

Correlating Structure and Photophysical Properties in Thiazolo[5,4-*d*]thiazole Crystal Derivatives for Use in Solid- State Photonic and Fluorescence-Based Optical Devices

*Abhishek Shibu, Sean Jones, P. Lane Tolley, David Diaz, Carly O. Kwiatkowski, Daniel S. Jones,
Jessica M. Shivas, Jonathan J. Foley IV, Thomas A. Schmedake, and Michael G. Walter**

Department of Chemistry, University of North Carolina at Charlotte, Charlotte, NC, 28223,

*Michael.Walter@uncc.edu

TABLE OF CONTENTS

Section 1: Spectra Characterization

Section 1.1: $^1\text{H-NMR}$ Spectra.....3

Section 1.2: MALDI-TOF-MS.....7

Section 2: X-Ray Characterization

Section 2.1: Single Crystal Structure Data and Refinement.....8

Section 2.2: Correlation between Volume of Unit Cell and Alkyl Chain Length.....9

Section 2.3: Pictorial representation of interaction energies in TTz crystals.....10

Section 2.4: Intermolecular interaction energies in TTz crystals.....11

Section 2.5: List of contacts from crystal packing data of $(\text{C}_n\text{OPh})_2\text{TTz}$ crystals....12

Section 2.6: Atom labelling sequence.....13

Section 2.7: Distribution of intermolecular contacts to Hirshfeld surface area.....14

Section 2.8: Shape index of the Hirshfeld surfaces.....15

Section 2.9: Powder X-Ray Diffractograms of TTz Crystals.....16

Section 3: Photophysical Characterization

Section 3.1: DFT Simulated Raman Spectra.....17

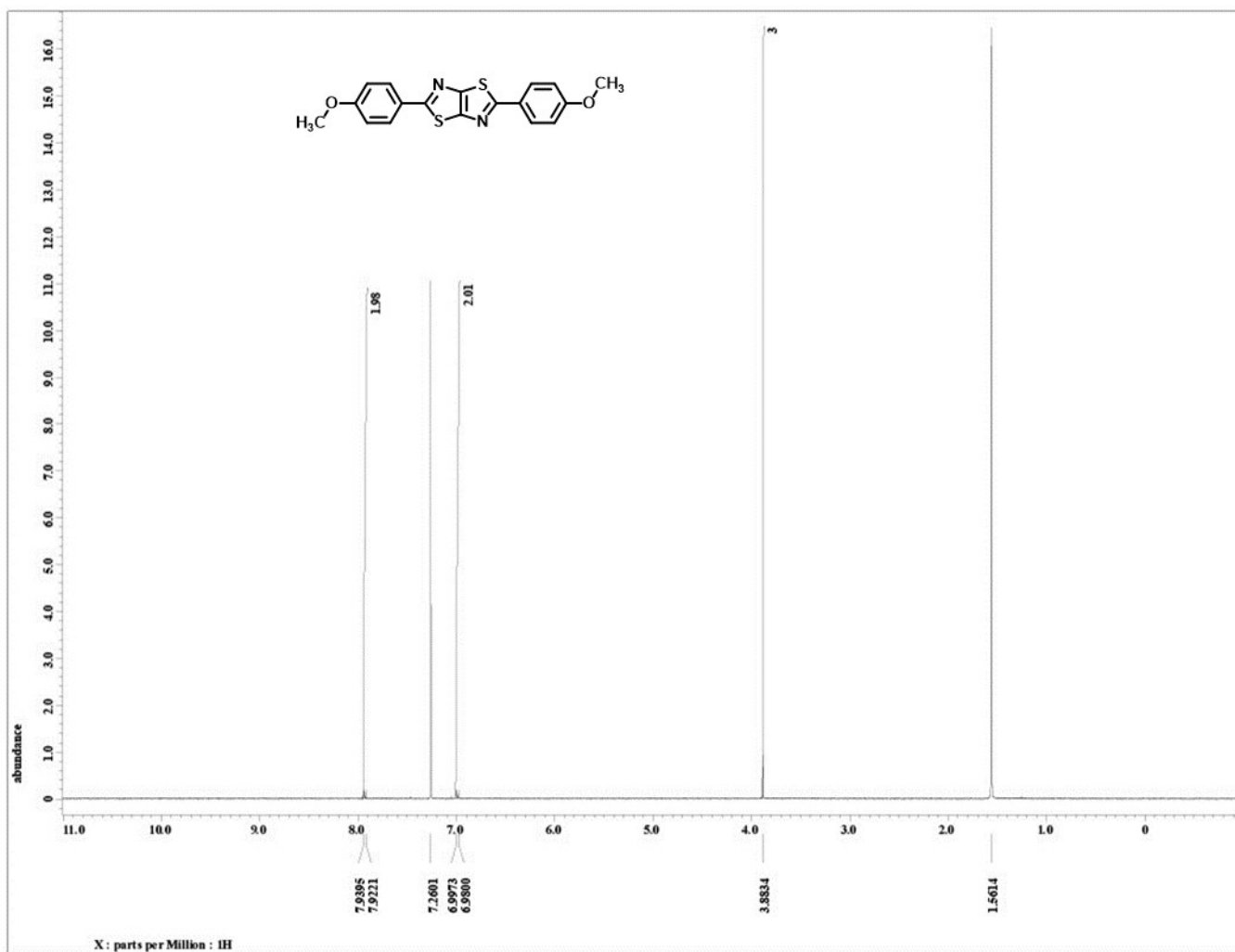
Section 3.2: Lattice-Phonon Raman Spectra of TTz Crystals.....17

