

1 Supplemental Information

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3 Membrane Coating Methods.

4 *Attachment of zeolite procedure using UV (Attachment – UV method).* Faujasite zeolite nano particles
5 were made using the following procedure. 1.1 g of $\text{Na}_2\text{Al}_2\text{SO}_4$ (Sigma Aldrich) in 10 mL of water was
6 mixed with 2.4 g of NaOH and 18.1 g of Na_2SiO_3 solution (Sigma Aldrich) were chilled in an ice bath
7 for 30 min. The $\text{Na}_2\text{Al}_2\text{SO}_4$ mixture was added slowly added to Na_2SiO_3 . After aging for 24 hours at
8 room temperature, the mixture was freeze dried and then crystalized for 2 days at 50 °C. Solution
9 was centrifuged, washed, and dried to produce a zeolite powder.

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11 PSf membranes were prepared as described by Binahmed et al.² Zeolite was attached to the
12 membranes using a similar procedure as described by Kulak et al.⁴ Briefly, zeolite was dried for 24
13 hours at 100 °C using a Schlenk Line (100 millitorr) and treated with 3-aminopropyltriethoxysilane
14 (APTES) (2 mM) (Sigma Aldrich) in toluene for 1 hour at 70 °C under nitrogen. PSf membranes
15 were irradiated with a UV lamp (Spetroline Model EF-160C) for 15 seconds to create carboxylic
16 groups on the membrane surface and treated with [3-(2,3epoxypropoxylpropyl]-trimethoxysilane
17 (GLYMO) (2 mM) (Sigma Aldrich) in iso-octane for 1 hour at room temperature under nitrogen.

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19 The treated zeolite was dried at 70 °C overnight and resuspended in iso-octane and applied to the
20 treated PSf membrane for 1 hour while mixing. Coated membranes were rinsed and sonicated for 20
21 seconds to remove non-attached particles. Membranes were stored at 4 °C in deionized (DI) water.

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23 *Attachment of zeolite procedure using UV and poly(acrylic acid (PAA) (Attachment – AA/UV*
24 *method).* The previous procedure was modified to apply 10% acrylic acid (AA) solution from stock
25 99% AA (Sigma Aldrich) in MilliQ to the PSf membrane immediately after membrane oxidation with

26 UV exposure for the growth of PAA and enhance the coverage of carboxylic groups on the membrane
27 surface.³ AA (10%) was applied for 5 minutes and immediately rinsed with MilliQ water. A more
28 detailed procedure is described in Wuolo-Journey et al.³ The procedure continued as described in the
29 attachment procedure using UV.

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31 *Attachment of zeolite procedure using EDC-NHS (Attachment – EDC/NHS).* Beginning with UV/AA
32 treatment, this procedure then uses 4 mM 1-ethyl-3-(3-dimethylaminopropyl)carbodiimide
33 hydrochloride (EDC) (Sigma Aldrich) and 10 mM N-hydroxysuccinimide (NHS) (Sigma Aldrich),
34 adjusted to pH 5 with 10 mM MES (Sigma Aldrich) supplemented with 0.5 M NaCl. This solution is
35 applied to PSf membrane while mixing at ambient conditions for one hour.^{3,5} Membrane was then
36 rinsed and zeolite applied as described above.

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38 *Attachment of zeolite procedure using polydopamine coating (Attachment – PDA).* PSf membranes
39 were prepared and stored as described earlier.² Dopamine solution was prepared using 4 g of 3,4-
40 dihydroxyphenylalanine (DOPA) (Sigma Aldrich) in 1 L of Trizma buffer (10 mM) (Sigma
41 Aldrich).^{2,8} Once mixed, 50 mg of zeolite, prepared as described above, was added to the solution.
42 The pH was adjusted to 8.5 using 1 M NaOH. Dopamine solution was immediately transferred to the
43 membrane to initiate PDA deposition for 15 minutes while mixing. Membranes were then
44 immediately rinsed thoroughly with MilliQ and stored in MilliQ at 4 °C.

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46 *Embedment of zeolite into PSf membranes (Embedment).* PSf membranes were prepared as described
47 previously.² 50 mg of zeolite nano powder (prepared as described previously) was added directly to
48 the dope solution after solution was stored overnight to remove air bubbles. After the membranes
49 were cast, they were stored in MilliQ at 4 °C until use.

