

## Supporting Information

# Uncovering Gold Nanoparticle Synthesis Using Microchip Laser System through Pulsed Laser Ablation in Aqueous Solution

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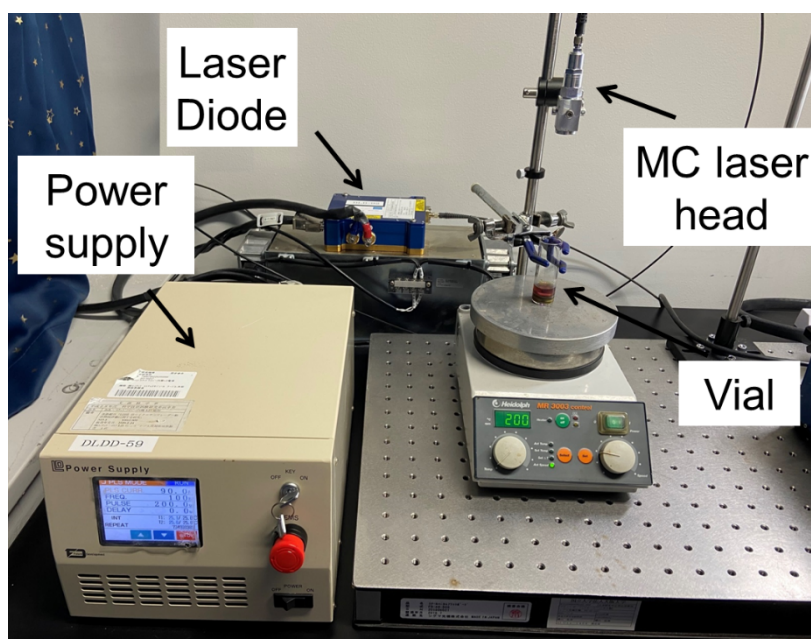
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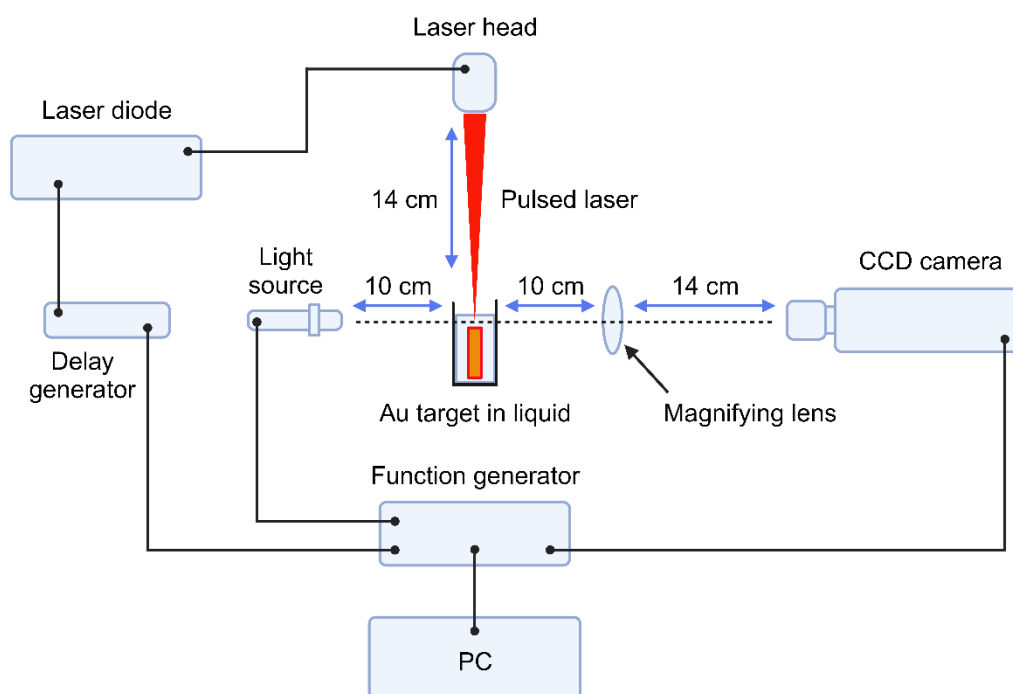
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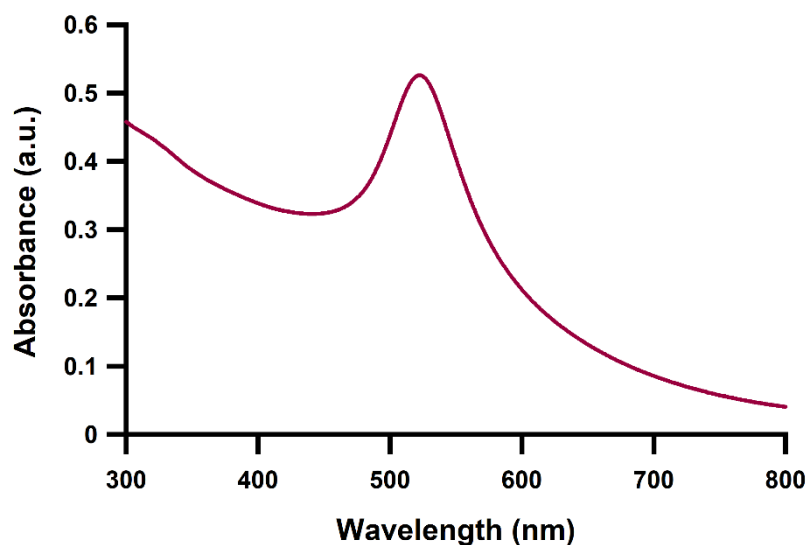
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**Figure S1.** Setup of PLAL system.



