

Supporting Information File

Understanding the activity of single atom catalysts for CO₂ reduction to C₂ products: A high throughput computational screening

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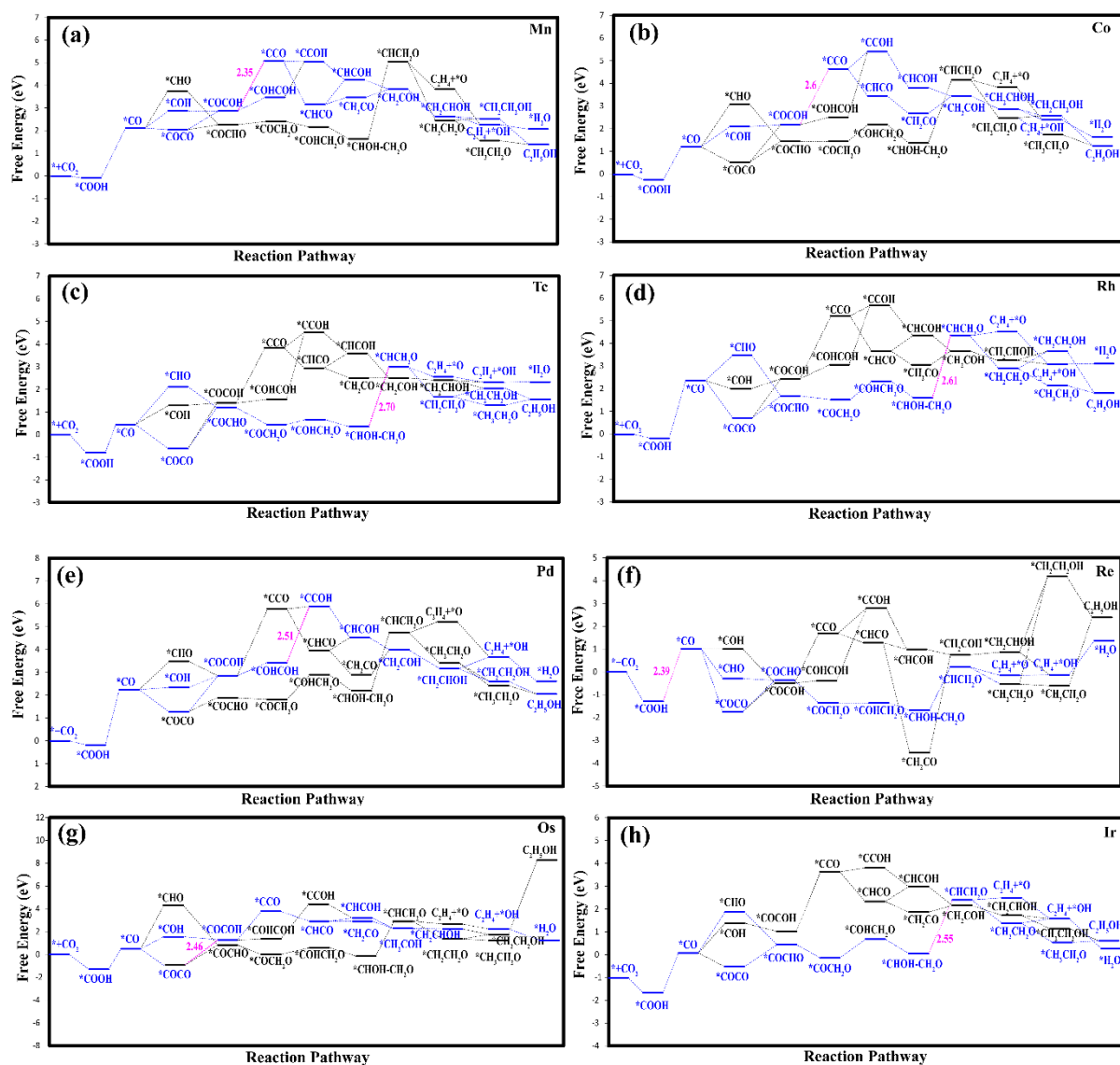
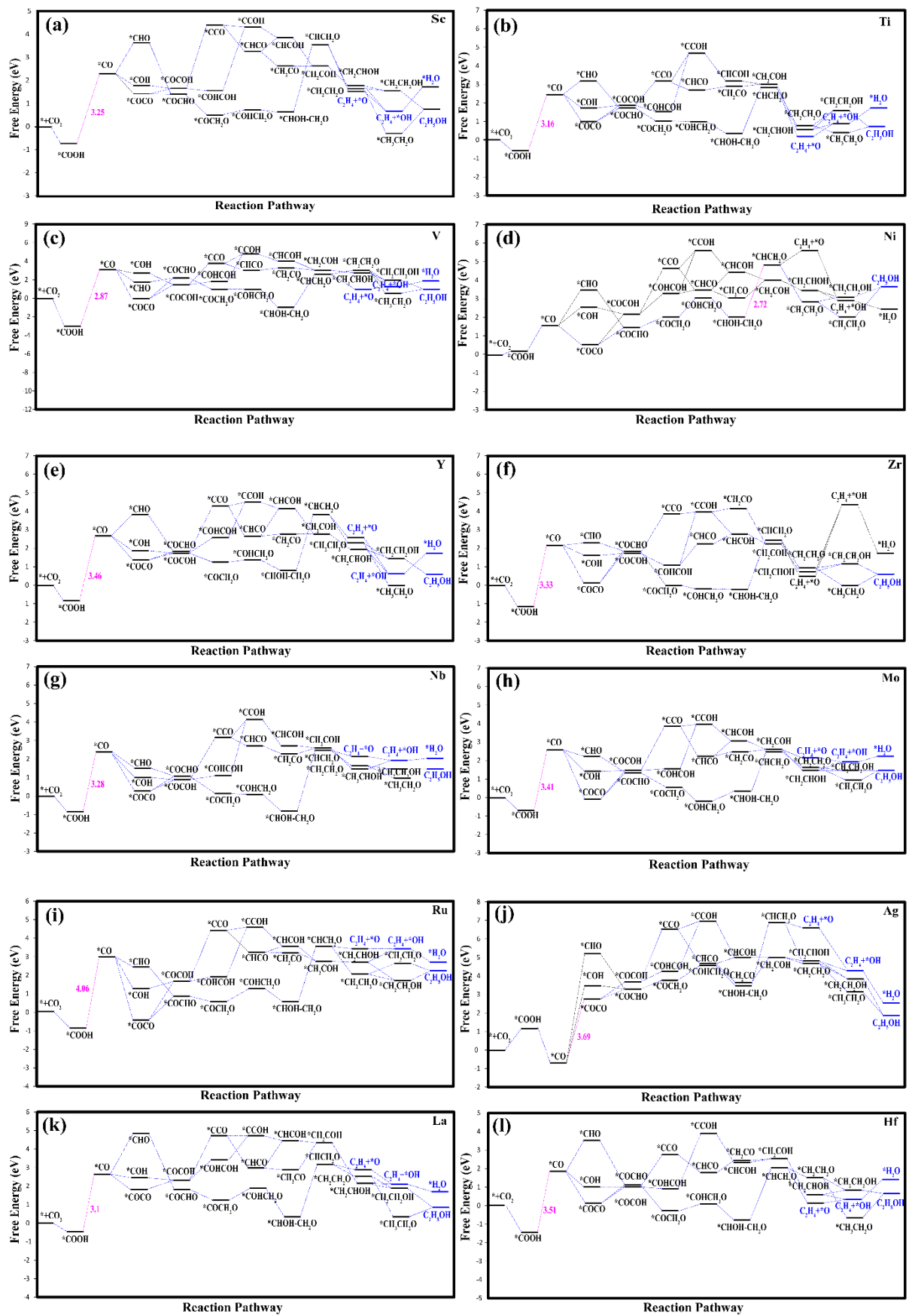


