

Methionine and its hydroxy analogues: the paths toward their sustainable chemical synthesis †

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SUPPORTING INFORMATION

Table S1: Detailed quantitative sustainability metrics for the synthesis of D,L-Met and HMTBA produced by industrial (1-3) and renewable (4-6) synthetic routes.

Metric	Yield (%)	AE (%)	RME (%)	OE (%)	PMI (g ·g ⁻¹)	PMI RRC (g ·g ⁻¹)	PMI Solv (g ·g ⁻¹)	PMI WU (g ·g ⁻¹)	RI (g ·g ⁻¹)	RP (%)	
Process	Step										
(1) Propene to D,L-Met (through hydantoin)		81.0	40.7	17.9	43.9	31.0	6.7	20.7	3.6	28.3	91.1
	1	90.0	75.7	60.0	79.2	9.9	1.8	3.2	5.0		
	2	91.1	100	90.9	90.9	3.7	2.6	1.1	0		
	3	99.8	90.6	58.5	64.6	5.5	1.7	3.8	0		
	4	99.1	60.0	30.9	51.6	15.7	3.2	11.2	1.2		
	5	100	72.7	62.4	85.8	1.6	1.6	0	0		
(2) Propene to D,L-Met (through cyanohydrin)		67.7	60.8	16.6	27.3	31.4	7.2	19.9	4.3	28.8	91.8
	1	90.0	75.7	60.0	79.2	9.9	1.8	3.2	5.0		
	2	97.9	100	92.1	92.1	3.6	2.6	1.0	0		
	3	98.1	100	79.5	79.5	1.8	1.3	0.6	0		
	4	77.0	65.6	18.7	28.5	24.1	5.3	17.0	1.8		
(3) Propene to D,L-HMTBA		71.8	50.0	37.2	74.4	11.1	3.9	3.3	3.9	7.3	66.0
	1	90.0	75.7	60.0	79.2	9.9	1.8	3.2	5.0		
	2	91.1	100	90.9	90.9	3.7	2.6	1.1	0		
	3	93.2	100	90.3	90.3	2.1	1.1	1.0	0		
	4	94.0	53.2	45.3	85.2	3.7	2.2	0	1.5		
(4) DHA to D,L-HMTBA (through 2-HBL)		68.0	66.3	38.3	57.7	144.8	3.5	122.1	19.1	45.7	92.5
	1	68.0	85.0	38.5	45.3	53.8	3.9	49.8	0		
	2	100	72.0	65.4	90.8	108.9	1.5	88.2	19.1		
(5) Tetrose to D,L-HMTBA (through MVG)		38.1	42.7	19.0	44.4	68.0	6.9	59.6	1.5	65.2	95.9
	1	50.0	76.3	29.0	58.4	4.8	53.5	0	0		
	2	81.0	100	78.1	78.1	5.0	1.5	3.6	0		
	3	94.0	47.6	40.8	85.6	4.0	2.5	0	1.5		
(6) GA dimer to D,L-HMTBA		35.2	42.7	12.7	29.6	650.5	14.5	634.5	1.5	641	98.5
	1	37.5	82.0	17.6	21.5	556.7	11.4	545.3	0		
	2	94.0	47.6	40.8	85.6	4.0	2.5	0	1.5		

AE: Atom economy. RME: Reaction mass efficiency. OE: Optimum efficiency. PMI: process mass intensity. PMI RRC: PMI for reactants, reagents and catalysts. PMI Solv: PMI of solvents. PMI WU: PMI of species used in reaction work-ups. RI: Renewables intensity. RP: Renewables percentage.

