New Journal of Chemistry

Supporting information

Synthesis and photophysical properties of blue emission maleimide molecules with Dual-State Emission (DSE) effects

Xiaodong Yang, Kazuhiro Yamabuki and Kenjiro Onimura†

Department of Applied Chemistry, Graduate School of Sciences and Technology for Innovation, Yamaguchi University

2-16-1 Tokiwadai, Ube, Yamaguchi 755-8611, Japan

†To whom correspondence should be addressed (TEL: +81-836-85-9283, FAX: +81-836-85-9201, E-mail: onimura@yamaguchi-u.ac.jp)

1. ¹H and ¹³C NMR spectra of compound 1.

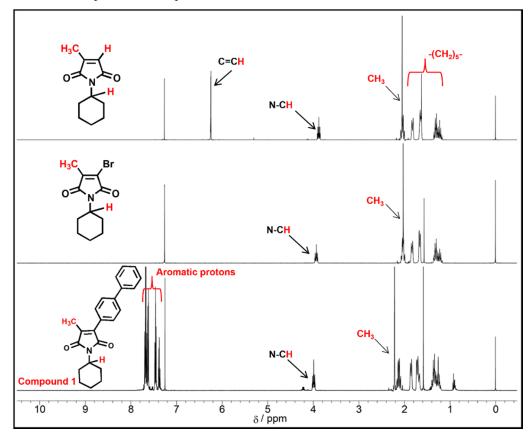


Figure S1. ¹H NMR Spectrum of compound 1 in CDCl₃.

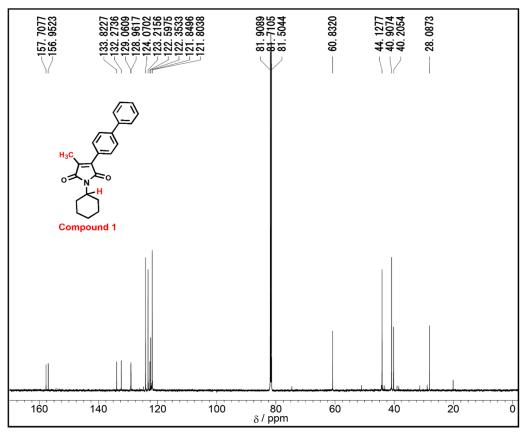


Figure S2. ¹³C NMR Spectrum of compound 1 in CDCl₃.

2. ¹H and ¹³C NMR spectra of compound 2.

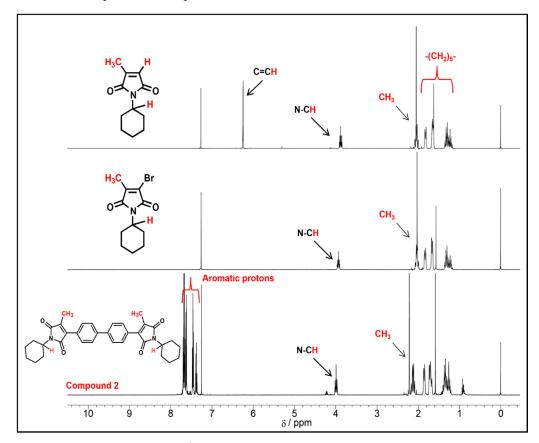


Figure S3. ¹H NMR Spectrum of compound 2 in CDCl₃.

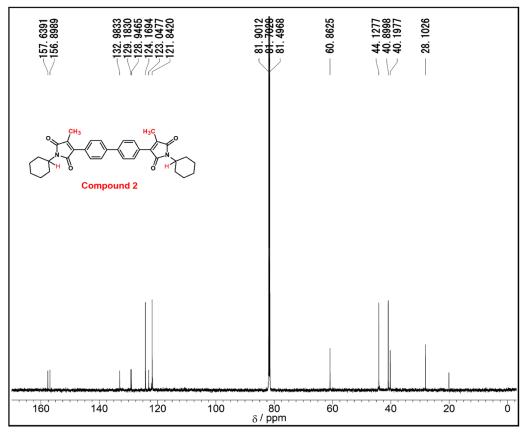


Figure S4. ¹³C NMR Spectrum of compound 2 in CDCl₃.

3. Optical properties of fluorophores in water-THF mixtures.

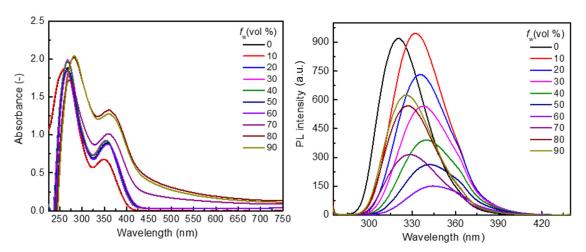


Figure S5. Absorption spectrum and emission spectrum of compound 1 in a mixture of water-THF.

