

*Supporting Information*

Molecular electrostatic potential and volume-aided drug design based on an isoindolinone-containing cyclopeptide S-PK6

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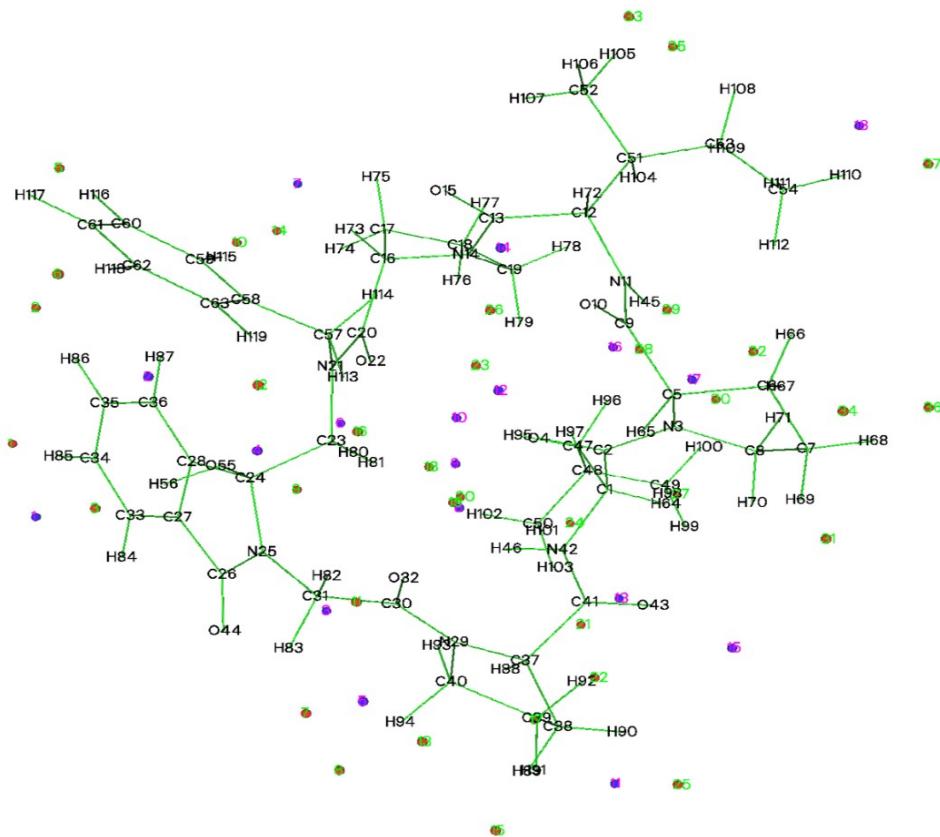
<sup>†</sup> Lei Zhao and Tingting Li contributed equally to this work.

Content	Page
<b>Table S1.</b> Electrostatic potential distribution diagram of compound <b>1</b> (S-PK6).	<b>S3</b>
<b>Fig.S1.</b> Distribution of the surface electrostatic potential of compound <b>1</b> (S-PK6) with numbered atoms. The blue points with pink numbers represent the minimum extreme, and the red points with green number represent the maximum extreme.	<b>S3</b>
<b>Table S2.</b> Electrostatic potential distribution diagram of compound <b>2</b> .	<b>S4</b>
<b>Fig.S2.</b> Distribution of the surface electrostatic potential extreme of compound <b>2</b> with numbered atoms. The blue points with pink numbers represent the minimum extreme, and the red points with green number represent the maximum extreme.	<b>S4</b>
<b>Table S3.</b> Electrostatic potential distribution diagram of compound <b>3</b> .	<b>S5</b>
<b>Fig.S3.</b> Distribution of the surface electrostatic potential extreme of compound <b>3</b> with numbered atoms. The blue points with pink numbers represent the minimum extreme, and the red points with green numbers represent the maximum extreme.	<b>S5</b>
<b>Fig.S4.</b> The 3D molecular docking between compounds <b>4(A), 5(B), 6(C), 7(D), 8(E),</b> and <b>9 (F)</b> with MDM2.	<b>S6</b>
<b>Fig.S5.</b> The 2D molecular docking sites for compounds <b>4(A), 5(B), 6(C), 7(D), 8(E),</b> and <b>9(F)</b> .	<b>S7</b>
<b>Fig.S6.</b> The 3D molecular docking for compounds <b>11(A), 12(B), 13(C), 14(D), 15(E), 16(F), 17(G),</b> and <b>18(H)</b> .	<b>S8</b>
<b>Fig.S7.</b> The 2D molecular docking sites for compounds <b>11(A), 12(B), 13(C), 14(D), 15(E), 16(F),</b> <b>17(G),</b> and <b>18(H)</b> .	<b>S9</b>
<b>Fig.S8.</b> The 3D molecular docking between compounds <b>19(A), 20(B), 21(C), 22(D), 23(E),</b> and <b>24(F)</b> with MDM2.	<b>S10</b>
<b>Fig.S9.</b> The 2D molecular docking sites between compounds <b>19(A), 20(B), 21(C), 22(D), 23(E),</b> and <b>24(F)</b> with MDM2.	<b>S11</b>
<b>Fig.S10.</b> Distribution of the surface electrostatic potential extreme of compound <b>19</b> with numbered atoms. The blue points with pink numbers represent the minimum extreme, and the red points with green number represent the maximum extreme.	<b>S12</b>
<b>Table S4.</b> Electrostatic potential distribution diagram of compound <b>19</b> .	<b>S12</b>
<b>Fig.S11.</b> Distribution of the surface electrostatic potential extreme of compound <b>20</b> with numbered atoms. The blue points with pink numbers represent the minimum extreme, and the red points with green number represent the maximum extreme.	<b>S13</b>
<b>Table S5.</b> Electrostatic potential distribution diagram of compound <b>20</b> .	<b>S13</b>
<b>Table S6.</b> Docking process parameter settings for those with relative high binding energy.	<b>S14</b>
<b>Table S7.</b> Drug-likeness prediction of compounds <b>1-24</b>	<b>S15</b>

