

## Supporting Information

### Hydrogen society: from present to future

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## Supplementary Note #1

Calculation model:

The supply chain of hydrogen is divided into four stages: production, transmission & storage, distribution and refueling. The total cost of hydrogen ( $C_{H_2}$ ) is shown in equation (S1).

$$C_{H_2} = C_p + C_t + C_d + C_r \quad (S1)$$

where  $C_p$  is hydrogen production cost,  $C_t$  is the transmission & storage cost,  $C_d$  is the distribution cost (from the transportation terminal to the refueling station), and  $C_r$  is the refueling cost.

In the four stages of hydrogen production, transmission & storage, distribution and refueling (in the case of sea routine, there are more stages, including production, conditioning, loading, transmission, unloading, reconditioning, distribution and refueling), hydrogen cost at each stage includes initial investment cost and operation and maintenance cost. The initial investment cost is mainly composed of related equipment and infrastructure (hydrogen production plant, hydrogen transport pipeline, etc.). The operation and maintenance cost mainly includes the operation and maintenance cost of the equipment as well as the salary of labors. Specially, for the hydrogen production, it also includes the cost of raw materials (the cost of purchasing electricity, coal, or natural gas) and replacement costs (mainly considering that the life of the electrolytic equipment is shorter than the project life, so the electrolytic equipment will be replaced during the project life cycle). Refueling station is mainly composed of storage system, filling system and control system. The main equipment includes

compressor, hydrogen storage tank, hydrogen filling machine, cooler and sequential control valve group. In our calculations, the scale of refueling station is uniformly set as 700 kg per day, and the capacity of high-pressure hydrogen storage tank is set as one day's reserve to reduce the construction cost of refueling station.

In order to estimate the cost of hydrogen more accurately and rationally, the levelized cost of hydrogen (*LCOH*) is calculated as follows (S2):

$$LCOH = \frac{C_{\text{capital}} \times \frac{i(1+i)^y}{(1+i)^y - 1} + C_{\text{feedstock}} + C_{\text{OM}} + C_{\text{replacement}}}{P} \quad (\text{S2})$$

where  $C_{\text{capital}}$  is the initial investment cost of hydrogen at each stage (USD, \$). For example, the  $C_{\text{capital}}$  of hydrogen production includes the cost of hydrogen production equipment, plant, and the infrastructure construction;  $C_{\text{feedstock}}$  is the raw-material cost of hydrogen production (electricity, coal, or natural gas, \$ per year);  $C_{\text{OM}}$  and  $C_{\text{replacement}}$  are the operation & maintenance cost and replacement cost, respectively;  $P$  is the hydrogen flow rate (kg per year);  $i$  is the discount rate, and  $y$  is the project life (year).

Considering that the life of electrolytic equipment is shorter than that of the project, there will be an additional replacement cost in the process of water electrolysis, which can be calculated by the equation (S3):

$$C_{\text{replacement}} = D_{\text{replacement}} \times N \times \left( \sum_{m=1}^{m=N} \frac{(1 - V_{\text{replacement}})^{m \times A}}{(1+i)^{m \times A}} \right) \times \frac{i(1+i)^y}{(1+i)^y - 1} \quad (\text{S3})$$

where  $D_{\text{replacement}}$  (\$) is the replacement cost of the hydrogen production plant,  $N$  is the times of replacement during a project's life cycle,  $V_{\text{replacement}}$  is the annual change rate of the replacement cost, and  $A$  is the replacement years for the hydrogen-production plants.

For water electrolysis, the feedstock cost is calculated following the equation S4:

$$C_{\text{feedstock}} = B_{\text{elec}} \times d \times \left( \frac{h_d}{\eta} \right) \times \left( \sum_{m=1}^{m=y} \frac{(1+V_{\text{elec}})^m}{(1+i)^m} \right) \times \frac{i(1+i)^y}{(1+i)^y - 1} \quad (\text{S4})$$

where  $h_d$  is the operation hours per day of electrolysis equipment;  $d$  is the operation days per year;  $\eta$  is the system efficiency;  $B_{\text{elec}}$  is the electricity price (\$ kWh<sup>-1</sup>) and  $V_{\text{elec}}$  is the annual change rate of  $B_{\text{elec}}$ .

For coal and natural gas, the feedstock cost is calculated by equation S5:

$$C_{\text{feedstock}} = B_{\text{coal,gas}} \times m_{\text{year}} \times \left( \sum_{m=1}^{m=y} \frac{(1+V_{\text{elec}})^m}{(1+i)^m} \right) \times \frac{i(1+i)^y}{(1+i)^y - 1} \quad (\text{S5})$$

where  $B_{\text{coal,gas}}$  is the price of coal (\$ per ton) or natural gas (\$ per m<sup>3</sup>),  $m_{\text{year}}$  is the raw material demand per year (coal: ton per year, natural gas: m<sup>3</sup> per year).

