.125412 8 -0.775470 -1.369380 -0.236340 8 -2.208064 -1.455681 -0.094286 9 -0.581278 0.131016 1.458852 9 1.767040 -1.081280 0.664166 9 1.500076 -0.340829 -1.358925 9 1.711964 1.054975 0.288796 6 -1.137958 2.174483 -0.141075 8 -0.976074 0.836339 -0.650734 1 -2.502510 -1.133336 -0.960161 1 -1.617729 2.721975 -0.946229 1 -1.775055 2.172851 0.740006 1 -0.176486 2.627805 0.090227	44.9757, 73.5984, 126.1443, 143.4334, 171.5709, 224.3379, 239.8204, 283.2250, 310.2977, 332.6826, 377.7186, 389.2323, 504.7051, 544.3302, 584.7558, 623.1479, 729.8109, 784.1960, 941.3039, 1035.3797, 1067.2085, 1104.9170, 1170.5624, 1179.6748, 1184.6668, 1202.5637, 1226.4720, 1325.3801, 1408.3648, 1484.1721, 1496.9524, 1509.2122, 3057.9428, 3133.2275, 3152.2286, 3722.1372

S9. Example of a Mesmer file

sCl 1 + HNO₃

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<?xmlstylesheet type='text/xsl' href='../../mesmer1.xsl' media='screen'?>
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  <atom id="a3" elementType="O" spinMultiplicity="2" x3="1.179333" y3="0.194958" z3="0.000000" />

  <atom id="a4" elementType="H" x3="-1.021445" y3="1.284352" z3="-0.000000" />

  <atom id="a5" elementType="H" x3="-1.976903" y3="-0.382830" z3="0.000002" />

</atomArray>

<bondArray>

  <bond atomRefs2="a1 a2" order="1" />

  <bond atomRefs2="a1 a4" order="1" />

  <bond atomRefs2="a1 a5" order="1" />

  <bond atomRefs2="a2 a3" order="1" />

</bondArray>

<propertyList>

<property title="program">
  <scalar>Gaussian 09, Revision D.01</scalar>
</property>

<property title="basis">
  <scalar>Aug-CC-pVTZ (5D, 7F)</scalar>
</property>

<property title="method">
  <scalar>B3LYP</scalar>
</property>

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```

<property title="File Format">
  <scalar>g03</scalar>
</property>

<property title="Energy" dictRef="me:ZPE">
  <scalar units="kJ/mol" convention="computational"
zeroPointVibEnergyAdded="true">0.00</scalar>
</property>

<property title="Vibrational Frequencies" dictRef="me:vibFreqs">
  <array units="cm-1">527.76 673.76 912.85 951.07 1242.06 1402.00 1543.54 3118.17
3267.96</array>
</property>

<property title="Rotational Constants" dictRef="me:rotConsts">
  <array units="cm-1">2.693 0.413 0.358</array>
</property>

<property title="Symmetry Number" dictRef="me:symmetryNumber">
  <scalar>1</scalar>
</property>

<property dictRef="me:frequenciesScaleFactor" default="true">
  <scalar>1</scalar>
</property>
</propertyList>

<me:DOSCMETHOD xsi:type="QMRotors" />
</molecule>

<molecule id="HNO3" spinMultiplicity="1" default="true">
<atomArray>
  <atom id="a1" elementType="N" x3="-0.152562" y3="0.032028" z3="-0.000002" />
  <atom id="a2" elementType="O" x3="1.152087" y3="-0.513293" z3="0.000000" />
  <atom id="a3" elementType="H" x3="1.719946" y3="0.275046" z3="0.000002" />
  <atom id="a4" elementType="O" x3="-0.214564" y3="1.238054" z3="0.000001" />
  <atom id="a5" elementType="O" x3="-1.019025" y3="-0.787166" z3="0.000001" />

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</atomArray>

<bondArray>

  <bond atomRefs2="a1 a2" order="1" />
  <bond atomRefs2="a1 a4" order="2" />
  <bond atomRefs2="a1 a5" order="2" />
  <bond atomRefs2="a2 a3" order="1" />

</bondArray>

<propertyList>

  <property title="program">
    <scalar>Gaussian 09, Revision D.01</scalar>
  </property>

  <property title="basis">
    <scalar>Aug-CC-pVTZ (5D, 7F)</scalar>
  </property>

  <property title="method">
    <scalar>B3LYP</scalar>
  </property>

  <property title="File Format">
    <scalar>g03</scalar>
  </property>

  <property title="Energy" dictRef="me:ZPE">
    <scalar units="kJ/mol" convention="computational"
zeroPointVibEnergyAdded="true">0.00</scalar>
  </property>

