Supplementary information

Simple microplasma reactor paired with indirect ultrasonication for aqueous phase synthesis of cobalt oxide nanoparticles

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Characterization of materials

Absorption spectra were recorded using a Scinco (Mega-2100) double-beam UV-Vis spectrophotometer in the spectral range of 325-700 nm. The microstructures and morphologies of the samples were observed by scanning electron microscopy (FE-SEM) (TESCAN, MIRA3) and transmission electron microscopy (TEM) (Talos F200X G2). The chemical compositions were evaluated by energy-dispersive spectroscopy (TEM-EDS) measurements. The phase composition of the material was identified by X-ray diffraction (XRD) analysis performed using a PAN analytical Empyrean X-ray diffractometer with a Cu K α (λ = 0.15405 nm) radiation source and operating at 40 kV, 30 mA, with a scan range (2 θ) of 5° to 90°. A surface analysis study using X-ray photoelectron spectroscopy (XPS) was conducted using a Theta Probe angleresolved XPS system (Thermo Fisher Scientific Inc., U.K.) with monochromatic Al K α in the range of 1486.6–154 eV at 15 kV.



Figure S1: A) Precursor treated by microdisharge plasma for 25min. B) Color of precursor before and after reaction, Cobalt 0.01 M before microplasma treatment: (a) pH 7, (b) pH 10, and (c) pH 12 and Cobalt 0.01 M and pH 10 reaction time 25 min.: (d) MPR, (e) MPR-MS, (f) MPR-Son, (g) MPR-urea, and (h) MPR-Suc.



Figure S2: Current-voltage waveform: Inlet gas used air :(a) without microplasma formed and (b) with microplasma formed.



Figure S3: UV-Vis spectrum of cobalt oxide with treatment time 25 min from 1st day to 4 weeks



Figure S4. XRD pattern cobalt oxide without calcination