## **Electronic Supplementary information**

### A novel bio-template route to synthesize enzyme-

# immobilized MOF/LDHs tubular magnetic micromotors and their application in water treatment

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#### 120000 (b) **(a)** 8000 Mg oxide 2p<sub>3/2</sub> 2p<sub>3/2</sub> (1303.9eV) 100000 (73.4eV) (74.4e) Intensity (a.u.) 0009 Intensity (a.u.) 80000 60000 2000 Mg 1s l 2p 40000 1308 1306 1304 1302 1300 74 70 78 76 72 68 Binding energy (eV) Binding energy (eV)

### 1. Figures

Fig. S1 XPS spectra of Al 2p (a), Mg 1s (b).



**Fig. S2** UV-Vis absorption spectra of HRP enzyme tested by the Kaumas Brilliant Blue method (a) and its standard curve (b).

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Fig. S3 Effect of different chromogenic substrates, time (a), pH (b), catalyst concentration (c), TMB concentration (d), temperature (e) and  $H_2O_2$  concentration (f) on the chromogenic reaction system.



**Fig. S4** Degradation rates of catechol, hydroquinone, and tetrabromobisphenol A under the same conditions.



**Fig. S5** Manganese ion leaching during degradation (a), iron ion leaching (b), zinc ion leaching (c).



Fig. S6 Effect of adding MgAl-LDH/MFZ@HRP micromotor on COD removal efficiency of catechol practical wastewater.



Fig. S7 XPS plots of degraded MgAl-LDH/MFZ@HRP micromotors, full spectrum (a), Mn (b), Fe (c).

### 2. Table

Table S1 Pore texture parameters of the obtained samples

Sample	<b>SBET</b> (m <sup>2</sup> g <sup>-1</sup> )	VTotal (cm <sup>3</sup> g <sup>-1</sup> )	Dp (nm)
Mn <sub>2</sub> O <sub>3</sub> /C	44.929	0.160341	14.4125
MgAl-LDH	65.7686	0.157659	10.5368
MgAl-LDH/MFZ@HRP	110.6669	0.25114	8.6179

Table S2 Comparison of speed with other reported micromotors

Micromotors	Concentration of H <sub>2</sub> O <sub>2</sub>	Velocity (µm s <sup>-1</sup> )	Ref.
CuS@Fe <sub>3</sub> O <sub>4</sub> /Pt	5%	423.8	1
Magnetic illite microspheres	5%	100.38	2
MnO <sub>2</sub> microparticles	5%	128	3
Au/ Ni/Pt	5%	37.57	4
Fe-zeolite	10%	84.96	5
rGO/ZnO/BiOI/Co-Pi/Pt	5%	63.1	6
PAPBA/Ni/Pt	3%	40	7
MgAl-LDHs/MFZ	3%	128.33	This work

Table S3 Comparison of various sensor platforms for the detection of catechol

Method	System	LOD (µM)	Linear(µM)	Ref.
Electrochemical	AuNP-MoS <sub>2</sub> -Lac/GCE	2.0	2-2000	8
Electrochemical	FYSSns-2-Lac/GCE	1.6	12.5-450	9
Electrochemical	Fe <sub>3</sub> O <sub>4</sub> -GO-AuNPs	0.8	2-145	10
Colorimetric	ТМВ- $\delta$ -MnO <sub>2</sub>	0.22	0.5-10	11
Colorimetric	Co <sub>1.5</sub> Mn <sub>1.5</sub> O <sub>4</sub>	0.35	1-1000	12
Colorimetric	MgAl-LDH/MFZ@HRP	0.69	0-200	This work

 Table S4 Comparison of COD before and after degradation of actual catechol

 wastewater

Actual water sample status	Testing Program	Test results (mg/L)
pre-degradation	CODCr	192.23
after degradation		49.21

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