**Table S1.** Electrostatic potential distribution extreme of compound **1** (S-PK6).

Number of surface minima	Potential (kcal·mol <sup>-1</sup> )	Number of surface maxima	Potential (kcal·mol <sup>-1</sup> )	Number of surface maxima	Potential (kcal·mol <sup>-1</sup> )
1	-12.18	1	14.69	20	-13.21
2	-22.58	2	16.249	21	3.01
3	-56.01	3	45.71	22	11.69
4	-22.11	4	11.29	23	-22.29
5	19.04	5	13.05	24	5.68
6	-50.26	6	9.72	25	7.82
7	2.34	7	21.24	26	-24.19
8	-30.23	8	21.06	27	10.25
9	-37.963	9	-23.02	28	32.25
10	-39.26	10	12.14	* 29	52.04
11	5.66	11	22.92	30	21.70
12	-48.61	12	3.86	31	10.87
13	-5.30	13	20.32	32	29.70
*14	-64.19	14	4.75	33	2.35
15	-33.459	15	19.06	34	12.30
16	29.61	16	-0.95	35	15.48
17	24.57	17	11.18	36	21.96
18	8.05	18	-26.68	37	14.54
		19	-30.99		

Note: The values in bold with \* indicate the maximum or minimum extreme values



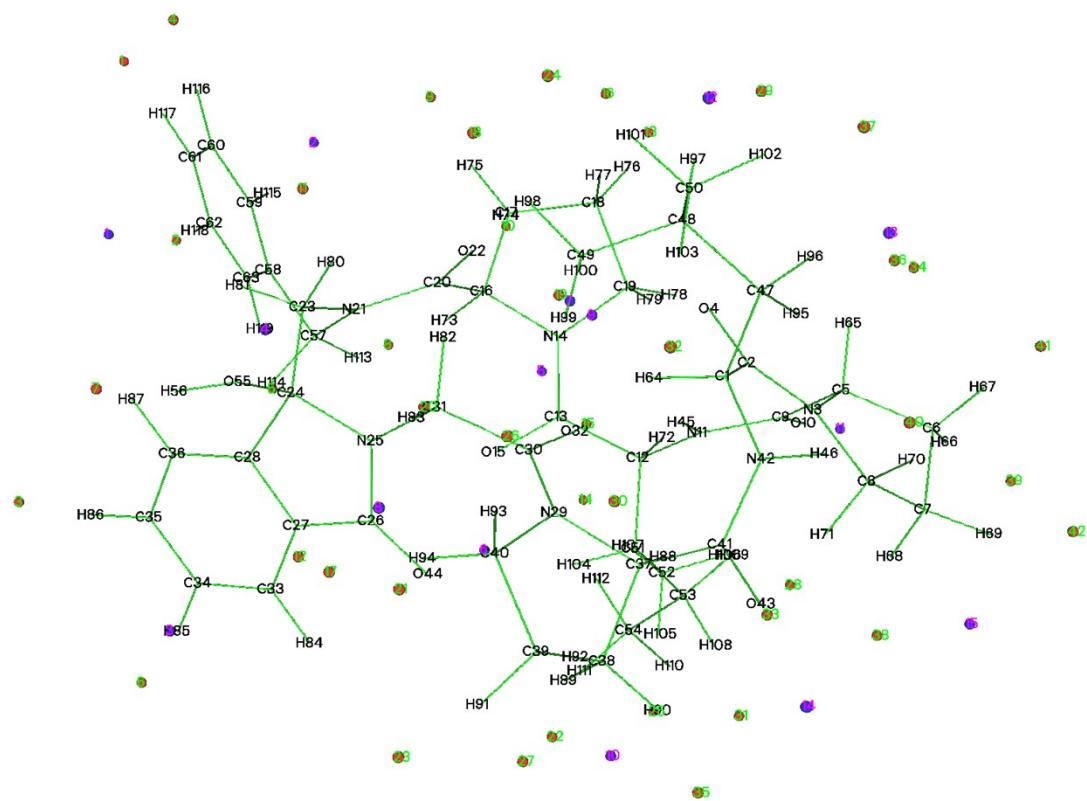
**Fig.S1.** Distribution of the surface electrostatic potential of compound **1** (S-PK6) with numbered atoms. The

blue points with pink numbers represent the minimum extreme, and the red points with green number represent the maximum extreme.

**Table S2.** Electrostatic potential distribution diagram of compound 2.

Number of surface minima	Potential (kcal·mol <sup>-1</sup> )	Number of surface maxima	Potential (kcal·mol <sup>-1</sup> )	Number of surface maxima	Potential (kcal·mol <sup>-1</sup> )
<b>1</b>	-7.78	<b>1</b>	15.39	<b>22</b>	-3.60
<b>2</b>	-10.43	<b>2</b>	15.84	<b>23</b>	7.64
<b>3</b>	-2.60	<b>3</b>	22.04	<b>24</b>	8.39
<b>4</b>	-21.45	<b>4</b>	15.58	<b>25</b>	0.03
<b>5</b>	-4.64	<b>5</b>	15.58	<b>26</b>	12.58
<b>6</b>	-49.93	<b>6</b>	18.75	<b>27</b>	-1.78
<b>7</b>	-59.07	* <b>7</b>	52.53	<b>28</b>	-9.49
<b>8</b>	6.77	<b>8</b>	17.44	<b>29</b>	4.44
* <b>9</b>	-59.80	<b>9</b>	6.36	<b>30</b>	10.09
<b>10</b>	-9.65	<b>10</b>	10.12	<b>31</b>	-2.44
<b>11</b>	-43.94	<b>11</b>	9.86	<b>32</b>	8.99
<b>12</b>	-1.49	<b>12</b>	7.03	<b>33</b>	10.31
<b>13</b>	5.41	<b>13</b>	6.79	<b>34</b>	8.81
<b>14</b>	-38.76	<b>14</b>	-0.13	<b>35</b>	-0.61
<b>15</b>	0.47	<b>15</b>	2.24	<b>36</b>	10.77
		<b>16</b>	-3.34	<b>37</b>	9.53
		<b>17</b>	10.30	<b>38</b>	4.16
		<b>18</b>	7.29	<b>39</b>	2.18
		<b>19</b>	-36.19	<b>40</b>	33.60
		<b>20</b>	11.54	<b>41</b>	12.67
		<b>21</b>	8.46	<b>42</b>	11.88

