

SUPPORTING INFORMATION – 6 pages

Contactless Measurement of Photovoltage in BiVO₄ Photoelectrodes

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Table S1. Semiconductor and junction information.

Semiconductors	n-BiVO ₄			
Band Gap / eV	2.4			
Dielectric Constant	68.0 ^a			
Thickness / μm	1.4			
Conduction Band edge / V vs RHE	0.041			
Valence band edge / V vs RHE	2.44			
E _F (Flatband potential) / V vs RHE	0.10 ^b			

Redox Couples	I ₃ /I ⁻	SO ₄ ²⁻ /SO ₃ ²⁻	O ₂ /H ₂ O ₂	HCF ^{3-/4-}
E ⁰ / V vs NHE	0.54	0.17	0.70	0.36
E ⁰ / V vs RHE (calculated)	0.85 at pH 6.7 (KI ₃ :KI 5:95)	0.17 at pH 9.6	0.70 at pH 4.2	0.70 at pH 5.6
E _F / V vs RHE (experimental)	0.89	0.17	0.80	0.70
Built-in potential (band bending) / V	0.79	0.07	0.70	0.60
Depletion layer width / nm ^c	11	3	10	9

^a from Wang et al. ¹

^b from Kim et al. ²

^c calculated from Poisson equation.

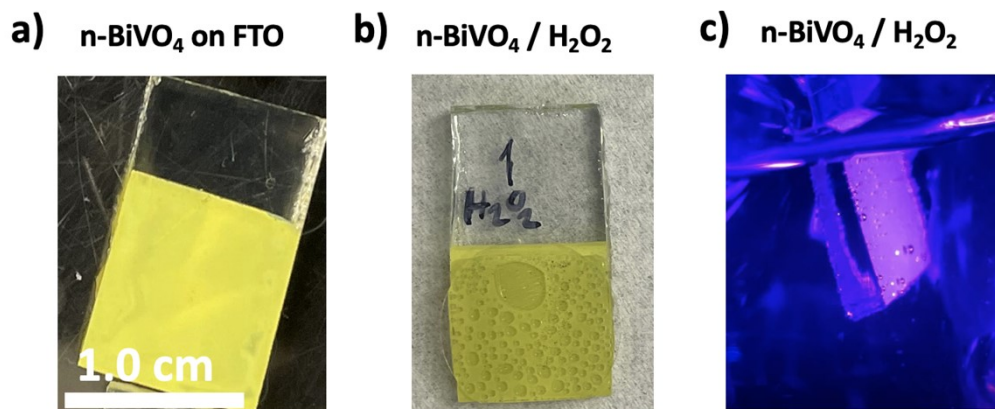


Figure S2. Photos of semiconductor electrodes. a) BiVO₄ on FTO. b) BiVO₄/H₂O₂ after SPV experiment, c) BiVO₄ electrode during short circuit current experiment in aqueous H₂O₂.