In this study, we mainly consider two transportation routes, namely land transmission and sea transmission routines. The land transmission routine is designed from Gansu, China to Shanghai, China, in which Gansu and Shanghai are selected as the place of hydrogen production and hydrogen consuming, respectively. The sea transmission is designed from Australia to Japan, where Australia and Japan serve as the place of hydrogen production and hydrogen consuming, respectively. We calculated the levelized cost of hydrogen from production to consuming.

**Table S1.** Detailed parameters of AWE, PEME and SOEC technologies at 2020 and 2050.<sup>1</sup>

Alkaline water electrolyzer			
Parameters	2020	2050	Focus
Voltage (V)	1.4-3.0	< 1.7	Catalysts
Current density (A cm <sup>-2</sup> )	0.2-0.8	> 2.0	Diaphragm
Cell pressure (bar)	< 30	> 70	Diaphragm, frames, cell
Cell temperature (°C)	70-90	> 90	Diaphragm, frames, balance of plant components
Electrode area (cm <sup>2</sup> )	10,000-30,000	30,000	Electrodes
Stack unit size (kW)	1000	10,000	Electrodes
Stack lifetime (h)	60,000	100,000	Electrodes
Time of cold start (min)	< 50	< 30	Design of insulation
H <sub>2</sub> purity (%)	99.9-99.9998	> 99.9999	Diaphragm
Load range (%)	15-100	5-300	Diaphragm
System capital costs (USD kW <sup>-1</sup> )	500-1000	< 200	Balance of plant
Stack capital costs	270	< 100	Electrodes

(USD kW <sup>-1</sup> )			
System electrical efficiency (kWh kg <sup>-1</sup> H <sub>2</sub> )	50-78	< 45	Balance of plant
Stack electrical efficiency (kWh kg <sup>-1</sup> H <sub>2</sub> )	47-66	< 42	Diaphragm, catalysts
Voltage efficiency (%)	50-68	> 70	Catalysts, temperature
Proton exchange membrane electrolyzer			
Parameters	2020	2050	Focus
Voltage (V)	1.4-2.5	< 1.7	Catalysts, membrane
Current density (A cm <sup>-2</sup> )	1-2	4-6	Design, membrane
Cell pressure (bar)	< 30	> 70	Membrane, reversion
Cell temperature (°C)	50-80	80	Effect on durability
Electrode area (cm <sup>2</sup> )	1500	> 10,000	Membrane electrode assembly, porous transport layers
Stack unit size (kW)	1000	10,000	Membrane electrode assembly, porous transport layers

Stack lifetime (h)	50,000-80,000	100,000-120,000	Membrane, catalysts, porous transport layers
Time of cold start (min)	< 20	< 5	Design of insulation
H <sub>2</sub> purity (%)	99.9-99.9999	99.9-99.9999	Membrane
Load range (%)	5-120	5-300	Membrane
System capital costs (USD kW <sup>-1</sup> )	700-1400	< 200	Rectifier, water purification
Stack capital costs (USD kW <sup>-1</sup> )	400	< 100	Membrane electrode assembly, porous transport layers, bipolar plates
System electrical efficiency (kWh kg <sup>-1</sup> H <sub>2</sub> )	50-83	< 45	Balance of plant
Stack electrical efficiency (kWh kg <sup>-1</sup> H <sub>2</sub> )	47-66	< 42	Catalysts, membrane
Voltage efficiency (%)	50-68	> 80	Catalysts
Solid oxide electrolysis cell			
Parameters	2020	2050	Focus
Voltage (V)	1.0-1.5	< 1.48	Catalysts

Current density (A cm <sup>-2</sup> )	0.3-1.0	> 2.0	Electrolyte, electrodes
Cell pressure (bar)	1	> 20	Electrolyte, electrodes
Cell temperature (°C)	700-850	< 600	Electrolyte
Electrode area (cm <sup>2</sup> )	200	500	All
Stack unit size (kW)	5	200	All
Stack lifetime (h)	< 20,000	80,000	All
Time of cold start (min)	> 600	< 300	Design of insulation
H <sub>2</sub> purity (%)	99.9	> 99.9999	Electrolyte, electrodes
Load range (%)	30-125	0-200	Electrolyte, electrodes
System capital costs (USD kW <sup>-1</sup> )	Unknown	< 300	All
Stack capital costs (USD kW <sup>-1</sup> )	> 2000	< 200	Electrolyte, electrodes
System electrical efficiency (kWh kg <sup>-1</sup> H <sub>2</sub> )	40-50	< 40	Balance of plant
Stack electrical efficiency (kWh kg <sup>-1</sup> H <sub>2</sub> )	35-50	< 35	Electrolyte, electrodes
Voltage efficiency (%)	75-85	> 85	Catalysts