**Figure S2.** Experimental setup for monitoring the bubble formation with the aid of videography.

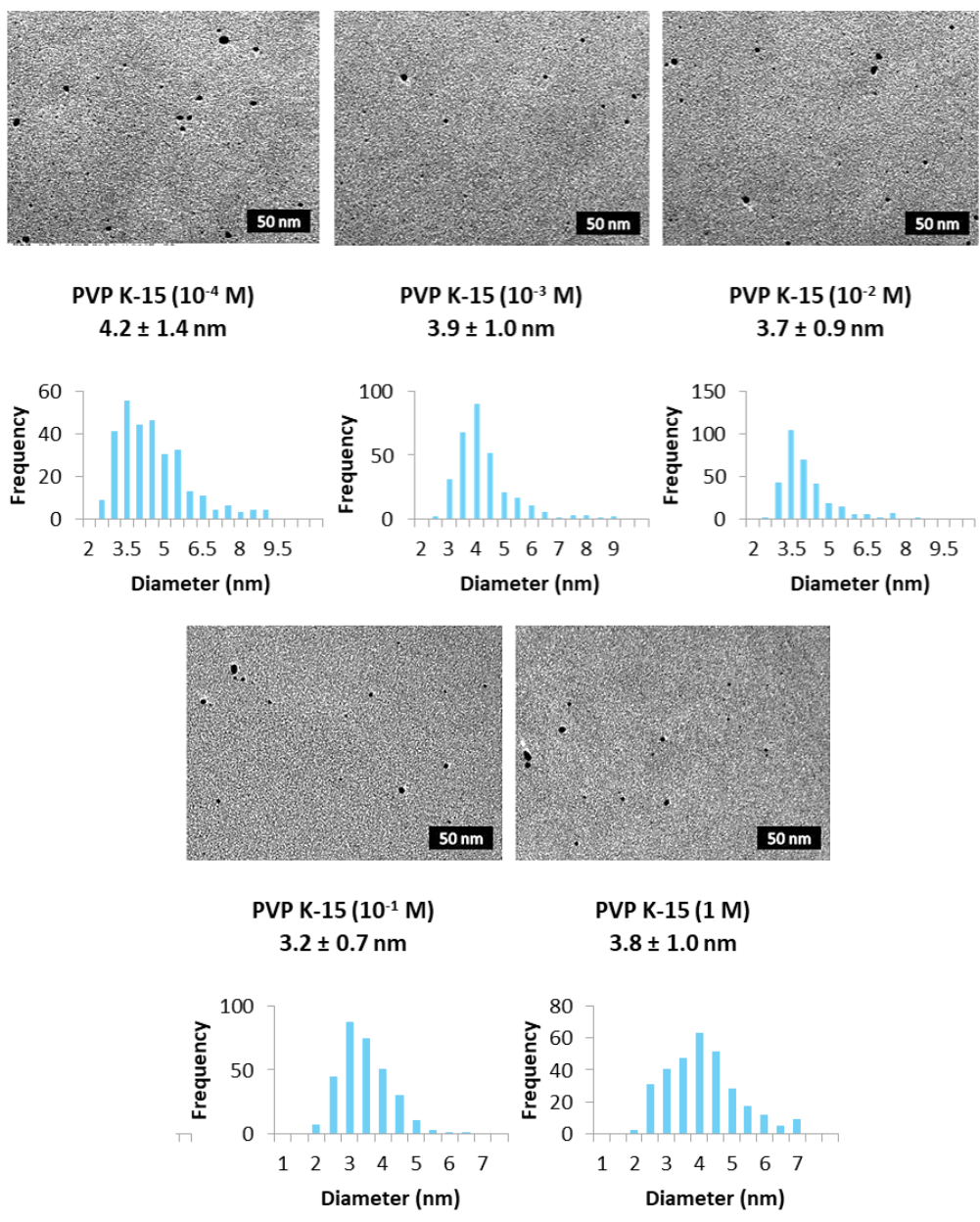


**Figure S3.