

Supporting Information

Solvent-free mechanochemical chlorination of pyrazoles with trichloroisocyanuric acid

*Chi-Min Chen, Jia-Xin Chen and Ching Tat To**

Department of Chemistry, Tunghai University, No. 1727, Sec. 4, Taiwan Boulevard, Xitun District, Taichung City 407224, Taiwan, Republic of China

Email: ctto@go.thu.edu.tw

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Experimental Section

Unless otherwise specified, all reagents were purchased from commercial suppliers and directly used without further purification. All reactions were carried out in 10 mL zirconia milling vessel designated for the Retsch MM400 shaker mill. All reactions were repeated once to ensure reproducibility. Silica gel (Merck, 70–230 mesh) was used.

^1H , $^{13}\text{C}\{^1\text{H}\}$ and $^{19}\text{F}\{^1\text{H}\}$ NMR spectra were recorded on a Bruker Ascend TM 400MHz (400 MHz for ^1H , 100 MHz for ^{13}C and 376.5 MHz for ^{19}F). Chemical shifts (ppm) were referenced internally to the residual solvent proton resonance in CDCl_3 ($\delta = 7.26$ ppm) and acetone- d_6 ($\delta = 2.04$ ppm) for ^1H NMR, and CDCl_3 ($\delta = 77.2$ ppm), acetone- d_6 ($\delta = 29.8$ ppm) for $^{13}\text{C}\{^1\text{H}\}$ NMR, and with reference externally with $\text{C}_6\text{H}_5\text{CF}_3$ ($\delta -63.7$ ppm) for $^{19}\text{F}\{^1\text{H}\}$ NMR. Coupling constants (J) are reported in hertz (Hz).

High-resolution mass spectrometry (HRMS) were performed by the Center for Advanced Instrumentation and Department of Applied Chemistry at National Yang Ming Chiao Tung University. Mass spectra of the sample solutions at pH 7.5 were acquired by direct infusion (2 μL). Electrospray ionization (ESI) experiments were carried out using a Bruker Impact HD Q-TOF mass spectrometer equipped with an ESI source operating in positive ion mode. The parameters of ESI(+) included 4.5kV for ion spray voltage, 200 $^\circ\text{C}$ for capillary temperature, and 6 L/min for sheath gas flow rate. The mass spectra were collected over the mass range of m/z 50-1500 at a resolving power of 40000. The collected data were analysed using Compass DataAnalysis 4.1.

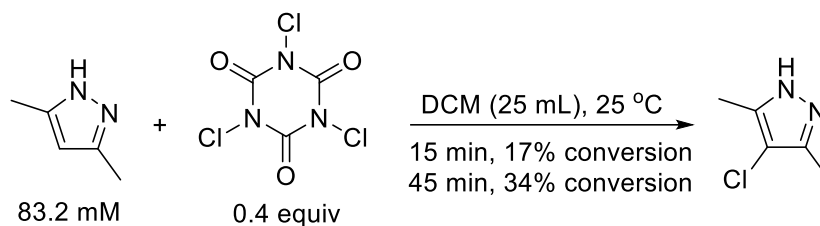
Melting points were determined on a Krüss M3000.

GC-MS analysis was conducted on a Shimadzu QP2010SE system using a Rtx-5MS column (30 m x 0.25 mm). The details of GC program are as follow: the column oven temperature and injection temperature were 100.0 and 230.0 $^\circ\text{C}$. Helium was used as the carrier gas. Flow control mode was chosen as linear velocity (36.6 cm s^{-1}) with pressure of 68.8 kPa. The total flow, column flow and purge flow were 32.5, 0.95 and 3.0 mL min^{-1} , respectively. Split mode injection with split ratio 30.0 was applied. Sample was injected as solutions in dichloromethane. After injection, the column oven temperature kept at 100.0 $^\circ\text{C}$ for 2 min and then elevated at a rate of 30 $^\circ\text{C min}^{-1}$ until 250.0 $^\circ\text{C}$. This temperature was kept for 3 min.

X-ray diffraction crystallography was performed on a Bruker D8 Venture diffractometer at National Chung Hsing University.

Compounds **1a-1c**,¹ **1d**,² **1e**,³ **1f-1g**,⁴ **1h**,⁵ **1i**,⁶ **1l**,⁷ **1m**,⁸ **1n**⁹ have been reported.

Procedure for the Chlorination of 3,5-Dimethylpyrazole with TCCA in CH₂Cl₂



In two separate runs, 3,5-dimethylpyrazole (200.0 mg, 2.08 mmol) and TCCA (193.4 mg, 0.83 mmol) were dissolved in CH₂Cl₂ (25 mL) in a 50 mL round bottom flask. The reactions were stirred at r.t for 15 min and 45 min, respectively. The aliquots were then rotary evaporated and the residue was immediately analysed with ¹H NMR spectroscopy in CDCl₃. The conversions were calculated from the ratio of methyl signals between the starting material and product.

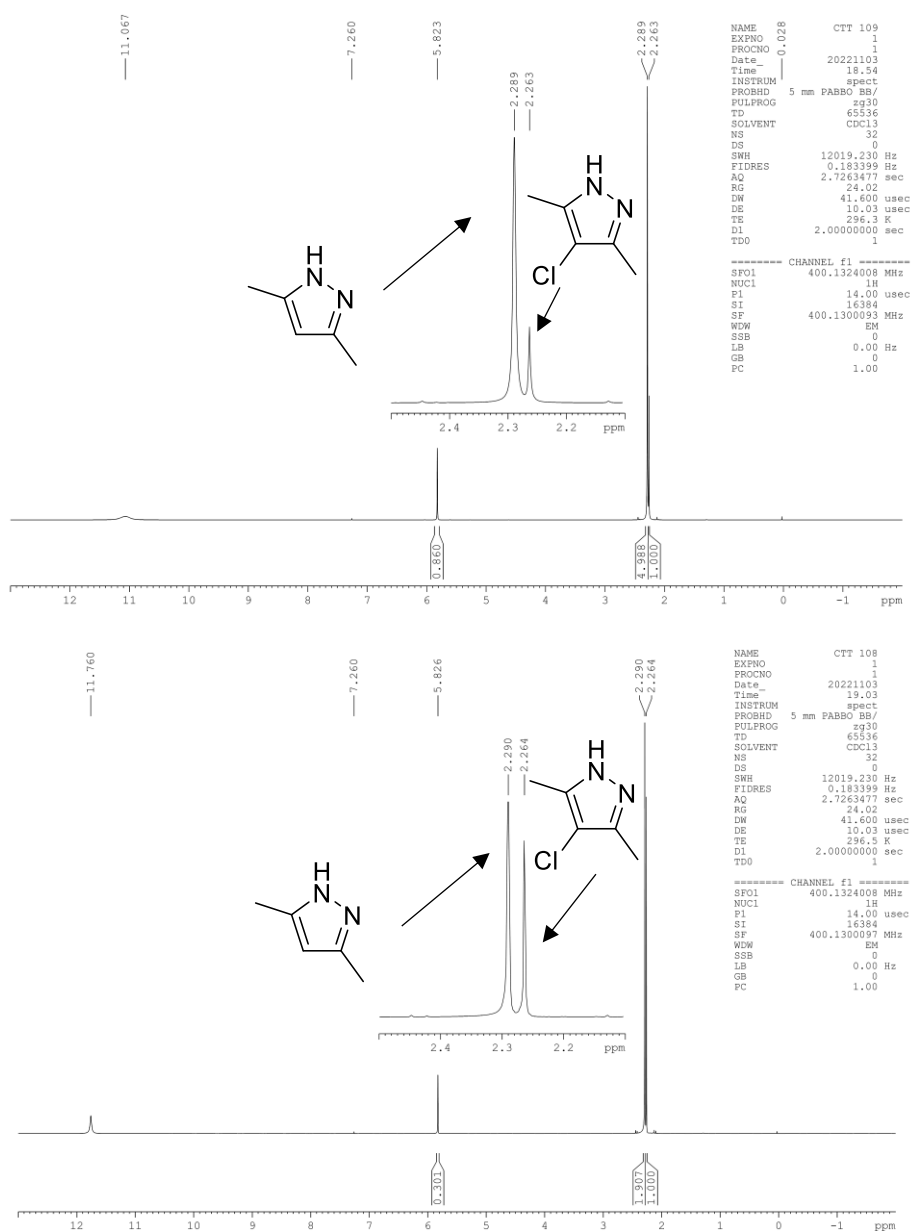


Figure S1. ¹H NMR spectrum for 15 min (up) and 45 min (down) reaction time.

General Procedure for the Mechanochemical Chlorination of Pyrazoles with TCCA

In a typical run, pyrazole (2.08 mmol), TCCA (0.83 mmol), silica gel (200.0 mg) and two ZrO₂ milling balls (5.0 mm diameter) were charged into a 10 mL ZrO₂ milling cell. This reaction vessel was then mounted on the shaker mill and oscillated at 30.0 Hz for designated reaction time. The progress of reaction was monitored by temporarily pausing the shaking and analyzed either by TLC or GC-MS until complete consumption of pyrazole. The resultant mixture was extracted with CH₂Cl₂ or MeOH and filtered through a short pad of silica on a frit (2 cm X 1 cm) to remove insoluble solid, followed by washing with 2% NaS₂O₃ solution to quench any remaining active chlorine content. Rotary evaporation of the solvent yielded the corresponding 4-chloropyrazole product.

4-Chloro-3,5-dimethylpyrazole **1a** [CAS 15953-73-8]. ¹H NMR (CDCl₃, 400 MHz) 2.24 (s, 6 H), 11.25 (br s, 1 H). ¹³C{¹H} NMR (CDCl₃, 100 MHz) 10.5, 107.9, 141.1. HRMS calcd. for [C₅H₇ClN₂+H]⁺: m/z 131.0371. Found: m/z 131.0371. m.p. Lit. 117-118 °C. m.p. found 114.7-115.9 °C.

4-Chloropyrazole **1b** [CAS 15878-00-9]. ¹H NMR (CDCl₃, 400 MHz) 7.57 (s, 2 H), 9.14 (br s, 1 H). ¹³C{¹H} NMR (CDCl₃, 100 MHz) 110.6, 132.1. HRMS calcd. for [C₃H₃ClN₂+H]⁺: m/z 103.0058. Found: m/z 103.0058. M.p. Lit. 75-79 °C. M.p. found 76.7-78.1 °C.

4-Chloro-3,5-diphenylpyrazole **1c** [CAS 71549-28-5]. ¹H NMR (CDCl₃, 400 MHz) 7.40-7.49 (m, 6 H), 7.81 (d, *J* = 7.2 Hz, 4 H), 10.78 (s, 1 H). HRMS calcd. for [C₁₅H₁₁ClN₂+H]⁺: m/z 255.0684. Found: m/z 255.0684. mp. Lit. 191-193 °C. M.p. found 204.8-205.7 °C.

4-Chloro-3,5-dimethyl-1H-pyrazole-1-carboxamide **1d** [CAS 1174305-03-3]. ¹H NMR (CDCl₃, 400 MHz) 2.22 (s, 3 H), 2.55 (s, 3 H), 5.27 (br s, 1 H), 7.08 (s, 1 H). ¹³C{¹H} NMR (CDCl₃, 100 MHz) 11.7, 12.1, 113.2, 139.4, 148.2, 151.8. HRMS calcd. for [C₆H₈ClN₃O+H]⁺: m/z 174.0429. Found: m/z 174.0424. M.p. lit. 141-142 °C. M.p. found 134.4-135.8 °C.

4-Chloro-1-(tetrahydro-2H-pyran-2-yl)-1H-pyrazole **1e** [CAS 1881288-80-7]. ¹H NMR (CDCl₃, 400 MHz) 1.51-1.57 (m, 3 H), 1.94 (broad s, 3 H), 3.58 (t, *J* = 10.5 Hz, 1 H), 3.93 (d, *J* = 11.5 Hz, 1 H), 5.22-5.23 (m, 1 H), 7.38 (s, 1 H), 7.53 (s, 1 H). ¹³C{¹H} NMR (CDCl₃, 100 MHz) 22.1, 24.8, 30.2, 67.6, 87.9, 110.6, 125.7, 137.9. HRMS calcd. for [C₈H₁₁ClN₂O+H]⁺: m/z 187.0633. Found: m/z 187.0627. (Green Chem. 2016, 18, 6209-6214). Pale yellow oil.

4-Chloro-3,5-dimethyl-1-phenyl-1H-pyrazole **1f** [CAS 861382-34-5]. Purified on silica gel column chromatography with hexane/DCM = 3:1 as the eluent. ¹H NMR (CDCl₃, 400 MHz) 2.27 (s, 3 H), 2.29 (s, 3 H), 7.32-7.45 (m, 5 H). HRMS calcd. for [C₁₁H₁₁ClN₂+H]⁺: m/z 207.0684. Found: m/z 207.0686. pale yellow oil.

Di-chlorinated side product of 3,5-dimethyl-1-phenylpyrazole **1f-2**. Purified on silica gel column chromatography with hexane/DCM = 3:1 as the eluent. ^1H NMR (CDCl_3 , 400 MHz) 2.32 (s, 3 H), 4.56 (s, 2 H), 7.42-7.57 (m, 5 H). HRMS calcd. for $[\text{C}_{11}\text{H}_{10}\text{Cl}_2\text{N}_2+\text{H}]^+$: m/z 241.0294. Found: m/z 241.0293.

Tri-chlorinated side product of 3,5-dimethyl-1-phenylpyrazole **1f-3**. Purified on silica gel column chromatography with hexane/DCM = 3:1 as the eluent. ^1H NMR (CDCl_3 , 400 MHz) 2.32 (s, 3 H), 6.70 (s, 1 H), 7.45-7.55 (m, 5 H). HRMS calcd. for $[\text{C}_{11}\text{H}_9\text{Cl}_3\text{N}_2+\text{H}]^+$: m/z 274.9904. Found: m/z 274.9902.

4-Chloro-1-phenylpyrazole **1g** [CAS 6831-92-1]. ^1H NMR (CDCl_3 , 400 MHz) 7.31 (t, $J = 7.6$ Hz, 1 H), 7.45 (t, $J = 7.8$ Hz, 2 H), 7.63 (d, $J = 7.6$ Hz, 2 H), 7.65 (s, 1 H), 7.90 (s, 1 H). $^{13}\text{C}\{^1\text{H}\}$ NMR (CDCl_3 , 100 MHz) 112.5, 119.1, 125.0, 127.1, 129.7, 139.6, 139.8. HRMS calcd. for $[\text{C}_9\text{H}_7\text{ClN}_2+\text{H}]^+$: m/z 179.0371. Found: m/z 179.0371. M.p. Lit. 72.5-74.3 °C. M.p. found 73.5-74.2 °C.

4-Chloro-1-(4-methoxyphenyl)-1H-pyrazole **1h** [CAS 1402566-05-5]. ^1H NMR (CDCl_3 , 400 MHz) 3.83 (s, 3 H), 6.96 (d, $J = 9.2$ Hz, 2 H), 7.52 (d, $J = 8.8$ Hz, 2 H), 7.60 (s, 1 H), 7.80 (s, 1 H). $^{13}\text{C}\{^1\text{H}\}$ NMR (CDCl_3 , 100 MHz) 55.7, 112.0, 114.7, 120.9, 125.1, 133.6, 139.1, 158.7. HRMS calcd. for $[\text{C}_{10}\text{H}_9\text{ClN}_2\text{O}+\text{H}]^+$: m/z 209.0476. Found: m/z 209.0476. M.p. found 94.5-95.2 °C.

1-(4-Bromophenyl)-4-chloro-1H-pyrazole **1i** [CAS 1248589-16-3]. ^1H NMR (CDCl_3 , 400 MHz) 7.52 (d, $J = 9.1$ Hz, 2 H), 7.58 (d, $J = 8.8$ Hz, 2 H), 7.64 (s, 1 H), 7.88 (s, 1 H). ^1H NMR (acetone- d_6 , 400 MHz) 7.69 (d, $J = 8.7$ Hz, 2 H), 7.75 (s, 1 H), 7.80 (d, $J = 8.7$ Hz, 2 H), 8.52 (s, 1 H). $^{13}\text{C}\{^1\text{H}\}$ NMR (CDCl_3 , 100 MHz) 112.9, 120.4, 124.8, 132.6, 138.7, 139.9 [one peak is overlapped]. $^{13}\text{C}\{^1\text{H}\}$ NMR (acetone- d_6 , 100 MHz) 113.1, 120.3, 121.2, 126.5, 133.4, 139.9, 140.3. HRMS calcd. for $[\text{C}_9\text{H}_6\text{BrClN}_2+\text{H}]^+$: m/z 256.9476. Found: m/z 256.9473. M.p. found 74.2-75.0 °C.

4-(4-chloro-1H-pyrazol-1-yl)benzaldehyde **1j** [CAS 1179758-89-4]. ^1H NMR (CDCl_3 , 400 MHz) 7.70 (s, 1 H), 7.83 (d, $J = 8.6$ Hz, 2 H), 7.99 (d, $J = 8.7$ Hz, 2 H), 8.01 (s, 1 H), 10.02 (s, 1 H). $^{13}\text{C}\{^1\text{H}\}$ NMR (CDCl_3 , 100 MHz) 113.9, 118.7, 125.0, 131.5, 134.6, 140.9, 143.8, 190.9. HRMS calcd. for $[\text{C}_{10}\text{H}_7\text{ClN}_2\text{O}+\text{H}]^+$: m/z 207.0320. Found: m/z 207.0316. M.p. found 122.8-123.3 °C.

4-(4-chloro-1H-pyrazol-1-yl)benzotrile **1k** [CAS 1179745-44-8]. ^1H NMR (CDCl_3 , 400 MHz) 7.69 (s, 1 H), 7.75 (d, $J = 9.0$ Hz, 2 H), 7.79 (d, $J = 9.0$ Hz, 2 H), 7.98 (s, 1 H). $^{13}\text{C}\{^1\text{H}\}$ NMR (CDCl_3 , 100 MHz) 110.4, 114.2, 118.3, 118.9, 124.9, 133.9, 141.0, 142.5. HRMS calcd. for $[\text{C}_{10}\text{H}_6\text{ClN}_3+\text{H}]^+$: m/z 204.0323. Found: m/z 204.0319. M.p. found 134.5-134.9 °C.

4-Chloro-1-(4-nitrophenyl)-1H-pyrazole **1l** [CAS 65041-53-4]. ^1H NMR (CDCl_3 , 400 MHz) 7.72 (s, 1 H), 7.83 (d, $J = 9.0$ Hz, 2 H), 8.02 (s, 1 H), 8.35 (d, $J = 9.0$ Hz, 2 H). $^{13}\text{C}\{^1\text{H}\}$ NMR (CDCl_3 , 100

MHz) 114.5, 118.6, 125.1, 125.6, 141.4, 143.9, 145.9. HRMS calcd. for $[\text{C}_9\text{H}_6\text{ClN}_3\text{O}_2+\text{H}]^+$: m/z 224.0221. Found: m/z 224.0220. M.p. Lit. 145.5-146.5 °C. M.p. found 144.7-146.3 °C.

2-(4-chloro-1H-pyrazol-1-yl)pyridine **1m** [CAS 77556-33-3]. ^1H NMR (CDCl_3 , 400 MHz) 7.18-7.21 (m, 1 H), 7.64 (s, 1 H), 7.81 (td, $J = 7.8$ Hz, 2.1 Hz, 1 H), 7.93 (d, $J = 8.0$ Hz, 1 H), 8.39 (d, $J = 3.2$ Hz, 1 H), 8.54 (s, 1 H). $^{13}\text{C}\{^1\text{H}\}$ NMR (CDCl_3 , 100 MHz) 112.0, 113.0, 121.9, 125.1, 138.9, 140.5, 148.2, 151.1. HRMS calcd. for $[\text{C}_8\text{H}_6\text{ClN}_3+\text{H}]^+$: m/z 180.0323. Found: m/z 180.0321. mp. Lit. 64-65 °C. M.p. found 69.6-70.3 °C.

4-chloro-celecoxib **1n** [2248154-36-9]. ^1H NMR (CDCl_3 , 400 MHz) 2.41 (s, 3 H), 4.89 (br s, 2 H), 7.16 (d, $J = 8.0$ Hz, 2 H), 7.24 (d, $J = 8.0$ Hz, 2 H), 7.41 (d, $J = 8.6$ Hz, 2 H), 7.88 (d, $J = 8.5$ Hz, 2 H). $^{13}\text{C}\{^1\text{H}\}$ NMR (CDCl_3 , 100 MHz) 21.6, 109.9, 120.5 (quartet, $^1J_{\text{CF}} = 268.5$ Hz), 123.4, 125.2, 127.7, 129.8, 130.1, 140.7, 140.8 (quartet, $^2J_{\text{CF}} = 37.5$ Hz), 141.6, 142.0, 142.4. $^{19}\text{F}\{^1\text{H}\}$ NMR (CDCl_3 , 376 MHz) -63.7 (s). HRMS calcd. for $[\text{C}_{17}\text{H}_{13}\text{ClF}_3\text{N}_3\text{O}_2\text{S}+\text{H}]^+$: m/z 416.0442. Found: m/z 416.0440. M.p. Lit. 169-171 °C. M.p. found 174.5-175.8 °C. Single crystal suitable for X-ray diffraction was obtained by slow evaporation of $\text{CH}_2\text{Cl}_2/\text{MeOH}$ solution.

Gram-Scale Mechanochemical Chlorination of Pyrazoles with TCCA

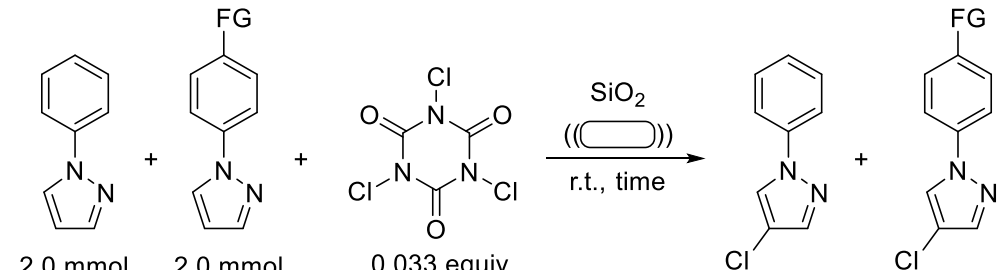
With 3,5-dimethylpyrazole. 3,5-dimethylpyrazole **1** (1.00 g, 10.4 mmol), TCCA (967.0 mg, 4.16 mmol), silica gel (200.0 mg) and one ZrO_2 milling ball (10.0 mm diameter) were charged into a ZrO_2 shaker cell. This reaction vessel was then mounted on the shaker mill and oscillated at 30.0 Hz for 60 min. The resultant mixture was extracted with CH_2Cl_2 and filtered through a short pad of silica on a frit to remove insoluble solid, followed by washing with 5% NaS_2O_3 solution to quench any remaining active chlorine content. Rotary evaporation of the solvent yielded 4-chloro-3,5-dimethylpyrazole **1a** (1.11 g, 8.50 mmol, 82%).

With 1-phenylpyrazole. 1-phenylpyrazole **1** (917 μL , 6.36 mmol), TCCA (591.0 mg, 2.54 mmol), silica gel (200.0 mg) and one ZrO_2 milling ball (10.0 mm diameter) were charged into a ZrO_2 shaker cell. This reaction vessel was then mounted on the shaker mill and oscillated at 30.0 Hz for 45 min. The resultant mixture was extracted with CH_2Cl_2 and filtered through a short pad of silica on a frit to remove insoluble solid, followed by washing with 5% NaS_2O_3 solution to quench any remaining active chlorine content. Rotary evaporation of the solvent yielded 4-chloro-1-phenylpyrazole **1g** (1.06 g, 5.93 mmol, 93%).

Competition Experiments between 1-Phenylpyrazole and 1-Arylpyrazoles

1-phenylpyrazole (2.0 mmol), 1-arylpyrazole (2.0 mmol), TCCA (0.066 mmol), silica gel (200.0 mg) and two ZrO₂ milling balls (5.0 mm diameter) were charged into a 10 mL ZrO₂ milling cell. This reaction vessel was then mounted on the shaker mill and oscillated at 30.0 Hz for designated reaction time. The resultant mixture was extracted with CH₂Cl₂ and filtered through a short pad of silica on a frit to remove insoluble solid, followed by washing with 5% NaS₂O₃ solution to quench any remaining active chlorine content. The ratio between 4-chloro-1-phenylpyrazole **1a** and 4-chloro-1-arylpyrazole was determined by analyzing the mixture with GC-MS.

Table S1. Competition Experiments



entry	FG	time / min	yield / %			log(k_{FG}/k_H) ^a
			FG	H	total yield	
1	OMe	30	62	20	82	0.49
2	OMe	45	64	23	89	0.44
3	Br	45	9	70	79	-0.89
4	Br	45	8	73	81	-0.96
5	CHO	30	4	86	91	-1.33
6	CHO	45	3	95	98	-1.50
7	CN	30	2	91	93	-1.66
8	CN	30	2	84	86	-1.62
9	NO ₂	30	1	91	91	-3.26
10	NO ₂	45	1	85	85	-3.23

Table S2. Hammett Plot of Mechanochemical Chlorination of 1-Arylpyrazoles with TCCA using the Hammett Constant σ_p .

entry	FG	σ_p^a	$\log(k_{FG}/k_H)$
1	OMe	-0.27	0.49
2	OMe	-0.27	0.44
3	Br	0.23	-0.89
4	Br	0.23	-0.96
5	CHO	0.42	-1.33
6	CHO	0.42	-1.50
7	CN	0.66	-1.66
8	CN	0.66	-1.62
9	NO ₂	0.78	-3.26
10	NO ₂	0.78	-3.23

^a Reference 10.

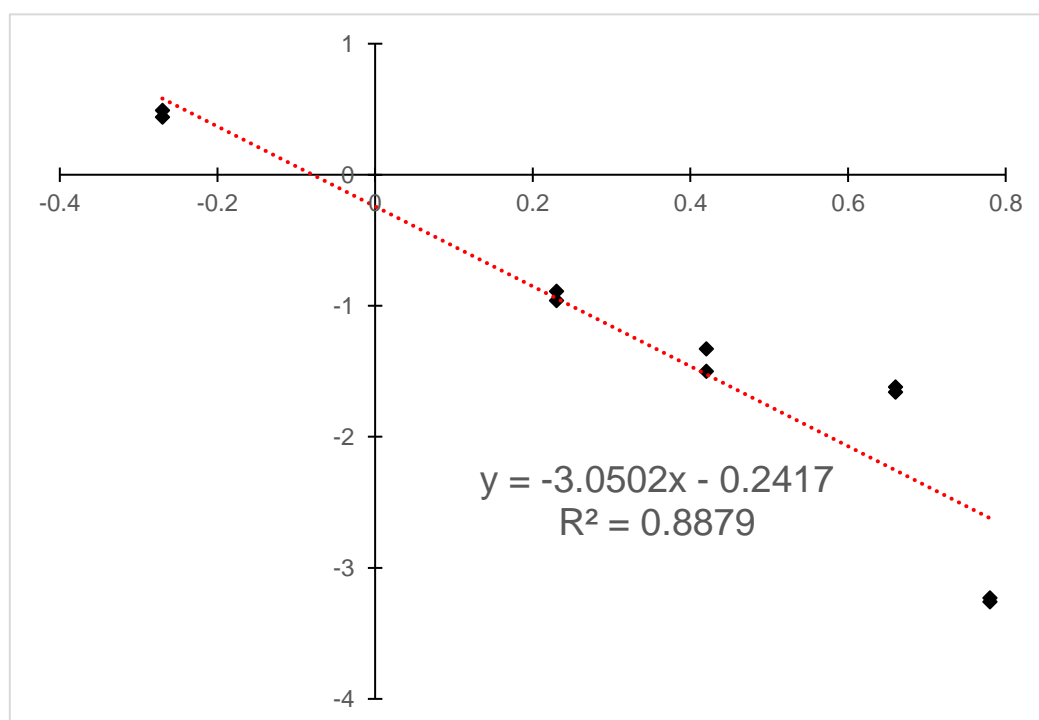
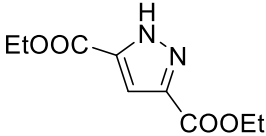
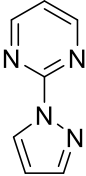
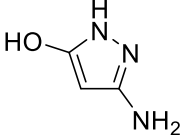
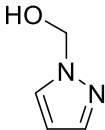
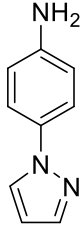
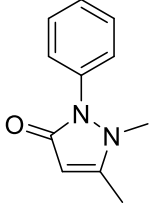
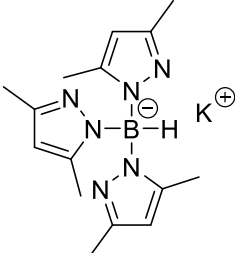
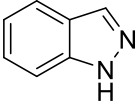

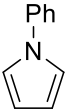
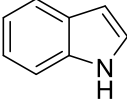
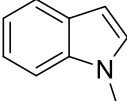
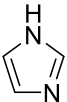
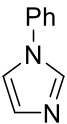
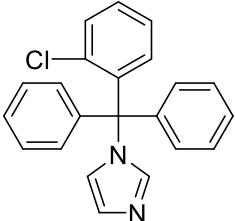


Figure S2. Hammett Plot of Mechanochemical Chlorination of 1-Arylpyrazoles with TCCA using the Hammett Constant σ_p .