**Table S2.** Input parameters of water electrolysis.<sup>2-4</sup>

	AWE	PEME	SOEC	SOEC (waste heat)
Plant lifetime (year)	20	20	20	20
Scale (MW)	1	1	1	1
Discount rate	8%	8%	8%	8%
Efficiency (%)	65	70	61.4	87.6
Stack lifetime (h)	80000	40000	20000	20000
System degradation (%/10000 hour)	1.30	2.70	5.90	5.9
Electricity cost (\$ kWh <sup>-1</sup> )	0.05	0.05	0.05	0.05
Deionized water (\$ kg <sup>-1</sup> )	0.01	0.01	0.01	0.01
Replacement times	1	2	4	4
<b>Capital cost:</b>				
Stock cost (\$ MW <sup>-1</sup> )	340000	420000	520000	520000
Electric heater	--	--	15000	--
<b>Balance of plant (BOP) cost:</b>				
Power supply (\$ MW <sup>-1</sup> )	27556	27556	27556	27556
Separator tank of water (\$ MW <sup>-1</sup> )	10000	10000	10000	10000
Circulation pump (\$ MW <sup>-1</sup> )	5500	5500	5500	5500
Piping (\$ MW <sup>-1</sup> )	7600	7600	7600	7600
Valves (control of water, \$ MW <sup>-1</sup> )	7700	7700	7700	7700
Hydrogen dryer (\$ MW <sup>-1</sup> )	13900	13900	13900	13900
Hydrogen separator (\$ MW <sup>-1</sup> )	5300	5300	5300	5300
Tubing (\$ MW <sup>-1</sup> )	3800	3800	3800	3800
<b>Electric heater:</b>				
Valves (control of hydrogen)	5200	5200	5200	5200
Heat exchanger	9000	9000	9000	9000
Cooling pump	15000	15000	15000	15000
Dry cooler	4000	4000	4000	4000
Valves control	3000	3000	3000	3000
Electric resistance heater	7500	7500	7500	7500
Other BOPs	6000	6000	6000	6000
<b>Operation and maintenance cost:</b>				
Operation and maintenance (%capital)	2	2	2	2
Other operation cost (%capital)	1	1	1	1

**Table S3.** Parameters of natural gas steam reforming.<sup>5</sup>

Parameters	Values
Life	20 years
Discount rate	8%
Annual load (h per year)	7884
Overall efficiency (%)	70-80
Feedstock input (m <sup>3</sup> (stp) h <sup>-1</sup> )	65000
Natural gas costs (\$ GJ <sup>-1</sup> )	5.40
Electricity cost (\$ kWh <sup>-1</sup> )	0.05
Auxiliary electricity input (MW)	5
Net by-products (MW)	140
<b>Capital cost:</b>	
Total capital investment (million \$)	150
Specific total capital investment (million \$ MW <sup>-1</sup> ) <sup>a</sup>	0.333
<b>Operation and maintenance cost:</b>	
Operation & maintenance costs (O&M) (million \$ per year)	7.93
Specific O&M (million \$ MW <sup>-1</sup> )	0.018

<sup>a</sup>Based on the annual hydrogen output of 450 MW for large scale and 3 MW for small scale.

**Table S4.** Parameters of coal gasification.<sup>5</sup>

Parameters	Values
Life	20 years
Discount rate	8%
Annual load (h per year)	7008
Overall efficiency (%)	53
Feedstock input (unit/h)	120 t
Auxiliary electricity input (MW)	36
Net by-products (MW)	43
Feedstock costs (\$ GJ <sup>-1</sup> )	2.10
Electricity cost (\$ kWh <sup>-1</sup> )	0.05
<b>Capital cost:</b>	
Total capital investment (million \$)	375.500
Specific total capital investment (million \$ MW <sup>-1</sup> ) <sup>a</sup>	0.834
<b>Operation &amp; maintenance:</b>	
Operation & maintenance costs (O&M) (million \$ per year <sup>-1</sup> )	17.310
Specific O&M (million \$ MW <sup>-1</sup> )	0.038

<sup>a</sup>Based on the annual hydrogen output of 450 MW for large scale and 3 MW for small scale.

