

Supporting information

Porous calcium copper titanate electrodes for paracetamol degradation by electro-oxidation via CuO-induced peroxymonosulfate activation

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S1. Characterization of the synthesized nanocomposites

The pellet surface morphology was analyzed using a Hitachi S4800 scanning electron microscope (SEM) and a three-dimensional (3D) optical microscope (VHX-7000, KEYENCE, Osaka, Japan). Elemental mapping and the crystalline phase were determined using a Zeiss EVO HD15 microscope with an Oxford X-MaxN EDX detector, and a PANalytical Xpert-PRO diffractometer with an Xcelerator detector using Ni-filtered Cu-radiation with a wavelength of 1.54 Å (scan step size = 0.0020889°/step, time per step = 200.660 sec, and $2\theta = 20^\circ$ – 80°), respectively. The electrode phase composition was determined using Rietveld refinement and the FullProf software⁴⁷ with the profile function 7. LaB₆ structure refinement was used as standard to obtain the instrument resolution function. Raman spectra and mapping images were recorded using dispersive Raman spectroscopy (HORIBA LABRAM, $\lambda = 659$ nm). The laser power was fixed at 20 W with the following acquisition conditions: continuous mode of 10 s, snapshot time of 7 s, and 2.5 accumulations set to 30 times. The pellet surface

elemental composition and the pellet porosity were determined by X-ray photoelectron spectroscopy (XPS) using a monochromatic X-ray source (Al-K α , 1486.6 eV – Resolution FWHM 0.45 eV), and mercury porosimetry (Quantachrome Autoscan 500 porosimeter), respectively. Room-temperature continuous wave (cw) Electron Paramagnetic Resonance (EPR) measurements were carried out at X-band frequencies (9.78 GHz) using a Bruker B-ER420 spectrometer upgraded with a Bruker ECS 041XG microwave bridge and a lock-in amplifier (Bruker ER023M) that relies on a Bruker TE₁₀₂ resonator with modulation amplitude of 1 G, modulation frequency of 100 kHz, and attenuation of 30 dB for the microwave power. Samples were measured in 2.9 mm outer diameter quartz tubes (filling height of ~10 mm) that contained 0.08-0.09 g powdered sample. The optical properties were studied using photoluminescence and an optical fiber spectrometer (Ocean Optics usb2000) after excitement with a nitrogen Nd:YAG laser (266 nm, 10 mW, 1 kHz) in the 350–900 nm range.

S2. Toxicity test

V. fischeri was reconstituted by adding 5 mL of reagent diluent at 5 °C. Then, the reagent was stabilized by transferring 200 μ L of the solution to the cuvettes at 15 °C for 15 min. *V. fischeri* bacteria were activated by adding 22% NaCl solution to the mixture to dilute the initial sample concentration to 0.27%. This dilution was selected to increase the bioluminescence measurement sensitivity and the detection of acute toxicity, even at extremely low concentrations of toxic compounds. Indeed, *V. fischeri* activity and consequently luminescence can be decreased by toxic compounds present in the solution.

