

Supporting Information

Behaviour of Advanced Materials in environmental aquatic media - Dissolution kinetic and dispersion stability of perovskite automotive catalysts

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Table of Contents

Fig. S1 (a)Uv-vis absorption spectrum of LaCoNi; (b) DLS of LaCoNi stock dispersion in water; (c) XRD spectrum showing the perovskite-like crystal lattice ABO_3 , with minor phases of La_2O_3 and La(OH)_3 ; (d)XPS analysis of LaCoNi showing Nickel enrichment in surface composition. (e), (f) and (g) are SEM images of LaCoNi, LaCoNi_Pd, LaCoNi_Pt, respectively	3
Fig. S2 Log-log linear dependency of Dissolution rates vs SA/V	3
Fig. S3 Dispersion stability of LaCoNi under TG318 conditions	4
Table S1 Basic perovskites physicochemical properties	5
Table S2 Dissolution rates at different flow rates and amount of loaded material in EPAs.....	5

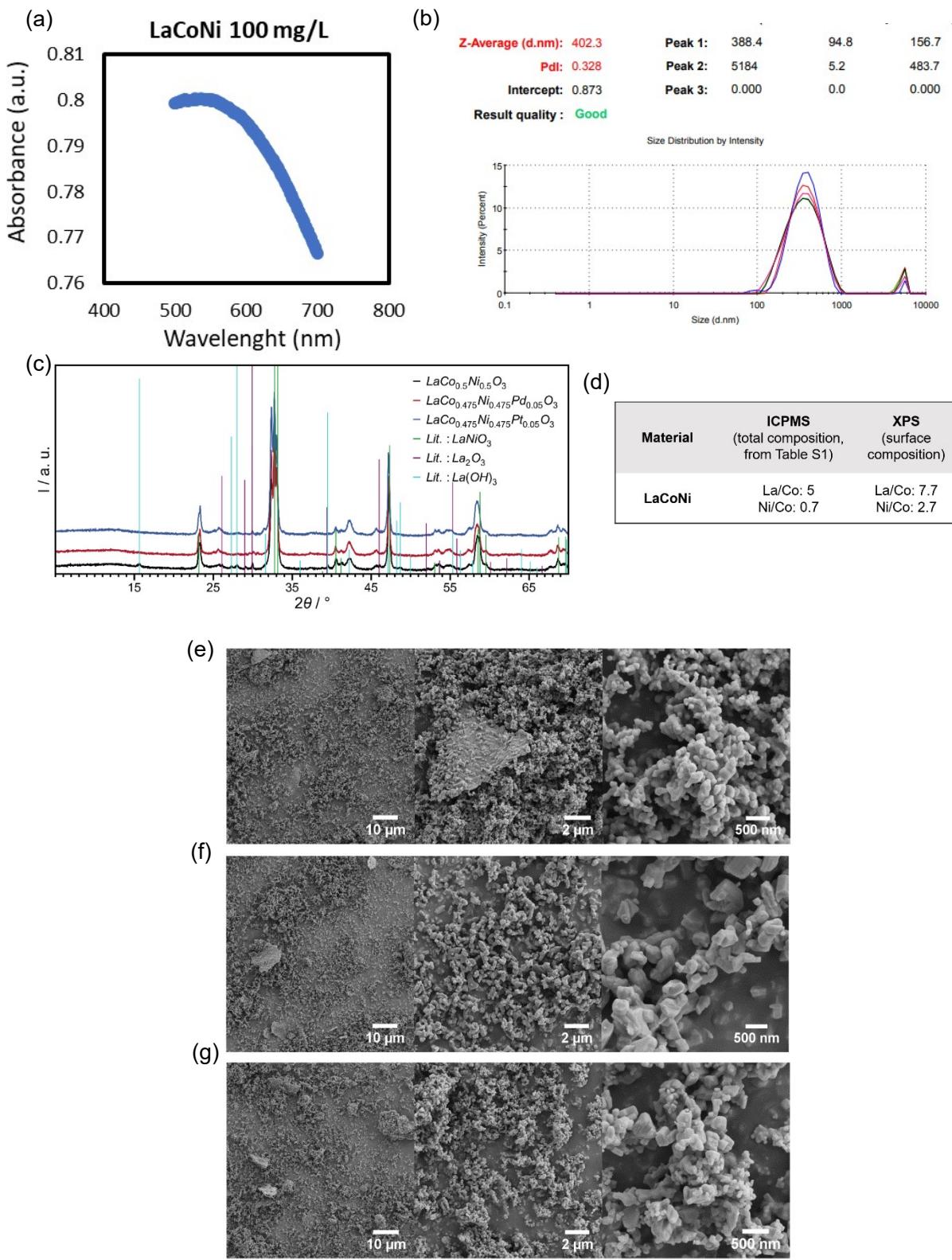


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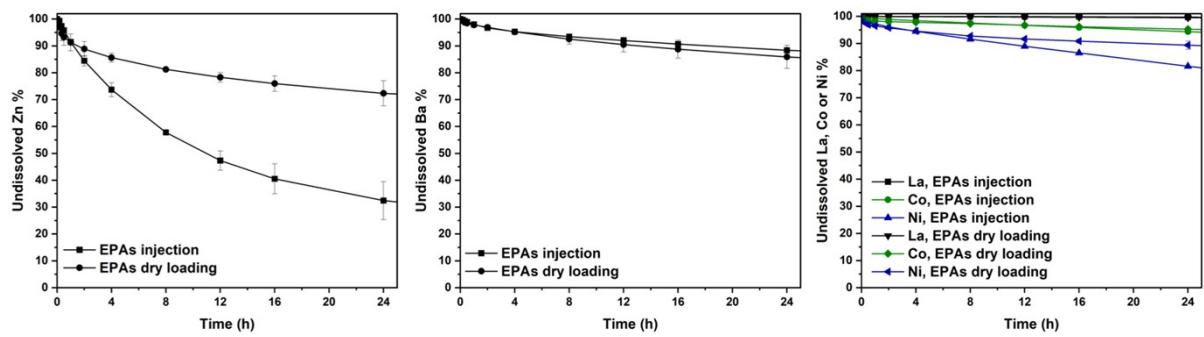


Fig. S2 Mass loss in time expressed in Undissolved metal percentage. Data are reported for EPAs in both dry loading and injection methods; from the left: ZnO, BaSO₄, LaCoNi.

