

Supporting Information

A Scalable and Eco-friendly Total Synthesis of Poly (ADP-Ribose) Polymerase Inhibitor Olaparib

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1. General Information. Unless otherwise noted, all commercial reagents were used without further purification. All the reactions were performed in round bottom flask, stirred with a magnetic bar under nitrogen atmosphere and monitored by Thin-layer chromatography (0.2 mm silica gel-coated GF 254 plates) and visualized under UV or by staining with dragendorff solution. Flash column chromatography was carried out on Silica Gel 60-120 and 100-200 mesh basified by triethylamine. Nuclear magnetic resonance (NMR) spectra were recorded using Bruker Avance 400 Spectrometer. ¹H and ¹³C chemical shifts are reported in ppm downfield of tetramethylsilane and referenced to residual solvent peak [CHCl₃; δ_H = 7.26 and δ_C = 77.0; DMSO-d₆; δ_H = 2.54 and δ_C = 39.5]. Multiplicities were given as: s (singlet); brs (broad singlet), d (doublet); t (triplet); q (quartet); dd (doublets of doublet); m (multiplets). High resolution mass spectra were taken with a 3000-mass spectrometer, using Waters Agilent 6520-Q-ToFMS/MS system and JEOL-Accu TOF JMST100LC. Melting points are uncorrected and were determined in capillary tubes on SMP 10 melting point apparatus.

2. Synthetic details and experimental section.

Cyclopropyl (piperazin-1-yl) methanone (15): Acetic acid (10 ml) was treated with piperazine (2.5g, 29.02 mmol) portion wise over 15 minutes with stirring under nitrogen. The reaction mixture was warmed to 40 °C and maintained at this temperature until a complete solution was obtained. Cyclopropane carbonyl chloride (2.89 ml, 31.92 mmol) was added over 15 minutes. The reaction mixture was stirred at room temperature for 8h. The reaction mixture was filtered and the filtrate distilled under reduced pressure until acetic acid was fully evaporated. Ethyl acetate was charged to the reaction mixture and filtrate distilled under reduced pressure. A further charge of ethyl acetate was added and reduced pressure distillation continued until white precipitation of (15). White solid (3.9g), Melting point-149 °C, yield-88%, ¹H NMR (400MHz, DMSO-d₆) δ 9.73 (1H, brs), 3.93-3.41 (5H, m), 3.09-3.06 (3H, s), 2.03-1.95 (1H, m), 0.80-0.74 (4H, m); ¹³C{¹H} NMR (100 MHz, DMSO-d₆) δ 171.9, 42.9, 42.2, 10.7, 7.6; **HRMS(ESI):** Calculated for C₈H₁₅N₂O (M+H)⁺-155.1179, found 155.1183.

(4-(5-bromo-2-fluorobenzoyl)piperazin-1-yl)(cyclopropyl)methanone (14): To a stirred mixture of 2-fluoro-5-bromo benzoic acid (5.54 g, 25.29 mmol) in anhydrous DMF (25 ml) was added DIPEA (13.22 ml, 75.87 mmol), HBTU (9.59 g, 25.29 mmol), followed by cyclopropyl (piperazin-1-yl) methanone (3.9 g, 25.29 mmol) at 0 °C under nitrogen atmosphere. After the addition, the mixture was allowed to stir at room temperature for 15h. After distilling DMF, the mixture was diluted with dichloromethane, washed with saturated citric acid, followed by saturated sodium bicarbonate, brine, dried over anhydrous sodium sulfate, and concentrated. White solid of (4-(5-bromo-2-fluorobenzoyl)piperazin-1-yl)(cyclopropyl)methanone was obtained by recrystallization using EtOAc. White solid (6.74 g), Melting point- 93 °C, yield-75%, ¹H NMR (400MHz, CDCl₃) δ 7.48-7.44 (2H, m), 6.97-6.93 (1H, m), 3.71-3.27 (8H, m), 1.68 (1H, brs), 0.96-0.93 (2H, m), 0.74 (2H, brs); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 172.3, 163.9, 157.1 (d, J = 248.3 Hz), 134.5 (d, J = 8.0 Hz), 132.1 (d, J =

3.5 Hz), 125.5 (d, $J = 21.7$ Hz), 117.7 (d, $J = 23.1$ Hz), 117.4 (d, $J = 3.1$ Hz), 47.2, 42.3, 11.0, 7.7; **HRMS(ESI)**: Calculated for $C_{15}H_{17}BrFN_2O_2$ (M+H)⁺- 355.0452, found 355.0462.

2-(2-(3-(4-(cyclopropanecarbonyl) piperazine-1-carbonyl)-4-fluorophenyl) acetyl) benzoic acid (12): In a 50 ml schlenk tube, 2-acetylbenzoic acid (9.34 g, 56.92 mmol), (4-(5-bromo-2-fluorobenzoyl)piperazin-1-yl)(cyclopropyl)methanone (6.74 g, 18.97 mmol) and sodium *tert* butoxide (9.12 g, 94.85 mmol) were taken and then 20 ml of DMF was added to it under argon atmosphere. After that, mixture was heated to 120 °C for 36-40h. After completion of the reaction was monitored by TLC or LCMS, the mixture was extracted with chilled ethyl acetate and the extract dried with Na₂SO₄. Crude NMR was taken and used subsequently without further purification for next step. The product (12) was obtained as Colourless oil (7.07 g), yield- 85%, ¹H NMR (400MHz, CDCl₃) δ 7.98 (1H, s), 7.84-7.83 (1H, m), 7.67-7.52 (2H, m), 7.41- 7.39 (1H, m), 7.33 (1H, s), 6.82-6.80 (1H, m), 3.78-3.74 (4H, m), 3.58 (2H,s), 3.30 (2H, brs), 2.96 (2H, s), 1.88-1.75 (1H, m), 0.99 (2H, s), 0.79 (2H, brs); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 174.1, 172.5, 168.4, 162.8, 156.5 (d, $J = 243.4$ Hz), 148.8, 133.1 (d, $J = 3.3$ Hz), 131.6 (d, $J = 7.8$ Hz), 130.2, 129.6 (d, $J = 18.5$ Hz), 129.3, 128.8, 127.2, 125.2, 122.1, 118.6 (d, $J = 20.4$ Hz), 113.4, 46.5, 45.2, 42.2, 41.8, 36.6, 11.0, 7.7; ¹⁹F NMR (376 MHz, CDCl₃) δ -117.03; **HRMS(ESI)**: Calculated for $C_{24}H_{24}FN_2O_5$ (M+H)⁺- 439.1664, found 439.1662.

4-(3-(4-(cyclopropanecarbonyl) piperazine-1-carbonyl)-4-fluorobenzyl) phthalazin-1(2H)-one (1): Add hydrazine hydrate (0.7 ml, 20.96 mmol) to a solution of 2-(2-(3-(4-(cyclopropanecarbonyl) piperazine-1-carbonyl)-4-fluorophenyl) acetyl) benzoic acid (7.07 g, 16.12 mmol) in EtOH (15 ml). Reflux the reaction mixture for 15h. Cool the mixture and distilled off the ethanol using rotavapour. The collected ethanol was reused. Then the reaction mixture was purified without any solvent extraction. The final product (1) was purified by flash chromatography. HPLC purity: 99.9 %. White solid (6.30 g), Melting point- 208 °C, yield-90%, IR ν (cm⁻¹): 3912.64, 3885.22, 3781.66, 3696.08, 3657.28, 3404.03, 3011.49, 2922.83, 1633.92, 1468.48, 1438.75, 1357.09, 1287.40, 1229.20, 1159.16, 1064.56, 1014.22, 841.68, 774.12, 646.79, 559.42, 485.55; ¹H NMR (400MHz, DMSO-d₆) δ 12.59 (1H, s), 8.28-8.26 (1H, m), 7.98-7.82 (3H, m), 7.46-7.37 (2H, m), 7.27-7.22 (1H, m), 4.34 (2H, s), 3.75-3.60 (6H, m), 3.23-3.17 (2H, m), 1.99 (1H,s), 0.75-0.72 (4H, m); ¹³C{¹H} NMR (100 MHz, DMSO-d₆) δ 171.8, 164.5, 159.9, 156.8 (d, $J = 244.2$ Hz), 145.3, 135.3 (d, $J = 2.5$ Hz), 133.9, 132.2 (d, $J = 7.7$ Hz), 132.0, 129.5, 129.4 (d, $J = 3.5$ Hz), 128.4, 126.5, 125.9, 124.0 (d, $J = 18.2$ Hz), 116.4 (d, $J = 21.6$ Hz), 46.7, 45.4, 42.2, 41.6, 36.9, 10.8, 7.6; ¹⁹F NMR (376 MHz, DMSO-d₆) δ -118.05; **HRMS (ESI)**: Calculated for $C_{24}H_{24}FN_4O_3$ (M+H)⁺- 435.1827, found 435.1833.

(5-bromo-2-fluorophenyl)(4-methoxypiperidin-1-yl)methanone (21): To a stirred mixture of 2-fluoro-5-bromo benzoic acid (200 mg, 0.91 mmol) in anhydrous DMF (5 ml) was added DIPEA (320 mg, 2.74 mmol), HBTU (346 mg, 0.91 mmol), followed by 4-methoxypiperidine (105 mg, 0.91 mmol) at 0 °C under nitrogen atmosphere. After the addition, the mixture was allowed to stir at room

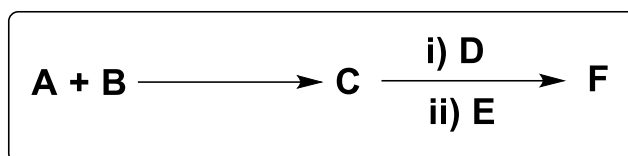
temperature for 15h. After distilling DMF, the mixture was diluted with dichloromethane (50 ml), washed with saturated citric acid (100 ml×2), followed by saturated sodium bicarbonate (100 ml×2), brine (50 ml), dried over anhydrous sodium sulfate, and concentrated. White solid (5-bromo-2-fluorophenyl)(4-methoxypiperidin-1-yl)methanone was obtained by recrystallization using EtOAc. White solid, Melting point- 89 °C, yield-78%, ¹H NMR (400MHz, CDCl₃) δ 7.50-7.47 (2H, m), 7.01-6.97 (1H, m), 3.96 (1H,brs), 3.61 (1H, brs), 3.51-3.46 (2H, m), 3.36 (3H, s), 3.16 (1H, brs), 1.96-1.80 (2H, m), 1.75-1.61 (2H, m); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 163.4, 157.1 (d, *J* = 249.0 Hz), 133.9 (d, *J* = 7.9 Hz), 131.7 (d, *J* = 3.6 Hz), 126.4 (d, *J* = 19.8 Hz), 117.6 (d, *J* = 23.1 Hz), 117.2 (d, *J* = 3.5 Hz), 74.9, 55.8, 44.1, 38.9, 31.0, 29.9; ¹⁹F NMR (376 MHz, CDCl₃) δ -117.46; HRMS(ESI): Calculated for C₁₃H₁₆BrFNO₂(M+H)⁺- 316.0343, found 316.0349.

4-(4-fluoro-3-(4-methoxypiperidine-1-carbonyl)benzyl)phthalazin-1(2H)-one (19): In a 30 ml glass vial, 2-acetylbenzoic acid (792mg, 4.8 mmol), (5-bromo-2-fluorophenyl)(4-methoxypiperidin-1-yl)methanone (500 mg, 1.58 mmol) and sodium *tert* butoxide (760 mg, 7.9 mmol) were taken and then 5ml of DMF was added to it under argon atmosphere. After that, the reaction mixture was heated to 120 °C for 36-40h. After completion of the reaction, DMF distilled off. Then the mixture was extracted with ethyl acetate and the extract dried with Na₂SO₄. The crude oily product (500mg) was used in next step without further purification. The crude oily product (500mg) was dissolved in EtOH and then adds hydrazine hydrate (480 μl, 1.54 mmol) to it. Reflux the reaction mixture for 15h. Cool the mixture and distilled off the ethanol using rotavapour. Then the mixture was extracted with ethyl acetate and the extract dried with Na₂SO₄. The final product (**19**) was purified by flash chromatography. HPLC purity: 99.9 %. White solid, Melting point-80 °C, yield-68%, IR ν (cm⁻¹): 3913.12, 3885.50, 3781.85, 3432.25, 2931.78, 2121.60, 1633.49, 1451.27, 1407.43, 1353.38, 1268.99, 1228.48, 1182.74, 1154.96, 1092.96, 1024.47, 938.29, 846.02, 754.33, 685.04, 647.23, 560.24, 485.67; ¹H NMR (400MHz, DMSO-d₆) δ 12.60 (1H, s), 8.28-8.25 (1H, m), 7.97-7.95 (1H, m), 7.89-7.79 (2H, m), 7.43-7.34 (2H, m), 7.23-7.18 (1H, m), 4.33 (2H, s), 3.91 (1H, brs), 3.42-3.38 (3H, m), 3.23 (3H,s), 3.05-2.99 (1H, m), 1.88-1.84 (1H, m), 1.72-1.69 (1H, m), 1.47-1.39 (1H, m), 1.32-1.29 (1H, m); ¹³C{¹H} NMR (100 MHz, DMSO-d₆) δ 163.7, 159.4, 156.4 (d, *J* = 244.1 Hz), 144.9, 134.8 (d, *J* = 3.2 Hz), 133.5, 131.5, 131.4 (d, *J* = 7.9 Hz), 129.1, 128.6 (d, *J* = 3.5 Hz), 127.9, 126.1, 125.5, 124.2 (d, *J* = 18.7 Hz), 115.8 (d, *J* = 21.6 Hz), 74.8, 55.0, 43.9, 38.5, 36.5, 30.7, 30.1; ¹⁹F NMR (376 MHz, DMSO-d₆) δ -120.16; HRMS (ESI): Calculated for C₂₂H₂₃FN₃O₃ (M+H)⁺- 396.1718, found 396.1711.

3. Calculations of Green Chemistry Metrics.

The green chemistry metrics of our method and literature reported method were calculated on the basis of Green Chemistry equations.¹

Steps involved in this process (linear reactions) are:

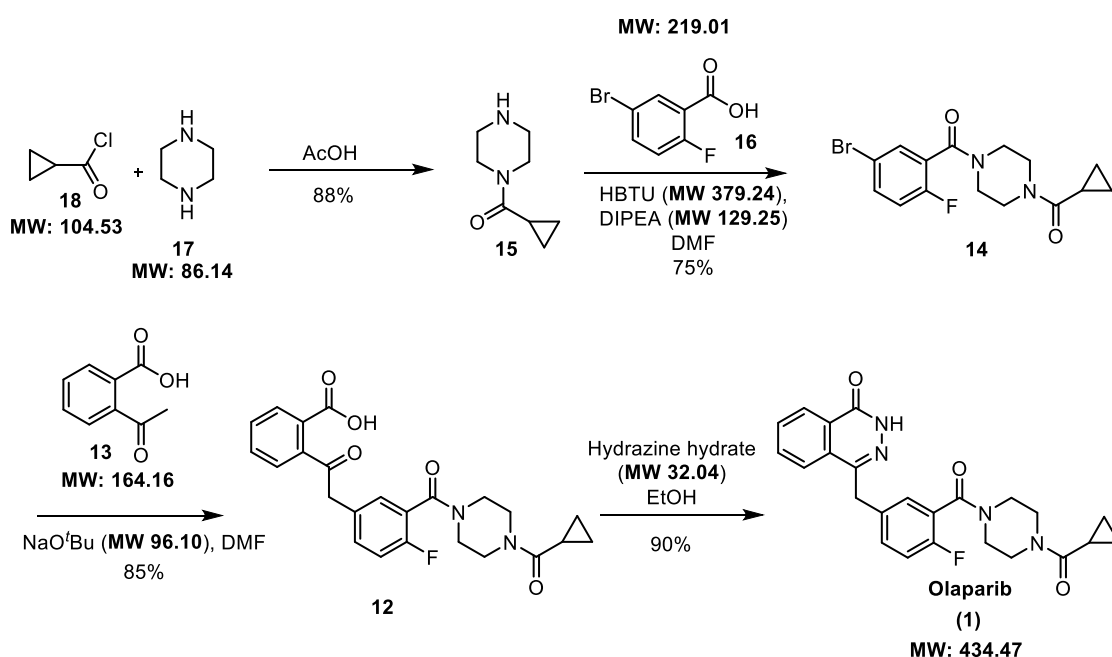


The reactants and reagents efficiently participate in product formation excluding intermediates. The Green Chemistry Metrics are calculated based on general formula¹ as below:

1. No. of steps = No. of steps involved in the process
2. Atom economy = $[(\text{M.W. of product F}) / (\text{M.W. of A} + \text{M.W. of B} + \text{M.W. of D} + \text{M.W. of E})] \times 100$
3. Overall yield = $(\text{Yield of formation of C} \times \text{Yield of formation of F}) / 100$
4. Atom efficiency = $\% \text{ yield} \times \text{Atom economy}$
5. Process Mass Intensity = $(\text{Total mass used in the process} / \text{Mass of the product})$
6. Mass productivity = $(1 / \text{Mass intensity}) \times 100$
7. E-factor = $(\text{Mass intensity} - 1)$
8. Effective Mass Yield = $(1 / \text{E-factor}) \times 100$
9. Reaction Mass Efficiency = $[(\text{Mass of product F}) / (\text{Mass of A} + \text{Mass of B} + \text{Mass of D} + \text{Mass of E})] \times 100$.

A) Calculation of Green Chemistry Metrics for our method

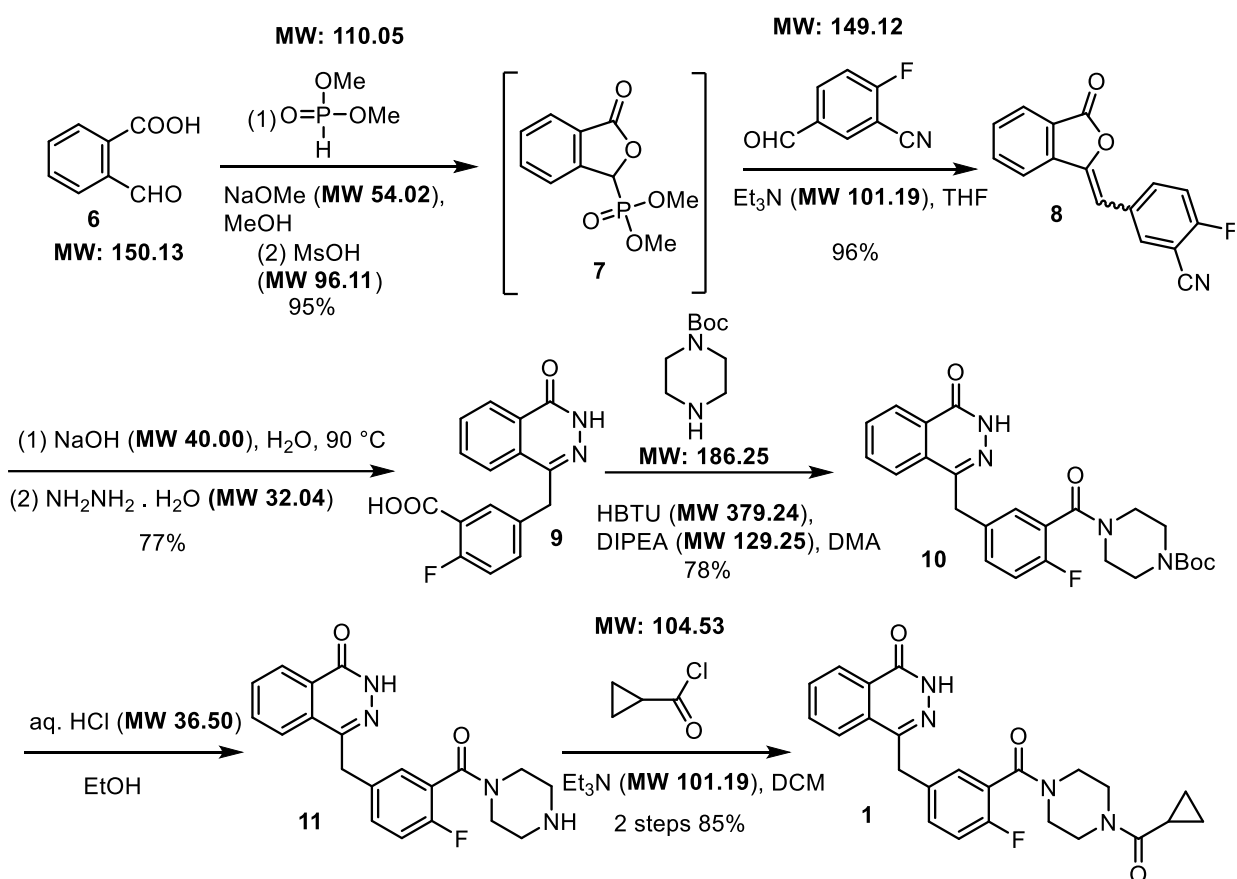
Steps involved in this process are:



1. No. of steps = 4.
2. Atom economy = $[(434.47) / (104.53+86.14+219.01+379.24+129.25+164.16+96.10+32.04)] \times 100 = [434.47/1210.47] \times 100 = 35.89\% \approx 35.9\%$.
3. Overall yield = 51.0%.
4. Atom efficiency = $[(51.0/100) \times 35.9] = 18.3\%$.
5. Process Mass Intensity (PMI) = $[(3.32+2.50+10.50+5.54+9.59+9.81+23.60+9.34+9.12+18.90+0.70+11.80)/6.30] = [114.72/6.30] = 18.21 \text{ g/g} \approx 18.2 \text{ g/g}$.
6. Mass productivity = $[(1/18.2) \times 100] = [0.0549 \times 100] = 5.5\%$.
7. E-factor = $18.2 - 1 = 17.2 \text{ g/g}$.
8. Effective Mass Yield = $[(1/17.2) \times 100] = [0.0581 \times 100] = 5.8\%$.
9. Reaction Mass Efficiency (RME) = $[(6.30 / (3.32+2.50+5.54+9.59+9.81+9.34+9.12+0.70)) \times 100] = [(6.30/49.92) \times 100] = 12.62\% \approx 12.6\%$.