Note: The value in bold with \* indicates the maximum or minimum value

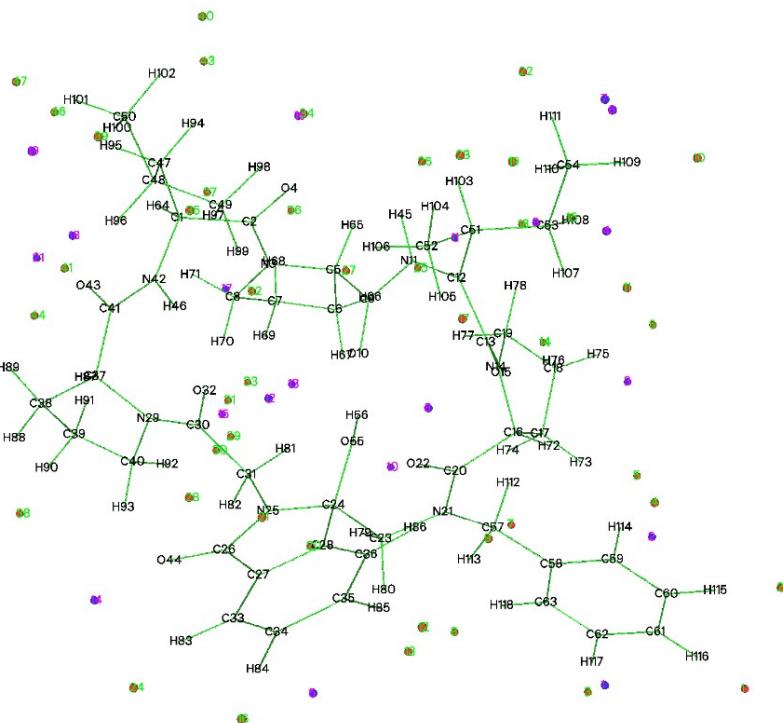


**Fig.S2.** Distribution of the surface electrostatic potential extreme of compound **2** with numbered atoms. The blue points with pink numbers represent the minimum extreme, and the red points with green number represent the maximum extreme.

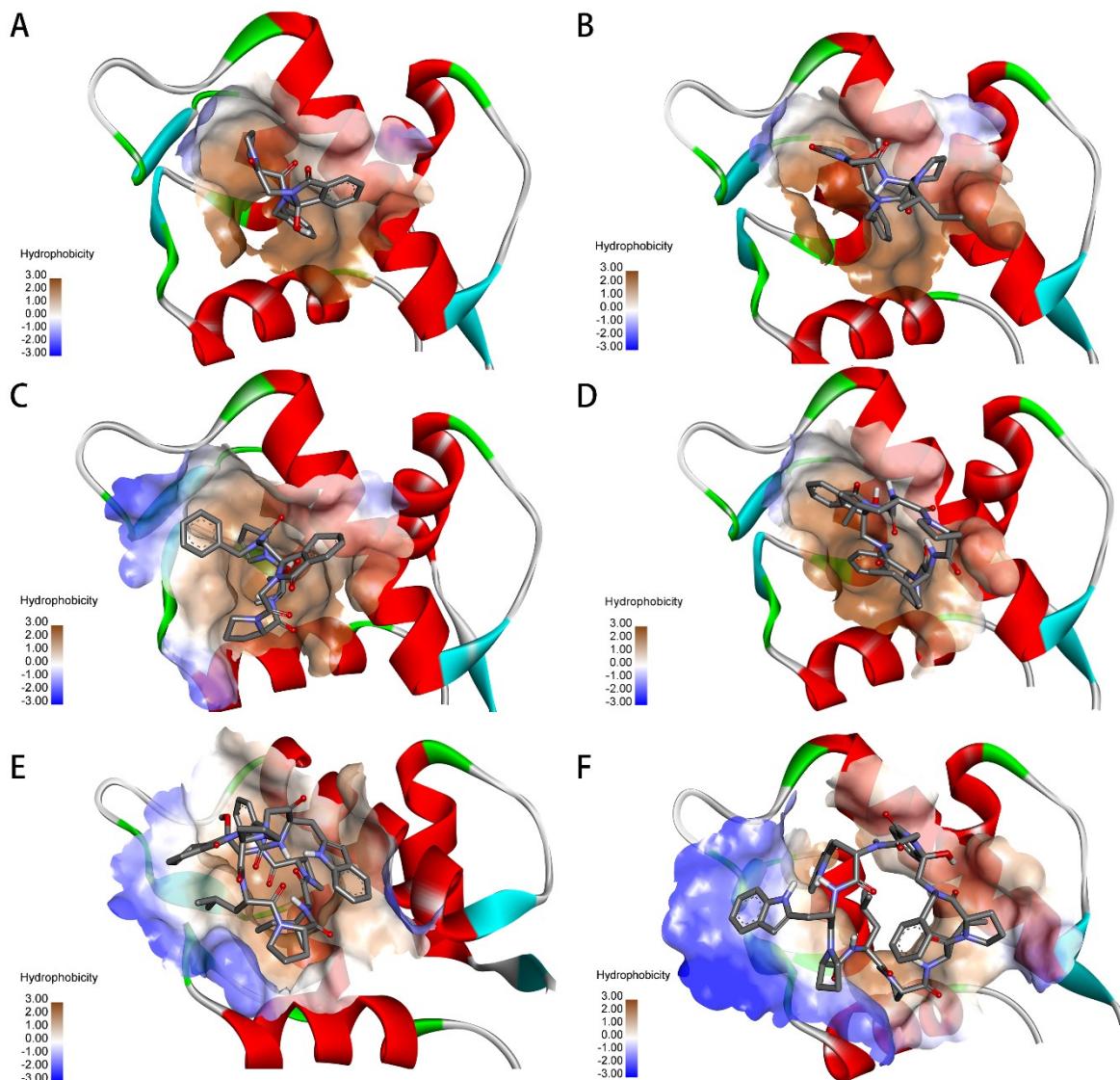
**Table S3.** Electrostatic potential distribution diagram of compound **3**.

Number of surface minima	Potential (kcal·mol <sup>-1</sup> )	Number of surface maxima	Potential (kcal·mol <sup>-1</sup> )	Number of surface maxima	Potential (kcal·mol <sup>-1</sup> )
1	-14.09	1	13.77	<b>25</b>	19.20
2	-11.27	2	13.60	* <b>26</b>	34.89
3	6.52	3	13.50	<b>27</b>	-6.88
4	-2.54	4	10.62	<b>28</b>	-15.33
5	-12.09	5	9.55	<b>29</b>	16.60
6	-1.77	6	12.49	<b>30</b>	17.44
7	-1.77	7	4.24	<b>31</b>	15.50
8	-47.38	8	10.72	<b>32</b>	2.75
9	9.84	9	10.99	<b>33</b>	16.28
10	-32.09	10	7.12	<b>34</b>	-2.29
11	10.99	11	-0.58	<b>35</b>	-1.11
*12	-55.87	12	12.67	<b>36</b>	15.67
13	-3.97	13	10.83	<b>37</b>	14.64
14	-40.75	14	14.83	<b>38</b>	11.83
15	14.89	15	10.72	<b>39</b>	2.22
16	-30.03	16	12.09	<b>40</b>	2.26
17	8.12	17	-6.35	<b>41</b>	0.38
18	-7.07	18	18.65	<b>42</b>	7.00
19	-5.28	19	5.54	<b>43</b>	0.81
20	0.85	20	7.02	<b>44</b>	8.98
21	-36.27	21	-18.06	<b>45</b>	4.13
		22	8.21	<b>46</b>	2.82
		23	4.69	<b>47</b>	3.30
		24	7.39		

