

Supporting Information for:

Feedstock nitrogen content mediates maximum possible Pb sorption capacity of biochars

Chinonso Ogbuagu¹, Steve Robinson¹, and Tom Sizmur¹

¹Soil Research Centre, Department of Geography and Environmental Science, University of Reading

This Supporting Information file contains text and references describing the assignment of FTIR peaks to functional groups, 4 Tables and 5 Figures:

Table S-1. Functional groups observed in the Fourier-transform infrared (FTIR) spectra of biochars produced at different pyrolysis temperatures

Table SI-2. Final pH of solutions after sorption

Table S-3. Biochar feedstock properties

Table S-4. The relationship between biochar feedstock properties and maximum possible Pb sorption capacity at optimum pyrolysis temperature

Figure S-1; Sorption isotherm for hay biochars pyrolyzed at 300 °C, 350 °C, 400 °C, 450 °C, 500 °C, 550 °C, 600 °C, 650 °C, 700 °C, and 750 °C shaken with 100, 200, 500, 1000, and 5000 mg/L of Pb and fitted to a Langmuir sorption isotherm model

Figure S-2; Sorption isotherm for wheat straw biochars pyrolyzed at 300 °C, 350 °C, 400 °C, 450 °C, 500 °C, 550 °C, 600 °C, 650 °C, 700 °C, and 750 °C shaken with 100, 200, 500, 1000, and 5000 mg/L of Pb and fitted to a Langmuir sorption isotherm model

Figure S-3; Sorption isotherm for coco coir biochars pyrolyzed at 300 °C, 350 °C, 400 °C, 450 °C, 500 °C, 550 °C, 600 °C, 650 °C, 700 °C, and 750 °C shaken with 100, 200, 500, 1000, and 5000 mg/L of Pb and fitted to a Langmuir sorption isotherm model

Figure S-4; Sorption isotherm for pine bark biochars pyrolyzed at 300 °C, 350 °C, 400 °C, 450 °C, 500 °C, 550 °C, 600 °C, 650 °C, 700 °C, and 750 °C shaken with 100, 200, 500, 1000, and 5000 mg/L of Pb and fitted to a Langmuir sorption isotherm model

Figure S-5. FTIR spectra for feedstocks (FD) and biochars made from coco coir (CC), hay, wheat straw (WS), and pine bark (PB) at pyrolysis temperatures between 300°C and 750°C

Assignment of FTIR (Fourier-transform infrared) spectroscopy peaks to functional groups

The peaks observed at 3380 cm^{-1} were attributed to O-H stretching which may belong to the hydroxyl, alcohol, carboxylic acid, and water groups (Figueredo et al., 2018; Keiluweit et al., 2010). The peaks around 2940 cm^{-1} were attributed to vibrations of asymmetric and symmetric aliphatic C-H stretching (Keiluweit et al., 2010; Li et al., 2020; Zhao et al., 2017). The peaks at 2158 cm^{-1} could be linked to overtones of aromatic ring bonds $\text{C}\equiv\text{C}$ (Korus et al., 2019). The stretching vibration at 1590 cm^{-1} could be attributed to aromatic compounds $\text{C}=\text{C}$ and carbonyl groups $\text{C}=\text{O}$ (Figueredo et al., 2017; Li et al., 2020). The peaks at 1416 cm^{-1} could be ascribed to asymmetric deformation vibration of CH_3 aliphatic group as seen in a study by Li et al., (2020). The peaks which appeared at 1351 cm^{-1} signify the phenolic stretching vibrations of O-H (Zhao et al., 2017). The stretching at 1160 cm^{-1} could arise from C-O-C oxygen containing functional group, while the stretching at 876 cm^{-1} were attributed to the out of plane deformation of aromatic C-H groups (Zhao et al., 2017).

