

## Supporting Information

### **Substituent and redox effects on the second-order NLO response of Ru(II) complexes with polypyridine ligands: A theoretical study**

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## General Comments

**Table S1** Electronic Energy (kJ/mol) calculated in two spin states of complexes 1-4 at the B3LYP/6-31G(d)/LANL2DZ level of theory.

**Table S2** Total first hyperpolarizabilities  $\beta_{\text{tot}}$  (a.u.) of all studied complexes calculated with two methods.

**Fig.S1** Spin density diagrams for one-electron reduced and oxidized species of complexes 1-4.

**Fig.S2** Simulated absorption spectra of complex 1 in acetonitrile solution by four functionals.

**Fig.S3** Molecular orbitals involved in crucial electronic transitions of complexes 1,  $1^+$  and  $1^-$ .

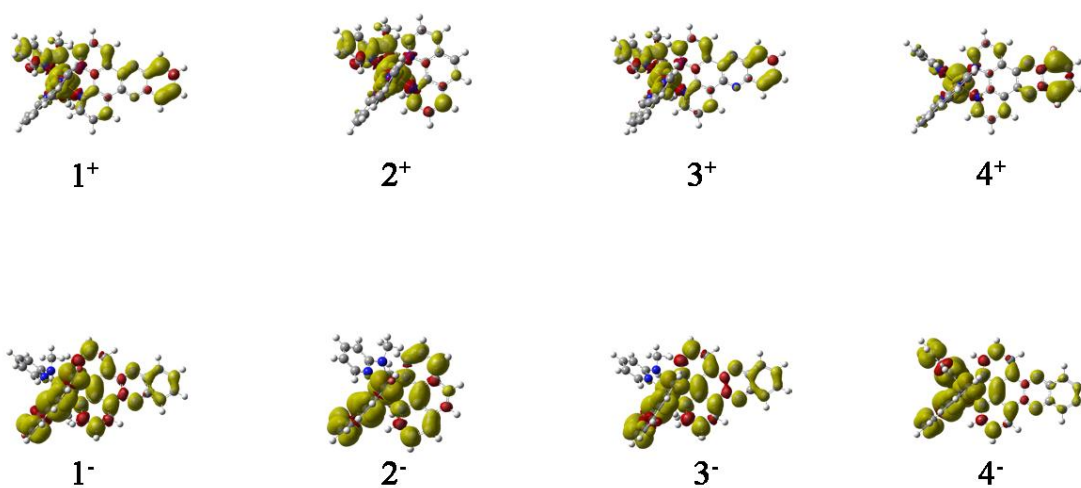
**Fig.S4** Molecular orbitals involved in crucial electronic transitions of complexes 2,  $2^+$  and  $2^-$ .

**Fig.S5** Molecular orbitals involved in crucial electronic transitions of complexes 3,  $3^+$  and  $3^-$ .

**Fig.S6** Molecular orbitals involved in crucial electronic transitions of complexes 4,  $4^+$  and  $4^-$ .

**Table S1** Electronic Energy (kJ/mol) calculated in two spin states of complexes 1-4 at the B3LYP/6-31G(d)/LANL2DZ level of theory.