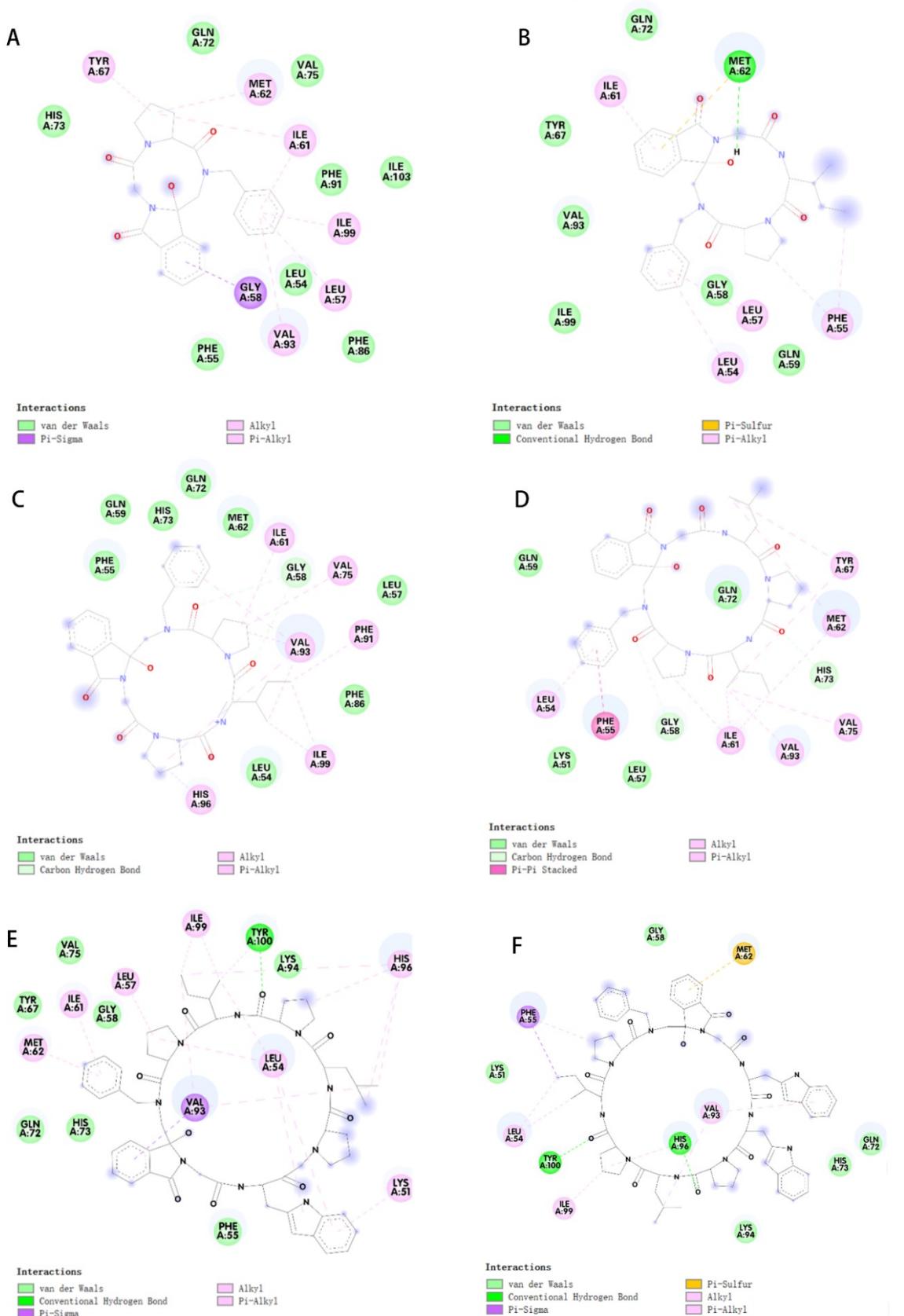
Note: The value in bold with \* indicates the maximum or minimum value



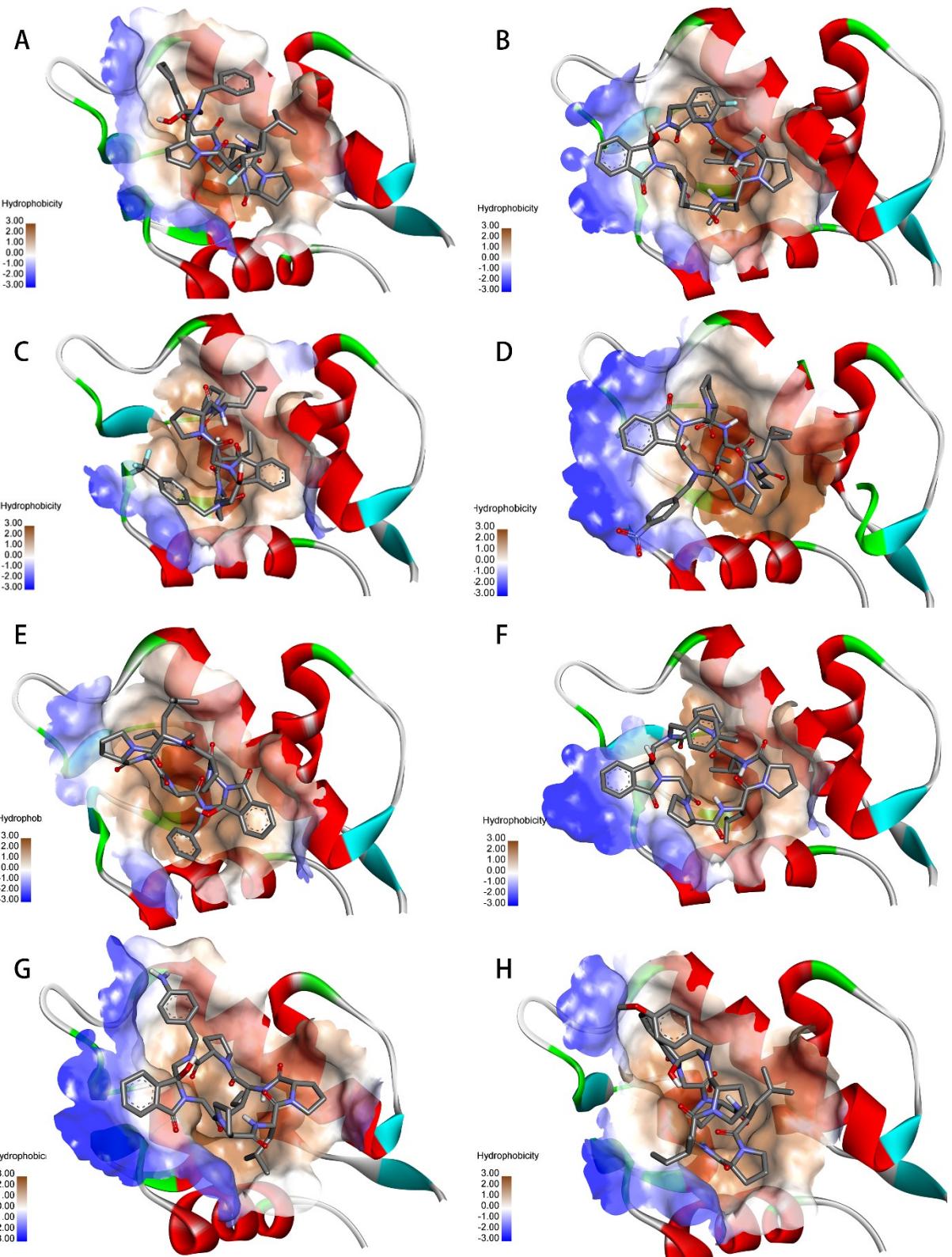
**Fig.S3.** Distribution of the surface electrostatic potential extreme of compound **3** with numbered atoms. The blue points with pink numbers represent the minimum extreme, and the red points with green numbers represent the maximum extreme.