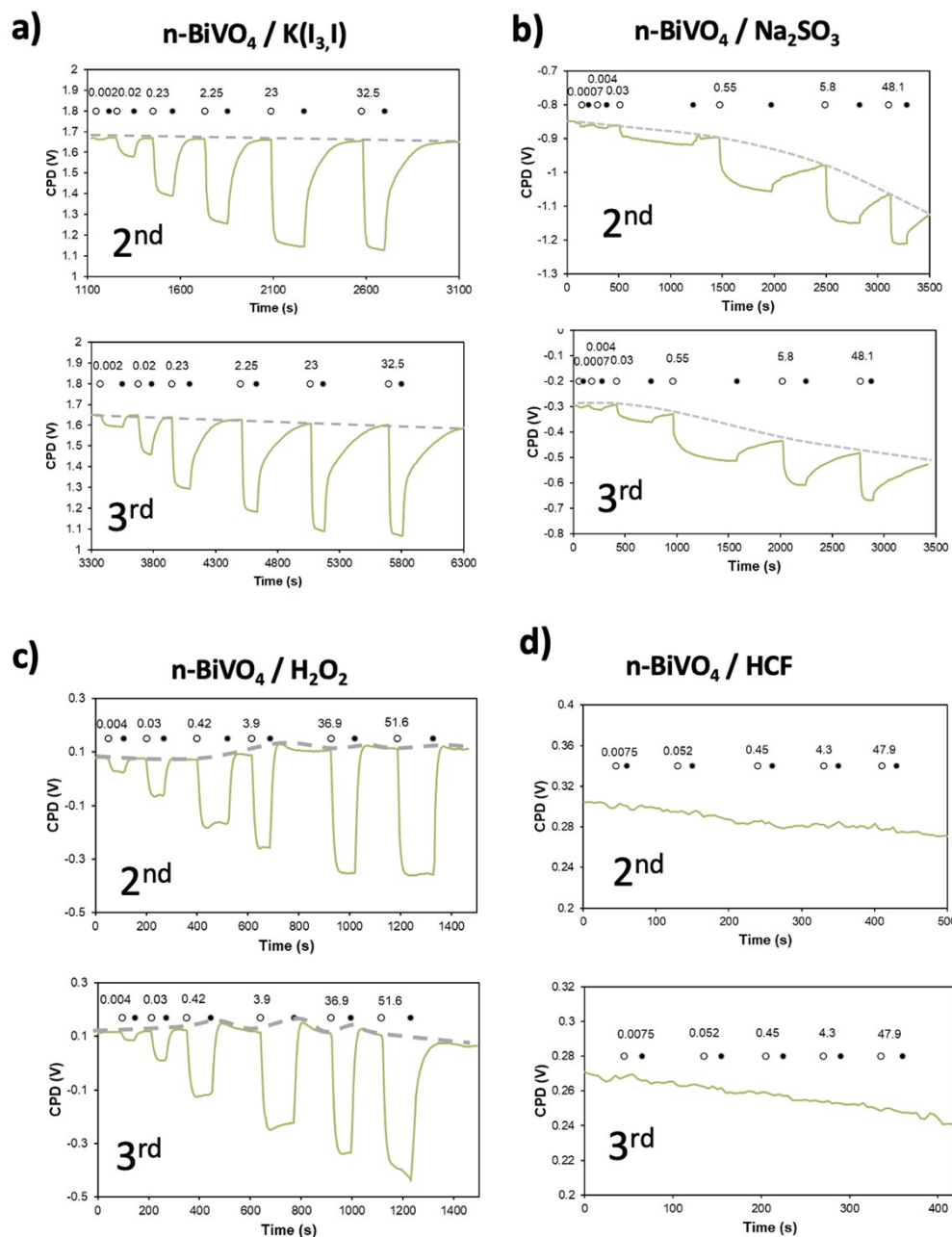


Figure S3. Additional surface photovoltage data under monochromatic illumination of variable irradiance. BiVO₄ film on FTO in contact with aqueous solutions of a) KI/KI₃ (470 nm) b) Na₂SO₃ (400 nm), c) H₂O₂ (400 nm), or d) K_{3/4}[Fe(CN)₆] (HCF) (400 nm). Empty circles are light and filled circles are dark periods.

Table S4. SPV statistics. Data from **Figures 3** and **S2**.

BiVO₄/ K(I₃,I)				BiVO₄/ Na₂SO₃			
Intensity / mW cm ⁻²	average ΔCPD / V	STD / V	STD / %	Intensity / mW cm ⁻²	ΔCPD / V	STD / V	STD / %
0.002	-0.03	0.02	66	0.0007	-0.012	0.006	50
0.02	-0.15	0.06	37	0.004	-0.03	0.02	67
0.23	-0.31	0.03	9	0.03	-0.075	0.01	13
2.25	-0.42	0.02	4	0.55	-0.158	0.02	13
23.0	-0.50	0.01	2	5.8	-0.167	0.006	4
32.5	-0.52	0.01	2	48.1	-0.17	0.01	6
BiVO₄/H₂O₂				BiVO₄/ HCF			
Intensity / mW cm ⁻²	average ΔCPD / V	STD / V	STD / %	Intensity / mW cm ⁻²	ΔCPD / V	STD / V	STD / %
0.004	-0.035	0.01	28	0.0007	0	-	-
0.03	-0.11	0.02	18	0.004	0	-	-
0.42	-0.24	0.01	4	0.03	0	-	-
3.9	-0.36	0.02	5	0.55	0	-	-
36.9	-0.46	0.02	4	5.8	0	-	-
51.6	-0.49	0.01	2	48.1	0	-	-

