Supplementary Information

A lab-made screen-printed sensing strip for sensitive and selective electrochemical detection of butylated hydroxyanisole

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 $[\]dagger$ Electronic supplementary information (ESI) available: cross-sectional SEM images of LabSPE before and after modification, FT-IR spectrum of g-C₃N₄-Fe₃O₄ nanocomposite, XRD patern of g-C₃N₄ and g-C₃N₄-Fe₃O₄, WCA measurements of electrodes, CVs of LabSPE and commercial SPE, selectivity test of sensor, DPVs of mixture of BHA and other antioxidants, and stability test of sensor.



Fig. S1. Cross-sectional SEM images of LabSPE before and after modification.



Fig. S2. FT-IR spectrum of g-C₃N₄-Fe₃O₄ nanocomposite.



Fig. S3. XRD patern of g-C₃N₄ and g-C₃N₄-Fe₃O₄ nanocomposite.



Fig. S4. WCA measurements of bare LabSPE (a), $g-C_3N_4$ -Fe₃O₄/LabSPE (b), and Pt/g-C₃N₄-Fe₃O₄/LabSPE (c) surfaces.



Fig. S5. CVs of LabSPE and commercial SPE in 0.1 M KCl solution containing 5 mM $Fe(CN)_6^{3-/4-}$.



Fig. S6. (a) Selectivity test of Pt/g-C₃N₄-Fe₃O₄/LabSPE in the presence of 25 μ M BHA, 25 μ M BHA + 25 μ M BHT, 25 μ M BHA + 25 μ M GA, 25 μ M BHA + 25 μ M PG, 25 μ M BHA + 25 μ M TBHQ in pH 4.0 citric acid-Na₂HPO₄ buffer solution. (b) DPVs of Pt/g-C₃N₄-Fe₃O₄/LabSPE in the presence of 25 μ M BHA + 25 μ M other antioxidant compound (BHT, GA, PG or TBHQ) in pH 4.0 citric acid-Na₂HPO₄ buffer solution.



Fig. S7. Stability test of Pt/g-C₃N₄-Fe₃O₄/LabSPE in the presence of 10 μ M BHA for 35 days.