** UV/Vis absorption spectrum of the Au NPs produced by MCL ablation in water.

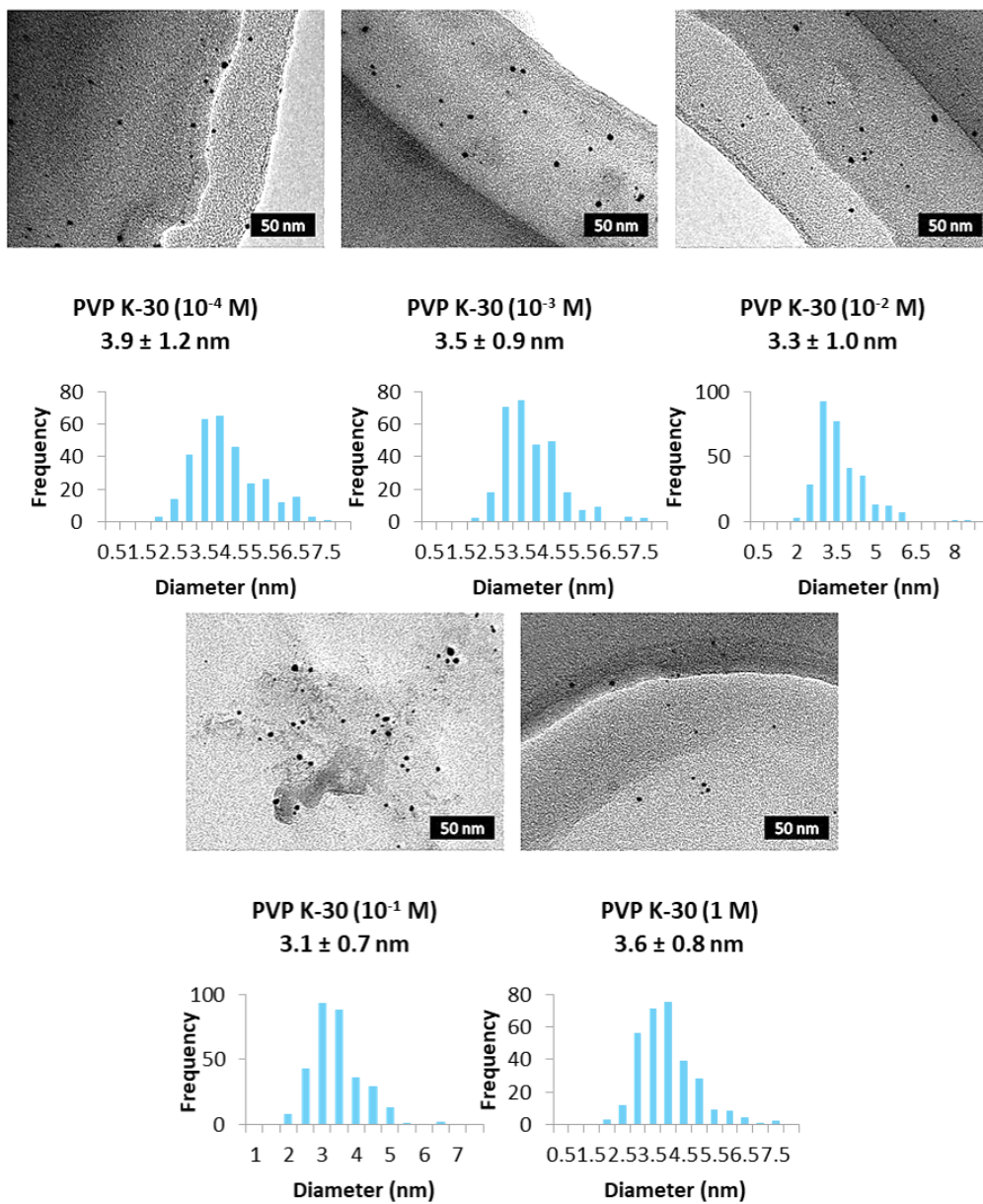
**Table S1.** Present and reported results of PLAL generated particle sizes.

