

Supporting Information

Non-volatile resistive switching mechanism in single-layer MoS₂ memristors: Insights from ab initio modelling of Au and MoS₂ interfaces

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1. Stability of defects in free-standing MoS₂

We assessed the effect of S vacancies, Au substitution, and different combinations of both on free-standing monolayer MoS₂ by computing formation energies of selected defects.

Table S1. Formation energies of selected defects in free-standing MoS₂.

Defect	E _F [MoS ₂] (eV)	E _F [V _S] (eV)
V _S	2.75	-
V _{S2} ^A	5.45	2.69
V _{S2} ^P	5.48	2.73
Au _S	3.48	0.72
Au2 _{S2} ^A	7.27	4.52
Au2 _{S2} ^P	6.64	3.88

The structural stability of defects can be assessed by computing their formation energies (see Table S1). To compute such a quantity, we considered two cases: i) a defect-free MoS₂, and ii) a MoS₂ sample with a V_S density of ~2%, in line with typical defect densities found in CVD-MoS₂. Formation energies computed with the latter approach (E_F[V_S]) give us an estimate of how likely it is to generate new S vacancies (and Au to fill such vacancies) when the material is already defective, as it is the case in MoS₂ nanodevices. Formation energies computed with defect-free MoS₂ (E_F[MoS₂]) confirmed that V_S is the most thermodynamically stable defect (E_F[MoS₂] = 2.75 eV), in accordance with previous studies. Additionally, it is worth noting that E_F[MoS₂] of an axial (V_{S2}^A) and an in-plane S di-vacancy (V_{S2}^P) is lower than the sum of E_F[MoS₂] for two infinitely separated S vacancies, which suggests the slight tendency of V_S to form clusters. We also found that Au_S has a higher E_F[MoS₂] (3.48 eV) than a simple S vacancy, which is the outcome of both the removal of an S atom and the addition of an Au atom from/to the defect-free MoS₂. We could also observe the tendency of Au atoms to cluster in-plane rather than axially. When considering the material with a V_S density of ~2%, we were able to confirm that additional S vacancies are slightly more likely to form in the proximity of

an already existing vacancy. Indeed, $E_F[V_S]$ for one additional S vacancy is slightly lower than $E_F[\text{MoS}_2]$ for the single S vacancy. Moreover, we found that the formation energy of Au_S significantly decreases ($E_F[V_S] = 0.72 \text{ eV}$), making this the most energetically favourable defect. Once again, we found that Au atoms tend to form small clusters in-plane rather than axially out-of-plane.

2. Partial Charges of Selected Interfaces

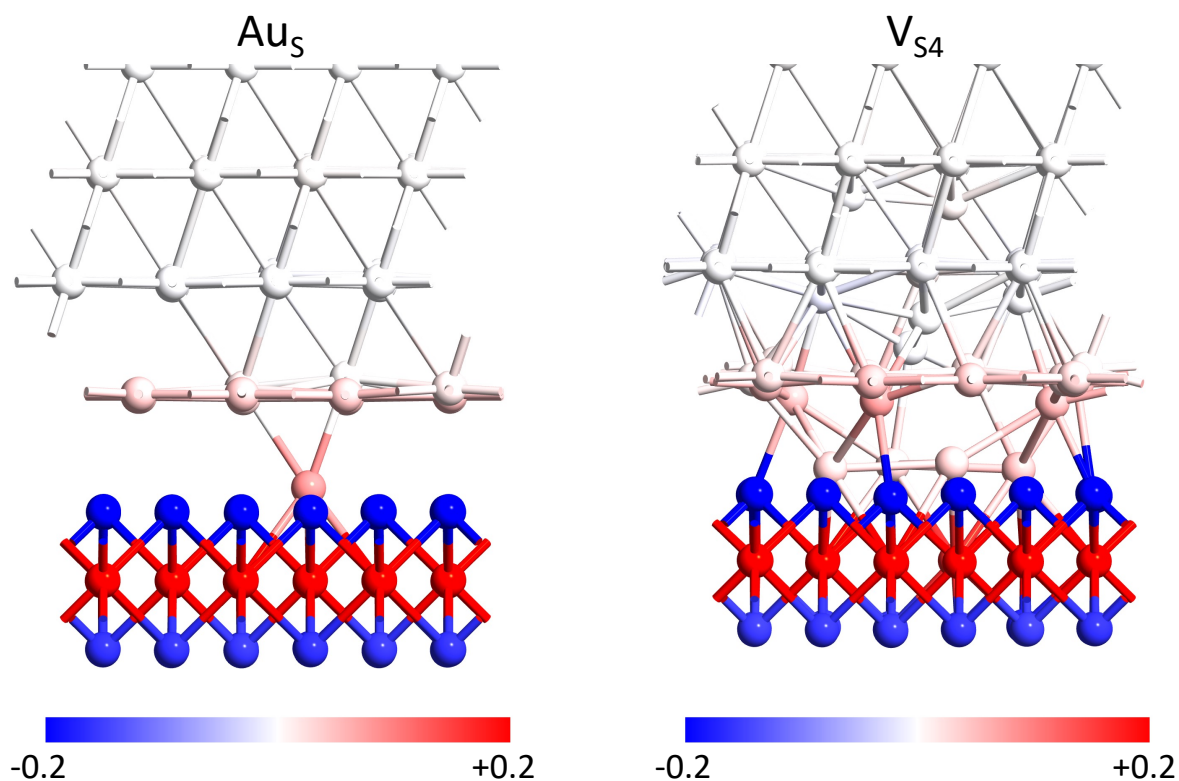


Figure S1. Mulliken population analysis of Au_S and V_{S4} interfaces. Negative partial charges are in blue, whereas positive partial charges are in red.