Single electron	1	2	3	4
S=0	0.0	0.0	0.0	0.0
S=4	427.0	488.2	437.8	1001.3



**Fig.S1** Spin density diagrams for one-electron reduced and oxidized species of complexes 1–4.

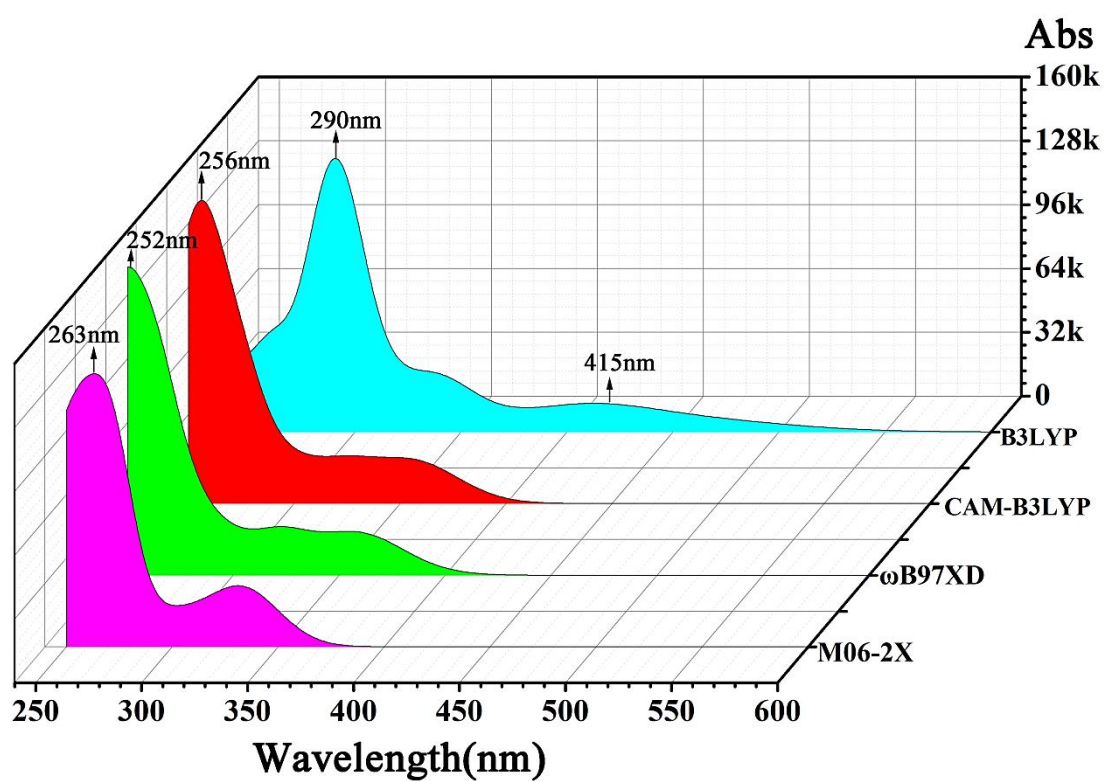


Fig.S2 Simulated absorption spectra of complex 1 in acetonitrile solution by four functionals.

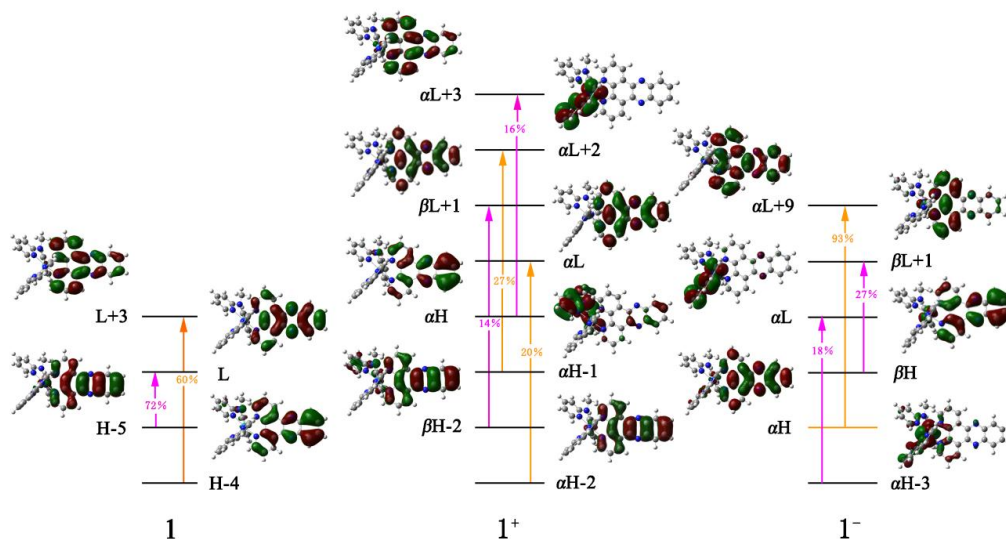


Fig.S3 Molecular orbitals involved in crucial electronic transitions of complexes 1, 1<sup>+</sup>, and 1<sup>-</sup>.