Substrates with Unsatisfactory Results^{a,b}

entry	substrate	remark
1 ^c		expected chlorination product observed but less than 50% conversion in 3 h
2 ^c		expected chlorination product observed but less than 50% conversion in 3 h
3		complicated mixture
4		complicated mixture
5		complicated mixture
6		complicated mixture
7		complicated mixture

8		two mono-chlorination products
9 ^c		mixture of mono, di, tri, tetra-chlorination products and poly-pyrrolic substances
10 ^c		two mono-chlorination products three di-chlorination products one tri-chlorination product one tetra-chlorination product
11 ^c		one mono-chlorination product one di-chlorination product
12 ^c		one mono-chlorination product one di-chlorination product some chlorinated indolione products
13		unknowns
14 ^c		two mono-chlorination products one di-chlorination product one tri-chlorination product
15 ^c		three products

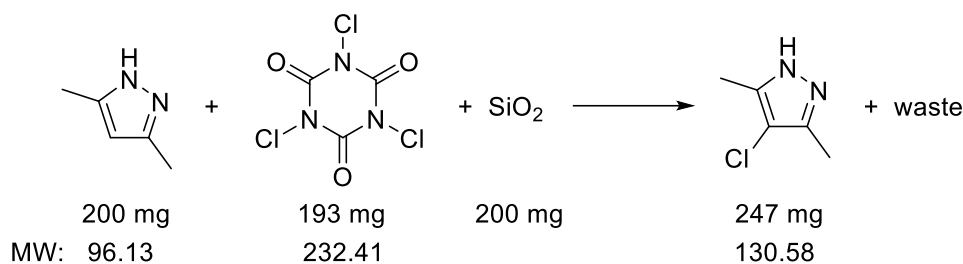
^a Standard conditions as indicated in the manuscript

^b Reaction was analyzed after 30 min with TLC, GC-MS or ¹H NMR; Regioselectivity not determined

^c Starting material was remained

Calculation of Green Chemistry Metrics

For mechanochemical chlorination of 3,5-dimethylpyrazole with TCCA



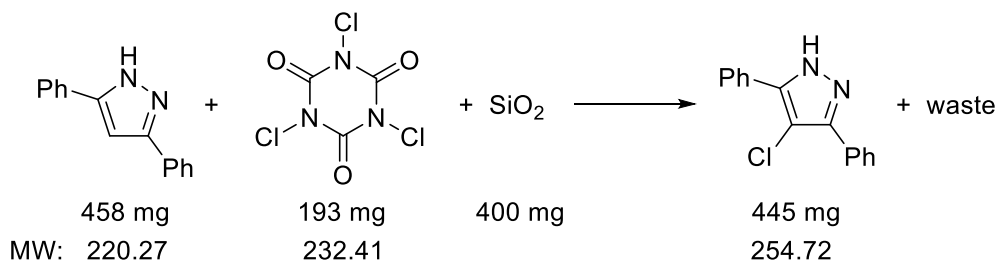
Atom Economy (AE) = $130.58 / (96.13 + 232.41/3) \times 100\% = 75.2\%$

Reaction Mass Efficiency (RME) = $247 / (200 + 193) \times 100\% = 62.8\%$

Process Mass Intensity (PMI) = $(200 + 193 + 200) / 247 = 2.40$

E-factor = $PMI - 1 = 2.40 - 1 = 1.40$

For mechanochemical chlorination of 3,5-diphenylpyrazole with TCCA



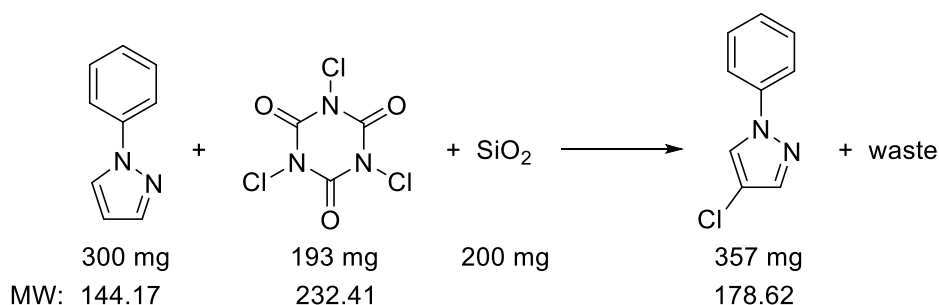
Atom Economy (AE) = $254.72 / (220.27 + 232.41/3) \times 100\% = 85.6\%$

Reaction Mass Efficiency (RME) = $445 / (458 + 193) \times 100\% = 68.4\%$

Process Mass Intensity (PMI) = $(458 + 193 + 400) / 445 = 2.36$

E-factor = $PMI - 1 = 2.36 - 1 = 1.36$

For mechanochemical chlorination of 1-phenylpyrazole with TCCA



Atom Economy (AE) = $178.62 / (144.17 + 232.41/3) \times 100\% = 80.6\%$

Reaction Mass Efficiency (RME) = $357 / (300 + 193) \times 100\% = 72.4\%$

Process Mass Intensity (PMI) = $(300 + 193 + 200) / 357 = 1.94$

E-factor = $PMI - 1 = 1.94 - 1 = 0.94$

The green chemistry metrics for the synthesis of 4-chloro-1-phenylpyrazole **1g** using other protocols were calculated according to the published literatures. The results were tabulated below together with this report.

Table S3. Comparison of Green Chemistry Metrics for the Chlorination of 1-phenylpyrazole of various Chlorinating System

Green Chemistry Metrics	this work TCCA	1-Chloro-1,2-benziodoxol-3-one ¹¹	CBMG ¹²	CFBSA ¹³	TCCA ¹⁴	NaCl/Oxone ¹⁵
AE	80.6%	41.9%	50.5%	50.5%	80.6%	35.0%
RME	72.4%	36.7%	45.4%	37.5%	44.1%	10.7%
<i>E</i> -factor	0.94	55.7	47.8	65.3	37.1	13.3
PMI	1.94	56.7	48.8	66.3	38.1	14.3
Purification:	filtration / recrystallization	column chromatography				

X-Ray Diffraction Data

Table S4. Crystal data and structural refinement for **1n**.

Compound	4-chloro-celecoxib 1n
Empirical formula	C ₁₇ H ₁₃ ClF ₃ N ₃ O ₂ S
Formula weight	415.81
Temperature	150(2) K
Wavelength	0.71073 Å
Crystal system	triclinic
Space group	P-1
Unit cell dimensions	a = 8.0656(3) Å α = 85.0509(15)° b = 8.5494(3) Å β = 86.9189(14)° c = 13.9078(6) Å γ = 79.3578(14)°
Volume	938.32(6) Å ³
Z	2
Density (calculated)	1.472 mg/cm ³
Absorption coefficient	0.361 mm ⁻¹
F(000)	424
Crystal size	0.380 x 0.260 x 0.220 mm ³
Theta range for data collection	2.913 to 27.952°
Limiting indices	-10 ≤ h ≤ 10, -11 ≤ k ≤ 11, -18 ≤ l ≤ 18
Reflections collected	18454
Independent reflections	4468 [R(int) = 0.0365]
Completeness to theta = 25.242°	99.2%
Absorption correction	Semi-empirical from equivalents
Max. and min. transmission	0.7456 and 0.6957
Refinement method	Full-matrix least-squares on F ²
Data / restraints / parameters	4468 / 0 / 252
Goodness-of fit on F ²	1.070
Final R indices [I > 2σ(I)]	R1 = 0.0394, wR2 = 0.1192
R indices (all data)	R1 = 0.0445, wR2 = 0.1242
Extinction coefficient	n/a
Largest diff. peak and hole	0.367 and -0.528 e.Å ⁻³

Table S5. Atomic coordinates ($\times 10^4$) and equivalent isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for 4-chlorocelecoxib. $U(\text{eq})$ is defined as one third of the trace of the orthogonalized U^{ij} tensor.

	x	y	z	U(eq)
Cl	6202(1)	1128(1)	710(1)	32(1)
S	2686(1)	8964(1)	5370(1)	16(1)
F(1)	1448(2)	210(2)	2608(1)	37(1)
F(2)	3769(2)	-1110(2)	2059(1)	50(1)
F(3)	2113(2)	444(2)	1102(1)	42(1)
O(1)	1144(2)	8800(2)	5900(1)	22(1)
O(2)	4192(2)	8905(2)	5889(1)	22(1)
N(1)	3013(2)	2665(2)	2786(1)	18(1)
N(2)	4066(2)	3740(2)	2700(1)	18(1)
N(3)	2350(2)	10661(2)	4744(1)	19(1)
C(1)	5318(2)	3414(2)	2002(1)	19(1)
C(2)	5041(2)	2051(2)	1631(1)	21(1)
C(3)	3613(2)	1639(2)	2134(1)	19(1)
C(4)	3768(2)	5007(2)	3332(1)	18(1)
C(5)	2113(2)	5542(2)	3662(2)	26(1)
C(6)	1787(2)	6751(3)	4287(2)	26(1)
C(7)	3107(2)	7417(2)	4575(1)	17(1)
C(8)	4755(2)	6882(2)	4243(1)	20(1)
C(9)	5087(2)	5658(2)	3626(1)	21(1)
C(10)	6622(2)	4370(2)	1691(1)	18(1)
C(11)	6200(3)	5820(2)	1150(2)	27(1)
C(12)	7439(3)	6702(2)	843(2)	32(1)
C(13)	9114(3)	6154(3)	1062(2)	28(1)
C(14)	9527(3)	4688(3)	1585(2)	29(1)
C(15)	8298(3)	3796(2)	1903(2)	26(1)
C(16)	10461(3)	7120(3)	742(2)	44(1)
C(17)	2731(3)	293(2)	1985(1)	23(1)

Table S6. Bond lengths [\AA] and angles [$^\circ$] for JXC041.

Bond	Bond Length / \AA	Bond	Bond Length / \AA
Cl-C(2)	1.7069(18)	C(8)-C(9)	1.389(2)
S-O(1)	1.4346(13)	C(8)-H(8A)	0.9500
S-O(2)	1.4372(13)	C(9)-H(9A)	0.9500
S-N(3)	1.6117(16)	C(10)-C(11)	1.389(3)
S-C(7)	1.7704(18)	C(10)-C(15)	1.388(3)
F(1)-C(17)	1.322(2)	C(11)-C(12)	1.389(3)
F(2)-C(17)	1.330(2)	C(11)-H(11A)	0.9500
F(3)-C(17)	1.337(2)	C(12)-C(13)	1.387(3)
N(1)-C(3)	1.327(2)	C(12)-H(12A)	0.9500
N(1)-N(2)	1.356(2)	C(13)-C(14)	1.388(3)
N(2)-C(1)	1.370(2)	C(13)-C(16)	1.509(3)
N(2)-C(4)	1.431(2)	C(14)-C(15)	1.390(3)
N(3)-H(3A)	0.85(3)	C(14)-H(14A)	0.9500
N(3)-H(3B)	0.85(3)	C(15)-H(15A)	0.9500
C(1)-C(2)	1.374(2)	C(16)-H(16A)	0.9800
C(1)-C(10)	1.473(2)	C(16)-H(16B)	0.9800
C(2)-C(3)	1.400(2)	C(16)-H(16C)	0.9800
C(3)-C(17)	1.494(2)	C(13)-C(16)	1.509(3)
C(4)-C(9)	1.384(2)	C(14)-C(15)	1.390(3)
C(4)-C(5)	1.395(3)	C(14)-H(14A)	0.9500
C(5)-C(6)	1.387(3)	C(15)-H(15A)	0.9500
C(5)-H(5A)	0.9500	C(16)-H(16A)	0.9800
C(6)-C(7)	1.388(3)	C(16)-H(16B)	0.9800
C(6)-H(6A)	0.9500	C(16)-H(16C)	0.9800
C(7)-C(8)	1.391(2)		

Bond	Bond Angle / $^\circ$	Bond	Bond Angle / $^\circ$
O(1)-S-O(2)	119.12(8)	C(9)-C(8)-H(8A)	120.1
O(1)-S-N(3)	107.22(8)	C(7)-C(8)-H(8A)	120.1
O(2)-S-N(3)	106.96(8)	C(4)-C(9)-C(8)	119.53(16)
O(1)-S-C(7)	106.97(8)	C(4)-C(9)-H(9A)	120.2
O(2)-S-C(7)	107.29(8)	C(8)-C(9)-H(9A)	120.2
N(3)-S-C(7)	109.01(8)	C(11)-C(10)-C(15)	119.39(17)
C(3)-N(1)-N(2)	104.47(14)	C(11)-C(10)-C(1)	120.40(17)
N(1)-N(2)-C(1)	112.83(14)	C(15)-C(10)-C(1)	120.15(16)
N(1)-N(2)-C(4)	118.15(14)	C(12)-C(11)-C(10)	120.21(18)

C(1)-N(2)-C(4)	129.01(15)	C(12)-C(11)-H(11A)	119.9
S-N(3)-H(3A)	112.2(17)	C(10)-C(11)-H(11A)	119.9
S-N(3)-H(3B)	110.6(18)	C(11)-C(12)-C(13)	120.95(19)
H(3A)-N(3)-H(3B)	117(2)	C(11)-C(12)-H(12A)	119.5
N(2)-C(1)-C(2)	105.00(15)	C(13)-C(12)-H(12A)	119.5
N(2)-C(1)-C(10)	127.07(16)	C(12)-C(13)-C(14)	118.33(18)
C(2)-C(1)-C(10)	127.88(16)	C(12)-C(13)-C(16)	121.1(2)
C(1)-C(2)-C(3)	106.27(16)	C(14)-C(13)-C(16)	120.5(2)
C(1)-C(2)-Cl	125.33(15)	C(13)-C(14)-C(15)	121.34(19)
C(3)-C(2)-Cl	128.37(14)	C(13)-C(14)-H(14A)	119.3
N(1)-C(3)-C(2)	111.43(16)	C(15)-C(14)-H(14A)	119.3
N(1)-C(3)-C(17)	120.28(16)	C(10)-C(15)-C(14)	119.76(18)
C(2)-C(3)-C(17)	128.23(17)	C(10)-C(15)-H(15A)	120.1
C(9)-C(4)-C(5)	120.85(16)	C(14)-C(15)-H(15A)	120.1
C(9)-C(4)-N(2)	121.00(15)	C(13)-C(16)-H(16A)	109.5
C(5)-C(4)-N(2)	118.14(15)	C(13)-C(16)-H(16B)	109.5
C(6)-C(5)-C(4)	119.50(17)	H(16A)-C(16)-H(16B)	109.5
C(6)-C(5)-H(5A)	120.2	C(13)-C(16)-H(16C)	109.5
C(4)-C(5)-H(5A)	120.2	H(16A)-C(16)-H(16C)	109.5
C(5)-C(6)-C(7)	119.74(17)	H(16B)-C(16)-H(16C)	109.5
C(5)-C(6)-H(6A)	120.1	F(1)-C(17)-F(2)	107.60(16)
C(7)-C(6)-H(6A)	120.1	F(1)-C(17)-F(3)	106.91(17)
C(6)-C(7)-C(8)	120.60(16)	F(2)-C(17)-F(3)	106.27(17)
C(6)-C(7)-S	119.62(13)	F(1)-C(17)-C(3)	112.68(15)
C(8)-C(7)-S	119.78(13)	F(2)-C(17)-C(3)	111.89(17)
C(9)-C(8)-C(7)	119.78(16)	F(3)-C(17)-C(3)	111.13(15)

Table S7. Anisotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for JXC041. The anisotropic displacement factor exponent takes the form: $-2p^2 [h^2 a^{*2}U^{11} + \dots + 2 h k a^* b^* U^{12}]$

	U^{11}	U^{22}	U^{33}	U^{23}	U^{13}	U^{12}
Cl	37(1)	28(1)	32(1)	-14(1)	14(1)	-11(1)
S	12(1)	20(1)	16(1)	-5(1)	0(1)	-3(1)
F(1)	41(1)	43(1)	33(1)	-14(1)	15(1)	-26(1)
F(2)	41(1)	16(1)	91(1)	-7(1)	1(1)	-6(1)
F(3)	65(1)	48(1)	25(1)	-5(1)	-5(1)	-35(1)
O(1)	15(1)	30(1)	20(1)	-4(1)	3(1)	-5(1)
O(2)	15(1)	30(1)	21(1)	-6(1)	-3(1)	-4(1)
N(1)	19(1)	16(1)	20(1)	-2(1)	0(1)	-5(1)
N(2)	18(1)	16(1)	20(1)	-3(1)	1(1)	-5(1)
N(3)	14(1)	21(1)	23(1)	-3(1)	-1(1)	-3(1)
C(1)	19(1)	18(1)	18(1)	-1(1)	1(1)	-2(1)
C(2)	23(1)	19(1)	20(1)	-4(1)	3(1)	-4(1)
C(3)	23(1)	17(1)	19(1)	-2(1)	-1(1)	-5(1)
C(4)	18(1)	16(1)	19(1)	-4(1)	-1(1)	-3(1)
C(5)	16(1)	31(1)	37(1)	-16(1)	0(1)	-7(1)
C(6)	13(1)	32(1)	36(1)	-16(1)	2(1)	-3(1)
C(7)	15(1)	19(1)	18(1)	-4(1)	0(1)	-2(1)
C(8)	15(1)	24(1)	23(1)	-7(1)	1(1)	-5(1)
C(9)	13(1)	25(1)	25(1)	-6(1)	1(1)	-3(1)
C(10)	20(1)	18(1)	19(1)	-4(1)	2(1)	-4(1)
C(11)	22(1)	19(1)	39(1)	3(1)	-1(1)	-2(1)
C(12)	32(1)	20(1)	43(1)	5(1)	2(1)	-9(1)
C(13)	28(1)	29(1)	30(1)	-7(1)	4(1)	-14(1)
C(14)	18(1)	37(1)	34(1)	-2(1)	-4(1)	-7(1)
C(15)	24(1)	26(1)	28(1)	3(1)	-2(1)	-3(1)
C(16)	41(1)	50(2)	50(2)	-5(1)	4(1)	-30(1)
C(17)	27(1)	21(1)	23(1)	-4(1)	4(1)	-8(1)

Table S8. Hydrogen coordinates ($\times 10^4$) and isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for JXC041.

	x	y	z	U(eq)
H(3A)	1440(30)	10800(30)	4439(18)	28(6)
H(3B)	3240(40)	10820(30)	4426(19)	30(6)
H(5A)	1217	5082	3460	32
H(6A)	665	7122	4517	32
H(8A)	5650	7352	4437	24
H(9A)	6213	5272	3407	25
H(11A)	5059	6209	990	32
H(12A)	7135	7696	477	38
H(14A)	10672	4287	1730	35
H(15A)	8605	2798	2263	31
H(16A)	11558	6545	961	66
H(16B)	10504	7288	36	66
H(16C)	10196	8155	1021	66

31 Y

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Z -61

JXC041

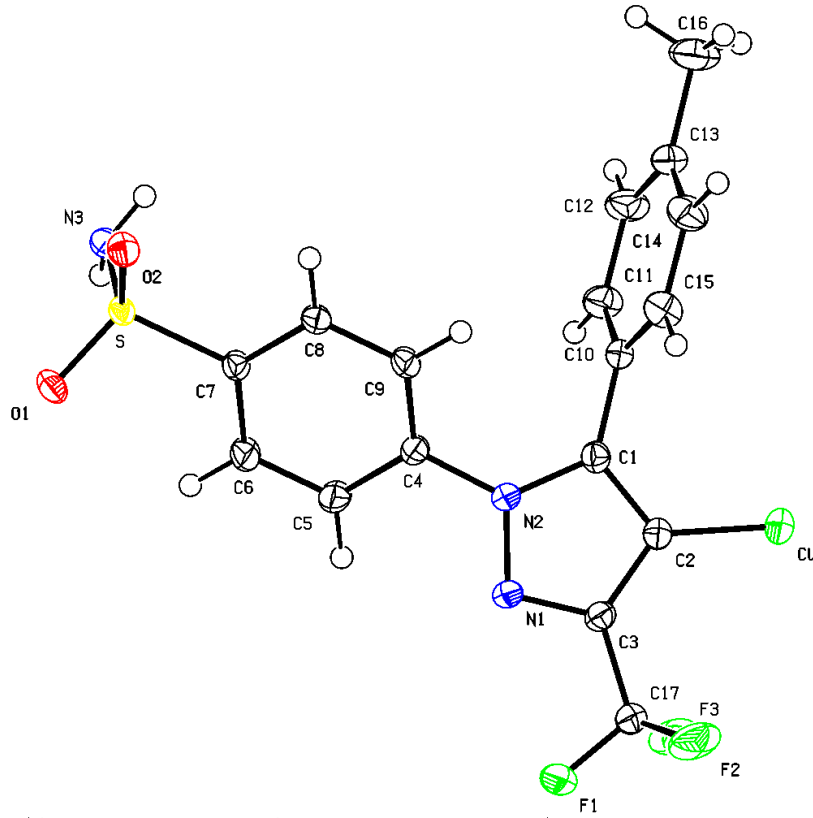
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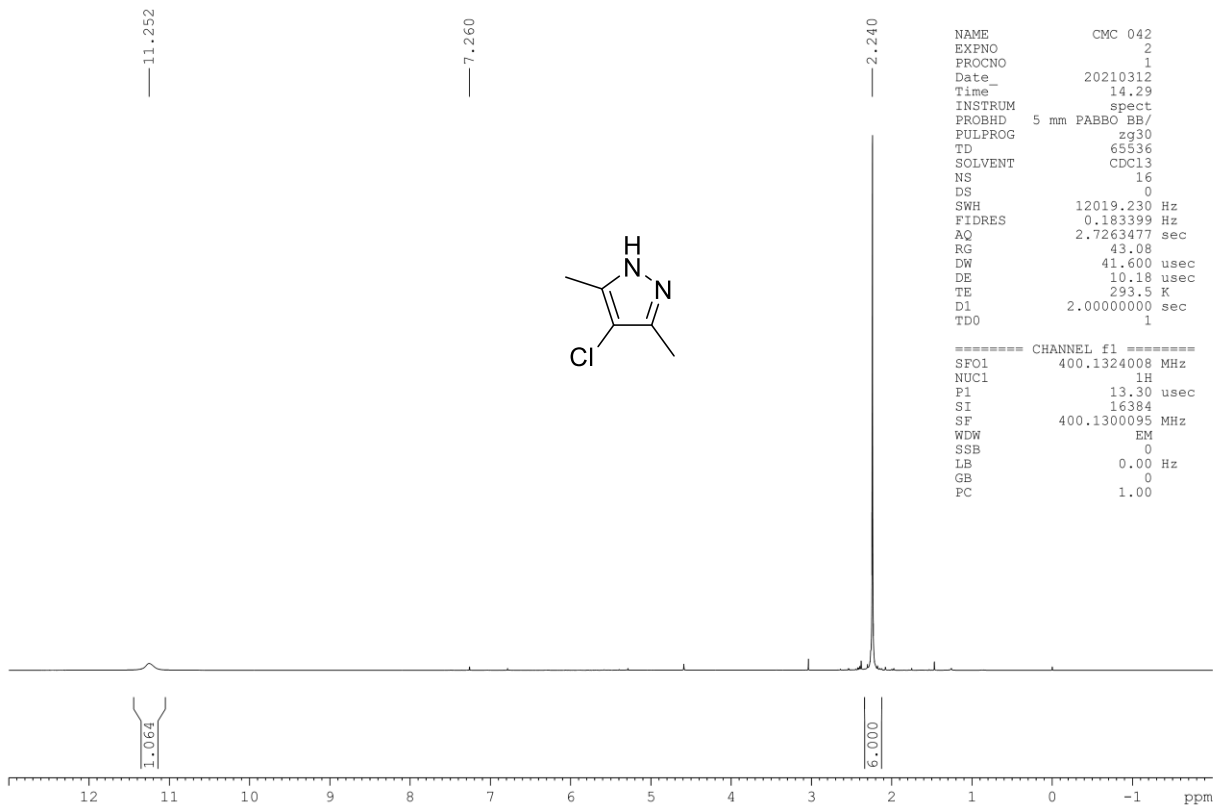


