Optimization of the NH₂-UiO-66@MoS₂ Heterostructure for Enhanced Photocatalytic Hydrogen Evolution Performance

Zenghuan Ren, ^b Xinghao Zhang, ^a Xiaofan Shi, ^a Di Yang, ^a Mei-Hui Yu, ^a Wenjun

Zheng^{bc} and Jijie Zhang*a

^a School of Materials Science and Engineering, Tianjin Key Laboratory of Metal and

Molecule-Based Material Chemistry, Nankai University, Tianjin 300350, China.

^b Department of Chemistry, Key Laboratory of Advanced Energy Materials Chemistry

(MOE), TKL of Metal and Molecule-based Materials Chemistry, College of Chemistry, Nankai University, Tianjin 300071, P. R. China.

^c Collaborative Innovation Center of Chemical Science and Engineering (Tianjin), Nankai University, Tianjin 300071, P. R. China.

*Corresponding author. *E-mail address:* zhangjijie@nankai.edu.cn



Figure S1. SEM images of (a) C-NUM-0.05; (b) C-NUM-0.1; (c) C-NUM-0.2 and (d)

C-NUM-0.3.



Figure S2. PXRD pattern of (a) NH_2 -UiO-66 and (b) simulated NH_2 -UiO-66.



Fig. S3. Survey XPS spectra of C-NUM-0.2 and P-NUM



Figure S4. Hydrogen evolution performance of C-NUM-0.05; C-NUM-0.1; C-NUM-

0.2 and C-NUM-0.3.



Figure S5. SEM image of (a) C-NUM-0.2 and (b) P-NUM after photocatalytic reaction;(c) PXRD pattern of C-NUM-0.2 and P-NUM before and after photocatalytic reaction.



Figure S6. UV-vis spectra of C-NUM-0.05; C-NUM-0.1; C-NUM-0.2 and C-NUM-0.3.



Figure S7. PL spectra of C-NUM-0.05; C-NUM-0.1; C-NUM-0.2 and C-NUM-0.3.



Figure S8. EIS plots of C-NUM-0.05; C-NUM-0.1; C-NUM-0.2 and C-NUM-0.3.



Figure S9. I-t curves of MoS₂; C-NUM-0.2; P-NUM and NH₂-UiO-66(inset).



Figure S10. (a) TEM image of C-NUM-0.2/Pt, (b) HRTEM image of Pt nanoparticle,
(c) TEM image of C-NUM-0.2/MnO_x, (d) EDS line scan result of C-NUM-0.2/MnO_x

| Samples | Zr (wt %) | Mo (wt %) |
|------------|-----------|-----------|
| C-NUM-0.05 | 10.96 | 28.15 |
| C-NUM-0.1 | 6.70 | 34.36 |
| C-NUM-0.2 | 3.71 | 38.53 |
| P-NUM | 3.91 | 38.01 |
| C-NUM-0.3 | 2.72 | 36.39 |

Table S1. ICP-OES results for the contents of Zr and Mo.

| Samples | Adsorption (cm ³ /g) | Desorption (cm ³ /g) |
|-------------------------|---------------------------------|---------------------------------|
| NH ₂ -UiO-66 | 0.080 | 0.080 |
| MoS ₂ | 0.144 | 0.145 |
| C-NUM-0.2 | 0.077 | 0.077 |
| P-NUM | 0.079 | 0.081 |

Table S2. BJH results of NH₂-UiO-66, MoS₂, C-NUM-0.2 and P-NUM.