B) Calculation of Green Chemistry Metrics for Literature reported method² (WO 2004/080976 A1 & WO 2008/047082 A2)

Consideration: All calculations are done based on two literature reported patents (WO 2004/080976 A1 & WO 2008/047082 A2). In our work, 6.30 g Olaparib was synthesized. The two above-mentioned patents disclosed all reactions procedures, such as equivalent of reactants, mmol, yield etc. But the scaling of all steps is different. So, for mass calculation purpose and comparison with our method, we assume the scale of Olaparib synthesis in 6.30 g by using reported patents data (such as equivalent of reactants, mmol, yield etc).



- No. of steps = 6.
- Atom economy = $[(434.47) / (150.13+110.05+54.02+96.11+149.12+101.19+40.00+32.04+186.25+379.24+129.25+36.50+104.53+101.19)] \times 100 = [434.47/1669.62] \times 100 = 26.02\% \approx 26.0\%$.
- Overall yield = 46.0%.
- Atom efficiency = $[(46.0/100) \times 26.0] = 11.96\% \approx 12.0\%$.
- Process Mass Intensity (PMI) = $[(4.82+7.11+13.96+6.83+36.21+4.57+3.10+64.16+3.54+1.42+15.91+5.07+11.19+6.46+26.49+12.91+5.14+1.68+1.63+94.40)/6.30] = [326.60/6.30] = 51.84 \text{ g/g} \approx 51.8 \text{ g/g}$.
- Mass productivity = $[(1/51.8) \times 100] = [0.0193 \times 100] = 1.9\%$.
- E-factor = $51.8 - 1 = 50.8 \text{ g/g}$
- Effective Mass Yield = $[(1/50.8) \times 100] = [0.0197 \times 100] = 1.97\% \approx 2.0\%$.
- Reaction Mass Efficiency (RME) = $[(6.30 / (4.82+7.11+13.96+6.83+4.57+3.10+3.54+1.42+5.07+11.19+6.46+12.91+1.68+1.63))] \times 100 = [(6.30/84.29) \times 100] = 7.47\% \approx 7.5\%$.

4. Table S1: Table of comparing the shifts (¹H NMR spectrum) of final compound Olaparib (1) with the literature reported³ values.

¹ H NMR (400MHz, DMSO-d ₆) Compound 1 (δ ppm)	¹ H NMR (400MHz, DMSO-d ₆) Literature reported compound ³ (δ ppm)
12.59 (1H, s)	12.53 (1H, s)
8.28-8.26 (1H, m)	8.25 (1H, dd)
7.98-7.82 (3H, m)	7.92 (1H, d), 7.83 (1H, dt), 7.77 (1H, dt)
7.46-7.37 (2H, m)	7.41 (1H, m), 7.34 (1H, dd)
7.27-7.22 (1H, m)	7.17 (1H, t)
4.34 (2H, s)	4.31 (2H, s)
3.75-3.60 (6H, m)	3.56 (6H, m)
3.23-3.17 (2H, m)	3.20 (2H, brs)
1.99 (1H, s)	1.88 (1H, brs)
0.75-0.72 (4H, m)	0.70 (4H, m)

5. A) Table S2: Table of comparing the shifts (¹H NMR spectrum) of final compound AZD2461(19) with the literature reported⁴ values.

¹ H NMR (400MHz, DMSO-d ₆) Compound 19 (δ ppm)	¹ H NMR (500MHz, CDCl ₃) Literature reported compound ⁴ (δ ppm)
12.60 (1H, s)	12.12 (1H, s)
8.28-8.25 (1H, m)	8.41-8.39 (1H, m)
7.97-7.95 (1H, m), 7.89-7.79 (2H, m)	7.66-7.65 (3H, m)
7.43-7.34 (2H, m)	7.28-7.24 (1H, m), 7.23-7.21 (1H, m)
7.23-7.18 (1H, m)	6.94-6.90 (1H, m)
4.33 (2H, s)	4.22 (2H, s)
3.91 (1H, brs)	3.91 (1H, brs)
3.42-3.38 (3H, m)	3.50-3.39 (1H, m), 3.31-3.27 (2H, m)

3.23 (3H, s)	3.12 (3H, s)
3.05-2.99 (1H, m)	3.05-2.81 (1H, m)
1.88-1.84 (1H, m)	1.86-1.82 (1H, m)
1.72-1.69 (1H, m)	1.69 (1H, brs)
1.47-1.39 (1H, m)	1.64-1.62 (1H, brs)
1.32-1.29 (1H, m)	1.60-1.59 (1H, m)

B) Table S3: Table of comparing the shifts $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum of final compound AZD2461(**19**) with the literature reported⁴ values.

^{13}C NMR (100MHz, DMSO- d_6) Compound 19 (δ ppm)	^{13}C NMR (125MHz, CDCl_3) Literature reported compound ⁴ (δ ppm)
163.73	164.75
159.44	161.10
157.61 & 155.18	157.96 & 155.99
144.96	145.55
134.83 & 134.80	134.20 & 134.18
133.48	133.46
131.55	131.34
131.44 & 131.36	131.04 & 130.97
129.10	129.48
128.61 & 128.57	128.81 & 128.78
127.93	128.16
126.10	126.93
125.46	125.02
124.29 & 124.10	124.51 & 124.36
115.96 & 115.75	116.03 & 115.86
74.76	74.97
55.04	55.86
43.90	44.08
38.52	38.81
36.48	37.86
30.72	30.91
30.10	29.94

6. References

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7. Spectra of compounds

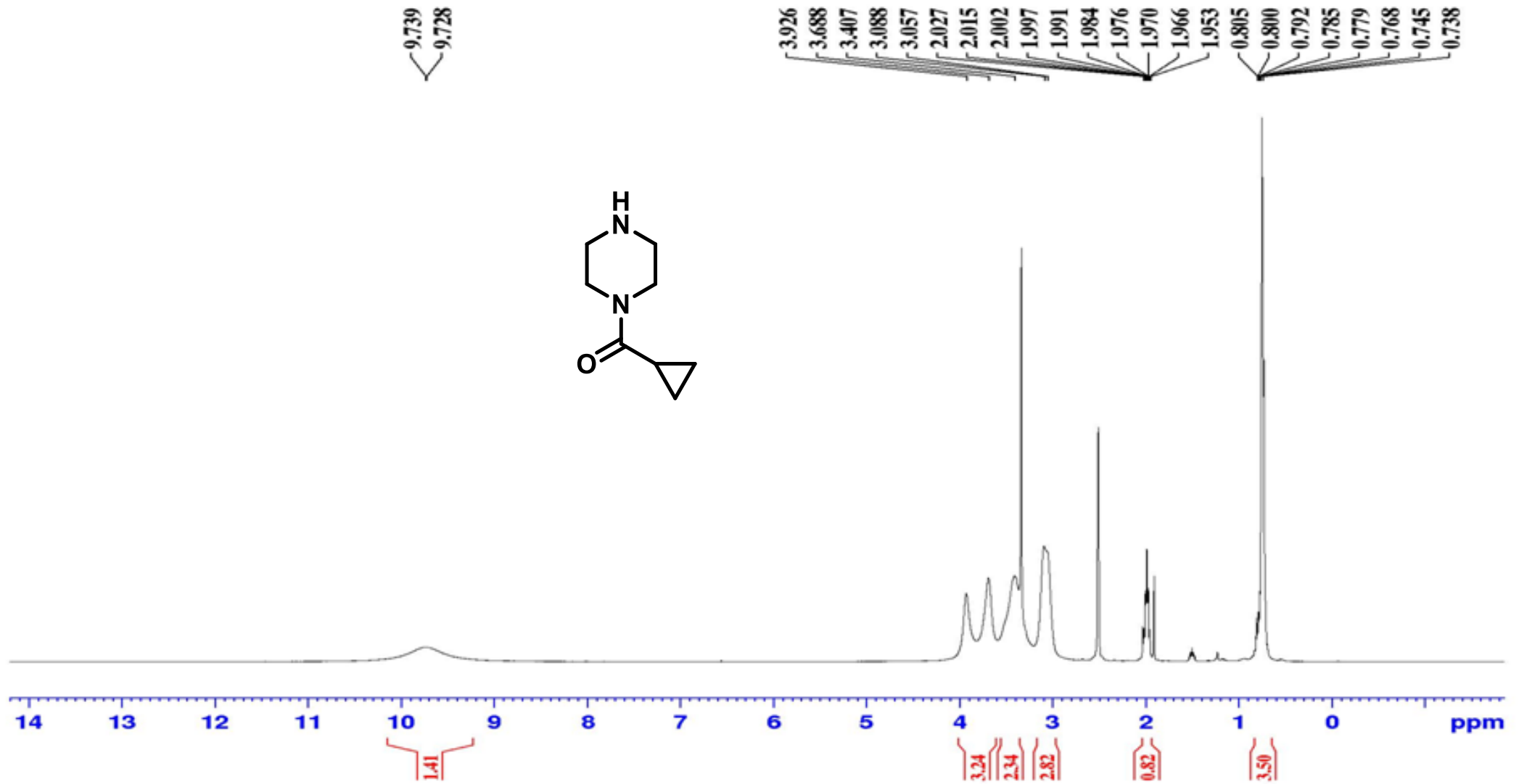


Figure S1. ¹H NMR of Cyclopropyl (piperazin-1-yl) methanone (15)

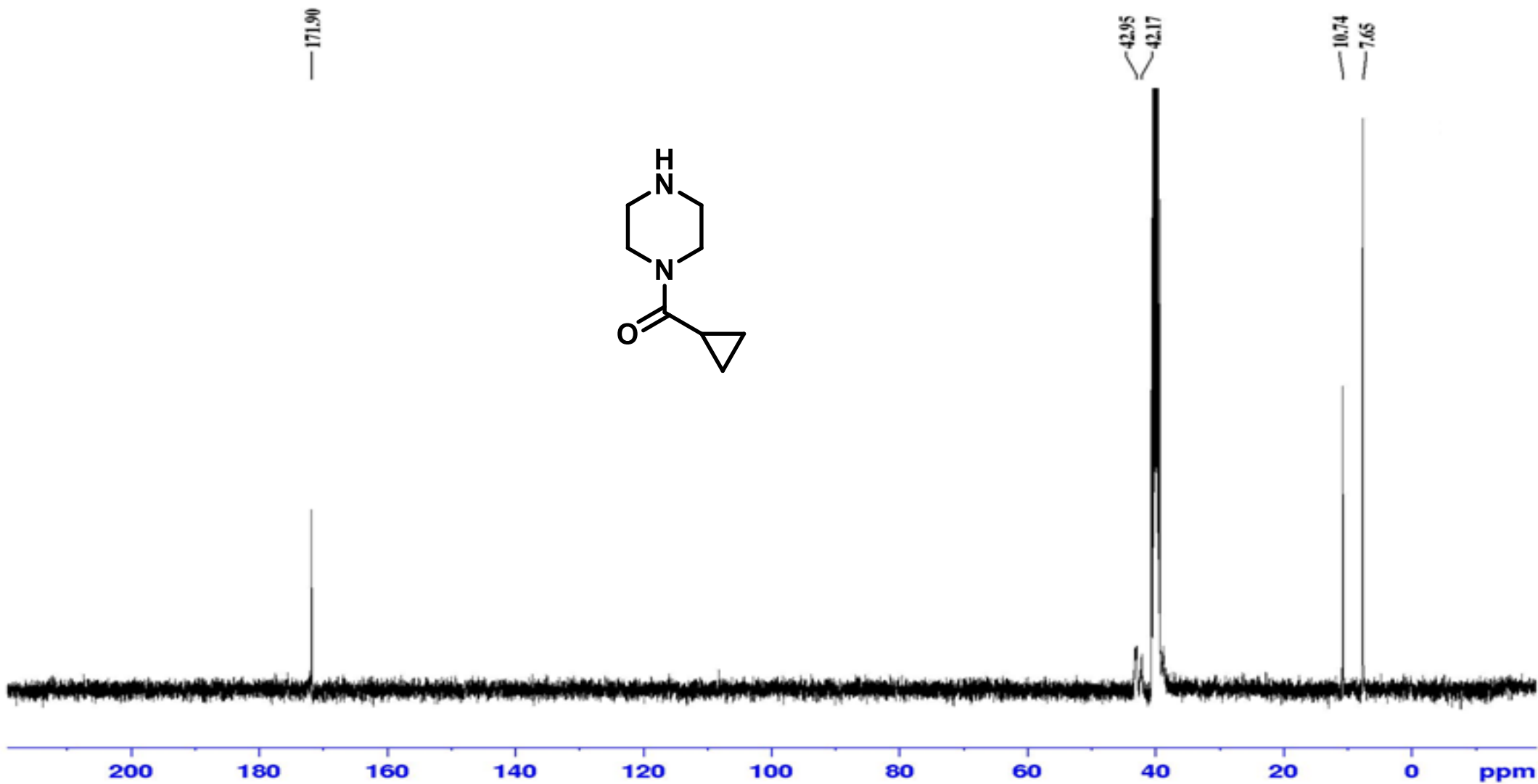


Figure S2. $^{13}\text{C}\{^1\text{H}\}$ NMR spectra of Cyclopropyl (piperazin-1-yl) methanone (15)

Sample Name	HRMS21I08DEC28	Position	Vial 28	Instrument Name	Instrument 1	User Name	
Inj Vol	1	InjPosition		SampleType	Sample	IRM Calibration Status	Some Ions Missed
Data Filename	AK-468.d	ACQ Method	ISOCRATIC.m	Comment		Acquired Time	12/8/2021 12:46:06 PM

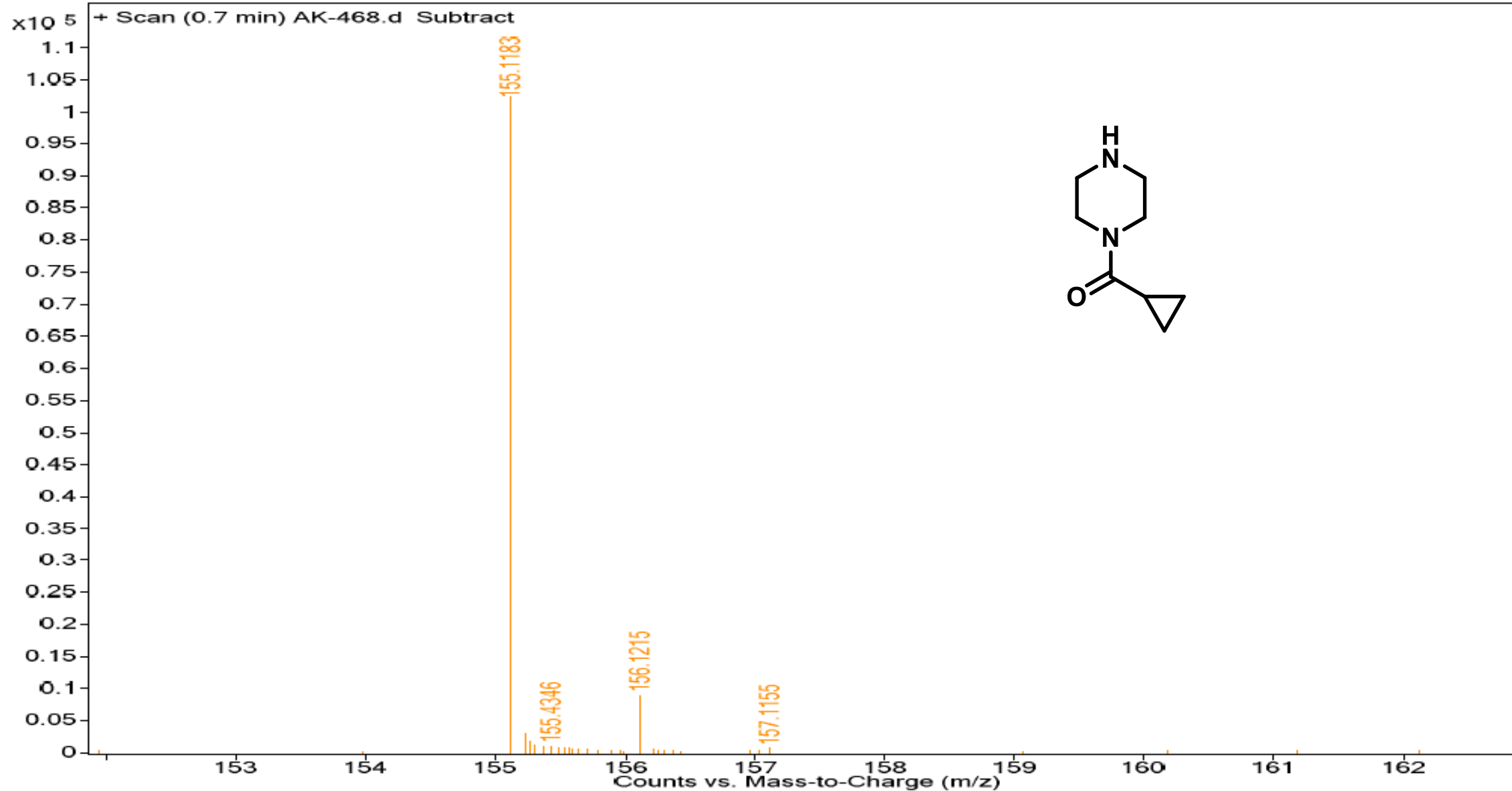


Figure S3. HRMS spectra of Cyclopropyl (piperazin-1-yl) methanone (15)