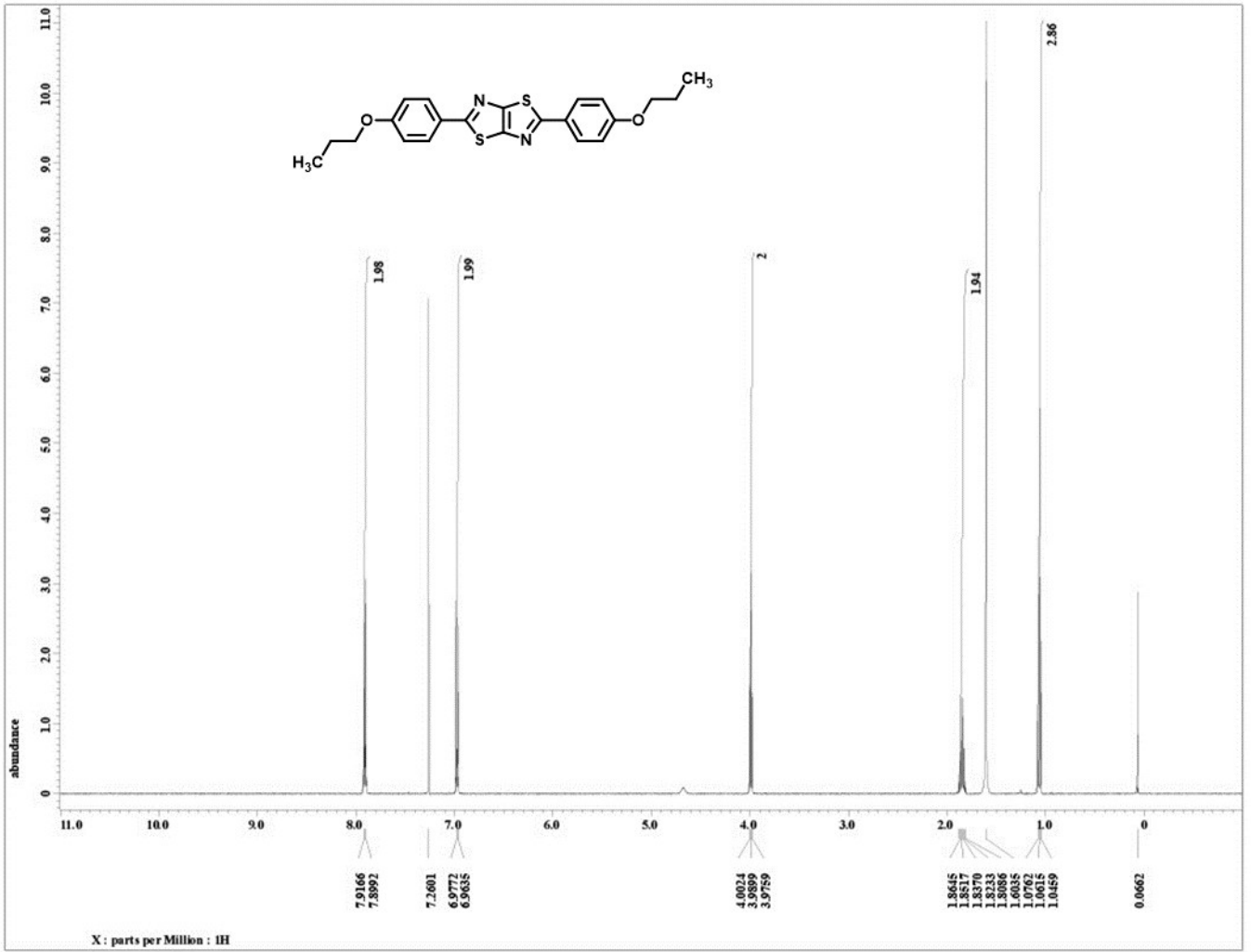
Section 3.3: Photostability of TTz Crystals.....18