Fig. S1 The reaction Pathways for CO₂ reduction to C₂H₄ and C₂H₅OH on (a) Mn@C₂N, (b) Co@C₂N, (c) Tc@C₂N, (d) Rh@C₂N, (e) Pd@C₂N, (f) Re@C₂N, (g) Os@C₂N and (h) Ir@C₂N catalysts. The most favourable pathways are highlighted in blue colour.



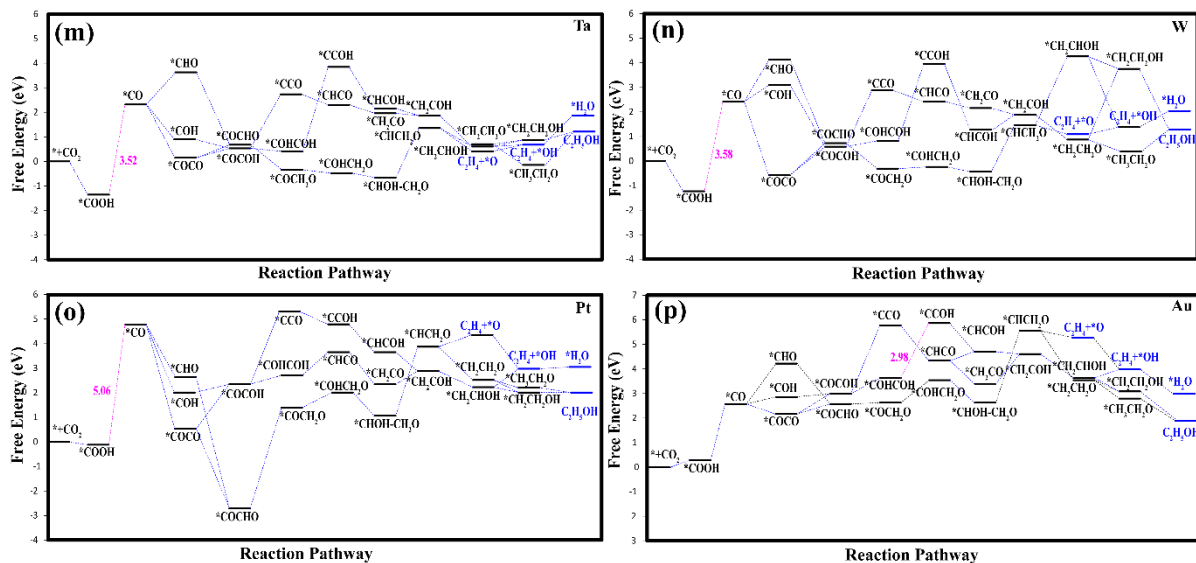


Fig. S2 The reaction Pathways for CO_2 reduction to C_2H_4 and $\text{C}_2\text{H}_5\text{OH}$ on (a) $\text{Sc@C}_2\text{N}$, (b) $\text{Ti@C}_2\text{N}$, (c) $\text{V@C}_2\text{N}$, (d) $\text{Ni@C}_2\text{N}$, (e) $\text{Y@C}_2\text{N}$, (f) $\text{Zr@C}_2\text{N}$, (g) $\text{Nb@C}_2\text{N}$ (h) $\text{Mo@C}_2\text{N}$, (i) $\text{Ru@C}_2\text{N}$, (j) $\text{Ag@C}_2\text{N}$, (k) $\text{La@C}_2\text{N}$, (l) $\text{Hf@C}_2\text{N}$, (m) $\text{Ta@C}_2\text{N}$, (n) $\text{W@C}_2\text{N}$, (o) $\text{Pt@C}_2\text{N}$ and (p) $\text{Au@C}_2\text{N}$ catalysts.

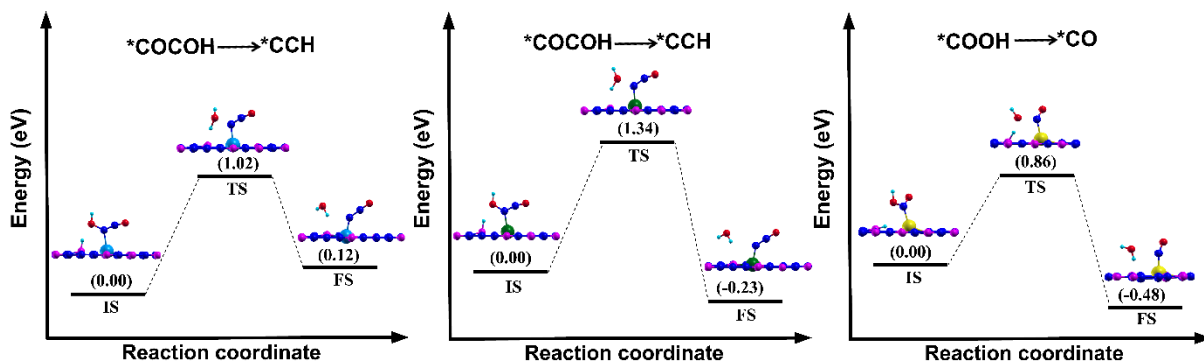


Fig. S3. Energy barrier for hydrogenating *COCHO to *CCO ($\text{Cr@C}_2\text{N}$ and $\text{Fe@C}_2\text{N}$) and *COOH to *CO ($\text{Cu@C}_2\text{N}$).