NMR Spectra

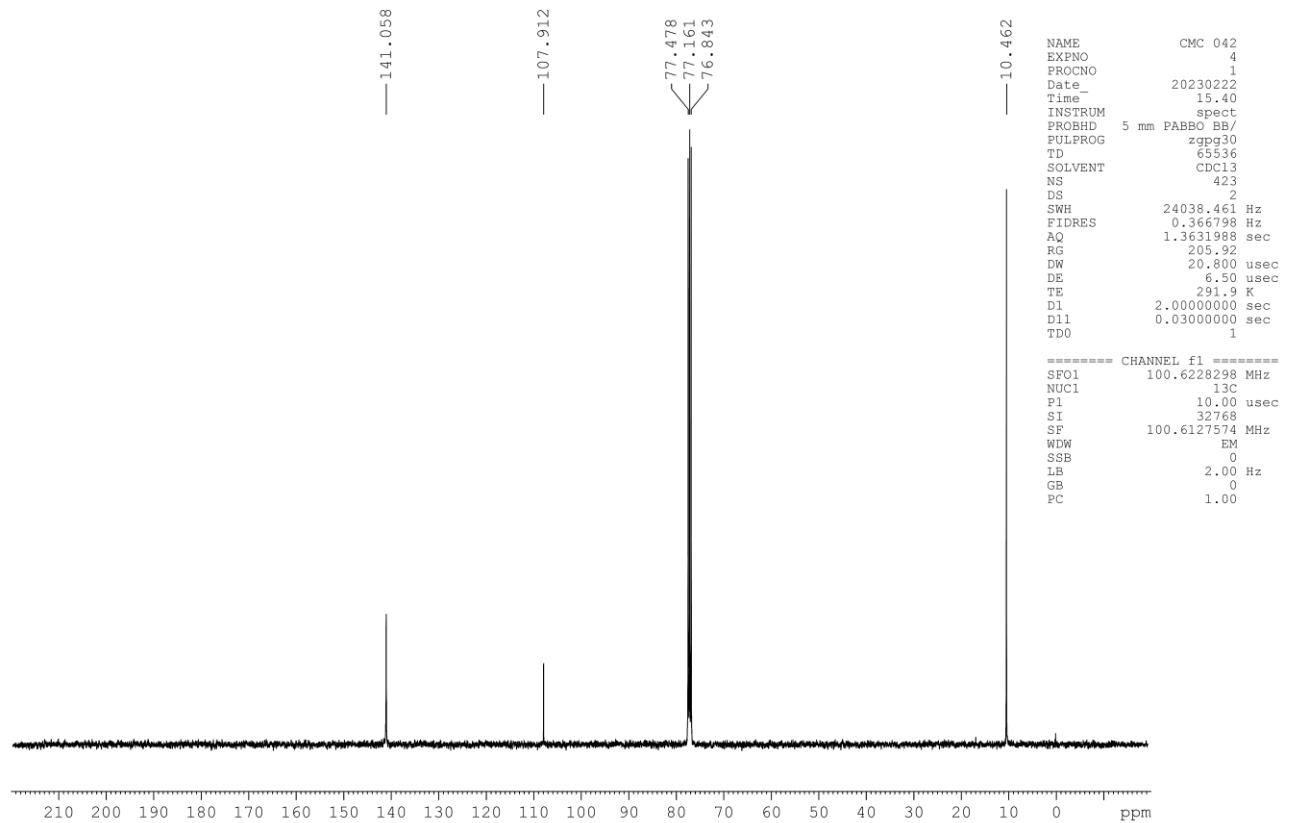
No.	NMR Spectra	Page
1	^1H and $^{13}\text{C}\{^1\text{H}\}$ NMR spectra of 4-chloro-3,5-dimethylpyrazole 1a	S21
2	^1H and $^{13}\text{C}\{^1\text{H}\}$ NMR spectra of 4-chloropyrazole 1b	S22
3	^1H and $^{13}\text{C}\{^1\text{H}\}$ NMR spectra of 4-chloro-3,5-diphenylpyrazole 1c	S23
4	^1H and $^{13}\text{C}\{^1\text{H}\}$ NMR spectra of 4-chloro-3,5-dimethyl-1 <i>H</i> -pyrazole-1-carboxamide 1d	S24
5	^1H and $^{13}\text{C}\{^1\text{H}\}$ NMR spectra of 4-chloro-1-(tetrahydro-2 <i>H</i> -pyran-2-yl)-1 <i>H</i> -pyrazole 1e	S25
6	^1H and $^{13}\text{C}\{^1\text{H}\}$ NMR spectra of 4-chloro-3,5-dimethyl-1-phenyl-1 <i>H</i> -pyrazole 1f	S26
7	^1H NMR spectra of side product 1f-2 and 1f-3	S27
8	^1H and $^{13}\text{C}\{^1\text{H}\}$ NMR spectra of 4-chloro-1-phenylpyrazole 1g	S28
9	^1H and $^{13}\text{C}\{^1\text{H}\}$ NMR spectra of 4-chloro-1-(4-methoxyphenyl)-1 <i>H</i> -pyrazole 1h	S29
10	^1H and $^{13}\text{C}\{^1\text{H}\}$ NMR spectra of 1-(4-bromophenyl)-4-chloro-1 <i>H</i> -pyrazole 1i	S30-S31
11	^1H and $^{13}\text{C}\{^1\text{H}\}$ NMR spectra of 4-(4-chloro-1 <i>H</i> -pyrazol-1-yl)benzaldehyde 1j	S32
12	^1H and $^{13}\text{C}\{^1\text{H}\}$ NMR spectra of 4-(4-chloro-1 <i>H</i> -pyrazol-1-yl)benzotrile 1k	S33
13	^1H and $^{13}\text{C}\{^1\text{H}\}$ NMR spectra of 4-chloro-1-(4-nitrophenyl)-1 <i>H</i> -pyrazole 1l	S34
14	^1H and $^{13}\text{C}\{^1\text{H}\}$ NMR spectra of 2-(4-chloro-1 <i>H</i> -pyrazol-1-yl)pyridine 1m	S35
15	^1H , $^{13}\text{C}\{^1\text{H}\}$ and $^{19}\text{F}\{^1\text{H}\}$ NMR spectra 4-chloro-celecoxib 1n	S36-S37

4-Chloro-3,5-dimethylpyrazole **1a**

^1H , CDCl_3

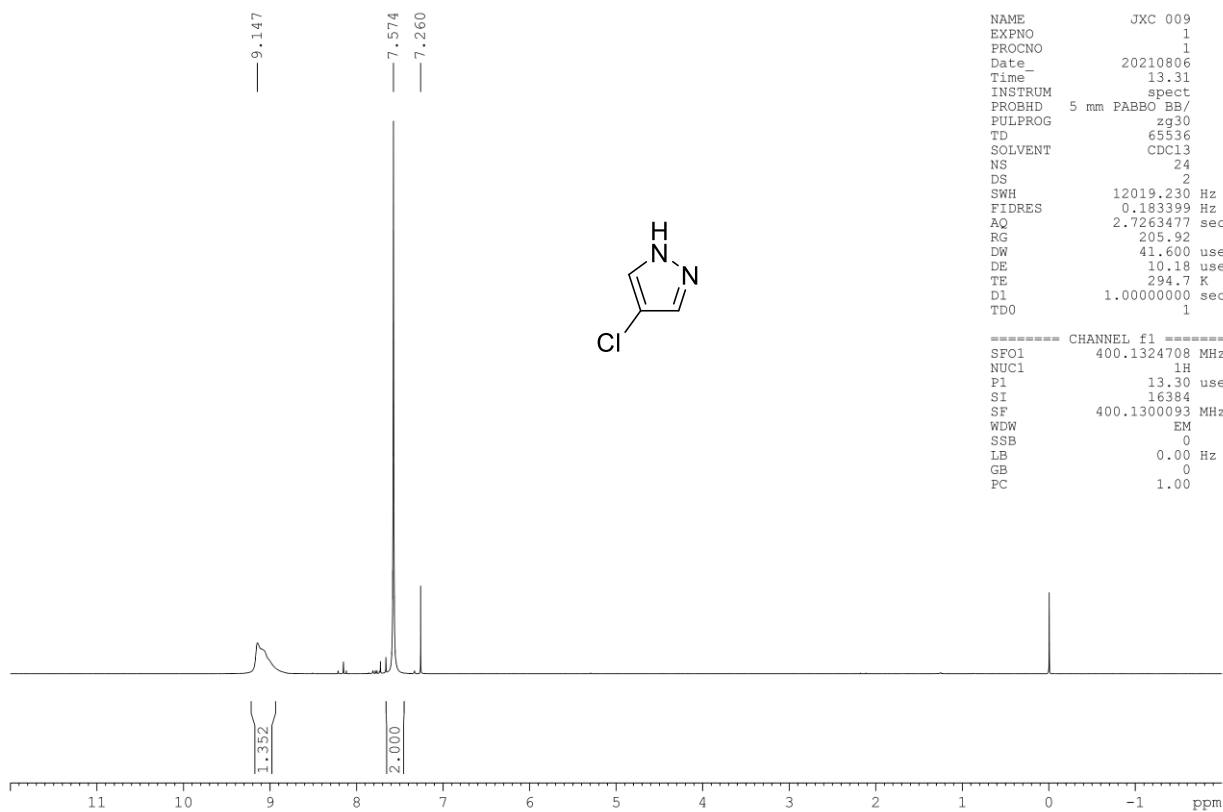


$^{13}\text{C}\{^1\text{H}\}$, CDCl_3

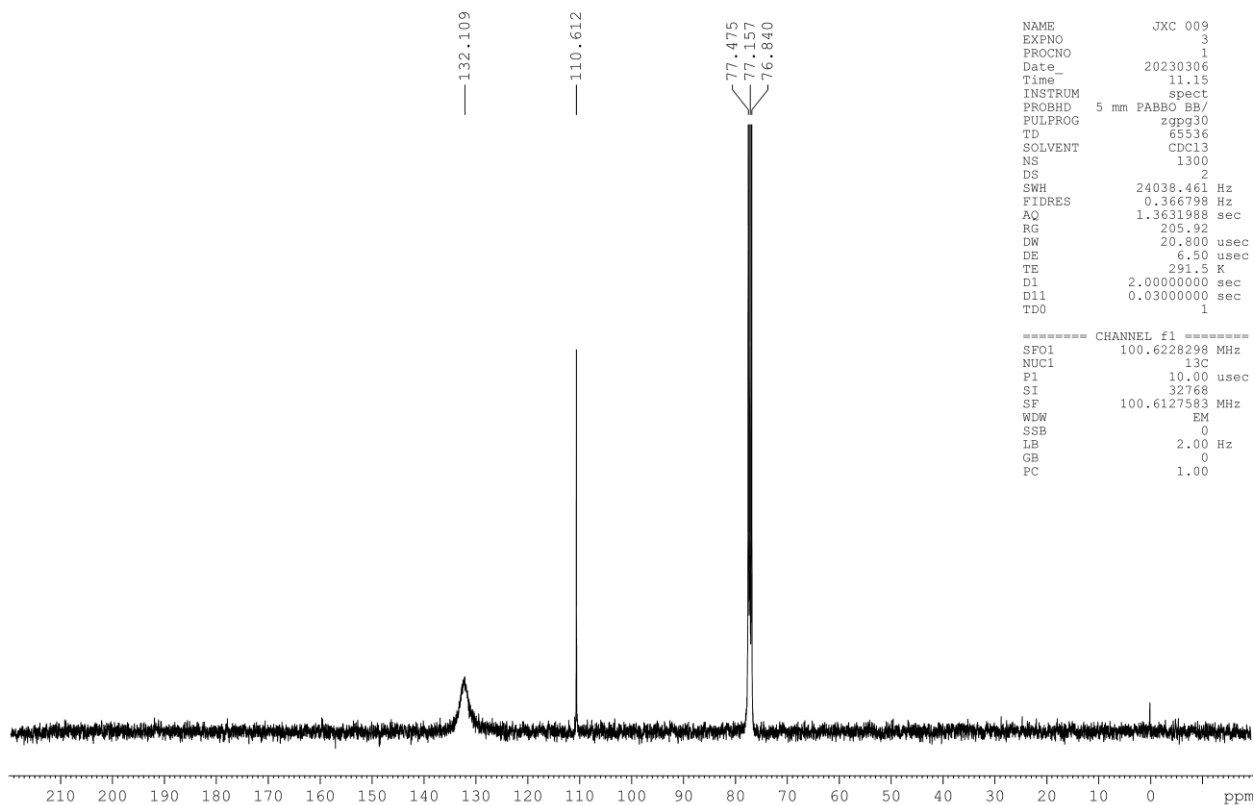


4-Chloropyrazole **1b**

^1H , CDCl_3

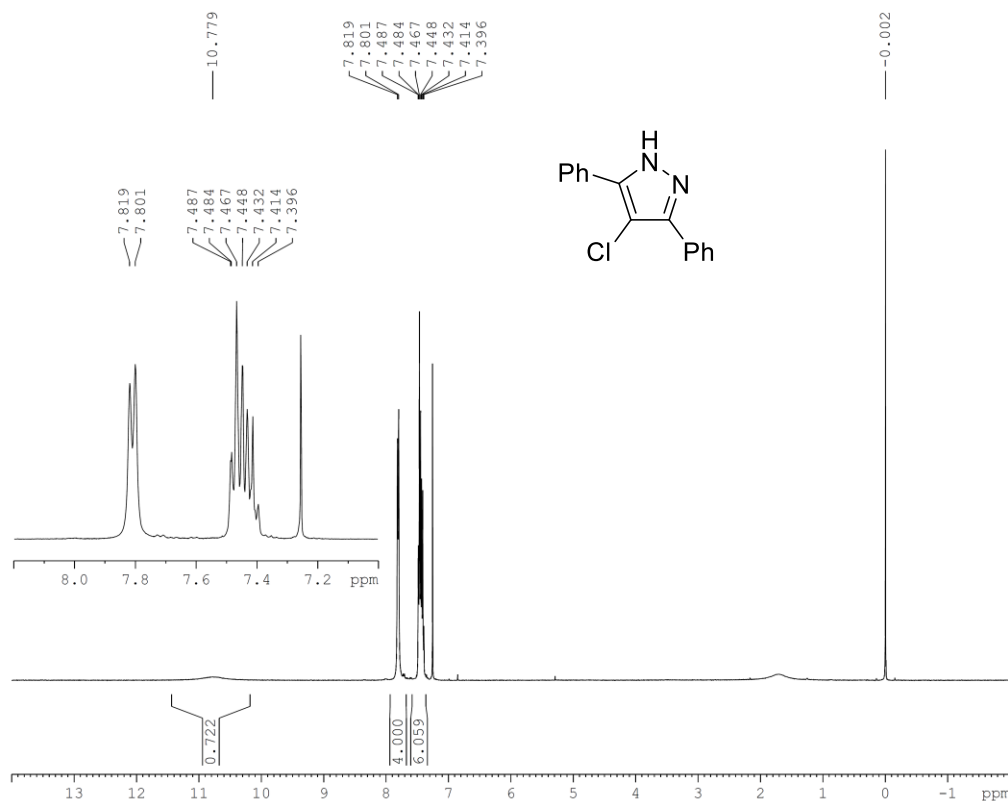


$^{13}\text{C}\{^1\text{H}\}$, CDCl_3



4-Chloro-3,5-diphenylpyrazole 1c

^1H , CDCl_3



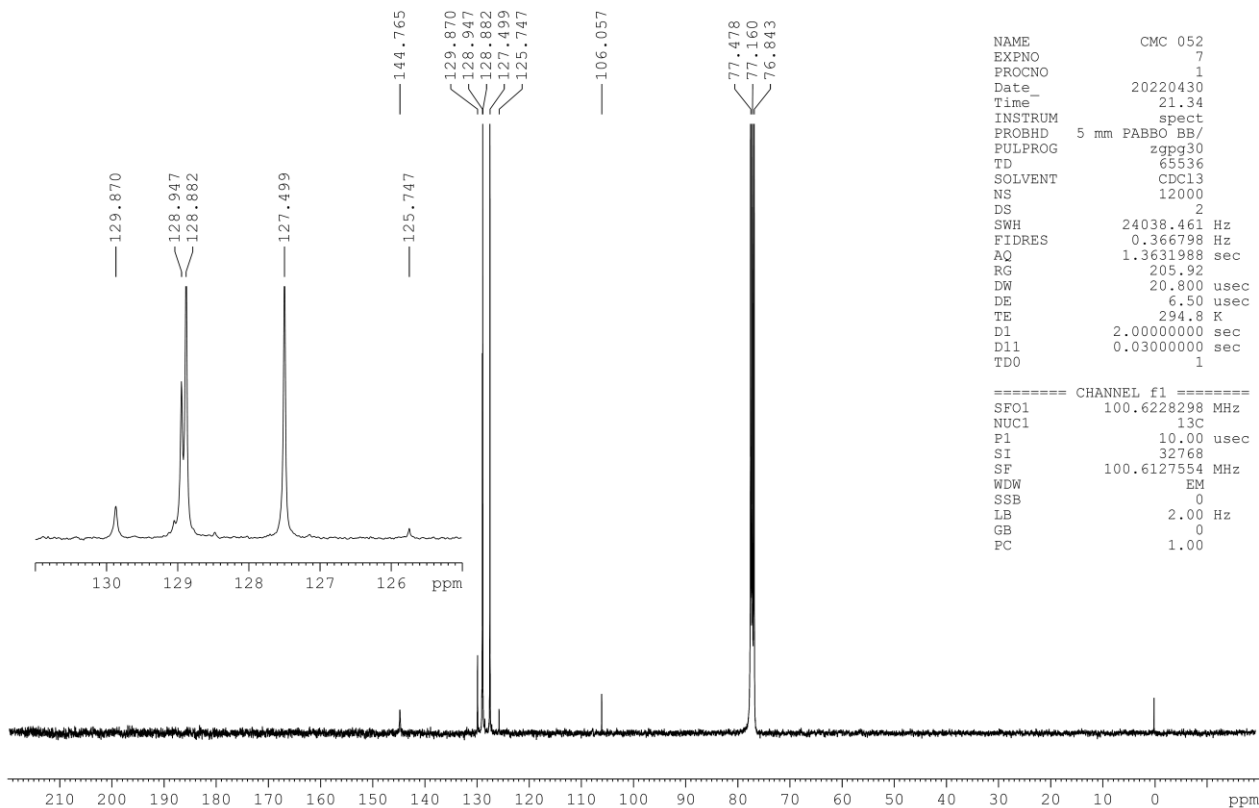
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FIDRES        0.183399 Hz
AQ            2.7263477 sec
RG            181.8
DW            41.600 usec
DE            10.18 usec
TE            293.9 K
D1            2.00000000 sec
D10           1
    
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P1            13.30 usec
SI            16384
SF            400.1300111 MHz
WDW           EM
SSB           0
LB            0.00 Hz
GB            0
PC            1.00
    
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$^{13}\text{C}\{^1\text{H}\}$, CDCl_3



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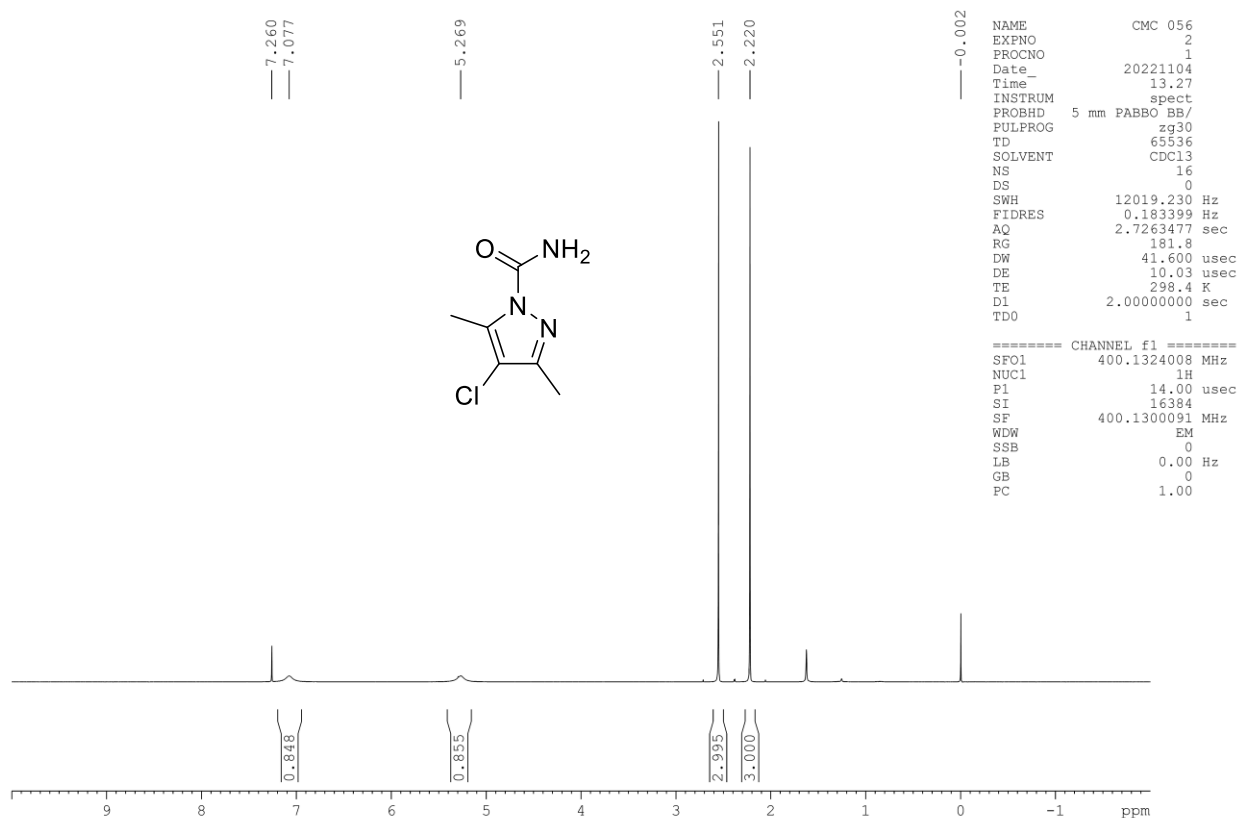
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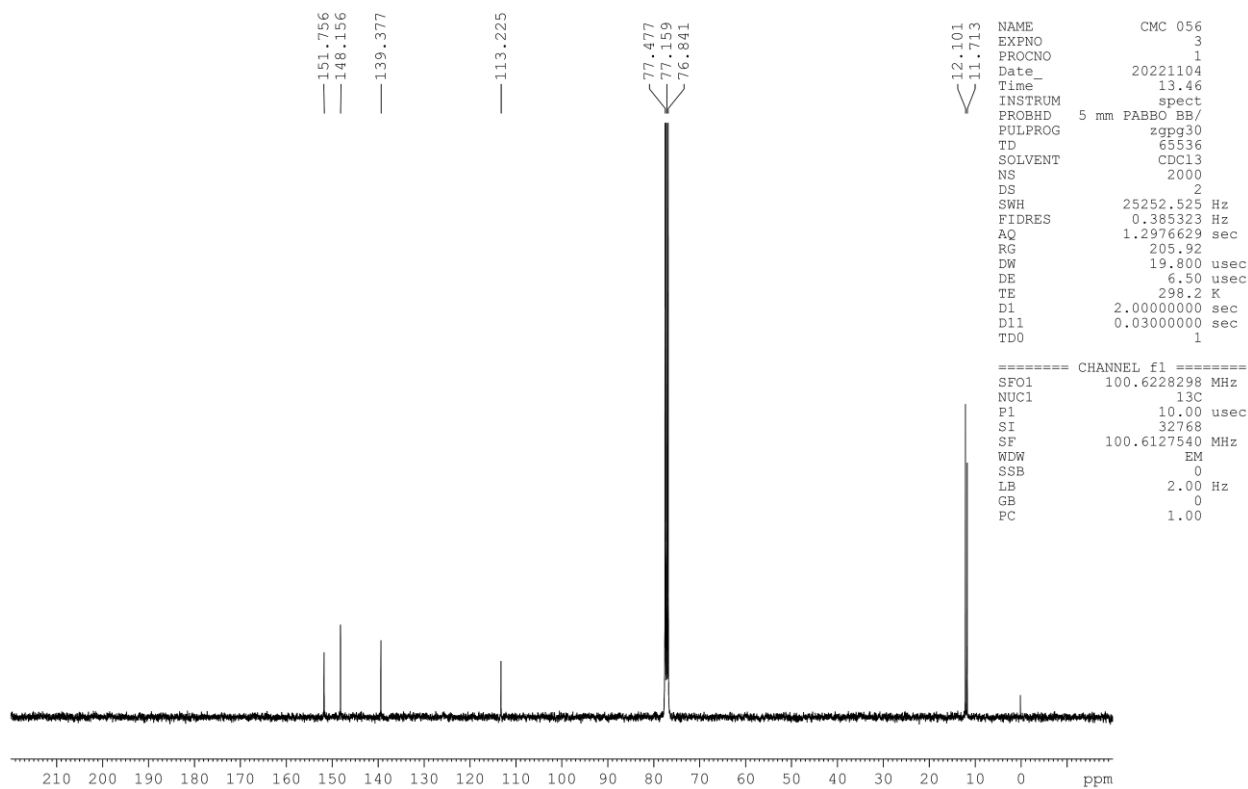
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P1            10.00 usec
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WDW           EM
SSB           0
LB            2.00 Hz
GB            0
PC            1.00
    
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4-Chloro-3,5-dimethyl-1H-pyrazole-1-carboxamide **1d**