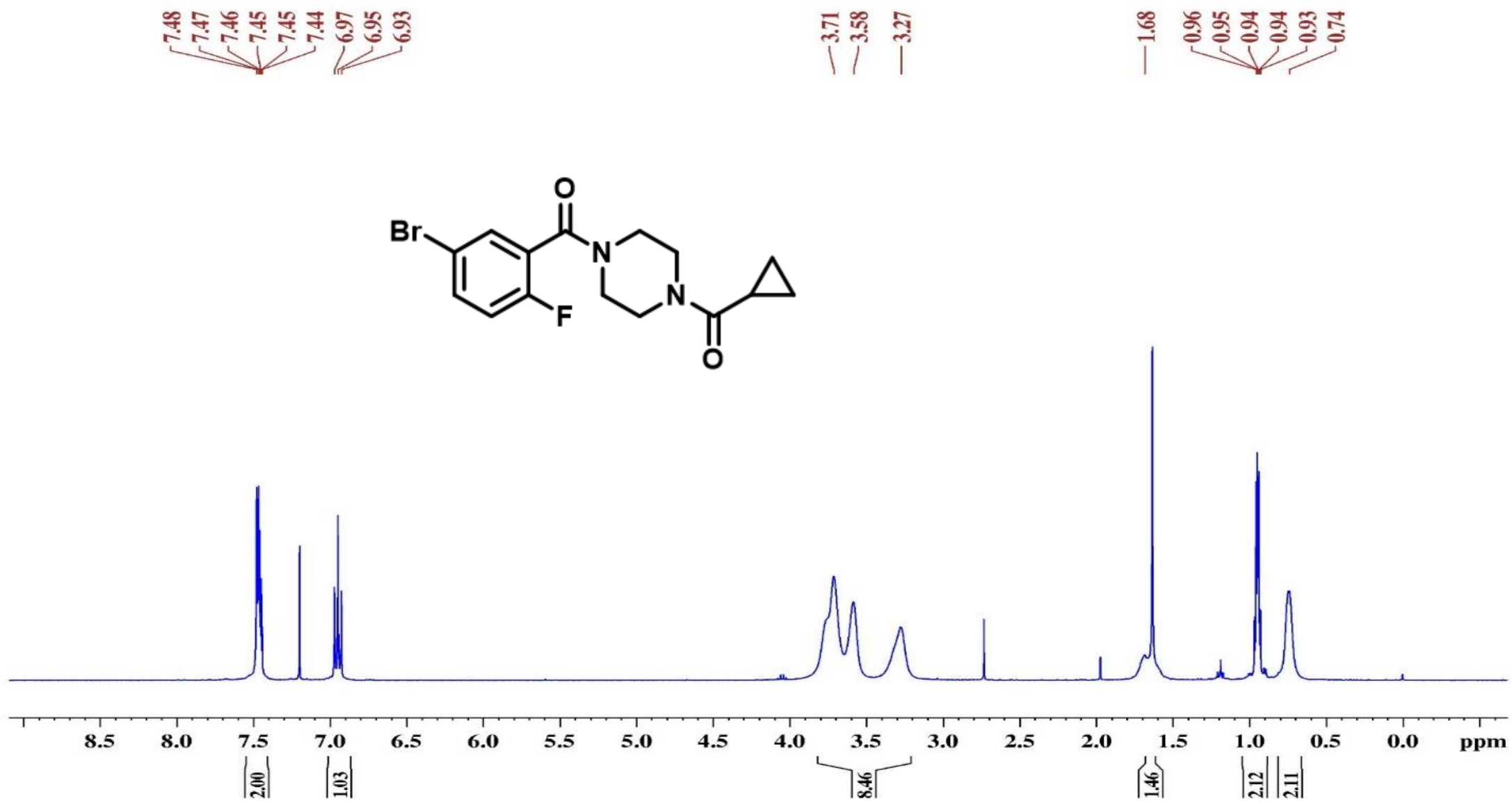


Figure S4. ¹H NMR spectra of (4-(5-bromo-2-fluorobenzoyl)piperazin-1-yl)(cyclopropyl)methanone (14)

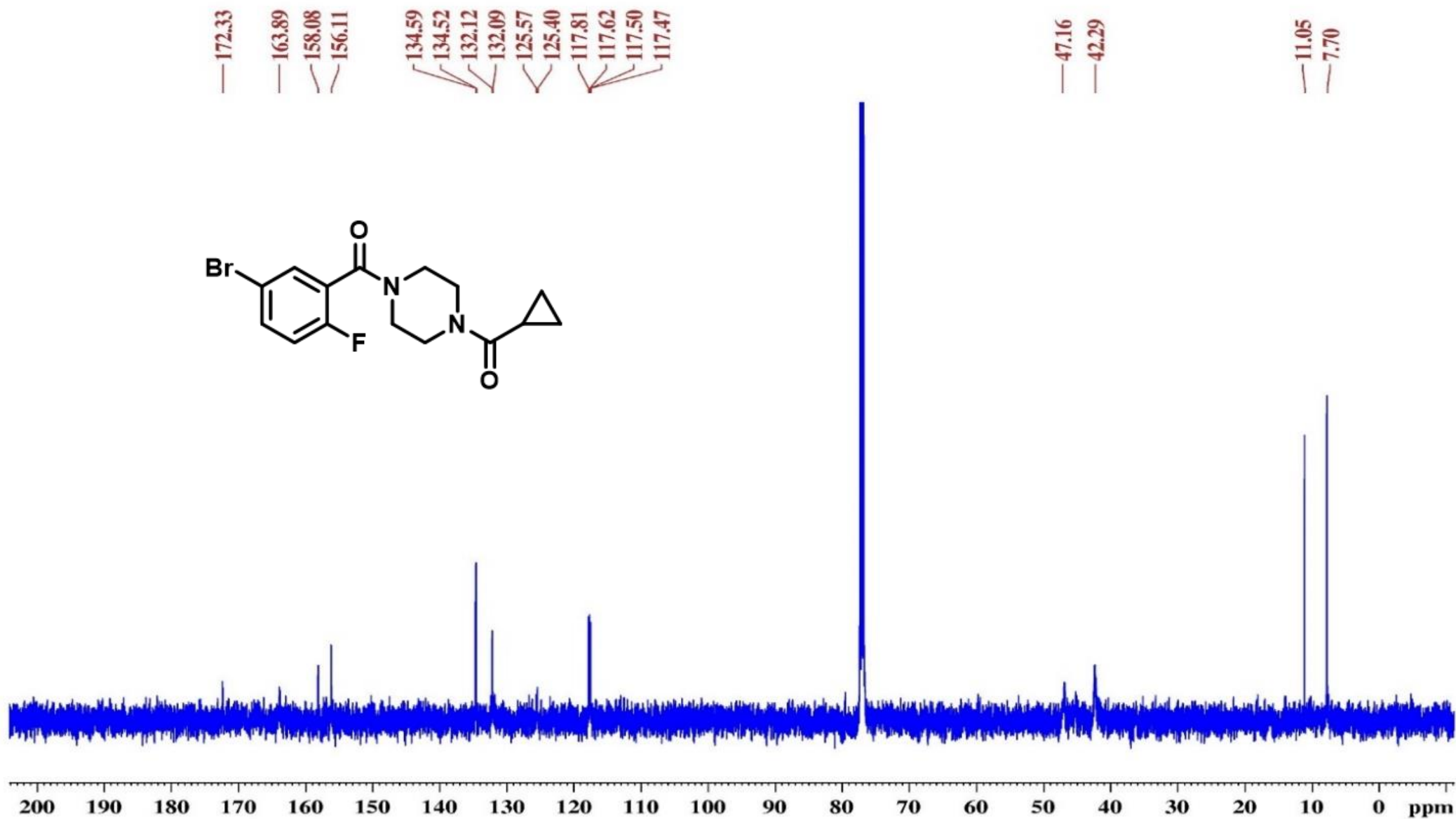


Figure S5. $^{13}\text{C}\{^1\text{H}\}$ NMR spectra of (4-(5-bromo-2-fluorobenzoyl)piperazin-1-yl)(cyclopropyl)methanone (14) S15

Sample Name	HRMS21I08DEC27	Position	Vial 27	Instrument Name	Instrument 1	User Name	
Inj Vol	1	InjPosition		SampleType	Sample	IRM Calibration Status	Some Ions Missed
Data Filename	AK-472.d	ACQ Method	ISOCRATIC.m	Comment		Acquired Time	12/8/2021 12:42:27 PM

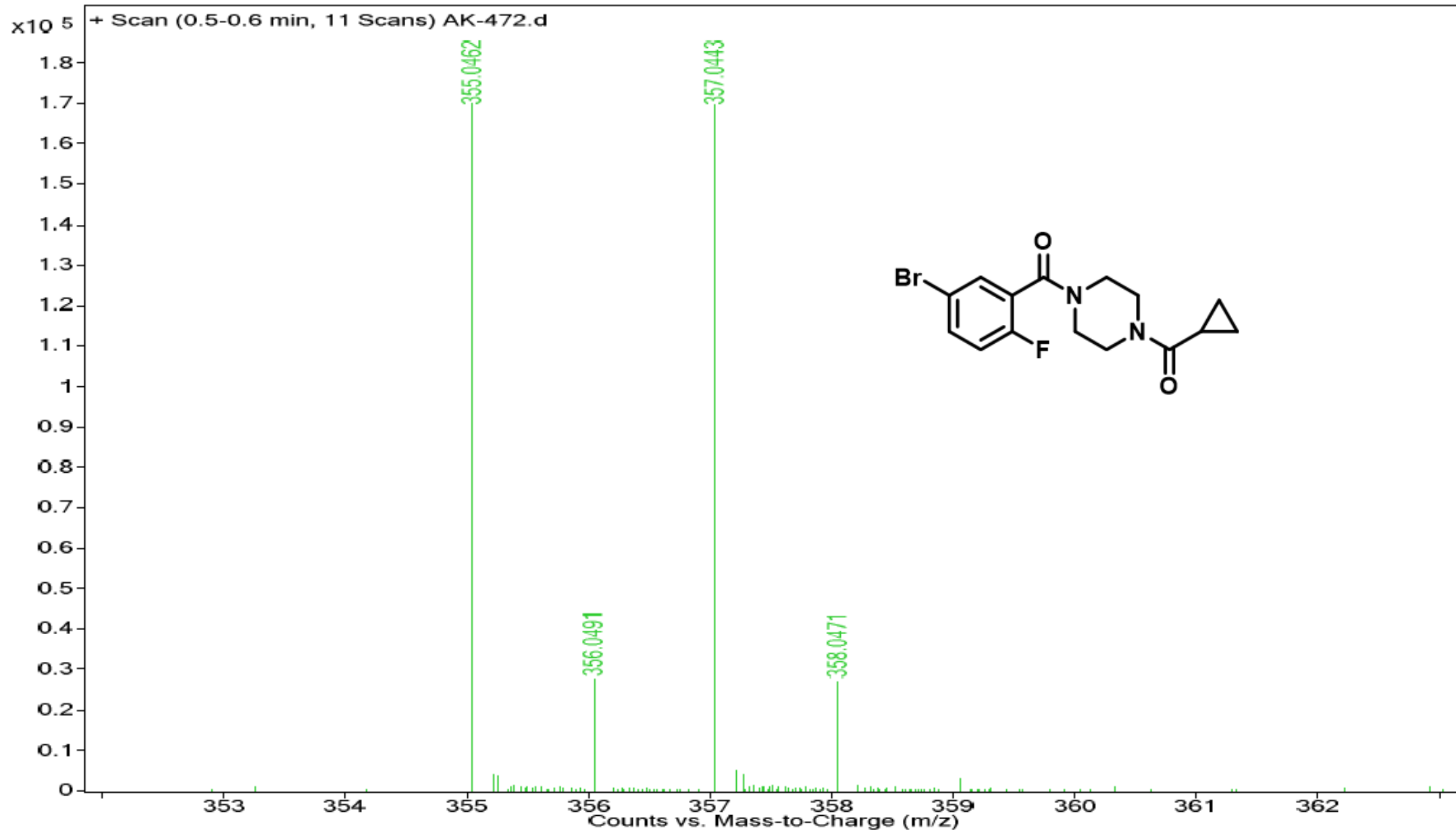


Figure S6. HRMS spectra of (4-(5-bromo-2-fluorobenzoyl)piperazin-1-yl)(cyclopropyl)methanone (14)

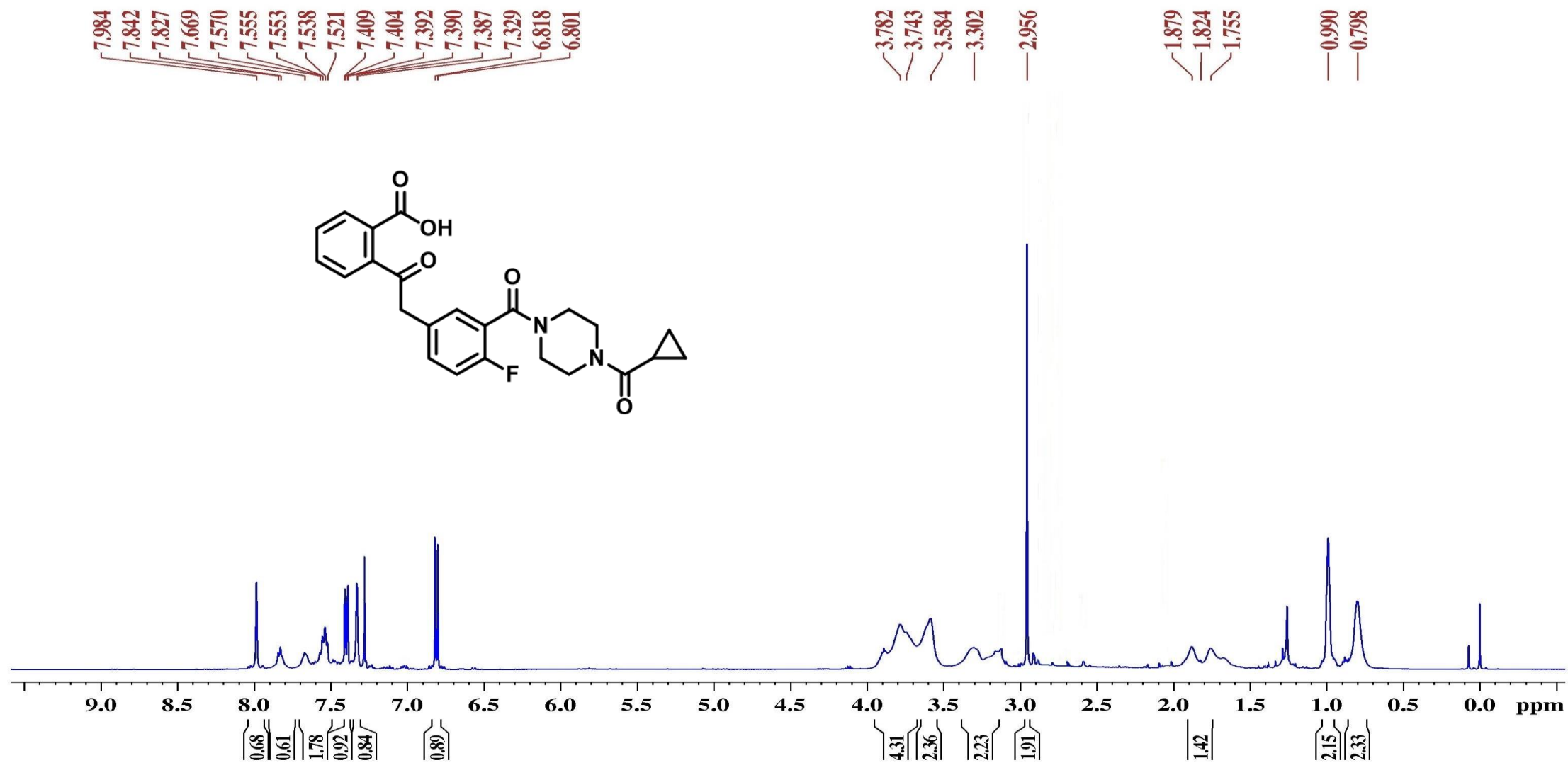


Figure S7. ¹H NMR spectra of 2-(2-(3-(4-(cyclopropanecarbonyl) piperazine-1-carbonyl)-4-fluorophenyl) acetyl) benzoic acid (12)

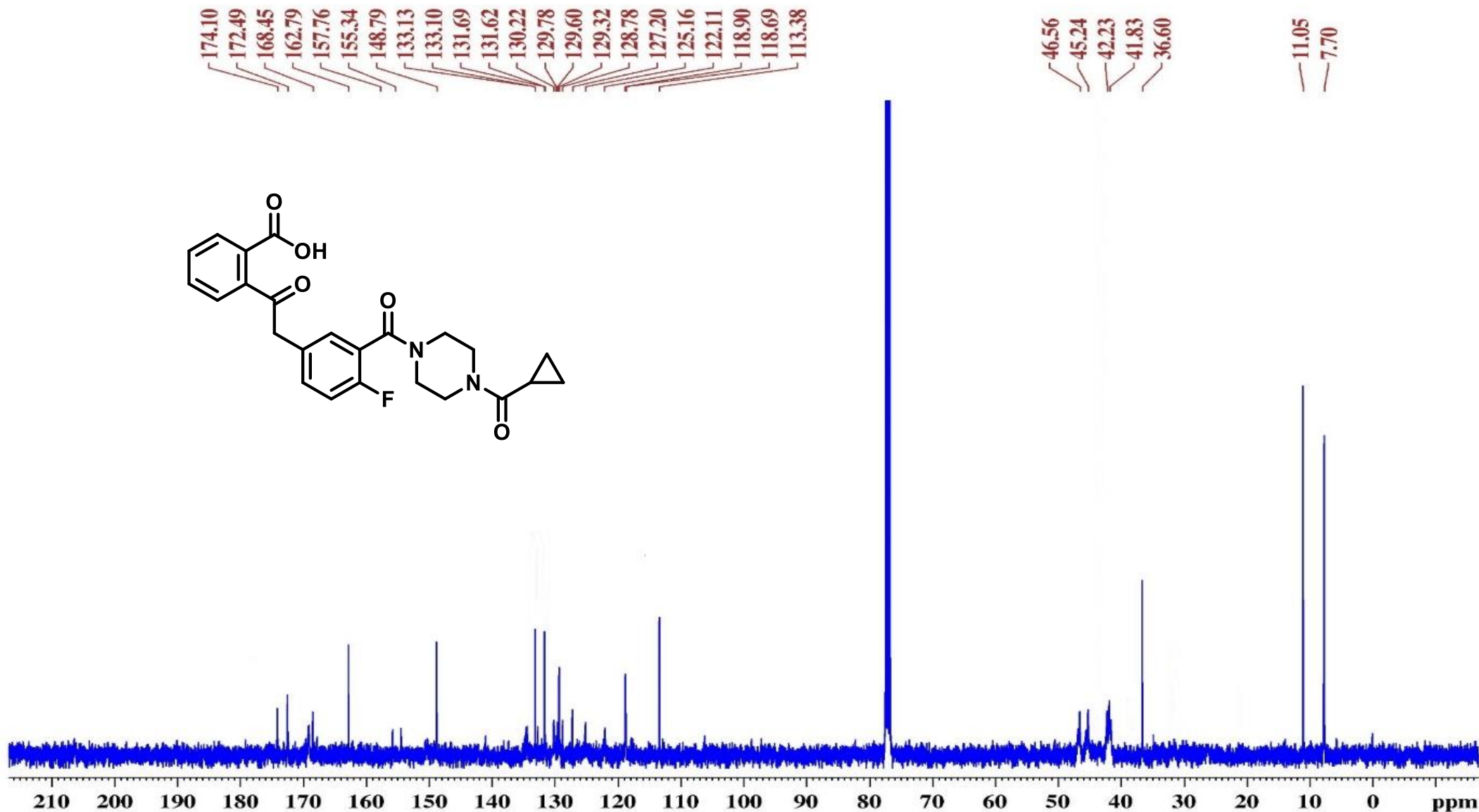


Figure S8. $^{13}\text{C}\{^1\text{H}\}$ NMR spectra of 2-(2-(3-(4-(cyclopropanecarbonyl) piperazine-1-carbonyl)-4-fluorophenyl) acetyl) benzoic acid (12)

