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Supplementary Information

Covalent Organic Frameworks with Triazine Units for Iodine

Capture via Weak Molecular Interaction

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Fig. S1. FT IR spectra of (a) T-COF-1 and (b) T-COF-2.



Fig. S2. FE SEM images of (a) (b) T-COF-1 and (c) (d) T-COF-2.



Fig. S3. PXRD pattens of (a) T-COF-1 and (b) T-COF-2 under different conditions.



Fig. S4. FT IR pattens of (a) T-COF-1 and (b) T-COF-2 under different conditions.



Fig. S5. The images of COFs before and after iodine capture: (a) T-COF-1 and (b) T-COF-2.



Fig. S6. The pseudo-first-order kinetic model of (c) T-COF-1 and (d) T-COF-2.



Fig. S7. Retention ability of (a) T-COF-1 and (b) T-COF-2.



Fig. S8. Raman spectra of (a) T-COF-1 and (b) T-COF-2.



Fig. S9. Iodine uptake of T-COF-2 (15 mg) in hexane (iodine: 1.5 mmol L⁻¹, 3 mL).



Fig. S10. Recycle ability of (a) T-COF-1 and (b) T-COF-2.



Fig. S11. FT IR spectra of (a) T-COF-1 and (b) T-COF-2.



Fig. S12. PXRD patterns of regenerated (a) T-COF-1 and (b) T-COF-2.



Fig. S13. FE SEM images of regenerated (a) T-COF-1 and (b) T-COF-2.

		C (%)	H (%)	N (%)
T-COF-1	Calcd.	86.79	4.40	8.81
	Found	86.01	4.92	8.16
T-COF-2	Calcd.	82.76	4.08	13.16
	Found	81.23	4.52	13.06

 Table S1. Elemental analysis of T-COF-1 and T-COF-2.

Samples	Temperature (°C)	I ₂ uptake (g g ⁻¹)	Rate constant	Ref.
T-COF-1	77	4.29		This
T-COF-2		4.72		work
TPB-DMTP COF		6.26	0.13	
TTA-TTB COF	75	4.95	0.14	S1
ETTA-TPA COF		4.79	0.12	
M-COF	77	4.56	-	S2
TPT-DHBD-COF	75	5.43	0.14	S3
TPT-Azine-COF	-	2.19	-	S4
SOF	-	4.46	-	S5
AzoPPN	77	2.90	0.18	S6
NTP	75	1.80	0.11	S7
NiP-CMP	75	2.02	0.17	S8

 Table S2. Iodine sorption capacity compared with other materials.

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