References

- Figueiredo, C., Lopes, H., Coser, T., Vale, A., Busato, J., Aguiar, N., Novotny, E., Canellas, L., 2018. Influence of pyrolysis temperature on chemical and physical properties of biochar from sewage sludge. *Arch. Agron. Soil Sci.* 64, 881–889. <https://doi.org/10.1080/03650340.2017.1407870>
- Keiluweit, M., Nico, P.S., Johnson, M.G., Kleber, M., 2010. Dynamic Molecular Structure of Plant Biomass-Derived Black Carbon (Biochar). *Environ. Sci. Technol.* 44, 1247–1253. <https://doi.org/10.1021/es9031419>
- Li, B., Liu, D., Lin, D., Xie, X., Wang, S., Xu, H., Wang, J., Huang, Y., Zhang, S., Hu, X., 2020. Changes in Biochar Functional Groups and Its Reactivity after Volatile-Char Interactions during Biomass Pyrolysis. *Energy and Fuels* 34, 14291–14299. <https://doi.org/10.1021/acs.energyfuels.0c03243>
- Zhao, S., Ta, N., Wang, X., 2017. Effect of Temperature on the Structural and Physicochemical Properties of Biochar with Apple Tree Branches as Feedstock Material. <https://doi.org/10.3390/en10091293>
- Korus, A., Szłęk, A., Samson, A., 2019. Physicochemical properties of biochars prepared from raw and acetone-extracted pine wood. *Fuel Process. Technol.* 185, 106–116. <https://doi.org/10.1016/j.fuproc.2018.12.004>

Table S-1. Functional groups observed in the Fourier-transform infrared (FTIR) spectra of biochars produced at different pyrolysis temperatures

FTIR Peak (cm ⁻¹)	Functional groups assigned
3380	O–H stretching (water, hydrogen-bonded hydroxyl)
2940	C–H stretching (aliphatic CH _x ; 2935-asymmetric), C-H stretching aliphatic CH _x
2158	C≡C Overtones of aromatic rings
1590	Aromatic C=C and C=O stretching of conjugated ketones and quinones, aromatic components and C]O of conjugated ketones and quinones
1416	Asymmetric deformation vibration of CH ₃ , CH ₂ added to carboxymethyl cellulose
1351	in plane bending of phenolic -OH
1160	C-O-C ester groups in cellulose and hemicellulose
876	C–H bending (aromatic C–H out-of-plane deformation), C=C polycyclic aromatic structures

Table SI-2. Final pH of solutions after sorption

Biochar Feedstock	Pyrolysis temp (°C)	100 mg Pb L ⁻¹	200 mg Pb L ⁻¹	500 mg Pb L ⁻¹	1000 mg Pb L ⁻¹	5000 mg Pb L ⁻¹
Coco Coir	300	6.31	5.43	4.81	4.35	3.82
Coco Coir	350	8.59	8.34	7.74	7.15	5.27
Coco Coir	400	9.56	9.34	8.66	7.15	5.27
Coco Coir	450	10.37	10.27	9.95	8.99	6.09
Coco Coir	500	10.57	10.41	10.13	9.59	5.63
Coco Coir	550	10.83	10.73	10.45	10.13	5.60
Coco Coir	600	10.8	10.67	10.48	10.23	5.34
Coco Coir	650	10.69	10.63	10.48	10.36	5.30
Coco Coir	700	11.03	10.96	10.8	10.62	5.28
Coco Coir	750	11.00	11.00	10.83	10.66	5.49
Hay	300	6.23	6.33	5.66	5.47	4.36
Hay	350	7.19	7.23	5.75	5.89	4.95
Hay	400	8.23	7.34	7.99	7.41	5.10
Hay	450	9.21	9.12	8.88	8.22	5.11
Hay	500	9.74	9.60	9.27	8.82	5.11
Hay	550	10.33	10.30	10.04	9.63	5.30
Hay	600	10.46	10.38	10.16	9.74	5.28
Hay	650	10.83	10.78	10.56	10.14	5.42
Hay	700	11.04	10.98	10.83	10.66	6.06
Hay	750	11.26	11.17	11.09	10.87	6.55
Wheat straw	300	6.27	6.13	4.84	4.63	4.03
Wheat straw	350	7.01	7.41	6.80	6.47	4.89
Wheat straw	400	8.80	8.63	8.21	7.40	5.19
Wheat straw	450	9.68	9.50	9.18	8.52	5.29
Wheat straw	500	10.44	10.37	10.18	9.68	5.32
Wheat straw	550	10.40	10.32	10.15	9.69	5.23
Wheat straw	600	10.38	10.31	10.19	9.85	5.32
Wheat straw	650	10.47	10.40	10.30	9.96	5.55
Wheat straw	700	10.30	10.24	10.17	9.93	5.53
Wheat straw	750	10.45	10.38	10.35	10.4	6.50
Pine bark	300	4.26	4.02	3.32	3.31	3.04
Pine bark	350	4.45	4.30	3.88	3.81	3.66
Pine bark	400	4.71	4.50	4.19	4.11	4.02
Pine bark	450	5.42	5.14	4.23	4.64	4.60
Pine bark	500	6.03	5.56	5.01	4.89	4.74
Pine bark	550	6.54	6.01	5.16	5.03	4.88
Pine bark	600	7.21	6.72	5.36	5.16	5.13
Pine bark	650	7.77	7.26	6.15	5.30	5.21
Pine bark	700	8.85	8.41	7.21	5.72	5.80
Pine bark	750	9.67	9.36	7.63	6.61	5.93