Metal	Matrix	Matrix conc. (mM)	Laser condition	Particle size (nm) <sup>a</sup>	Ref.
Au	PVP	0	900 ps, 1064 nm, 100 Hz, 27 J/pulse	4.9 ± 1.3	This work
		0.1			
		1			
		10			
		100			
1000					
Au <sup>b</sup>	PVP	0.0001 <sup>c</sup> 0.001 <sup>c</sup> 0.01 <sup>c</sup>	100 fs, 780 nm, 10 Hz, 6 mJ/pulse	3-5	1
Au <sup>d</sup>	PVP	0.01 <sup>c</sup>	100 fs, 800 nm, 1 kHz, 1 mJ/pulse	7-27	2
Au	PVP	0.5	7 ns, 1064 nm, 10 Hz, 11 mJ/pulse	20-50 <sup>e</sup>	3
Au	PVP	0.4	8 ns, 335 nm, 10 Hz, 17 mJ/pulse <sup>f</sup>	720 <sup>g</sup>	4
Au	SDS <sup>h</sup>	10	12 ns, 1064 nm, 10 Hz, 80 mJ/pulse	4.6	5
		0.1		14.4	
Au	$\beta$ -CD <sup>i</sup>	0	110 fs, 800 nm, 1 kHz, 1 mJ/pulse	40-60	6
		0.1		10	
		1		5.3	
		10		2.3	
Ag	PVP	0	8 ns, 1064 nm, 10 Hz, 12 mJ/pulse	18	7
		2		13	
		6		10	
		12		11	
		18		11	

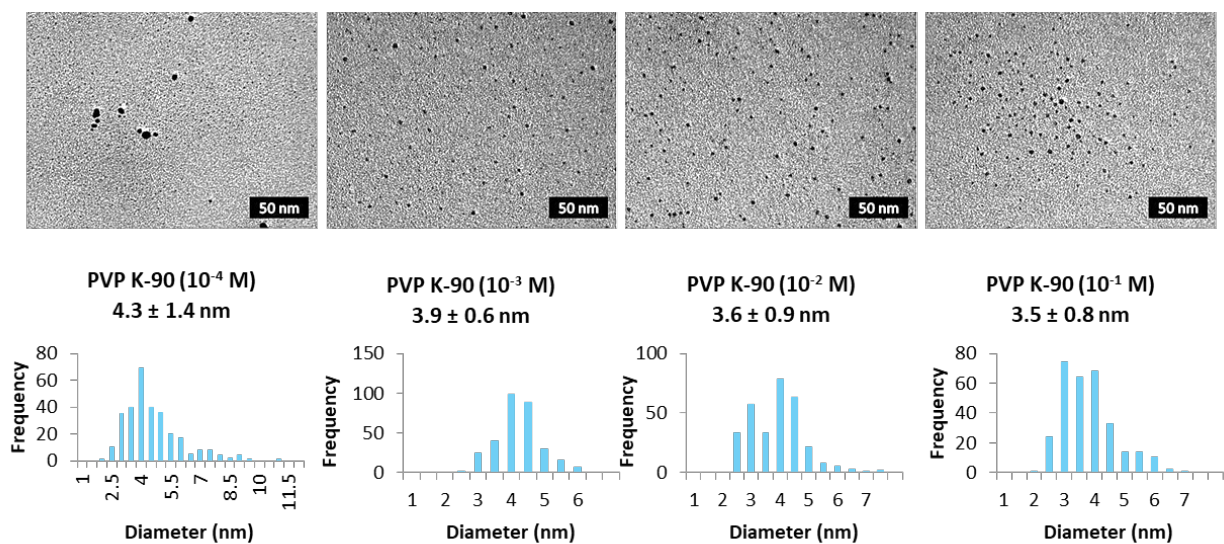
<sup>a</sup>Determined from TEM image. <sup>b</sup>wt%. <sup>c</sup>Laser ablation in H<sub>2</sub>AuCl<sub>4</sub> solution. <sup>d</sup>Laser ablation in KAuCl<sub>4</sub> solution. <sup>e</sup>Hydrodynamic diameter. <sup>f</sup>Estimated from 0.5 mm diameter and 17 J/cm<sup>2</sup> fluence. <sup>g</sup>Determined from SEM image. <sup>h</sup>Sodium dodecyl sulfate. <sup>i</sup>β-cyclodextrin.



