

Supplementary material

For

Degradation of ciprofloxacin in UV/NH₂Cl process: kinetics, mechanism, pathways and DBPs formation

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Tables

Table S1. Trihalomethanes and haloacetonitriles analysis – gas chromatograph operating conditions

Parameter	Description
Injector Temperature	Splitless 200°C 15 sec purge activation time
Detector Temperature	300°C
Temperature Program	35°C for 9.0 min 10°C/min temperature ramp to 40°C 40°C for 3.0 min 15°C/min temperature ramp to 150°C 150°C for 1.0 min
Carrier Gas	Helium
Flow Rate	24.8 cm/sec at 150°C

Table S2. Haloacetic acids analysis – gas chromatograph operating conditions

Parameter	Description
Injector Temperature	Splitless 210°C 15 sec purge activation time
Detector Temperature	300°C
Temperature Program	40°C for 2.0 min 1°C/min temperature ramp to 65°C 65°C for 2.0 min 10°C/min temperature ramp to 90°C 150°C for 0 min 30°C/min temperature ramp to 210°C 210°C for 2.0 min
Carrier Gas	Helium
Flow Rate	24.8 cm/sec at 150°C

Table S3. Comparison of degradation efficiency between different systems

Degradation system	Target pollution	Initial concentration	UV lamp power	Oxidant concentration	Degradation time	Reference
UV/Cl ₂	Metoprolol	0.1 mg/L	1.1 mW/cm ²	1.0 mM	120 min	(Nam et al. 2015)
UV/H ₂ O ₂	Cefixime	9 mg/L	36 W	0.85 mM	180 min	(Belghadr et al. 2014)
UV/PS	Thiamphenicol	0.14 mg/L	20 W	1.0 mM	60 min	(Wang et al. 2017)
UV/NH ₂ Cl	Ciprofloxacin	5 mg/L	65 uW/cm ²	0.5 mM	90 min	/

Reference

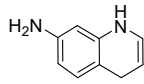
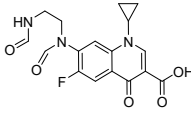
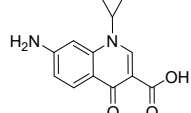
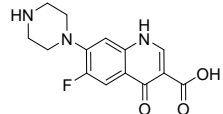
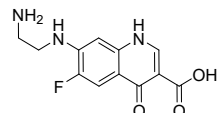
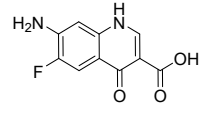
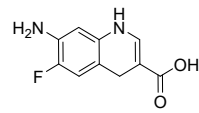
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- [3] Wang, Feige, Wenjing Wang, Shoujun Yuan, Wei Wang, and Zhen-Hu Hu. 2017. 'Comparison of UV/H₂O₂ and UV/PS processes for the degradation of thiamphenicol in aqueous solution', *Journal of Photochemistry and Photobiology A: Chemistry*, 348: 79-88.

Table S4. pH change of UV/NH₂Cl system in pH experiments

	pH				
Before reaction	3.0	5.0	7.0	9.0	11.0
After reaction (Parallel experiment I)	3.12	5.89	6.26	6.85	8.76
After reaction (Parallel experiment II)	3.25	5.74	6.13	7.23	9.12

Table S5. Degradation products of CIP during the UV/NH₂Cl process

Compound	Retention Time(min)	m/z	Formula	Possible chemical structure
CIP	3.59	311.13	C ₁₇ H ₁₈ FN ₃ O ₃	
A	3.18	305.12	C ₁₅ H ₁₆ FN ₃ O ₃	
B	6.98	262.08	C ₁₃ H ₁₁ FN ₂ O ₃	
C	1.61	248.10	C ₁₃ H ₁₃ FN ₂ O ₂	
D	1.61	204.11	C ₁₂ H ₁₃ FN ₂	

