SUPPLEMENTARY MATERIAL

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Figure S1 Particle size distribution of the composite.



Figure S2 IFFT images of the composite.





Figure S3 IFFT images of CQDs.

Figure S4 The surface area plot of the composite



Figure S5 DSC analysis of the composite



Figure S6 Bar graph representation of variation of dye concentration with decay rate.



Figure S7 Representing the effect of light intensity on the decay rate.



Figure S8 Test of different scavengers on decay rate.





Figure S9 Kinetic study of the decay of CR.

Figure S10 GCMS Spectra of Degraded Dye Products.



Sr. No.	Dye	Dye structure	Molecular Formula	M.W.	The Wavelength of Maximum absorbance (λ _{max})	Procured from	CI No.
1.	Thymol Blue	HO HO HO SO3"	C ₂₇ H ₃₀ O ₅ S	466.5 9 g/mol	595 nm	Himedia	510 10
2.	Congo Red		$\begin{array}{c} C_{32}H_{22}Na_2N_6\\ O_6S_2 \end{array}$	696.6 6 g/mol	497 nm	Himedia	221 20

Table S1: Details of dyes including structure, CI number, and M.W.

Table S2:	Various b	ounds of water	specimen	evaluated	before and	after pho	otocatalytic
decay of T	Ъ.						

Bounds	Prior decay	After photocatalytic decay
рН	5.5	6.9
DO (ppm)	0.3	7.5
Salinity (ppt)	0.07	0.89
TDS (ppm)	39.8	282
Conductance (µS)	78	512
COD (mg/L)	495	47.8

Table S3:	Various	bounds of v	water specimen	evaluated	before and	after pho	otocatalytic
decay of C	CR.						

Bounds	Prior decay	After photocatalytic decay
рН	6.5	7.2
DO (ppm)	0.2	7.4
Salinity (ppt)	0.08	0.86
TDS (ppm)	48.3	276
Conductance (µS)	83	509
COD (mg/L)	432	38.8

Scheme S1: Mechanism trail of TB's decay via GCMS examination.