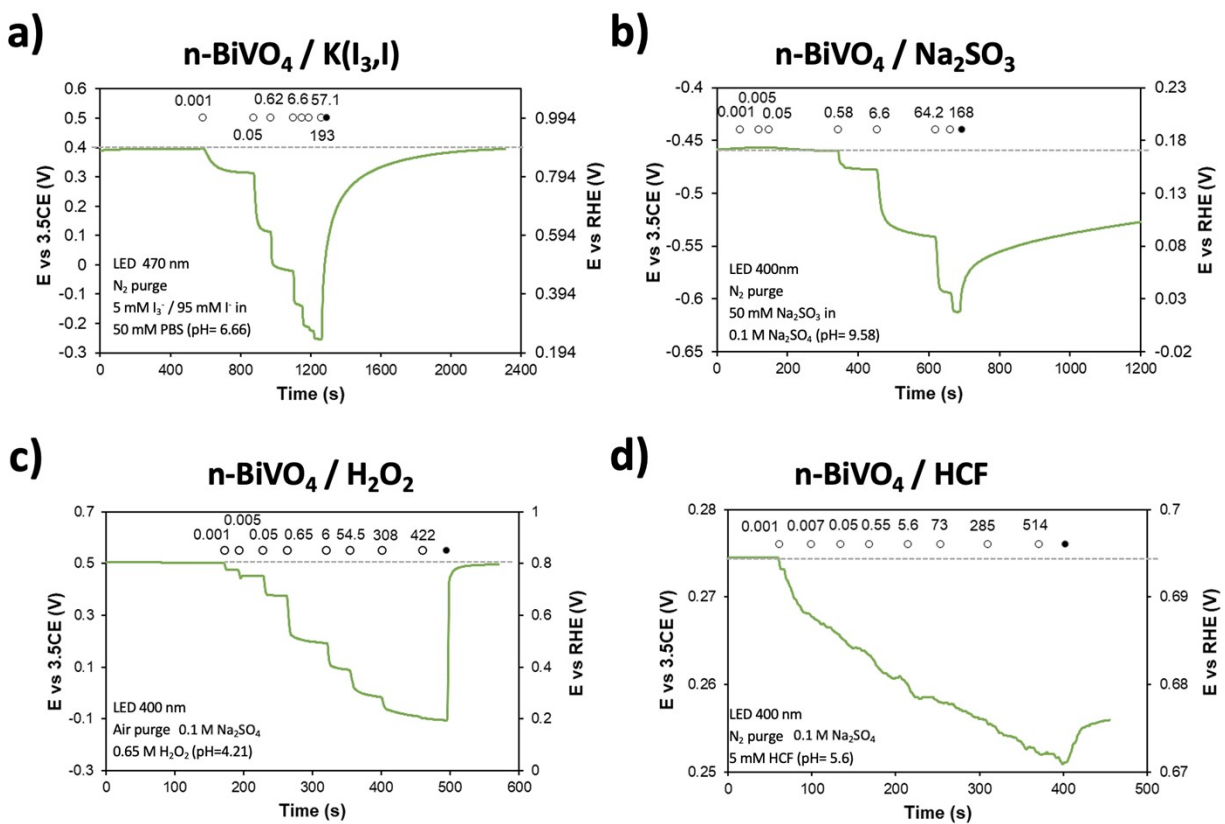


Figure S5. FTO/BiVO₄ Fermi levels from OCP measurements versus calomel electrode in the dark and under monochromatic illumination (irradiances given in mW cm⁻²). For the K(I₃,I) electrolyte a 470 nm LED was used, while all other systems used a 400 nm LED. All electrolytes (except for c) were de-aerated with N₂ gas, except for c. Data in b) was obtained after exposing the BiVO₄ photoelectrode to a light-dark cycle (168 mW cm⁻² of 400 nm) to establish a stable Fermi level in the dark.