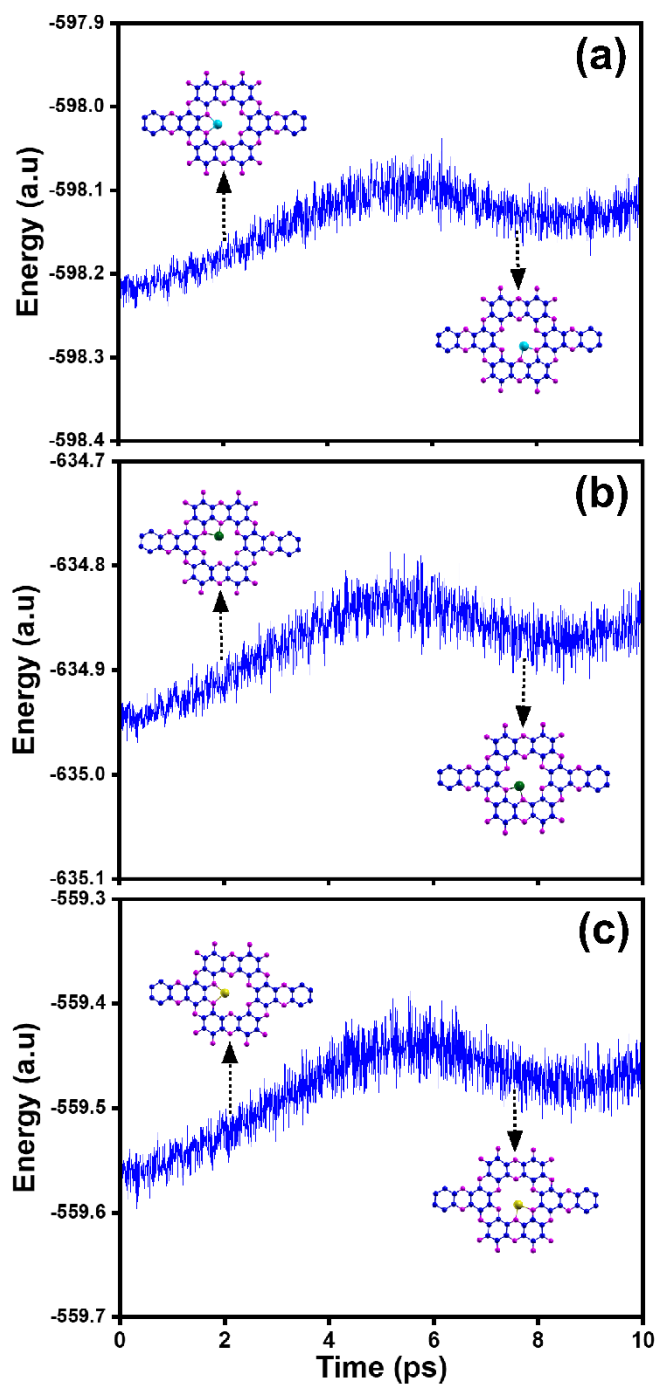


Fig. S4. Variation of potential energy with respect to the AIMD simulation time for a period of 10 picoseconds obtained using of NVT ensemble and Nose-Hoover thermostat for (a) Cr@C₂N, (b) Fe@C₂N and (c) Cu@C₂N catalysts at 600 K.