  <property title="Vibrational Frequencies" dictRef="me:vibFreqs">
    <array units="cm-1">479.28 584.84 646.82 783.11 896.08 1315.31 1341.55 1744.38
3712.10</array>
  </property>

  <property title="Rotational Constants" dictRef="me:rotConsts">
    <array units="cm-1">0.436 0.402 0.209</array>
  </property>

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</property>

<property title="Symmetry Number" dictRef="me:symmetryNumber">
  <scalar>1</scalar>
</property>

<property dictRef="me:frequenciesScaleFactor" default="true">
  <scalar>1</scalar>
</property>

</propertyList>

<me:DOSCMETHOD xsi:type="QMRotors" />

</molecule>

<molecule id="PRC" spinMultiplicity="1" default="true">

  <atomArray>

    <atom id="a1" elementType="N" x3="1.496914" y3="-0.134725" z3="-0.004950" />
    <atom id="a2" elementType="O" x3="0.667866" y3="-1.033940" z3="0.097505" />
    <atom id="a3" elementType="O" x3="2.688525" y3="-0.258885" z3="-0.068858" />
    <atom id="a4" elementType="O" x3="1.032740" y3="1.140905" z3="-0.058034" />
    <atom id="a5" elementType="H" x3="0.016643" y3="1.094684" z3="0.054113" />
    <atom id="a6" elementType="C" spinMultiplicity="2" x3="-2.071887" y3="-1.012931"
z3="0.121133" />

    <atom id="a7" elementType="H" x3="-2.497662" y3="-1.823013" z3="-0.458347" />
    <atom id="a8" elementType="H" x3="-1.657407" y3="-1.119954" z3="1.114924" />
    <atom id="a9" elementType="O" x3="-2.091387" y3="0.101106" z3="-0.429087" />
    <atom id="a10" elementType="O" spinMultiplicity="2" x3="-1.536325" y3="1.159431"
z3="0.283118" />

  </atomArray>

  <bondArray>

    <bond atomRefs2="a7 a6" order="1" />
    <bond atomRefs2="a9 a6" order="1" />
    <bond atomRefs2="a9 a10" order="1" />
    <bond atomRefs2="a3 a1" order="2" />

  </bondArray>

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<bond atomRefs2="a4 a1" order="1" />
<bond atomRefs2="a4 a5" order="1" />
<bond atomRefs2="a1 a2" order="2" />
<bond atomRefs2="a6 a8" order="1" />
</bondArray>

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<property title="program">
  <scalar>Gaussian 09, Revision D.01</scalar>
</property>

<property title="basis">
  <scalar>Aug-CC-pVTZ (5D, 7F)</scalar>
</property>

<property title="method">
  <scalar>B3LYP</scalar>
</property>

<property title="File Format">
  <scalar>g03</scalar>
</property>

<property title="Energy" dictRef="me:ZPE">
  <scalar units="kJ/mol" convention="computational" zeroPointVibEnergyAdded="true">-54.38</scalar>
</property>

<property title="Vibrational Frequencies" dictRef="me:vibFreqs">
  <array units="cm-1">51.88 79.16 103.33 140.18 197.95 266.46 512.02 650.00 676.78 699.25
  799.76 841.22 972.46 981.87 1060.98 1235.72 1307.83 1425.90 1483.69 1597.68 1708.48 2730.22
  3121.31 3266.04</array>
</property>

<property title="Rotational Constants" dictRef="me:rotConsts">
  <array units="cm-1">0.198 0.049 0.040</array>
</property>

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<property title="Symmetry Number" dictRef="me:symmetryNumber">
  <scalar>1</scalar>
</property>

<property dictRef="me:sigma" default="true">
  <scalar>5.0</scalar>
</property>

<property dictRef="me:epsilon" default="true">
  <scalar>50.0</scalar>
</property>

<property dictRef="me:frequenciesScaleFactor" default="true">
  <scalar>1</scalar>
</property>

</propertyList>

<me:DOSCMETHOD xsi:type="QMRotors" />
<me:DistributionCalcMethod default="true" name="Boltzmann" />
<me:energyTransferModel name="ExponentialDown" default="true" />
<me:deltaEDown units="cm-1">300.0</me:deltaEDown>
</molecule>

