

Electronic Supplementary Information for

**Hollow Ru/RuO₂ nanospheres with nanoparticulate shells
for high performance electrocatalytic oxygen evolution reactions**

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Fig. S1 Low magnification SEM images of H-MOP-Ru and HN-Ru/RuO₂.

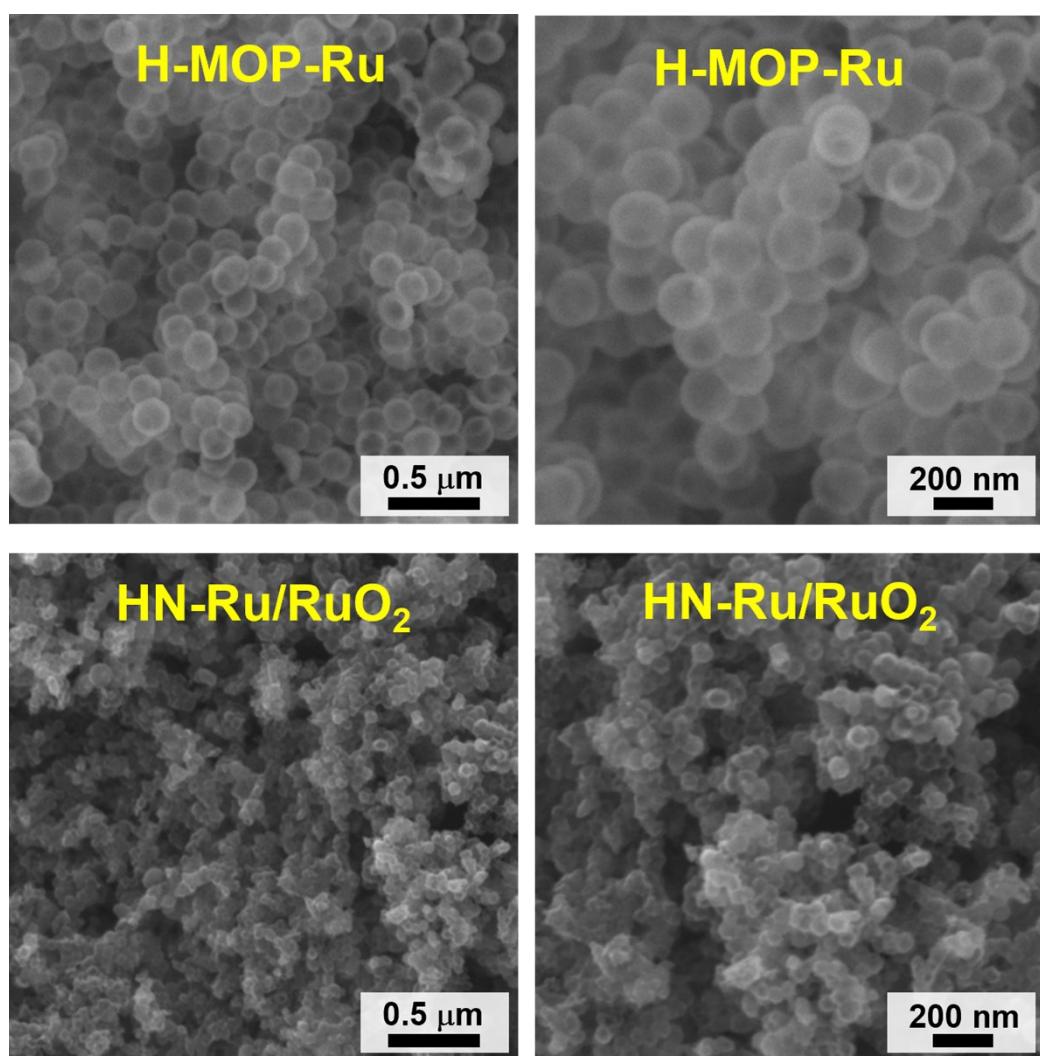


Fig. S2 Powder XRD patterns of H-MOP-CatP, H-MOP-Cat, and H-MOP-Ru.