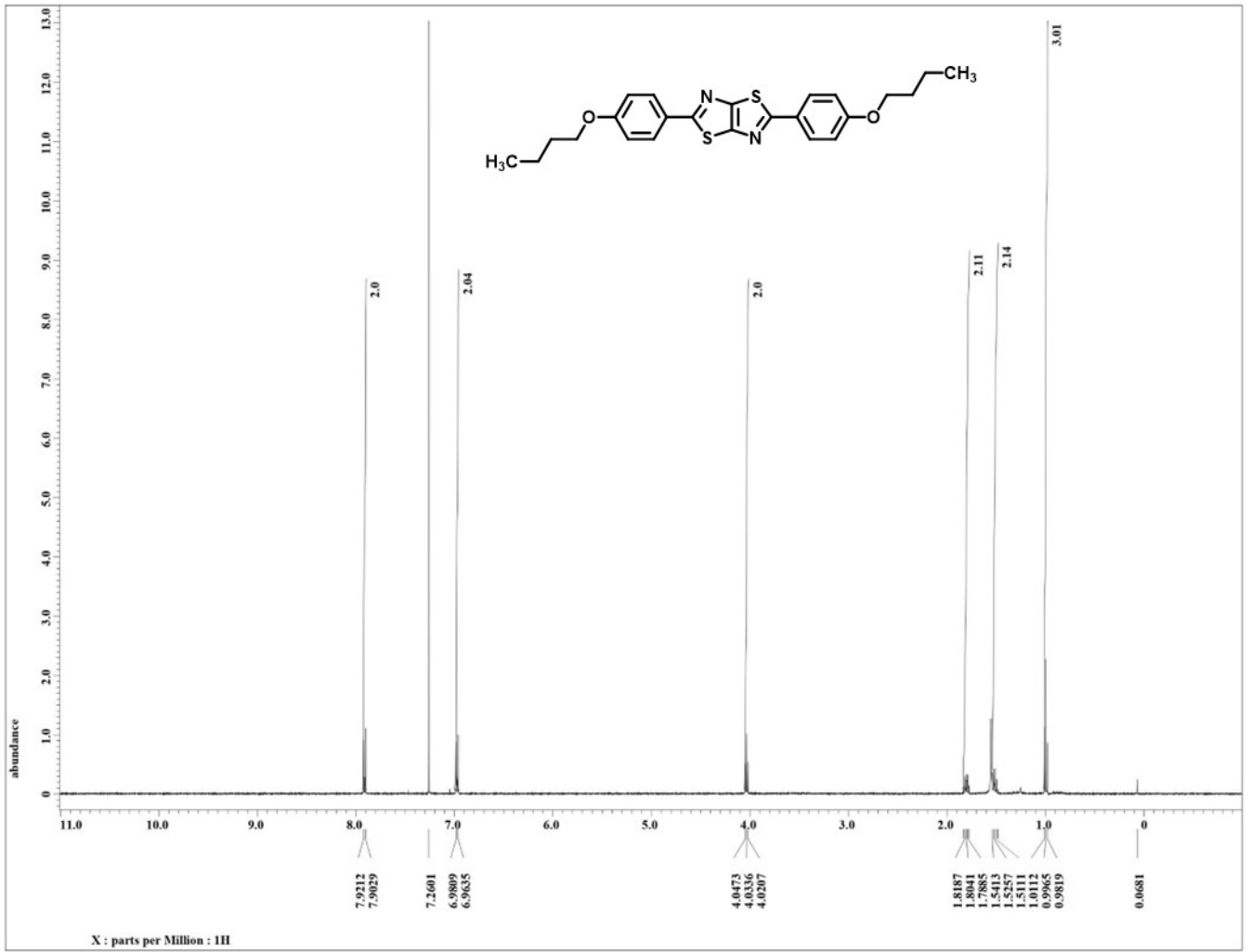
Section 4: References.....18

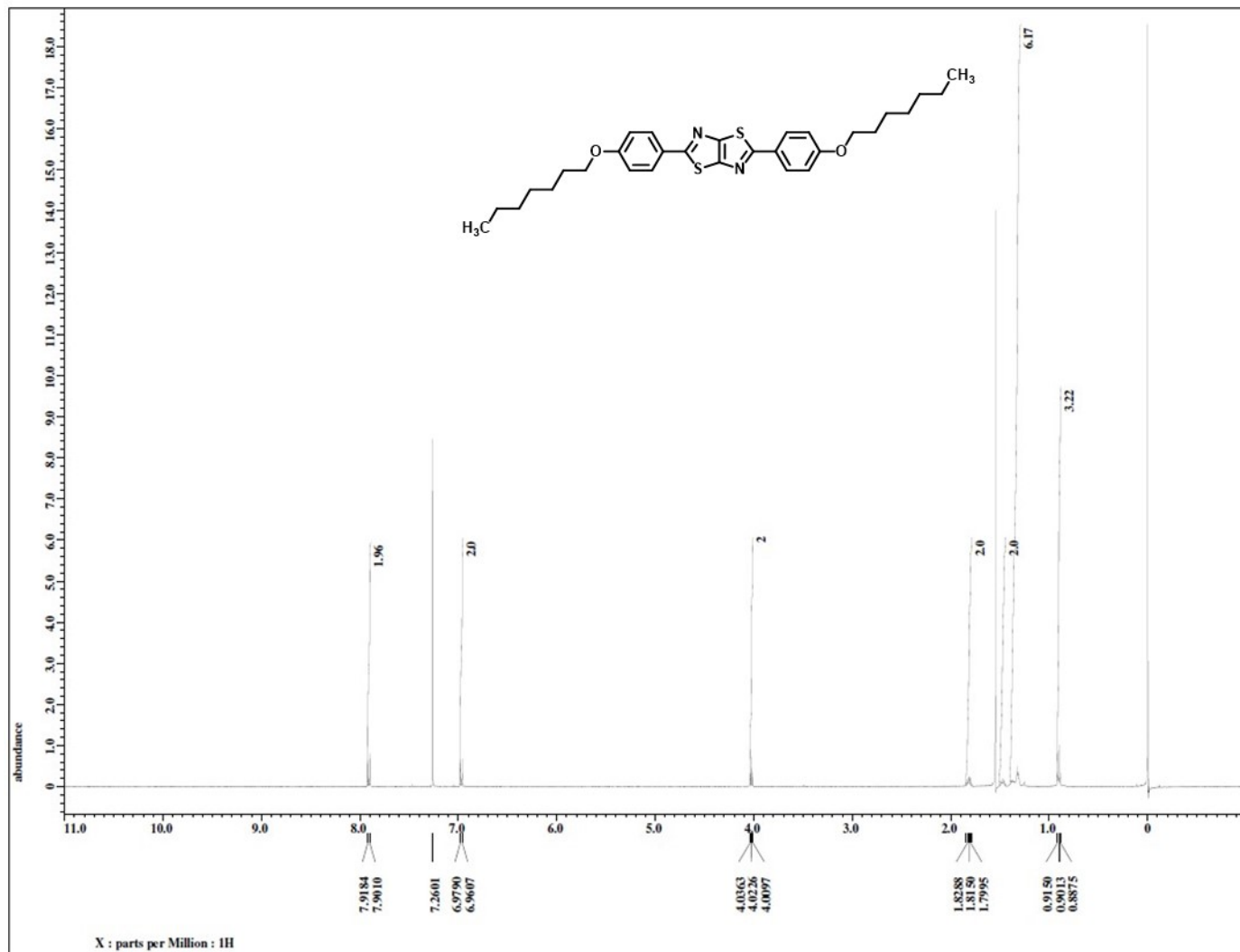
Section 1: Spectra Characterization

Section 1.1: $^1\text{H-NMR}$ Spectra









Section 1.2: MALDI-TOF-MS

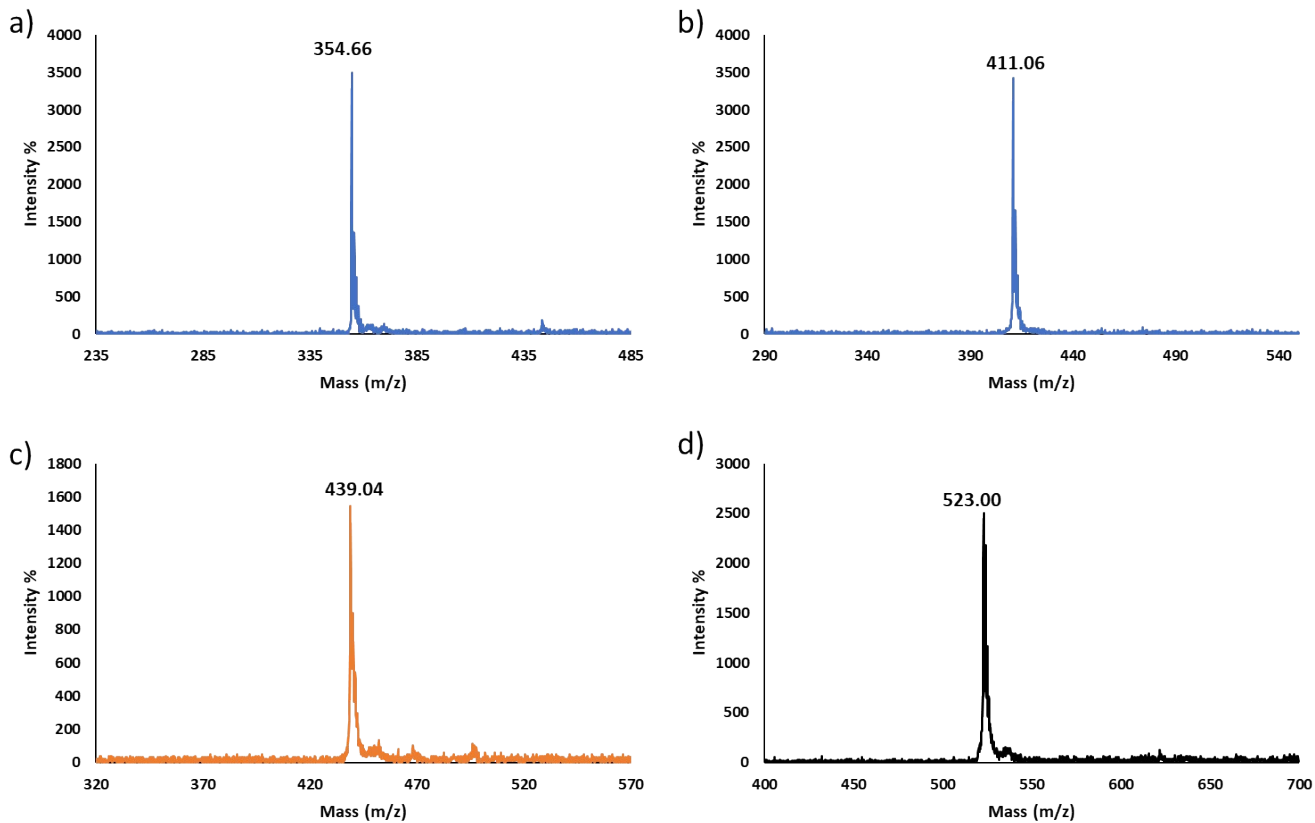


Figure S1: Maldi-TOF-MS Spectra of a) (MeOPh)₂TTz, b) (PrOPh)₂TTz, c) (BuOPh)₂TTz, and d) (HepOPh)₂TTz

Section 2: X-Ray Characterization

Section 2.1: Single Crystal Structure Data and Refinement

Table S1: Summary of crystallographic data of dialkoxyphenyl TTz

Identification code	(PrOPh) ₂ TTz	(BuOPh) ₂ TTz	(HepOPh) ₂ TTz
CSD Deposition Number	2257069	2257068	2257067
Empirical formula	C ₂₂ H ₂₂ N ₂ O ₂ S ₂	C ₂₄ H ₂₆ N ₂ O ₂ S ₂	C ₃₀ H ₃₈ N ₂ O ₂ S ₂
Formula weight	410.564	438.618	522.781
Temperature/K	100.00(10)	100.00(10)	100.00(10)
Crystal system	monoclinic	triclinic	monoclinic
Space group	P2 ₁ /c	P-1	P2 ₁ /c
a/Å	16.2976(4)	5.4441(5)	31.0223(11)
b/Å	7.46099(19)	7.4759(7)	7.1900(3)
c/Å	7.9920(2)	14.1023(13)	6.0107(2)
α/°	90	84.330(8)	90
β/°	96.418(2)	82.922(8)	91.400(3)
γ/°	90	70.044(8)	90
Volume/Å ³	965.71(4)	534.39(9)	1340.29(9)
Z	2	1	2
ρ _{calc} /cm ³	1.412	1.363	1.295
μ/mm ⁻¹	2.669	2.446	2.032
F(000)	434.6	233.3	562.9