**Table S5.** Calculated regional hydrogen production cost (USD kg<sup>-1</sup> H<sub>2</sub>) for AWE, PEME and SOEC in representative areas worldwide based on the latest electricity prices.<sup>6</sup>

Region	Electricity price (USD kWh <sup>-1</sup> )	AWE (USD kg <sup>-1</sup> H <sub>2</sub> )	PEME (USD kg <sup>-1</sup> H <sub>2</sub> )	SOEC (USD kg <sup>-1</sup> H <sub>2</sub> )
China	0.095	5.979	7.215	9.770
India	0.104	6.423	7.698	10.258
Japan	0.17	9.682	11.240	13.841
South Korea	0.095	5.979	7.215	9.770
Thailand	0.105	6.472	7.752	10.313
Saudi Arabia	0.068	4.645	5.766	8.304
Russia	0.132	7.805	9.201	11.778
France	0.132	7.805	9.201	11.778
Germany	0.334	17.780	20.042	22.743
United Kingdom	0.263	14.274	16.232	18.889
Italy	0.231	12.694	14.514	17.152
USA	0.115	6.966	8.288	10.855
Canada	0.098	6.127	7.376	9.933
Brazil	0.155	8.941	10.435	13.027
Argentina	0.035	3.016	3.995	6.513
Australia	0.144	8.398	9.845	12.430
Egypt	0.061	4.300	5.390	7.924

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DR Congo	0.097	6.077	7.322	9.878
South Africa	0.075	4.991	6.141	8.684
Libya	0.006	1.584	2.438	4.938
Ethiopia	0.02	2.275	3.190	5.698

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**Table S6.** Summary of hydrogen storage methods or materials mentioned in Fig. 4.

Name	Gravimetric capacity (wt.%)	Volumetric capacity (g H <sub>2</sub> L <sup>-1</sup> )	Operation temperature (°C)	Type
Liq. H (Liquid hydrogen) <sup>7</sup>	10	40	-253	Physical method
Comp. H (Compressed hydrogen) <sup>8</sup>	6	30	25	Physical method
CA (Carbon Aerogel) <sup>9</sup>	5.3	10.6	-196	Physical adsorption
AC (Activated Carbon) <sup>10</sup>	3.7	26	-196	Physical adsorption
CNT (Carbon Nanotubes) <sup>11</sup>	0.5	18	-196	Physical adsorption
MOF-74 <sup>12</sup>	2.8	/	-196	Physical adsorption
IRMOF-1 <sup>13</sup>	4.3	/	-196	Physical adsorption
LaNi <sub>5</sub> H <sub>6</sub> <sup>14</sup>	1.5	125.9	150	Interstitial hydride
TiFeH <sub>1.9</sub> <sup>15</sup>	1.8	123.6	30	Interstitial hydride
PdH <sub>0.7</sub> <sup>16</sup>	0.7	78.1	120	Interstitial hydride
ZrMn <sub>2</sub> H <sub>3.6</sub> <sup>17</sup>	1.7	130	60	Interstitial hydride
Ti-Cr-Mn <sup>18</sup>	2	/	-10	Interstitial hydride
NaH <sup>19</sup>	4.2	58	440	Metal

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				hydride
$\text{BeH}_2^{19}$	18.1	118	260	Metal hydride
$\text{LiH}^{19}$	12.6	103	740	Metal hydride
$\text{CaH}_2^{19}$	4.8	82	600	Metal hydride
$\text{AlH}_3^{20}$	10.1	150	160	Metal hydride
$\text{MgH}_2^{21}$	7.6	100	290	Metal hydride
$\text{Mg}_2\text{NiH}_4^{22}$	3.6	100	240	Metal hydride
$\text{Mg}(\text{BH}_4)_2^{23}$	14.8	77	320	Complex hydride
$\text{LiBH}_4^{24}$	18.4	120	380	Complex hydride
$\text{NaBH}_4^{24}$	10.7	118	400	Complex hydride
$\text{NaAlH}_4^{25}$	7.5	96	180	Complex hydride
$\text{NH}_3\text{BH}_3^{26}$	19.5	152	500	Chemical hydride
$\text{C}_{10}\text{H}_{18}$ (Decalin) <sup>27</sup>	7.2	65	210	Chemical hydride
$\text{NH}_3^{28}$	17.6	105	700	Chemical hydride
$\text{C}_6\text{H}_{12}$ (Cyclohexane) <sup>29</sup>	7.1	55	300	Chemical hydride

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C<sub>7</sub>H<sub>14</sub> (Cycloheptane)<sup>30</sup>

6.1

47

350

Chemical  
hydride

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**Table S7.** Parameters of pipeline transmission.<sup>31-33</sup>