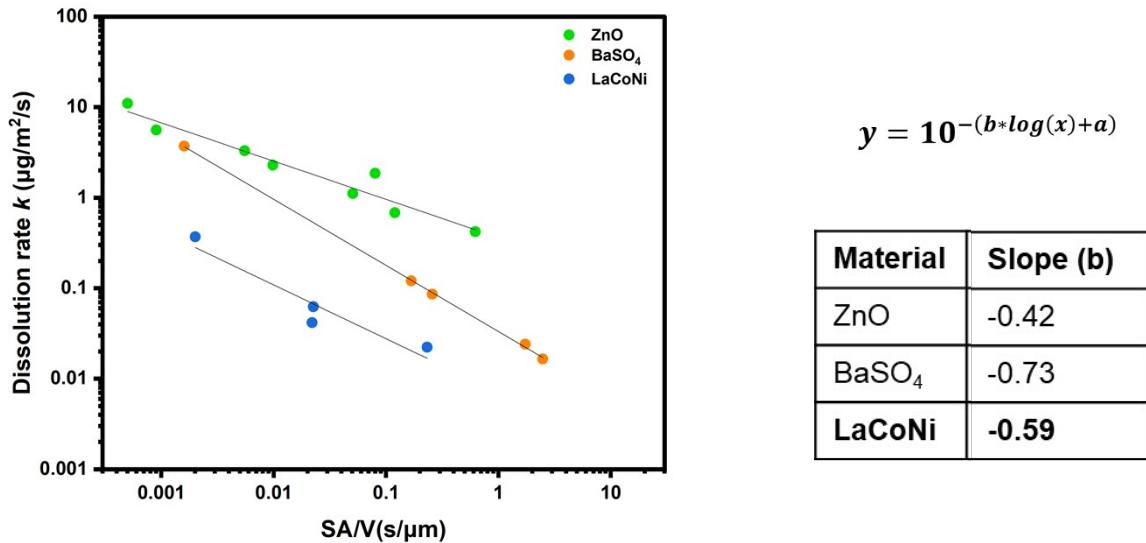


Fig. S3 Log-log linear dependency of Dissolution rates vs SA/V

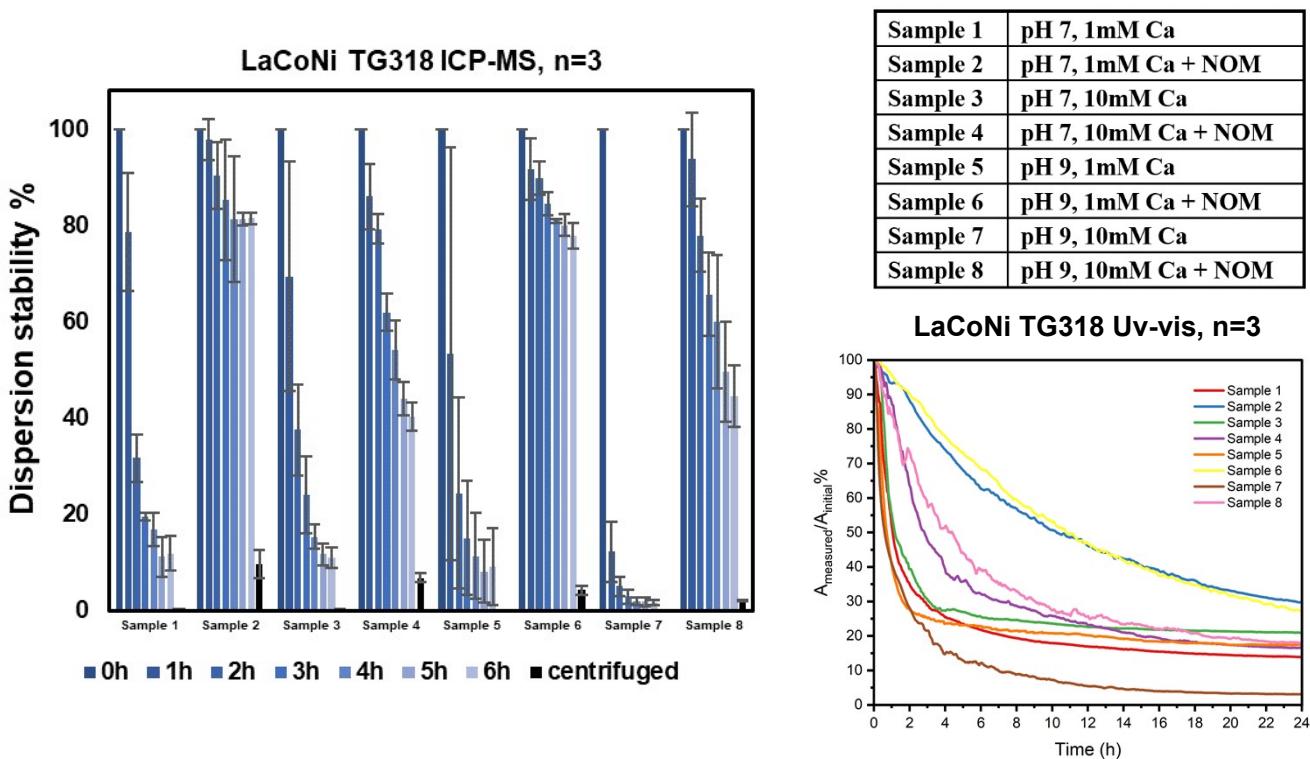


Fig. S4 Dispersion stability of LaCoNi under selected TG318 conditions

Table S1 Basic perovskites physicochemical properties: BET surface area, Min Feret diameter from SEM images, ICPMS characterization of the pristine particles, including dopants.

Material	BET (m ² /g)	Min Feret diameter in nm (SEM)*	ICP-MS				
			Co/wt.-%	La/wt.-%	Ni/wt.-%	Pd/wt.-%	Pt/wt.-%
LaCoNi	3.8	96.1	12.0	60	8.0	n.d.	n.d.
LaCoNi Pd	2.3	206.5	12.0	60	7.9	0.59	n.d.
LaCoNi Pt	4.6	106.6	12.0	60	7.5	n.d.	1.1

* $\sigma_g = 1.05$

Table S2 Dissolution rates at different flow rates and amount of loaded material in EPAs

	Flow rate (mL/min)	ZnO_1µg	ZnO_10µg	ZnO_100µg	ZnO_256µg	BaSO ₄ _1µg	BaSO ₄ _10µg	BaSO ₄ _100µg	LaCoNi_10µg	LaCoNi_100µg
Dissolution rate k [µg.m ⁻² .s ⁻¹]	1	11.3±0.6	3.32±0.3	0.85±0.10	0.42±0.12	3.71±0.4	0.12±0.08	0.024±0.004	0.37±0.06	0.031±0.008
Dissolution rate k [µg.m ⁻² .s ⁻¹]	0.5	5.57±0.5	2.29±0.1	0.68±0.11	0.36±0.12	/	/	/	/	/
Dissolution rate k [µg.m ⁻² .s ⁻¹]	0.1	/	1.86±0.06	0.42±0.01	/	/	0.085±0.009	0.016±0.001	0.042±0.003	0.022±0.003

* dissolution rates are the average rates prior reaching 10% of undissolved single metal ion, according to Keller et al. 2020