Crystal size/mm ³	0.378 × 0.364 × 0.197	0.565 × 0.511 × 0.036	0.165 × 0.141 × 0.035
Radiation	Cu Kα (λ = 1.54184)	Cu Kα (λ = 1.54184)	Cu Kα (λ = 1.54184)
2θ range for data collection/°	10.92 to 133.34	12.62 to 133.6	8.56 to 133.84
Index ranges	-19 ≤ h ≤ 19, -7 ≤ k ≤ 8, -9 ≤ l ≤ 8	-6 ≤ h ≤ 5, -8 ≤ k ≤ 6, -16 ≤ l ≤ 16	-36 ≤ h ≤ 36, -8 ≤ k ≤ 7, -6 ≤ l ≤ 7
Reflections collected	8742	4008	12196
Independent reflections	1698 [R _{int} = 0.0335, R _{sigma} = 0.0206]	1873 [R _{int} = 0.0374, R _{sigma} = 0.0367]	2371 [R _{int} = 0.0772, R _{sigma} = 0.0466]
Data/restraints/parameters	1698/0/141	1873/0/137	2371/0/176
Goodness-of-fit on F ²	1.075	1.075	1.103
Final R indexes [I >= 2σ (I)]	R ₁ = 0.0300, wR ₂ = 0.0733	R ₁ = 0.0366, wR ₂ = 0.0945	R ₁ = 0.0744, wR ₂ = 0.2012
Final R indexes [all data]	R ₁ = 0.0358, wR ₂ = 0.0780	R ₁ = 0.0431, wR ₂ = 0.1008	R ₁ = 0.0971, wR ₂ = 0.2214
Largest diff. peak/hole / e Å ⁻³	0.22/-0.28	0.40/-0.28	0.82/-0.43

Section 2.2: Correlation between Volume of Unit Cell and Alkyl Chain Length

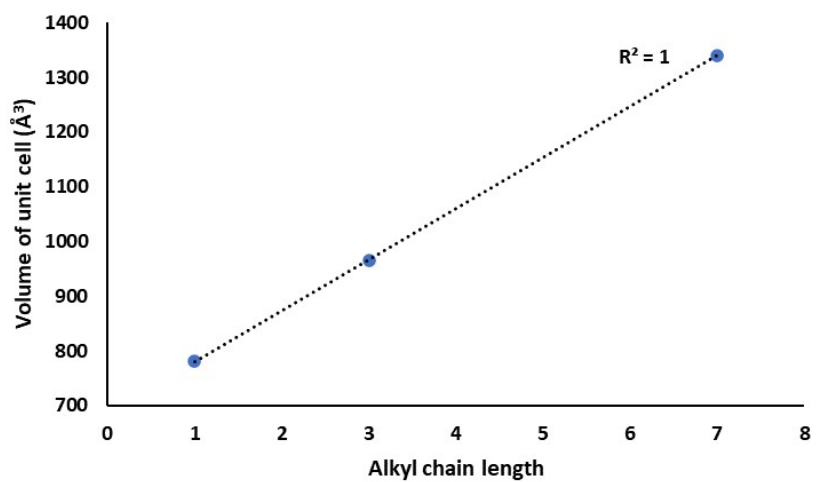


Figure S2: Crystals packed in monoclinic crystal system showed a perfect correlation between volume of unit cell and alkyl chain length.