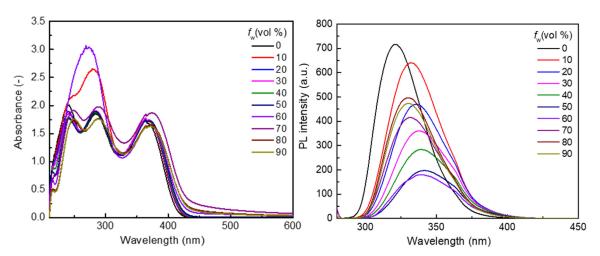


Figure S6. Absorption spectrum and emission spectrum of compound 2 in a mixture of water-THF.

4. Optical properties of fluorophores in different solvent.

Table S1

_										
	1					2				
Solvent	$\lambda_{abs}^{a)}$	λ em ^a	Δs	CIE (x,y)	$\mu_e - \mu_g^{\ b}$	$\lambda_{abs}^{a)}$	λ em a	Δs	CIE (x,y)	$\mu_e - \mu_g^{\ \mathrm{b}}$
	(nm)	(nm)	(cm ⁻¹)			(nm)	(nm)	(cm ⁻¹)		
Toluene	353	460	6590	0.14, 0.14	1.23 D	368	458	5340	0.14,0.13	2.09 D
THF	349	461	6961	0.14, 0.15	5.03 D	365	461	5705	0.15,0.16	8.62 D
Acetone	348	469	7414	0.14, 0.19	6.09 D	360.5	474	6642	0.15,0.23	10.9 D
Acetonitrile	349	472	7467	0.15, 0.24	6.36 D	359.5	480	6983	0.15,0.28	11.5 D
DMF	351	478	7569	0.15, 0.25	6.13 D	364.5	484	6774	0.16,0.30	11.1 D

a) 1.0 x 10⁻⁵ mol/L

b) μ_e and μ_g represents the Dipolar moments in the excited (e) and ground (g) states, respectively.

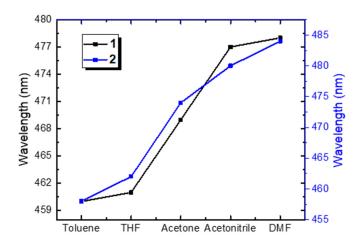


Figure S7. The change of the emission maximum wavelength of fluorophore **1** and **2** in solvent of different polarity (*conc*.: 1.0 x 10⁻⁵ mol/L).

5. ¹H NMR Spectra of fluorophores on addition of TFA in CDCl₃.

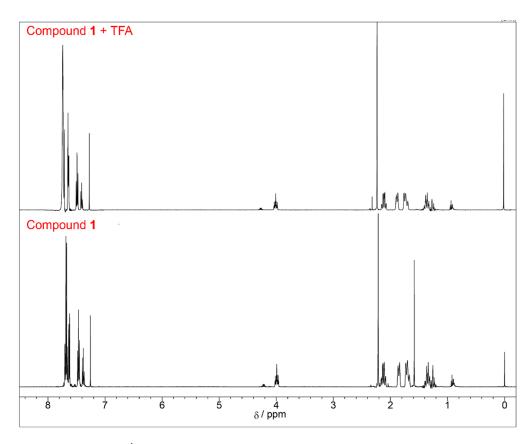


Figure S8. ¹H NMR Spectrum of compound 1 on addition of TFA in CDCl₃.

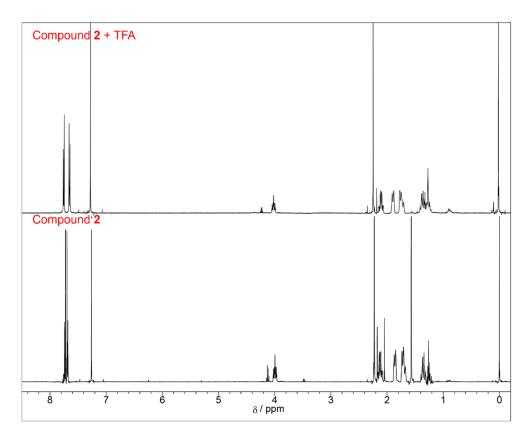


Figure S9. ¹H NMR Spectrum of compound 2 on addition of TFA in CDCl₃.