| | | | | | - |
|---|-------------------------------------|------------------------------|---|-----------|---|
| Sample | Test conditions | Light source | H ₂ production rate | Ref. | |
| C-NUM-0.2 | H2O/Na2S/Na2SO3 | Full spectrum | 3.509 mmol g ⁻¹ h ⁻¹ | This work | |
| 1TMoS ₂ /HCN-4 | TEOA | AM1.5G | 2724.2 mmol g ⁻¹ h ⁻¹ | [1] | |
| CN/MoS ₂ -1 | TEOA | λ≥400 nm | 441.3 µmol g ⁻¹ h ⁻¹ | [2] | |
| ac-MoS _x /TiO ₂ | Ethanol/H ₂ O | 365 nm LED | 3.43 mmol h ⁻¹ g ⁻¹ | [3] | |
| 1T-MoS ₂ /MIL-125-NH ₂ | MeCN/TEA/H ₂ O | $\lambda \ge 420 \text{ nm}$ | 1454 µmol g ⁻¹ h ⁻¹ | [4] | |
| MoS ₂ -MoC@rGO/TiO ₂ | CH ₃ OH/H ₂ O | UV light | 575 μ mol h ⁻¹ g ⁻¹ | [5] | |
| MoS ₂ /ZnO | H2O/Na2S/Na2SO3 | Visible light | 235 µmol g ⁻¹ h ⁻¹ | [6] | |
| CdS/MoS ₂ | H2O/Na2S/Na2SO3 | UV light | 5587 μmol g ⁻¹ h ⁻¹ | [7] | |
| g-C ₃ N ₄ -MoS ₂ -ZnNi-ZIF | TEOA | $\lambda \ge 420 \text{ nm}$ | 77.8 μmol g ⁻¹ h ⁻¹ | [8] | |

 Table S3. Similar reported photocatalyst

| Samples | $\mathrm{R}_{1}\left(\Omega ight)$ | $\mathrm{R}_{2}\left(\Omega ight)$ |
|-------------------------|------------------------------------|------------------------------------|
| NH ₂ -UiO-66 | 42.94 | $1.27*10^{7}$ |
| MoS_2 | 42.53 | 2456 |
| C-NUM-0.2 | 35.27 | 1479 |
| P-NUM | 42.85 | 2343 |

Table S4. Parameters of equivalent circuit for the impedance data of NH₂-UiO-66,

MoS₂, C-NUM-0.2 and P-NUM.

Table S5. Parameters of equivalent circuit for the impedance data of C-NUM-0.05;

| Samples | $\mathrm{R}_{1}\left(\Omega ight)$ | $\mathrm{R}_{2}\left(\Omega ight)$ |
|------------|------------------------------------|------------------------------------|
| C-NUM-0.05 | 42.68 | 4434 |
| C-NUM-0.1 | 35.29 | 1897 |
| C-NUM-0.2 | 35.27 | 1479 |
| C-NUM-0.3 | 38.63 | 2538 |

C-NUM-0.1; C-NUM-0.2 and C-NUM-0.3

Reference

- [1] Y. Xiong, T. Liu, W. Liu, X. Wang, Y. Xue and J. Tian, *Int. J. Hydrogen Energy*, 2023, 48, 7284-7293.
- [2] Y. Xu, J. Ouyang, L. Zhang, H. Long, Y. Song and Y. Cui, *Chem. Phys. Lett.*, 2023, 814, 140331.
- [3] L. Li, D. Gao, F. Chen, X. Wang and H. Yu, Appl. Surf. Sci., 2023, 608,155173.

- [4] T. Nguyen, S. Kampouri, B. Valizadeh, W. Luo, D. Ongari, O. Planes, A. Zuttel, B. Smit and K. Stylianou, ACS Appl. Mater. Interfaces, 2018, 10, 30035-30039.
- [5] M. Pan, L. Gao, P. Wang, X. Wang and H. Yu, J. Alloys Compd., 2023, 939, 168721.
- [6] Y. Hunge, A. Yadav, S. Kang, S. Jun Lim and H. Kim, *J. Photochem. Photobiol.*, *A*, 2023, **434**, 114250.
- [7] L. Lin, S. Huang, Y. Zhu, B. Du, Z. Zhang, C. Chen, X. Wang and N. Zhang, *Dalton Trans.*, 2019, 48, 2715-2721.
- [8] N. Arif, Y. Z. Lin, K. Wang, Y. Dou, Y. Zhang, K. Li, S. Liu and F. T. Liu, RSC Adv., 2021, 11, 9048-9056.