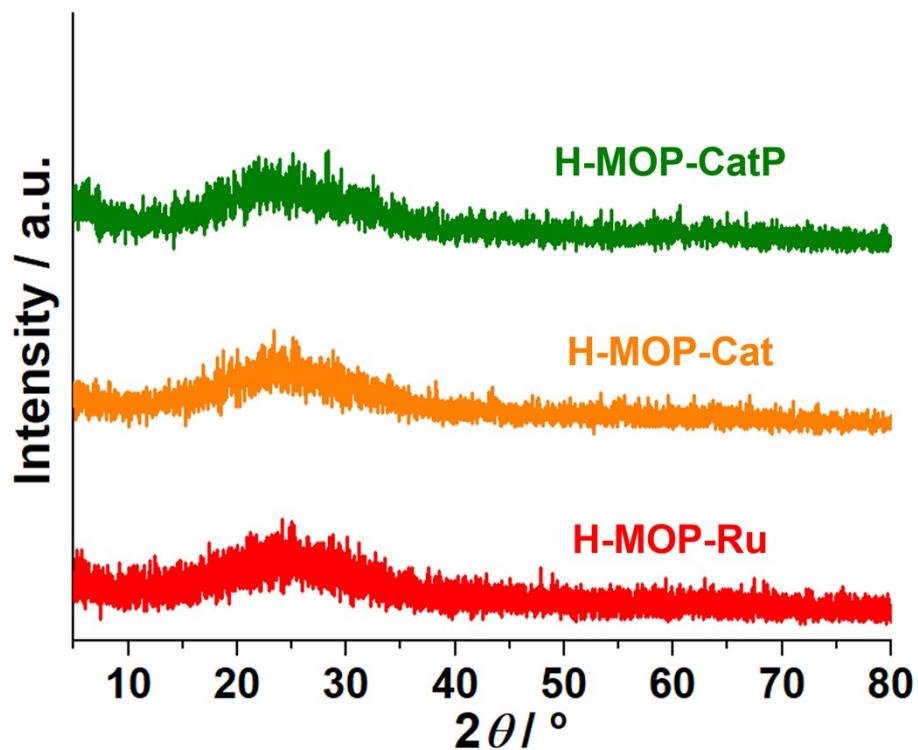


Fig. S3 TGA curves of H-MOP-CatP, H-MOP-Cat, and H-MOP-Ru.

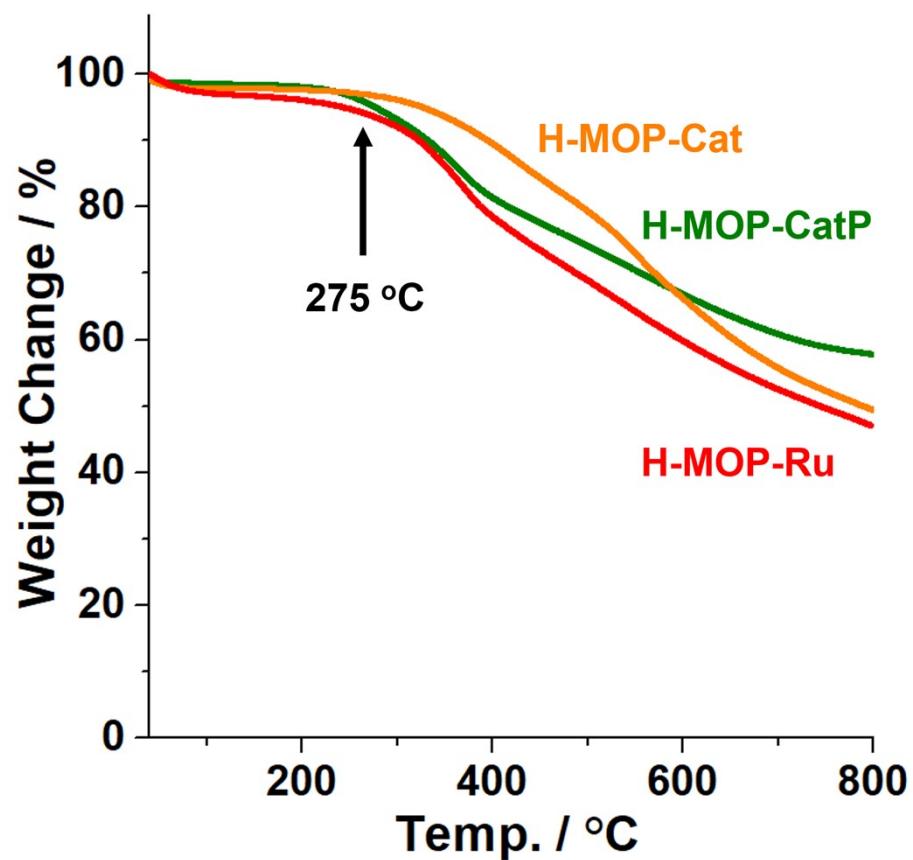
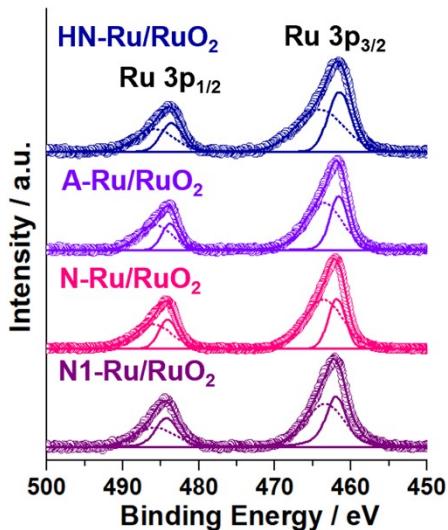
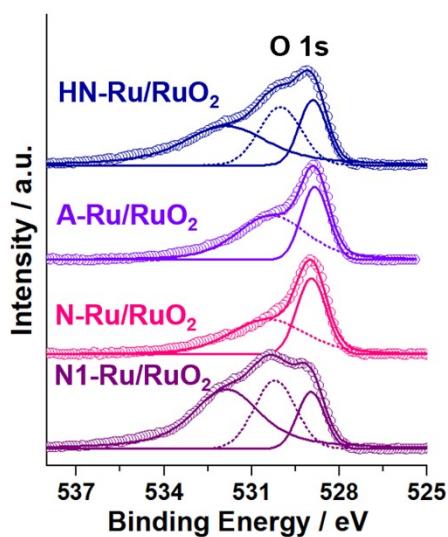