<molecule id="Pr">
  <atomArray>
    <atom id="a1" elementType="C" x3="0.811606" y3="0.963884" z3="0.434921" />
    <atom id="a2" elementType="O" x3="-0.557612" y3="1.030658" z3="-0.030897" />
    <atom id="a3" elementType="N" x3="-1.293667" y3="-0.183308" z3="0.009949" />
    <atom id="a4" elementType="O" x3="-2.436401" y3="-0.043762" z3="-0.299483" />
    <atom id="a5" elementType="O" x3="-0.701594" y3="-1.186614" z3="0.335941" />
    <atom id="a6" elementType="H" x3="1.046451" y3="2.024122" z3="0.518457" />
    <atom id="a7" elementType="H" x3="0.872854" y3="0.446444" z3="1.388182" />
    <atom id="a8" elementType="O" x3="1.680810" y3="0.406987" z3="-0.493329" />
    <atom id="a9" elementType="O" x3="2.126689" y3="-0.896290" z3="-0.049719" />
  </atomArray>
</molecule>

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<atom id="a10" elementType="H" x3="1.371600" y3="-1.458544" z3="-0.285911" />
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<bond atomRefs2="a8 a9" order="1" />
<bond atomRefs2="a8 a1" order="1" />
<bond atomRefs2="a4 a3" order="2" />
<bond atomRefs2="a10 a9" order="1" />
<bond atomRefs2="a2 a3" order="1" />
<bond atomRefs2="a2 a1" order="1" />
<bond atomRefs2="a3 a5" order="2" />
<bond atomRefs2="a1 a6" order="1" />
<bond atomRefs2="a1 a7" order="1" />
</bondArray>
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</property>
<property title="basis">
<scalar>Aug-CC-pVTZ (5D, 7F)</scalar>
</property>
<property title="method">
<scalar>B3LYP</scalar>
</property>
<property title="File Format">
<scalar>g03</scalar>
</property>
<property title="Energy" dictRef="me:ZPE">
<scalar units="kJ/mol" convention="computational" zeroPointVibEnergyAdded="true">-179.86</scalar>
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<property title="Vibrational Frequencies" dictRef="me:vibFreqs">
    <array units="cm-1">76.14 109.04 149.89 274.39 331.02 399.04 504.07 596.45 671.09 778.42
848.25 892.71 929.91 1066.43 1143.07 1299.64 1322.74 1400.83 1417.85 1455.02 1703.96 3076.26
3148.97 3703.26</array>
</property>

<property title="Rotational Constants" dictRef="me:rotConsts">
    <array units="cm-1">0.199 0.062 0.051</array>
</property>

<property title="Symmetry Number" dictRef="me:symmetryNumber">
    <scalar>1</scalar>
</property>
</propertyList>
</molecule>

<molecule id="TS" spinMultiplicity="1" default="true">
    <atomArray>
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        <atom id="a2" elementType="O" x3="0.628369" y3="-1.058966" z3="0.034369" />
        <atom id="a3" elementType="O" x3="2.580436" y3="-0.135427" z3="-0.065503" />
        <atom id="a4" elementType="O" x3="0.846896" y3="1.149168" z3="0.004906" />
        <atom id="a5" elementType="H" x3="-0.251112" y3="1.114260" z3="0.091169" />
        <atom id="a6" elementType="C" spinMultiplicity="2" x3="-1.671765" y3="-1.057011"
z3="0.209054" />
        <atom id="a7" elementType="H" x3="-1.838608" y3="-2.006498" z3="-0.286142" />
        <atom id="a8" elementType="H" x3="-1.395054" y3="-0.967906" z3="1.249688" />
        <atom id="a9" elementType="O" x3="-1.962928" y3="-0.044666" z3="-0.454559" />
        <atom id="a10" elementType="O" spinMultiplicity="2" x3="-1.613108" y3="1.171046"
z3="0.199461" />
    </atomArray>
    <bondArray>
        <bond atomRefs2="a9 a10" order="1" />
        <bond atomRefs2="a9 a6" order="1" />

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<bond atomRefs2="a7 a6" order="1" />
<bond atomRefs2="a3 a1" order="2" />
<bond atomRefs2="a1 a4" order="1" />
<bond atomRefs2="a1 a2" order="2" />
<bond atomRefs2="a4 a5" order="1" />
<bond atomRefs2="a6 a8" order="1" />
</bondArray>
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<scalar>Gaussian 09, Revision D.01</scalar>
</property>
<property title="basis">
<scalar>Aug-CC-pVTZ (5D, 7F)</scalar>
</property>
<property title="method">
<scalar>B3LYP</scalar>
</property>
<property title="File Format">
<scalar>g03</scalar>
</property>
<property title="Energy" dictRef="me:ZPE">
<scalar units="kJ/mol" convention="computational" zeroPointVibEnergyAdded="true">-48.32</scalar>
</property>
<property title="Vibrational Frequencies" dictRef="me:vibFreqs">
<array units="cm-1">39.67 92.49 198.57 324.04 328.85 500.71 660.32 702.16 736.22 801.49
819.39 1019.93 1047.07 1133.66 1160.20 1235.81 1425.02 1489.92 1598.78 1614.23 1818.43
3129.37 3269.89</array>
</property>
<property title="ImaginaryFrequency" dictRef="me:imFreqs">