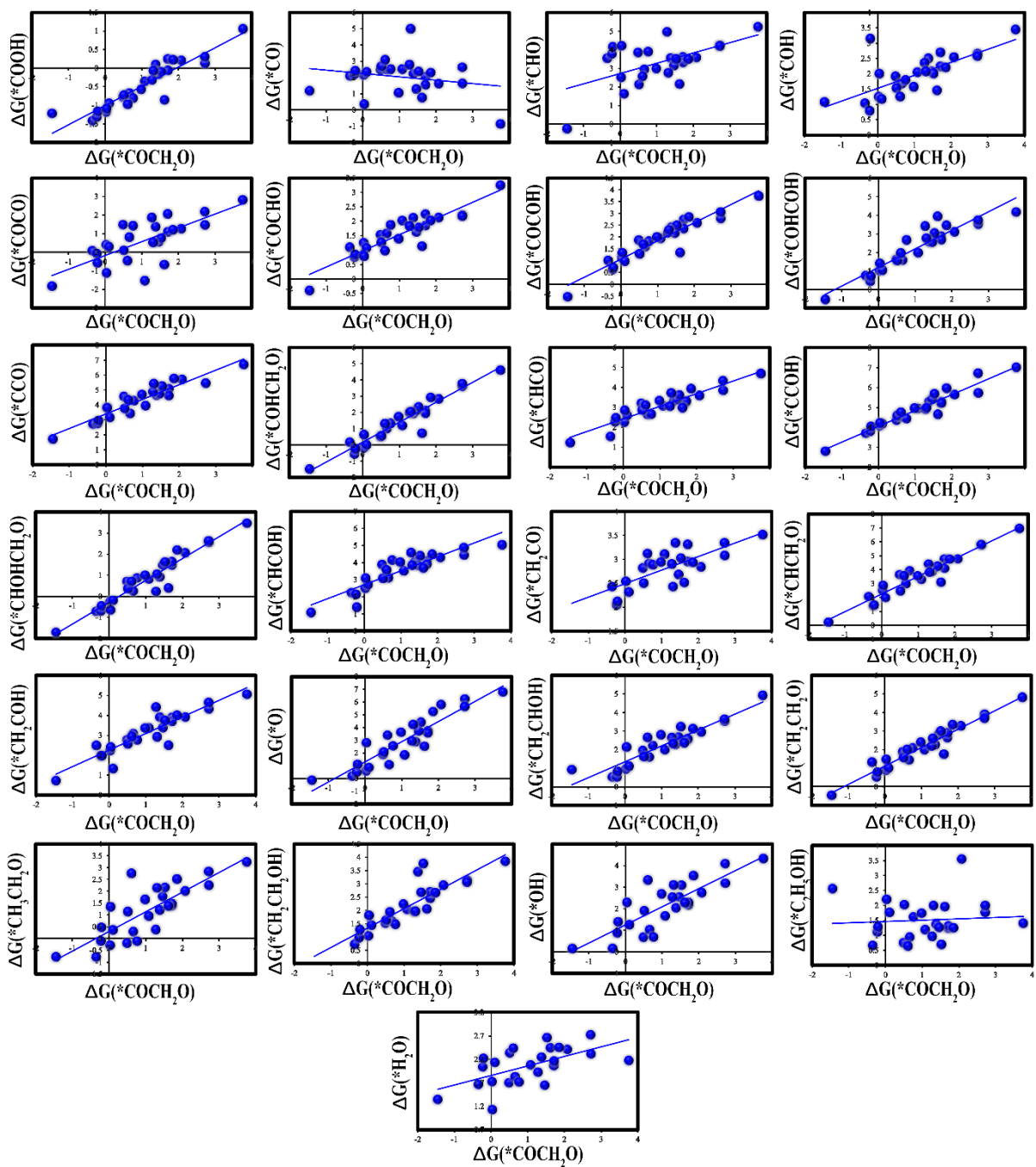


Fig. S5. Correlation between adsorption free energies of CO₂RR intermediates versus ΔG(*COCH₂O).