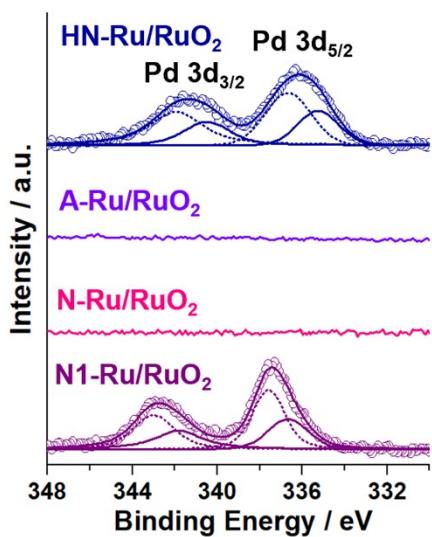
Fig. S4 XPS spectra of Ru 3p orbitals, O 1s orbitals, and Pd 3d orbitals of HN-Ru/RuO₂, A-Ru/RuO₂, N-Ru/RuO₂, and N1-Ru/RuO₂.



	Peak	p1	p2	p3	p4
HN-Ru/RuO ₂	Position (eV)	461.48	464.08	483.58	485.75
	FWHM	3.57	7.39	3.30	5.93
	Area (%)	28.2	41.5	12.4	17.9
	r ²			0.996	
A-Ru/RuO ₂	Peak	p1	p2	p3	p4
	Position (eV)	461.60	463.57	483.78	485.72
	FWHM	2.79	5.77	2.56	5.21
	Area (%)	24.6	44.7	10.5	20.2
N-Ru/RuO ₂	Peak	p1	p2	p3	p4
	Position (eV)	461.84	463.58	484.06	486.00
	FWHM	2.72	5.97	2.80	5.78
	Area (%)	20.4	45.2	12.3	22.1
N1-Ru/RuO ₂	Peak	p1	p2	p3	p4
	Position (eV)	461.95	463.41	484.23	485.72
	FWHM	3.23	6.04	3.46	6.21
	Area (%)	29.1	37.7	16.1	17.1
r ²					
0.997					