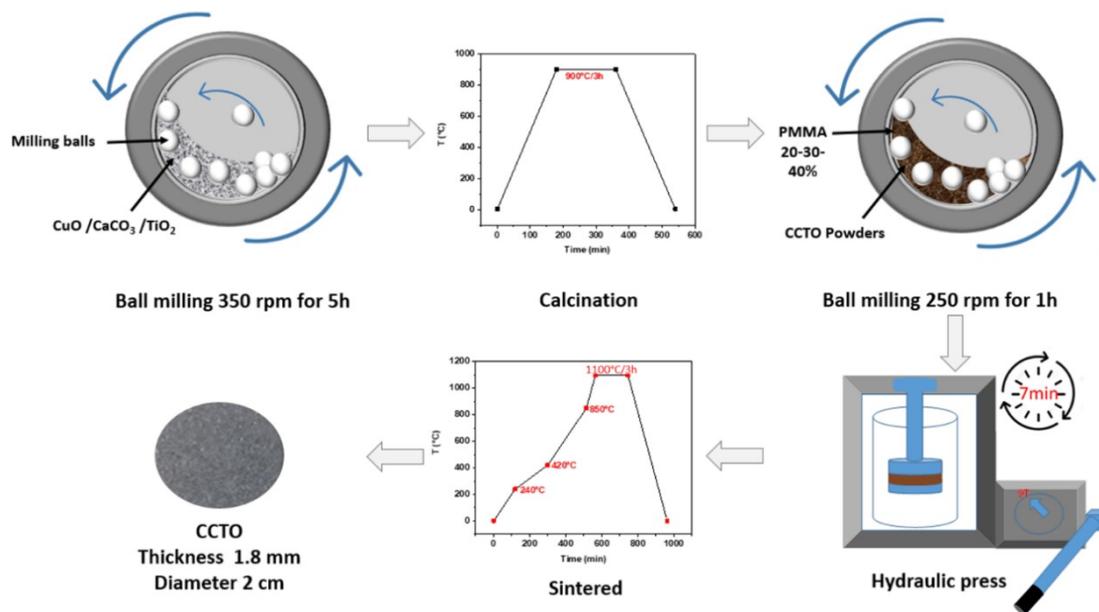


Figure S1. Schematic illustration the production of CCTO samples using different amounts of PMMA as pore-forming agent.

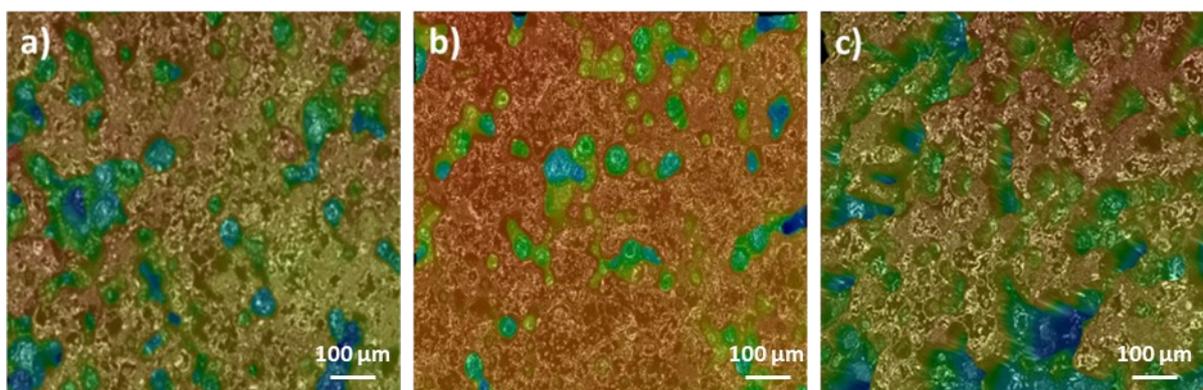


Figure S2. 3D optical microscopy images of a) CCTO-20, b) CCTO-30, and c) CCTO-40.