Table S-3. Biochar feedstock properties

Feedstock	N (%)	C (%)	C/N ratio	Ca (mg/kg)	K (mg/kg)	Mg (mg/kg)	P (mg/kg)	S (mg/kg)	Na (mg/kg)	Al (mg/kg)	Fe (mg/kg)	Cu (mg/kg)	Pb (mg/kg)	Mn (mg/kg)	Zn (mg/kg)
Hay	1.11	41.4	37.3	4490	16032	1303	2008	1308	996	53.3	87.3	3.40	1.14	27.2	19.7
Wheat straw	0.73	44.1	60.7	5361	8794	328	695	733	343	98.0	397	2.50	1.64	29.2	9.9
Pine bark	0.09	45.9	534	1356	776	262	101	441	12	80.3	141	1.94	1.89	35.6	10.4
Coco coir	0.44	44.2	101	2307	12584	1763	364	765	4046	660	1774	1.49	1.39	68.6	16.7

Table S-4. The relationship between biochar feedstock properties and maximum possible Pb sorption capacity at optimum pyrolysis temperature

Feedstock property	R ²	p-value
N	0.954	0.023
C	0.813	0.099
C/N ratio	0.863	0.071
Ca	0.782	0.116
K	0.759	0.129
Mg	0.135	0.633
P	0.706	0.160
S	0.763	0.127
Na	0.007	0.914
Cu	0.525	0.276
Pb	0.656	0.190
Mn	0.086	0.708
Zn	0.302	0.451
Al	0.016	0.872
Fe	0.006	0.920