	Peak	p1	p2	p3
HN-Ru/RuO ₂	Position (eV)	528.91	530.02	531.90
	FWHM	1.11	1.71	3.47
	Area (%)	21.7	29.8	48.5
	r ²			0.997
A-Ru/RuO ₂	Peak	p1	p2	-
	Position (eV)	528.84	530.32	-
	FWHM	1.10	2.82	-
	Area (%)	36.5	63.5	-
N-Ru/RuO ₂	Peak	p1	p2	-
	Position (eV)	528.96	530.44	-
	FWHM	1.17	3.06	-
	Area (%)	42.8	57.2	-
N1-Ru/RuO ₂	Peak	p1	p2	p3
	Position (eV)	528.97	530.20	531.86
	FWHM	1.09	1.67	2.77
	Area (%)	16.5	30.1	53.4
r ²				
0.997				



	Peak	p1	p2	p3	p4
HN-Ru/RuO ₂	Position (eV)	335.24	336.66	340.48	341.94
	FWHM	2.28	2.55	2.58	2.99
	Area (%)	20.2	34.5	17.8	27.5
	r ²			0.992	
N1-Ru/RuO ₂	Peak	p1	p2	p3	p4
	Position (eV)	336.66	337.58	341.80	342.99
	FWHM	2.23	1.81	2.63	2.30
	Area (%)	19.2	33.3	18.0	29.5
r ²					
0.991					

Fig. S5 Scan rate-dependent cyclic voltammograms of HN-Ru/RuO₂, A-Ru/RuO₂, N-Ru/RuO₂, and N1-Ru/RuO₂.