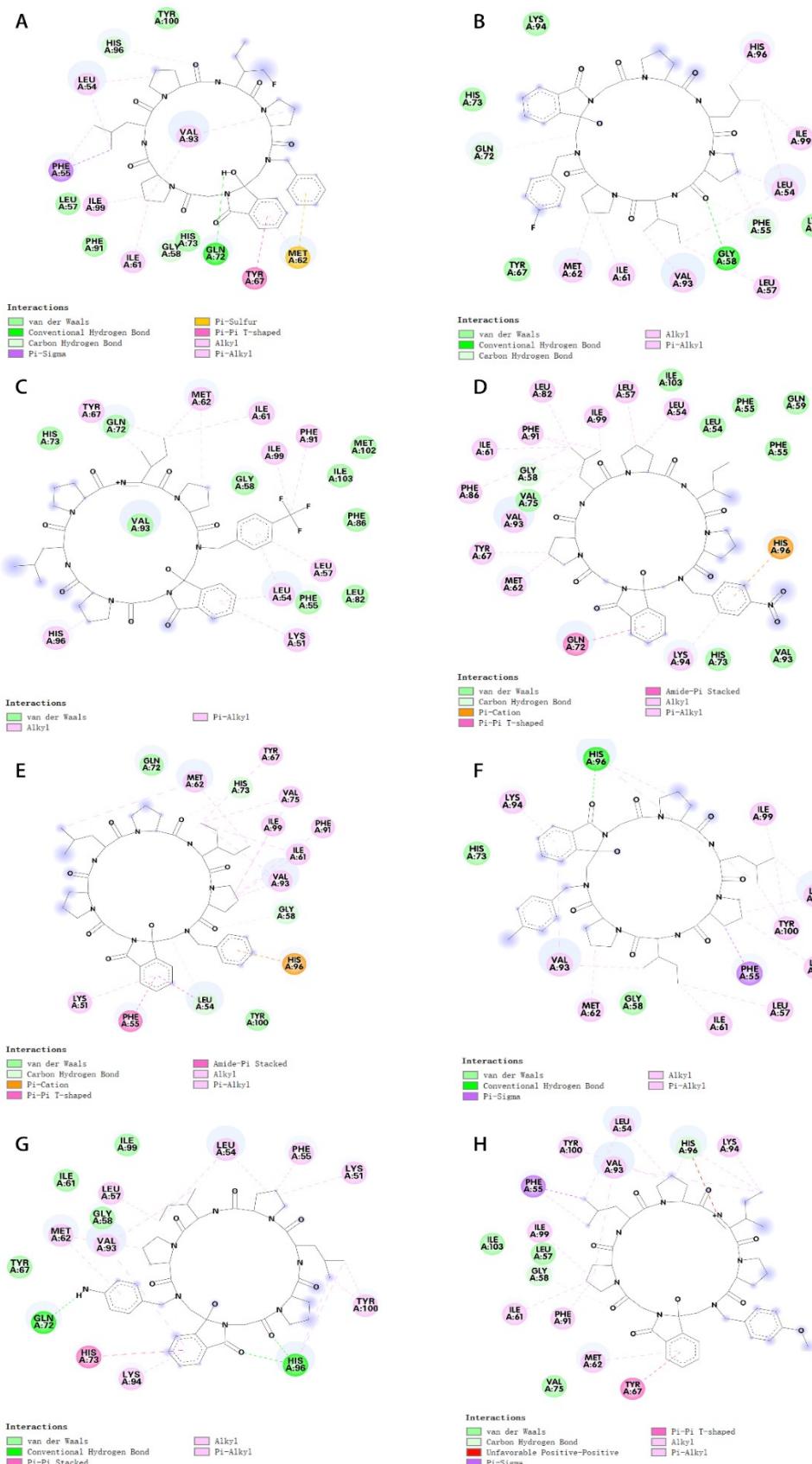
**Fig.S4.** The three-dimensional (3D) molecular docking between compounds **4(A)**, **5(B)**, **6(C)**, **7(D)**, **8(E)**, and **9(F)** with MDM2.



