Supporting Information

Electrochemical Oxidative Dehydrogenation Aromatization of cyclohex-2-enone and Amine to 1,4-Phenylenediamine

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General information

All glassware was oven dried at 110 °C for hours and cooled down under vacuum. The instrument for electrolysis was dual display potentiostat (XINRUI® DJS-292B) (made in China). Chronoamperometry and cyclic voltammograms were obtained on a CorrTest[®] CS2350H bipotentiostat. GC yields were recorded by SHIMADZU™ GC-2014 gas chromatography. Molecular weights of products were determined by SHIMADZU™ GCMS-QP2010 SE gas chromatgraphy mass spectrometrometry. Online Electrochemical mass spectra data were acquired by LTQ Velos Orbitrap Elite mass spectrometer (Thermo Fisher Scientific Co., Ltd., USA). All new compounds were characterized by High resolution mass spectra (HRMS). All undivided cells were purchased from Jiehengda® limited liability company (https://www.whjiehengda.com). All the electrochemical oxidations were performed in an oven-dried two-necked undivided cell equipped with a carbon felt anode ($1.5 \times 1.5 \times 0.30 \text{ cm}^3$) and a Pt plate cathode ($1.5 \times 0.75 \times 0.10 \text{ cm}^3$) unless otherwise noted. Unless otherwise noted, materials and solvents were obtained from commercial suppliers and used without further purification. Thin layer chromatography (TLC) employed glass 0.25 mm silica gel plates. Visualization of spots on TLC plate was accomplished with UV light (254 nm). Flash chromatography columns were packed with 200-300 aluminium oxide neutral in petroleum (b.p. 60-90 °C). ¹H and ¹³C NMR data were recorded with Bruker Advance III (400 MHz) spectrometers with tetramethylsilane as an internal standard. All chemical shifts (δ) are reported in ppm and coupling constants (J) in Hz. All chemical shifts are reported relative to tetramethylsilane and d-solvent peaks (77.00 ppm, chloroform, 39.60 ppm, dimethylsulfoxide), respectively.

Experimental procedures for the electrolysis



Figure S1. Reaction equipment



General procedures for electrochemical oxidative dehydrogenation aromatization of cyclohex-2-enone with amine:



3a-3c, **3g-3h**, **3k-3m**, **3o-3q**, **3z-3aa** (condition A): In a 15 mL oven-dried two-necked undivided cell equipped with a stir bar, TEAI (77.2 mg, 0.3 mmol), *p*-CIPhCOOH (94.0 mg, 0.6 mmol) were added into the tube. The tube was equipped with carbon felt $(1.5 \times 1.5 \times 0.30 \text{ cm}^3)$ as the anode and Pt plate $(1.5 \times 0.75 \times 0.10 \text{ cm}^3)$ as the cathode and then charged with nitrogen to replace air atmosphere (purged three times). cyclohex-2-enone (0.30 mmol), **2** (1.80 mmol), MeCN (4.0 mL) and PhMe (1.0 mL) were added to the tube through a syringe. The mixtures were stirred at a constant current of 3.0 mA at 60 °C for 14 h ($J = 2.67 \text{ mA} \cdot \text{cm}^2$, 5.2 F·mol⁻¹). At the end of the reaction, the reaction was quenched by brine and extracted with EtOAc three times. The combined organic layer was dried over anhydrous Na₂SO₄ and evaporated in vacuum. The desired products were obtained in the corresponding yields after purification by flash chromatography on 200-300 aluminium oxide neutral (petroleum: EtOAc = 3:1).

3d-3f, 3i-3j, 3n, 3r-3y, 3ab (condition B): Otherwise above general procedure, *p*-CIPhCOOH (0.6 mmol) was replaced by TFA (0.6 mmol) and the mixtures were stirred at a constant current of 3.0 mA at 60 °C for 12 h ($J \approx 2.67$ mA·cm², 4.5 F·mol⁻¹).

3ac (condition C): Otherwise above general procedure, TEAI (0.3 mmol) was replaced by TBAI (110.8 mg, 0.3 mmol) and the mixtures were stirred at a constant current of 3.0 mA at 60 °C for 12 h ($J \approx 2.67$ mA cm², 4.5 F mol⁻¹).

3ad (condition D): Otherwise above general procedure, 2-cyclohexen-1-one (0.3 mmol) was replaced by 3-methyl-2-cyclohexen-1-one (0.3 mmol), morpholine (1.80 mmol), *p*-ClPhCOOH (0.6 mmol) was replaced by TFA (0.6 mmol) and the mixtures were stirred at a constant current of 3.0 mA at 60 °C for 14 h ($J \approx 2.67$ mA·cm², 5.2 F·mol⁻¹).