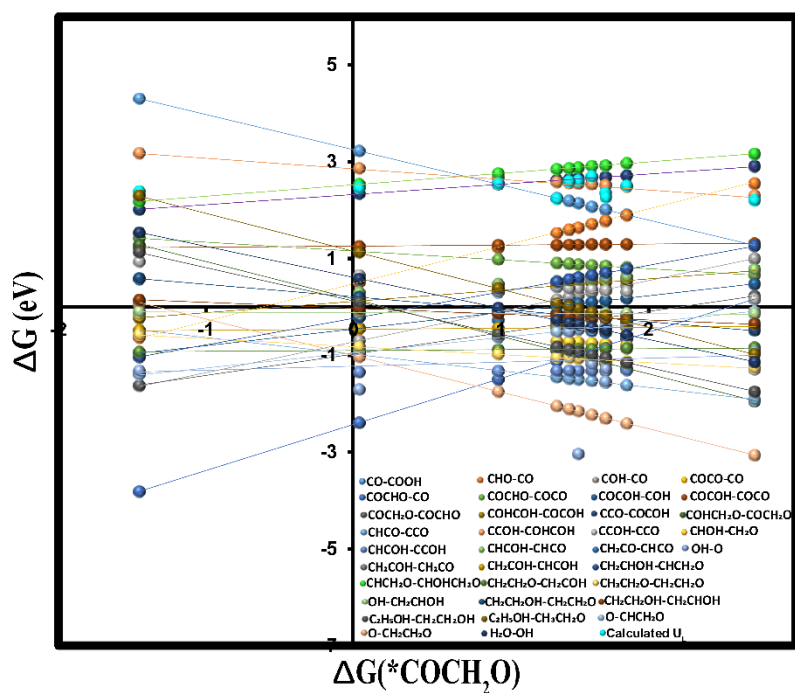


Fig. S6. Free energy change for each reaction step, predicted based on linear correlation between adsorption free energy of intermediates and $\Delta G_{(*\text{COCH}_2\text{O})}$.

Table S1. Adsorption free energies of CO₂ reduction intermediates on SACs supported on g-C₂N monolayer.

Catalyst	*COOH	*CO	*CHO	*COH	*COCO	*COCHO	*COCOHO
Sc	-0.78	2.47	3.86	1.91	1.48	1.52	1.87
Ti	-0.67	2.49	2.95	1.68	0.81	1.57	1.59
V	-0.36	2.51	0.01	2.05	-1.54	2.02	1.95
Cr	0.09	2.23	3.56	2.50	1.37	2.12	2.42
Mn	-0.06	2.04	3.62	2.70	2.06	2.25	2.74
Fe	0.25	1.54	3.32	2.24	1.09	1.85	2.36
Co	-0.16	1.27	3.14	2.08	0.56	1.62	2.16
Ni	0.21	1.61	3.58	2.54	1.26	2.13	2.60
Cu	0.13	1.62	4.24	2.65	1.47	2.21	2.79
Y	-0.81	2.51	3.90	1.79	1.43	1.87	1.82
Zr	-1.18	2.15	2.52	1.25	0.39	1.25	1.12
Nb	-0.95	2.33	1.62	1.17	0.30	1.08	0.98
Mo	-0.73	2.68	2.14	1.53	0.10	1.27	1.29
Tc	-0.86	0.75	2.15	1.46	-0.67	1.13	1.34
Ru	-0.98	3.08	2.58	1.25	-0.46	0.97	1.69
Rh	-0.10	2.37	3.50	2.01	0.76	1.78	2.50
Pd	0.22	2.29	3.49	2.21	1.20	2.03	2.85
Ag	1.06	-0.87	5.24	3.45	2.82	3.24	3.74
La	-0.32	2.78	4.96	2.36	1.86	1.83	2.18
Hf	-1.42	2.09	3.55	1.03	0.08	1.09	1.00
Ta	-1.31	2.21	3.78	0.79	-0.08	0.76	0.64
W	-1.16	2.42	4.16	3.14	-0.58	0.85	0.72
Re	-1.22	1.17	-0.25	1.07	-1.83	-0.40	-0.53
Os	-1.09	0.35	4.22	2.00	-1.12	0.78	1.33
Ir	-0.57	1.05	2.99	1.57	0.01	1.39	2.00
Pt	-0.07	4.99	2.76	2.08	0.53	0.01	2.30
Au	0.30	2.62	4.20	2.58	2.19	2.18	3.06