^1H , CDCl_3

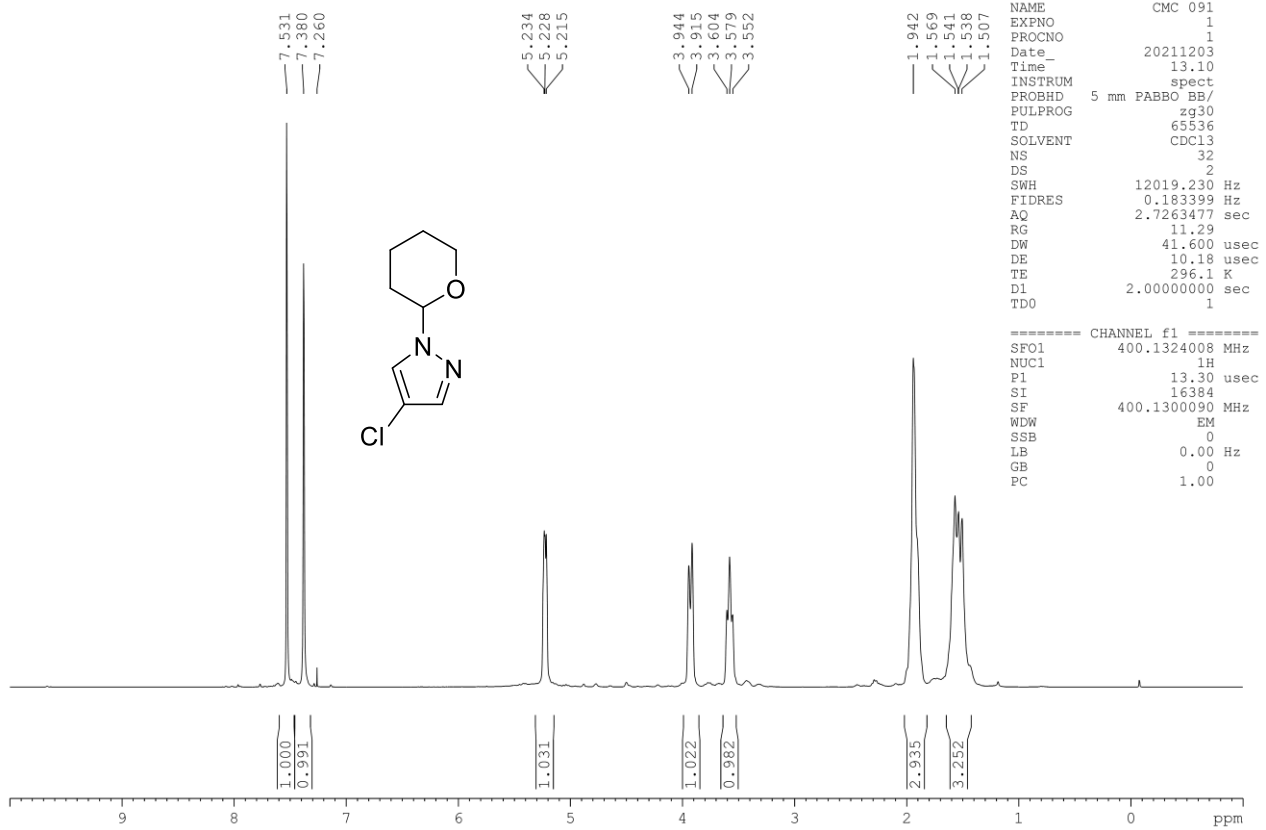


$^{13}\text{C}\{^1\text{H}\}$, CDCl_3

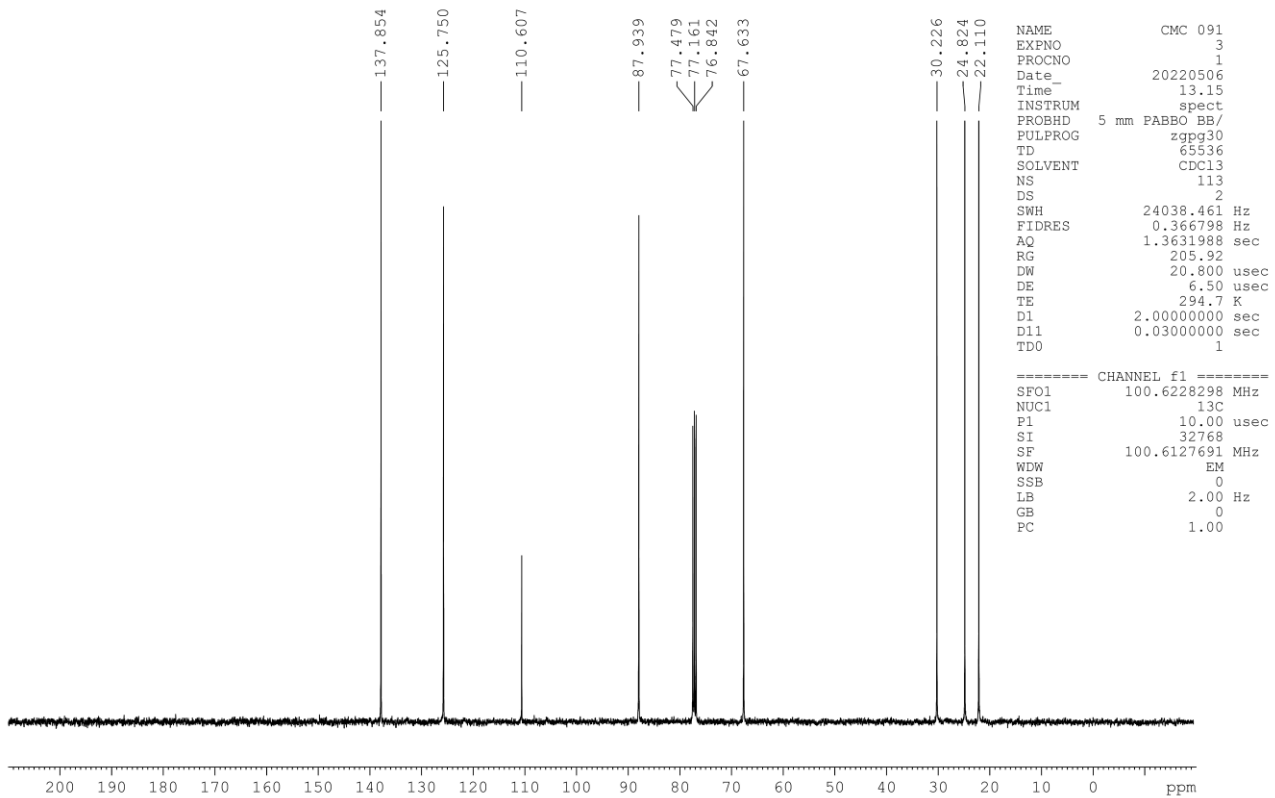


4-Chloro-1-(tetrahydro-2H-pyran-2-yl)-1H-pyrazole **1e**

^1H , CDCl_3

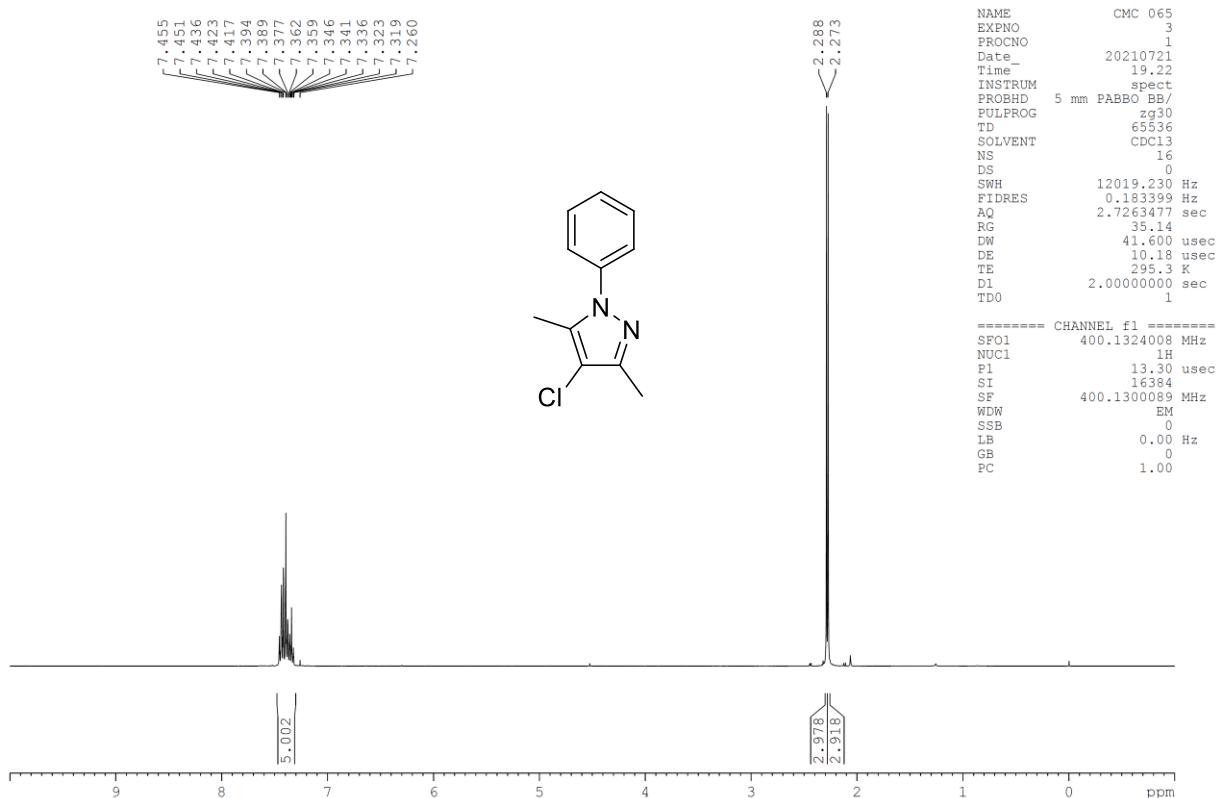


$^{13}\text{C}\{^1\text{H}\}$, CDCl_3

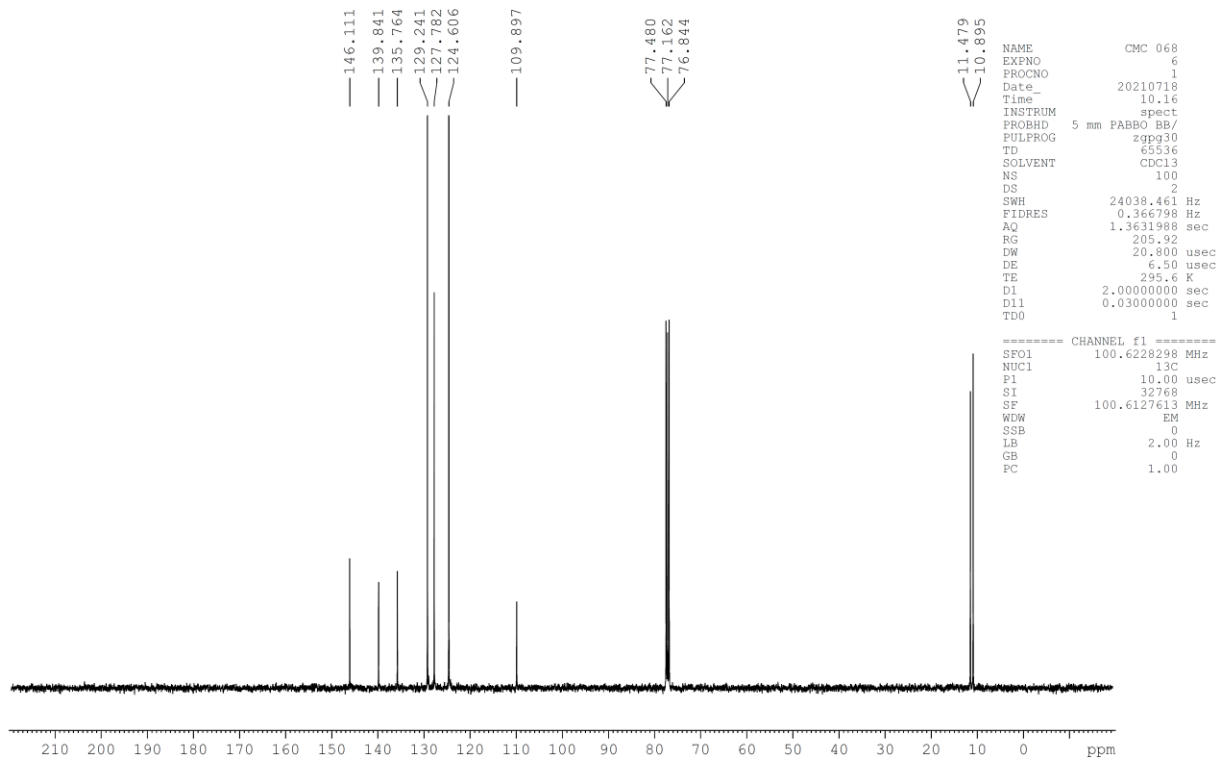


4-Chloro-3,5-dimethyl-1-phenylpyrazole **1f**

^1H , CDCl_3

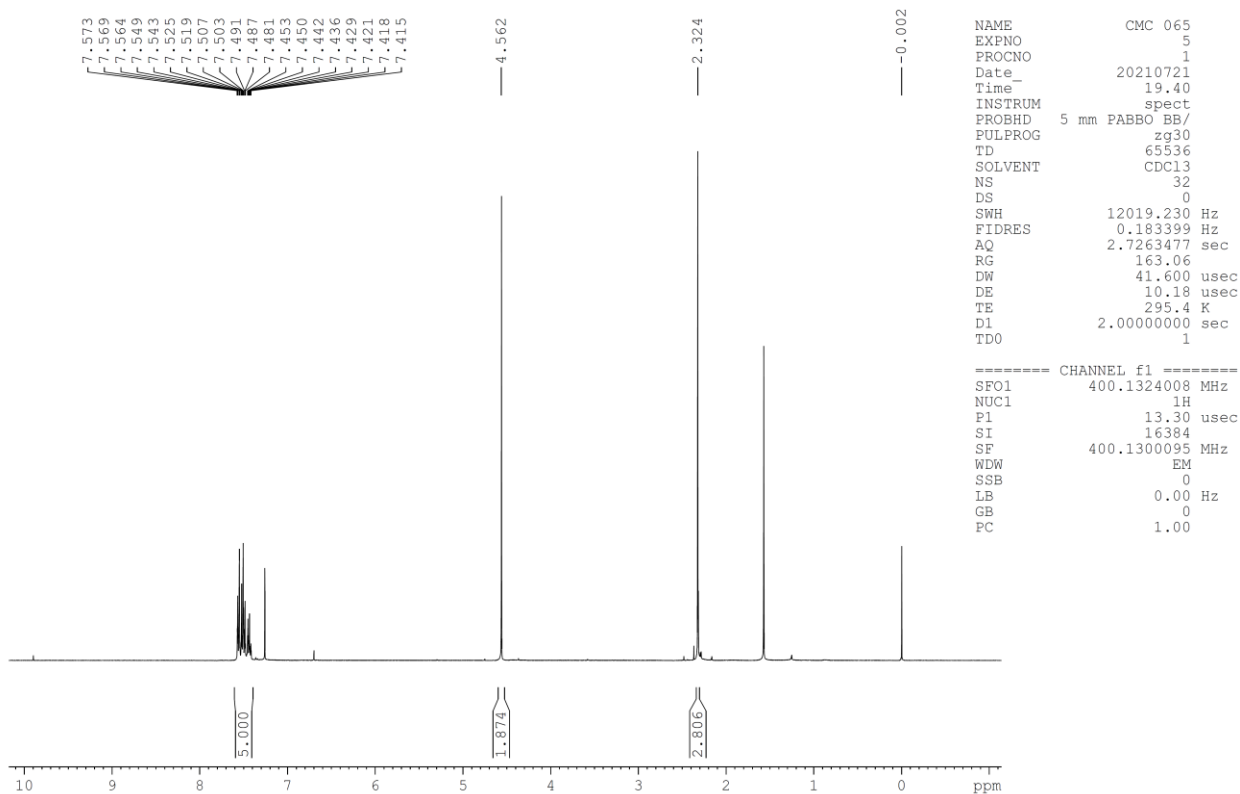


$^{13}\text{C}\{^1\text{H}\}$, CDCl_3



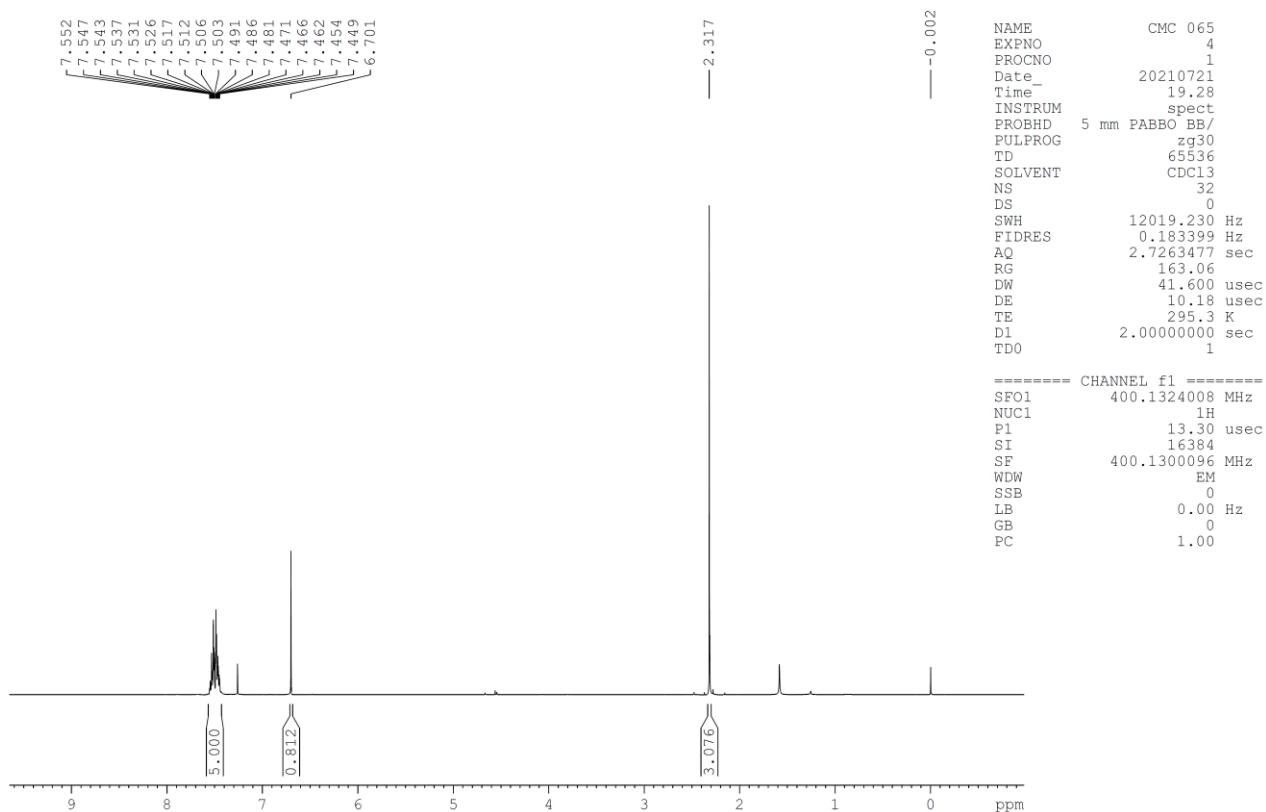
di-chlorinated side product of 3,5-dimethyl-1-phenylpyrazole **1f-2**

¹H, CDCl₃



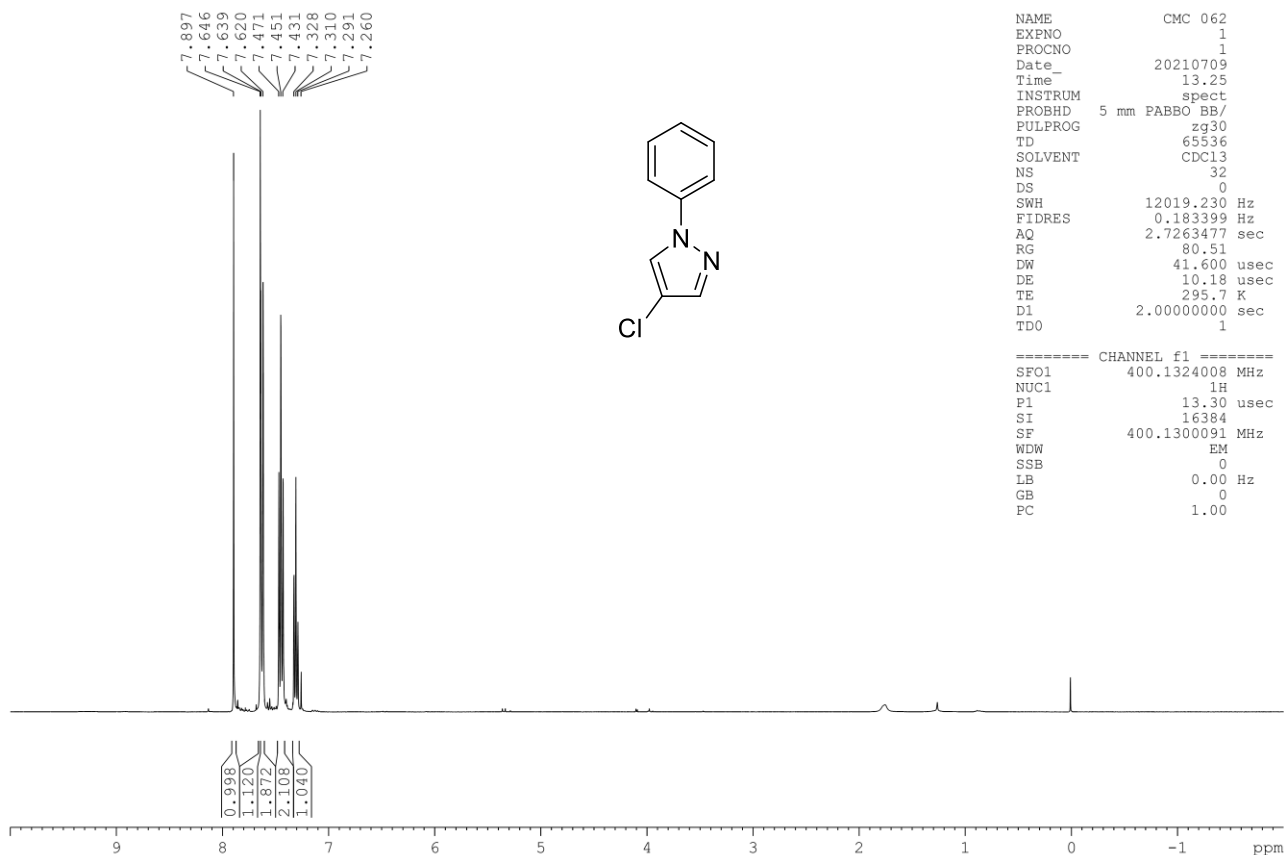
di-chlorinated side product of 3,5-dimethyl-1-phenylpyrazole **1f-3**

¹H, CDCl₃

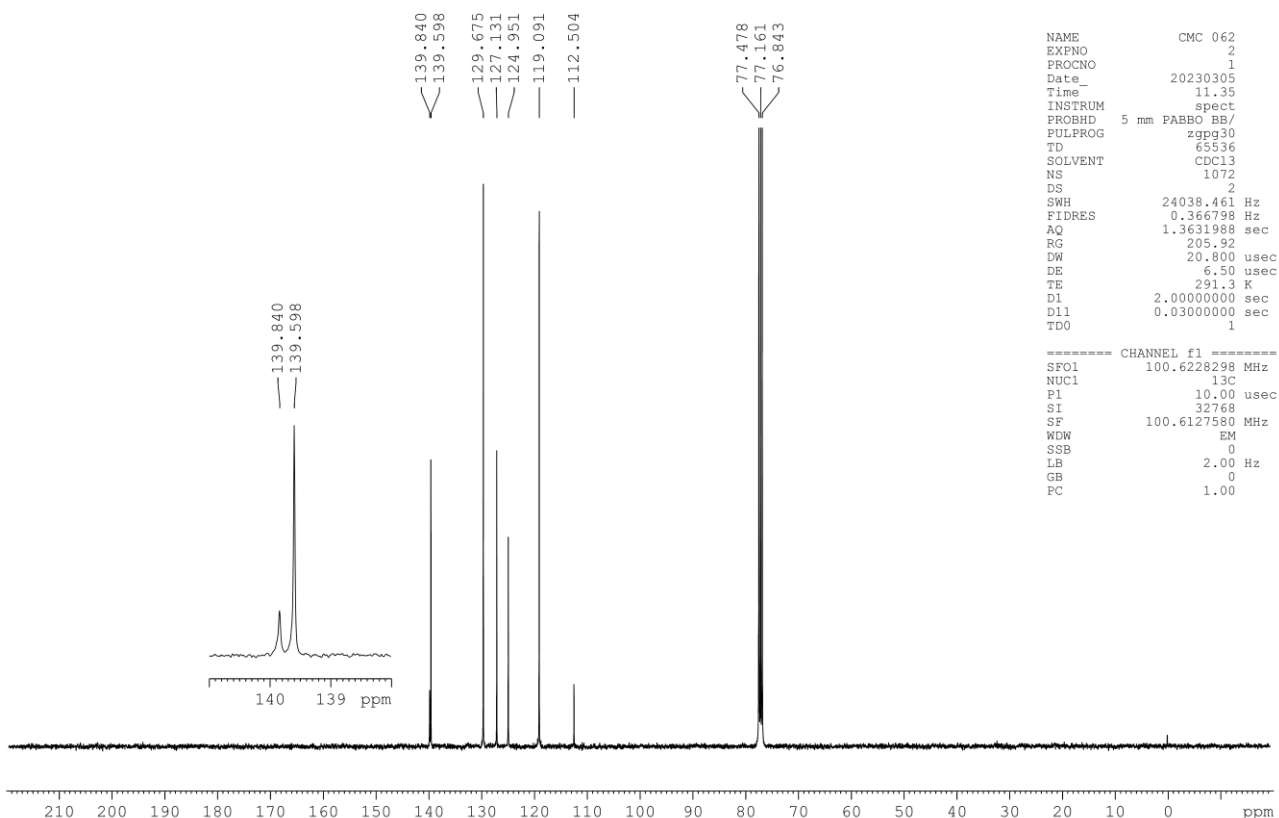


4-Chloro-1-phenylpyrazole **1g**

^1H , CDCl_3

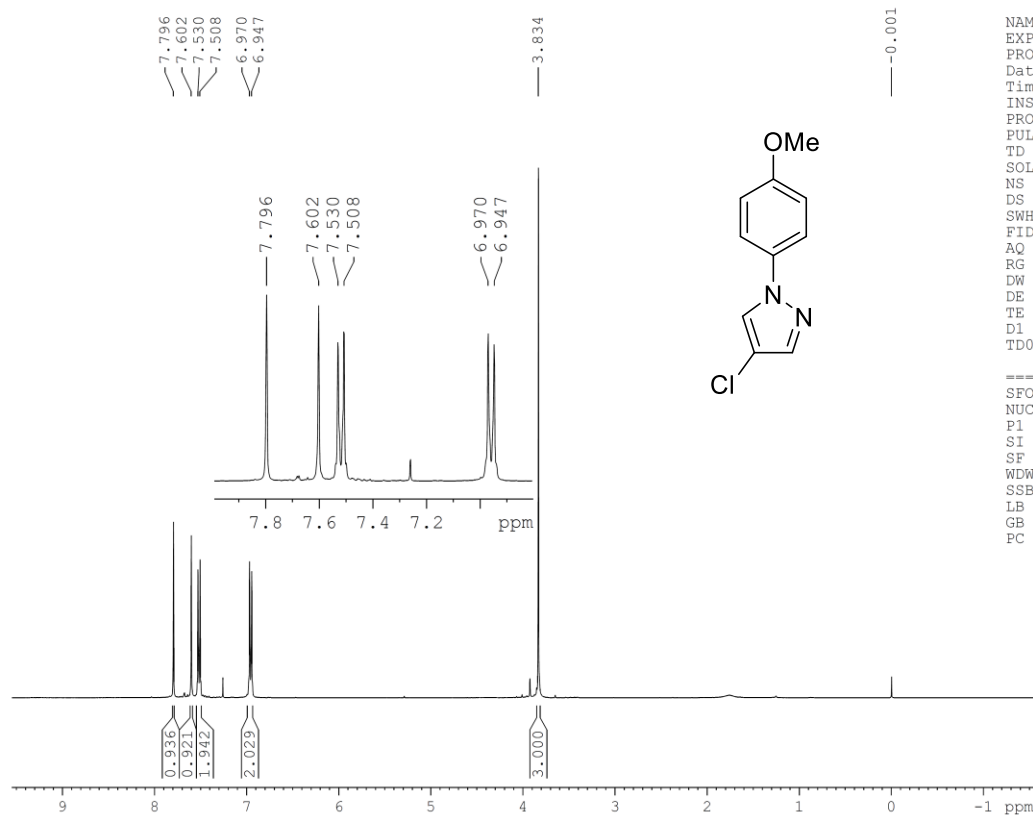


$^{13}\text{C}\{^1\text{H}\}$, CDCl_3



4-chloro-1-(4-methoxyphenyl)pyrazole **1h**

^1H , CDCl_3



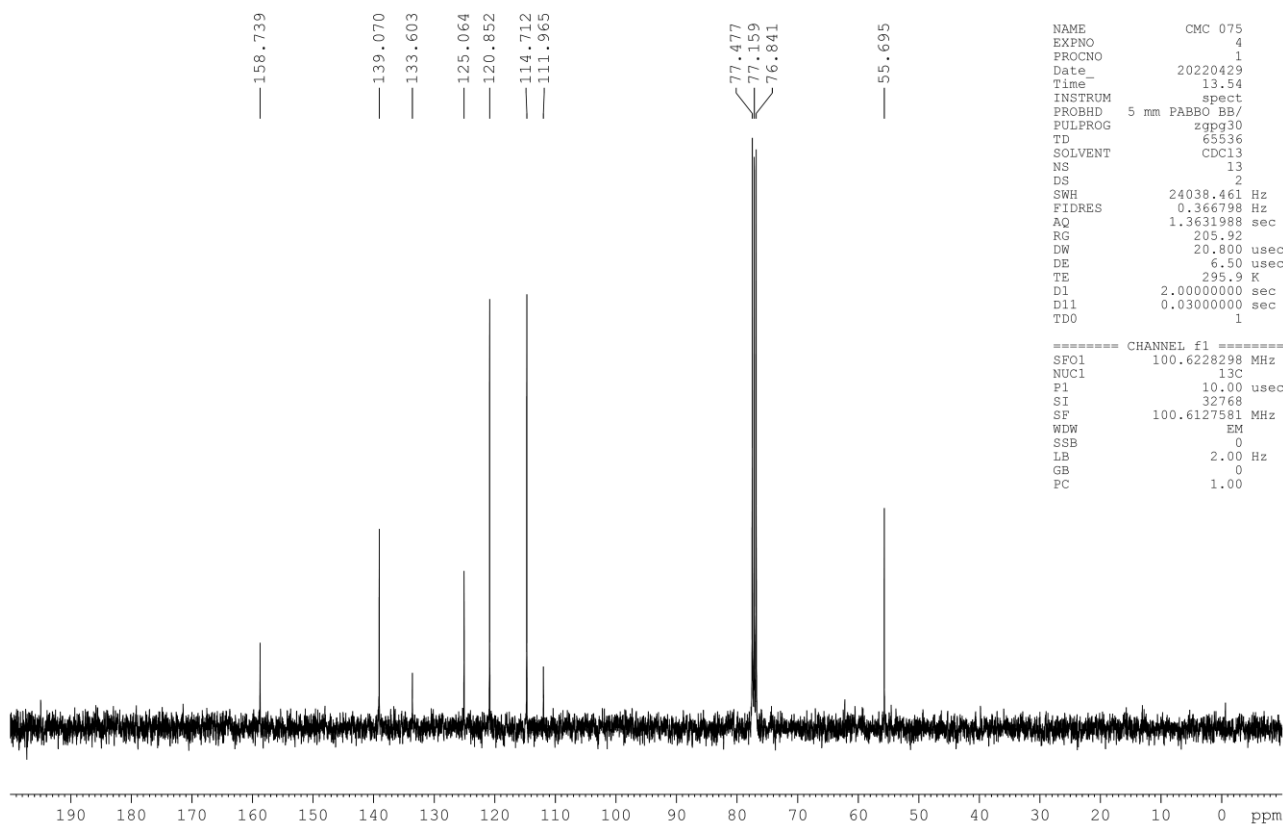
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Time_         13.41
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PULPROG       zg30
TD            65536
SOLVENT       CDCl3
NS            32
DS            0
SWH           12019.230 Hz
FIDRES        0.183399 Hz
AQ            2.7263477 sec
RG            91.1
DW            41.600 usec
DE            10.18 usec
TE            294.7 K
D1            2.00000000 sec
TD0           1
    
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GB            0
PC            1.00
    
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$^{13}\text{C}\{^1\text{H}\}$, CDCl_3



