Selective oxidative upgrade of waste polystyrene plastics by nitric acid to produce benzoic acid

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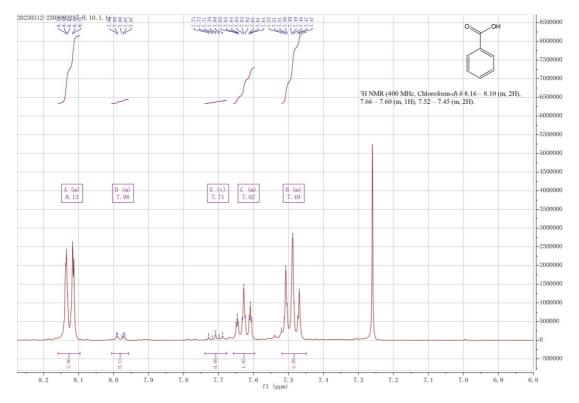


Fig. S1 ¹H-NMR spectra of the product (1g PS, 10ml 20% HNO₃, 180 °C, 3 h).

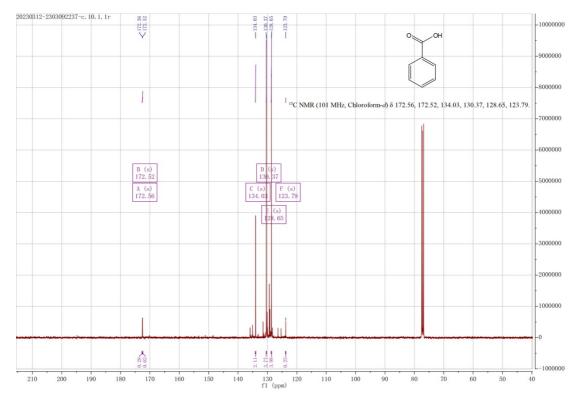


Fig. S2 ¹³C-NMR spectra of the product (1g PS, 10ml 20% HNO₃, 180 °C, 3 h).

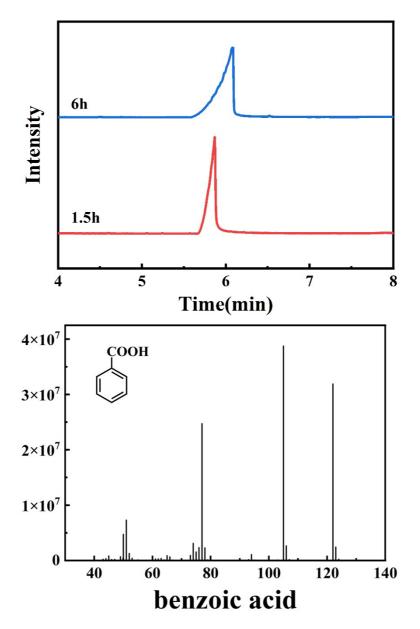


Fig. S3 GC-MS spectra of pure benzoic acid oxidation process (1g benzoic acid, 10ml

20% HNO₃, 180 °C).



Fig. S4 Reddish brown gas generated after the reaction.

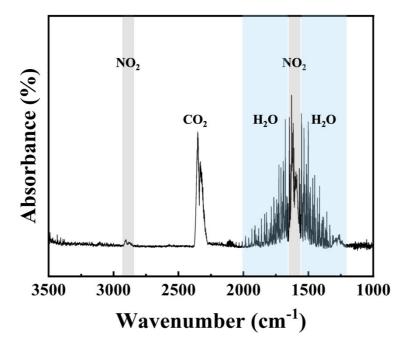


Fig. S5 FTIR spectra of the gas collected after the reaction (CO_2 and H_2O are

impurity peaks generated during the detection process).