Parameters	Values
Life	30 years
Pipeline length	2000 km
Discount rate	8%
Electricity price	0.015 \$ kWh <sup>-1</sup>
Mass flow rate	151893 kg H <sub>2</sub> per day
Capital recovery period	15 years
Hydrogen loss rate	486.25 kg per km per year
<b>Capital cost:</b>	
Material cost	74649 \$ km <sup>-1</sup>
Miscellaneous	264245 \$
Reconstruction of natural gas pipeline	12% of new pipeline
<b>Operation and maintenance:</b>	
Labor cost	240015 \$ km <sup>-1</sup>
Labor hours	9019 hours per year
Salary	33.54 \$ per hour
Operation and maintenance cost	5% of capital investment cost

**Table S8.** Parameters of pipeline distribution <sup>a,31</sup>

Parameters	Values
Life	30 years
Pipeline length	830.4 km
Discount rate	8%
Capital recovery period	15 years
Hydrogen loss rate	97.5 kg per km per year
Mass flow rate	151893 kg per day
<b>Capital cost:</b>	
Material cost	119380 \$ km <sup>-1</sup>
Miscellaneous	149315 \$
<b>Operation and maintenance:</b>	
Labor cost	88805 \$ km <sup>-1</sup>
Electricity price	0.05 \$ kWh <sup>-1</sup>
Labor hours	9019 hours per year
Salary	33.54 \$ hour <sup>-1</sup>
Operation and maintenance cost	5% of capital cost

<sup>a</sup>In the sea-transmission scenario, pipeline distribution methods are used for all routes. The transmission cost from port to receiving station is included in the receiving cost.

**Table S9.** Parameters of compressor.<sup>31,34</sup>

Parameter	Value
Z (Compress factor)	1.28216
m (Mass flow rate, kg/s)	700/ (12 × 3600) <sup>a</sup>
R (Universal gas constant)	8.3144 kJ (kg.mole.K) <sup>-1</sup>
T (Inlet temperature)	300 K
n (Compressor stage)	2
K (k is the ratio of specific heats)	1.4
$\eta$ (The isentropic efficiency)	0.75
$P_{\text{outlet}}$	20 MPa (Trailer) 45 MPa (High-pressure tank)
$P_{\text{inlet}}$	2 MPa (Trailer) 10.3 MPa (Operation pressure of pipeline)

<sup>a</sup>Work 12 hours per day.

**Table S10.** Parameters of refueling station.<sup>31,34</sup>

Parameters	Values
Scale	700 kg per day
Life year	20
Discount rate	8%
Inlet pressure	20 MPa (Tank trailer), 10 MPa (Pipeline)
Output pressure	45 MPa
Capacity of hydrogen tank (45 MPa)	700×1.5
<b>Capital cost:</b>	
Compressor cost	7500*P <sub>compressor</sub> (power of compressor)
Filling machine cost	100000×2 \$
Tank cost	1495 \$ per kg hydrogen
<b>Operation and maintenance:</b>	
Operation and maintenance cost of compressor	5% of compressor cost
Operation and maintenance cost of tank	3.5% of tank cost
Worker	2
Salary	1500 (\$/month)×2 (worker)×12 (month)×20 (year)

**Table S11.** Input parameters of methanol synthesis.<sup>35</sup>

Parameters	Values
Project life (year)	20
Discount rate	8%
Electricity (\$ kWh <sup>-1</sup> )	0.05
Scale (ton methanol per year)	10000
Operation hours	8400
Exchange rate of RMB to \$	6.8
Exchange rate of RMB to €	7.73
Methanol purity	99.92 wt.%
CO <sub>2</sub> usage (ton per ton methanol)	1.59
H <sub>2</sub> usage (ton per ton methanol)	0.22
H <sub>2</sub> compressor power (kW)	77.2
Recycle compressor power (kW)	115.17
Electricity usage (kWh per ton-MeOH)	161.75
Steam usage (ton/ton-MeOH)	0.43
Colling water usage (ton per ton methanol)	618.77
Heat duty of reboiler1 (kW)	281.22
Heat duty of reboiler2 (kW)	204.3
<b>Capital cost:</b>	
Methanol reactor (\$)	$1.6 \times 10^7 \times ((\text{Mass in kg h}^{-1})/54000)^{0.65}$
Compressor (\$)	$4.0 \times 10^4 \times P(\text{kW})^{0.6038}$
Boiler (\$)	$4.6397 \times 10^6 \times ((\text{Heat duty MW})/55.6)^{0.6}$
CO <sub>2</sub> (\$ ton <sup>-1</sup> )	58.1618
Fresh water (\$ ton <sup>-1</sup> )	0.22
Cooling water (\$ ton <sup>-1</sup> )	0.05
Steam (\$ ton <sup>-1</sup> )	14.71
Cooler (million \$)	0.028
Heat exchanger (million \$)	0.038
Recycle pump (million \$)	0.703
Compressor (million \$)	0.553
Reactor (million \$)	4.385
Heat duty of reboiler (million \$)	0.393
Separator (million \$)	0.046
Boiler (million \$)	0.276
<b>Indirect capital cost:</b>	
General facilities (% capital cost)	10%
Engineering permitting and start up	10%
Contingencies (% capital cost)	5%
Working capital, land, and miscellaneous (% capital cost)	3%
<b>Operation and maintenance cost:</b>	
Water (million \$ per year)	1.856