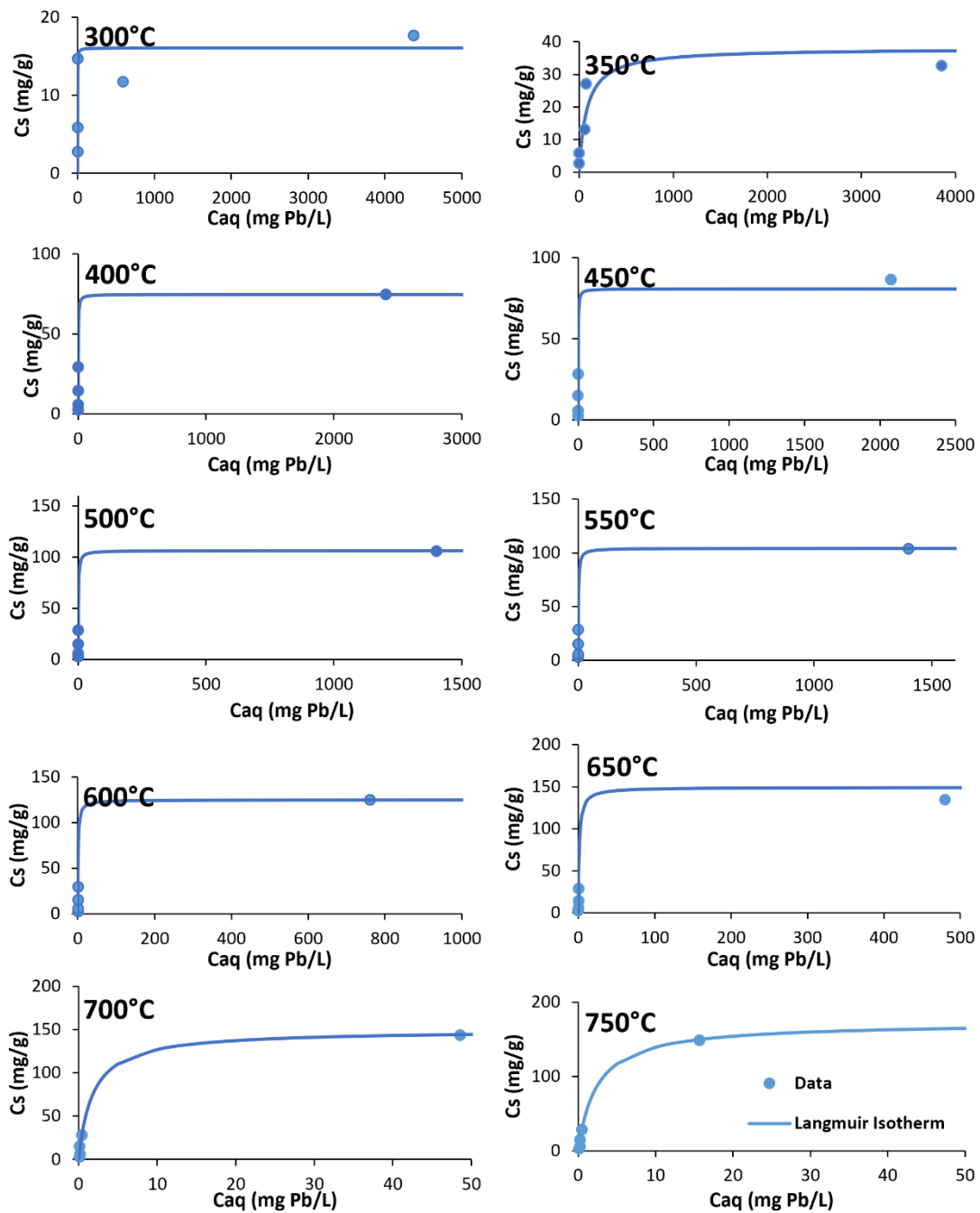


Figure S-1; Sorption isotherm for hay biochars pyrolyzed at 300 °C, 350 °C, 400 °C, 450 °C, 500 °C, 550 °C, 600 °C, 650 °C, 700 °C, and 750 °C shaken with 100, 200, 500, 1000, and 5000 mg/L of Pb and fitted to a Langmuir sorption isotherm model