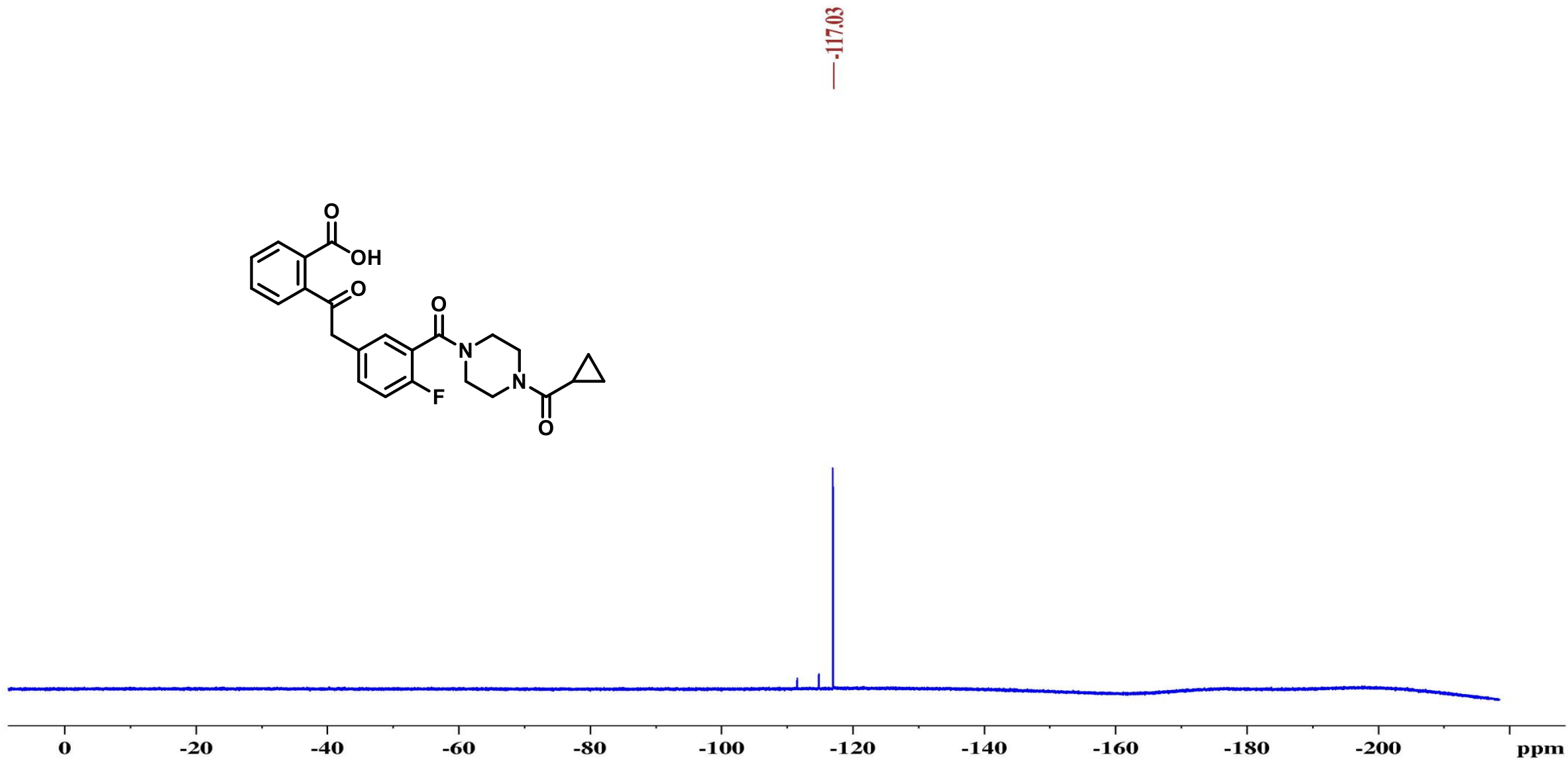


Figure S9. ^{19}F NMR spectra of 2-(2-(3-(4-(cyclopropanecarbonyl) piperazine-1-carbonyl)-4-fluorophenyl) acetyl) benzoic acid (12)

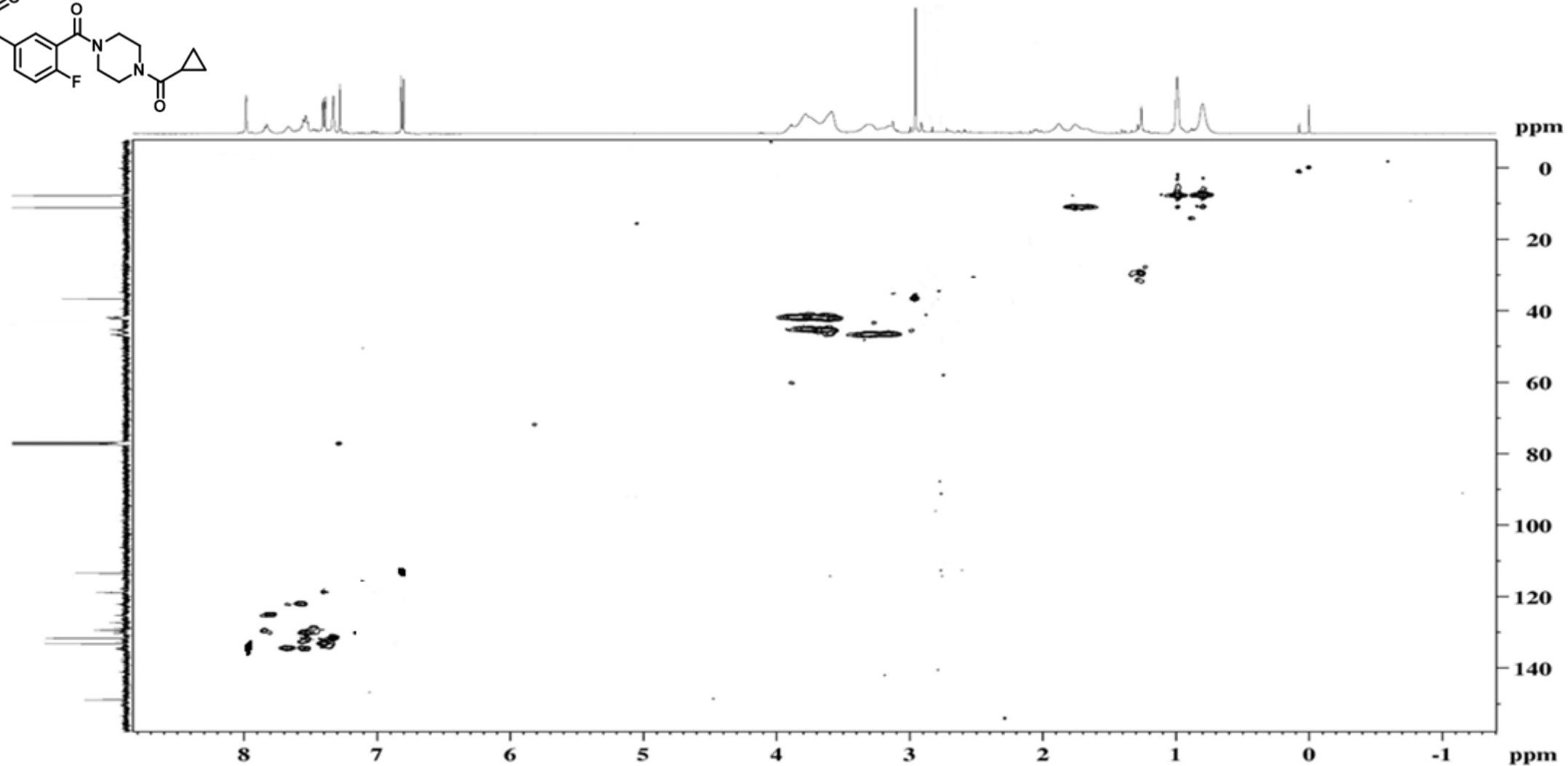
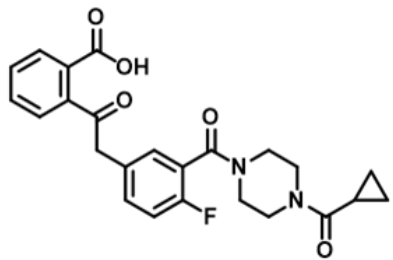


Figure S10. HSQC spectra of 2-(2-(3-(4-(cyclopropanecarbonyl) piperazine-1-carbonyl)-4-fluorophenyl) acetyl) benzoic acid (12)

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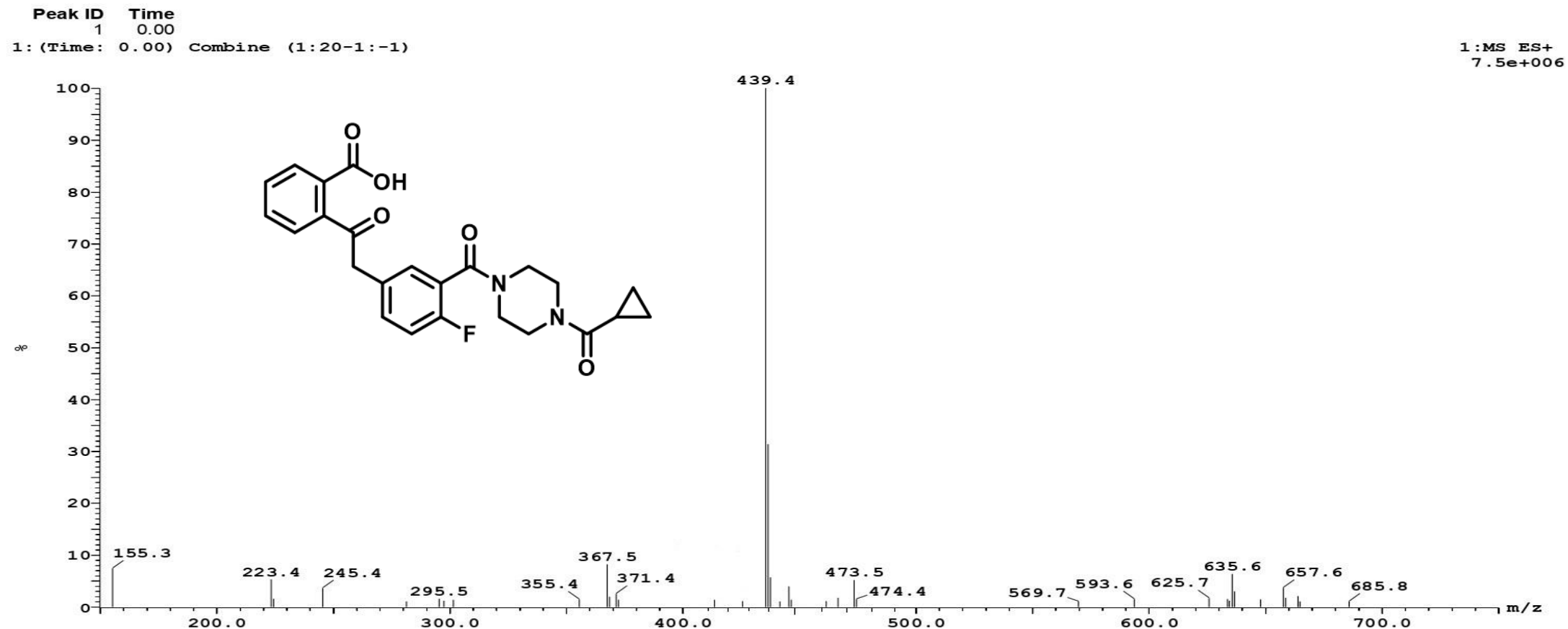


Figure S11. ESMS spectra of 2-(2-(3-(4-(cyclopropanecarbonyl) piperazine-1-carbonyl)-4-fluorophenyl) acetyl) benzoic acid (12)

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FTMS + c ESI Full ms [100.00-750.00]

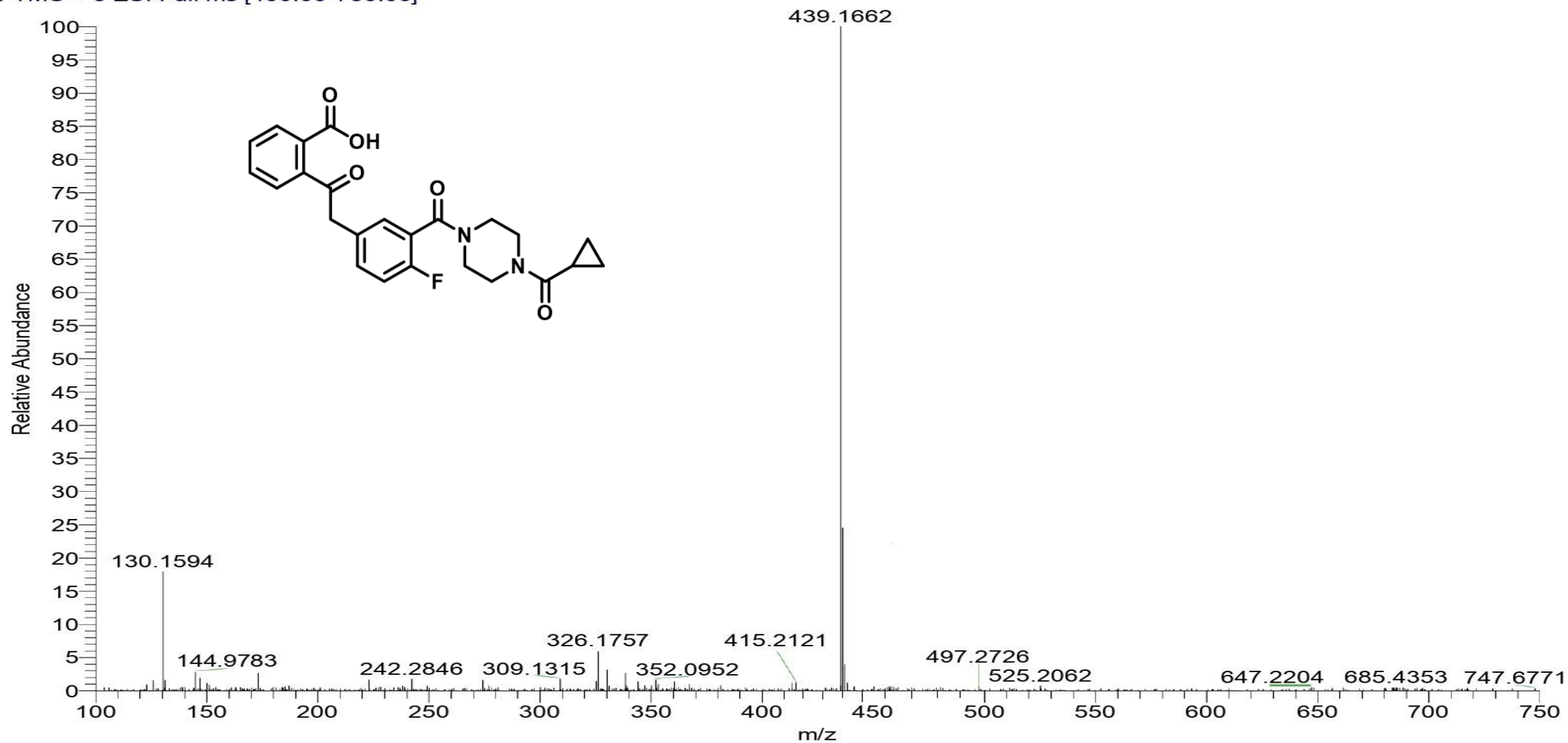


Figure S12. HRMS spectra of 2-(2-(3-(4-(cyclopropanecarbonyl) piperazine-1-carbonyl)-4-fluorophenyl) acetyl) benzoic acid (12)

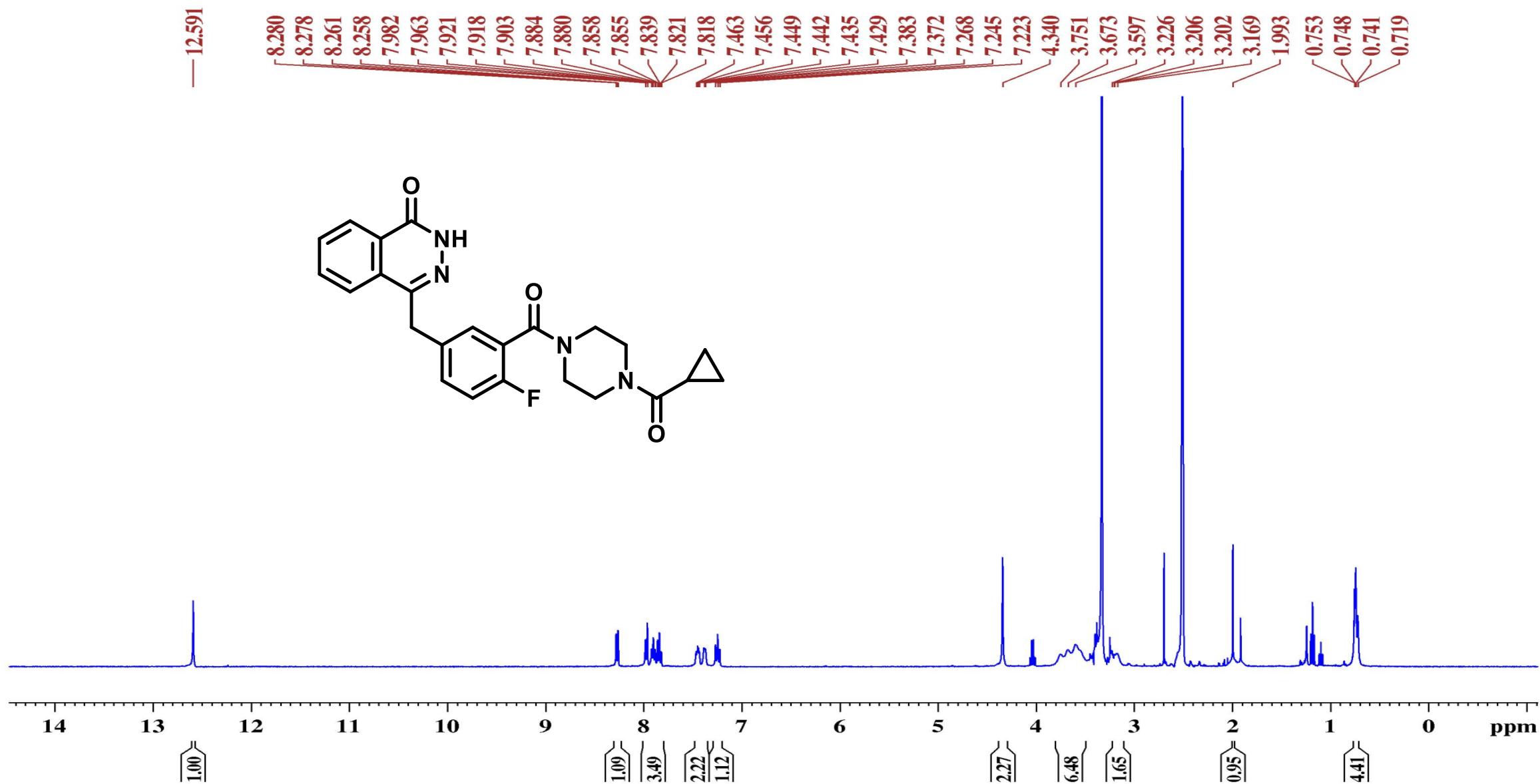


Figure S13. ¹H NMR spectra of 4-(3-(4-(cyclopropanecarbonyl) piperazine-1-carbonyl)-4-fluorobenzyl) phthalazin-1(2H)-one (1)