General procedures for electrochemical oxidative dehydrogenation aromatization of cyclohex-2-enones with morpholine:



In a 15 mL oven-dried two-necked undivided cell equipped with a stir bar, TEAI (77.2 mg, 0.3 mmol), *p*-CIPhCOOH (94.0 mg, 0.6 mmol) were added into the tube. The tube was equipped with carbon felt $(1.5 \times 1.5 \times 0.30 \text{ cm}^3)$ as the anode and Pt plate $(1.5 \times 0.75 \times 0.10 \text{ cm}^3)$ as the cathode and then charged with nitrogen to replace air atmosphere (purged three times). **1** (0.30 mmol), **2a** (1.80 mmol), MeCN (4.0 mL) and PhMe (1.0 mL) were added to the tube through a syringe. The mixtures were stirred at a constant current of 3.0 mA at 60 °C for 14 h ($J = 2.67 \text{ mA·cm}^2$, 5.2 F·mol⁻¹). At the end of the reaction, the reaction was quenched by brine and extracted with EtOAc three times. The combined organic layer was dried over anhydrous Na₂SO₄ and evaporated in vacuum. The desired products were obtained in the corresponding yields after purification by flash chromatography on 200-300 aluminium oxide neutral (petroleum: EtOAc = 3:1).

General procedure for the gram-scale reaction via anodic oxidation:



In a 100 mL three-necked flask oven-dried beaker equipped with a stir bar, Et₄NI (2.57 g, 10 mmol), *p*-Cl-PhCOOH (3.13 g, 20 mmol) were added into the tube. The tube was equipped with carbon felt ($3.0 \times 2.0 \times 0.30$ cm³) as the anode and Pt plate ($3.0 \times 1.5 \times 0.10$ cm³) as the cathode. **1a** (10 mmol), **2a** (60 mmol), MeCN (80 mL) and PhMe (20 mL) were added to the flask. The mixtures were stirred at a constant current of 20.0 mA at 60 °C for 70 h (J = 4.44 mA·cm², 5.2 F·mol⁻¹). At the end of the reaction, the reaction was quenched by brine and extracted with EtOAc three times. The combined organic layer was dried over anhydrous Na₂SO₄ and evaporated in vacuum. The desired products were obtained in the corresponding

yields after purification by flash chromatography on 200-300 aluminium oxide neutral (petroleum: EtOAc = 3:1). The white solid (1.56 g) was obtained with 63% isolated yield.

Table of additional oxidant experiment



Table S1. Different oxidants

Entry	Oxidants	Yield
1	$(NH_4)_2S_2O_8$	n.d.
2	DTBP	Trace
3	TBHP	Trace
4	O ₂ (1 atm)	10%

Reaction conditions: 1a (0.30 mmol), 2a (1.8 mmol), Et₄NI (0.3 mmol), p-ClPhCOOH (0.6 mmol), MeCN: PhMe=4:1, 5 mL in a

reaction tube.

Experimental of different current or reaction time experiment



Table S2. Different current or reaction time

Entry	x mA, y h	Yield ^a	Faradaic efficiency
1	2 mA, 21 h	75%	57%
2	3 mA, 10 h	54%	57%
3	3 mA, 14 h	71%	42%
4	3 mA, 18 h	56%	43%
5	6 mA, 7 h	47%	36%

Reaction conditions: **1a** (0.30 mmol), **2a** (1.8 mmol), Et₄NI (0.3 mmol), *p*-Cl-PhCOOH (0.6 mmol), MeCN: PhMe= 4:1, 5 mL in a reaction tube. ^{*a*}Yield was determined by ¹H NMR using 1,3, 5-trimethoxybenzene as the internal standard.

Additional discussion of online electrochemical mass spectra data



In a 15 mL oven-dried two-necked undivided cell equipped with a stir bar, HI (0.05 M in CH₃CN), PhCOOH (73.3 mg, 0.6 mmol) were added into the tube. **1a** (0.3 mmol), **2a** (1.2 mmol), MeCN (2 mL) were added to the flask. Stir at 60 °C for 3 h and then cool to room temperature. Samples were taken for electrochemical on-line mass spectrometry test.



Figure S2. Proposed mechanism





Figure S3. Monitoring during cyclohex-2-enone and morpholine. (a) High resolution tandem mass spectra of **4** (m/z 166.1224), (b) High resolution tandem mass spectra of **7** ($[M]^+$, m/z 251.1748), (c) High resolution tandem mass spectra of **8** ($[M+H]^+$, m/z 249.1592).