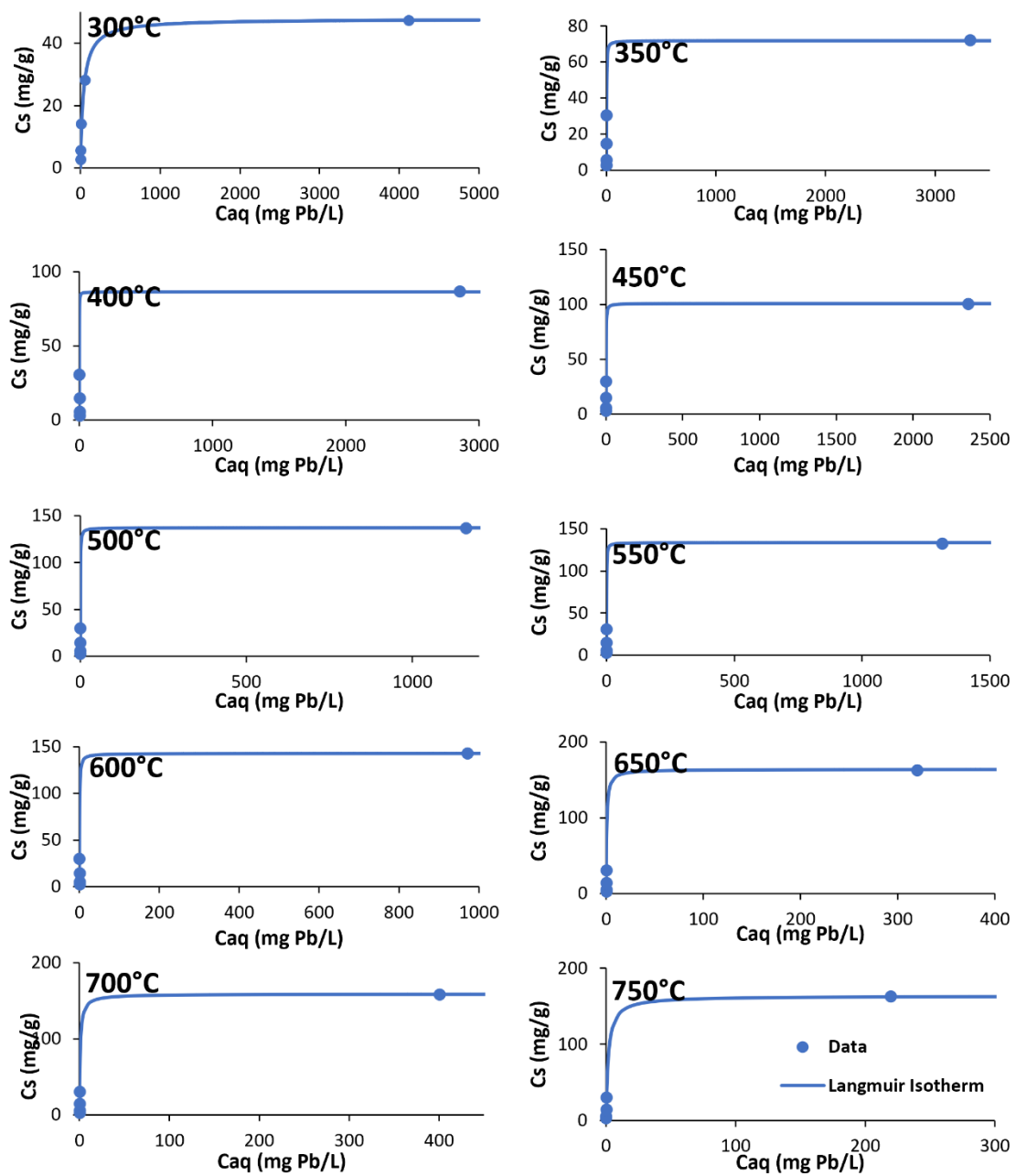


Figure S-2; Sorption isotherm for wheat straw biochars pyrolyzed at 300 °C, 350 °C, 400 °C, 450 °C, 500 °C, 550 °C, 600 °C, 650 °C, 700 °C, and 750 °C shaken with 100, 200, 500, 1000, and 5000 mg/L of Pb and fitted to a Langmuir sorption isotherm model