Catalyst	*COCH₂O	*COHCOH	*CCO	*COHCH₂O	*CHCO	*CCOH
Sc	0.49	1.62	4.56	0.52	3.20	4.36
Ti	0.66	1.59	3.44	1.01	2.65	4.53
V	1.08	1.97	3.96	1.19	3.08	4.95
Cr	1.38	2.56	4.65	1.93	3.31	5.30
Mn	1.71	2.88	5.08	2.12	3.26	5.28
Fe	1.71	2.68	4.63	1.94	3.32	5.23
Co	1.46	2.55	4.76	2.14	3.38	5.41
Ni	2.08	3.11	5.71	2.83	3.59	5.64
Cu	2.72	3.53	5.46	3.60	3.85	5.73
Y	0.76	2.67	4.28	1.33	2.66	4.44
Zr	0.03	1.02	3.82	-0.19	2.26	4.09
Nb	0.11	1.04	3.16	0.01	2.61	4.18
Mo	0.51	1.57	3.76	0.55	2.87	4.59
Tc	1.62	3.94	0.01	0.71	2.98	4.65
Ru	0.61	1.97	4.34	1.31	3.11	4.75
Rh	1.52	3.04	5.26	2.33	3.62	5.68
Pd	1.85	3.46	5.76	2.92	3.94	5.96
Ag	3.76	4.18	6.70	4.60	4.70	7.01
La	1.28	3.42	4.87	1.99	3.05	4.91
Hf	-0.35	0.75	2.71	0.17	1.54	3.70
Ta	-0.22	0.45	2.77	-0.57	2.27	3.81
W	-0.21	0.72	2.96	-0.25	2.48	4.05
Re	-1.45	-0.53	1.71	-1.49	1.23	2.81
Os	0.04	1.41	3.80	0.63	2.86	4.23
Ir	0.98	0.01	4.695	1.73	3.33	4.96
Pt	1.31	2.77	5.42	2.04	3.72	4.98
Au	2.72	3.74	0.01	3.77	4.34	6.72

Catalyst	*CHOH-CH ₂ O	*CHCOH	*CH ₂ CO	*CHCH ₂ O	*CH ₂ COH	*O
Sc	0.70	3.89	2.82	3.65	2.77	1.87
Ti	0.25	3.11	2.91	2.98	3.08	1.07
V	0.81	3.56	3.11	3.30	3.36	1.84
Cr	0.927	4.04	3.34	3.87	3.90	2.87
Mn	1.47	4.09	3.31	4.77	3.87	3.73
Fe	1.65	3.91	2.95	4.10	3.69	3.57
Co	1.47	3.85	2.68	4.17	3.38	3.73
Ni	2.06	4.30	2.84	4.78	3.91	5.79
Cu	2.55	4.45	3.08	5.77	4.32	6.23
Y	0.87	4.15	2.89	3.93	2.77	2.54
Zr	-0.26	2.51	2.54	2.52	2.23	0.58
Nb	-0.20	2.71	2.31	1.98	1.33	0.86
Mo	0.36	3.09	2.50	2.46	2.57	2.06
Tc	0.39	3.69	2.56	3.09	2.50	2.51
Ru	0.70	3.58	3.12	3.55	2.94	3.40
Rh	1.63	4.41	3.03	4.25	3.74	4.41
Pd	2.19	4.49	2.94	4.76	4.00	5.24
Ag	3.46	5.04	3.51	6.96	5.05	6.78
La	0.23	4.59	2.90	4.41	4.41	2.94
Hf	-0.72	2.23	2.43	2.07	2.48	0.18
Ta	-0.71	2.14	2.05	1.41	1.98	0.48
W	-0.46	1.39	2.12	1.46	1.97	1.09
Re	-1.71	1.08	0.01	0.22	0.72	-0.15
Os	-0.66	3.08	0.01	2.87	2.41	2.81
Ir	1.00	4.02	2.94	3.55	3.36	3.62
Pt	1.05	3.85	2.43	3.81	2.92	4.22
Au	2.63	4.88	3.35	5.82	4.65	5.64