No.	Measured Mass	Calculated Mass	Δppm	Chemical Formula	Fragment ion Molecular Formula
1	166.1224	166.1226	-1.2	$C_{10}H_{16}NO^+$	
2	122.0963	122.0964	-0.8	C ₈ H ₁₂ N ^{2+•}	$C_2H_4O^2$
3	94.0648	94.0651	-3.2	C ₆ H ₈ N ^{2+•}	C ₄ H ₈ O ²
4	251.1748	251.1754	-2.4	$C_{14}H_{23}N_2O_2^+$	
5	223.1437	223.1441	-1.8	C ₁₂ H ₁₉ N ₂ O ₂ ²	+• $C_2H_4^2$ •
6	207.1490	207.1492	-1.0	C ₁₂ H ₁₉ N ₂ O ²⁺	• C ₂ H ₄ O ² •
7	249.1592	249.1598	-2.4	$C_{14}H_{21}N_2O_2^+$	
8	205.1333	205.1335	-1.0	C ₁₂ H ₁₇ N ₂ O ²⁺	• C ₂ H ₄ O ² •
9	177.1021	177.1022	-0.6	C ₁₀ H ₁₃ N ₂ O ²⁺	• C ₄ H ₈ O ² •

Table S3. Fragmentation ions from 4, 7, 8.

Verification experiments of intermediate 5



In a 15 mL oven-dried two-necked undivided cell equipped with a stir bar, "Et₄NI (77.2 mg, 0.3 mmol), 4-chlorobenzoic acid (94.0 mg, 0.6 mmol) were added into the tube. The tube was equipped with carbon felt $(1.5 \times 1.5 \times 0.30 \text{ cm}^3)$ as the anode and Pt plate $(1.5 \times 0.75 \times 0.10 \text{ cm}^3)$ as the cathode and then charged with nitrogen to replace air atmosphere (purged three times). **1a** (0.30 mmol), **2a** (1.80 mmol), **4a** (1.80 mmol), MeCN (4.0 mL) and PhMe (1.0 mL) were added to the tube through a syringe. Then the mixtures were stirred at a constant current of 3.0 mA at 60 °C for 14 h ($J = 2.67 \text{ mA·cm}^2$, 5.2 F·mol⁻¹) and then cool to room temperature. Samples were taken for high resolution mass spectrometry test.



Figure S4. HRMS result

Electrochemical procedures for cyclic voltammetry

Cyclic voltammograms were obtained on a CHI 605E potentiostat at room temperature. "Et4NBF4 (217.1 mg, 1.0 mmol) was used as the supporting electrolyte, with glassy carbon as working electrode (\sim 7.0 mm²) and a platinum wire (1.0 cm, spiral wire) as counter electrode, the scan rate was 100 mV/s. The scan range is between 0 V and 2 V.

b:
$$0.5 \text{ mmol TEABF}_4$$
 Cyclic Voltammetry
1 mL PhMe, 4 mL CH₃CN r.t.

1a, 0.15 mmol

c:
$$0$$
 0.5 mmol TEABF_4 Cyclic Voltammetry
1 mL PhMe, 4 mL CH₃CN r.t.

2a, 0.3 mmol



Figure S5. Cyclic voltammograms for different substrates.





Figure S6. Cyclic voltammograms for the different concentration solution of 1a, 2a, and 4-chlorophenol with/without iodide

We further investigated the interaction between iodine and the intermediate under electrochemical conditions through cyclic voltammetry. In this CV graph, (d) exhibits a more pronounced peak current at 0.78 V compared to (c), suggesting that the addition of iodine may participate in and facilitate the oxidation process of the intermediate. Contrasting (d) with (e) and (f), the increase in peak current at 0.78 V implies that the species undergoing oxidation at the anode originate from intermediates formed after the reaction of pyrrolidine, 2-cyclohexen-1-one, and 4-chlorobenzoic acid. When compared to (d), (g) shows the disappearance of the oxidation peak at 0.78 V, indicating that I₂ can oxidize the reaction intermediate.

Detail descriptions for products

1,4-dimorpholinobenzene (3a)^[1]: white solid was obtained with 75% isolated yield. ¹H NMR (400 MHz, CDCl₃) δ 6.89 (s, 4H), 3.89 – 3.82 (m, 8H), 3.10 – 3.03 (m, 8H). ¹³C NMR (101 MHz, CDCl₃) δ 145.35, 117.20, 66.89, 50.31.