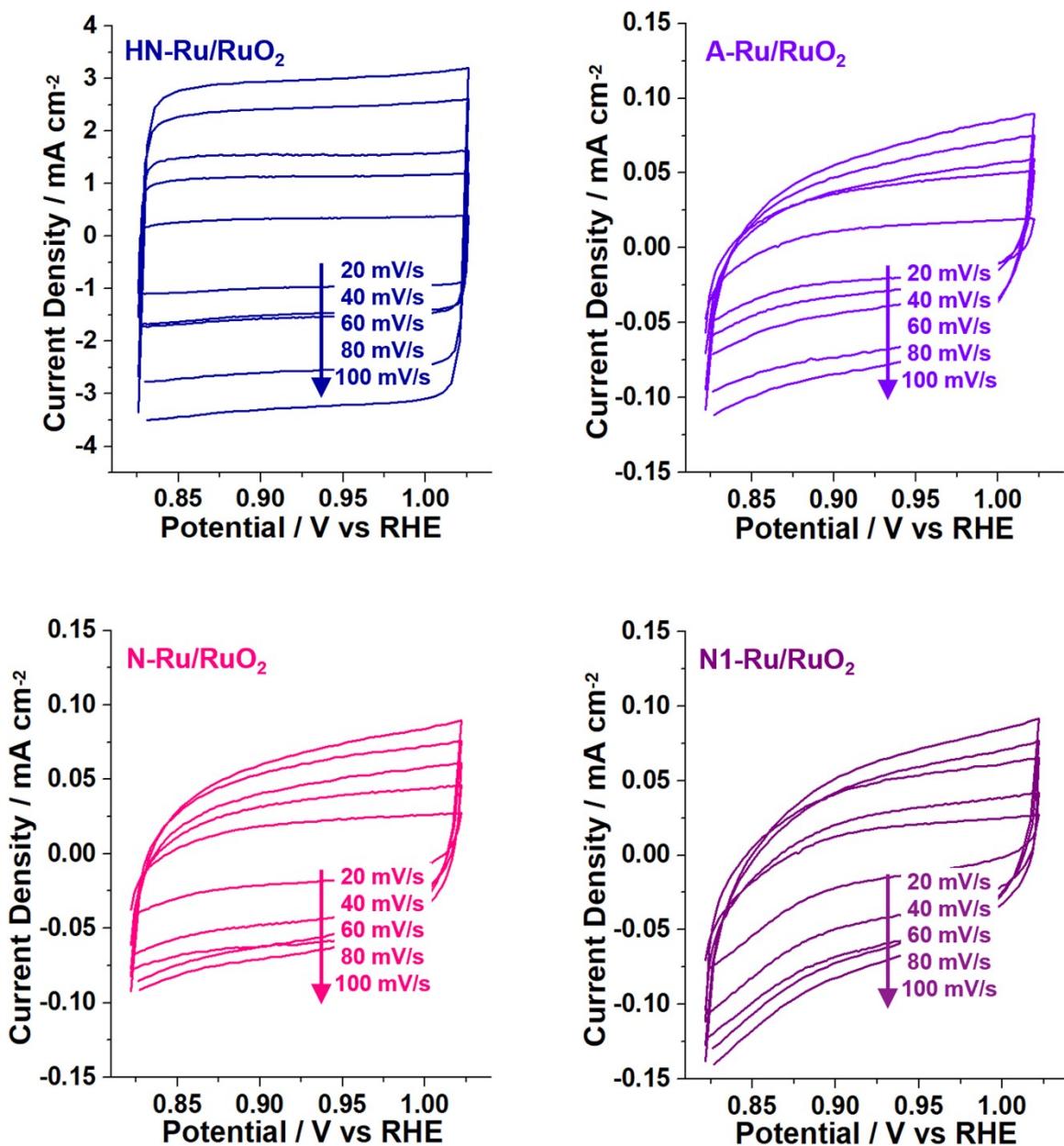


Fig. S6 Nyquist plots of HN-Ru/RuO₂, A-Ru/RuO₂, N-Ru/RuO₂, and N1-Ru/RuO₂ (Electrochemical impedance spectroscopy was conducted at the onset potential of 1.4 V vs RHE within a frequency range of 0.1~100 kHz with a signal amplitude of 10 mV).

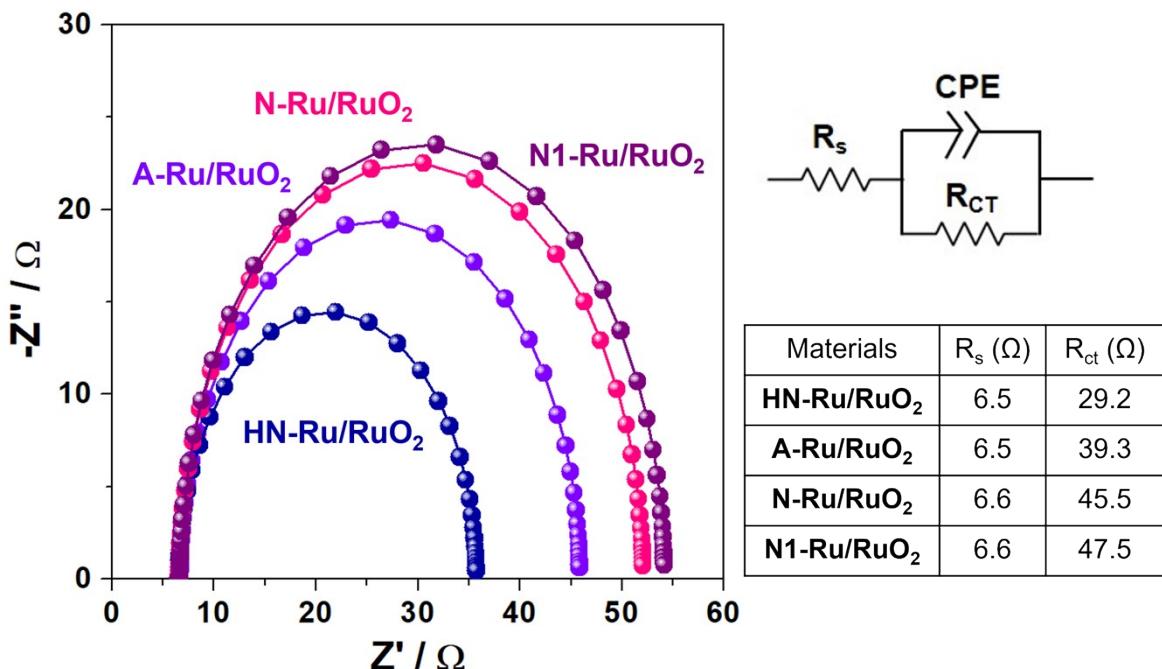


Fig. S7 (a) TEM and (b) XPS analysis of HN-Ru/RuO₂ retrieved after durability tests.