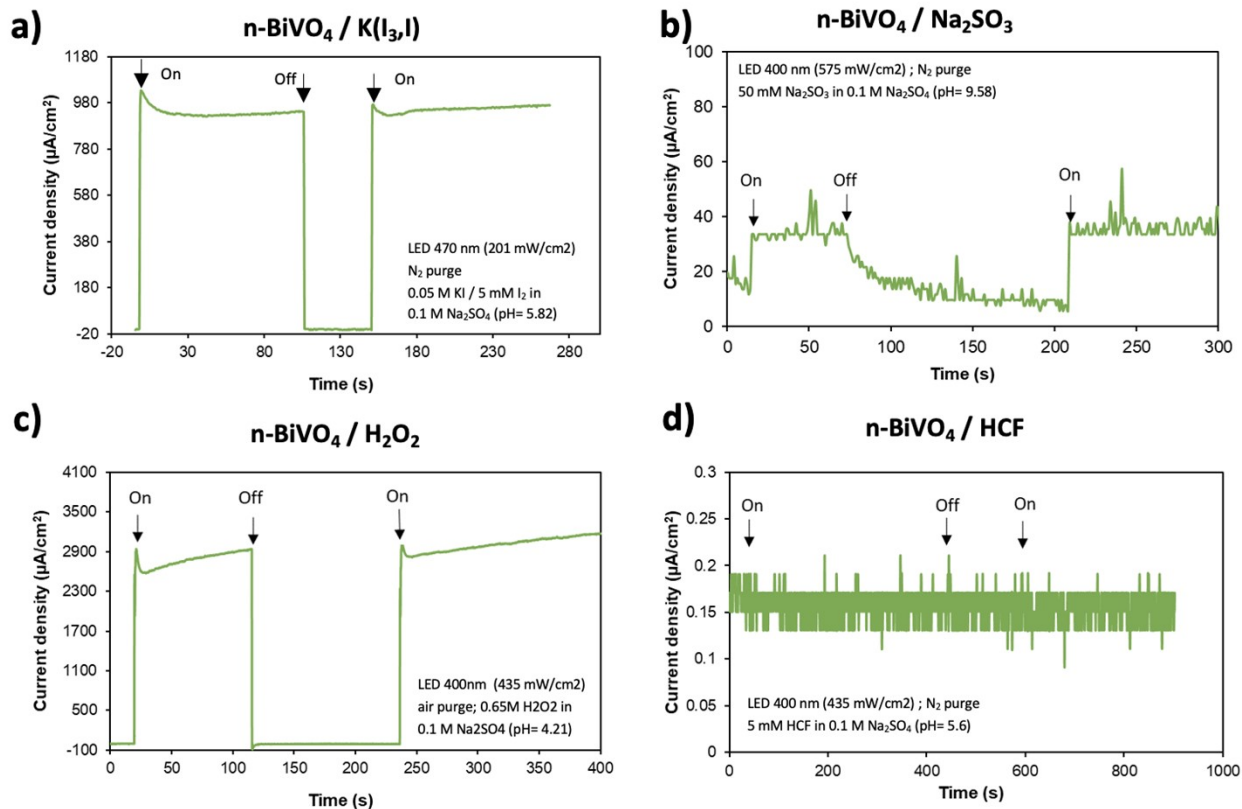


Figure S6. Short circuit photocurrent against Pt counter electrode in aqueous electrolyte. a) n-BiVO₄/K(I₃,I) under LED light (470 nm, 201 mW cm⁻²), b) BiVO₄/ Na₂SO₃ under LED light (400 nm, 575 mW cm⁻²), c) BiVO₄/ H₂O₂ under LED light (400 nm, 435 mW cm⁻²), d) BiVO₄/ HCF under LED light (400 nm, 435 mW cm⁻²). The weak 10 µA cm⁻² current is attributed to reduction of O₂ traces at the counter electrode. Electrolytes were de-aerated with bubbling N₂ gas, except for c), which was recorded in the presence of air.

References

1. Wang, Q. *et al.* Particulate Photocatalyst Sheets Based on Carbon Conductor Layer for Efficient Z-Scheme Pure-Water Splitting at Ambient Pressure. *Journal of the American Chemical Society* **139**, 1675-1683, (2017).
2. Kim, T. W. & Choi, K.-S. Nanoporous BiVO₄ Photoanodes with Dual-Layer Oxygen Evolution Catalysts for Solar Water Splitting. *Science* **343**, 990-994, (2014).