1,4-dithiomorpholinobenzene (3b): white solid was obtained with 57% isolated yield. ¹H NMR (400 MHz, CDCl₃) δ 6.87 (s, 4H), 3.72 – 3.15 (m, 8H), 2.89 – 2.48 (m, 8H). ¹³C NMR (101 MHz, CDCl₃) δ 146.13, 119.07, 53.19, 27.39. HRMS (ESI) calcd for C₁₄H₂₁N₂S₂ [M+H]⁺: 281.1141 found: 281.1137.

1,4-di(piperidin-1-yl)benzene (3c)^[2]: colorless oil was obtained with 39% isolated yield. ¹H NMR (400 MHz, CDCl₃) δ 6.89 (s, 4H), 3.24 – 2.80 (m, 8H), 1.87 – 1.64 (m, 8H), 1.58 – 1.36 (m, 4H). ¹³C NMR (101 MHz, CDCl₃) δ 146.25, 118.11, 51.94, 26.09, 24.21.

1,4-bis(4-chloropiperidin-1-yl)benzene (3d): white solid was obtained with 47% isolated yield. ¹H NMR (400 MHz, CDCl₃) δ 6.89 (s, 4H), 4.26 – 4.04 (m, 2H), 3.53 – 3.29 (m, 4H), 3.02 – 2.83 (m, 4H), 2.26 – 2.15 (m, 4H), 2.09 – 1.95 (m, 4H). ¹³C NMR (101 MHz, CDCl₃) δ 145.25, 118.20, 57.12, 48.62, 35.29. HRMS (ESI) calcd for C₁₆H₂₃Cl₂N₂ [M+H]⁺: 313.1233 found: 313.1223.

1,4-bis(4-fluoropiperidin-1-yl)benzene (3e): white solid was obtained with 38% isolated yield. ¹H NMR (400 MHz, CDCl₃) δ 6.91 (s, 4H), 5.00 – 4.79 (m, 1H), 4.77 – 4.54 (m, 1H), 3.42 – 3.18 (m, 4H), 3.10 – 2.94 (m, 4H), 2.24 – 1.80 (m, 8H). ¹³C NMR (101 MHz, CDCl₃) δ 145.37, 118.30, 89.21, 87.51, 47.14, 47.08, 31.47, 31.28. ¹⁹F NMR (377 MHz, CDCl₃) δ -180.90. HRMS (ESI) calcd for C₁₆H₂₃F₂N₂ [M+H]⁺: 281.1824 found: 281.1819.

1,4-bis(4-methoxypiperidin-1-yl)benzene (3f): white solid was obtained with 40% isolated yield. ¹H NMR (400 MHz, CDCl₃) δ 6.89 (s, 4H), 3.37 (m, 12H), 2.85 – 2.76 (m, 4H), 2.06 – 1.98 (m, 4H). ¹³C NMR (101 MHz, CDCl₃) δ 145.41, 118.12, 76.15, 55.52, 48.52, 30.86. HRMS (ESI) calcd for C₁₈H₂₉N₂O₂ [M+H]⁺: 305.2224 found: 305.2219.

1,4-bis(4-benzylpiperidin-1-yl)benzene (3g): white solid was obtained with 51% isolated yield. ¹H NMR (400 MHz, CDCl₃) δ 7.28 (t, J = 7.4 Hz, 4H), 7.21 – 7.13 (m, 6H), 6.86 (s, 4H), 3.54 – 3.27 (m, 4H), 2.72 – 2.36 (m, 8H), 1.73 (d, J = 12.9 Hz, 4H), 1.68 – 1.56 (m, 2H), 1.49 – 1.33 (m, 4H). ¹³C NMR (101 MHz, CDCl₃) δ 145.82, 140.55, 129.11, 128.15, 125.79, 118.10, 51.23, 43.17, 37.74, 32.26. HRMS (ESI) calcd for C₃₀H₃₇N₂ [M+H]⁺: 429.2951 found: 429.2938.

1,4-bis(4-phenoxypiperidin-1-yl)benzene (3h): white solid was obtained with 45% isolated yield. ¹H NMR (400 MHz, CDCl₃) δ 7.33 – 7.24 (m, 4H), 7.02 – 6.87 (m, 10H), 4.48 – 4.38 (m, 2H), 3.45 – 3.35 (m, 4H), 3.03 – 2.93 (m, 4H), 2.16 – 2.01 (m, 4H), 2.01 – 1.88 (m, 4H). ¹³C NMR (101 MHz, CDCl₃) δ 157.29, 145.40, 129.47, 120.80, 118.14, 116.08, 72.20, 47.95, 30.75. HRMS (ESI) calcd for C₂₈H₃₃N₂O₂ [M+H]⁺: 429.2537 found: 429.2531.