E	3.48	146.08	$C_9H_{12}N_2$	
F	5.57	361.11	$C_{17}H_{16}FN_3O_5$	
G	3.79	244.06	$C_{13}H_9FN_2O_2$	
H	5.45	291.10	$C_{14}H_{14}FN_3O_3$	
I	6.98	265.09	$C_{12}H_{12}FN_3O_3$	
J	5.98	222.04	$C_{10}H_7FN_2O_3$	
K	4.42	208.06	$C_{10}H_9FN_2O_2$	

Figures

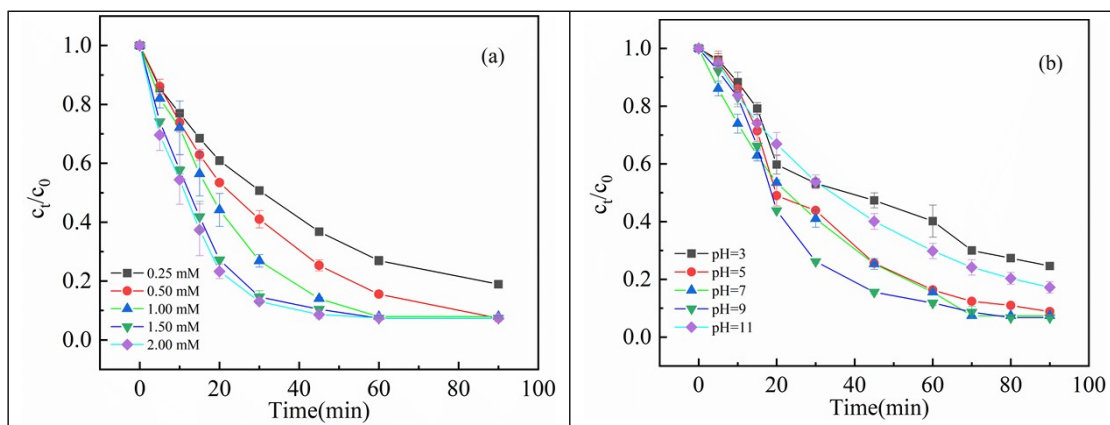


Fig. S2 Effect of (a) initial NH_2Cl concentration, (b) initial pH, of different concentrations on the degradation of CIP by UV/ NH_2Cl . Conditions: $[\text{CIP}] = 19.58 \mu\text{M} = 5\text{mg/L}$, $[\text{NH}_2\text{Cl}] = 0.25 \sim 2 \text{ mM}$, $\text{pH} = 5.0 \sim 10.0$, $[\text{NOM}] = 0 \sim 4 \text{ mg/L}$, $[\text{HCO}_3^-] = 0 \sim 4 \text{ mM}$, $[\text{CO}_3^{2-}] = 0 \sim 4 \text{ mM}$, $[\text{Cl}^-] = 0 \sim 4 \text{ mM}$, $[\text{NO}_3^-] = 0 \sim 4 \text{ mM}$, $[\text{TBA}] = 0 \sim 50 \text{ mM}$, $I_s = 65 \mu\text{W}/\text{cm}^2$.