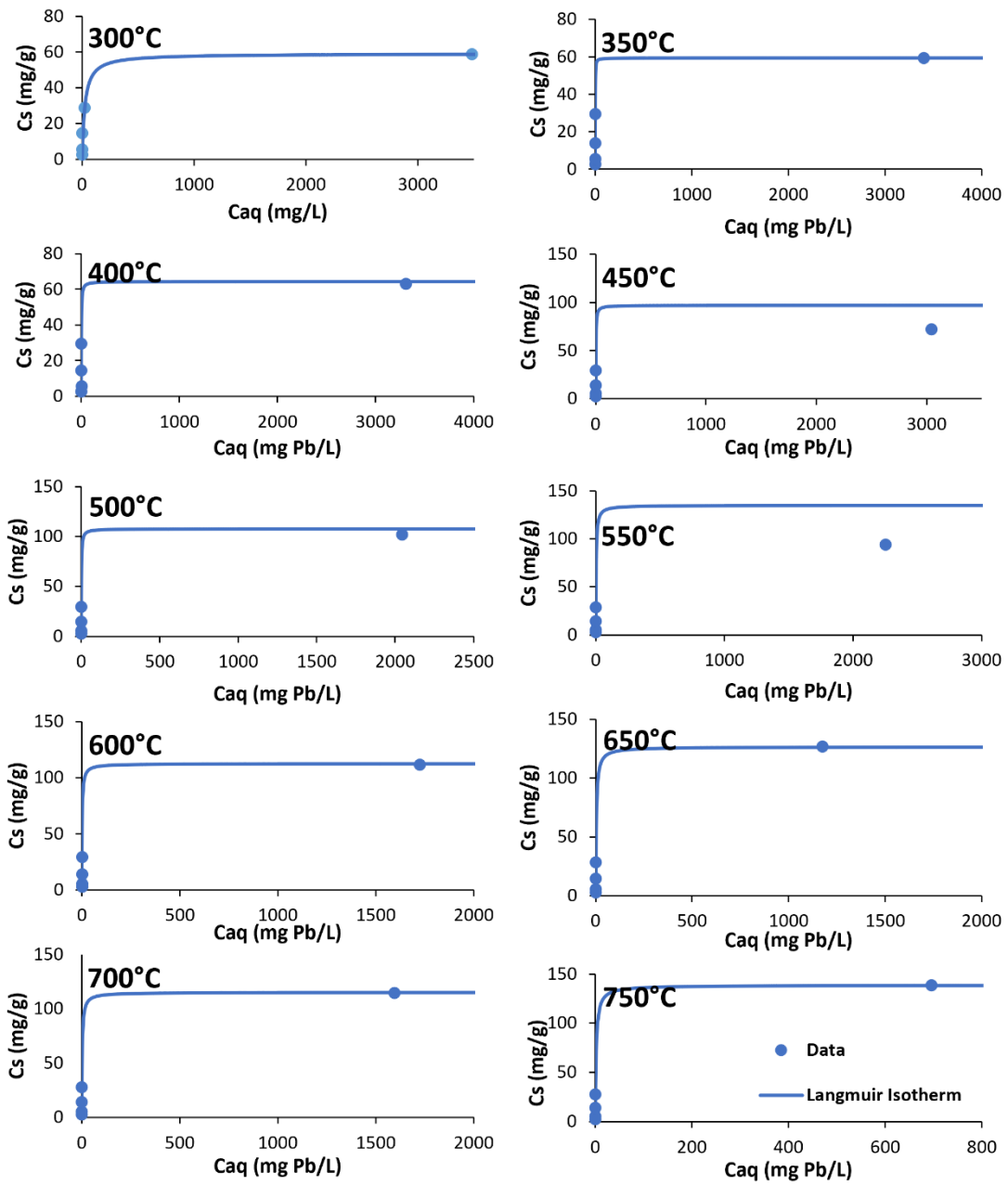


Figure S-3; Sorption isotherm for coco coir biochars pyrolyzed at 300 °C, 350 °C, 400 °C, 450 °C, 500 °C, 550 °C, 600 °C, 650 °C, 700 °C, and 750 °C shaken with 100, 200, 500, 1000, and 5000 mg/L of Pb and fitted to a Langmuir sorption isotherm model