Catalyst	*CH ₂ CHOH	*CH ₂ CH ₂ O	*CH ₃ CH ₂ O	*CH ₂ CH ₂ OH	*OH	*C ₂ H ₅ OH	*H ₂ O
Sc	1.96	1.88	-0.20	1.52	0.68	0.75	1.70
Ti	1.61	1.44	0.29	1.61	1.03	0.94	1.83
V	1.99	1.99	0.95	2.04	1.67	1.19	2.08
Cr	2.29	2.27	1.21	3.45	2.04	1.37	2.24
Mn	2.55	2.64	1.39	2.46	2.18	1.24	2.06
Fe	2.59	2.92	1.47	2.69	2.29	1.31	2.16
Co	2.66	2.81	1.79	2.68	2.55	1.26	1.65
Ni	2.95	3.28	2.02	2.95	2.75	3.55	2.41
Cu	3.53	3.90	2.25	3.13	3.19	1.77	2.32
Y	2.21	2.09	-0.11	1.48	0.69	1.61	1.71
Zr	0.96	0.88	-0.28	1.04	0.01	2.20	1.73
Nb	1.11	1.01	0.36	1.45	1.26	1.76	2.13
Mo	1.63	1.57	1.14	1.64	1.91	2.02	2.34
Tc	2.35	1.76	1.38	2.05	2.27	1.95	2.44
Ru	2.65	2.01	2.76	1.94	3.33	0.64	2.43
Rh	3.22	3.00	2.16	3.76	3.10	0.70	2.66
Pd	3.13	3.35	2.51	2.67	3.53	1.25	2.46
Ag	4.9	4.81	3.24	3.84	4.32	1.41	2.18
La	2.64	2.19	0.38	1.96	2.54	0.96	1.93
Hf	0.53	1.32	-0.7	0.73	0.16	0.66	1.67
Ta	0.52	0.52	-0.10	0.97	0.87	1.11	2.04
W	0.83	0.79	0.47	1.28	1.40	1.28	2.22
Re	0.92	-0.46	-0.77	0.01	0.15	2.56	1.34
Os	2.14	1.48	1.35	1.81	2.30	0.01	1.13
Ir	2.80	2.40	1.65	2.24	2.69	1.74	0.01
Pt	2.33	2.60	2.15	1.97	3.10	2.00	0.01
Au	3.65	3.69	2.84	3.07	4.09	1.99	2.72

Table S2. The linear correlation $y = a*x + b$ between adsorption free energy of intermediate species and that of $*COCH_2O$. R^2 represents the mean square error.

Species	a	b	R ²
*COOH	0.52	1.00	0.84
*CO	-0.21	2.25	0.05
*CHO	0.55	2.72	0.32
*COH	0.42	1.52	0.50
*COCO	0.74	-0.18	0.52
*COCHO	0.56	0.97	0.81
*COCOHO	0.77	1.07	0.89
*COHCOH	1.00	1.19	0.87
*CCO	0.98	3.40	0.87
*COHCH ₂ O	1.23	0.16	0.92
*CHCO	0.63	2.42	0.86
*CCOH	0.79	4.06	0.92
*CHOH-CH ₂ O	1.03	-0.28	0.92
*CHCOH	0.81	2.71	0.79
*CH ₂ CO	0.28	2.49	0.54
*CHCH ₂ O	1.27	2.25	0.91
*CH ₂ COH	0.83	2.26	0.82
*O	1.58	1.31	0.83
*CH ₂ CHOH	0.85	1.36	0.84
*CH ₂ CH ₂ O	0.99	1.12	0.92
*CH ₃ CH ₂ O	0.83	0.30	0.65
*CH ₂ CH ₂ OH	0.73	1.33	0.75
*OH	0.84	1.24	0.67
*C ₂ H ₅ OH	0.04	1.46	0.01
*H ₂ O	0.20	1.85	0.35

Table S3. Comparison between limiting potential values [in Volts (V)] for Cr, Mn, Fe, Co, Cu, Tc, Rh, Pd, Re, Os and Ir based catalysts predicted using linear correlation between adsorption free energies of different intermediates vs adsorption free energy of *COCH₂O and limiting potentials computed using the free energy pathways.

System	U _L (Predicted)	U _L (Calculated)
Cr	-2.85	-2.23
Mn	-2.93	-2.35
Fe	-2.93	-2.27
Co	-2.87	-2.60
Cu	-3.17	-1.50
Tc	-2.91	-2.70
Rh	-2.88	-2.61
Pd	-2.96	-2.51
Re	-4.30	-2.39
Os	-3.22	-2.46
Ir	-2.76	-2.55

*The predicted and calculated limiting potential is given by the following equation

$$U_L = \frac{-\Delta G_{Max.}}{e}$$