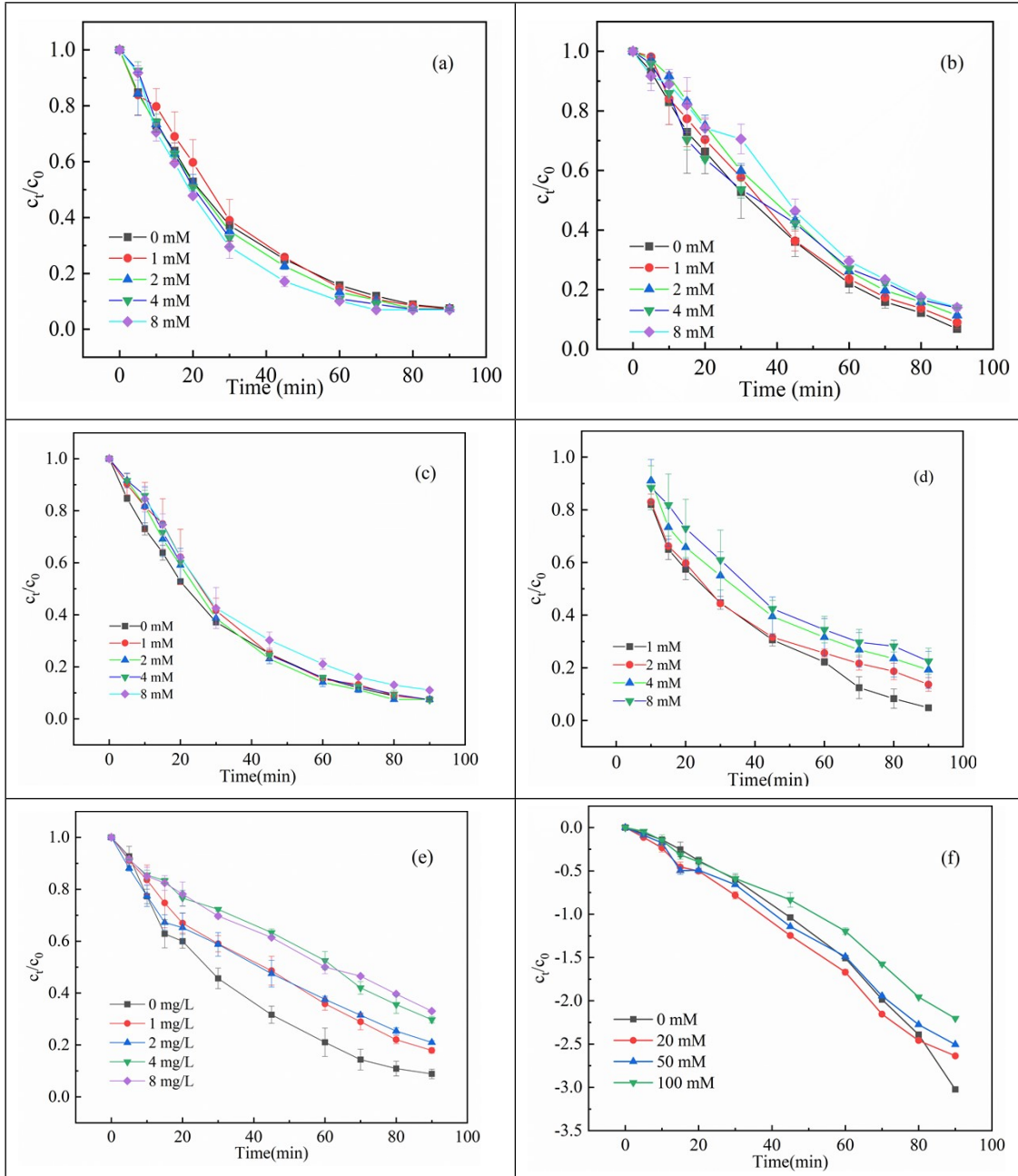
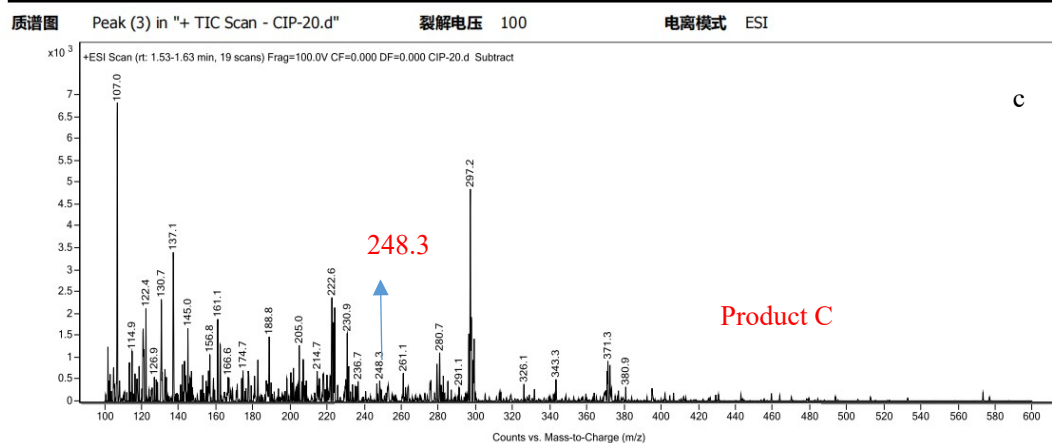
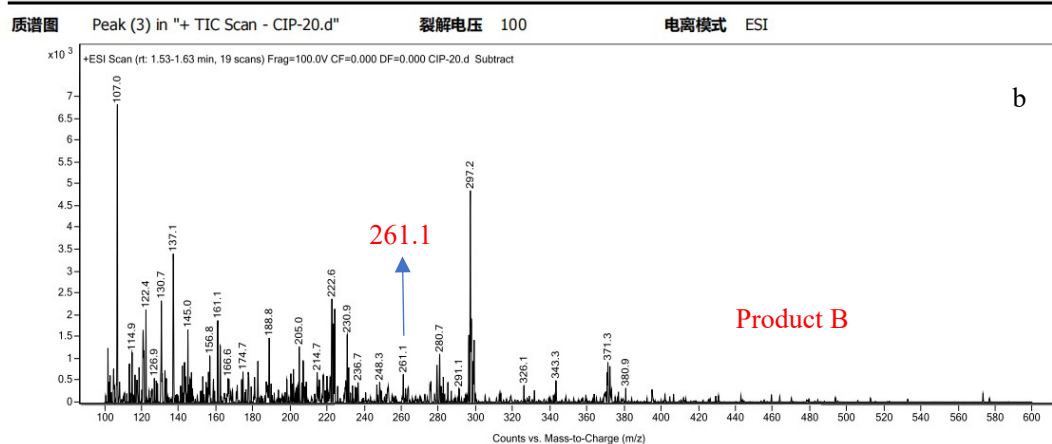