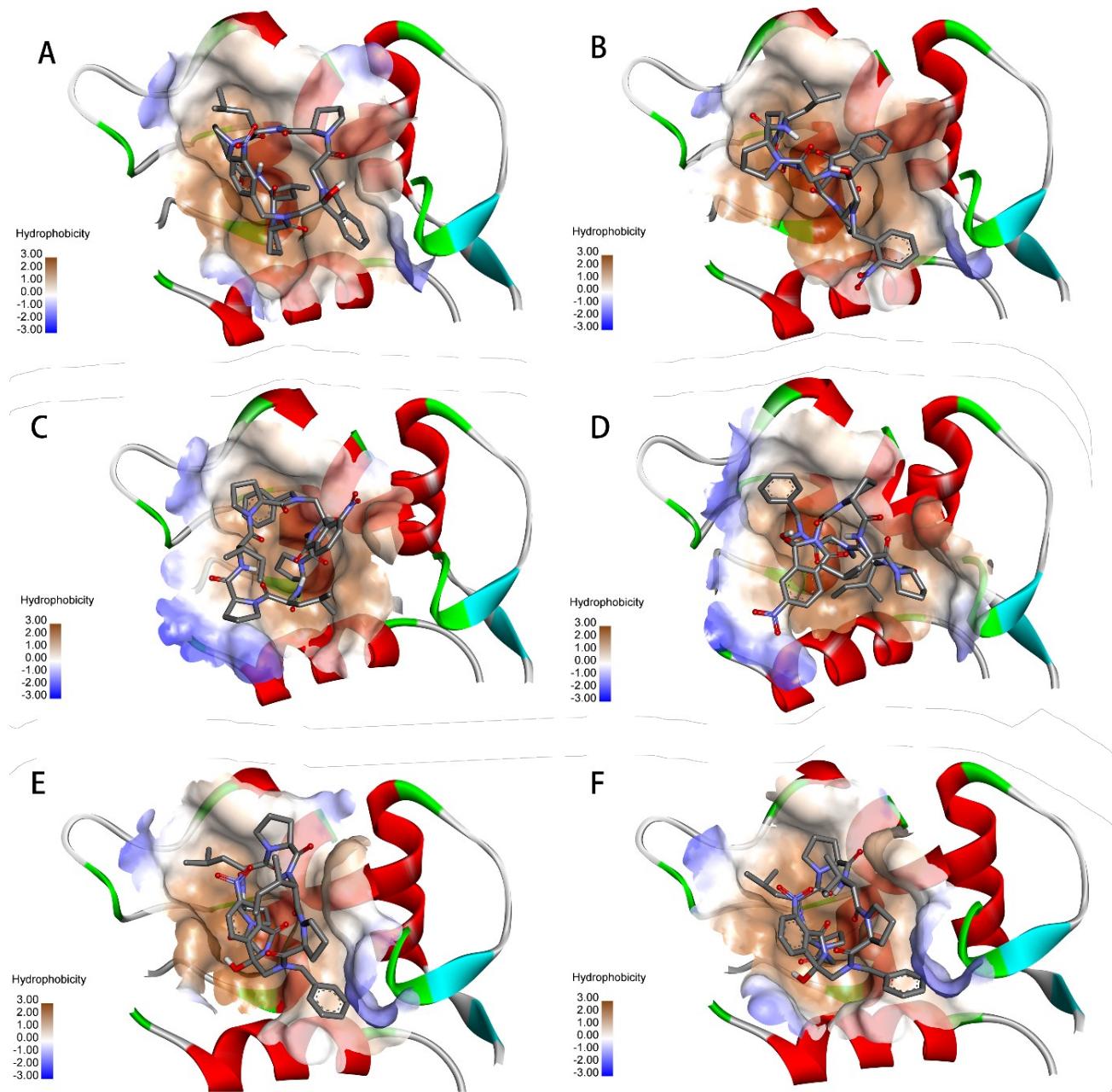
**Fig.S5.** The 2D molecular docking sites between compounds **4(A)**, **5(B)**, **6(C)**, **7(D)**, **8(E)**, and **9(F)** with MDM2.



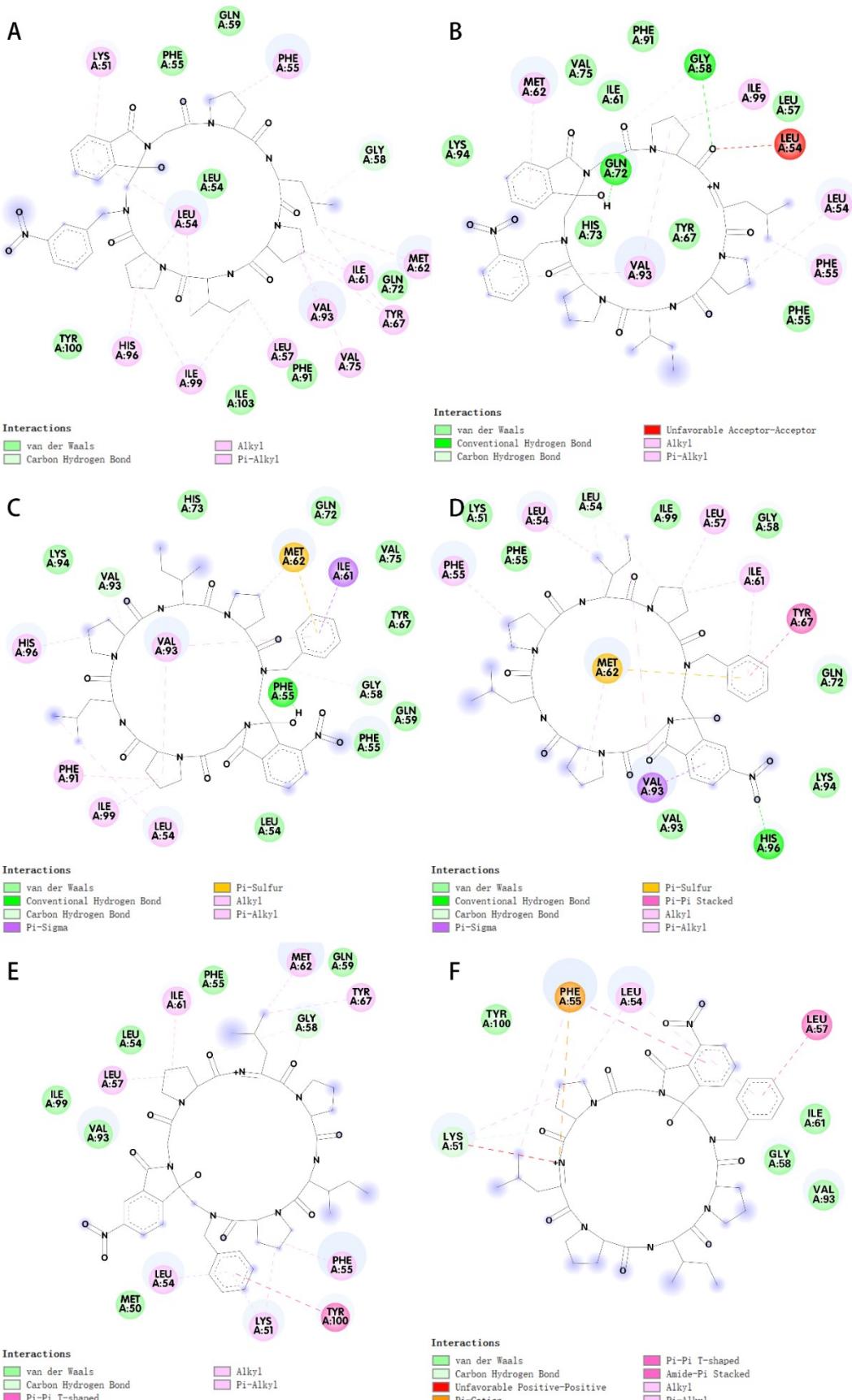
**Fig.S6.** The three-dimensional (3D) molecular docking between compounds **11(A)**, **12(B)**, **13(C)**, **14(D)**, **15(E)**, **16(F)**, **17(G)**, and **18(H)** with MDM2.