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<scalar units="cm-1">248.93</scalar>
</property>

<property title="Rotational Constants" dictRef="me:rotConsts">
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</property>

<property title="Symmetry Number" dictRef="me:symmetryNumber">
  <scalar>1</scalar>
</property>

<property dictRef="me:frequenciesScaleFactor" default="true">
  <scalar>1</scalar>
</property>
</propertyList>

<me:DOSCMETHOD xsi:type="QMRotors" />
</molecule>
</moleculeList>

<reactionList>
  <reaction id="R1" reversible="true">
    <reactantList>
      <reactant>
        <molecule ref="Cl 1" role="deficientReactant" />
      </reactant>
      <reactant>
        <molecule ref="HNO3" role="excessReactant" />
      </reactant>
    </reactantList>
    <productList>
      <product>
        <molecule ref="PRC" role="modelled" />
      </product>
    </productList>
  </reaction>
</reactionList>

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    </productList>

    <rateParameters reactionType="arrhenius" reversible="true">
        <A>1.000e-010</A>
        <n>0</n>
        <E>0</E>
    </rateParameters>

    <me:MCRCMethod default="CanonicalRateCoefficient, DefinedSumOfStates,
LandauZenerCrossing, MesmerILT, SimpleBimolecularSink, SimpleILT, MesmerILT, WKBCrossing,
ZhuNakamuraCrossing" name="MesmerILT" />

    <me:excessReactantConc default="true">1.0E16</me:excessReactantConc>

    <me:TInfinity default="true">298</me:TInfinity>

</reaction>

<reaction id="R2">

    <reactantList>
        <reactant>
            <molecule ref="PRC" role="modelled" />
        </reactant>
    </reactantList>

    <productList>
        <product>
            <molecule ref="Pr" role="sink" />
        </product>
    </productList>

    <me:transitionState>
        <molecule ref="TS" role="transitionState" />
    </me:transitionState>

    <me:MCRCMethod name="SimpleRRKM" />

    <me:tunneling>Eckart</me:tunneling>

</reaction>

</reactionList>

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  <me:bathGas>N2</me:bathGas>
  <me:PTs>
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    <me:PTpair units="Torr" P="31" T="295." precision="d" default="true" bathGas="N2" />
    <me:PTpair units="Torr" P="35" T="295." precision="d" default="true" bathGas="N2" />
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    <me:PTpair units="Torr" P="760" T="275." precision="d" default="true" bathGas="N2" />
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    <me:PTpair units="Torr" P="760" T="325." precision="d" default="true" bathGas="N2" />
    <me:PTpair units="Torr" P="760" T="400." precision="d" default="true" bathGas="N2" />
  </me:PTs>
</me:conditions>
<me:modelParameters>
  <me:grainSize units="cm-1">10</me:grainSize>
  <me:energyAboveTheTopHill>30.0</me:energyAboveTheTopHill>
</me:modelParameters>
<me:control>
  <!--<me:calcMethod xsi:type="me:marquardt">
    <me:MarquardtIterations>50</me:MarquardtIterations>
    <me:MarquardtTolerance>0.1</me:MarquardtTolerance>
    <me:testMicroRates Tstep="25" default="true" Tmin="200" Tmax="400" />
  </me:calcMethod>-->
  <me:printSpeciesProfile />
  <me:testRateConstants />
  <!--<me:printGrainBoltzmann />-->
  <me:printGrainedSpeciesProfile />
  <me:eigenvalues>3</me:eigenvalues>

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<!-- <me:hideInactive/> Molecules and reactions with attribute active="false" are not shown-->
<me:diagramEnergyOffset>0</me:diagramEnergyOffset>
<!--Adjusts displayed energies to this values for the lowest species. -->
<me:calcMethod default="true" name="simpleCalc" />
<me:ForceMacroDetailedBalance default="true">true</me:ForceMacroDetailedBalance>
</me:control>
<metadataList xmlns:dc="http://purl.org/dc/elements/1.1/">
  <dc:title>HCHO+HNO3</dc:title>
  <dc:source>sCI 1_HNO3.xml</dc:source>
  <dc:creator>Mesmer v5.2</dc:creator>
  <dc:date>20190405_183821</dc:date>
  <dc:contributor>c1675612</dc:contributor>
</metadataList>
</me:mesmer>
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