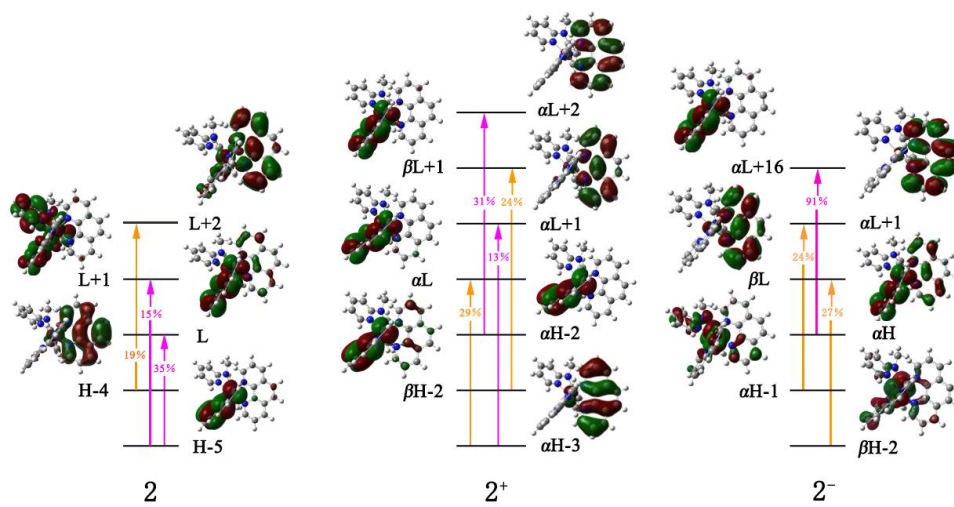


Fig.S4 Molecular orbitals involved in crucial electronic transitions of complexes 2, 2<sup>+</sup>, and 2<sup>-</sup>.