CO <sub>2</sub> (million \$ per year)	6.29
Electricity (million \$ per year)	0.57
Others (million \$ per year)	4% of capital cost

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**Table S12.** Input parameters of methanol cracking.<sup>36</sup>

Parameters	Values
Scale	532 kg H <sub>2</sub> per day
Project life (year)	20
Electricity (\$ kWh <sup>-1</sup> )	0.05
Discount rate	8%
<b>Capital cost:</b>	
Methanol storage (million \$)	0.004
Methanol reformer (million \$)	0.455
Hydrogen compressor (million \$)	0.09
Hydrogen storage (million \$)	0.115
Hydrogen dispenser (million \$)	0.021
Total process unit (million \$)	0.685
<b>Indirect capital cost:</b>	
General facilities (million \$)	0.137
Engineering permitting and start up (million \$)	0.069
Contingencies (million \$)	0.069
Working capital, land and miscellaneous (million \$)	0.034
<b>Operation and maintenance:</b>	
Variable non-fuel O&M (\$ per kg H <sub>2</sub> )	0.028
Labor (\$ per kg H <sub>2</sub> )	0.069
Electricity (\$ per kg H <sub>2</sub> )	0.142
Variable operating cost (\$ per kg H <sub>2</sub> )	0.239
Fixed operating cost (\$ per kg H <sub>2</sub> )	0.171

**Table S13.** Capital Cost parameters of ammonia synthesis.<sup>36</sup>

Parameters	Values
Scale	300 tons ammonia per day
Project life (year)	20
Electricity (\$ kWh <sup>-1</sup> )	0.05
Discount rate	8%
Natural gas costs (\$ GJ <sup>-1</sup> )	5.40
Water Electrolysis +Haber-Bosch	12000 (kWh per ton NH <sub>3</sub> )
<b>Capital Cost of installing cryogenic air distillation:</b>	
Heat exchangers (million \$)	2.63
Compressors (million \$)	8.21
Columns (million \$)	6.04
Pump (million \$)	0.02
Indirect capital Cost:	
Contingency and fees (%capital cost)	18%
Auxiliary facility, land, infrastructure (%capital cost)	27%
<b>Capital cost Haber Bosch Process:</b>	
Heat exchangers (million \$)	11.31
Compressors (million \$)	21.06
Reactors (million \$)	2.46
Separator (million \$)	2.55
Furnace (million \$)	2.39
Indirect capital Cost:	
Contingency and fees (%capital cost)	18%
Auxiliary facility, land, infrastructure (%capital cost)	18.70%
<b>Capital cost of Electric Haber Bosch Process:</b>	
Heat exchangers (million \$)	11.31
Compressors (million \$)	21.06
Reactors (million \$)	2.46
Separator (million \$)	2.55
Indirect capital Cost:	
Contingency and fees (%capital cost)	18%
auxiliary facility, land, infrastructure (%capital cost)	21.40%
<b>Capital cost of installing ammonia storage system:</b>	
Heat exchanger (million \$)	0.16
Compressors (million \$)	0.09
Tank (Inner and Outer) (million \$)	5.33
Separator (million \$)	0.07
Indirect capital Cost:	
Contingency and fees (%module cost) (million \$)	18%



Others (million \$)

31%

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**Table S14.** Operation and maintenance cost parameters of ammonia synthesis.<sup>36</sup>

Operation and maintenance cost	Haber Bosch Process	Storage system	Air separation unit
Water Consumption (m <sup>3</sup> per year)	1371018	6669	185157
Cost of Water (million \$ per year)	3.06	0.01	0.41
Electric Consumption (MWh per year)	71655	185	25829
Natural Gas Consumption (tons per year)	1860		
Total number of operators	16	12	14
Labor cost (million \$ year <sup>-1</sup> )	0.87	0.65	0.76
Salvage value (million \$ year <sup>-1</sup> )	5.58	0.84	2.45
Depreciation (million \$ year <sup>-1</sup> )	2.51	0.38	1
Miscellaneous cost (million \$ year <sup>-1</sup> )	9.2	2.02	4.5
Manufacture cost	31.55	3.06	6.77