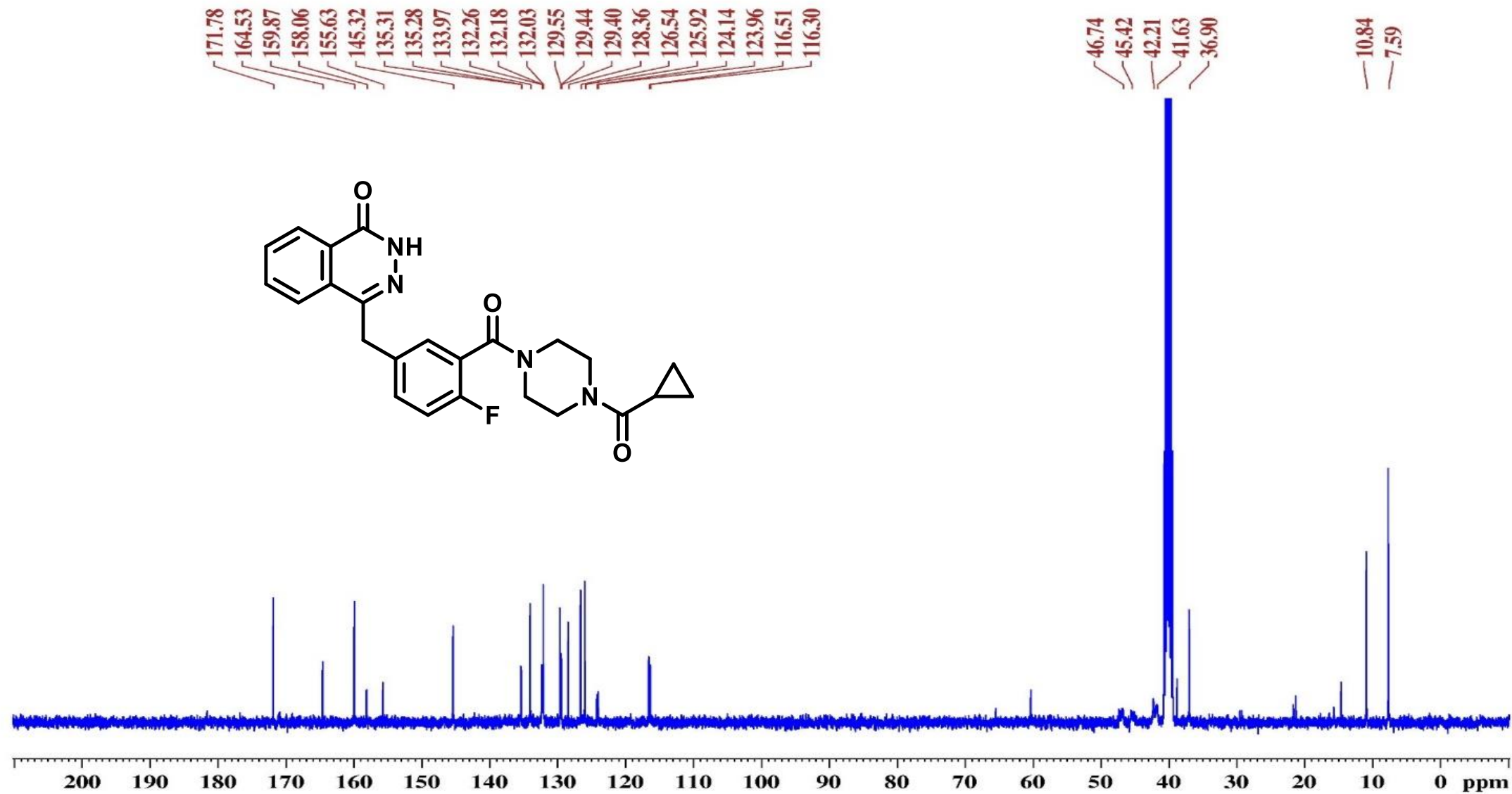


Figure S14. $^{13}\text{C}\{^1\text{H}\}$ NMR spectra of 4-(3-(4-(cyclopropanecarbonyl) piperazine-1-carbonyl)-4-fluorobenzyl) phthalazin-1(2H)-one (1)

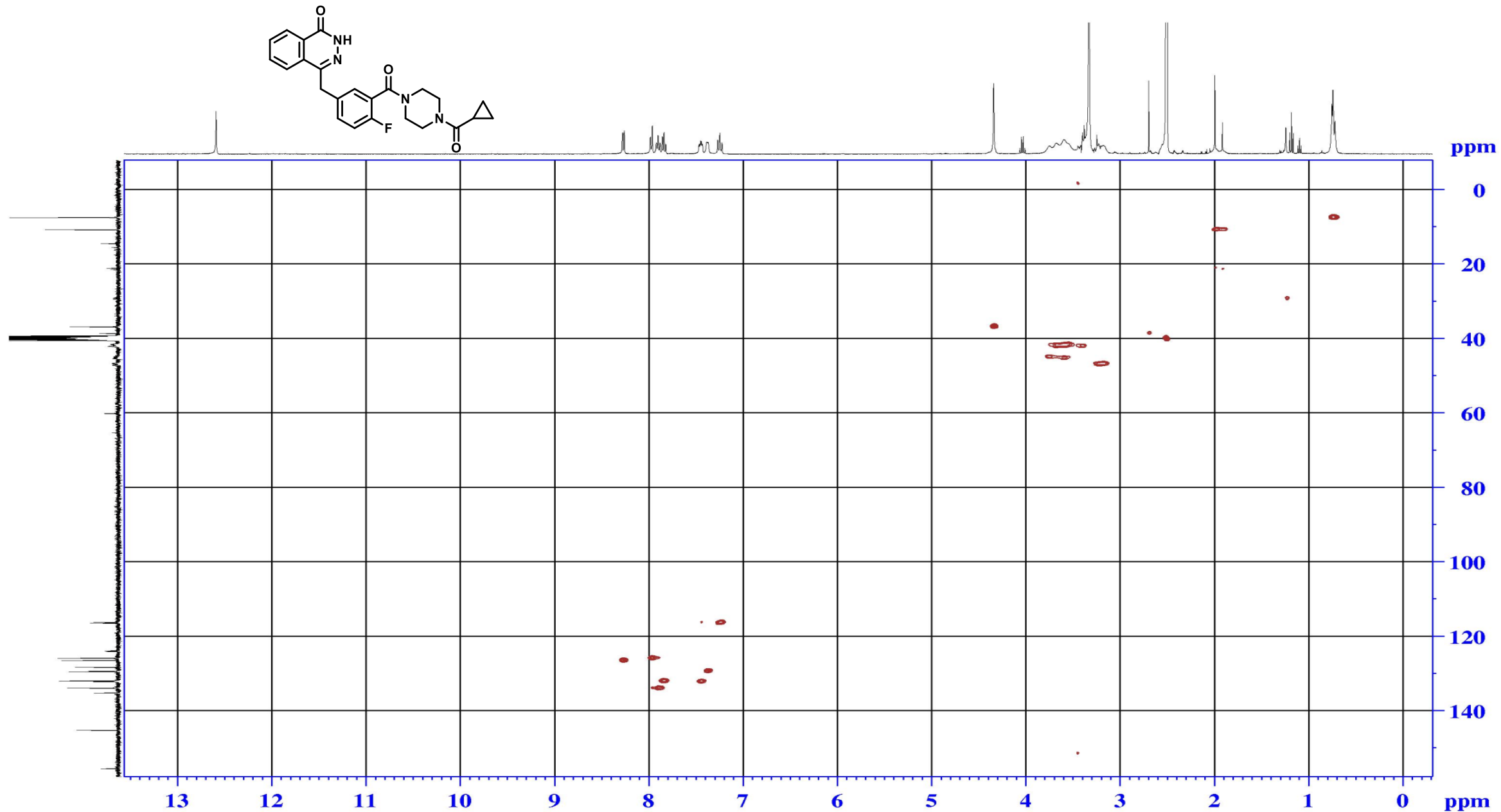


Figure S15. HSQC spectra of 4-(3-(4-(cyclopropanecarbonyl) piperazine-1-carbonyl)-4-fluorobenzyl) phthalazin-1(2H)-one (1)

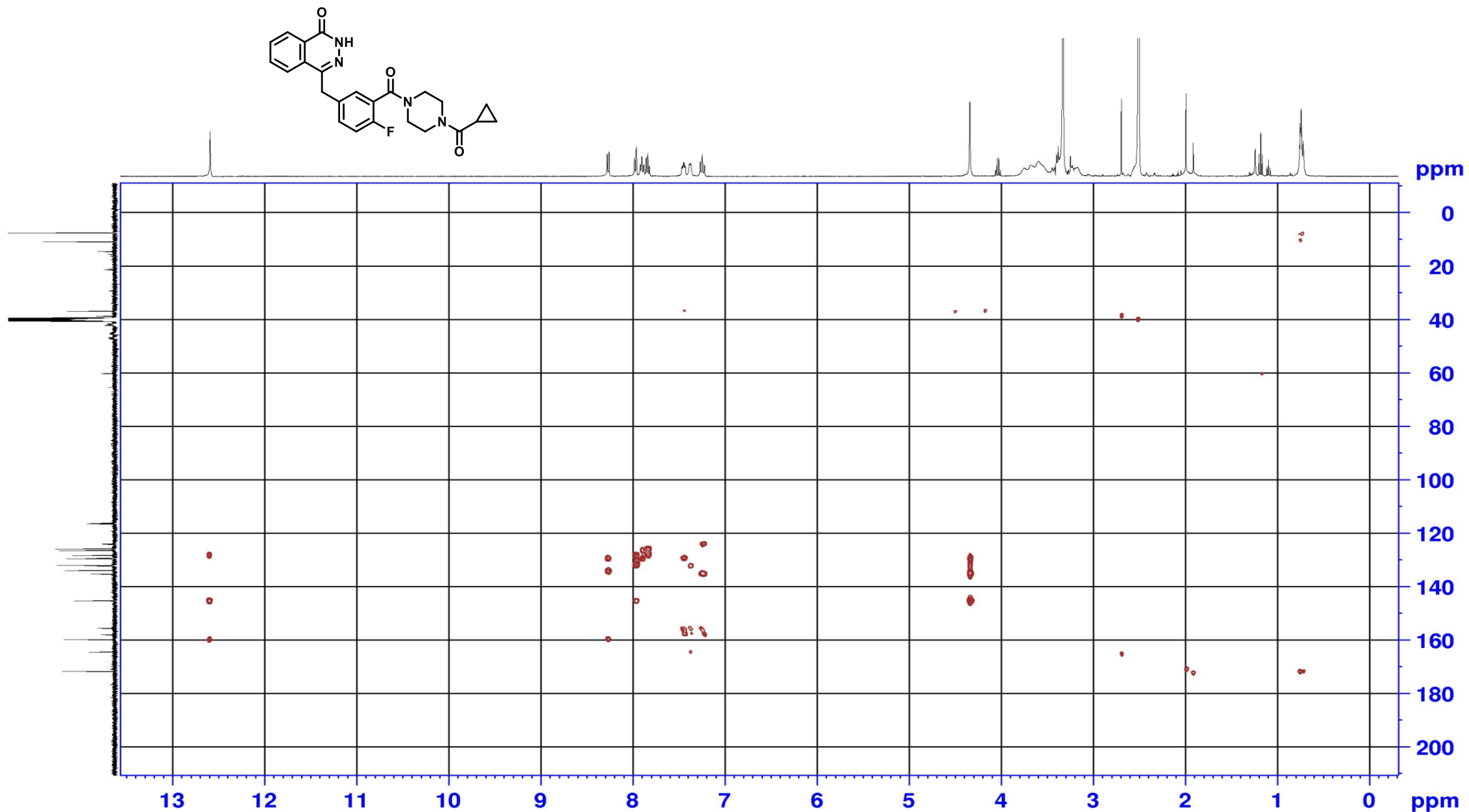


Figure S16. HMBC spectra of 4-(3-(4-(cyclopropanecarbonyl) piperazine-1-carbonyl)-4-fluorobenzyl) phthalazin-1(2H)-one (1)

Peak ID Time
1 0.00

1: (Time: 0.00) Combine (1:20-1:-1)

1: MS ES+
1.2e+006

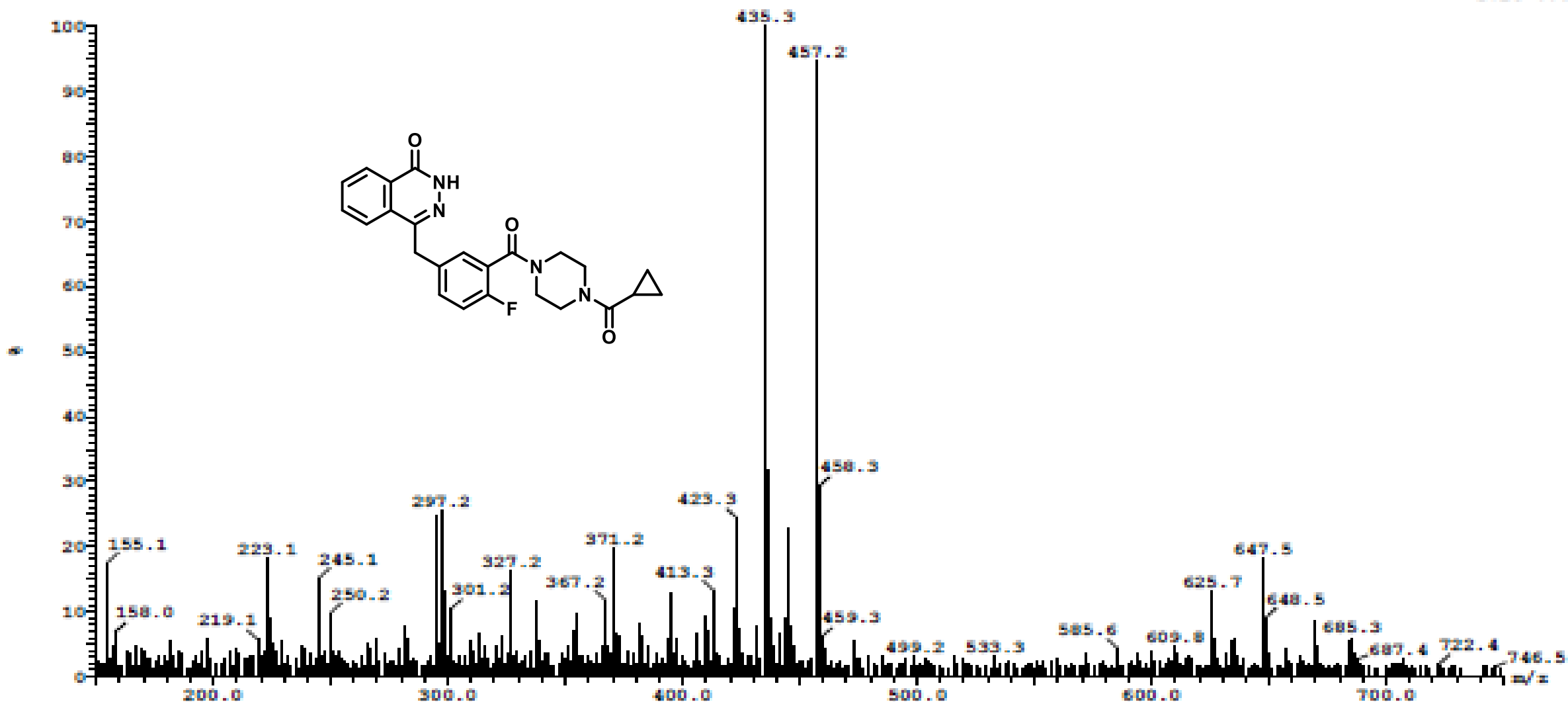


Figure S17. ESMS spectra of 4-(3-(4-(cyclopropanecarbonyl) piperazine-1-carbonyl)-4-fluorobenzyl) phthalazin-1(2H)-one (1)

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T: FTMS + c ESI Full ms [100.00-750.00]

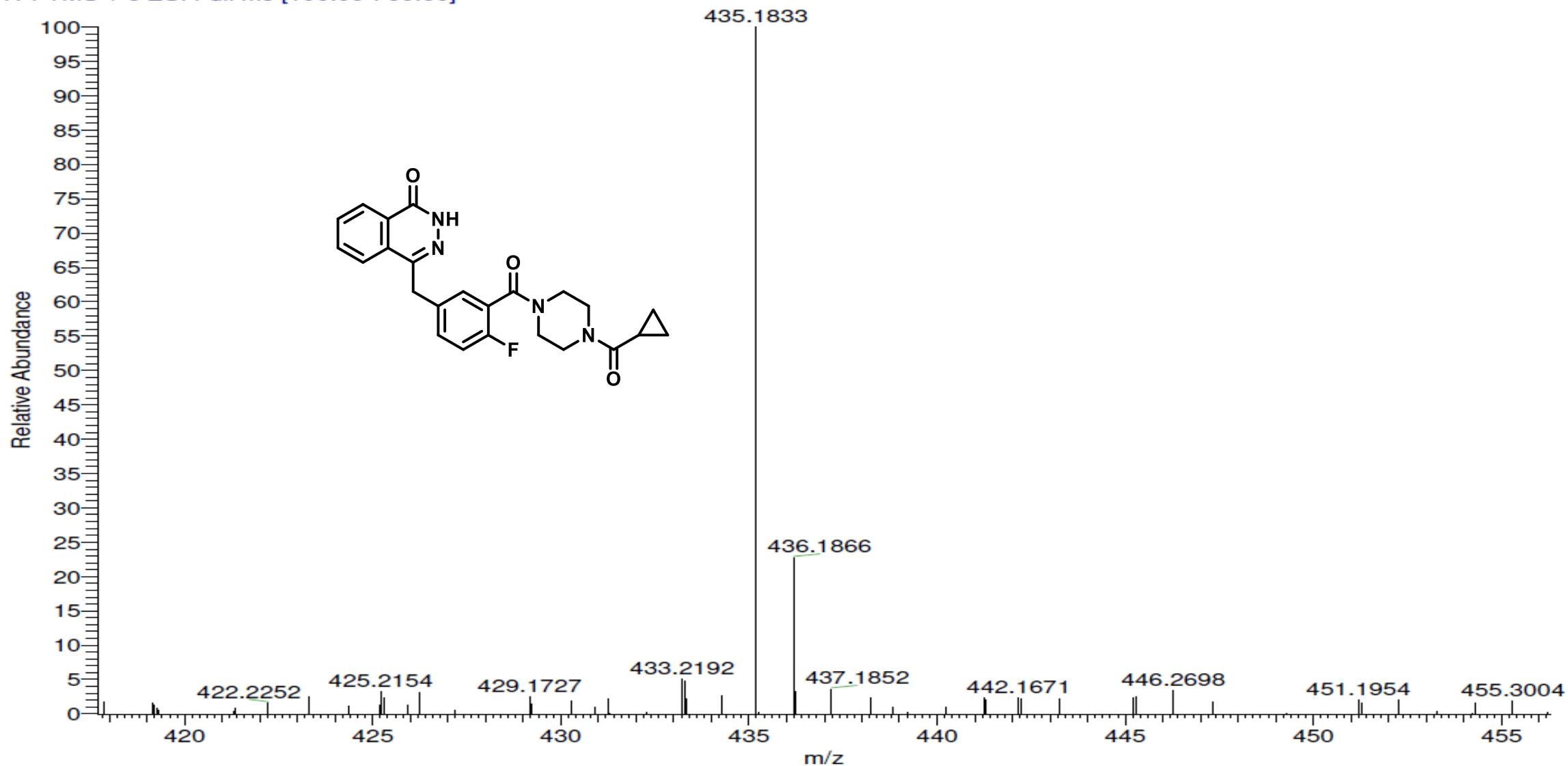
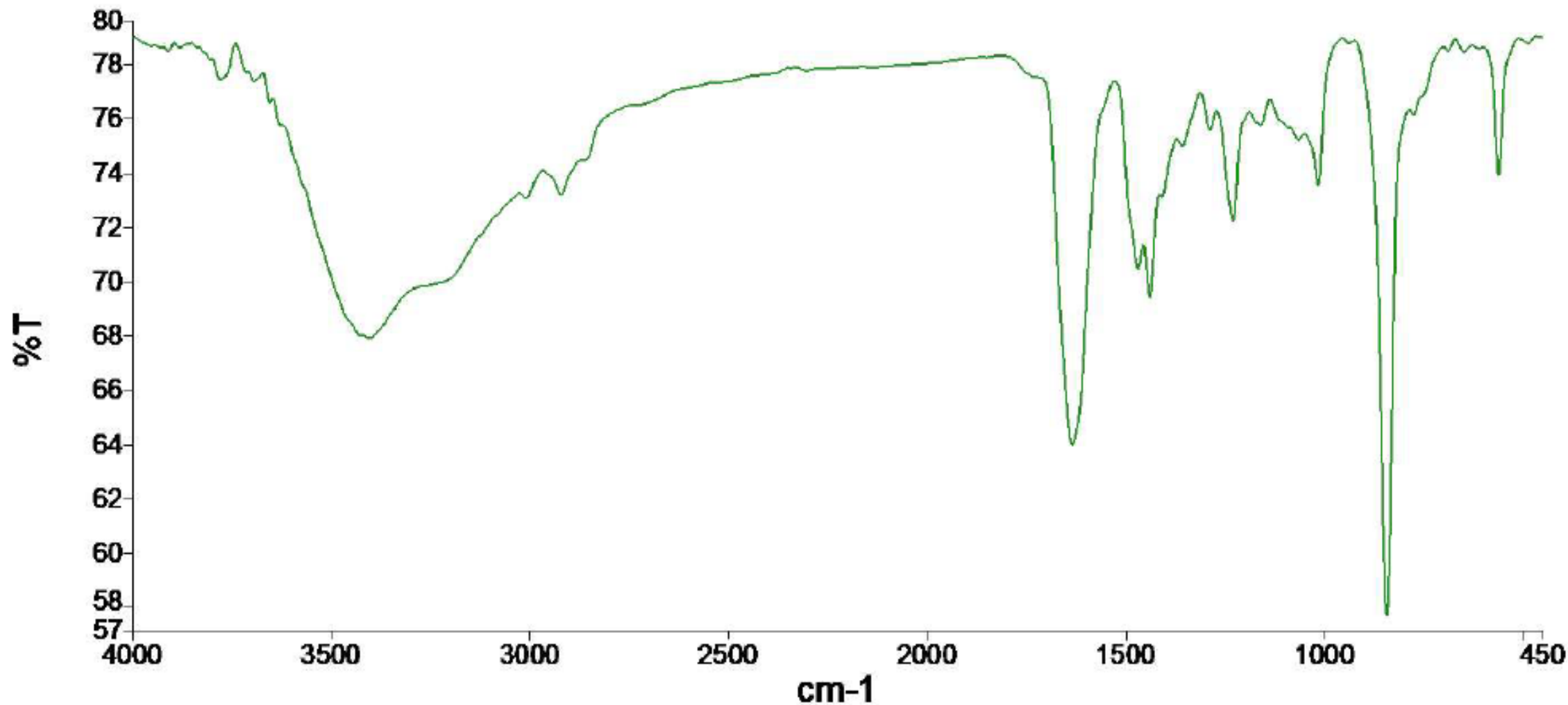