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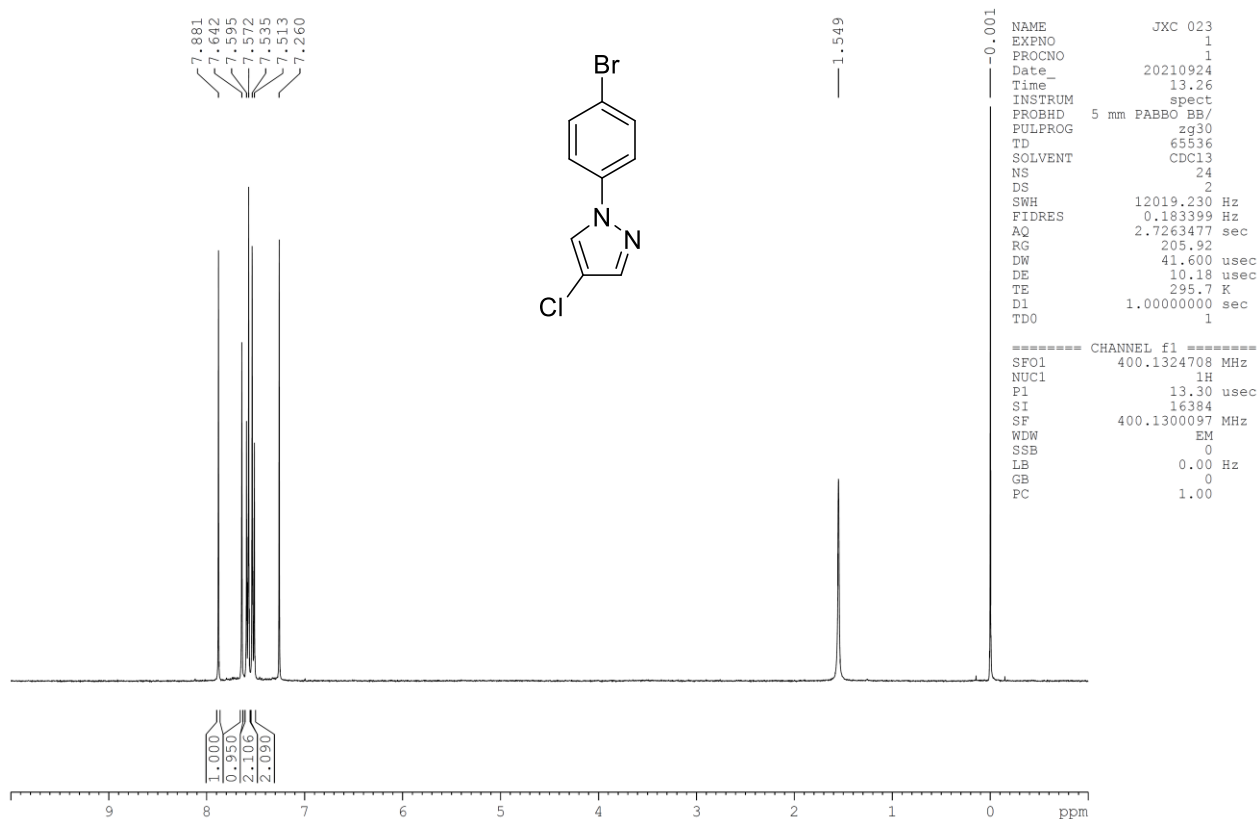
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PROCNO        1
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PULPROG       zgpg30
TD            65536
SOLVENT       CDCl3
NS            13
DS            2
SWH           24038.461 Hz
FIDRES        0.366798 Hz
AQ            1.3631988 sec
RG            205.92
DW            20.800 usec
DE            6.50 usec
TE            295.9 K
D1            2.00000000 sec
D11           0.03000000 sec
TD0           1
    
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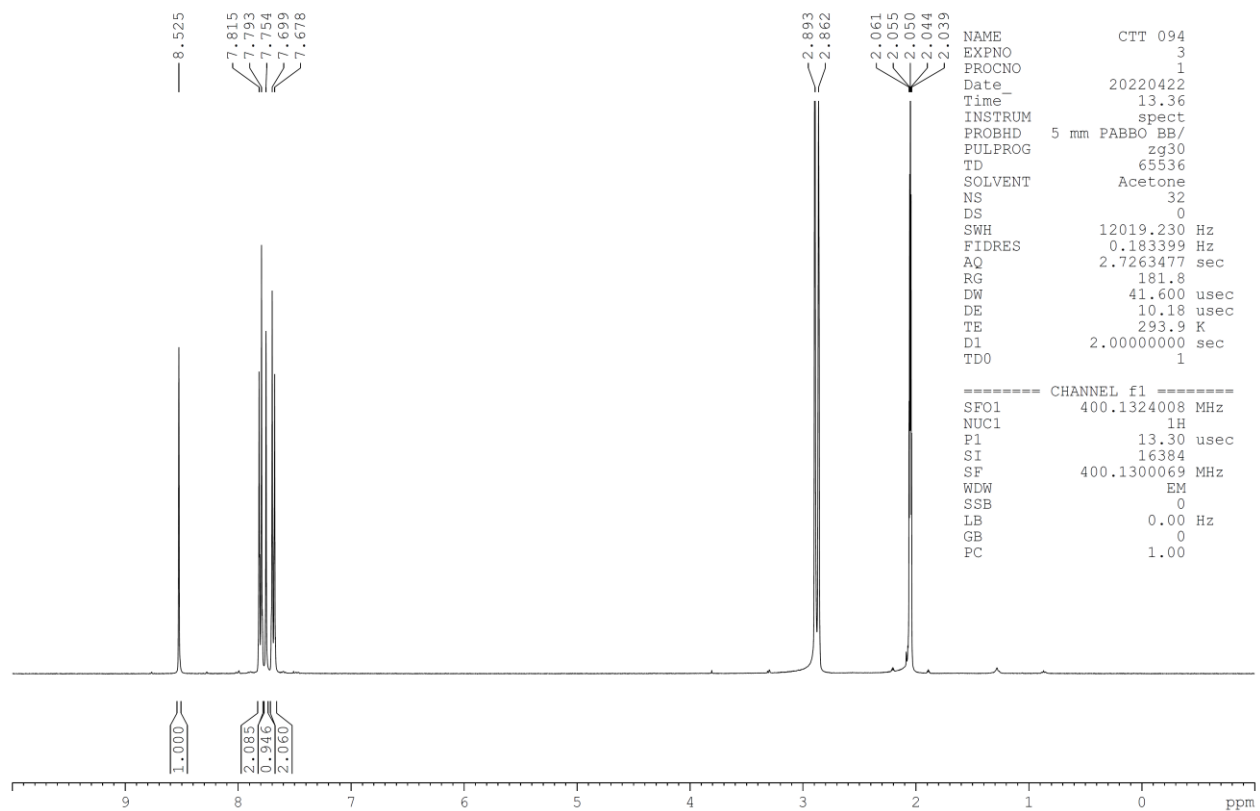
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SI            32768
SF            100.6127581 MHz
WDW           EM
SSB           0
LB            2.00 Hz
GB            0
PC            1.00
    
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4-Chloro-1-(4-bromophenyl)pyrazole **1i**

¹H, CDCl₃

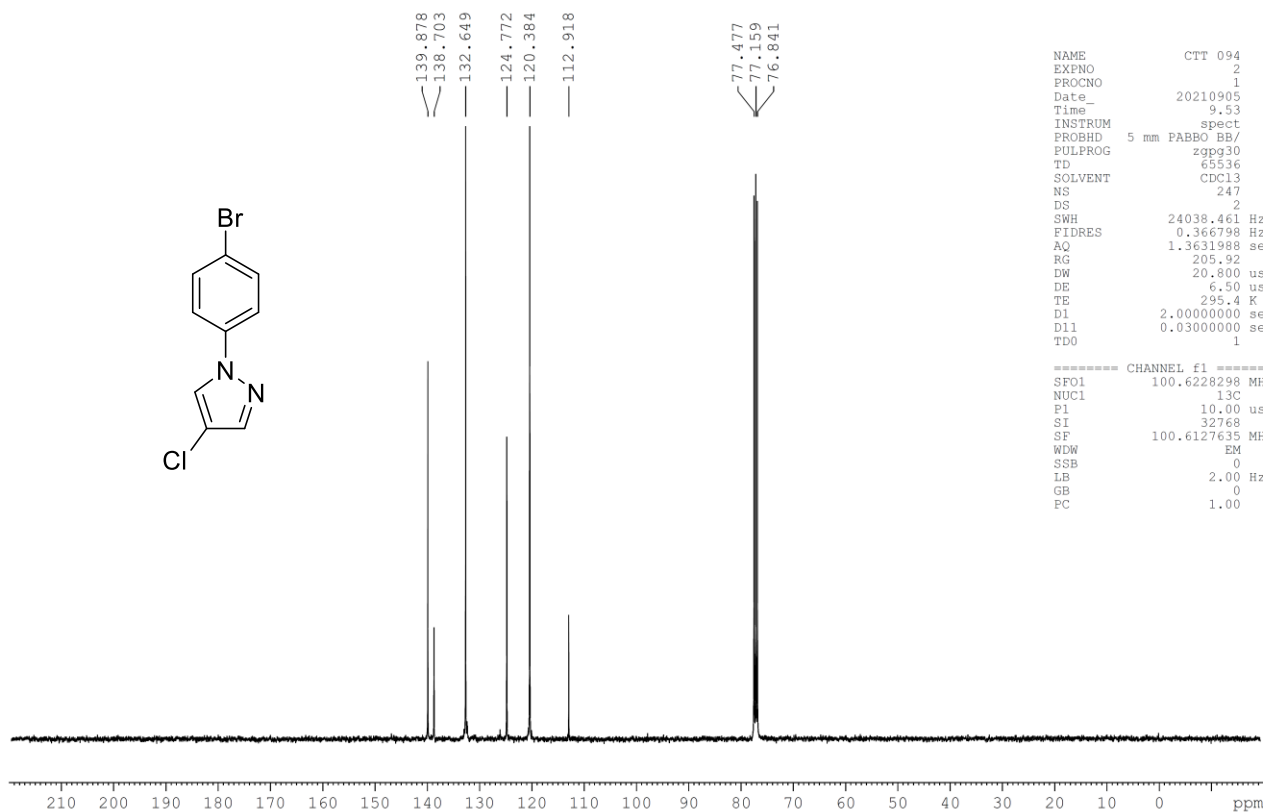
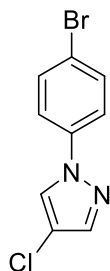