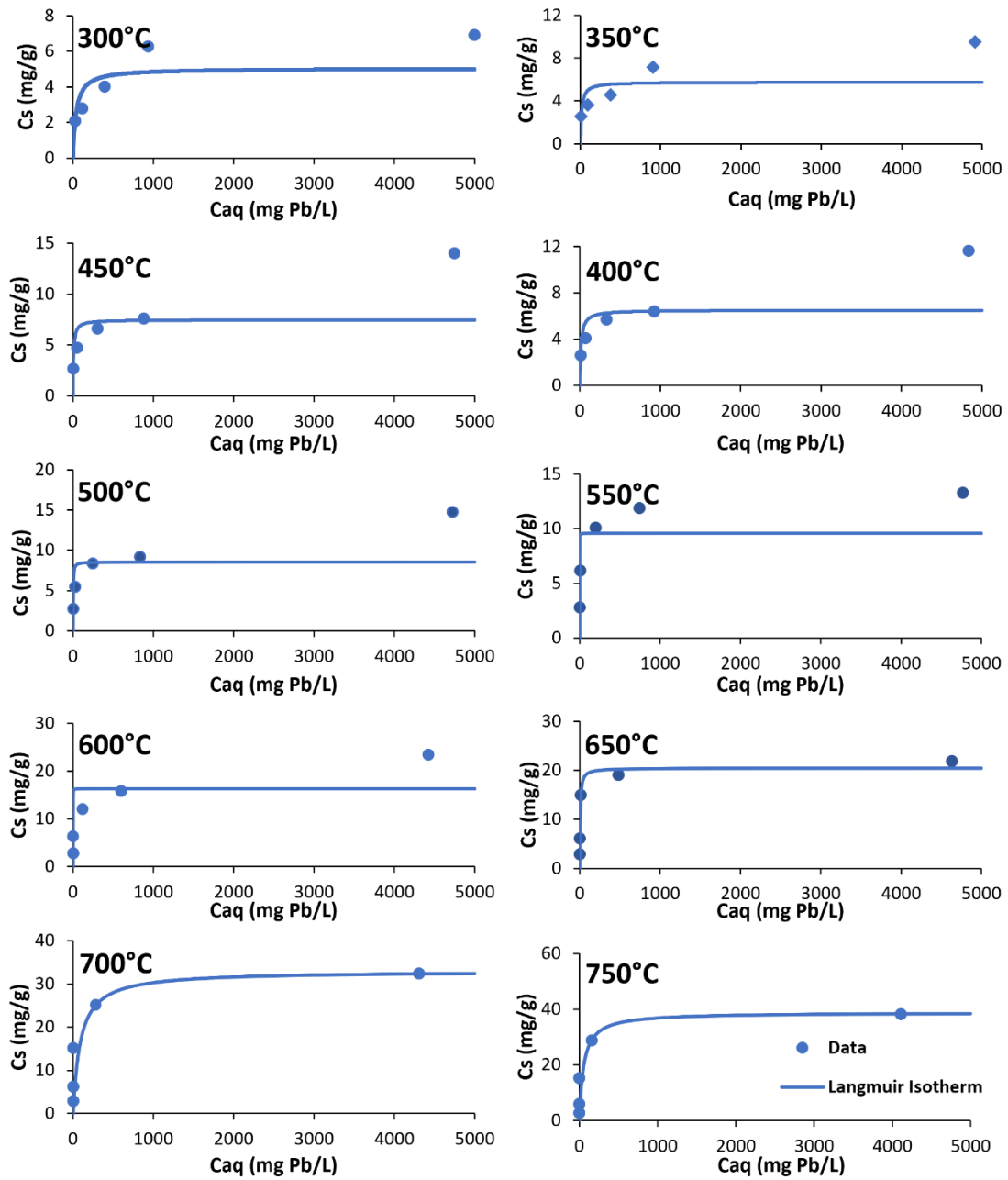


Figure S-4; Sorption isotherm for pine bark biochars pyrolyzed at 300 °C, 350 °C, 400 °C, 450 °C, 500 °C, 550 °C, 600 °C, 650 °C, 700 °C, and 750 °C shaken with 100, 200, 500, 1000, and 5000 mg/L of Pb and fitted to a Langmuir sorption isotherm model