Figure S18. HRMS spectra of 4-(3-(4-(cyclopropanecarbonyl) piperazine-1-carbonyl)-4-fluorobenzyl) phthalazin-1(2H)-one (1)

Spectrum



Name

Description

———— DHARMESH 97_1_1 Sample IR22I12DEC01 OLA By DHARMESH Date Monday, December 12 2022 S29

Figure S19. IR spectra of 4-(3-(4-(cyclopropanecarbonyl) piperazine-1-carbonyl)-4-fluorobenzyl) phthalazin-1(2H)-one (1)

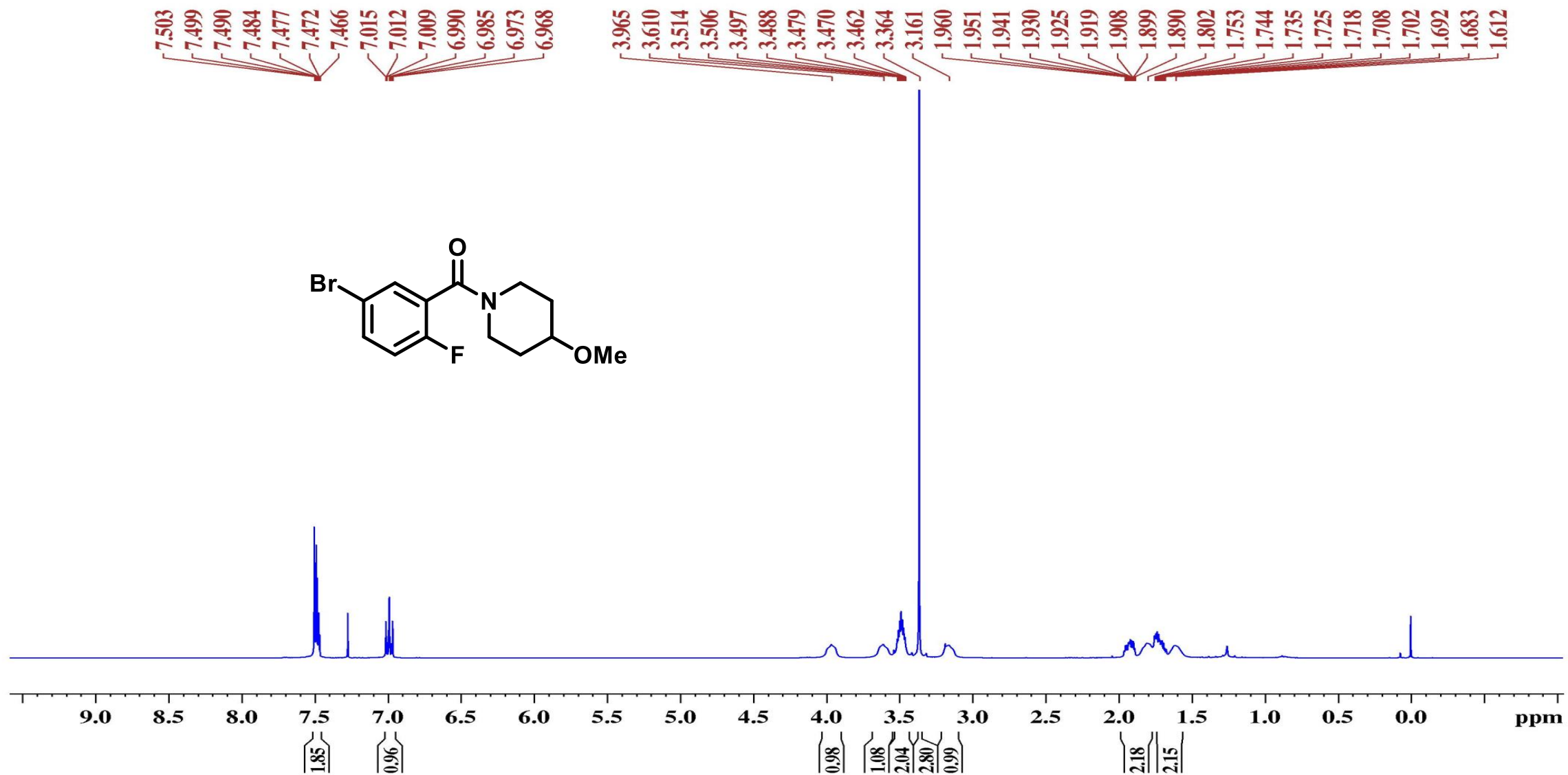


Figure S20. ¹H NMR spectra of (5-bromo-2-fluorophenyl)(4-methoxypiperidin-1-yl)methanone (21)

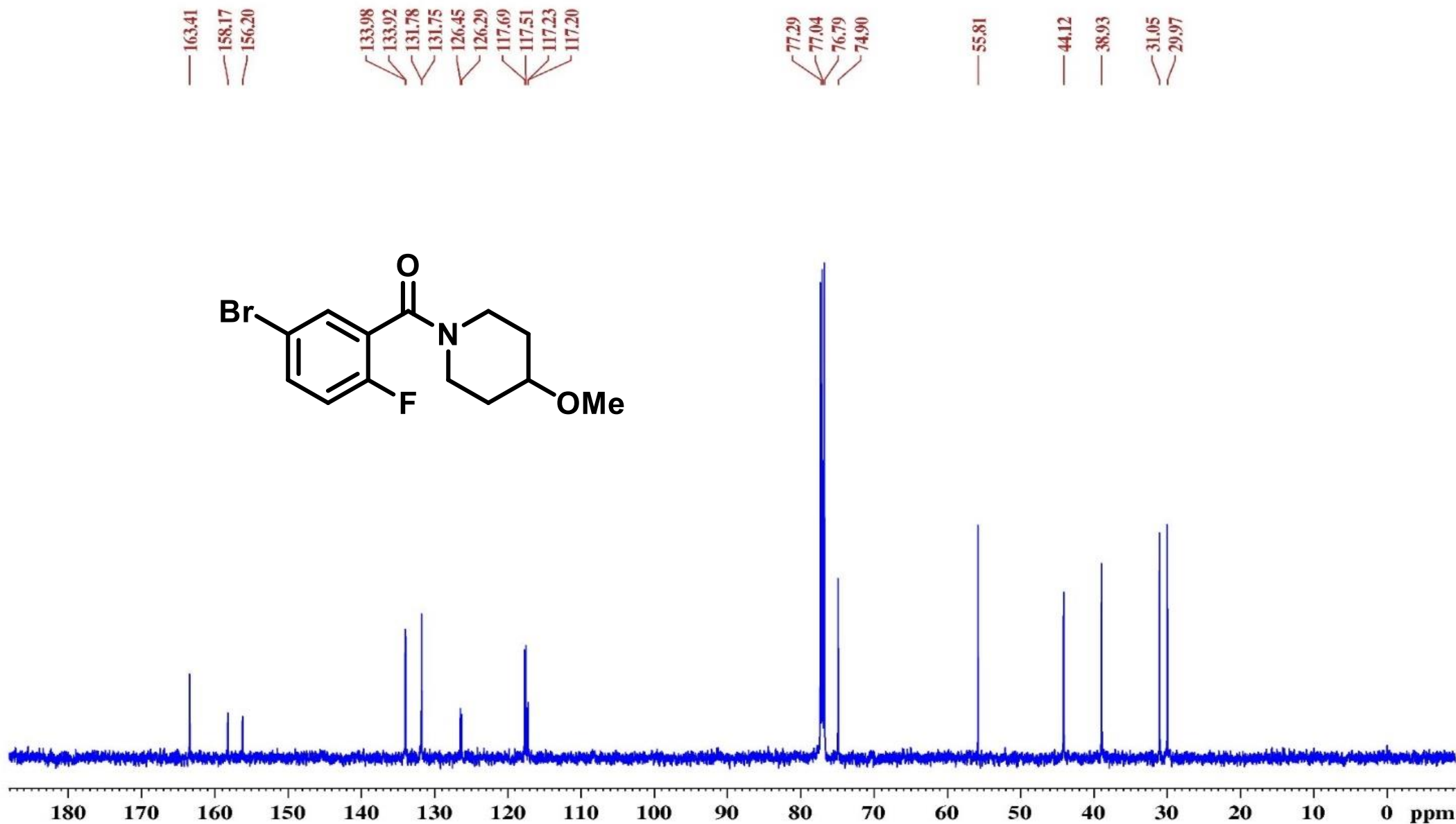


Figure S21. $^{13}\text{C}\{^1\text{H}\}$ NMR spectra of (5-bromo-2-fluorophenyl)(4-methoxypiperidin-1-yl)methanone (21) S31

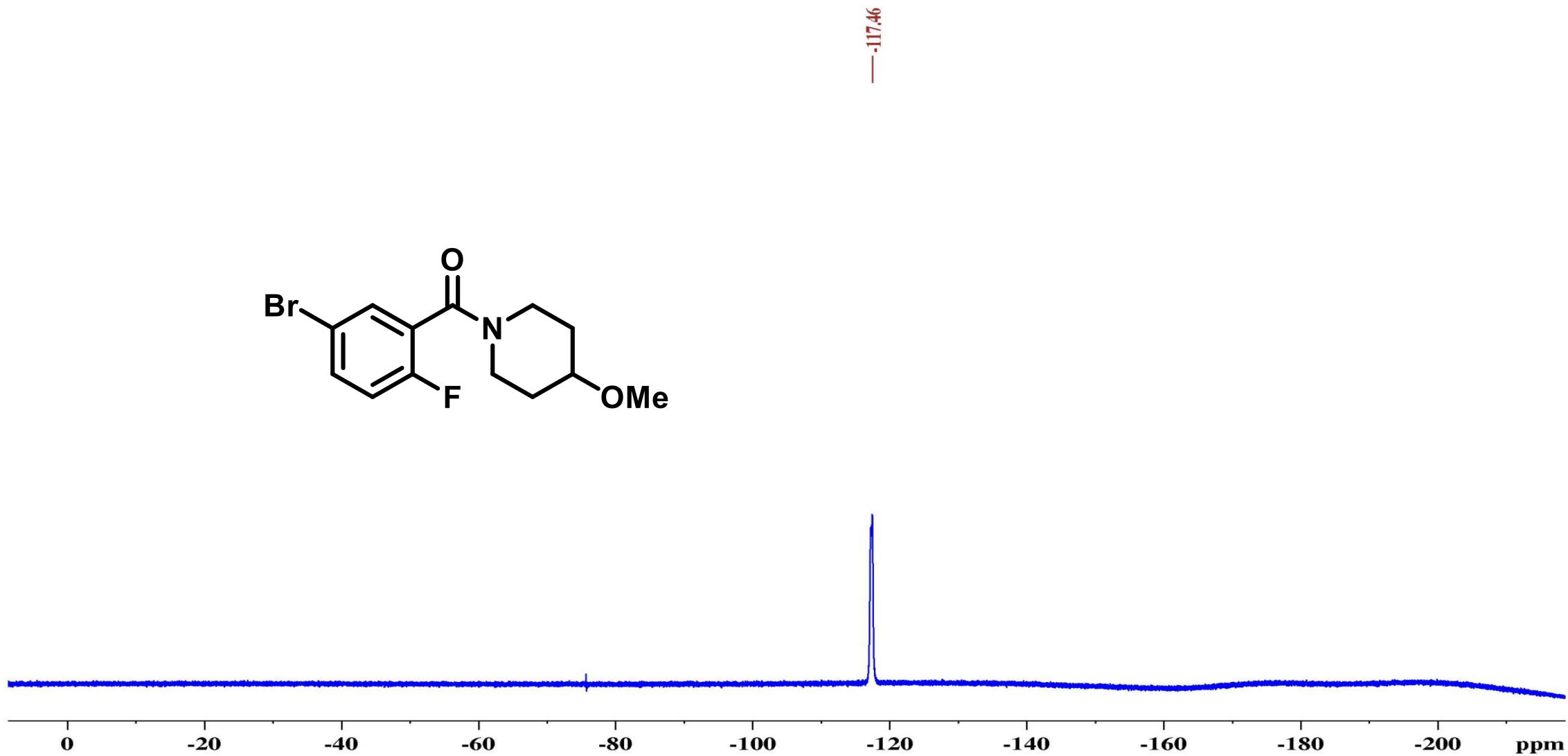


Figure S22. ^{19}F NMR spectra of (5-bromo-2-fluorophenyl)(4-methoxypiperidin-1-yl)methanone (21)

HRMS22I4NOV03 #31 RT: 0.24 AV: 1 NL: 3.44E6T:
FTMS + c ESI Full ms [100.00-750.00]

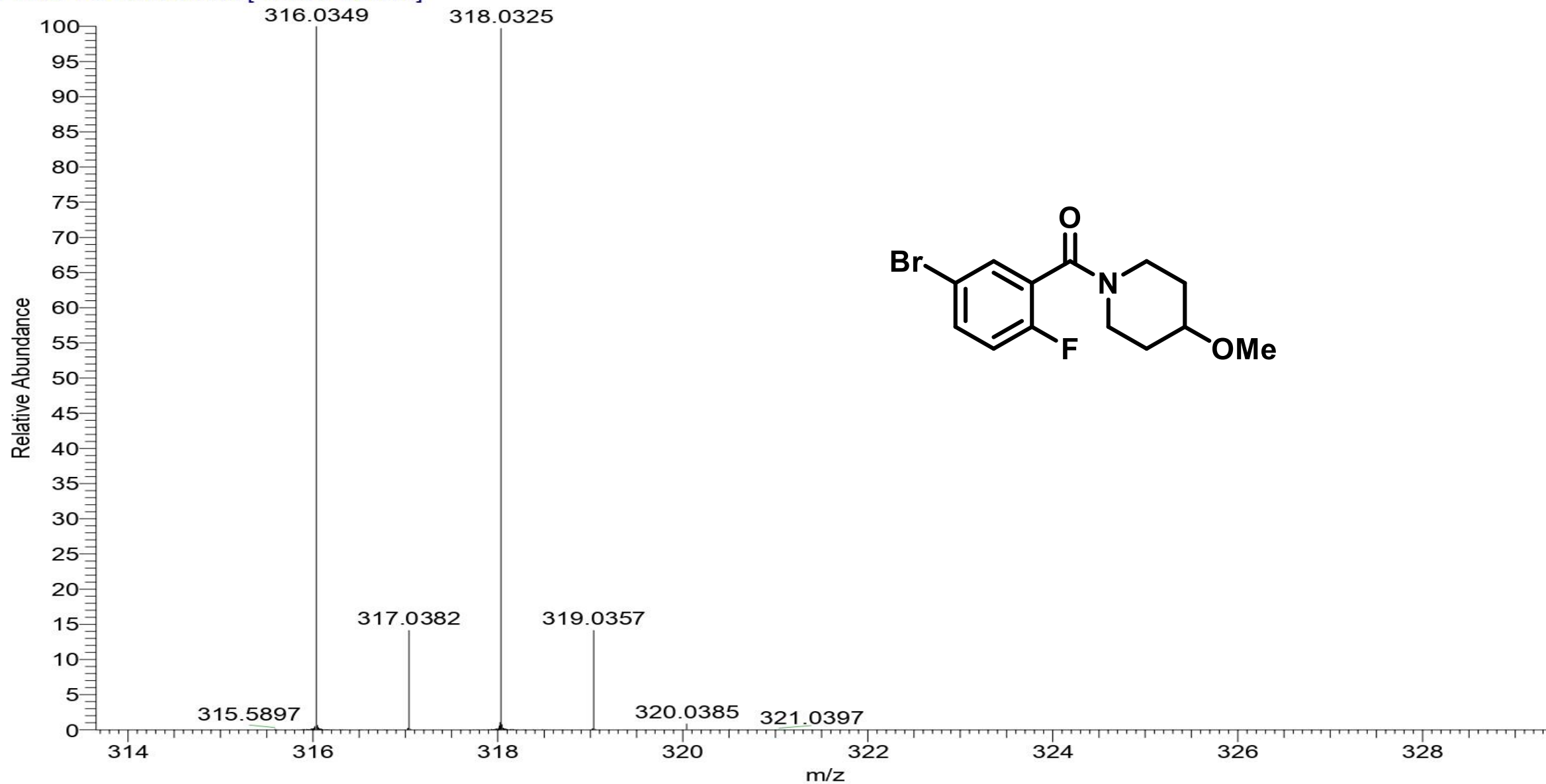


Figure S23. HRMS spectra of (5-bromo-2-fluorophenyl)(4-methoxypiperidin-1-yl)methanone (21)