1,1'-(1,4-phenylene)bis(piperidin-4-ol) (3i): white solid was obtained with 53% isolated yield. ¹H NMR (400 MHz, DMSO-d₆) δ 6.80 (s, 4H), 4.64 (d, J = 4.2 Hz, 2H), 3.55 (td, J = 9.0, 4.4 Hz, 2H), 3.32 (dd, J = 11.6, 6.0 Hz, 4H), 2.72 – 2.61 (m, 4H), 1.85 – 1.74 (m, 4H), 1.54 – 1.40 (m, 4H).¹³C NMR (101 MHz, DMSO-d₆) δ 144.73, 117.46, 66.22, 47.98, 34.26. HRMS (ESI) calcd for C₁₆H₂₅N₂O₂ [M+H]⁺: 277.1911 found: 277.1911.

(**1,4-phenylenebis(piperidine-1,4-diyl))dimethanol (3j):** white solid was obtained with 51% isolated yield. ¹H NMR (400 MHz, DMSO-d₆) δ 6.81 (s, 4H), 4.47 (t, J = 5.3 Hz, 2H), 3.52 – 3.43 (m, 4H), 3.32 – 3.25 (m, 4H), 2.63 – 2.35 (m, 4H), 1.77 – 1.68 (m, 4H), 1.50 – 1.37 (m, 2H), 1.30 – 1.16 (m, 4H). ¹³C NMR (101 MHz, DMSO-d₆) δ 145.25, 117.57, 66.02, 50.23, 38.33, 28.79. HRMS (ESI) calcd for C18H29N2O2 [M+H]⁺: 305.2224 found: 305.2219.

dibenzyl 4,4'-(1,4-phenylene)bis(piperazine-1-carboxylate) (3k): white solid was obtained with 61% isolated yield. ¹H NMR (400 MHz, CDCl₃) δ 6.88 (s, 4H), 3.70 (s, 6H), 3.54 – 3.45 (m, 4H), 2.68 (t, *J* = 11.7 Hz, 4H), 2.40 (m, *J* = 10.9, 3.5 Hz, 2H), 2.06 – 1.98 (m, 4H), 1.95 – 1.83 (m, 4H) ¹³C NMR (101 MHz, CDCl₃) δ 175.34, 145.64, 118.17, 51.62, 50.35, 40.79, 28.23. HRMS (ESI) calcd for C₂₀H₂₉N₂O₄ [M+H]⁺: 361.2122 found: 361.2115.

1,4-bis(4-methylpiperidin-1-yl)benzene (31): white solid was obtained with 57% isolated yield. ¹H NMR (400 MHz, CDCl₃) δ 6.88 (s, 4H), 3.49 (m, J = 12.5, 3.4 Hz, 4H), 2.65 – 2.54 (m, 4H), 1.76 – 1.68 (m, 4H), 1.53 – 1.23 (m, 6H), 0.97 (d, J = 6.1 Hz, 6H). ¹³C NMR (101 MHz, CDCl₃) δ 145.89, 118.06, 51.25, 34.39, 30.57, 21.87. HRMS (ESI) calcd for

 $C_{20}H_{29}N_2O_4$ [M+H]⁺: 273.2325 found: 273.2322

1,4-bis(3-methylpiperidin-1-yl)benzene (3m): white solid was obtained with 53% isolated yield. ¹H NMR (400 MHz, CDCl₃) δ 6.90 (s, 4H), 3.50 – 3.37 (m, 4H), 2.58 – 2.47 (m, 2H), 2.26 – 2.17 (m, 2H), 1.04 – 0.95 (m, 2H), 0.93 (d, *J* = 6.4 Hz, 6H). ¹³C NMR (101 MHz, CDCl₃) δ 145.98, 118.17, 76.68, 59.26, 51.34, 32.81, 31.16, 25.62, 19.61. HRMS (ESI) calcd for C₁₈H₂₉N₂ [M+H]⁺: 273.2325 found: 273.2319.



dimethyl 1,1'-(1,4-phenylene)bis(piperidine-3-carboxylate) (3n): white solid was obtained with 44% isolated yield. ¹H NMR (400 MHz, CDCl₃) δ 6.90 (s, 4H), 3.71 (s, 6H), 3.62 – 3.54 (m, 2H), 3.35 (d, *J* = 11.9 Hz, 2H), 2.91 (t, *J* = 10.8 Hz, 2H), 2.76 – 2.66 (m, 4H), 2.06 – 1.93 (m, 2H), 1.87 – 1.77 (m, 2H), 1.74 – 1.55 (m, 4H). ¹³C NMR (101 MHz, CDCl₃) δ 174.43, 145.69, 118.46, 53.34, 51.69, 51.06, 41.59, 26.75, 24.40. HRMS (ESI) calcd for C₂₀H₂₉N₂O₄ [M+H]⁺: 361.2122 found: 361.2118.