Section 2.3: Pictorial representation of interaction energies in TTz crystals

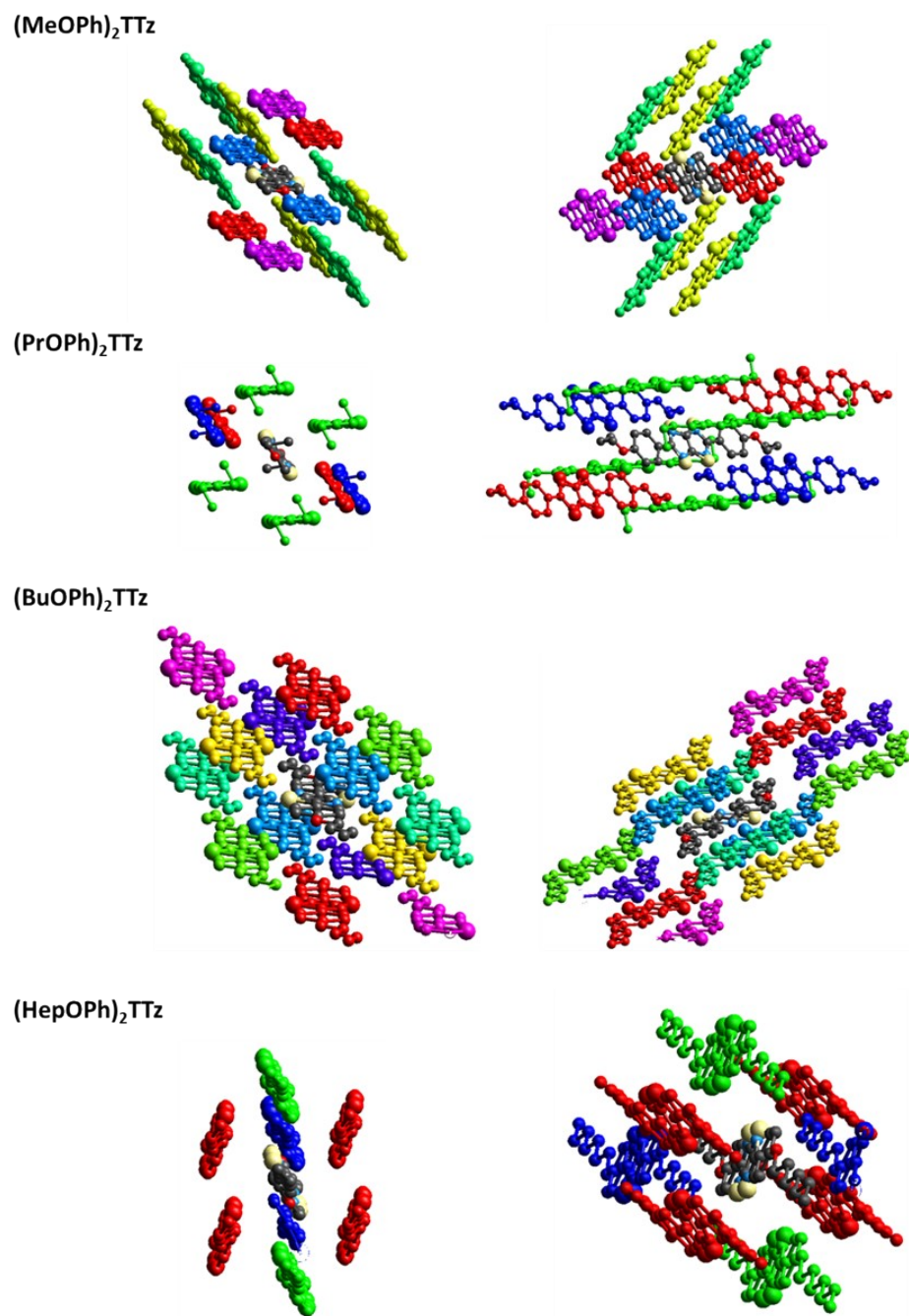


Figure S3: Color-coded pictorial representation of interaction energy in TTz crystals with respect to one central reference molecule.

Section 2.4: Intermolecular interaction energies in TTz crystals

Table S2: Details of intermolecular interaction energies in TTz crystals

	Pair label (<i>cf</i> Fig 3)	Orientation	Color	N	R (Å)	E _{ele} (kJ/mol)	E _{pol} (kJ/mol)	E _{dis} (kJ/mol)	E _{rep} (kJ/mol)	E _{tot} (kJ/mol)
(MeOPh)₂TTz	a)	Cofacial	Red	2	3.98	-6.98	-2.44	-80.74	36.34	-53.70
	b)	Orthogonal	Green	4	10.67	-11.73	-1.48	-25.69	14.28	-24.60
	c)	Parallel	Blue	2	19.04	-7.82	-1.18	-8.80	0.00	-17.90
	d)	Orthogonal	Yellow	4	11.47	-5.60	-0.89	-16.03	6.30	-16.20
(PrOPh)₂TTz		Cofacial	Purple	2	18.52	3.28	-0.37	-5.75	0.00	-2.80
	e)	Orthogonal	Green	4	5.47	-54.65	-7.03	-101.65	93.75	-69.70
	f)	Cofacial	Blue	2	7.99	-28.12	-5.25	-58.62	75.52	-16.60
	g)	Cofacial	Red	2	18.94	-0.11	-0.07	-8.97	0.00	-9.20
(BuOPh)₂TTz	h)	Cofacial	Red	2	15.29	-11.20	-1.41	-58.53	0.00	-71.10
		Cofacial	Blue	2	14.1	1.16	-1.70	-68.29	0.00	-68.80
		Cofacial	Green	2	15.96	-29.81	-1.26	-27.35	0.00	-58.50
		Cofacial	Teal	2	14.48	-10.36	-2.00	-33.01	0.00	-45.40
		Cofacial	Yellow	2	5.44	-10.36	-5.25	-44.68	22.93	-37.40
		Cofacial	Purple	2	28.46	0.85	0.00	-4.88	0.00	-4.00
		Cofacial	Pink	2	28.8	0.11	0.00	-0.70	0.00	-0.60
(HepOPh)₂TTz	i)	Orthogonal	Red	4	4.69	-72.51	0.00	-137.88	99.37	-111.00
	j)	Cofacial	Green	2	6.01	-79.06	-7.62	-78.39	110.44	-54.70
		Cofacial	Blue	2	33.54	0.21	0.00	-7.06	0.00	-6.80

R = distance between molecular centroids

$$E_{\text{tot}} = E_{\text{ele}} + E_{\text{pol}} + E_{\text{dis}} + E_{\text{rep}}$$

Where,

E_{ele} = classical electrostatic energy of interaction between monomer charge distributions

E_{rep} = exchange–repulsion energy

E_{dis} = dispersion energy

E_{pol} = Polarization energy

Since k_BT = 2.5 kJ/mol and the error to which the sublimation enthalpy can be measured is

approximately 5 kJ/mol for organic compounds, molecular pairs with E_{tot} ~ 5 kJ/mol should be

ignored.