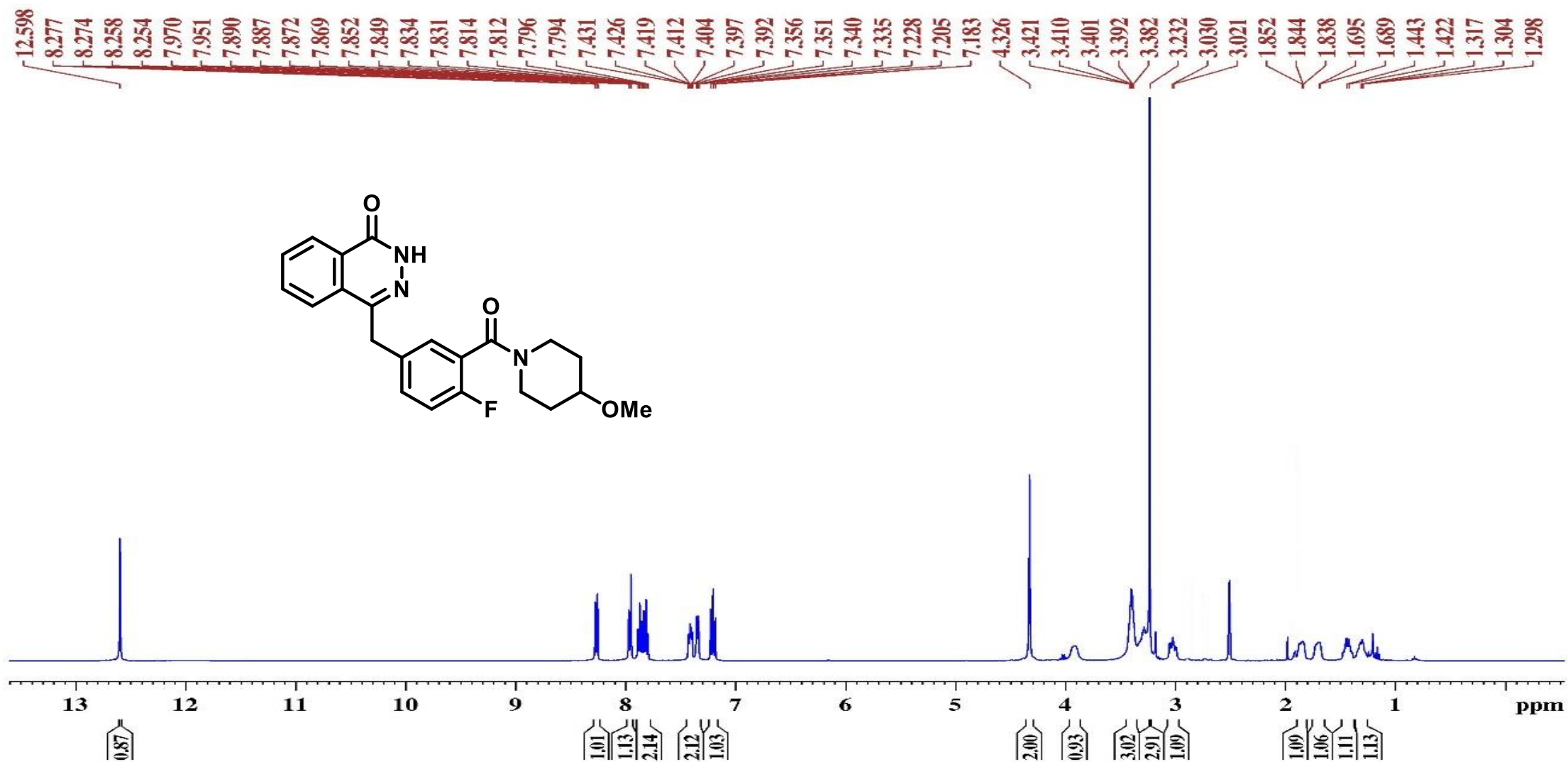


Figure S24. ¹H NMR spectra of 4-(4-fluoro-3-(4-methoxypiperidine-1-carbonyl) benzyl) phthalazin-1(2H)-one (19)

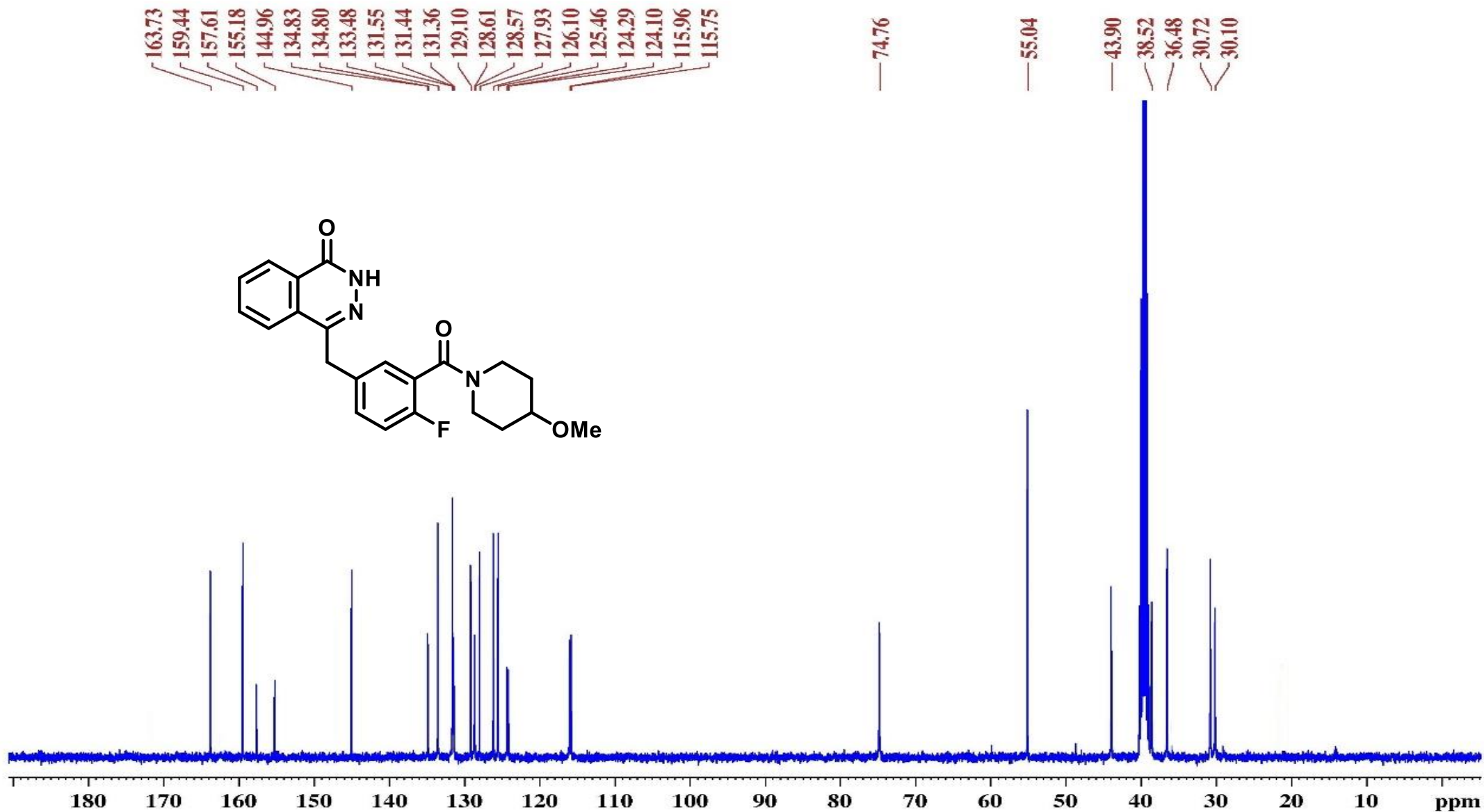


Figure S25. $^{13}\text{C}\{^1\text{H}\}$ NMR spectra of 4-(4-fluoro-3-(4-methoxypiperidine-1-carbonyl) benzyl) phthalazin-1(2H)-one (19)

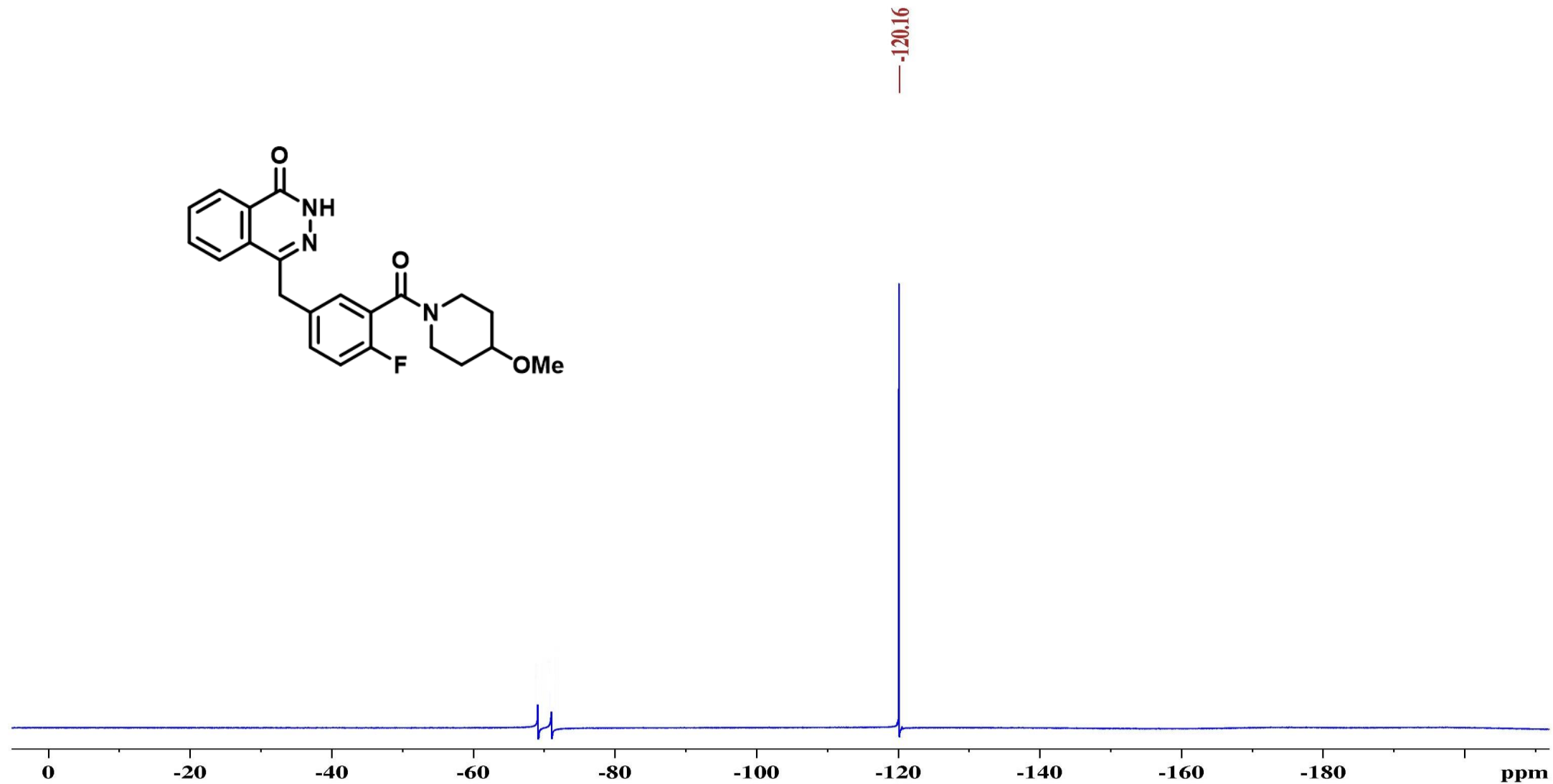


Figure S26. ^{19}F NMR spectra of 4-(4-fluoro-3-(4-methoxypiperidine-1-carbonyl) benzyl) phthalazin-1(2H)-one (19)

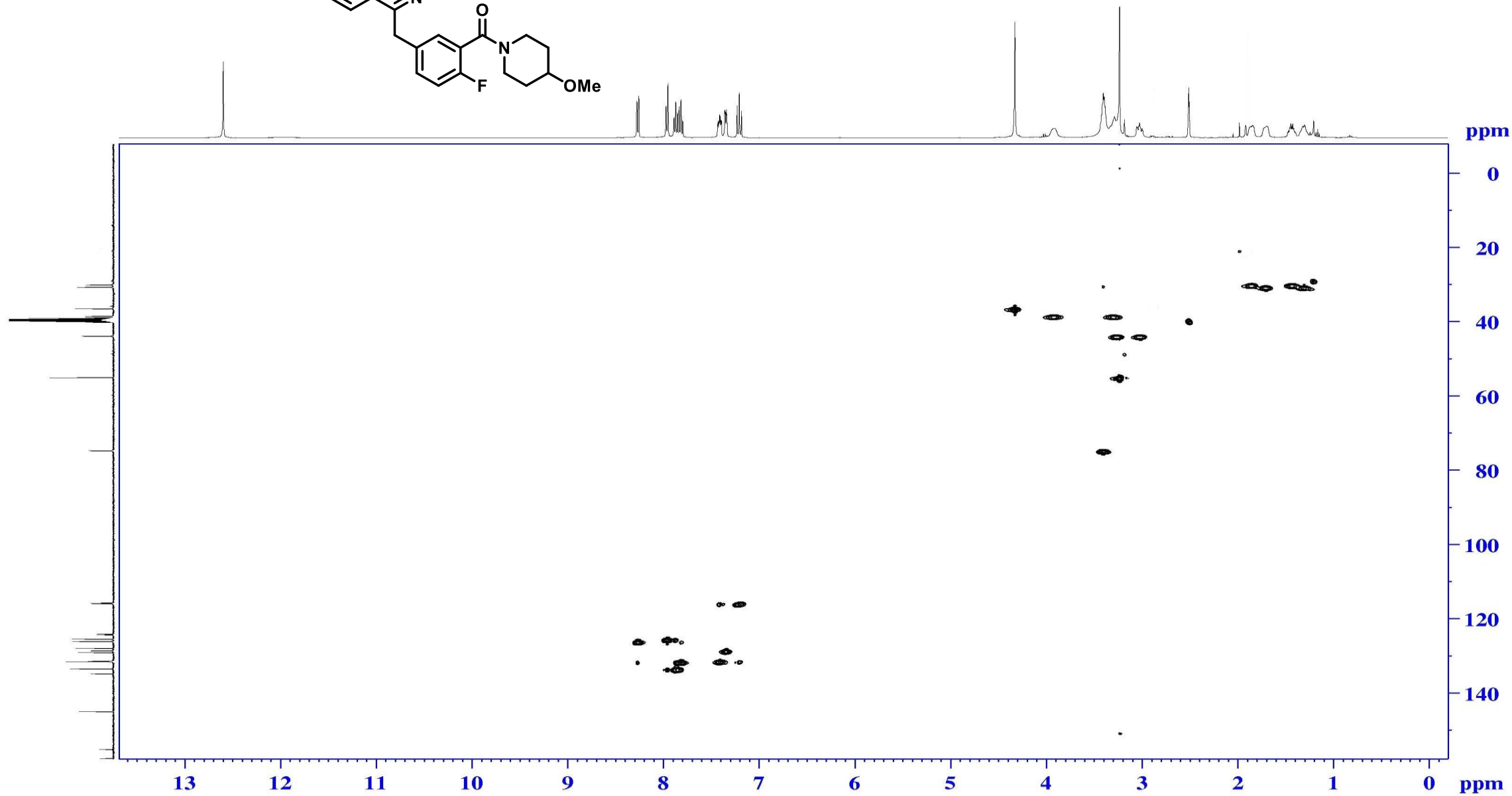
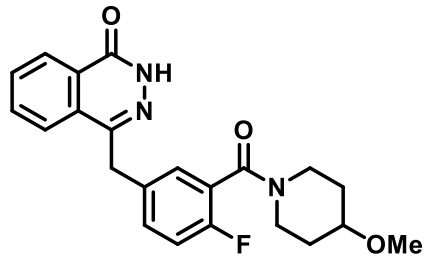


Figure S27. HSQC spectra of 4-(4-fluoro-3-(4-methoxypiperidine-1-carbonyl) benzyl) phthalazin-1(2H)-one (19) S37

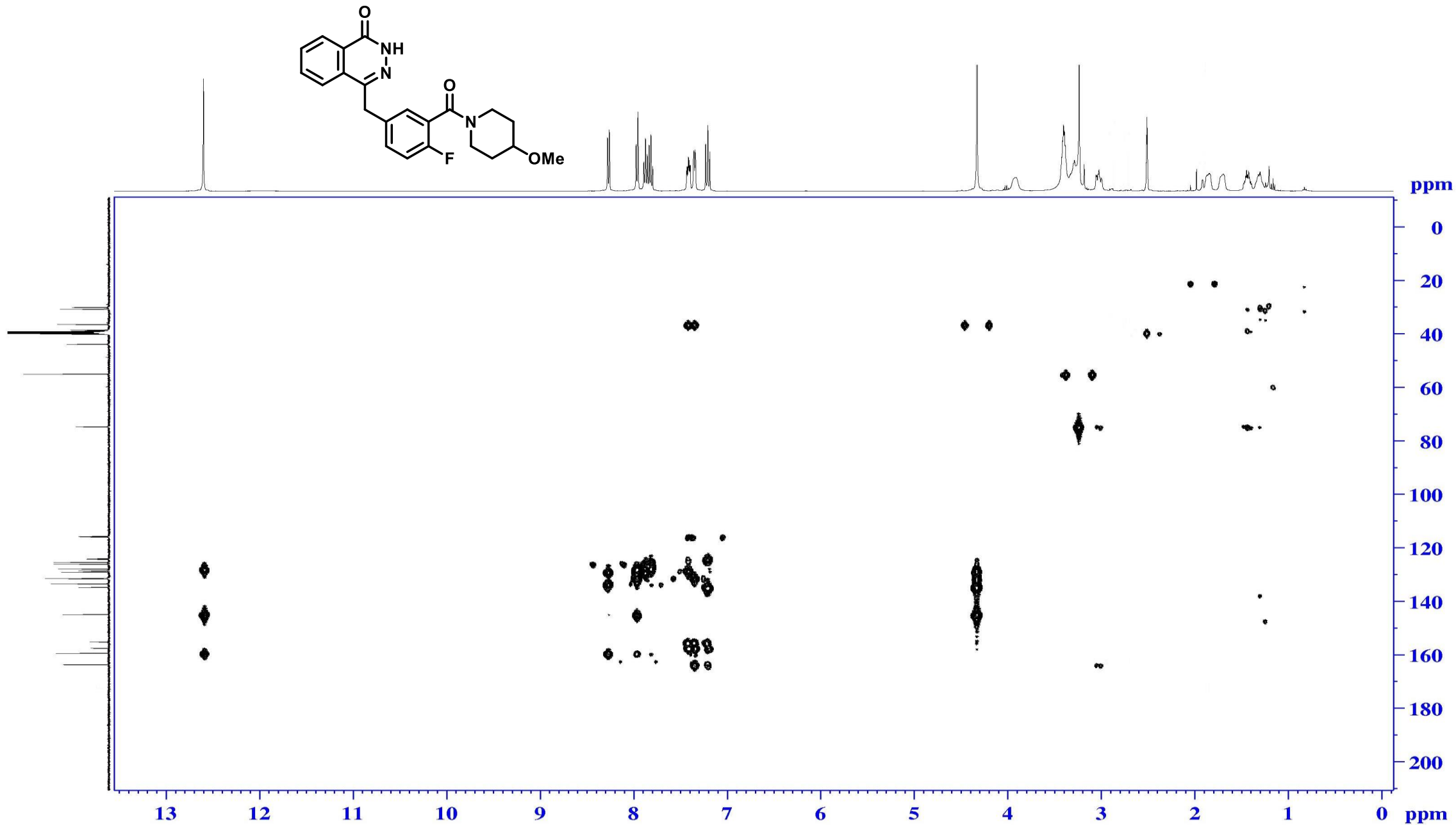


Figure S28. HMBC spectra of 4-(4-fluoro-3-(4-methoxypiperidine-1-carbonyl) benzyl) phthalazin-1(2H)-one (19) 538