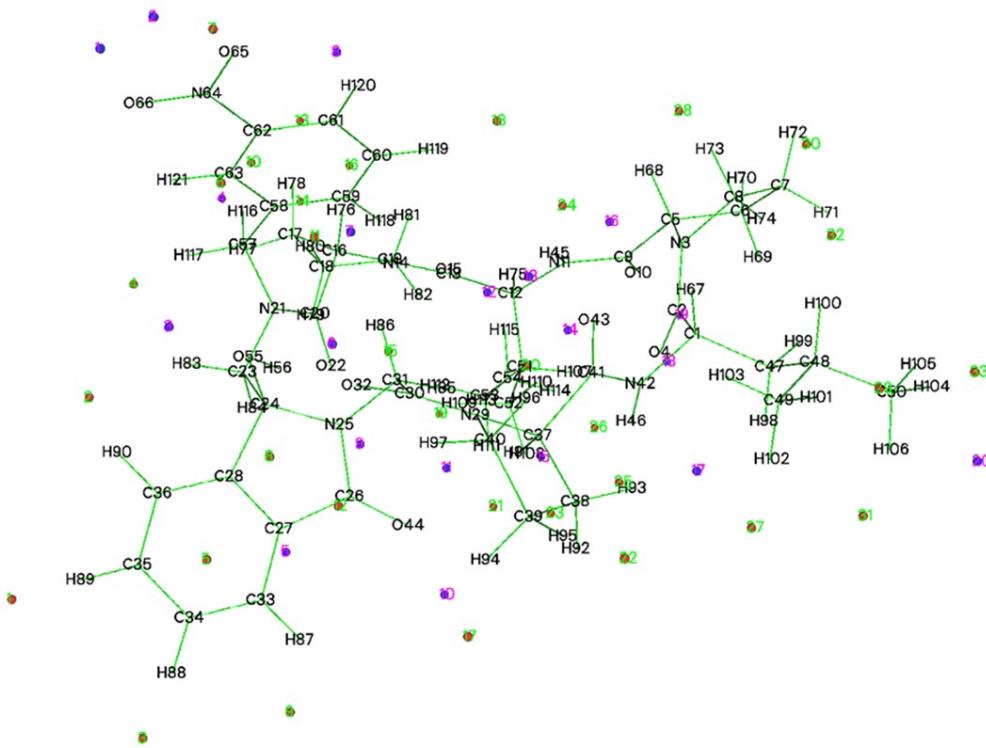
**Fig.S7.** The two-dimensional (2D) molecular docking sites between compounds **11(A)**, **12(B)**, **13(C)**, **14(D)**, **15(E)**, **16(F)**, **17(G)**, and **18(H)** with MDM2.



**Fig.S8.** The three-dimensional (3D) molecular docking between compounds **19(A)**, **20(B)**, **21(C)**, **22(D)**, **23(E)**, and **24(F)** with MDM2.



**Fig.S9.** The two-dimensional (2D) molecular docking sites between compounds **19(A)**, **20(B)**, **21(C)**, **22(D)**, **23(E)**, and **24(F)** with MDM2.

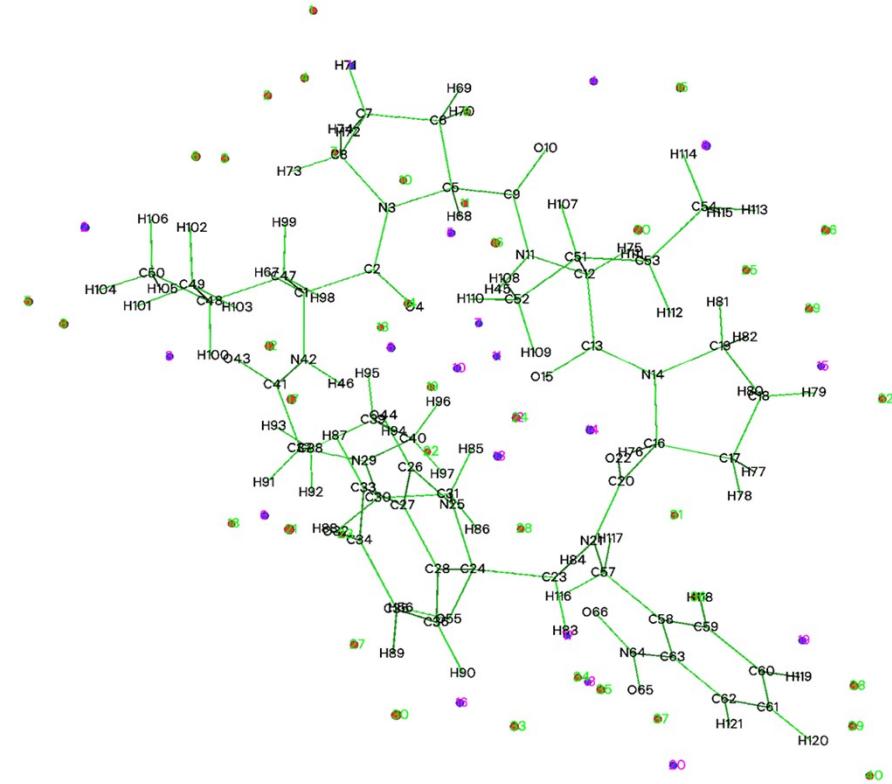


**Fig.S10.** Distribution of the surface electrostatic potential extreme of compound **19** with numbered atoms. The blue points with pink numbers represent the minimum extreme, and the red points with green number represent the maximum extreme.