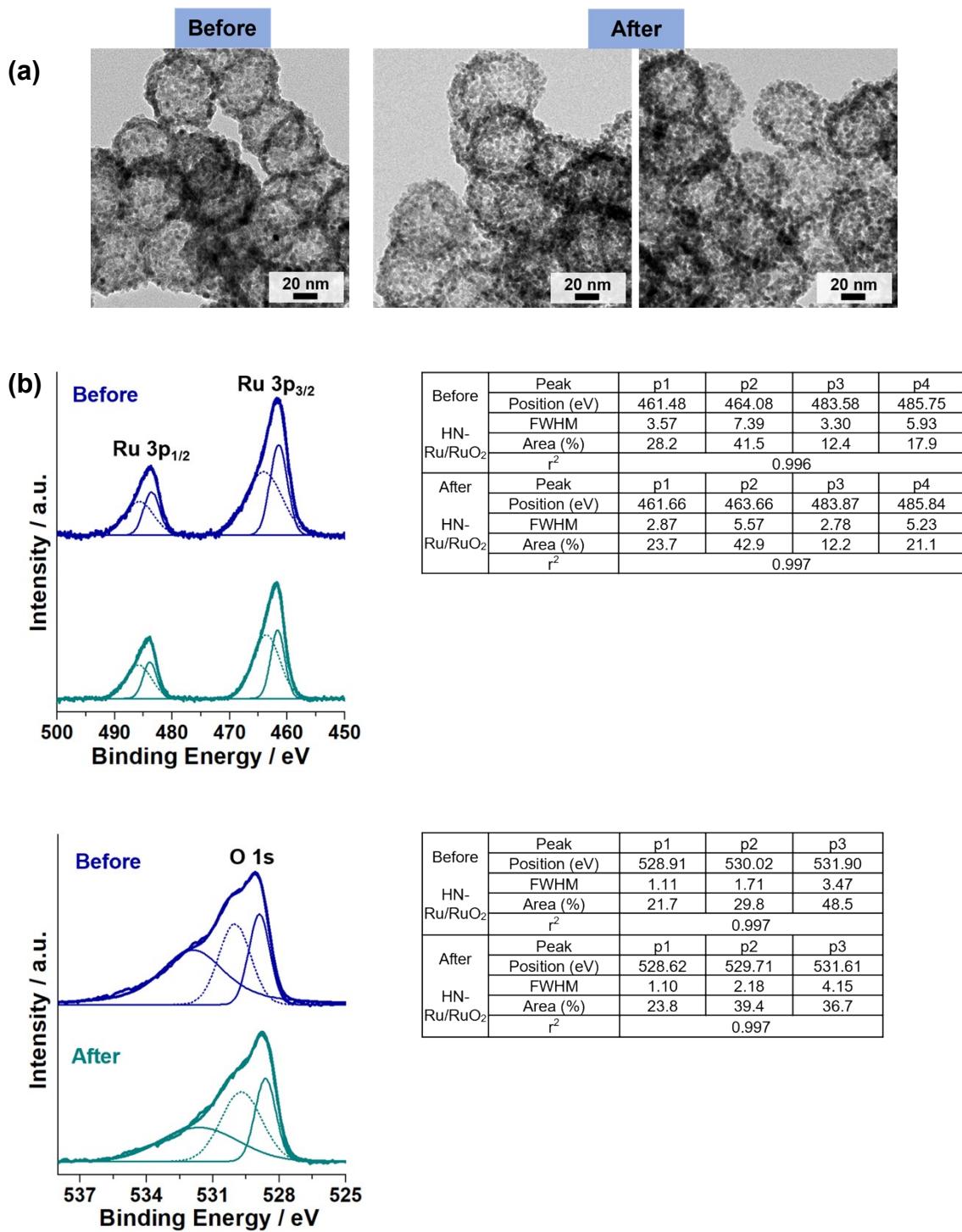


Fig. S8 HER performance of HN-Ru/RuO₂ and N-Ru/RuO₂ (Ar-saturated 1.0 M KOH, working electrode: graphite, reference electrode: Ag/AgCl): (a) HER polarization curves and overpotentials at 10 mA/cm² (scan rate: 10 mV/s) and (b) Tafel plots.