dibenzyl 4,4'-(1,4-phenylene)bis(piperazine-1-carboxylate) (30): white solid was obtained with 76% isolated yield. ¹H NMR (400 MHz, CDCl₃) δ 7.46 – 7.08 (m, 10H), 6.88 (s, 4H), 5.15 (s, 4H), 3.79 – 3.31 (m, 8H), 3.02 (s, 8H). ¹³C NMR (101 MHz, CDCl₃) δ 155.03, 145.46, 136.48, 128.37, 128.25, 127.92, 127.78, 127.78, 127.71, 118.15, 67.05, 50.24, 43.70. HRMS (ESI) calcd for C₃₀H₃₅N₄O₄ [M+H]⁺: 515.2653 found: 515.2639.

di-tert-butyl 4,4'-(1,4-phenylene)bis(piperazine-1-carboxylate) (3p): white solid was obtained with 55% isolated yield. ¹H NMR (400 MHz, CDCl₃) δ 6.90 (s, 4H), 3.57 (t, *J* = 5.1 Hz, 8H), 3.03 (d, *J* = 5.1 Hz, 8H), 1.48 (s, 9H). ¹³C NMR (101 MHz, CDCl₃) δ 154.68, 145.61, 118.18, 81.34, 50.40, 45.31, 29.86. HRMS (ESI) calcd for C₂₄H₃₉N₄O₄ [M+H]⁺: 447.2966 found: 447.2959.

1,4-bis(4-(tert-butyl)piperazin-1-yl)benzene (3q): white solid was obtained with 47% isolated yield. ¹H NMR (400 MHz, CDCl₃) δ 6.89 (s, 4H), 3.11 (t, *J* = 4.8 Hz, 8H), 2.74 (t, *J* = 4.9 Hz, 8H), 1.11 (s, 18H). ¹³C NMR (101 MHz, CDCl₃) δ 145.27, 117.49, 53.68, 50.89, 45.77, 25.79. HRMS (ESI) calcd for C₂₂H₃₈N₄ [M+H]⁺: 359.3169 found: 359.3168.



N1,N4-dibenzyl-N1,N4-dimethylbenzene-1,4-diamine (3r): white solid was obtained with 47% isolated yield. ¹H NMR (400 MHz, CDCl₃) δ 7.35 – 7.15 (m, 10H), 6.75 (s, 4H), 4.36 (s, 4H), 2.85 (s, 6H). 13C NMR (101 MHz, CDCl3) δ 142.84, 139.49, 128.35, 127.23, 126.71, 115.07, 58.27, 38.99. HRMS (ESI) calcd for C₂₂H₂₅N₂ [M+H]⁺: 317.2012 found: 317.2002.



N1,N4-dimethyl-N1,N4-bis(4-methylbenzyl)benzene-1,4-diamine (3s): white solid was obtained with 45% isolated yield. ¹H NMR (400 MHz, CDCl₃) δ 7.16 – 7.06 (m, 8H), 6.75 (s, 4H), 4.32 (s, 6H), 2.83 (s, 6H), 2.31 (s, 4H). ¹³C NMR (101 MHz, CDCl₃) δ 142.88, 136.24, 129.02, 127.24, 115.13, 58.02, 38.88, 21.04. HRMS (ESI) calcd for C₂₄H₂₉N₂ [M+H]⁺: 345.2325 found: 345.2317.



N1,N4-bis(4-methoxybenzyl)-N1,N4-dimethylbenzene-1,4-diamine (3t): white solid was obtained with 45% isolated yield. ¹H NMR (400 MHz, CDCl₃) δ 7.21 – 7.13 (m, 4H), 6.87 – 6.79 (m, 4H), 6.77 (s, 4H), 4.30 (s, 4H), 3.77 (s, 6H), 2.82 (s, 6H). ¹³C NMR (101 MHz, Chloroform-*d*) δ 158.44, 142.96, 131.32, 128.49, 115.33, 113.69, 57.75, 55.18, 38.80. HRMS (ESI) calcd for C₂₄H₂₉N₂O₂ [M+H]⁺: 371.2224 found: 377.2217.