**Table S15.** Input parameters of Ammonia cracking.<sup>36</sup>

Parameters	Values
Life	30 years
Scale	100 tons H <sub>2</sub> per day
Electricity (\$ kWh <sup>-1</sup> )	0.05
Discount rate	8%
<b>Capital cost:</b>	
Equipment purchase (\$)	77417838.39
Installation (\$)	36386383.66
Piping (\$)	52644129.81
Electrical System (\$)	8515962.423
Instrumentation & controls (\$)	27870421.23
Buildings (\$)	13935211.14
Yard improvement (\$)	7741783.734
Service facilities (\$)	54192487.19
Indirect cost:	
Engineering & supervision (\$)	25547886.22
Construction expenses (\$)	31741313.63
Legal expenses (\$)	3096713.704
Contractor's fee (\$)	17031924.85
Contingency (\$)	34063848.64
Working capital (\$)	58527885.96
<b>Operation and maintenance (O &amp; M) cost:</b>	
O & M fixed cost:	
Operating labor (\$)	467821.0685
Supervision labor (\$)	116955.2671
Direct salary overhead (\$)	233910.5343
Maintenance cost (\$)	19509294.97
Insurance cost (\$)	7803717.987
General plant overhead (\$)	12196789.1
O & M variable cost:	
Water (\$)	48723.00
Heating utilities (\$)	3953070.00
Colling Utilities (\$)	3922179.00
Cracker catalyst (\$)	4320953.00
Electricity (\$)	18986406.00

**Table S16.** Parameters of methanol shipment.<sup>37</sup>

Parameters	Values
Transport distance	7500 km (Japan-Australia)
Load of ship	110000 tons
Discount rate	8%
Average speed	32 km h <sup>-1</sup>
Fuel consumption	99 ton per day
Material loss rate	0.01% per day
Loading period	2 days
Life of ship	15 years
<b>Capital cost:</b>	
Capital investment of ship	76000000 \$
Indirect capital cost	45% of capital cost
<b>Operation and maintenance (O &amp; M) cost:</b>	
Diesel cost	358.3 \$ ton <sup>-1</sup>
Operation cost of ship	5550 \$ per day
Labor cost	20×15×100000 <sup>a</sup>
Load cost	37.5 \$ per ton methanol
Receiving cost	37.5 \$ per ton methanol

<sup>a</sup>20 crews of one ship and the salary each crew is 10000 \$ per year.

**Table S17.** Parameters of ammonia shipment.<sup>37</sup>

Parameters	Values
Transport distance	7500 km (Japan-Australia)
Load of ship	53000 ton
Discount rate	8%
Average speed	32 km h <sup>-1</sup>
Fuel consumption	61 tons per day
Material loss rate	0.04% per day
Loading period	2 days
Life of ship	15 years
<b>Capital cost:</b>	
Capital investment of ship	85000000 \$
Indirect capital cost	45% of capital cost
<b>Operation and maintenance (O &amp; M) cost:</b>	
Operation cost of ship	5550 \$ day <sup>-1</sup>
Diesel cost	358.3 \$ ton <sup>-1</sup>
Labor cost	20×15×100000
Load cost	33.5 \$ per ton methanol
Receiving cost	35.3 \$ per ton methanol

## References

- 1 Green hydrogen cost reduction: scaling up electrolysers to meet the 1.5 °C climate goal, International Renewable Energy Agency, 2020.
- 2 D. Jang, J. Kim, D. Kim, W.-B. Han and S. Kang, *Energy Convers. Manag.*, 2022, **258**, 115499.
- 3 M. D. Rashid, M. K. Al Mesfer, H. Naseem and M. Danish, *Int. J. Eng. Adv. Tech.*, 2015, **4**, 80-93.
- 4 F. M. Sapountzi, J. M. Gracia, C. J. Weststrate, H. O. A. Fredriksson and J. W. Niemantsverdriet, *Prog. Energy Combust. Sci.*, 2017, **58**, 1-35.
- 5 F. Mueller-Langer, E. Tzimas, M. Kaltschmitt and S. Peteves, *Int. J. Hydrogen Energy*, 2007, **32**, 3797-3810.
- 6 Electricity market report, International Energy Agency, 2022.
- 7 E. Rivard, M. Trudeau and K. Zaghbi, *Materials*, 2019, **12**, 1973.
- 8 H. Barthelemy, M. Weber and F. Barbier, *Int. J. Hydrogen Energy*, 2017, **42**, 7254-7262.
- 9 H. Kabbour, T. F. Baumann, J. H. Satcher, A. Saulnier and C. C. Ahn, *Chem. Mater.*, 2006, **18**, 6085-6087.
- 10 R. J. Anderson, T. P. McNicholas, A. Kleinhammes, A. Wang, J. Liu and Y. Wu, *J. Am. Chem. Soc.*, 2010, **132**, 8618-8626.
- 11 Y. Ye, C. C. Ahn, C. Witham, B. Fultz, J. Liu, A. G. Rinzler, D. Colbert, K. A. Smith and R. E. Smalley, *Appl. Phys. Lett.*, 1999, **74**, 2307-2309.
- 12 Y. Liu, C. M. Brown, D. A. Neumann, H. Kabbour and C. C. Ahn, *Mater. Res.*