¹H, acetone-d₆

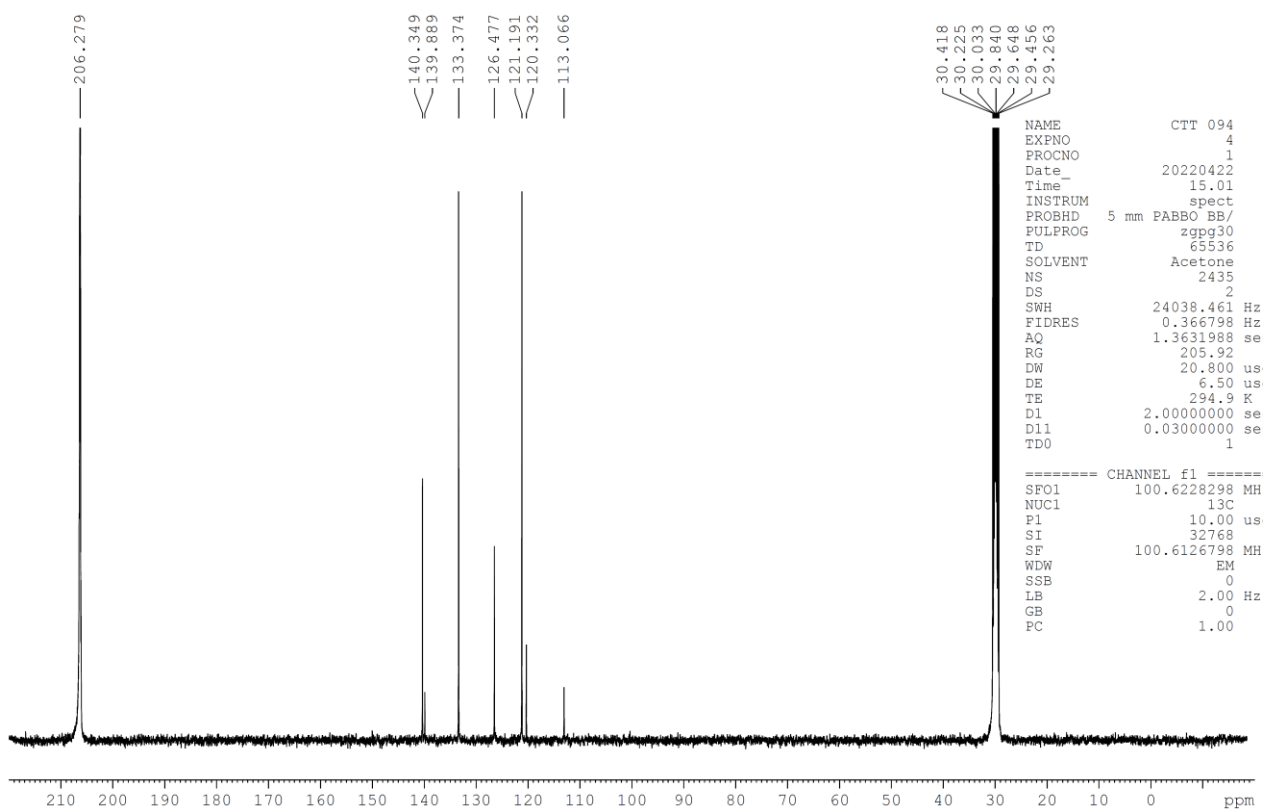


4-Chloro-1-(4-bromophenyl)pyrazole **1i**

$^{13}\text{C}\{^1\text{H}\}$, CDCl_3

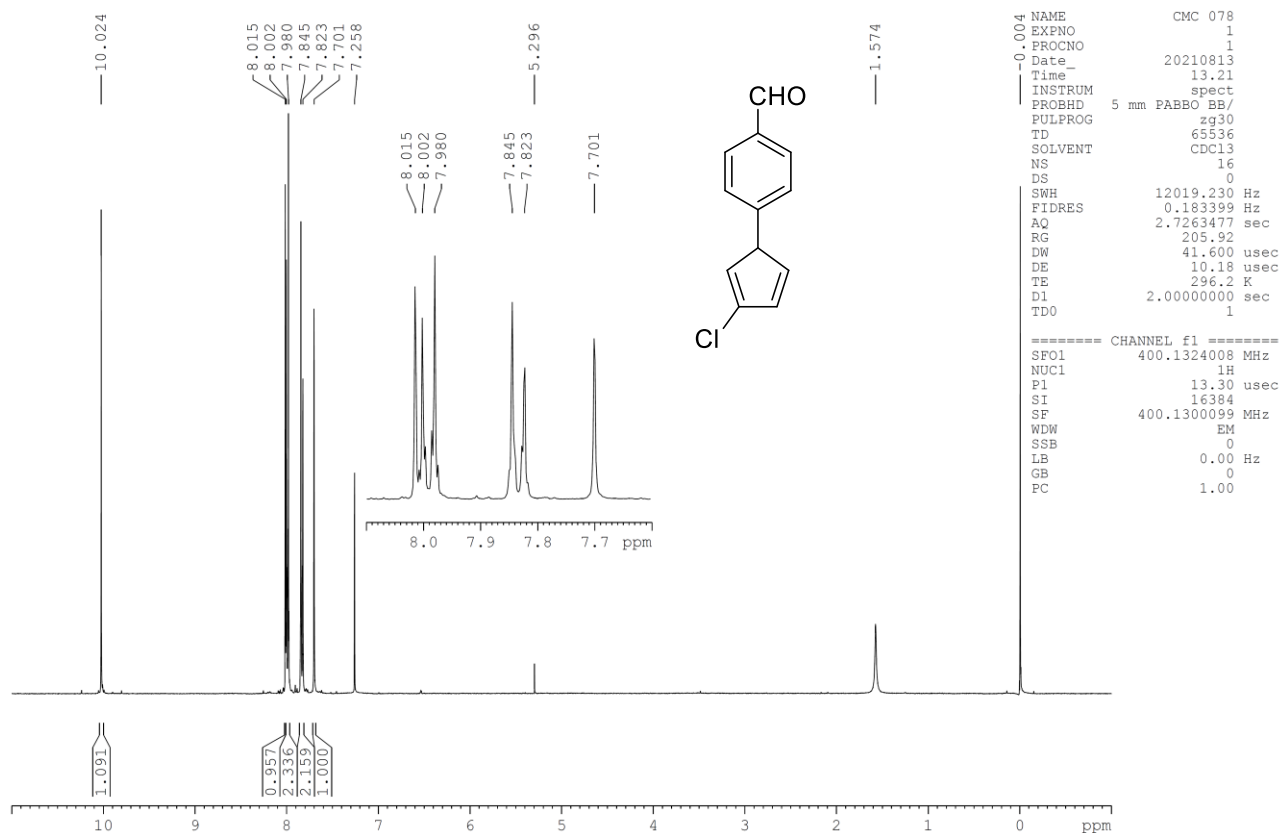


$^{13}\text{C}\{^1\text{H}\}$, acetone- d_6

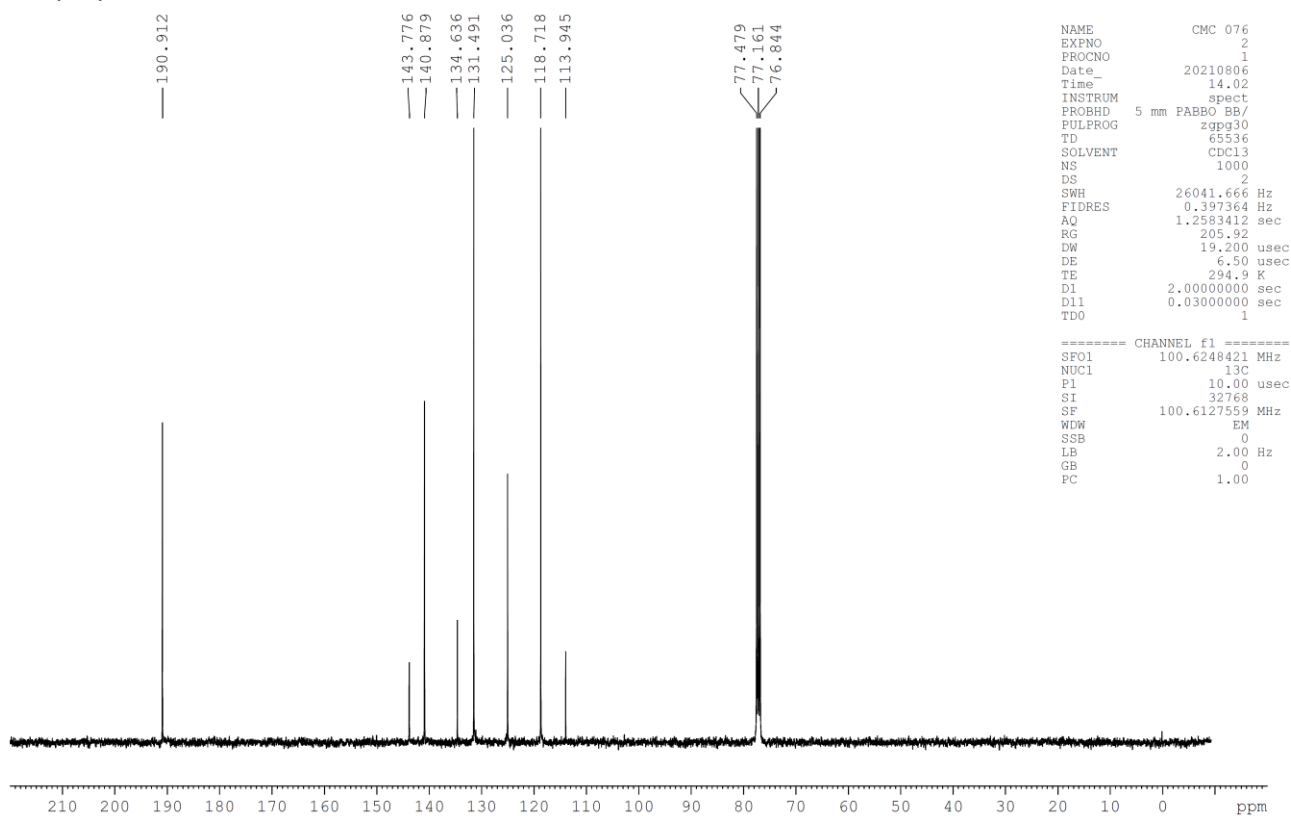


4-(4-chloro-1H-pyrazol-1-yl)benzaldehyde **1j**

¹H, CDCl₃

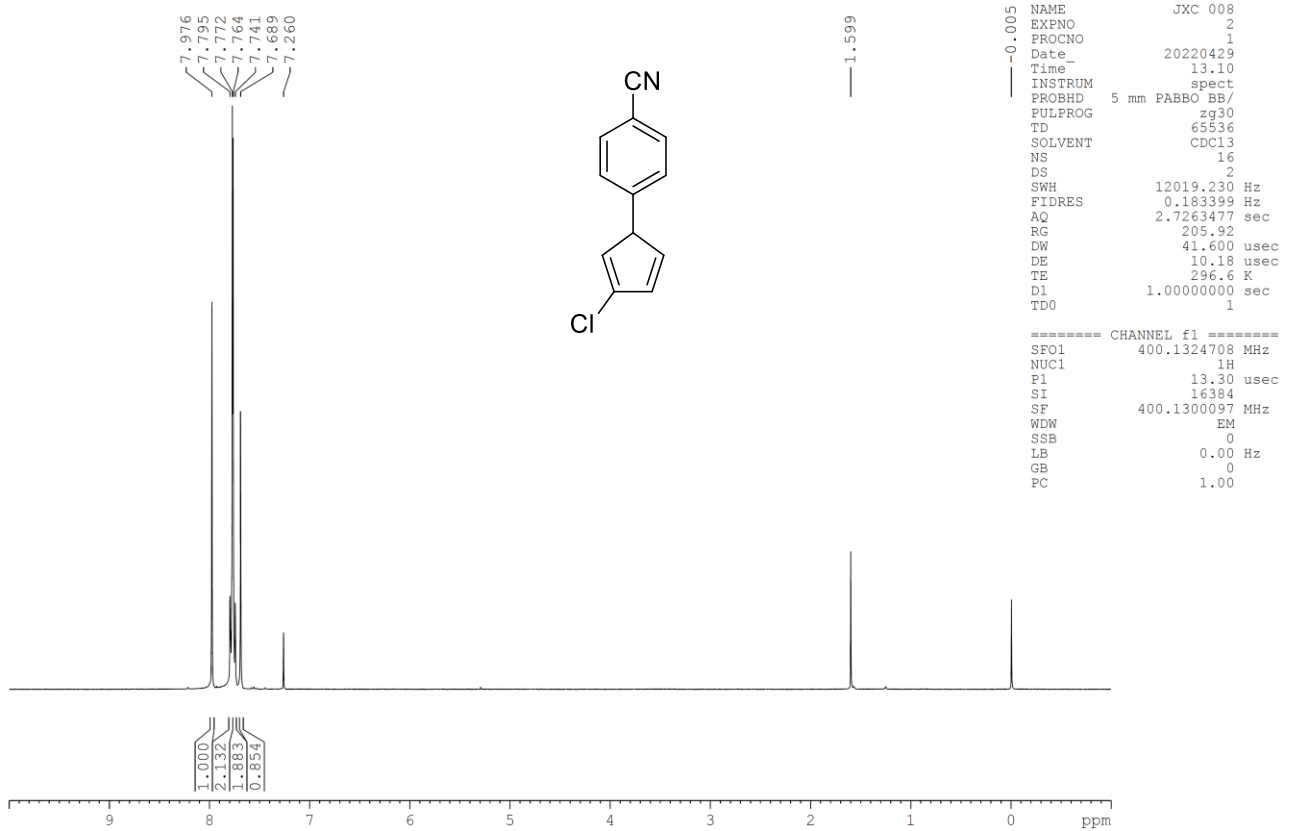


¹³C{¹H}, CDCl₃

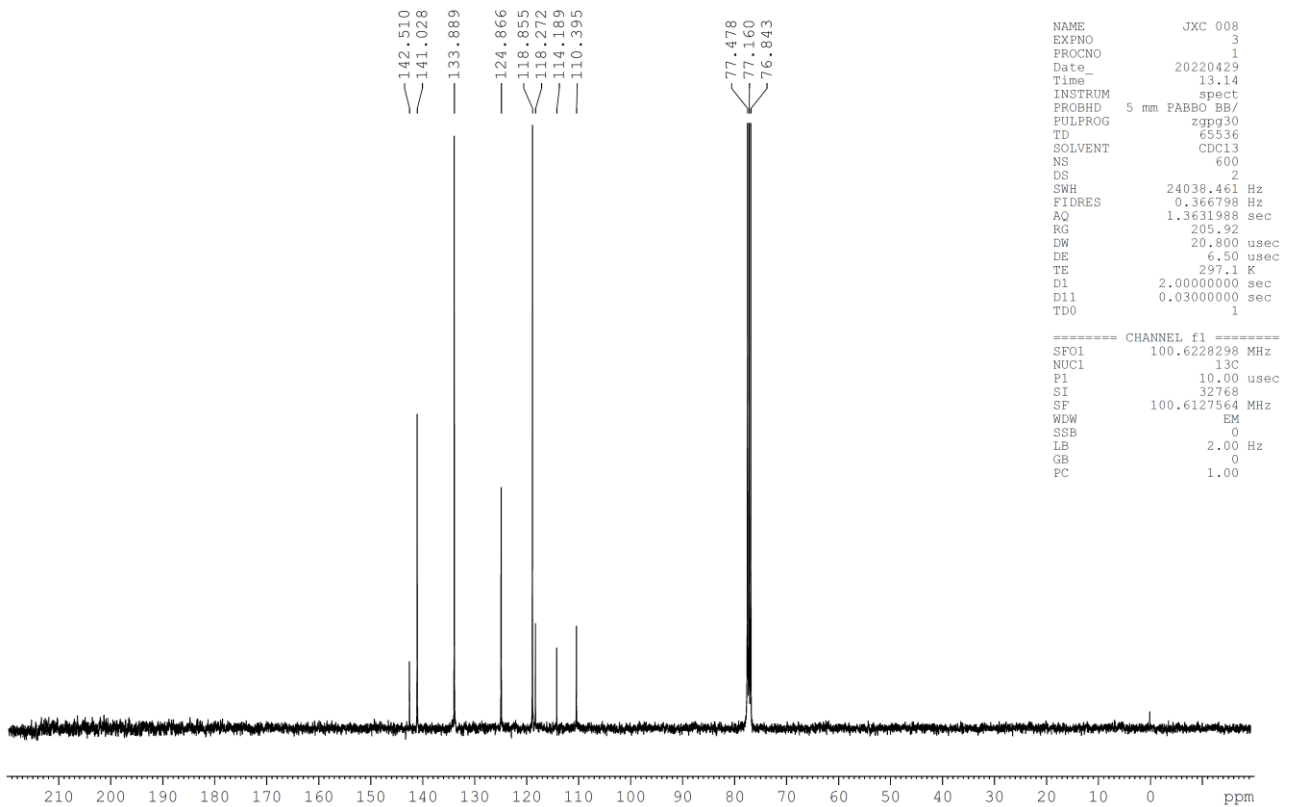


4-(4-chloro-1H-pyrazol-1-yl)benzotrile **1k**

¹H, CDCl₃

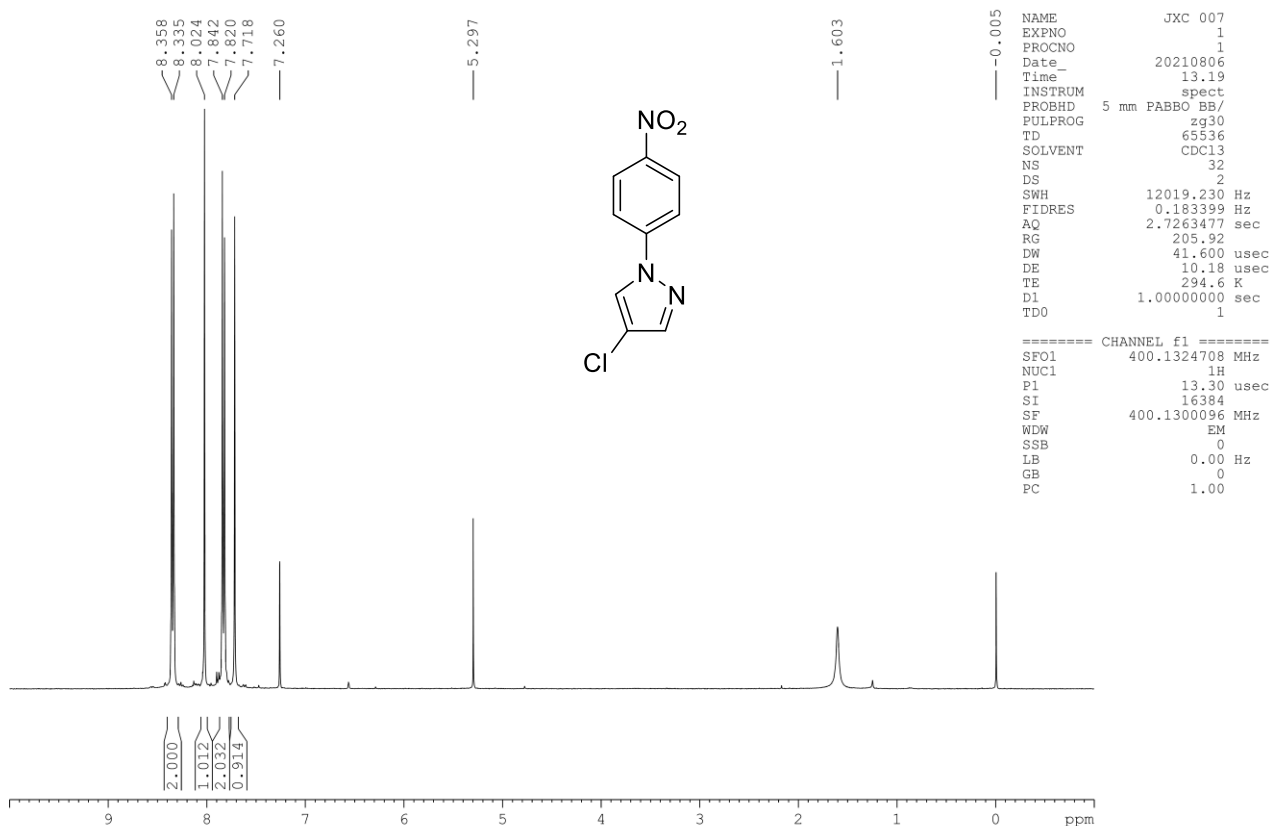


¹³C {¹H}, CDCl₃

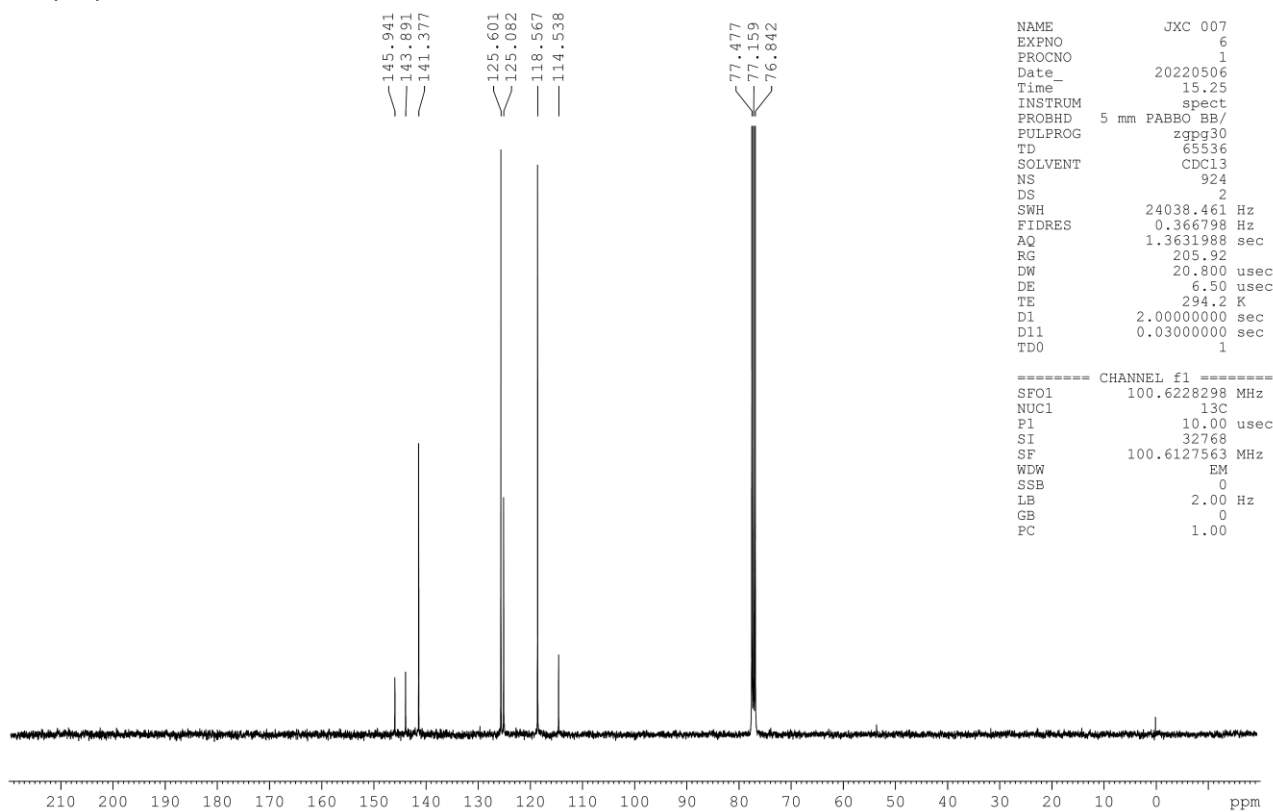


4-chloro-1-(4-nitrophenyl)-1H-pyrazole **11**

¹H, CDCl₃

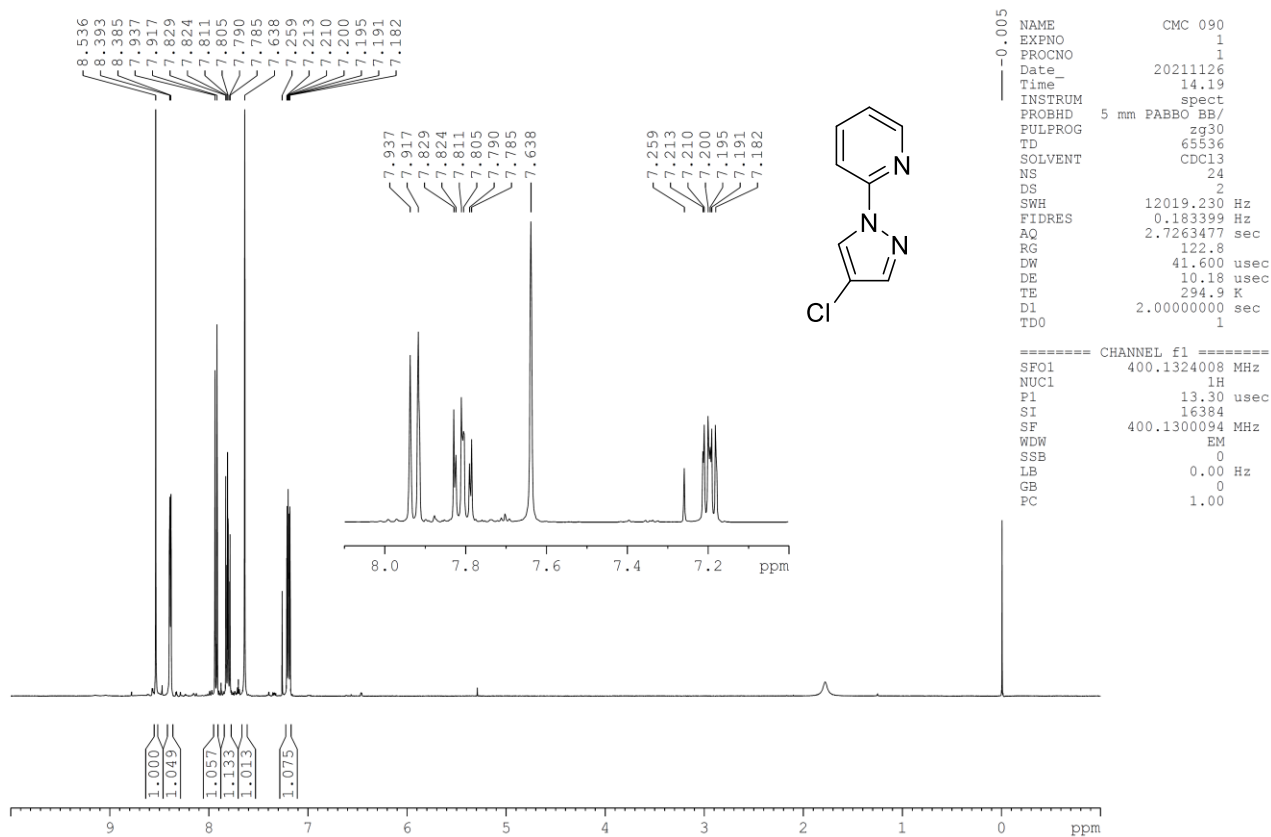


¹³C {¹H}, CDCl₃

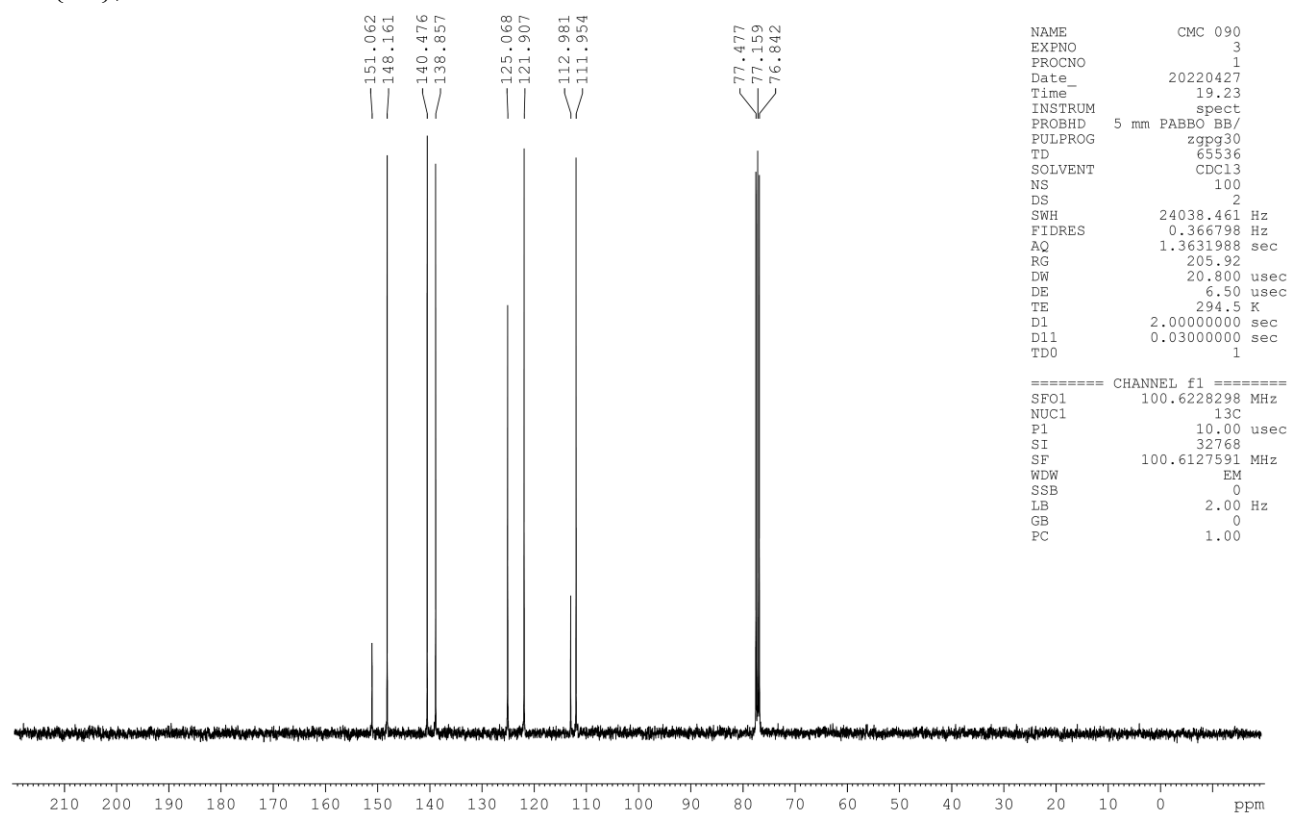


2-(4-chloro-1H-pyrazol-1-yl)pyridine **1m**

¹H, CDCl₃

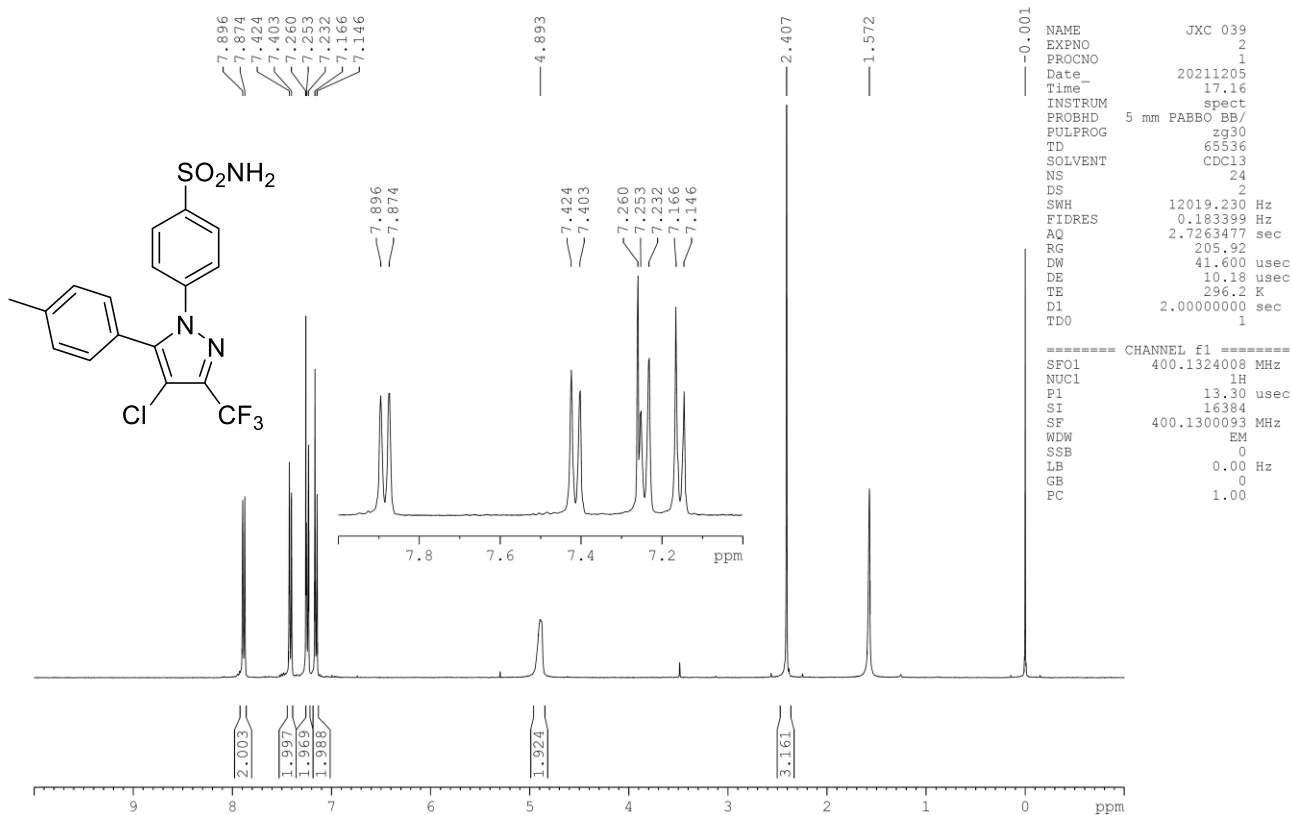


¹³C{¹H}, CDCl₃

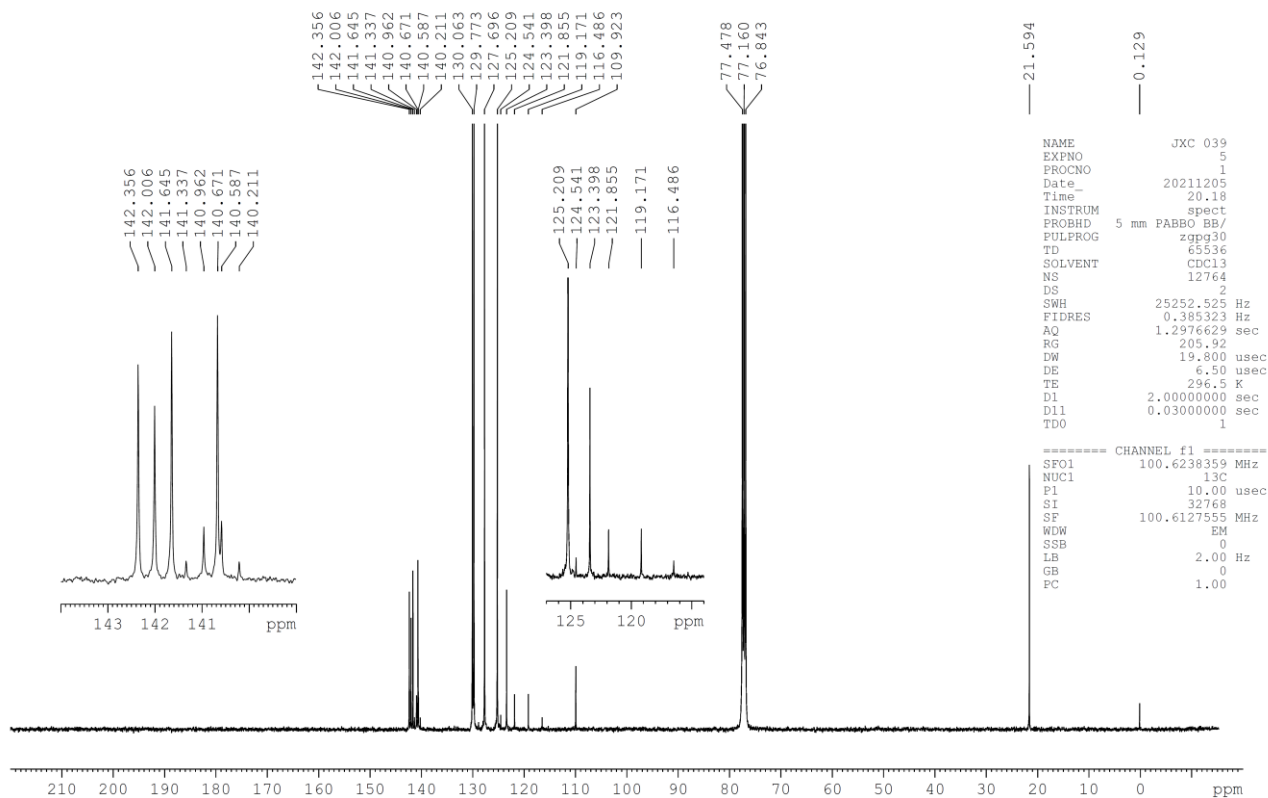


4-chloro-celecoxib **1n**

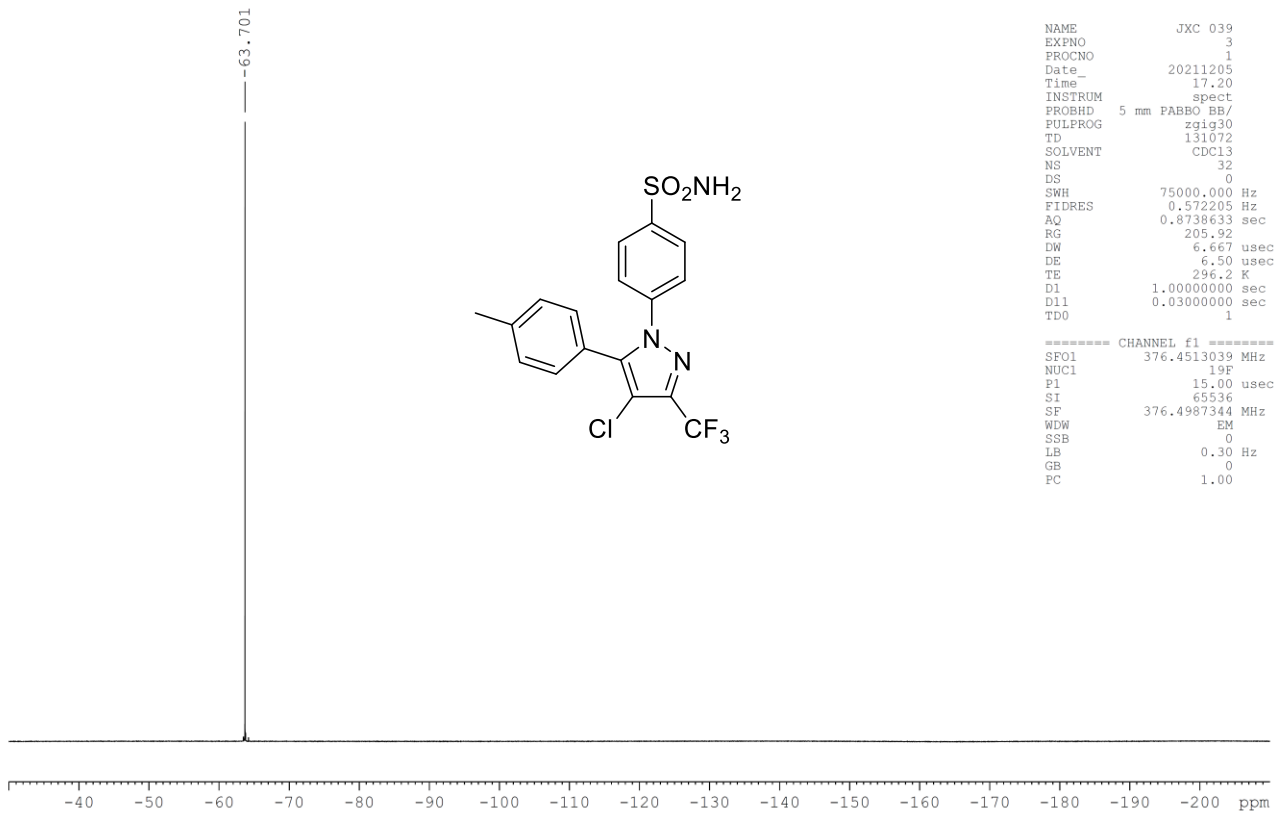
^1H , CDCl_3



$^{13}\text{C}\{^1\text{H}\}$, CDCl_3



$^{19}\text{F}\{^1\text{H}\}$, CDCl_3



HRMS Spectra

No.	HRMS Spectra	Page
1	4-Chloro-3,5-dimethylpyrazole 1a	S39
2	4-Chloropyrazole 1b	S40
3	4-Chloro-3,5-diphenylpyrazole 1c	S41
4	4-Chloro-3,5-dimethyl-1H-pyrazole-1-carboxamide 1d	S42
5	4-Chloro-1-(tetrahydro-2H-pyran-2-yl)-1H-pyrazole 1e	S43
6	4-Chloro-3,5-dimethyl-1-phenyl-1H-pyrazole 1f	S44
7	Di-chlorinated side product of 3,5-dimethyl-1-phenylpyrazole 1f-2	S45
8	Tri-chlorinated side product of 3,5-dimethyl-1-phenylpyrazole 1f-3	S46
9	4-Chloro-1-phenylpyrazole 1g	S47
10	4-Chloro-1-(4-methoxyphenyl)-1H-pyrazole 1h	S48
11	1-(4-Bromophenyl)-4-chloro-1H-pyrazole 1i	S49
12	4-(4-chloro-1H-pyrazol-1-yl)benzaldehyde 1j	S50
13	4-(4-chloro-1H-pyrazol-1-yl)benzotrile 1k	S51
14	4-chloro-1-(4-nitrophenyl)-1H-pyrazole 1l	S52
15	2-(4-chloro-1H-pyrazol-1-yl)pyridine 1m	S53
16	4-chloro-celecoxib 1n	S54

4-Chloro-3,5-dimethylpyrazole **1a**

Display Report

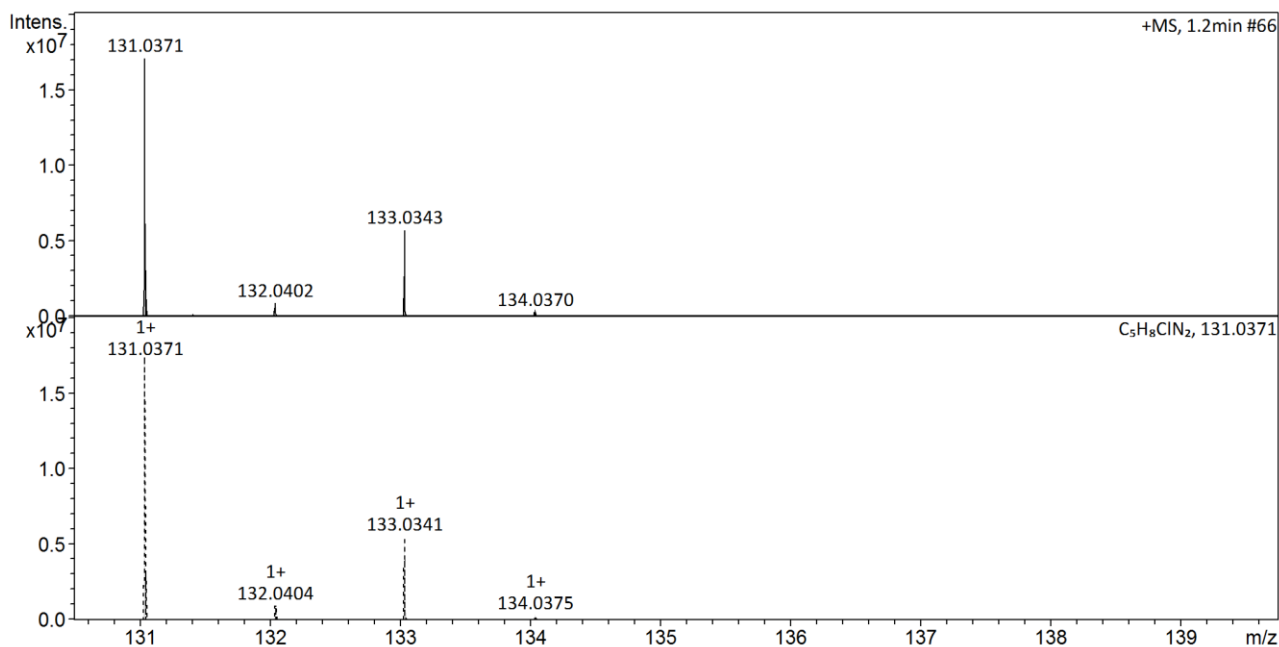
Analysis Info

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Method	Small molecule.m	Operator	NCTU	
Sample Name	CMC046	Instrument	impact HD	1819696.00164
Comment				

Acquisition Parameter

Source Type	ESI	Ion Polarity	Positive	Set Nebulizer	1.0 Bar
Focus	Active	Set Capillary	4500 V	Set Dry Heater	200 °C
Scan Begin	50 m/z	Set End Plate Offset	-500 V	Set Dry Gas	6.0 l/min
Scan End	1500 m/z	Set Charging Voltage	2000 V	Set Divert Valve	Waste
		Set Corona	0 nA	Set APCI Heater	0 °C