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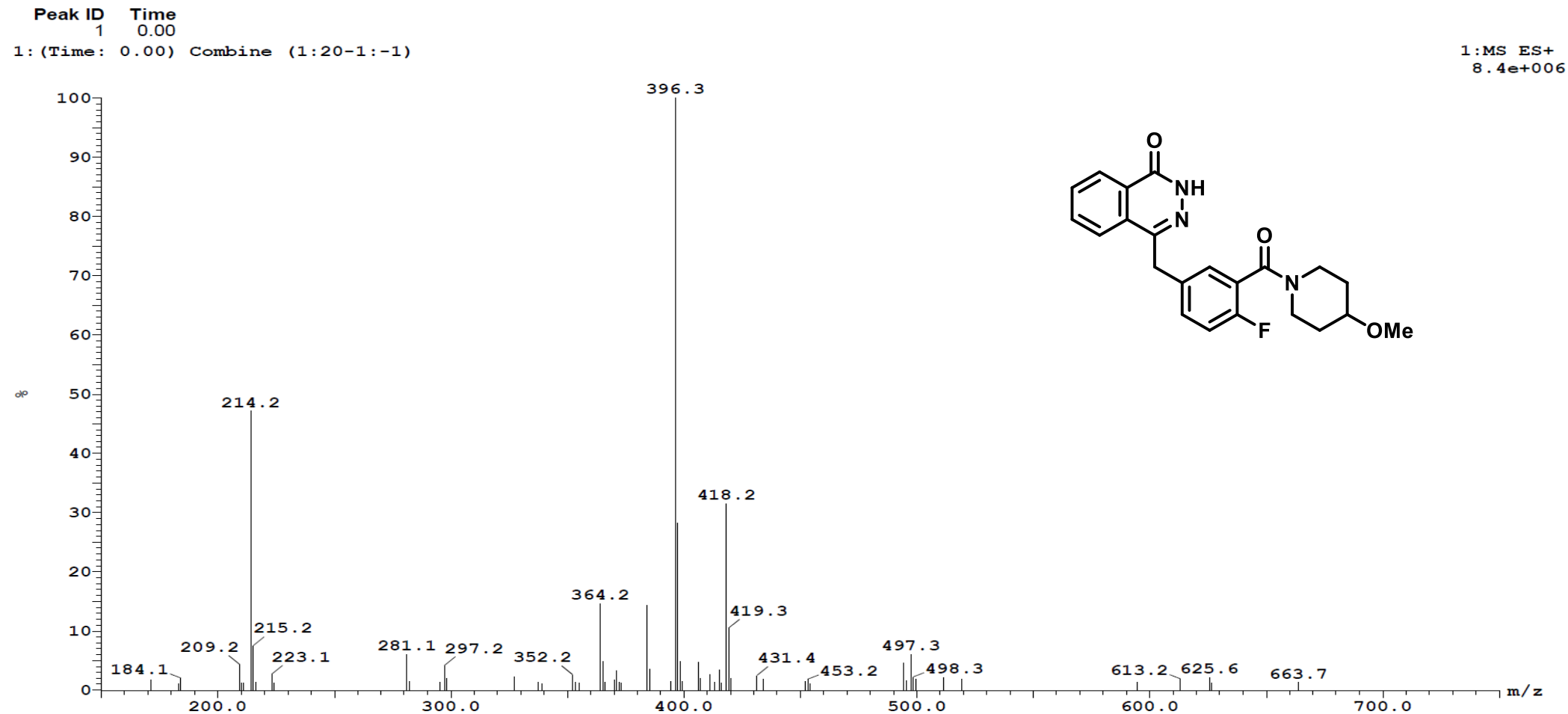


Figure S29. ESMS spectra of 4-(4-fluoro-3-(4-methoxypiperidine-1-carbonyl) benzyl) phthalazin-1(2H)-one (19)

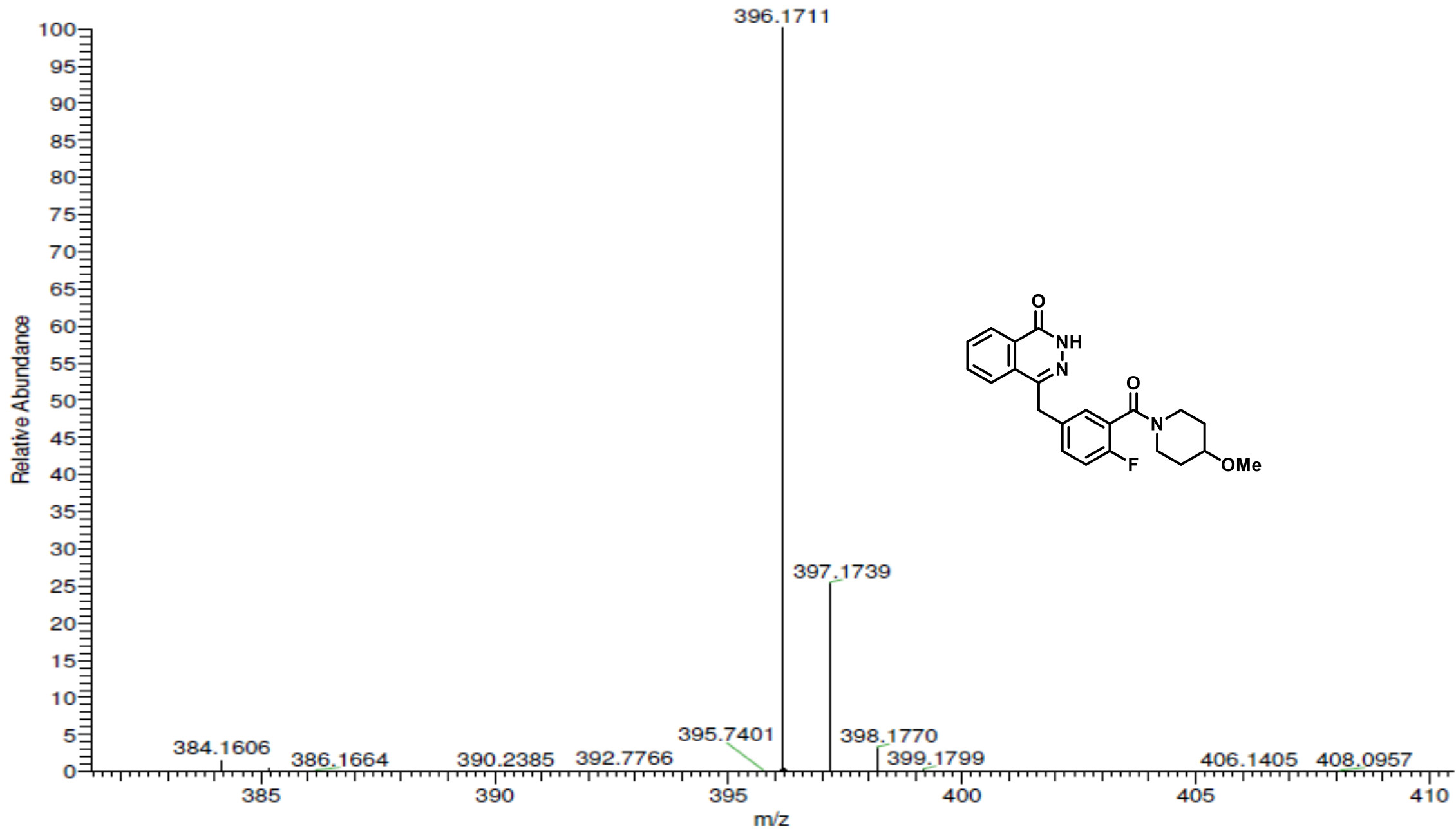
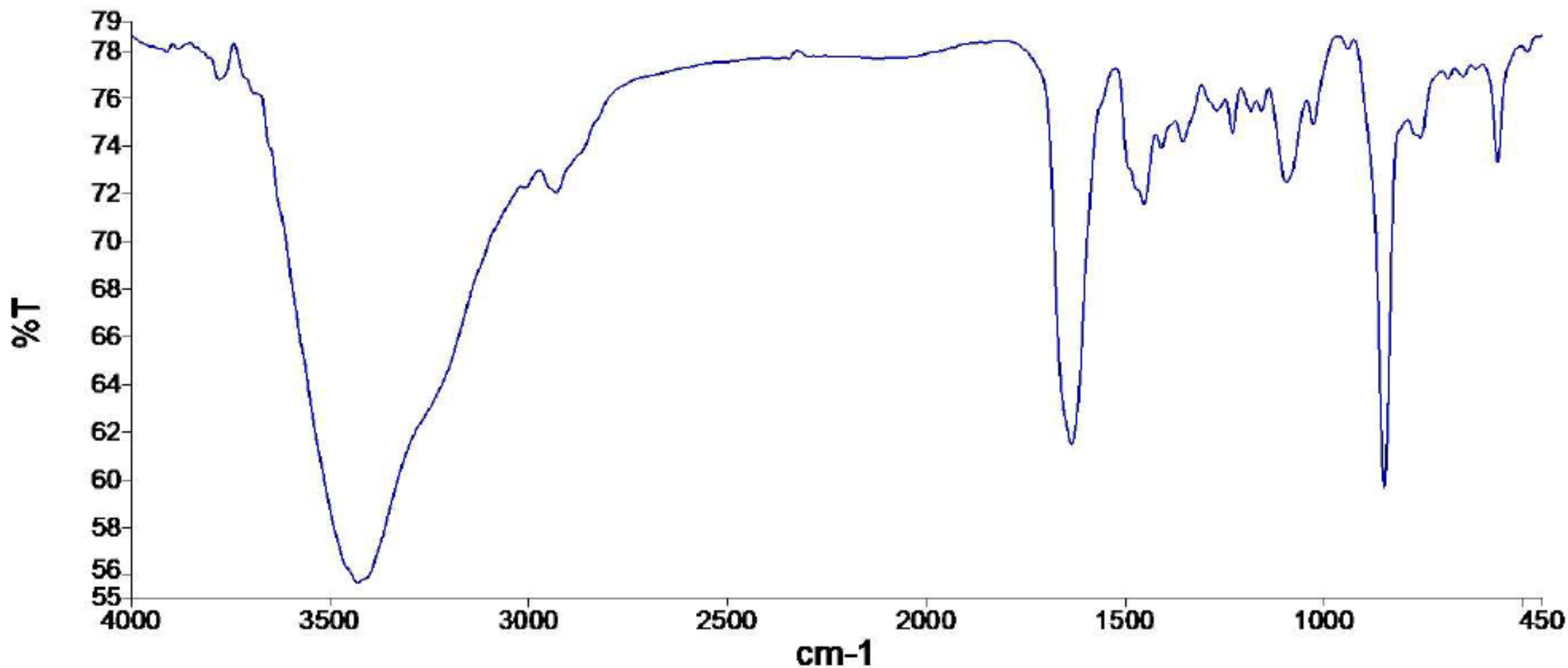


Figure S30. HRMS spectra of 4-(4-fluoro-3-(4-methoxypiperidine-1-carbonyl) benzyl) phthalazin-1(2H)-one (19)

Spectrum



Name

Description

DHARMESH 98_1_1 Sample IR22I12DEC02 INDR-201 By DHARMESH Date Monday, December 12 202

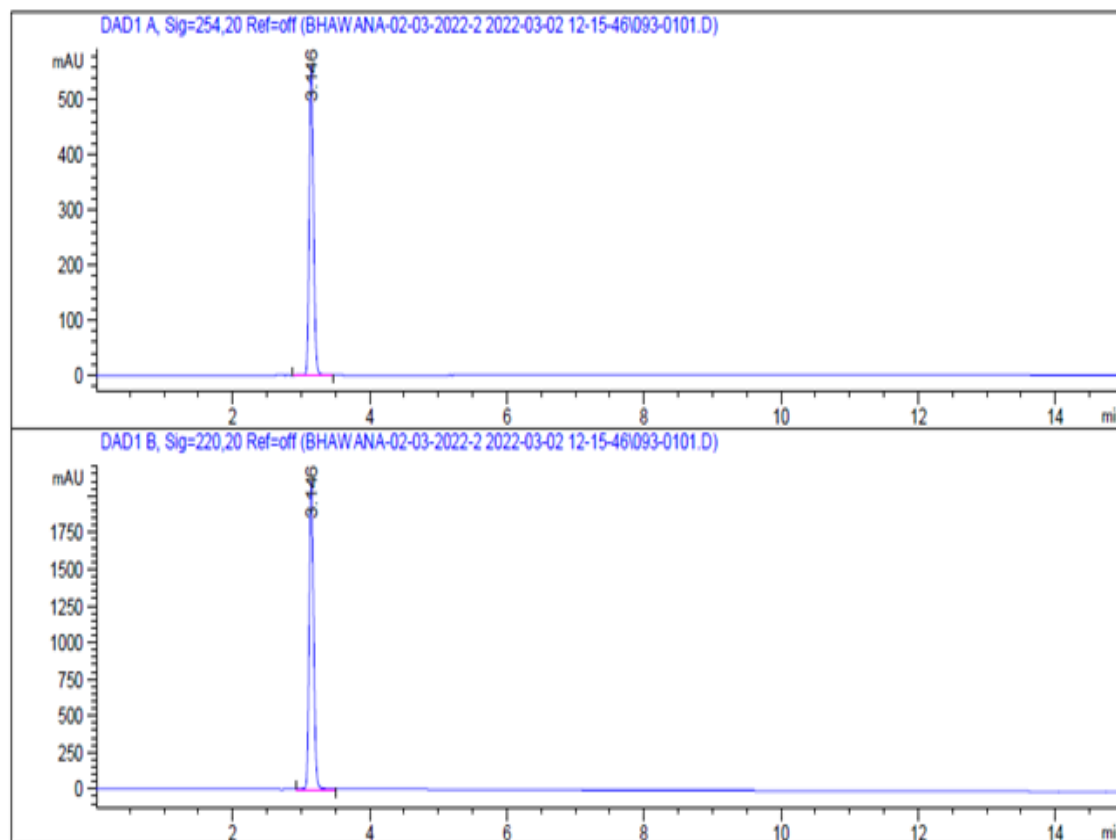
Figure S31. IR spectra of 4-(4-fluoro-3-(4-methoxypiperidine-1-carbonyl) benzyl) phthalazin-1(2H)-one (19)

8. Analytical information of compounds.

```

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Acq. Instrument : Instrument 1              Location  : Vial 93
Injection Date  : 3/2/2022 12:17:39 PM      Inj       :    1
                                           Inj Volume: 5.000 µl

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Analysis Method : C:\CHEM32\1\METHODS\ACN-WATER--GRAD.M
Last changed    : 2/28/2022 4:11:07 PM by CBR5-REPOSITORY
Method Info     : OSDD
  
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Totals :				2377.38477	564.30432	

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Peak #	RetTime [min]	Type	Width [min]	Area [mAU*s]	Height [mAU]	Area %
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Totals :				9319.29199	2104.14697	

System: Waters e2695-2998 Series Mobile phase: Mobile

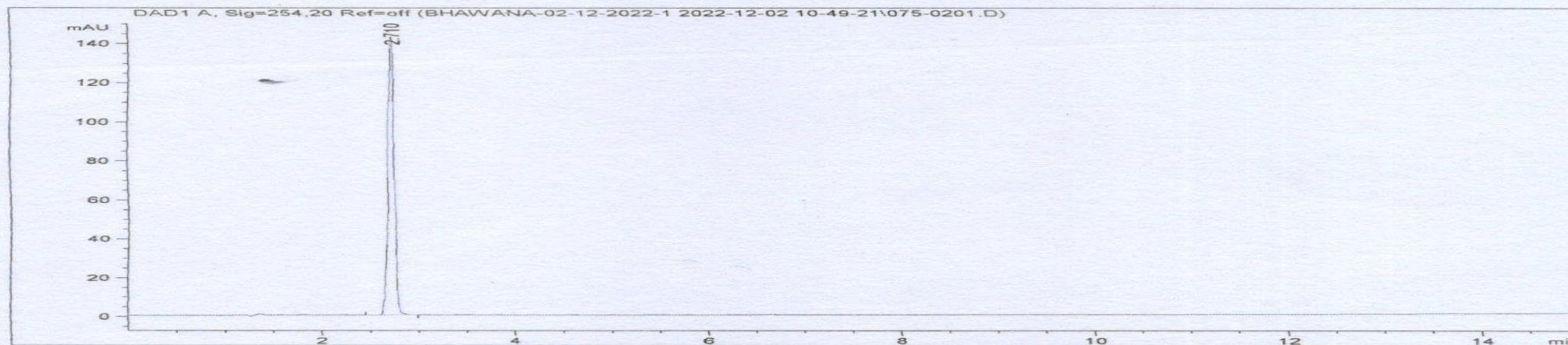
phase A: H₂O, Mobile phase B: MeCN

Detector: UV at 254 nm

Column: Thermo C-18 4.6 ×250 mm, 5µ

Column Temperature: 25°C


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Acq. Operator   : CBRS-REPOSITORY                      Seq. Line :    2
Acq. Instrument : Instrument 1                          Location  : Vial 75
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                                                    Inj Volume: 5.000 µl
Acq. Method     : C:\CHEM32\1\DATA\BHAWANA-02-12-2022-1 2022-12-02 10-49-21\ACN-WATER-40-60-
                  150MMGRAD.M
Last changed    : 11/28/2022 1:24:40 PM by CBRS-REPOSITORY
Analysis Method : C:\CHEM32\1\METHODS\ACN-WATER-40-60-150MMGRAD.M
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=====
Area Percent Report
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Sorted By      : Signal
Multiplier     : 1.0000
Dilution       : 1.0000
Use Multiplier & Dilution Factor with ISTDs
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Totals : 664.58521 142.96146

Figure S33. HPLC spectra of 4-(4-fluoro-3-(4-methoxypiperidine-1-carbonyl) benzyl) phthalazin-1(2H)-one (19) S43

9. Biological evaluation of Olaparib (1)

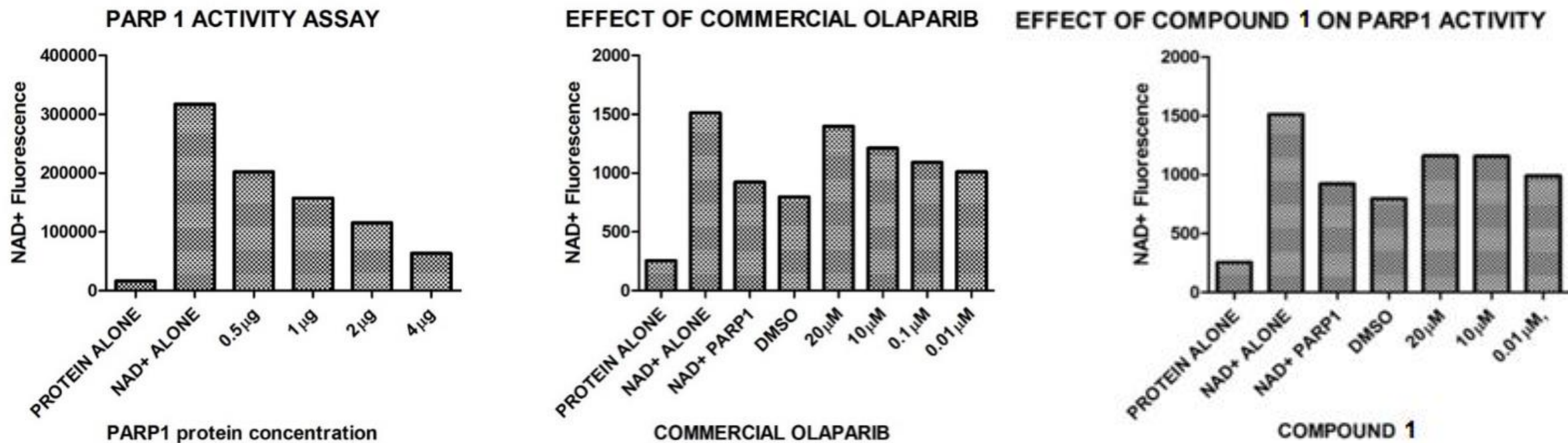


Figure S34. (A) PARP1 activity was measured at different concentrations and concentration dependent activity was observed. (B) The fluorescence intensity of NAD⁺ (substrate of PARP1) was measured in each reaction in the presence and absence of inhibitors. In Fig B, lane 4, NAD⁺ intensity was lower than lane 3 as it was used up by PARP1 for its activity. In lanes 5-7, the concentration of NAD⁺ was higher in the presence of Olaparib, indication inhibition of PARP1 activity. (C) Our synthesized compound 1 showed more or less similar activity with commercially sourced olaparib.