**Table S4.** Electrostatic potential distribution diagram of compound **19**.

Number of surface minima	Potential (kcal·mol <sup>-1</sup> )	Number of surface maxima	Potential (kcal·mol <sup>-1</sup> )	Number of surface maxima	Potential (kcal·mol <sup>-1</sup> )
<b>1</b>	-42.84	<b>1</b>	18.91	<b>21</b>	-1.43
<b>2</b>	-43.21	<b>2</b>	18.95	<b>22</b>	13.67
<b>3</b>	-33.61	<b>3</b>	17.33	<b>23</b>	-4.58
<b>4</b>	22.23	<b>4</b>	27.27	<b>24</b>	23.31
<b>5</b>	-13.75	<b>5</b>	-14.77	<b>25</b>	34.21
<b>6</b>	-43.51	<b>6</b>	-2.36	<b>26</b>	-3.46
<b>7</b>	-19.42	<b>7</b>	7.24	<b>27</b>	13.99
<b>8</b>	-2.09	<b>8</b>	-11.87	<b>28</b>	26.93
<b>9</b>	-43.33	<b>9</b>	9.97	<b>29</b>	10.76
<b>10</b>	-42.19	<b>*10</b>	35.31	<b>30</b>	24.88
<b>11</b>	-25.42	<b>11</b>	-17.15	<b>31</b>	12.57
<b>12</b>	-24.66	<b>12</b>	-9.26	<b>32</b>	28.06
<b>13</b>	-43.37	<b>13</b>	29.11	<b>33</b>	12.52
<b>14</b>	-3.13	<b>14</b>	25.32		
<b>15</b>	-6.17	<b>15</b>	-20.27		
<b>16</b>	-3.32	<b>16</b>	28.24		
<b>17</b>	6.28	<b>17</b>	4.34		
<b>18</b>	-47.05	<b>18</b>	16.13		
<b>*19</b>	-49.73	<b>19</b>	7.13		
<b>20</b>	3.68	<b>20</b>	9.25		

Note: The value in bold with \* indicates the maximum or minimum value



**Fig.S11.** Distribution of the surface electrostatic potential extreme of compound **20** with numbered atoms. The blue points with pink numbers represent the minimum extreme, and the red points with green number represent the maximum extreme.

**Table S5.** Electrostatic potential distribution diagram of compound **20**.

Number of surface minima	Potential (kcal·mol <sup>-1</sup> )	Number of surface maxima	Potential (kcal·mol <sup>-1</sup> )	Number of surface maxima	Potential (kcal·mol <sup>-1</sup> )
<b>1</b>	6.81	<b>1</b>	14.70	<b>21</b>	16.10
<b>2</b>	-6.76	<b>2</b>	15.68	<b>22</b>	11.14
<b>3</b>	-51.19	<b>3</b>	16.47	<b>23</b>	6.35
<b>4</b>	-47.28	<b>4</b>	13.90	<b>24</b>	-39.66
<b>5</b>	-16.60	<b>5</b>	-0.98	<b>25</b>	23.07
<b>6</b>	-40.20	<b>6</b>	4.84	<b>26</b>	9.081
<b>7</b>	-28.01	<b>7</b>	12.13	<b>27</b>	8.97
<b>8</b>	-4.43	<b>8</b>	0.71	<b>28</b>	15.75
* <b>9</b>	-49.05	<b>9</b>	11.07	<b>29</b>	22.99
<b>10</b>	-57.72	<b>10</b>	10.66	<b>30</b>	19.52
<b>11</b>	-32.13	<b>11</b>	10.10	<b>31</b>	31.04
<b>12</b>	-42.55	<b>12</b>	0.57	<b>32</b>	26.21
<b>13</b>	-22.01	<b>13</b>	0.49	<b>33</b>	20.86
<b>14</b>	-41.13	<b>14</b>	-19.63	<b>34</b>	20.24
<b>15</b>	18.16	<b>15</b>	-1.48	<b>35</b>	21.77
<b>16</b>	-25.06	<b>16</b>	-2.60	<b>36</b>	22.93
<b>17</b>	-30.34	<b>17</b>	7.60	<b>37</b>	11.51
<b>18</b>	-32.10	<b>18</b>	-2.24	* <b>38</b>	34.18
<b>19</b>	12.14	<b>19</b>	-38.87	<b>39</b>	32.47
<b>20</b>	6.99	<b>20</b>	0.66	<b>40</b>	32.37

Note: The value in bold with \* indicates the maximum or minimum value

**Table S6.** Docking process parameter settings for those with relative high binding energy.

Parameter	Compound <b>1</b>	Compound <b>2</b>	Compound <b>3</b>	Compound <b>8</b>	Compound <b>14</b>	Compound <b>19</b>
Grid box (nm)	40×40×40	40×40×40	40×40×40	40×40×40	40×40×40	40×40×40
spacing	0.375	0.375	0.375	0.375	0.375	0.375
grid center x	-7.442	-7.442	-7.442	-7.442	-7.442	-7.442
grid center y	0.621	0.621	0.621	0.621	0.621	0.621
grid center z	6.892	6.892	6.892	6.892	6.892	6.892
GA runs	20	20	20	20	20	20
population size	150	150	150	150	150	150

**Table S7.** Drug-likeness prediction of compounds **1-24**

Compounds	Number of HBA	HBD	MolLogP	MolPSA (A <sup>2</sup> )	BBB Score	Drug-likeness model score
Compound <b>1</b>	8	3	1.86	150.95	1.00	0.36
Compound <b>2</b>	8	3	1.86	150.95	1.00	0.36
Compound <b>3</b>	8	3	1.86	150.95	1.00	0.36
Compound <b>4</b>	4	1	1.71	66.26	3.55	0.22
Compound <b>5</b>	5	2	1.95	91.64	2.11	0.47
Compound <b>6</b>	6	2	1.59	108.58	1.64	0.47
Compound <b>7</b>	7	3	2.31	134.02	1.00	0.36
Compound <b>8</b>	9	5	3.18	185.97	0.20	0.3
Compound <b>9</b>	10	7	4.39	221.21	0.00	0.3
Compound <b>10*</b>	-	-	-	-	-	-
Compound <b>11</b>	8	3	1.48	150.95	1.00	0.49
Compound <b>12</b>	8	3	1.92	150.95	1.00	0.88
Compound <b>13</b>	8	3	2.79	150.95	1.00	0.44
Compound <b>14</b>	10	3	1.78	184.34	1.00	0.18
Compound <b>15</b>	8	3	2.46	150.95	1.00	0.49
Compound <b>16</b>	8	5	2.30	150.95	1.00	0.47
Compound <b>17</b>	8	5	0.77	171.76	1.17	0.51
Compound <b>18</b>	9	3	1.81	158.50	1.00	0.54
Compound <b>19</b>	10	3	1.71	184.34	1.00	0.24
Compound <b>20</b>	10	3	2.01	184.03	1.00	-0.07
Compound <b>21</b>	10	3	1.16	184.03	1.00	-0.29
Compound <b>22</b>	10	3	1.97	184.03	1.00	-0.31
Compound <b>23</b>	10	3	1.91	184.34	1.00	-0.06
Compound <b>24</b>	10	3	1.27	184.03	1.00	0.12

\* The druglikeness of compound 10 could be not predicted by MolSoft programme due to it large molecular weight.