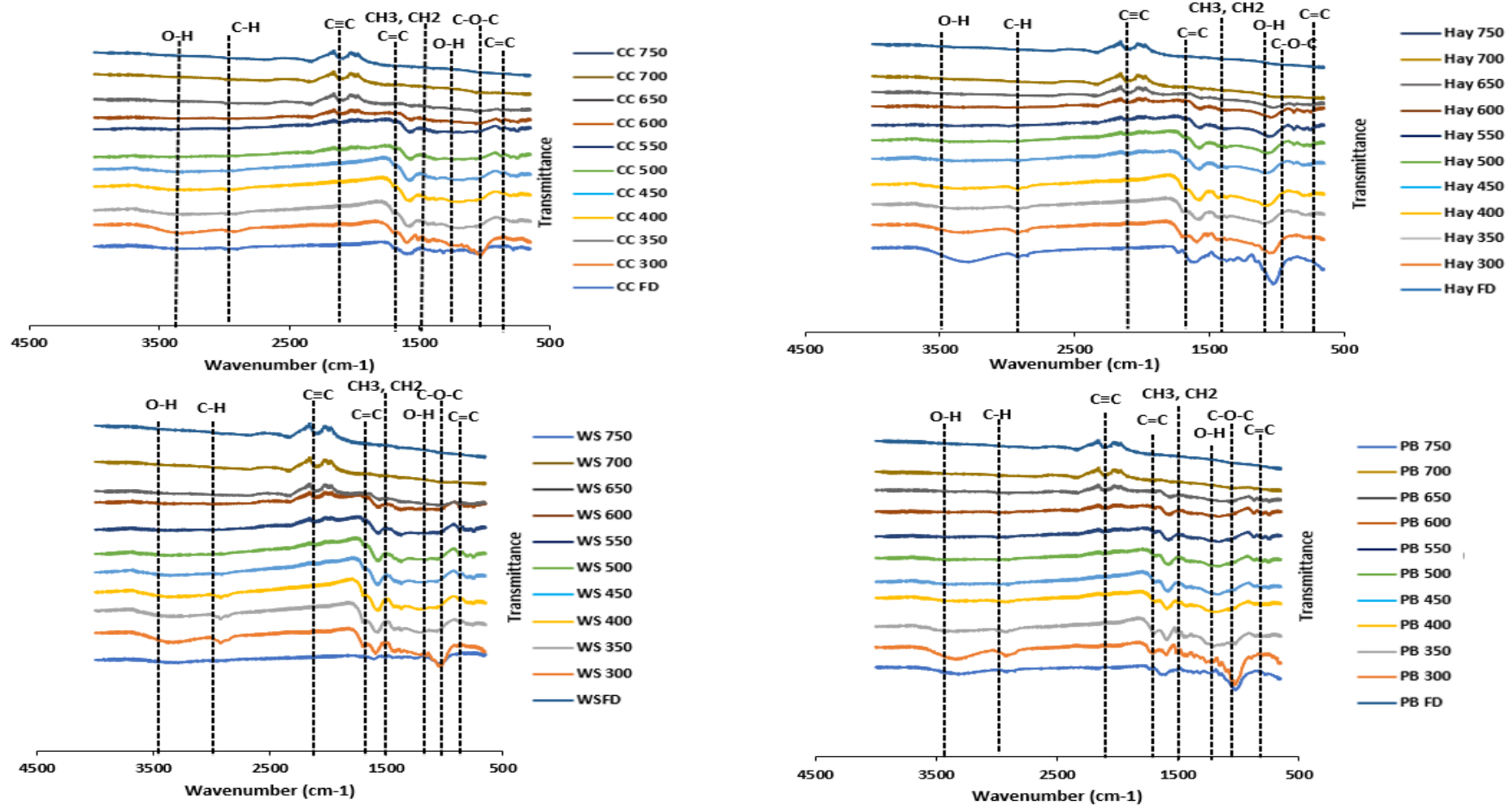


Figure S-5. FTIR spectra for feedstocks (FD) and biochars made from coco coir (CC), hay, wheat straw (WS), and pine bark (PB) at pyrolysis temperatures between 300°C and 750°C