Meas. m/z	#	Ion Formula	m/z	err [ppm]	mSigma	# Sigma	Score	rdb	e ⁻ Conf	N-Rule	Adduct
131.0371	1	C5H8ClN2	131.0371	0.4	9.2	1	100.00	2.5	even	ok	M+H



4-Chloropyrazole 1b

Display Report

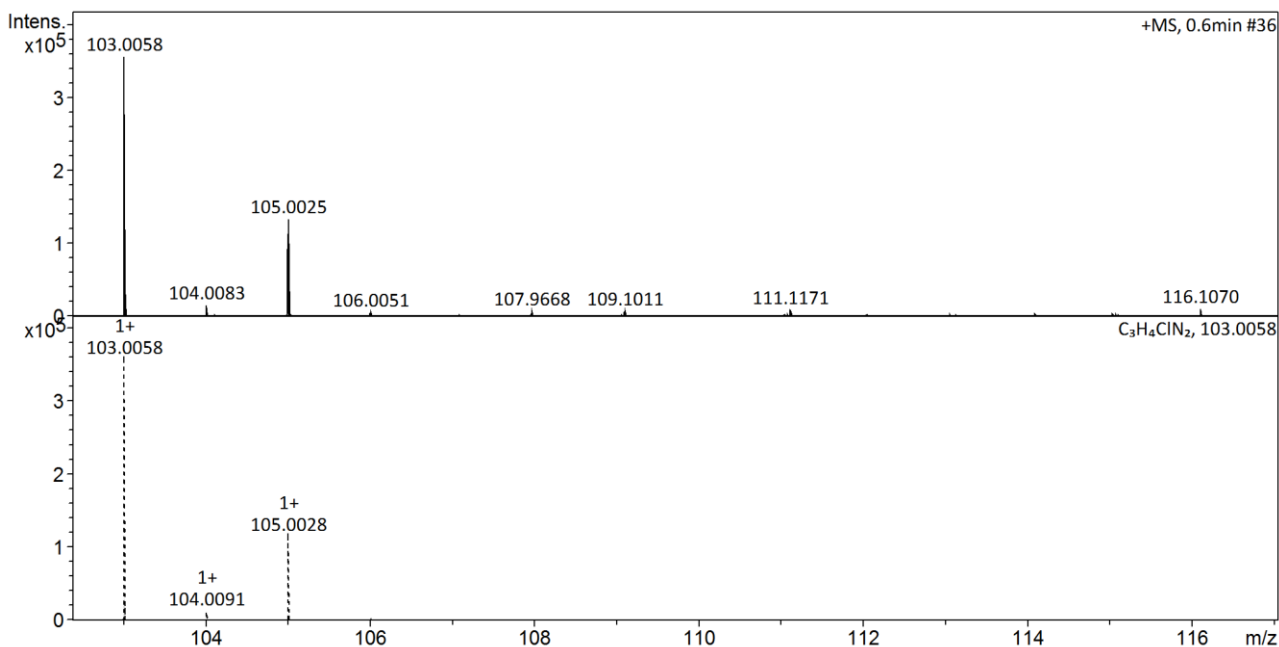
Analysis Info

Analysis Name	D:\Data\nctu service\data\2022\20220304\JXC013_GC3_01_28948.d	Acquisition Date	3/4/2022 3:55:31 PM	
Method	Small molecule.m	Operator	NCTU	
Sample Name	JXC013	Instrument	impact HD	1819696.00164
Comment				

Acquisition Parameter

Source Type	ESI	Ion Polarity	Positive	Set Nebulizer	1.0 Bar
Focus	Active	Set Capillary	4500 V	Set Dry Heater	200 °C
Scan Begin	50 m/z	Set End Plate Offset	-500 V	Set Dry Gas	6.0 l/min
Scan End	1500 m/z	Set Charging Voltage	2000 V	Set Divert Valve	Waste
		Set Corona	0 nA	Set APCI Heater	0 °C

Meas. m/z	#	Ion Formula	m/z	err [ppm]	mSigma	# Sigma	Score	rdb	e ⁻ Conf	N-Rule	Adduct
103.0058	1	C ₃ H ₄ ClN ₂	103.0058	-0.5	27.2	1	100.00	2.5	even	ok	M+H



4-Chloro-3,5-diphenylpyrazole 1c

Display Report

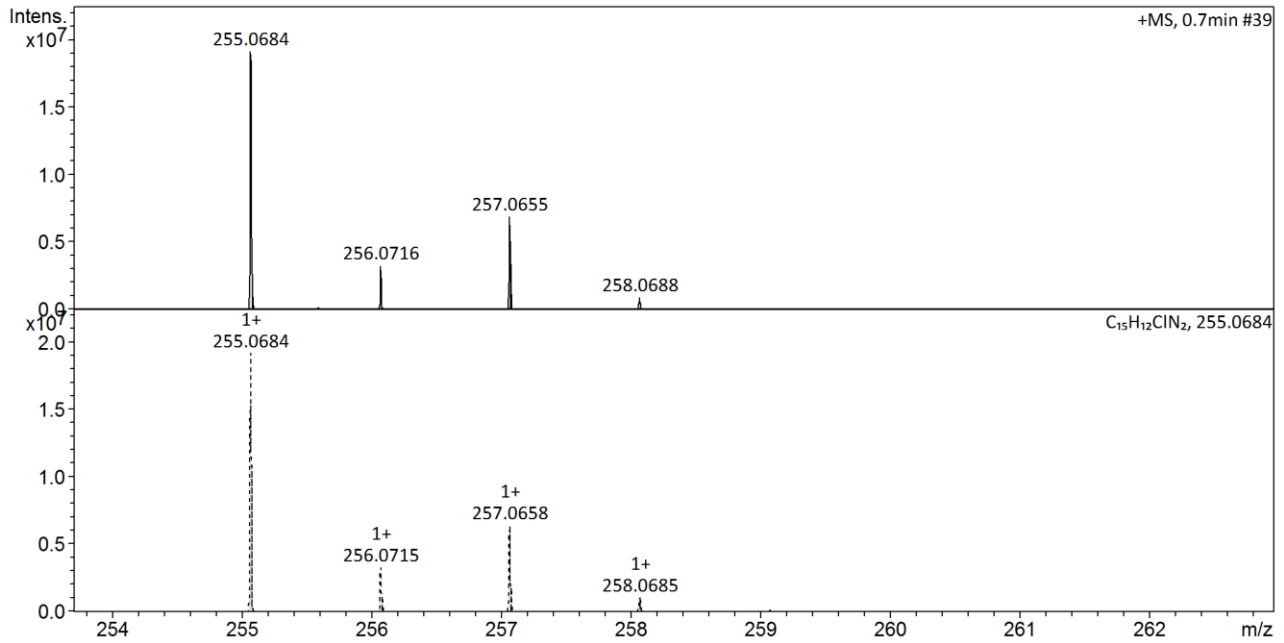
Analysis Info

Analysis Name	D:\Data\nctu service\data\2022\20220304\CMC053_GA5_01_28934.d	Acquisition Date	3/4/2022 2:55:15 PM	
Method	Small molecule.m	Operator	NCTU	
Sample Name	CMC053	Instrument	impact HD	1819696.00164
Comment				

Acquisition Parameter

Source Type	ESI	Ion Polarity	Positive	Set Nebulizer	1.0 Bar
Focus	Active	Set Capillary	4500 V	Set Dry Heater	200 °C
Scan Begin	50 m/z	Set End Plate Offset	-500 V	Set Dry Gas	6.0 l/min
Scan End	1500 m/z	Set Charging Voltage	2000 V	Set Divert Valve	Waste
		Set Corona	0 nA	Set APCI Heater	0 °C

Meas. m/z	#	Ion Formula	m/z	err [ppm]	mSigma	# Sigma	Score	rdb	e ⁻ Conf	N-Rule	Adduct
255.0684	1	C ₁₅ H ₁₂ ClN ₂	255.0684	0.3	12.1	1	100.00	10.5	even	ok	M+H



4-Chloro-3,5-dimethyl-1H-pyrazole-1-carboxamide **1d**

Display Report

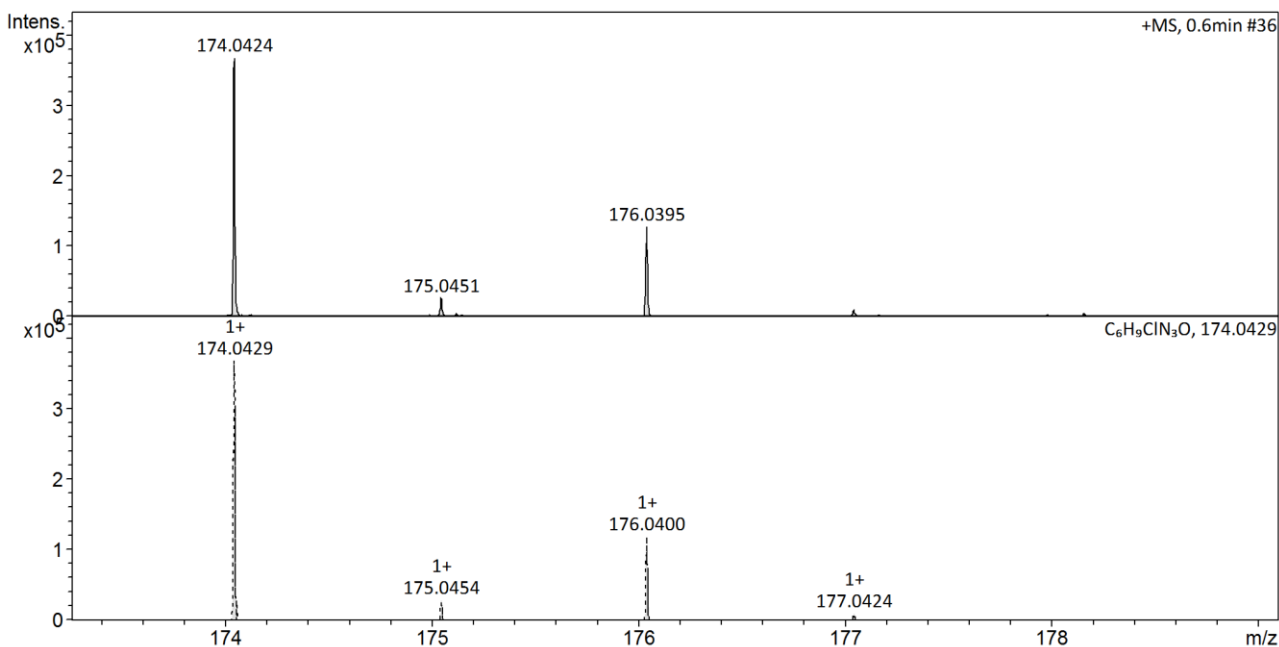
Analysis Info

Analysis Name	D:\Data\nctu service\data\2022\20221127\CMC056_GB3_01_33255.d	Acquisition Date	11/27/2022 3:37:14 PM	
Method	Small molecule.m	Operator	NCTU	
Sample Name	CMC056	Instrument	impact HD	1819696.00164
Comment				

Acquisition Parameter

Source Type	ESI	Ion Polarity	Positive	Set Nebulizer	1.0 Bar
Focus	Active	Set Capillary	4500 V	Set Dry Heater	200 °C
Scan Begin	50 m/z	Set End Plate Offset	-500 V	Set Dry Gas	6.0 l/min
Scan End	1500 m/z	Set Charging Voltage	2000 V	Set Divert Valve	Waste
		Set Corona	0 nA	Set APCI Heater	0 °C

Meas. m/z	#	Ion Formula	m/z	err [ppm]	mSigma	# Sigma	Score	rdb	e ⁻ Conf	N-Rule	Adduct
174.0424	1	C6H9ClN3O	174.0429	-2.5	17.0	1	100.00	3.5	even	ok	M+H



4-Chloro-1-(tetrahydro-2H-pyran-2-yl)-1H-pyrazole 1e

Display Report

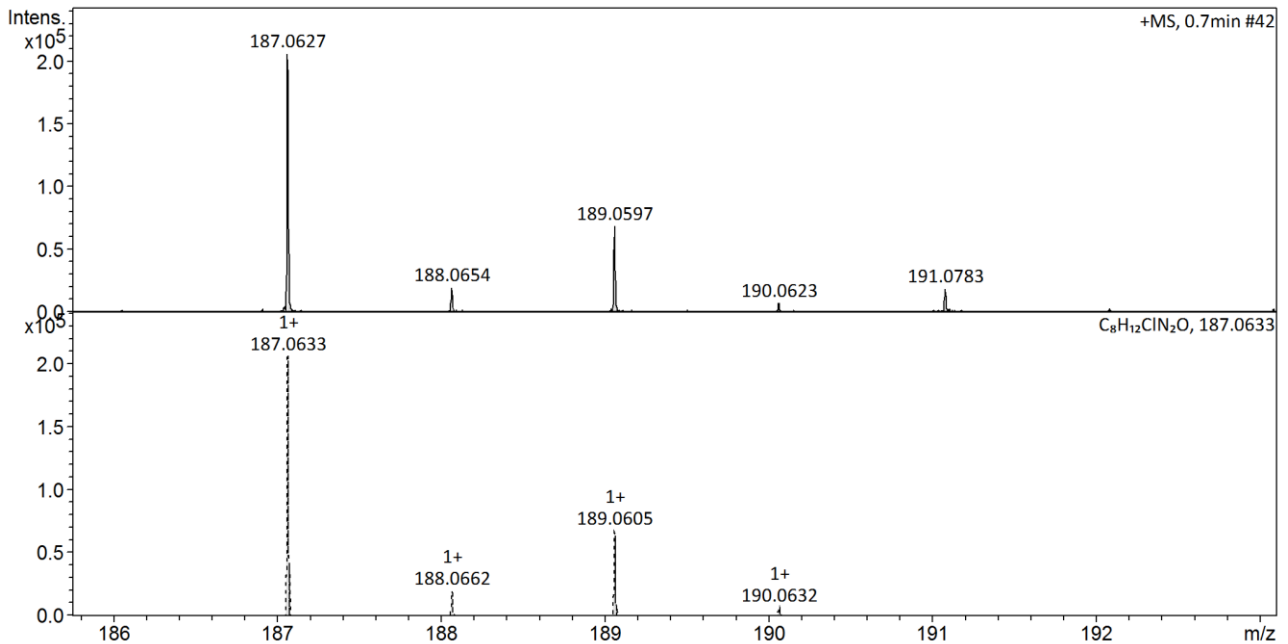
Analysis Info

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Method	Small molecule.m	Operator	NCTU	
Sample Name	CMC091	Instrument	impact HD	1819696.00164
Comment				

Acquisition Parameter

Source Type	ESI	Ion Polarity	Positive	Set Nebulizer	1.0 Bar
Focus	Active	Set Capillary	4500 V	Set Dry Heater	200 °C
Scan Begin	50 m/z	Set End Plate Offset	-500 V	Set Dry Gas	6.0 l/min
Scan End	1500 m/z	Set Charging Voltage	2000 V	Set Divert Valve	Waste
		Set Corona	0 nA	Set APCI Heater	0 °C

Meas. m/z	#	Ion Formula	m/z	err [ppm]	mSigma	# Sigma	Score	rdb	e ⁻ Conf	N-Rule	Adduct
187.0627	1	C ₈ H ₁₂ ClN ₂ O	187.0633	-3.0	4.4	1	100.00	3.5	even	ok	M+H



4-Chloro-3,5-dimethyl-1-phenylpyrazole 1f

Display Report

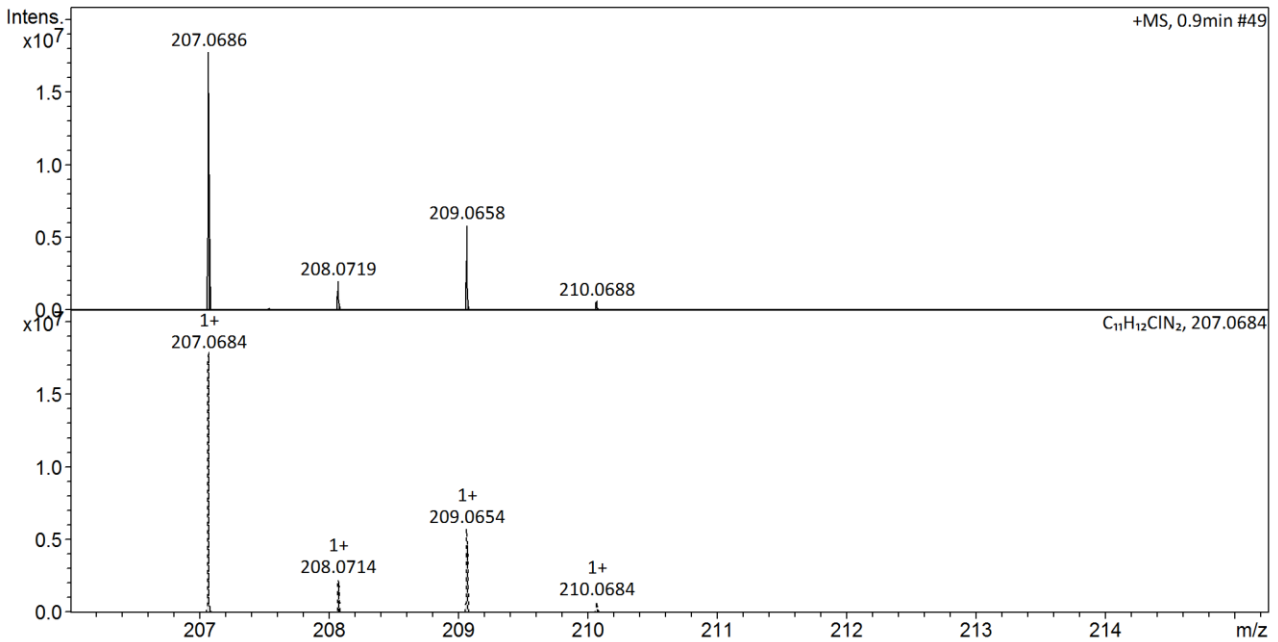
Analysis Info

Analysis Name	D:\Data\nctu service\data\2022\20220304\CMC065_GB7_01_28944.d	Acquisition Date	3/4/2022 3:38:18 PM	
Method	Small molecule.m	Operator	NCTU	
Sample Name	CMC065	Instrument	impact HD	1819696.00164
Comment				

Acquisition Parameter

Source Type	ESI	Ion Polarity	Positive	Set Nebulizer	1.0 Bar
Focus	Active	Set Capillary	4500 V	Set Dry Heater	200 °C
Scan Begin	50 m/z	Set End Plate Offset	-500 V	Set Dry Gas	6.0 l/min
Scan End	1500 m/z	Set Charging Voltage	2000 V	Set Divert Valve	Waste
		Set Corona	0 nA	Set APCI Heater	0 °C

Meas. m/z	#	Ion Formula	m/z	err [ppm]	mSigma	# Sigma	Score	rdb	e ⁻ Conf	N-Rule	Adduct
207.0686	1	C ₁₁ H ₁₂ ClN ₂	207.0684	-1.3	8.8	1	100.00	6.5	even	ok	M+H



Di-chlorinated side product of 3,5-dimethyl-1-phenylpyrazole **1f-2**

Display Report

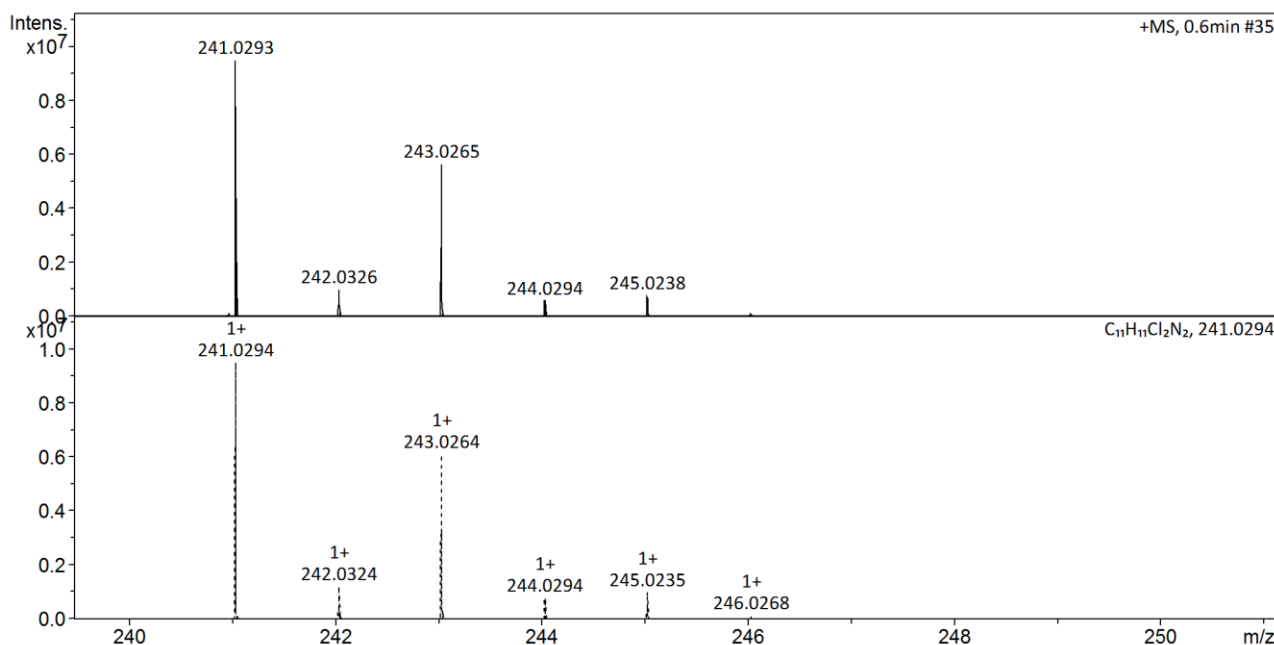
Analysis Info

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Method	Small molecule.m	Operator	NCTU
Sample Name	CMC065 F4	Instrument	impact HD 1819696.00164
Comment			

Acquisition Parameter

Source Type	ESI	Ion Polarity	Positive	Set Nebulizer	1.0 Bar
Focus	Active	Set Capillary	4500 V	Set Dry Heater	200 °C
Scan Begin	50 m/z	Set End Plate Offset	-500 V	Set Dry Gas	6.0 l/min
Scan End	1500 m/z	Set Charging Voltage	2000 V	Set Divert Valve	Waste
		Set Corona	0 nA	Set APCI Heater	0 °C

Meas. m/z	#	Ion Formula	m/z	err [ppm]	mSigma	# Sigma	Score	rdb	e ⁻ Conf	N-Rule	Adduct
241.0293	1	C ₁₁ H ₁₁ Cl ₂ N ₂	241.0294	-0.3	26.6	1	100.00	6.5	even	ok	M+H



Tri-chlorinated side product of 3,5-dimethyl-1-phenylpyrazole **1f-3**

Display Report

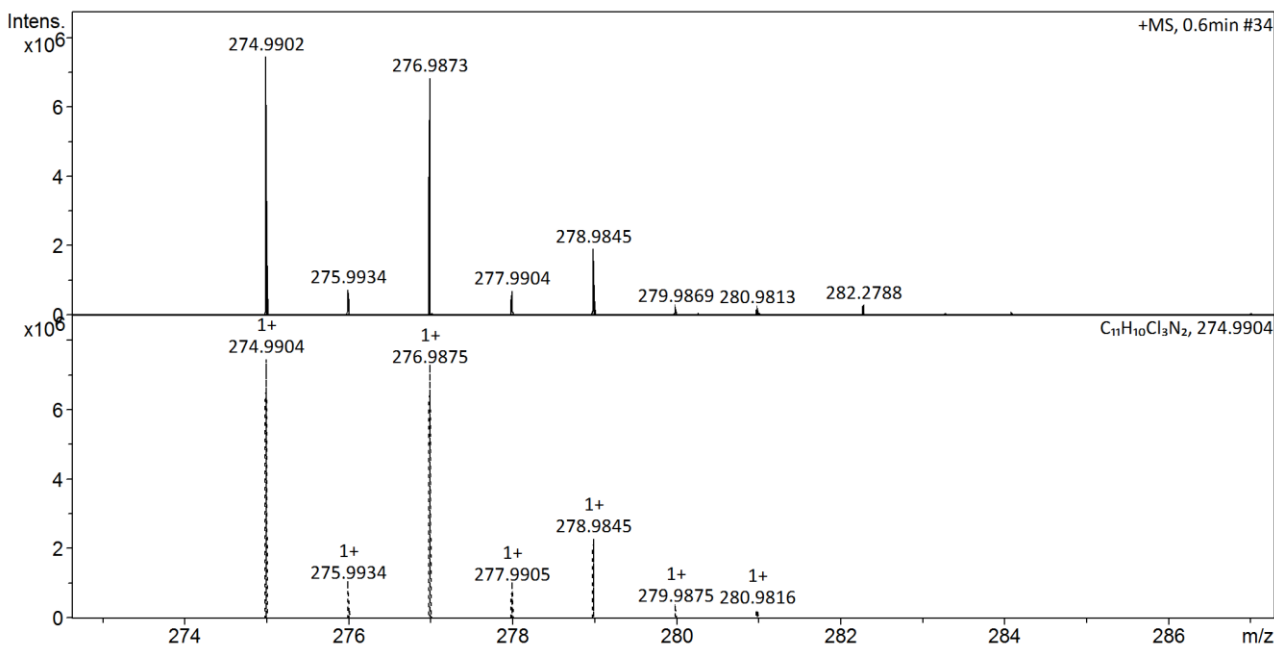
Analysis Info

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Method	Small molecule.m	Operator	NCTU
Sample Name	CMC065 F2	Instrument	impact HD 1819696.00164
Comment			

Acquisition Parameter

Source Type	ESI	Ion Polarity	Positive	Set Nebulizer	1.0 Bar
Focus	Active	Set Capillary	4500 V	Set Dry Heater	200 °C
Scan Begin	50 m/z	Set End Plate Offset	-500 V	Set Dry Gas	6.0 l/min
Scan End	1500 m/z	Set Charging Voltage	2000 V	Set Divert Valve	Waste
		Set Corona	0 nA	Set APCI Heater	0 °C

Meas. m/z	#	Ion Formula	m/z	err [ppm]	mSigma	# Sigma	Score	rdb	e ⁻ Conf	N-Rule	Adduct
274.9902	1	C ₁₁ H ₁₀ Cl ₃ N ₂	274.9904	-0.7	30.4	1	100.00	6.5	even	ok	M+H



4-Chloro-1-phenylpyrazole 1g

Display Report

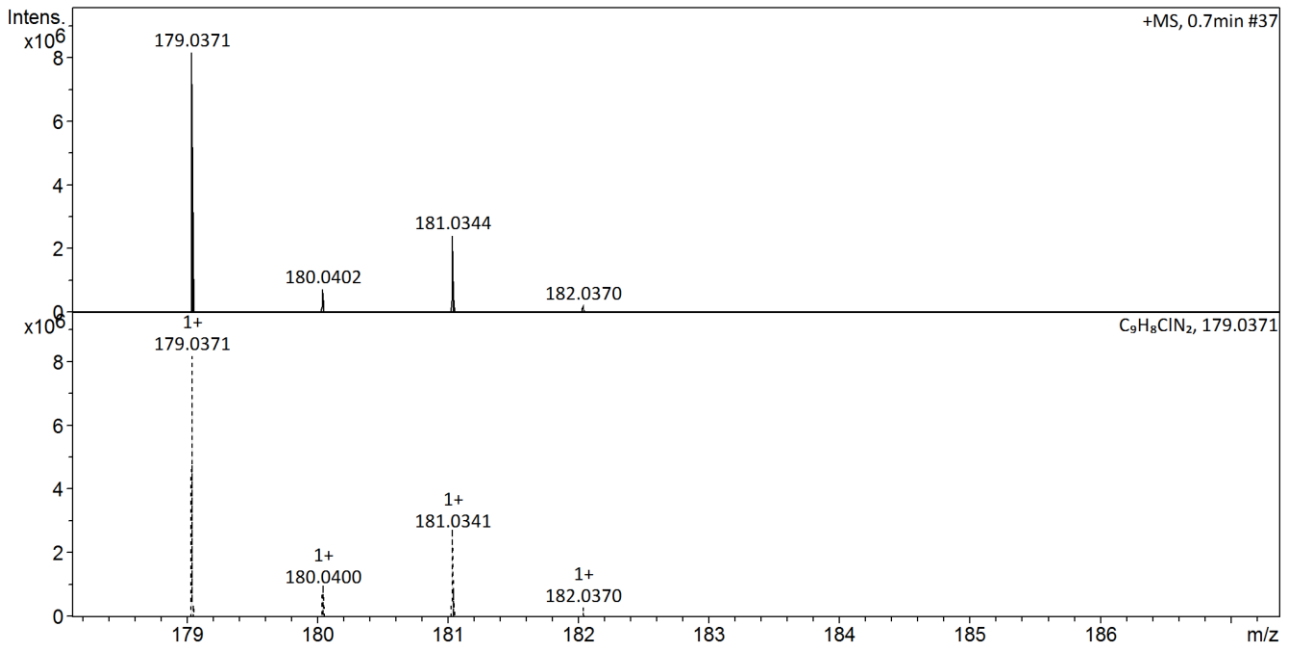
Analysis Info

Analysis Name D:\Data\nctu service\data\2022\20220304\JXC029_GA7_01_28936.d
Method Small molecule.m
Sample Name JXC029
Comment
Acquisition Date 3/4/2022 3:03:50 PM
Operator NCTU
Instrument impact HD 1819696.00164