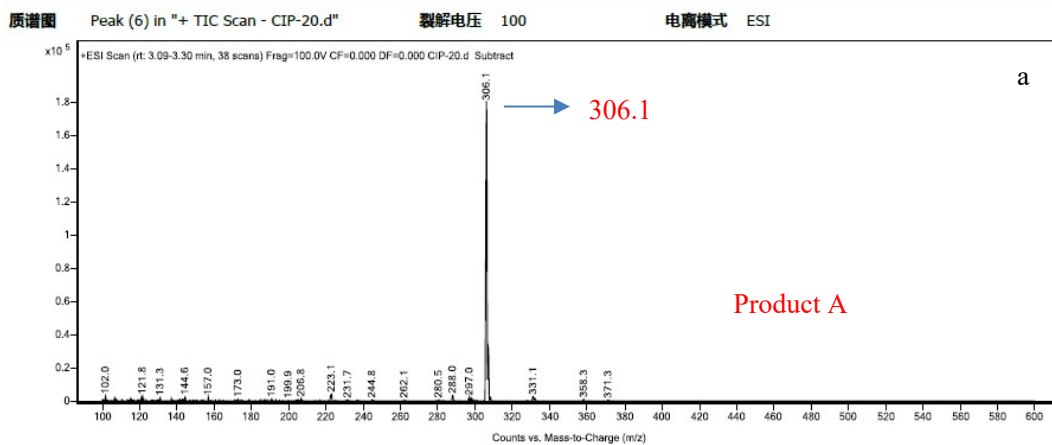
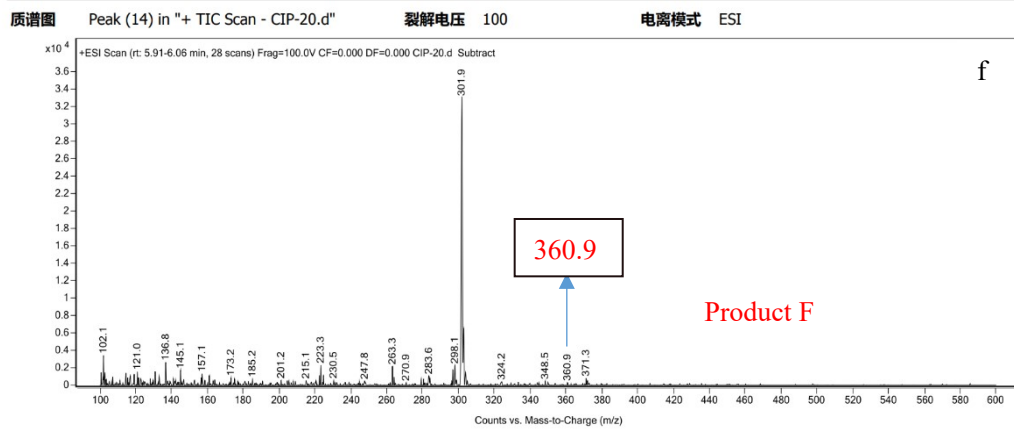