N1,N4-bis(4-(tert-butyl)benzyl)-N1,N4-dimethylbenzene-1,4-diamine (3u): white solid was obtained with 43% isolated yield. ¹H NMR (400 MHz, CDCl₃) δ 7.39 – 7.24 (m, 4H), 7.23 – 7.16 (m, 4H), 6.77 (s, 4H), 4.34 (s, 4H), 1.30 (s, 18H). ¹³C NMR (101 MHz, CDCl₃) δ 149.56, 142.94, 136.49, 126.95, 125.26, 115.04, 57.93, 38.96, 34.41, 31.39. HRMS (ESI) calcd for C₃₀H₄₁N₂ [M+H]⁺: 429.3264 found: 429.3257.



dimethyl 4,4'-((1,4-phenylenebis(methylazanediyl))bis(methylene))dibenzoate (3v): white solid was obtained with 47%

isolated yield. ¹H NMR (400 MHz, CDCl₃) δ 7.97 (d, *J* = 8.3 Hz, 4H), 7.33 (d, *J* = 8.3 Hz, 4H), 6.72 (s, 4H), 3.90 (s, 6H), 2.89 (s, 6H). ¹³C NMR (101 MHz, CDCl₃) δ 167.03, 145.13, 142.61, 129.78, 128.75, 127.07, 115.04, 76.68, 58.12, 52.01, 39.30. HRMS (ESI) calcd for C₂₆H₂₉N₂O₄ [M+H]⁺: 433.2122 found: 433.2116.



N1,N4-dimethyl-N1,N4-bis(thiophen-2-ylmethyl)benzene-1,4-diamine (3w): white solid was obtained with 33% isolated yield. ¹H NMR (400 MHz, CDCl₃) δ 7.16 (dd, J = 5.0, 1.3 Hz, 2H), 7.00 – 6.90 (m, 2H), 6.90 – 6.86 (m, 2H), 6.82 (s, 4H), 4.54 (s, 4H), 2.84 (s, 6H). ¹³C NMR (101 MHz, CDCl₃) δ 142.61, 142.30, 126.49, 125.06, 124.28, 115.95, 53.54, 38.70. HRMS (ESI) calcd for C₁₈H₂₁N₂S₂ [M+H]⁺: 329.1141 found: 329.1131.

N1,N4-dimethyl-N1,N4-diphenethylbenzene-1,4-diamine (3x): colorless oil was obtained with 33% isolated yield. ¹H NMR (400 MHz, CDCl₃) δ 7.37 – 7.25 (m, 4H), 7.23 – 7.15 (m, 6H), 6.78 (s, 1H), 3.48 – 3.43 (m, 4H), 2.86 – 2.77 (m, 10H). ¹³C NMR (101 MHz, CDCl₃) δ 141.70, 140.13, 128.72, 128.39, 125.97, 115.25, 56.09, 39.19, 32.73. HRMS (ESI) calcd for C₂₄H₂₉N₂ [M+H]⁺: 345.2325 found: 345.2315.



N1,N4-bis(2,2-dimethoxyethyl)-N1,N4-dimethylbenzene-1,4-diamine (3y): colorless oil was obtained with 40% isolated yield. ¹H NMR (400 MHz, CDCl₃) δ 6.75 (s, 4H), 4.50 (t, *J* = 5.1 Hz, 2H), 3.39 (s, 12H), 3.36 – 3.32 (m, 4H), 2.94 – 2.89 (m, 6H). ¹³C NMR (101 MHz, CDCl₃) δ 142.20, 114.64, 103.38, 56.58, 54.32, 39.87. HRMS (ESI) calcd for C₁₆H₂₉N₂O₄ [M+H]⁺: 313.2122 found: 313.2119.



1,4-di(1,4-dioxa-8-azaspiro[4.5]decan-8-yl)benzene (3z): white solid was obtained with 56% isolated yield. ¹H NMR (400 MHz, CDCl₃) δ 6.90 (s, 4H), 3.98 (s, 8H), 3.20 (t, *J* = 5.7 Hz, 8H), 1.85 (t, *J* = 5.8 Hz, 8H). ¹³C NMR (101 MHz, CDCl₃) δ 145.06, 118.28, 107.06, 64.22, 48.90, 34.73. HRMS (ESI) calcd for C₂₀H₂₉N₂O₄ [M+H]⁺: 361.2122 found: 361.2111.