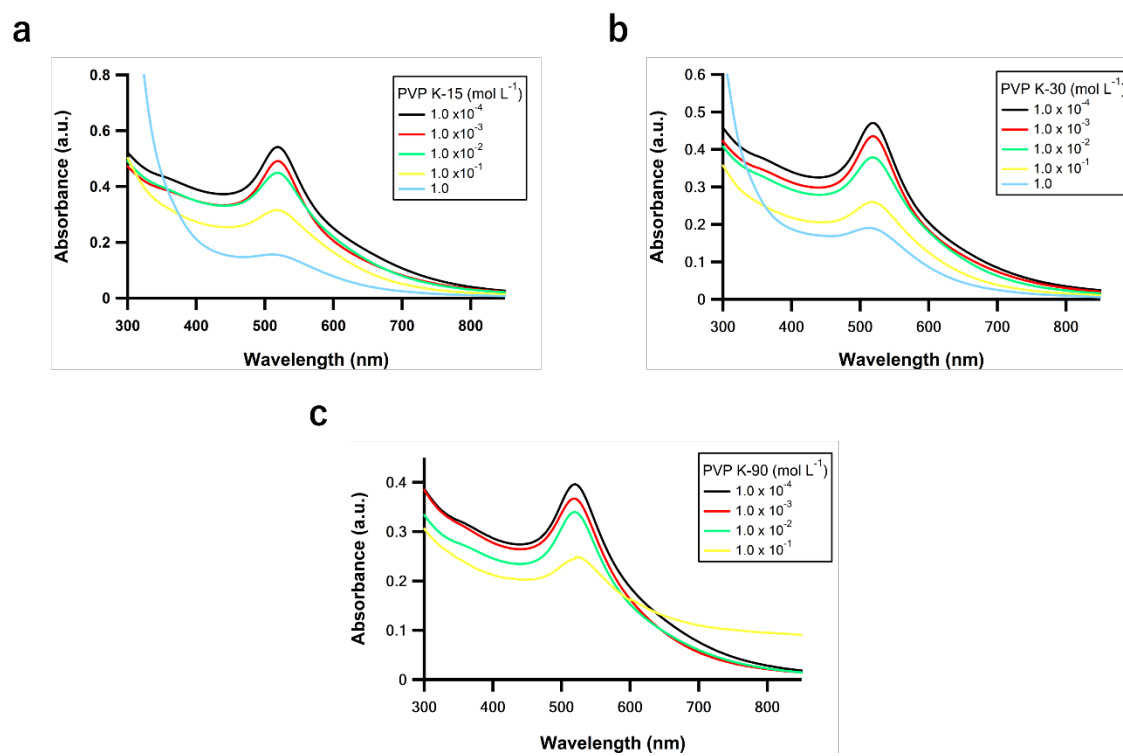
**Figure S4.** TEM images of Au:PVP (K-15) prepared in water.



**Figure S5.** TEM images of Au:PVP (K-30) prepared in water.



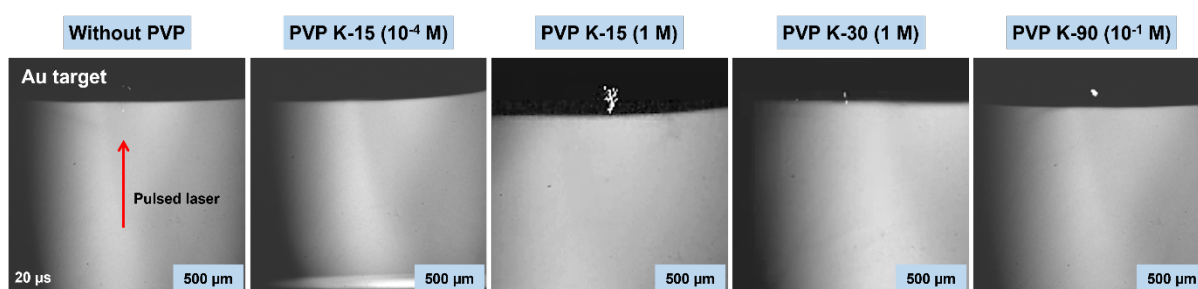
**Figure S6.** TEM images of Au:PVP (K-90) prepared in water.



**Figure S7.** UV/Vis absorption spectrum of Au NPs in different concentration of (a) PVP K-15, (b) PVP K-30 and (c) PVP K-90.

**Table S2.** Ablated total Au amount in 15 mL solvent with the same viscosity.

PVP type	Viscosity (mPa·s)	Au productivity ( $\mu\text{g h}^{-1}$ )
PVP K-15	$2.20 \pm 0.01$	93
PVP K-30	$2.23 \pm 0.01$	92
PVP K-90	$2.14 \pm 0.12$	104



**Figure S8.** Snapshot images of the target surface corrected at 2  $\mu\text{s}$  after the induction of the laser pulse. From left, without PVP, PVP K-15 ( $10^{-4}$  M), PVP K-15 (1 M), PVP K-30 (1 M), and PVP K-90 ( $10^{-1}$  M).

## References

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