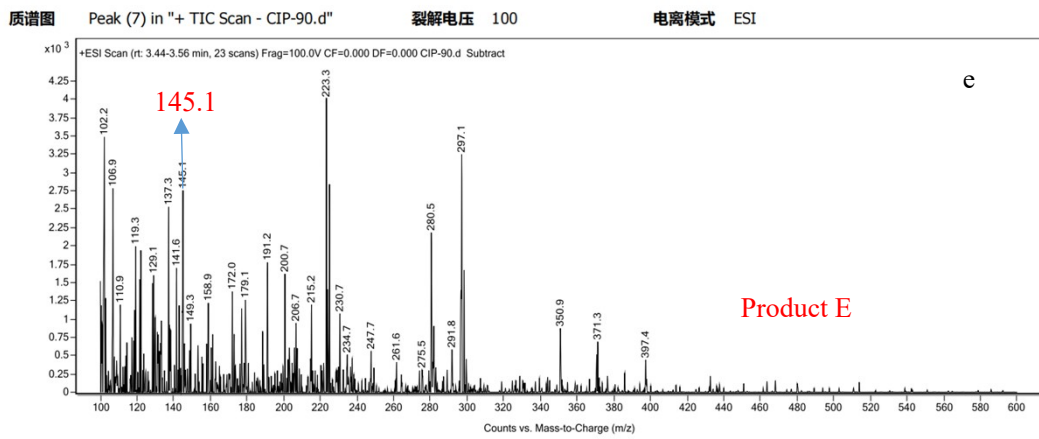
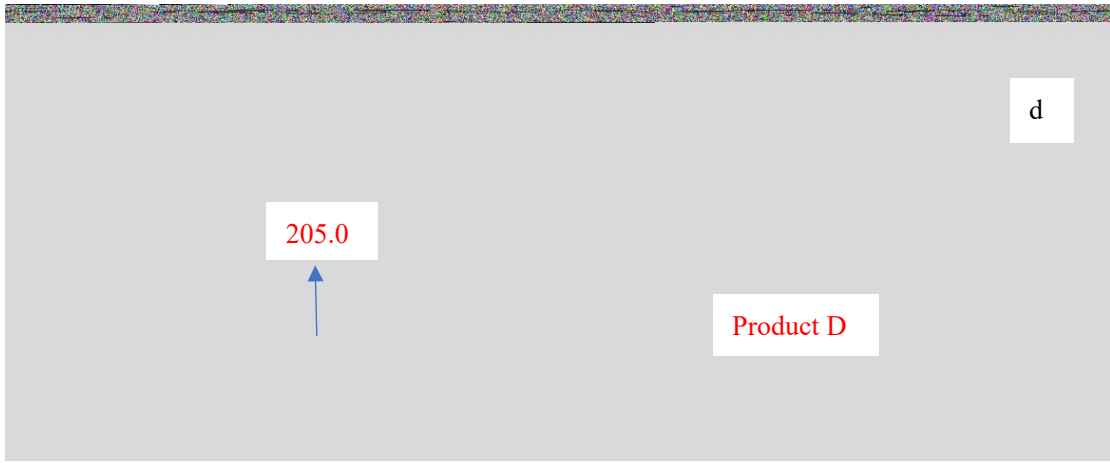
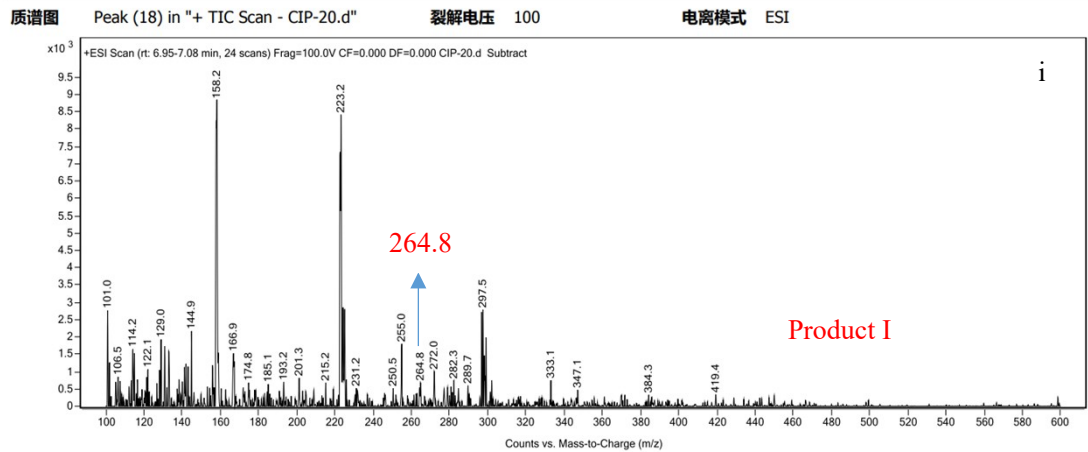