Section 2.5: List of contacts from crystal packing data of (C_nOPh)₂TTz crystals

Table S3: (C_nOPh)₂TTz crystal packing data with emphasis on interchromophoric interactions. For atom labelling sequence *cf.* Figure S4.

	Pair label (<i>cf.</i> Fig 3)	Orientation	Color	Interaction	Distance, (Å)	Angle, (degrees)	Frequency of Interaction	Number of molecules	Total				
(MeOPh)₂TTz	a)	Cofacial	Red	C ₉ -H—O ₁	2.935	149.48	2	2	4				
				π - π _{phenyl} centroid	3.985		2	2	4				
		b)	Orthogonal	Green	C ₅ -H—S ₁	3.519	120.35	1	4	4			
					C ₅ -H—N ₁	2.964	178.48	1	4	4			
					C ₉ —S ₁	3.468	173.66	1	4	4			
		c)		Yellow	C ₉ -H—N ₁	2.912	161.62	1	4	4			
					C ₇ -H—S ₁	3.851	169.5	1	4	4			
		d)	Parallel	Blue	C ₉ -H—O ₁	2.537	144.74	2	2	4			
	(PrOPh)₂TTz	e)	Orthogonal	Green	C ₈ -H—S ₁	3.839		1	4	4			
					C ₁₀ -H—O ₁	3.249	127.93	1	4	4			
C ₁₁ -H—O ₁					3.44	122.36	1	4	4				
C ₁₀ -H—O ₁					3.922	135.47	1	4	4				
C ₂ —S ₁					3.38		1	4	4				
S—S					3.488		1	4	4				
C ₇ -H—C ₃					2.88	136.06	1	4	4				
S ₁ —C ₃					3.327		1	4	4				
C ₄ —S ₁					3.379		1	4	4				
					f)	Cofacial	Blue	C ₇ -H—N ₁	3.35	126.41	2	2	4
								C ₉ -H—N ₁	2.986	72.88	2	2	4
								C ₈ —S ₁	3.456		2	2	4
					g)	Cofacial	Red	C ₁₁ -H—O ₁	3.799	142.93	2	2	4
(BuOPh)₂TTz		h)	Cofacial	Red	C ₁₂ -H—S ₁	3.266	123.54	2	14	28			
	C ₉ -H—O ₁				3.321	124.83	2	14	28				
	C ₁₂ -H—N ₁				3.255	170.66	2	14	28				
	C ₁₀ -H— π _{phenyl} centroid				2.856	138.1	2	14	28				
	C ₁₁ -H— π _{phenyl} centroid				3.591		2	14	28				
	C ₁₂ -H—TTZ _{centroid}				3.601	142.8	2	14	28				

(HepOPh) ₂ TTz	i)	Orthogonal	Red	C ₁₀ -H-O ₁	3.11	141.73	1	4	4
				C ₅ -H-O ₁	3.893	147.75	1	4	4
				C ₁₁ -H-O ₁	3.519	131.74	1	4	4
				C ₉ -H-O ₁	3.016	144.94	1	4	4
				C ₄ -H-S ₁	3.599	123.69	1	4	4
				C ₇ -H-C ₆	2.826	124.37	1	4	4
				S ₁ -C ₂	3.311		1	4	4
				S ₁ -C ₁	3.392		1	4	4
				C ₄ -H-C ₃	2.895	124.17	1	4	4
				C ₄ -H-C ₂	2.892	144.2	1	4	4
	j)	Cofacial	Green	C ₁₁ -H-O ₁	3.062	170.29	2	2	4
				C ₇ -H-H-C ₉	2.345	157.81	2	2	4
				S-S	3.118		1	2	2

Section 2.6: Atom labelling sequence

The atom labelling sequence of (HepOPh)₂TTz is shown in Figure S3. Truncated labelling sequence can be used for other TTz derivatives.

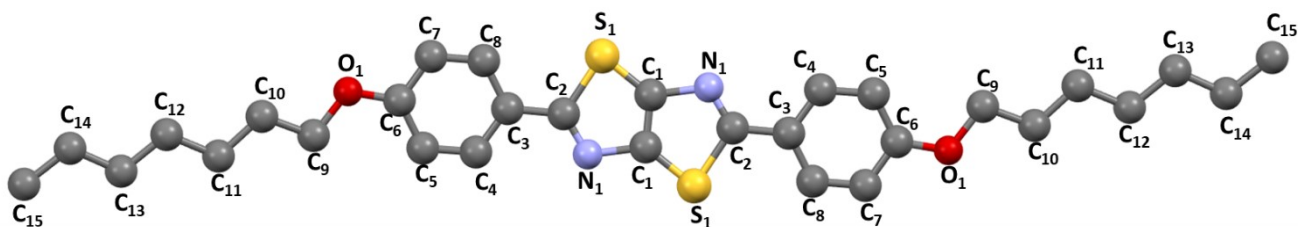


Figure S4. (HepOPh)₂TTz with labelled non-hydrogen atoms

Section 2.7: Distribution of intermolecular contacts to Hirshfeld surface area

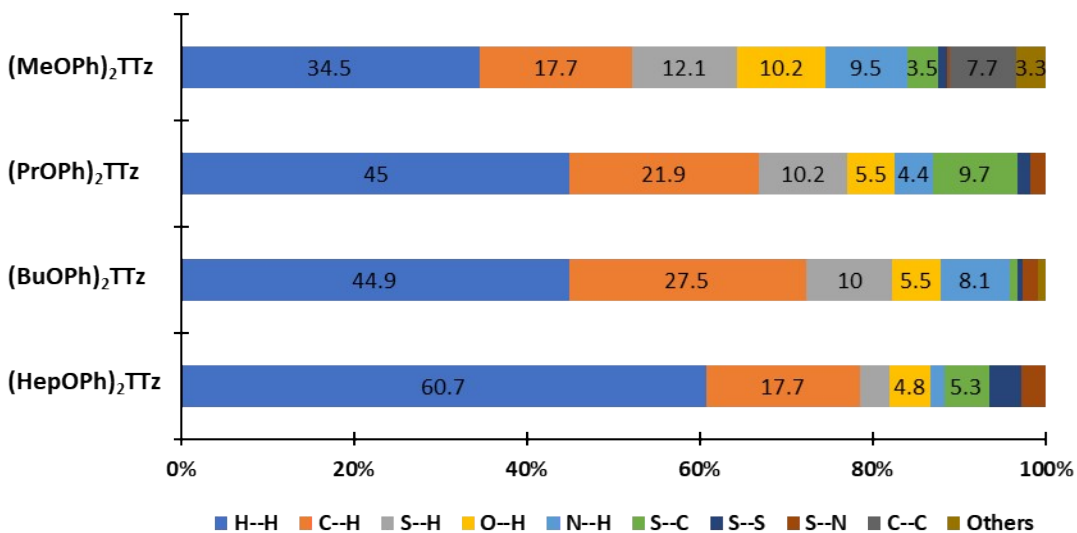


Figure S5: Relative contribution of the intermolecular contacts to the Hirshfeld surface area for (CnOPh)₂TTz crystals.