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51 *Growth of zeolite on alumina hollow fiber membranes (Growth).* Growth of faujasite zeolite on
 52 alumina hollow fiber membranes (0.7 mm diameter) is described in previous work.⁷ Briefly, raw
 53 fibers of 38.0% N-Methyl-2-pyrrolidone (NMP), 6.8% polyethersulfone (PES), 54.7% Al₂O₃, and
 54 05% polyvinylpyrrolidone (PVP) were sintered to make alumina hollow fiber membranes. Zeolite
 55 growth on the membrane surface occurred by immersing in a solution of 7.7 M NaOH, colloidal silica
 56 (Ludox TM-40, Sigma Alrich), and 0.15 of aluminum for 24 hours at 75 °C. Plain and zeolite-coated
 57 membranes were stored dry in sterile containers at ambient conditions until use.

58

59 **Table S1. Synthetic wastewater components in mg per liter for CFTR and ACFTR**

	CFTR	ACFTR	
NH ₄ Cl	133.75	133.75	mg
NaNO ₂	103.5	0	mg
Magnesium Phosphate Dibasic Trihydrate	25	25	mg
Potassium Phosphate Tribasic	20	20	mg
NaHCO ₃	275	275	mg
Sodium acetate	221	221	mg
Bacteriological peptone	24	24	mg
Dry meat extract	12	12	mg
Potato starch	42	42	mg
Low fat milk powder	50	50	mg
Glycerine	34	34	mg

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61 **Table S2. Synthetic wastewater components in mg per liter for MFTR and Bioavailability Test**

	MFTR	Bioavailability Test	
NH ₄ Cl	133.75	0	mg

NaNO ₂	0	236.4	mg
KH ₂ PO ₄	27.2	27.2	mg
KHCO ₃	500	500	mg
Trace solution 1	1	1	mL
Trace solution 2	1	1	mL
Mg solution	1	1	mL
Ca solution	1	1	mL

62 †Trace solutions are detailed in Peterson et al.²²

63 Table S3. Constants of isotherm equations

q_e	Ammonium sorbed per unit mass of membrane or carrier
q_{max}	Maximum sorption capacity
k	Adsorption equilibrium constant
c_f	Concentration of ammonium in solution
k_f	Fruendlich constant
n	Empirical isotherm constant

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66 References

- 67 1. J. Gascon, F. Kapteijn, B. Zornoza, V. Sebastián, C. Casado and J. Coronas, Practical Approach
68 to Zeolitic Membranes and Coatings: State of the Art, Opportunities, Barriers, and Future
69 Perspectives, *Chemistry of Materials*, 2012, **24**, 2829-2844.
- 70 2. S. Binahmed, A. Hasane, Z. Wang, A. Mansurov and S. Romero-Vargas Castrillón, Bacterial
71 Adhesion to Ultrafiltration Membranes: Role of Hydrophilicity, Natural Organic Matter, and
72 Cell-Surface Macromolecules, *Environmental Science and Technology*, 2018, **52**, 162-172.
- 73 3. K. Wuolo-Journey, S. Binahmed, E. Linna and S. Romero-Vargas Castrillón, Do graphene oxide
74 nanostructured coatings mitigate bacterial adhesion?, *Environmental Science: Nano*, 2019, **6**,
75 2863-2875.

- 76 4. A. Kulak, Y.-J. Lee, Y. S. Park and K. B. Yoon, Orientation-Controlled Monolayer Assembly of
77 Zeolite Crystals on Glass and Mica by Covalent Linkage of Surface-Bound Epoxide and Amine
78 Groups, *Angewandte Chemie*, 2000, **112**, 980-983.
- 79 5. H. Croll, A. Soroush, M. E. Pillsbury and S. R.-V. Castrillón, Graphene oxide surface
80 modification of polyamide reverse osmosis membranes for improved N-nitrosodimethylamine
81 (NDMA) removal, *Separation and Purification Technology*, 2019, **210**, 973-980.
- 82 6. E. C. Feinberg, A. L. H. Chester, P. J. Novak and M. A. Hillmyer, Porous Polyethylene-
83 Supported Zeolite Carriers for Improved Wastewater Deammonification, *ACS ES&T*
84 *Engineering*, 2021, **1**, 1104-1112.
- 85 7. A. L. Huff Chester, K. Eum, M. Tsapatsis, M. A. Hillmyer and P. J. Novak, Enhanced Nitrogen
86 Removal and Anammox Bacteria Retention with Zeolite-Coated Membrane in Simulated
87 Mainstream Wastewater, *Environmental Science & Technology Letters*, 2021, **8**, 468-473.
- 88 8. J. Xue, S. Binahmed, Z. Wang, N. G. Karp, B. L. Stottrup and S. Romero-Vargas Castrillón,
89 Bacterial Adhesion to Graphene Oxide (GO)-Functionalized Interfaces Is Determined by
90 Hydrophobicity and GO Sheet Spatial Orientation, *Environmental Science & Technology Letters*,
91 2018, **5**, 14-19.
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