N1,N4-diallyl-N1,N4-dimethylbenzene-1,4-diamine (3aa): colorless oil was obtained with 30% isolated yield. ¹H NMR (400 MHz, CDCl₃) δ 6.75 (s, 4H), 5.86 (m, *J* = 17.3, 10.7, 5.5 Hz, 2H), 5.23 – 5.09 (m, 4H), 3.81 – 3.77 (m, 4H), 2.82 (s, 6H). ¹³C NMR (101 MHz, CDCl₃) δ 142.57, 134.71, 116.39, 115.35, 56.93, 38.77. HRMS (ESI) calcd for C₁₄H₂₁N₂ [M+H]⁺: 217.1699 found: 217.1700.



di-tert-butyl 8,8'-(1,4-phenylene)(1R,1'R,5S,5'S)-bis(3,8-diazabicyclo[3.2.1]octane-3-carboxylate) (3ab): white soild was obtained with 56% isolated yield. ¹H NMR (400 MHz, CDCl₃) δ 6.74 (s, 4H), 4.05 (d, *J* = 21.1 Hz, 4H), 3.72 (d, *J* = 12.7 Hz, 2H), 3.58 (d, *J* = 12.7 Hz, 2H), 3.25 (dd, *J* = 34.6, 12.6 Hz, 4H), 2.00 – 1.95 (m, 4H), 1.79 (dd, *J* = 15.1, 7.4 Hz, 4H), 1.45 (s, 18H). ¹³C NMR (101 MHz, CDCl₃) δ 156.08, 139.39, 117.26, 79.49, 55.37, 55.15, 47.07, 45.83, 28.38, 26.75. HRMS (ESI) calcd for C₂₈H₄₃N₄O₄ [M+H]⁺: 499.3279 found: 499.3270.



1,4-bis(**4-(8-chloro-5,6-dihydro-11H-benzo**[**5,6**]**cyclohepta**[**1,2-b**]**pyridin-11-ylidene**)**piperidin-1-yl**)**benzene** (**3ac**)**:** white soild was obtained with 63% isolated yield. ¹H NMR (400 MHz, CDCl₃) δ 8.41 (dd, J = 4.9, 1.7 Hz, 2H), 7.43 (dd, J = 7.8, 1.7 Hz, 2H), 7.21 – 7.05 (m, 8H), 6.86 (s, 4H), 3.48 – 3.30 (m, 8H), 2.90 – 2.73 (m, 8H), 2.72 – 2.60 (m, 2H), 2.60 – 2.40 (m, 6H). ¹³C NMR (101 MHz, CDCl₃) δ 157.45, 146.63, 139.53, 138.13 (d, J = 71.1 Hz), 137.28, 133.37, 132.97, 132.67, 130.77, 128.92, 126.01, 122.08, 118.13, 52.22, 52.11, 31.59 (d, J = 32.6 Hz), 30.85 (d, J = 25.7 Hz). HRMS (ESI) calcd for C₄₄H₄₁N₄Cl₂ [M+H]⁺: 695.2703 found: 695.2683.

1,4-dimorpholino-2-methyl-benzene (3ad): white solid was obtained with 51% isolated yield. ¹H NMR (400 MHz, CDCl₃) δ 6.98 (d, J = 8.6 Hz, 1H), 6.79 (d, J = 2.9 Hz, 1H), 6.74 (dd, J = 8.6, 3.0 Hz, 1H), 3.93 – 3.79 (m, 8H), 3.09 (dd, J = 5.8, 3.6 Hz, 4H), 2.88 – 2.81 (m, 4H), 2.30 (s, 3H). ¹³C NMR (101 MHz, CDCl₃) δ 147.52, 144.52, 133.81, 119.89, 118.92,

 $113.92,\,67.51,\,66.98,\,52.62,\,49.93,\,17.96.\ HRMS\ (ESI)\ calcd\ for\ C_{44}H_{41}N_4Cl_2\ [M+H]^+:\,263.1754\ found:\ 263.1755.$

References

- [1] A. Bismuto, T. Delcaillau, P. Müller and B. Morandi, ACS Catal., 2020, 10, 4630-4639.
- [2] K. Chen, Q.-K. Kang, Y. Li, W.-Q. Wu, H. Zhu and H. Shi, J. Am. Chem. Soc., 2022, 144, 1144-1151.

Copies of product NMR spectra



210 200 190 180 170 160 150 140 130 120 110 100 90 80 70 60 50 40 30 20 10 fl (ppm)

Spectrum of 3b



210 200 190 180 170 160 150 140 130 120 110 100 90 80 70 60 50 40 30 20 10 fl (ppm)

Spectrum of 3c



fl (ppm)

Spectrum of 3d



210 200 190 180 170 160 150 140 130 120 110 100 90 80 70 60 50 40 30 20 10 fl (ppm)

Spectrum of 3e



fl (ppm)

¹⁹F NMR (377 MHz, CDCl₃)





---180.90

Spectrum of 3f











¹³C NMR (101 MHz, DMSO-d₆)











31





Spectrum of 3n





















38







fl (ppm)















Spectrum of 3ac