Section 2.8: Shape index of the Hirshfeld surfaces

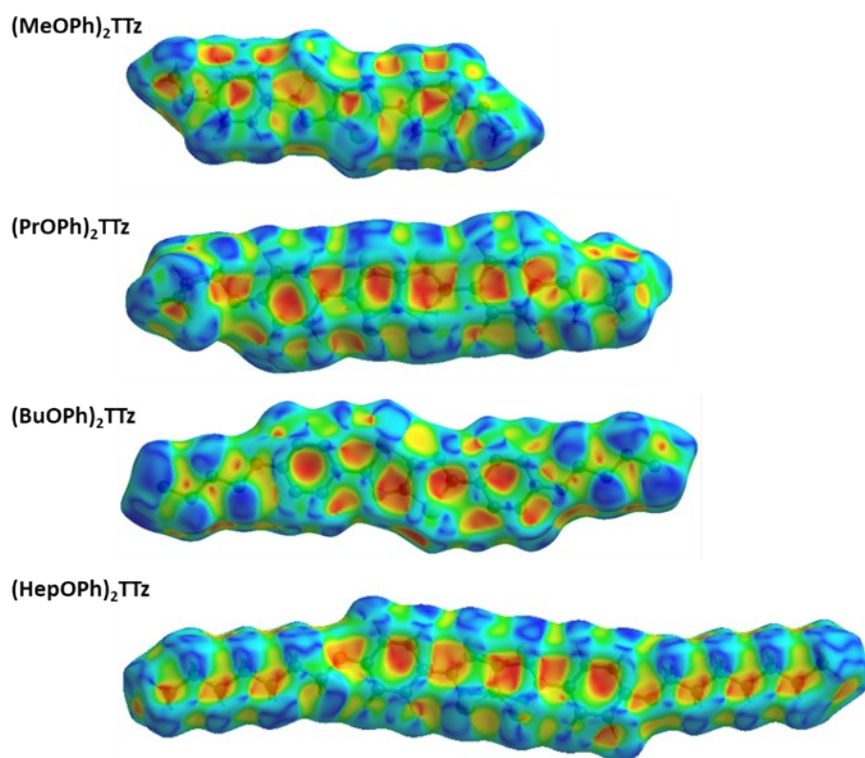


Figure S6. Shape index of the Hirshfeld surfaces of the $(C_nOPh)_2TTz$ crystals. π - π orbital overlap is represented by adjacent red and blue triangles on the Hirshfeld surface. The absence of adjacent red and blue triangles is indicative of lack of π - π orbital overlap.¹

Section 2.9: Powder X-Ray Diffractograms of TTz Crystals

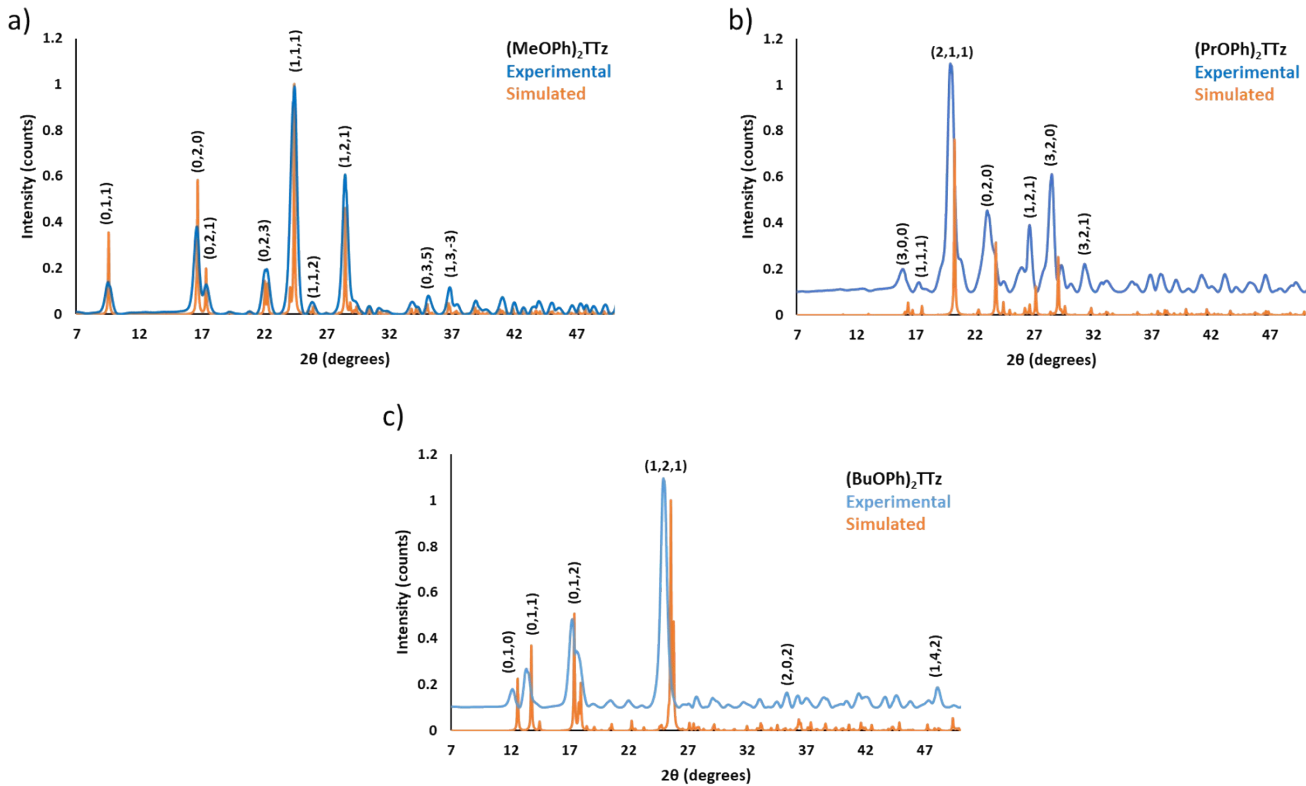


Figure S7: Comparison of experimental and simulated powder XRD patterns of a) (MeOPh)₂TTz, b) (PrOPh)₂TTz, c) (BuOPh)₂TTz.