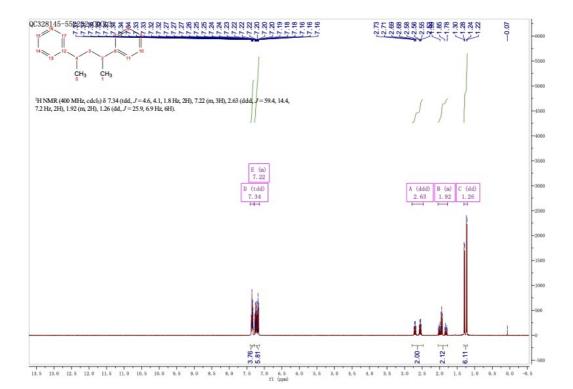
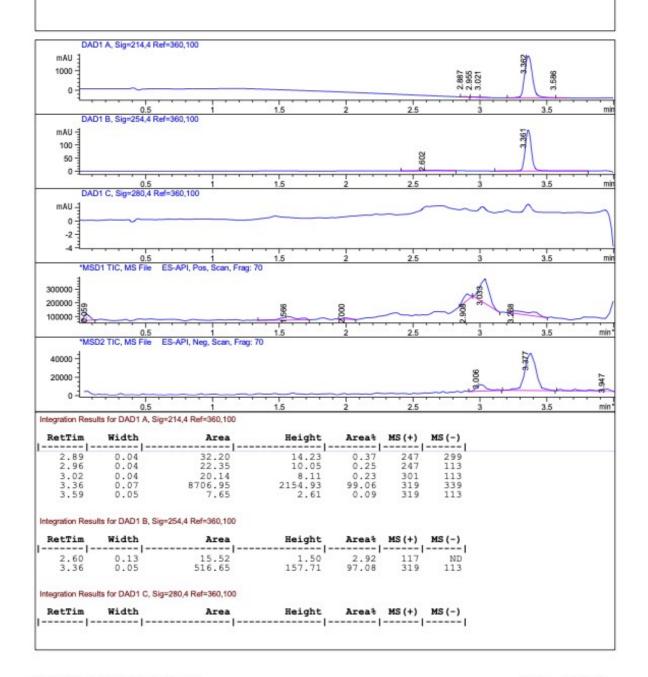


Fig. S6 ¹H-NMR spectra of 2,4-Diphenylpentane.

MS Report from Instrument: LCMS

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Acq. Operator	: SYSTEM	Inj : 1
Spec. Reported	: MS Integration	Inj Volume : 1 ul
Acq. Method	: D:\Chem32\1\Data\MAFANGYA\02	2023-03-03 09-21-10\20171123FA02-M.M
Analysis Method	: D:\Chem32\1\Data\MAFANGYA\02	2023-03-03 09-21-10\20171123FA02-M.M
Sample Info :		
Method Info :		



LCMS 3/3/2023 11:06:57 AM SYSTEM

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Fig. S7 LC-MS report of 2,4-Diphenylpentane.

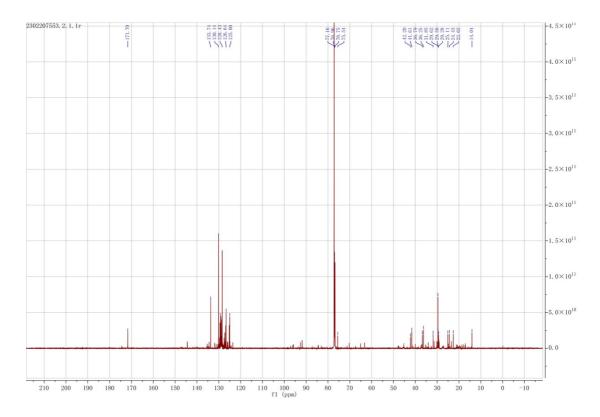


Fig. S8¹³C-NMR spectra of oxidative upgrade of 2,4-Diphenylpentane (0.1g 2,4-

Diphenylpentane, 3ml 20% HNO₃, 180 °C, 10 min).

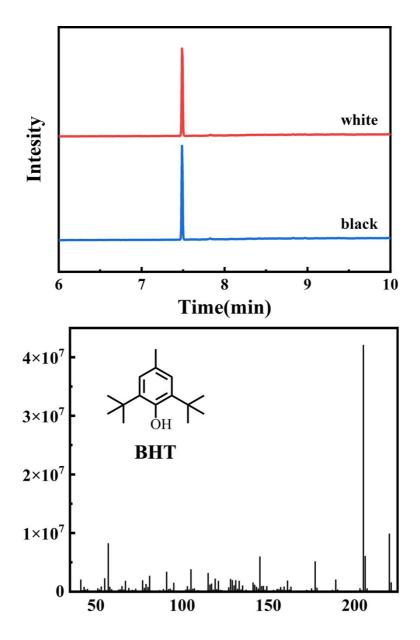


Fig. S9 GC/MS spectra of the solid residue generated after oxidative upgrade of the

cup lids.