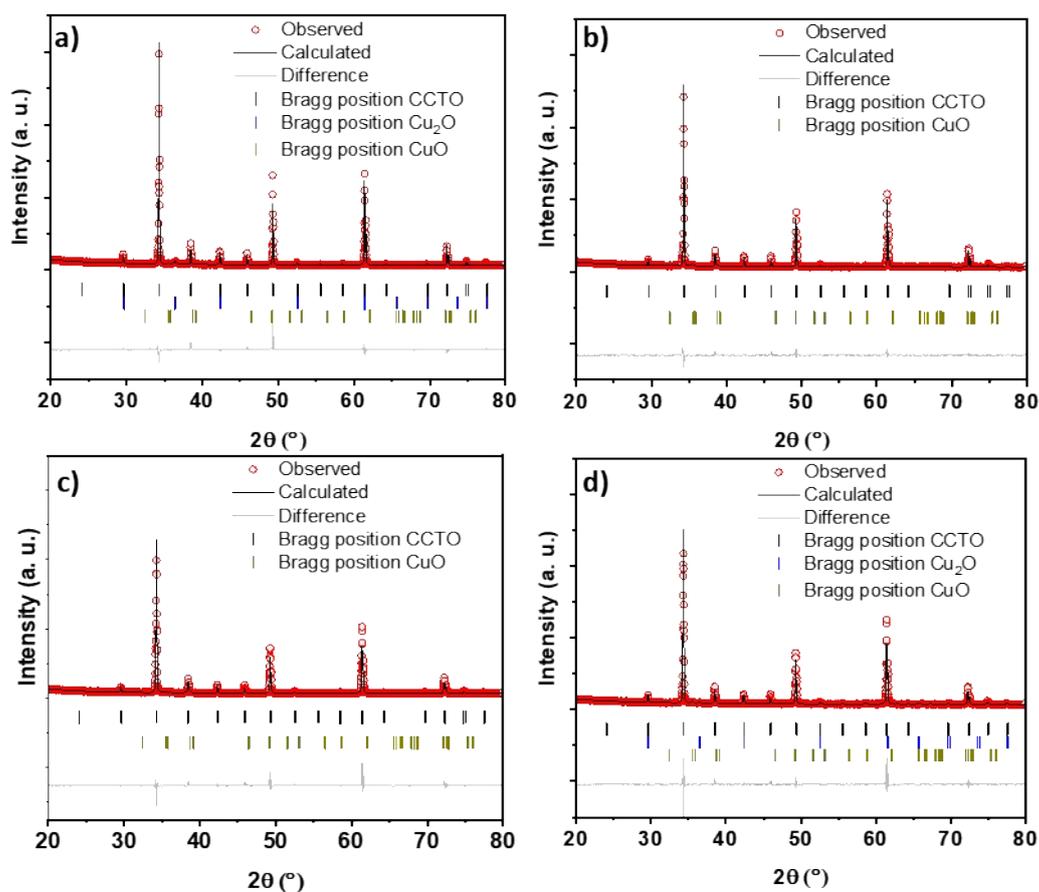


Figure S3. Structure refinement analysis based on the room-temperature XRD data for a) CCTO, b) CCTO-20, c) CCTO-30, and d) CCTO-40 powdered samples obtained by grinding the sintered pellets.

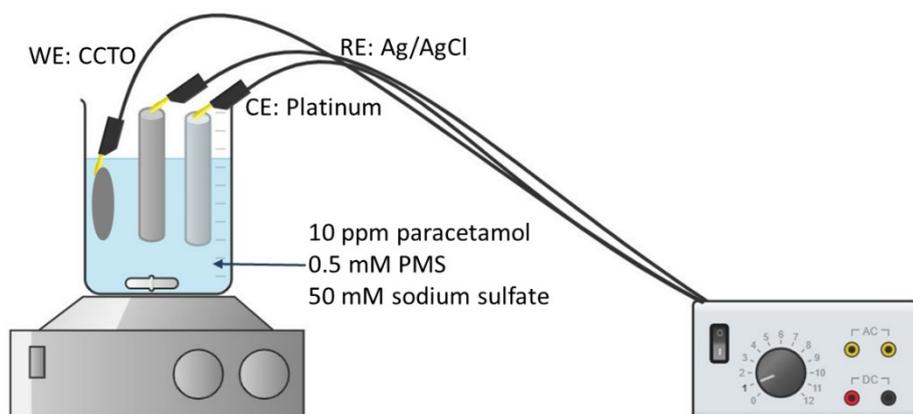


Figure S4. Electrocatalysis process. WE: working electrode; RE: reference electrode; CE: counter electrode.