Section 3: Photophysical Characterization

Section 3.1: DFT Simulated Raman Spectra

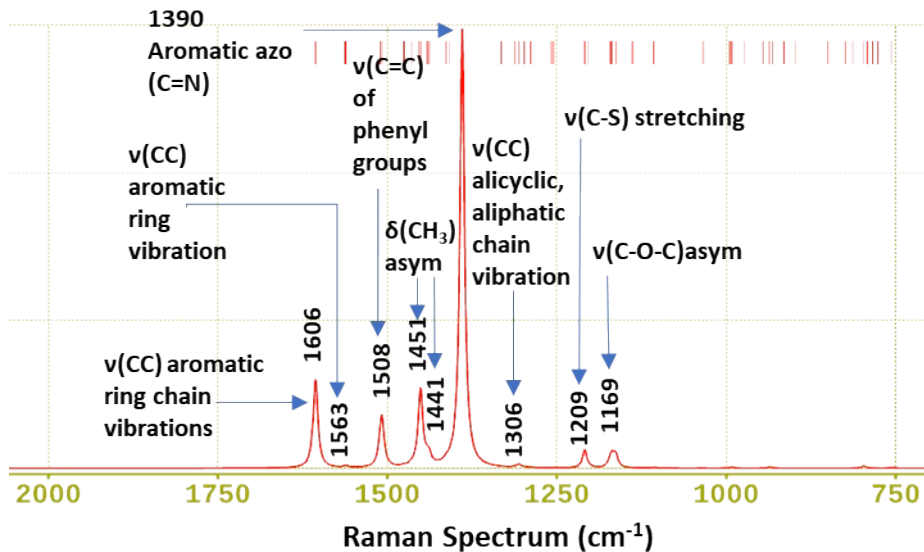


Figure S8: DFT Simulated Raman Spectra with peak assignments.

Section 3.2: Lattice-Phonon Raman Spectra of TTz Crystals

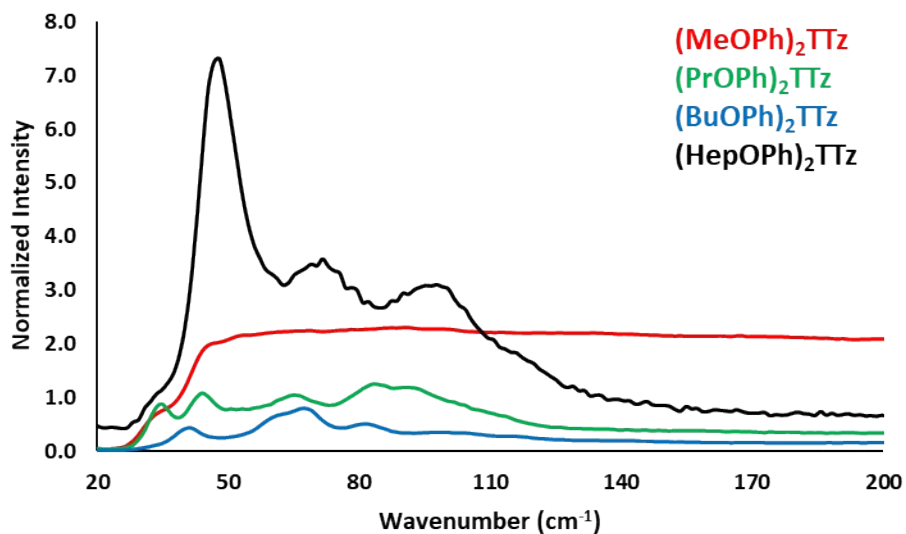


Figure S9: Lattice-Phonon Raman Spectra of TTz Crystals.

Section 3.3: Time Dependent Photostability Measurement of TTz Crystals

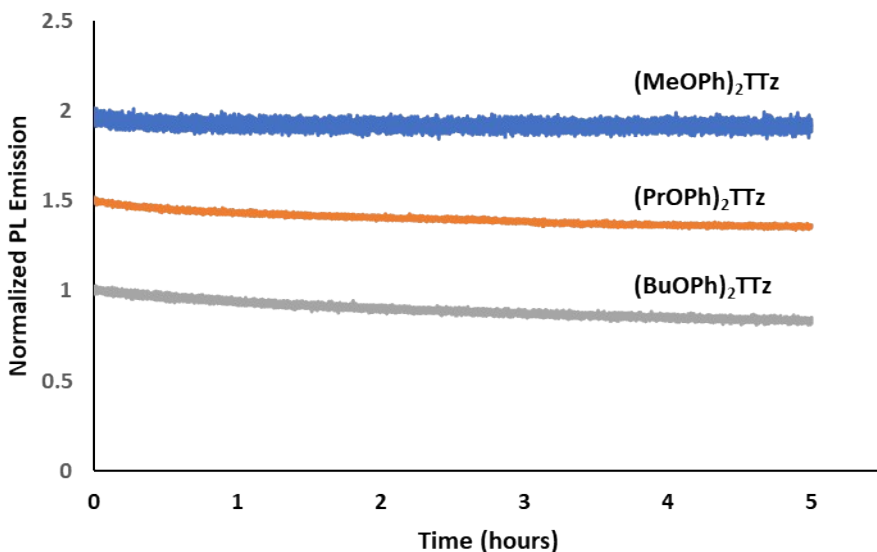


Figure S10: Time dependent photostability measurement of TTz Crystals under constant illumination by 400 nm in ambient conditions.

Material	Decrease in PL Intensity after 5 hours of illumination
(MeOPh) ₂ TTz	8 %
(PrOPh) ₂ TTz	14 %
(BuOPh) ₂ TTz	15 %

Section 4: References

(1) Shikhaliyev, N. Q.; Çelikesir, S. T.; Akkurt, M.; Bagirova, K. N.; Suleymanova, G. T.; Toze, F. A. Crystal structure and Hirshfeld surface analysis of (E)-1-(4-chlorophenyl)-2-[2, 2-dichloro-1-(4-fluorophenyl) ethenyl] diazene. *Acta Crystallographica Section E: Crystallographic Communications* **2019**, 75 (4), 465-469.