- Soc. Symp. Proc.*, 2007, 1041.
- 13 A. Dailly, J. J. Vajo and C. C. Ahn, *J. Phys. Chem. B*, 2006, **110**, 1099-1101.
  - 14 P. D. Goodell and P. S. Rudman, *J. Less-common Met.*, 1983, **89**, 117-125.
  - 15 J. J. Reilly and R. H. Wiswall, *Inorg. Chem.*, 1974, **13**, 218-222.
  - 16 F. D. Manchester, A. San-Martin and J. M. Pitre, *J. Phase Equilibria*, 1994, **15**, 62-83.
  - 17 J. M. Park and J. Y. Lee, *J. Alloys Comp.*, 1992, **182**, 43-54.
  - 18 Y. Komazaki, M. Uchida, S. Suda, A. Suzuki, S. Ono and N. Nishimiya, *J. Less-common Met.*, 1983, **89**, 269-274.
  - 19 Q. Lai, Y. Sun, T. Wang, P. Modi, C. Cazorla, U. B. Demirci, J. R. Ares Fernandez, F. Leardini and K. F. Aguey - Zinsou, *Adv. Sustainable Syst.*, 2019, **3**, 1900043.
  - 20 J. Graetz and J. J. Reilly, *J. Phys. Chem. B*, 2005, **109**, 22181-22185.
  - 21 L. Schlapbach and A. Züttel, *Nature*, 2002, **414**, 353-358.
  - 22 L. Li, T. Akiyama and J. Yagi, *J. Alloys Comp.*, 2000, **308**, 98-103.
  - 23 I. Saldan, *Int. J. Hydrogen Energy*, 2016, **41**, 11201-11224.
  - 24 H. I. Schlesinger, H. C. Brown, H. R. Hoekstra and L. R. Rapp, *J. Am. Chem. Soc.*, 1953, **75**, 199-204.
  - 25 B. Bogdanović and M. Schwickardi, *J. Alloys Comp.*, 1997, **253**, 1-9.
  - 26 R. P. Shrestha, H. V. K. Diyabalanage, T. A. Semelsberger, K. C. Ott and A. K. Burrell, *Int. J. Hydrogen Energy*, 2009, **34**, 2616-2621.
  - 27 A. N. Kalenchuk and L. M. Kustov, *Molecules*, 2022, **27**, 2236.

- 28 S. Mukherjee, S. V. Devaguptapu, A. Sviripa, C. R. F. Lund and G. Wu, *Appl. Catal. B Environ.*, 2018, **226**, 162-181.
- 29 G. Cacciola, N. Giordano and G. Restuccia, *Int. J. Hydrogen Energy*, 1984, **9**, 411-419.
- 30 F. Alhumaidan, D. Cresswell and A. Garforth, *Energy Fuels*, 2011, **25**, 4217-4234.
- 31 DOE H<sub>2</sub>: a delivery analysis, [https://www.hydrogen.energy.gov/h2a\\_delivery.html](https://www.hydrogen.energy.gov/h2a_delivery.html).
- 32 B. Miao, L. Giordano and S. H. Chan, *Int. J. Hydrogen Energy*, 2021, **46**, 18699-18718.
- 33 Transporting pure hydrogen by repurposing existing gas infrastructure: overview of existing studies and reflections on the conditions for repurposing, European Union Agency for the Cooperation of Energy Regulators, 2021.
- 34 Y. Gu, Q. Chen, J. Xue, Z. Tang, Y. Sun and Q. Wu, *Energy Convers. Manag.*, 2020, **223**, 113240.
- 35 Y. Gu, D. Wang, Q. Chen and Z. Tang, *Int. J. Hydrogen Energy*, 2022, **47**, 5085-5100.
- 36 S. A. Noshervani and R. C. Neto, *J. Energy Storage*, 2021, **34**, 102201.
- 37 J.-S. Lee, A. Cherif, H.-J. Yoon, S.-K. Seo, J.-E. Bae, H.-J. Shin, C. Lee, H. Kwon and C.-J. Lee, *Renew. Sust. Energy Rev.*, 2022, **165**, 112556.