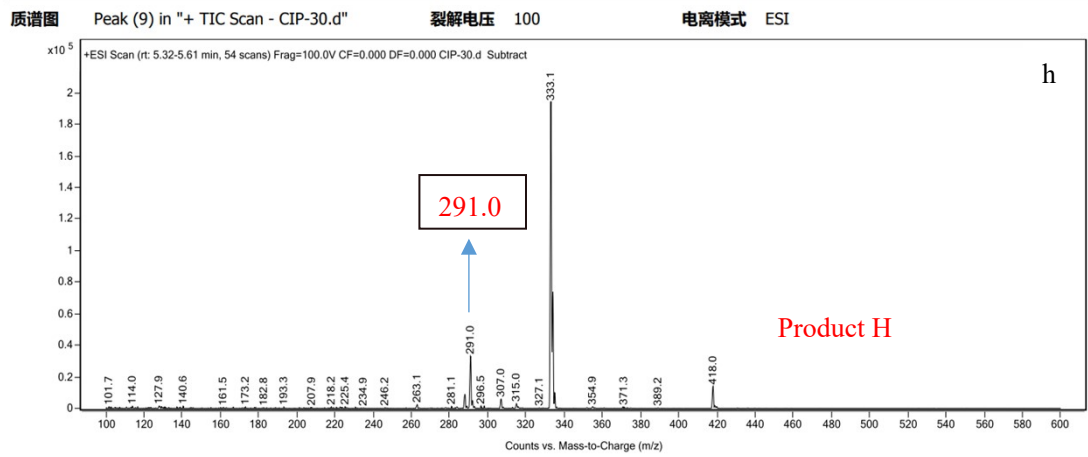
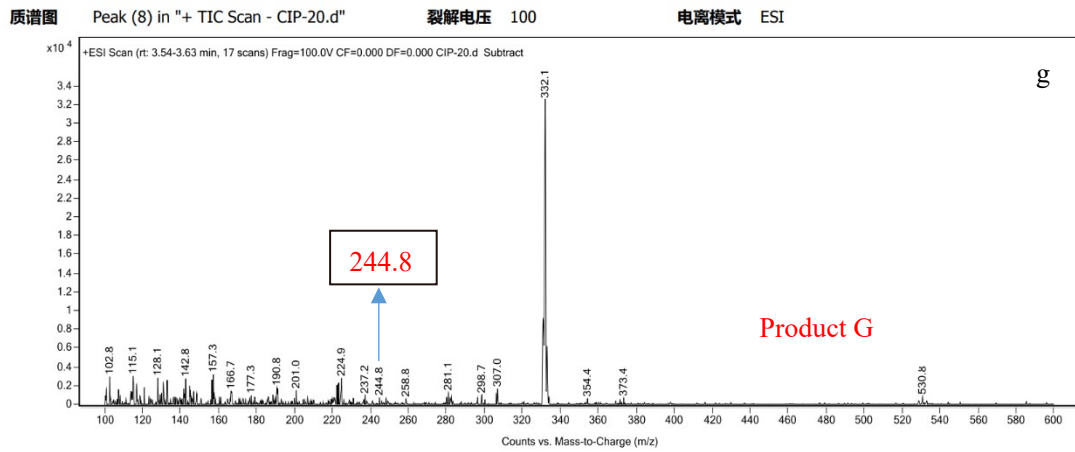
