Electronic Supplementary Material (ESI) for Environmental Science: Water Research & Technology. This journal is © The Royal Society of Chemistry 2023

## Comparison of different coagulants to improve membrane

## distillation performance for landfill leachate concentrate treatment

## **Supplementary Data**

Revised manuscript to Environmental Science: Water Research & Technology

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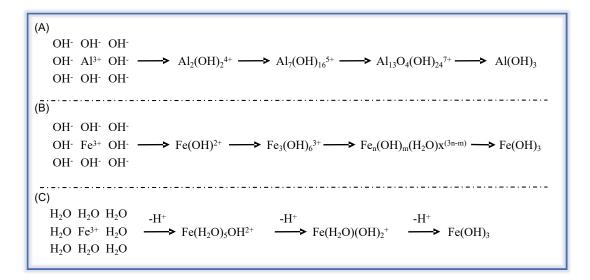
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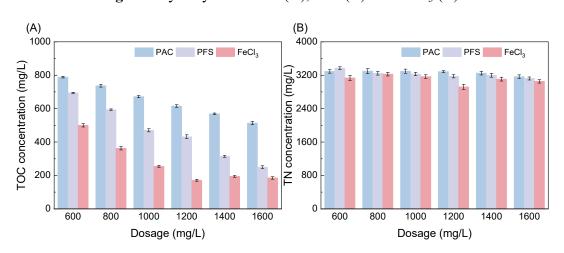
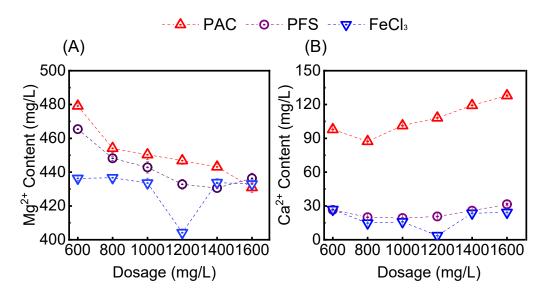
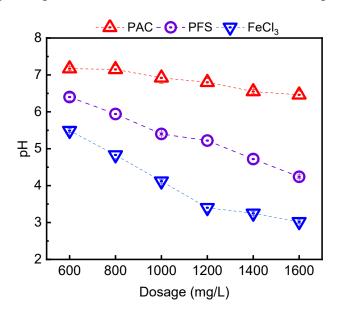


Fig. S1: Hydrolysis of PAC (A), PFS (B) and FeCl<sub>3</sub> (C).

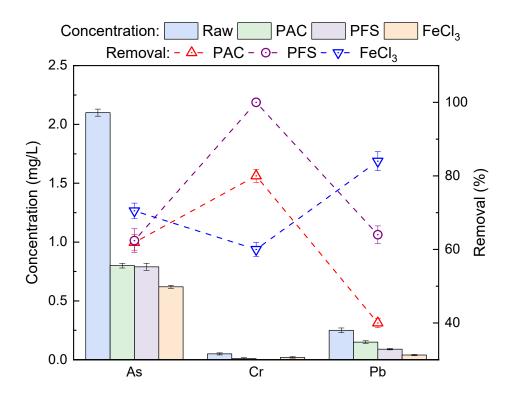
**Fig. S2:** Effect of coagulation on the concentration of TOC (A) and TN (B) in landfill leachate concentrate. Three coagulants, including PAC, PFS and FeCl<sub>3</sub>, were added to leachate concentrate and then mixed rapidly at 250 r/min for 2 min, slowly at 60 r/min for 10 min before natural settlement for approximately 30 min under the room temperature.



**Fig. S3:** Removal of  $Mg^{2+}$  and  $Ca^{2+}$  from landfill leachate concentrate via coagulation at different dosages. Experimental conditions were shown in the caption of Fig. S2.



**Fig. S4:** Effect of coagulation on the pH of landfill leachate concentrate. Experimental conditions were shown in the caption of Fig. S2.



**Fig. S5:** Effect of coagulation on the removal of heavy metals in landfill leachate concentrate by coagulation and their concentration at 1200 mg/L. Experimental conditions were shown in the caption of Fig. S2.

Element/ Wt (%)	Pristine	Raw	PAC	PFS	FeCl <sub>3</sub>
F	70.1	0.6	35.9		65.7
С	29.9	6.1	24.3		27.2
0		31.1	15.5	38.6	4.5
Na		17.4	5.8	18.1	1.3
Cl		20.5	5.4	16.1	0.6
Ν		11.8	4.2		
K		10.7	2.8	10.9	0.5
Ca		1.1	0.9		
Mg		0.7	0.7	2.0	0.3
Si			3.6	1.0	
S			0.2	1.3	
Fe				5.1	
Al			0.5		

**Table S1:** Elements onto the pristine and fouled membrane after MD operation for
 landfill leachate concentrate treatment via EDS measurement