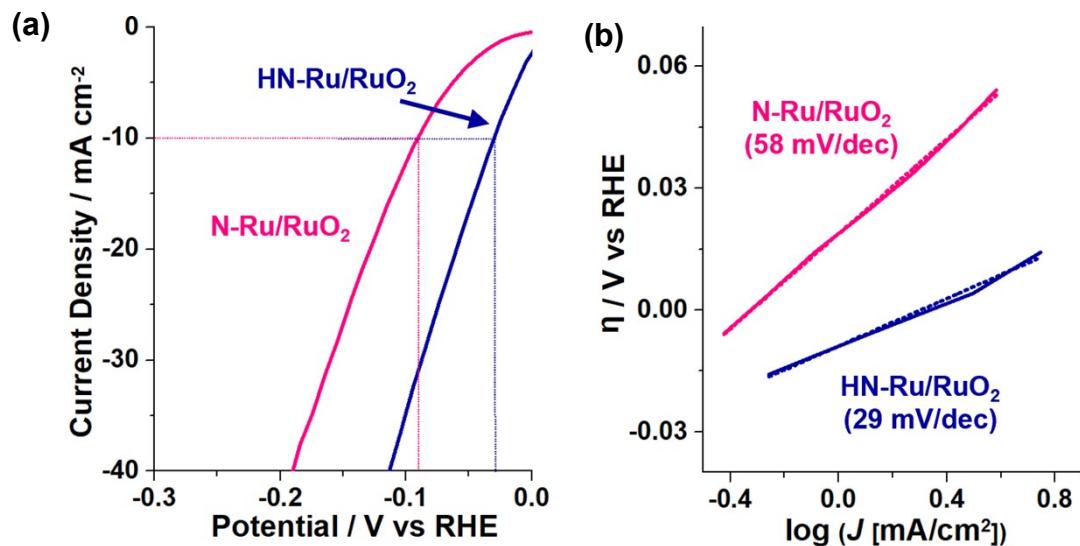


Table S1 Physical parameters of H-MOP-CatP, H-MOP-Cat, H-MOP-Ru, HN-Ru/RuO₂, A-Ru/RuO₂, N-Ru/RuO₂, and N1-Ru/RuO₂.

Materials	S _A _{BET} ^a (m ² /g)	V _{mic} ^b (cm ³ /g)	V _t ^c (cm ³ /g)
H-MOP-CatP	339	0.088	0.30
H-MOP-Cat	569	0.14	0.38
H-MOP-Ru	361	0.087	0.24
HN-Ru/RuO ₂	80	-	0.18
A-Ru/RuO ₂	10	-	0.019
N-Ru/RuO ₂	6	-	0.010
N1-Ru/RuO ₂	11	-	0.031

^a Surface area obtained by a BET plot. ^b Micropore volume obtained by a t-plot. ^c Total pore volume.

Table S2 Electrocatalytic performance of RuO₂-based materials for OER in alkaline electrolytes.

Entry	Materials	Strategy	Overpotential	Tafel slope	Electrolyte	Year	Ref
			@10 mA/cm ² (mV)	(mV/dec)			
1	RuO ₂ /NCNTs	Carbon composites	350	65	0.1 M KOH	2016	1
2	RuO ₂ /C nanoparticles	Carbon composites	-	69	0.1 M KOH	2017	2
3	RuO ₂ /N-C	Carbon composites	280	56	1 M KOH	2018	3
4	Ru/RuO ₂ nanorods	Nanoengineering	320	86	0.1 M KOH	2019	4
5	RuO ₂ /C	Carbon composites	300	116	0.1 M KOH	2019	5
6	Ru/RuO ₂ /CNT	Carbon composites	209	64	1 M KOH	2020	6
7	RuO ₂ /CeO ₂ nanoparticles	Metal oxide composites	350	74	1 M KOH	2020	7
8	RuO ₂ /F-doped graphene	Carbon composites	239	56	1 M KOH	2020	8
9	Ru/RuO ₂ /N-rGO	Carbon composites	255	89.4	1 M KOH	2020	9
10	CoOx-RuO ₂ /NF	Carbon composites	420	69.6	1 M KOH	2020	10
11	RuO ₂ /Co ₃ O ₄	Metal oxide composites	302	74.37	0.1 M KOH	2020	11
12	RuO ₂ nanoplates	Nanoengineering	400	82.5	1 M KOH	2022	12
13	HN-Ru/RuO ₂	Nanoengineering	295	46	1 M KOH	This work	

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