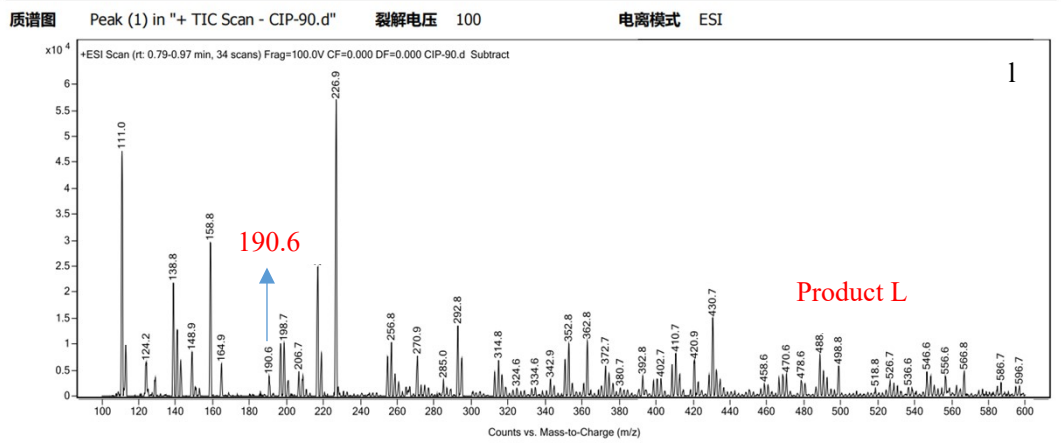
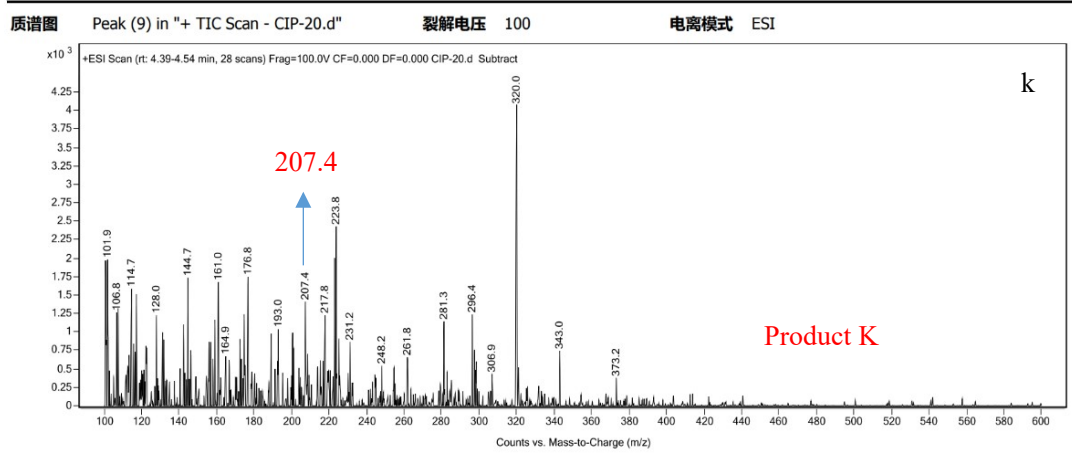
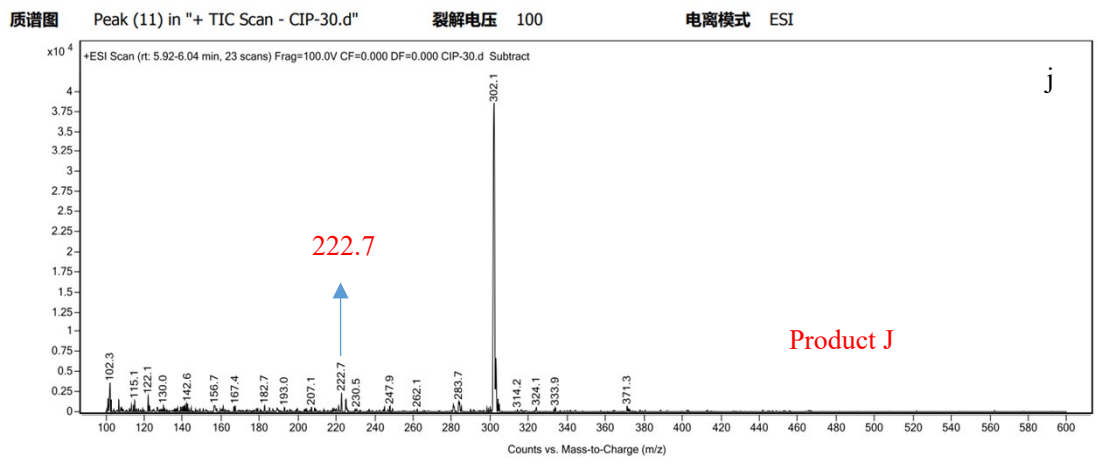