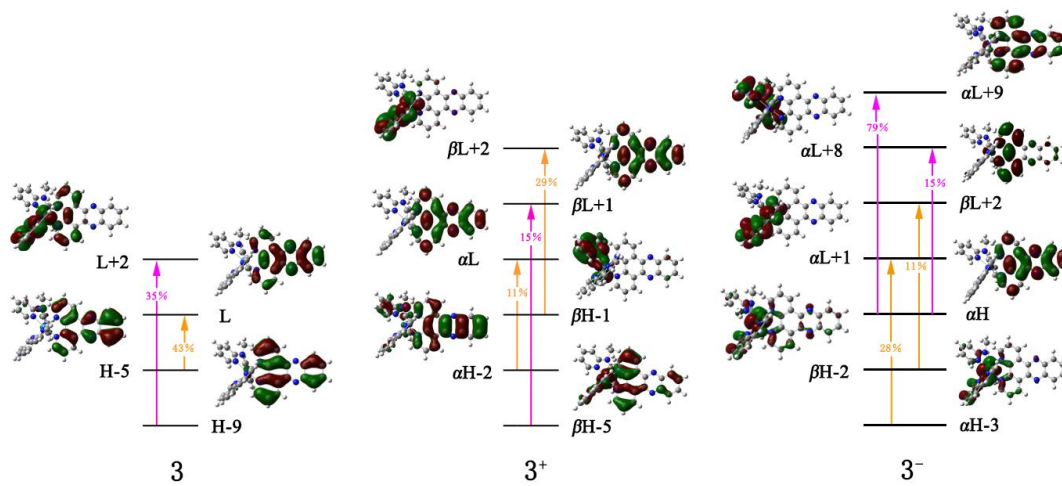
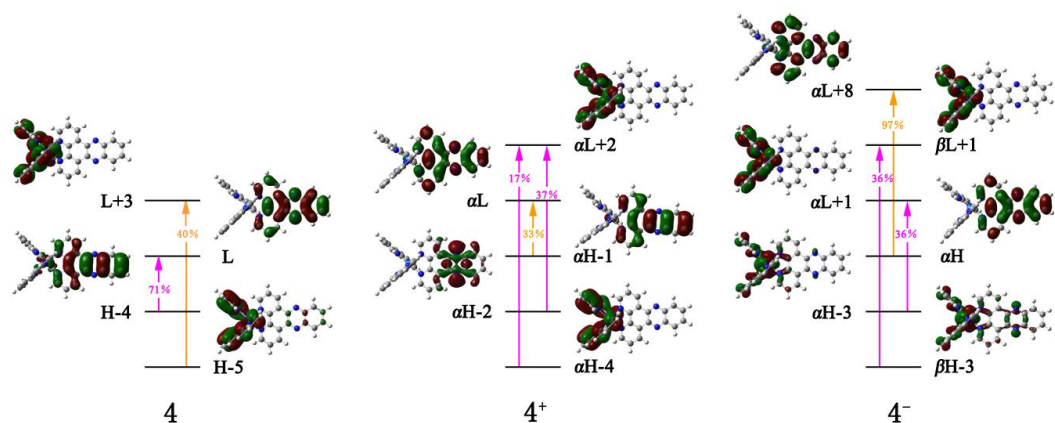


Fig.S5 Molecular orbitals involved in crucial electronic transitions of complexes 3, 3<sup>+</sup>, and 3<sup>-</sup>.



**Fig.S6** Molecular orbitals involved in crucial electronic transitions of complexes 4, 4<sup>+</sup>, and 4<sup>-</sup>.

**Table S2** Total first hyperpolarizabilities  $\beta_{\text{tot}}$  (a.u.) of all studied complexes calculated with two methods.

Complex	method	$\beta_{\text{tot}}$	Complex	method	$\beta_{\text{tot}}$
1 <sup>+</sup>	CAM-B3LYP	$6.24 \times 10^3$	3 <sup>+</sup>	CAM-B3LYP	$4.36 \times 10^3$
	$\omega$ B97XD	$6.83 \times 10^3$		$\omega$ B97XD	$6.18 \times 10^3$
1	CAM-B3LYP	$1.79 \times 10^3$	3	CAM-B3LYP	$1.67 \times 10^3$
	$\omega$ B97XD	$1.76 \times 10^3$		$\omega$ B97XD	$1.65 \times 10^3$
1 <sup>-</sup>	CAM-B3LYP	$1.44 \times 10^5$	3 <sup>-</sup>	CAM-B3LYP	$1.89 \times 10^5$
	$\omega$ B97XD	$1.09 \times 10^5$		$\omega$ B97XD	$1.58 \times 10^5$
2 <sup>+</sup>	CAM-B3LYP	$4.75 \times 10^3$	4 <sup>+</sup>	CAM-B3LYP	$3.12 \times 10^3$
	$\omega$ B97XD	$5.54 \times 10^3$		$\omega$ B97XD	$2.92 \times 10^3$
2	CAM-B3LYP	$1.53 \times 10^3$	4	CAM-B3LYP	$1.02 \times 10^3$
	$\omega$ B97XD	$1.57 \times 10^3$		$\omega$ B97XD	$8.36 \times 10^2$
2 <sup>-</sup>	CAM-B3LYP	$9.15 \times 10^3$	4 <sup>-</sup>	CAM-B3LYP	$5.36 \times 10^5$
	$\omega$ B97XD	$6.52 \times 10^3$		$\omega$ B97XD	$1.05 \times 10^5$