Acquisition Parameter

Source Type ESI Ion Polarity Positive Set Nebulizer 1.0 Bar
Focus Active Set Capillary 4500 V Set Dry Heater 200 °C
Scan Begin 50 m/z Set End Plate Offset -500 V Set Dry Gas 6.0 l/min
Scan End 1500 m/z Set Charging Voltage 2000 V Set Divert Valve Waste
Set Corona 0 nA Set APCI Heater 0 °C

Meas. m/z	#	Ion Formula	m/z	err [ppm]	mSigma	# Sigma	Score	rdb	e ⁻ Conf	N-Rule	Adduct
179.0371	1	C ₉ H ₈ ClN ₂	179.0371	-0.4	17.4	1	100.00	6.5	even	ok	M+H



4-Chloro-1-(4-methoxyphenyl)pyrazole 1h

Display Report

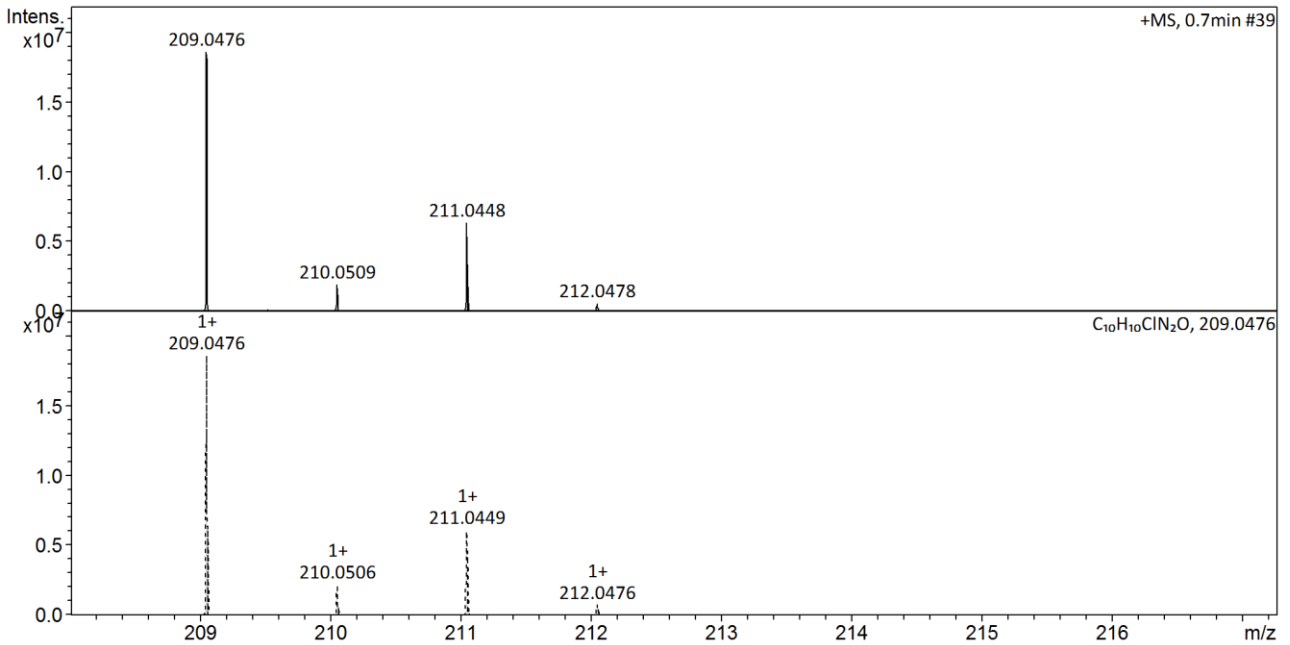
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Method	Small molecule.m	Operator	NCTU	
Sample Name	CMC075	Instrument	impact HD	1819696.00164
Comment				

Acquisition Parameter

Source Type	ESI	Ion Polarity	Positive	Set Nebulizer	1.0 Bar
Focus	Active	Set Capillary	4500 V	Set Dry Heater	200 °C
Scan Begin	50 m/z	Set End Plate Offset	-500 V	Set Dry Gas	6.0 l/min
Scan End	1500 m/z	Set Charging Voltage	2000 V	Set Divert Valve	Waste
		Set Corona	0 nA	Set APCI Heater	0 °C

Meas. m/z	#	Ion Formula	m/z	err [ppm]	mSigma	# Sigma	Score	rdb	e ⁻ Conf	N-Rule	Adduct
209.0476	1	C ₁₀ H ₁₀ ClN ₂ O	209.0476	-0.1	9.9	1	100.00	6.5	even	ok	M+H



4-Chloro-1-(4-bromophenyl)pyrazole **1i**

Display Report

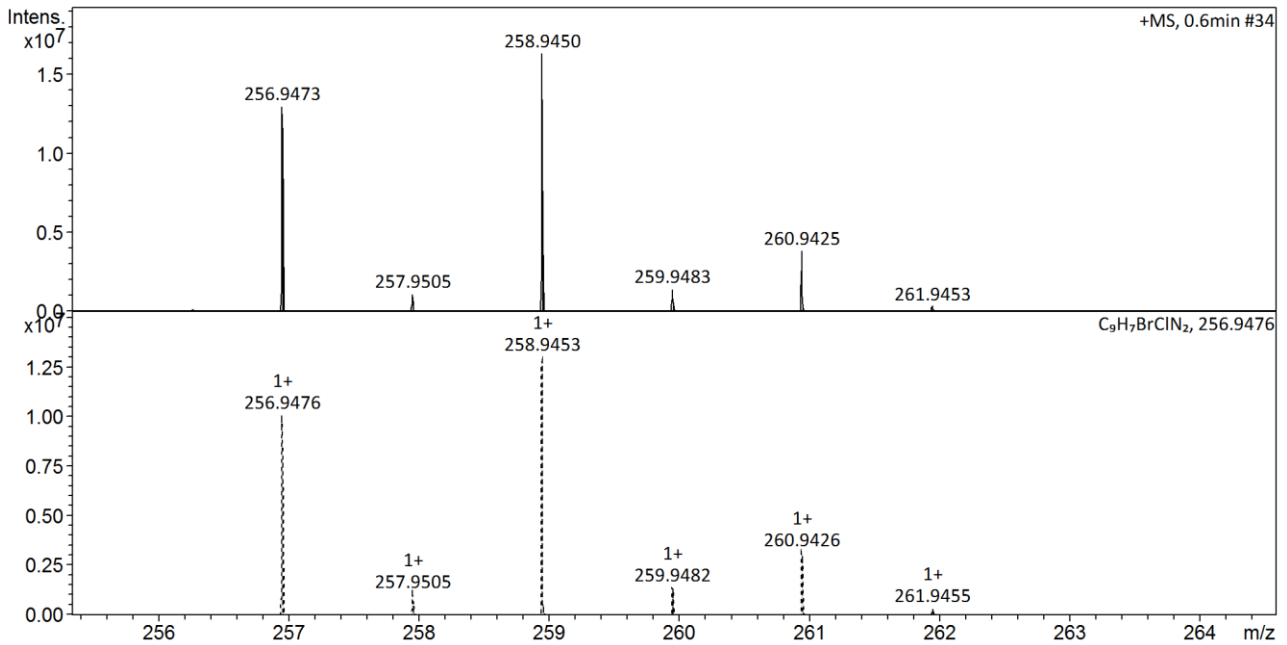
Analysis Info

Analysis Name	D:\Data\nctu service\data\2022\20221127\JXC023_GB4_01_33256.d	Acquisition Date	11/27/2022 3:41:32 PM
Method	Small molecule.m	Operator	NCTU
Sample Name	JXC023	Instrument	impact HD 1819696.00164
Comment			

Acquisition Parameter

Source Type	ESI	Ion Polarity	Positive	Set Nebulizer	1.0 Bar
Focus	Active	Set Capillary	4500 V	Set Dry Heater	200 °C
Scan Begin	50 m/z	Set End Plate Offset	-500 V	Set Dry Gas	6.0 l/min
Scan End	1500 m/z	Set Charging Voltage	2000 V	Set Divert Valve	Waste
		Set Corona	0 nA	Set APCI Heater	0 °C

Meas. m/z	#	Ion Formula	m/z	err [ppm]	mSigma	# Sigma	Score	rdb	e ⁻ Conf	N-Rule	Adduct
256.9473	1	C9H7BrClN2	256.9476	1.2	14.8	1	100.00	6.5	even	ok	M+H



4-(4-chloro-1H-pyrazol-1-yl)benzaldehyde 1j

Display Report

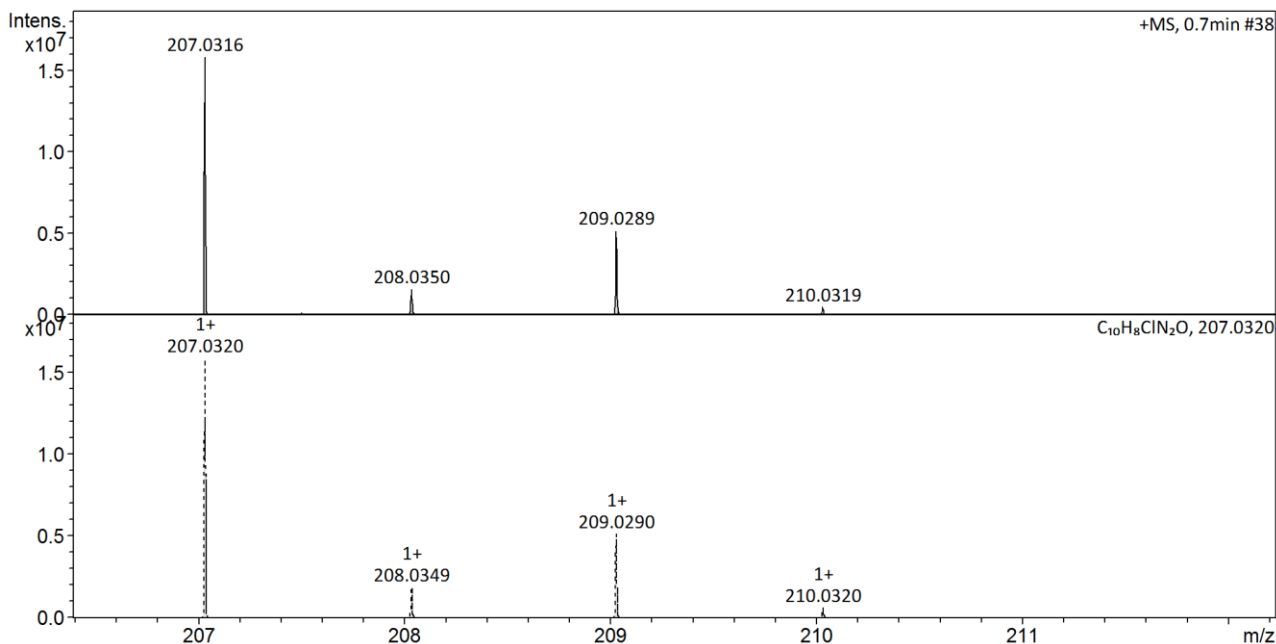
Analysis Info

Analysis Name	D:\Data\nctu service\data\2022\20221127\CMC076_GB7_01_33259.d	Acquisition Date	11/27/2022 3:54:38 PM
Method	Small molecule.m	Operator	NCTU
Sample Name	CMC076	Instrument	impact HD 1819696.00164
Comment			

Acquisition Parameter

Source Type	ESI	Ion Polarity	Positive	Set Nebulizer	1.0 Bar
Focus	Active	Set Capillary	4500 V	Set Dry Heater	200 °C
Scan Begin	50 m/z	Set End Plate Offset	-500 V	Set Dry Gas	6.0 l/min
Scan End	1500 m/z	Set Charging Voltage	2000 V	Set Divert Valve	Waste
		Set Corona	0 nA	Set APCI Heater	0 °C

Meas. m/z	#	Ion Formula	m/z	err [ppm]	mSigma	# Sigma	Score	rdb	e ⁻ Conf	N-Rule	Adduct
207.0316	1	C ₁₀ H ₈ ClN ₂ O	207.0320	-1.8	10.2	1	100.00	7.5	even	ok	M+H



4-(4-chloro-1H-pyrazol-1-yl)benzotrile 1k

Display Report

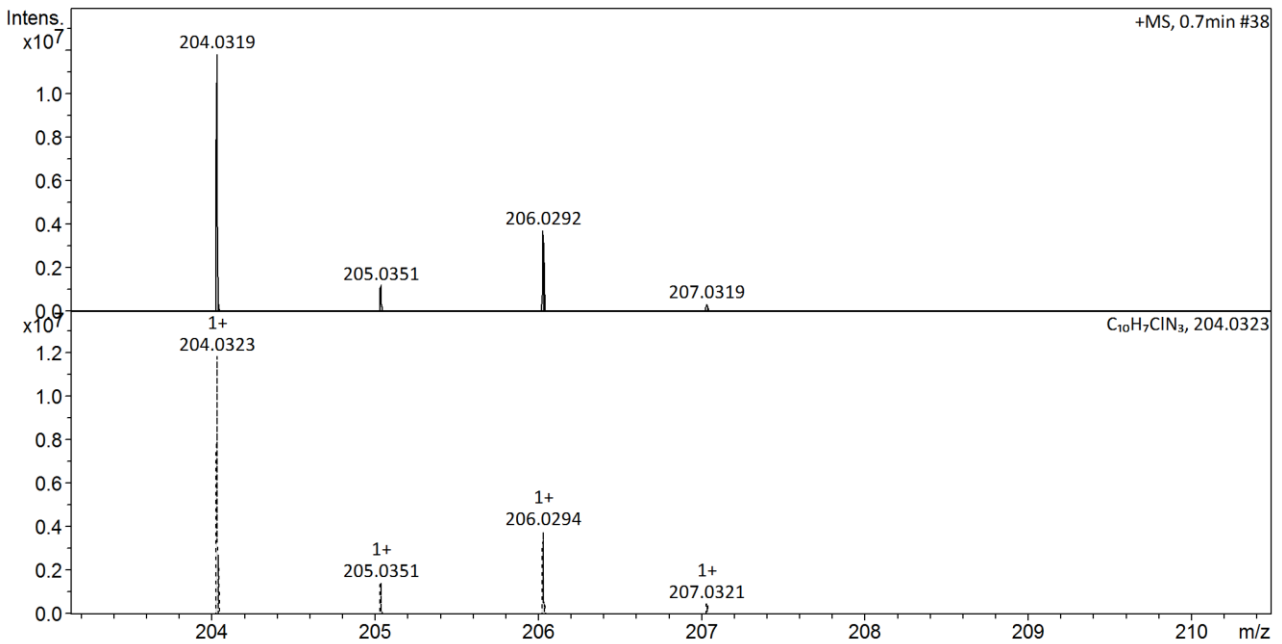
Analysis Info

Analysis Name	D:\Data\nctu service\data\2022\20221127\JXC055_GB1_01_33253.d	Acquisition Date	11/27/2022 3:28:28 PM	
Method	Small molecule.m	Operator	NCTU	
Sample Name	JXC055	Instrument	impact HD	1819696.00164
Comment				

Acquisition Parameter

Source Type	ESI	Ion Polarity	Positive	Set Nebulizer	1.0 Bar
Focus	Active	Set Capillary	4500 V	Set Dry Heater	200 °C
Scan Begin	50 m/z	Set End Plate Offset	-500 V	Set Dry Gas	6.0 l/min
Scan End	1500 m/z	Set Charging Voltage	2000 V	Set Divert Valve	Waste
		Set Corona	0 nA	Set APCI Heater	0 °C

Meas. m/z	#	Ion Formula	m/z	err [ppm]	mSigma	# Sigma	Score	rdb	e ⁻ Conf	N-Rule	Adduct
204.0319	1	C ₁₀ H ₇ ClN ₃	204.0323	2.1	13.0	1	100.00	8.5	even	ok	M+H



4-chloro-1-(4-nitrophenyl)-1H-pyrazole **11**

Display Report

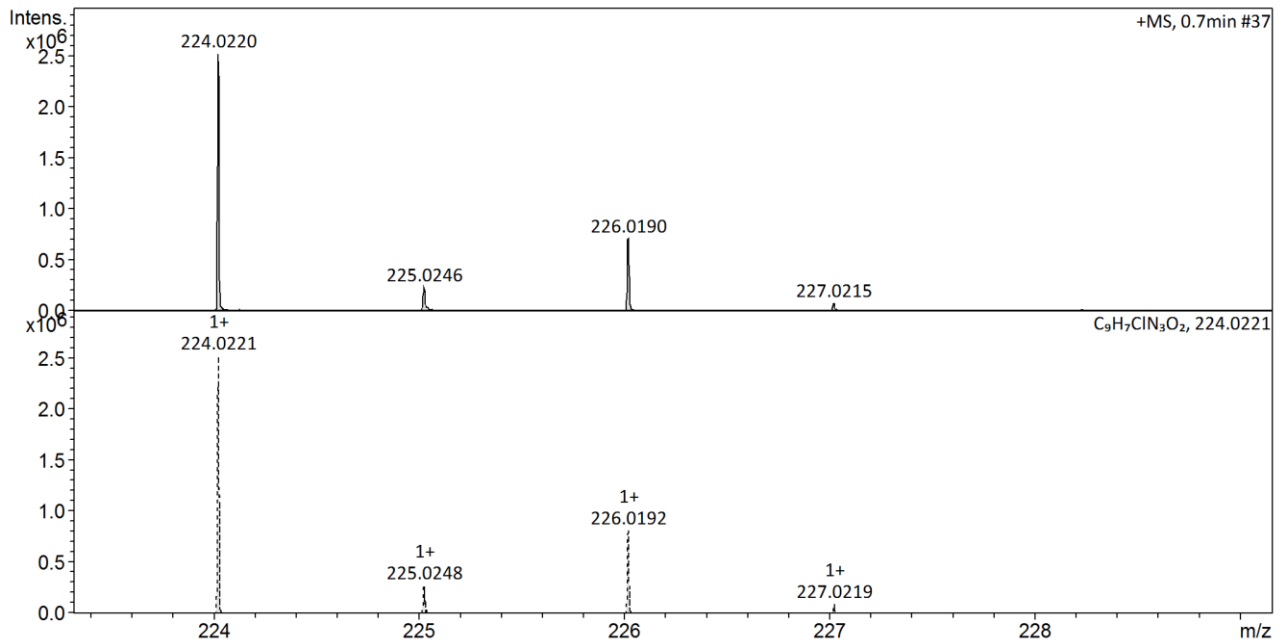
Analysis Info

Analysis Name	D:\Data\nctu service\data\2022\20221127\JXC007_GB2_01_33254.d	Acquisition Date	11/27/2022 3:32:51 PM	
Method	Small molecule.m	Operator	NCTU	
Sample Name	JXC007	Instrument	impact HD	1819696.00164
Comment				

Acquisition Parameter

Source Type	ESI	Ion Polarity	Positive	Set Nebulizer	1.0 Bar
Focus	Active	Set Capillary	4500 V	Set Dry Heater	200 °C
Scan Begin	50 m/z	Set End Plate Offset	-500 V	Set Dry Gas	6.0 l/min
Scan End	1500 m/z	Set Charging Voltage	2000 V	Set Divert Valve	Waste
		Set Corona	0 nA	Set APCI Heater	0 °C

Meas. m/z	#	Ion Formula	m/z	err [ppm]	mSigma	# Sigma	Score	rdb	e ⁻ Conf	N-Rule	Adduct
224.0220	1	C9H7ClN3O2	224.0221	-0.7	25.7	1	100.00	7.5	even	ok	M+H



2-(4-chloro-1H-pyrazol-1-yl)pyridine 1m

Display Report

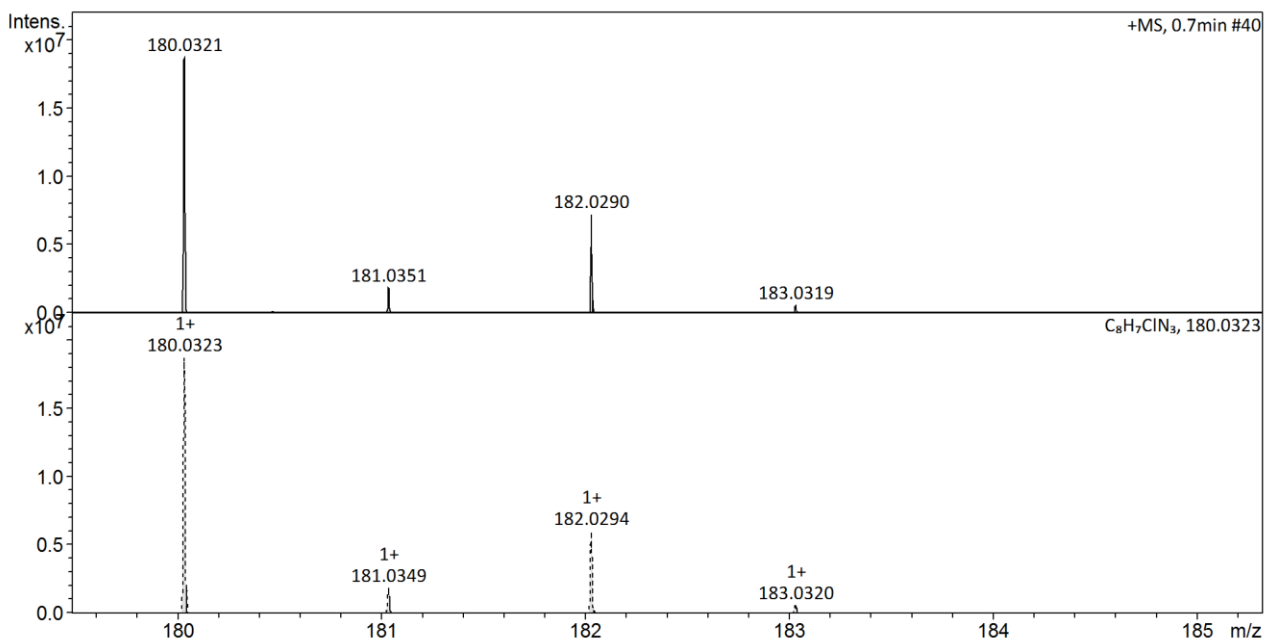
Analysis Info

Analysis Name	D:\Data\nctu service\data\2022\20221127\CMC090_GB5_01_33257.d	Acquisition Date	11/27/2022 3:45:53 PM
Method	Small molecule.m	Operator	NCTU
Sample Name	CMC090	Instrument	impact HD 1819696.00164
Comment			

Acquisition Parameter

Source Type	ESI	Ion Polarity	Positive	Set Nebulizer	1.0 Bar
Focus	Active	Set Capillary	4500 V	Set Dry Heater	200 °C
Scan Begin	50 m/z	Set End Plate Offset	-500 V	Set Dry Gas	6.0 l/min
Scan End	1500 m/z	Set Charging Voltage	2000 V	Set Divert Valve	Waste
		Set Corona	0 nA	Set APCI Heater	0 °C

Meas. m/z	#	Ion Formula	m/z	err [ppm]	mSigma	# Sigma	Score	rdb	e ⁻ Conf	N-Rule	Adduct
180.0321	1	C ₈ H ₇ ClN ₃	180.0323	-1.3	31.7	1	100.00	6.5	even	ok	M+H



4-chloro-celecoxib **1n**

Display Report

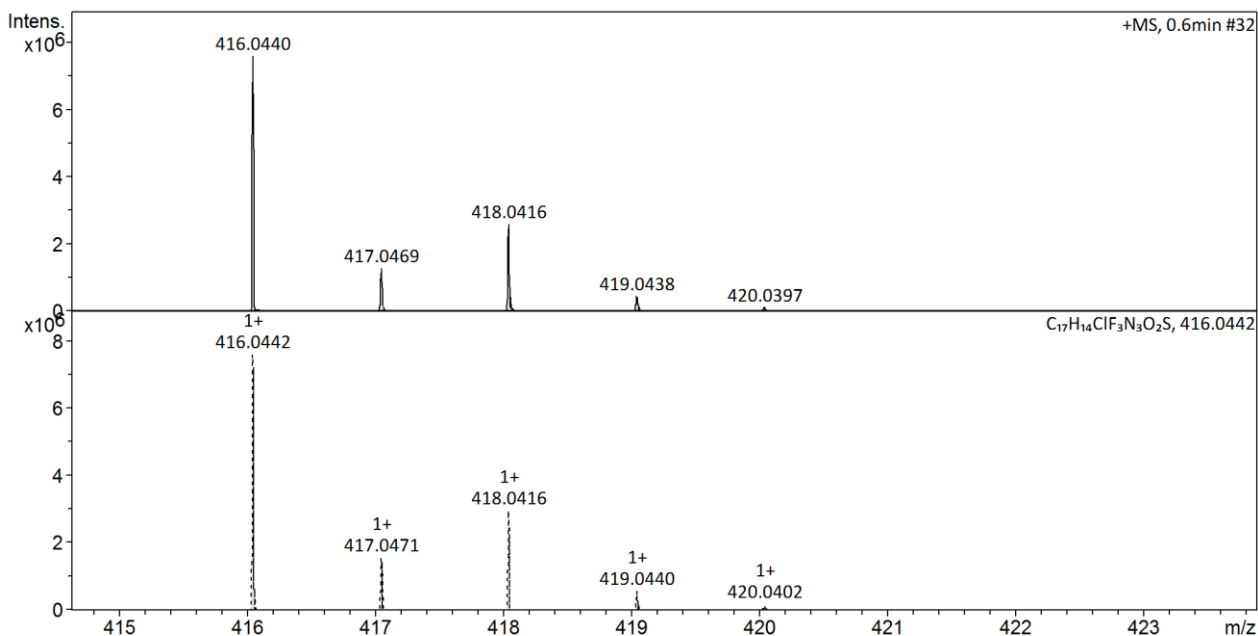
Analysis Info

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Method	Small molecule.m	Operator	NCTU
Sample Name	JXC039	Instrument	impact HD 1819696.00164
Comment			

Acquisition Parameter

Source Type	ESI	Ion Polarity	Positive	Set Nebulizer	1.0 Bar
Focus	Active	Set Capillary	4500 V	Set Dry Heater	200 °C
Scan Begin	50 m/z	Set End Plate Offset	-500 V	Set Dry Gas	6.0 l/min
Scan End	1500 m/z	Set Charging Voltage	2000 V	Set Divert Valve	Waste
		Set Corona	0 nA	Set APCI Heater	0 °C

Meas. m/z	#	Ion Formula	m/z	err [ppm]	mSigma	# Sigma	Score	rdb	e ⁻ Conf	N-Rule	Adduct
416.0440	1	C ₁₇ H ₁₄ ClF ₃ N ₃ O ₂ S	416.0442	-0.5	26.4	1	100.00	10.5	even	ok	M+H



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