Fig. S3 Effect of (a) Cl^- , (b) NO_3^- , (c) HCO_3^- , (d) CO_3^{2-} , (e) NOM, (f) TBA of different concentrations on the degradation of CIP by UV/ NH_2Cl . Conditions: $[\text{CIP}] = 19.58 \mu\text{M} = 5\text{mg/L}$, $[\text{NH}_2\text{Cl}] = 0.25 \sim 2 \text{ mM}$, $\text{pH} = 5.0 \sim 10.0$, $[\text{NOM}] = 0 \sim 4 \text{ mg/L}$, $[\text{HCO}_3^-] = 0 \sim 4 \text{ mM}$, $[\text{CO}_3^{2-}] = 0 \sim 4 \text{ mM}$, $[\text{Cl}^-] = 0 \sim 4 \text{ mM}$, $[\text{NO}_3^-] = 0 \sim 4 \text{ mM}$, $[\text{TBA}] = 0 \sim 50 \text{ mM}$, $I_s = 65 \mu\text{W}/\text{cm}^2$.









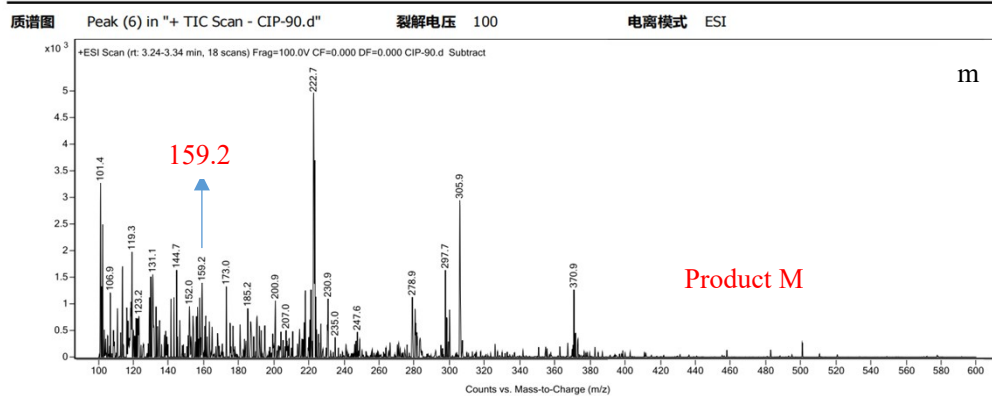


Fig. S3. The mass spectrograms of intermediate products of A to M (a-m).