Supplementary Information

A latent reversible ratiometric optical pH sensing probe based phenylboronic acid for alkaline pH detection and applications in test paper and alkalotic HK-2 cells

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Scheme 1 The synthetic procedure of DDPB.



Scheme 2 The proposed reaction pathways of visible light mediated aerobic hydroxylation of DDPB.



Figure S1. ¹H NMR spectrum of DDPB.



Figure S2. ¹³C NMR spectrum of DDPB.



Figure S3. ESI-MS spectrum of DDPB.



Figure S4. UV-vis spectrum (black line) and fluorescence emission spectrum (red line) of DDPB (dissoved in 20% ethanol solution). The excitation wavelength was 430 nm.



Figure S5. Absorbance spectrum (black line) and fluorescence emission spectrum (red line) of DDPB (incubated in PBS-ethanol solution with pH 10.23). The excitation wavelength was 490 nm.



Figure S6. Fluorescence spectra responses of DDPB (30 μ M) incubate with 20% ethanol 20 mM phosphate buffer solution with different pH value (from 7.00 to 10.23). Fluorescent spectra were recorded at the excitation wavelength of 490 nm with 5/10 nm slit widths.



Figure S7. The pH titration curve of DDPB based ratiometric fluorescence emission intensity ($I_{667 nm}/I_{590 nm}$). The excitation wavelengths were 490 nm (A) and 430 nm (B) with 5/10 nm slit widths. (C) ExpDec1 curve fitting for absorbance intensity ratio ($A_{510 nm}/A_{420 nm}$) in response to pH value.



Figure S8 Ratiometric fluorescence responses of DDPB (30 μ M) towards 100 μ M Hg²⁺ and Cu²⁺ with the masking effect of 500 μ M EDTA and S₂O₃²⁻ in 20% ethanol 20 mM phosphate buffer solution with pH 9.30. Excitation wavelength was 430 nm with 5/10 nm slit widths. Error bars represent the standard deviations (n = 3).



Figure S9. ESI-MS spectrum of the product obtained from reaction mixture of



DDPB with 200 µM NaOH.

Figure S10. ¹H NMR spectrum of the product obtained from reaction mixture of

DDPB with 2 µM NaOH.



Figure S11. ¹H NMR spectrum of the product obtained from reaction mixture of DDPB with 200 μ M NaOH.



Figure S12 Photographs of DDPB paper-based sensor in response to phosphate buffer solutions with different pH.



Figure S13 Fluorescence signal ($I_{667 nm}/I_{590 nm}$) responses of DDPB (30 μ M) incubated in 20 mM phosphate buffer solution with different ethanol concentrations. The excitation wavelength was 430 nm with 5/10 nm slit widths.

Sample	Integrated emission	Absorbance	Refractive	Quantum
	intensity area (I)	(A)	index (η)	yield (Φ)
Rhodamine B	41689005	0.023	1.36	89%
DDPB	255467	0.038	1.36	0.33%
HDM	2795215	0.045	1.36	3.05%

Table S1 Fluorescent quantum yield calculation.

Probes	Method	Туре	pKa	pH range	Application	Ref.
IECBT	Fluorescent	Turn off	9.75	9.48-10.07	E. coli cells imaging	[1]
BTNO	Colorimetric and	Ratiometric	7.91	7.00-9.50	HeLa cells imaging	[2]
	fluorescent					
FQ-5	Fluorescent	Turn on	/	2-13	1	[3]
SypHer3s	Colorimetric and	Ratiometric	7.8	7.5-9.5	HEK293 cells imaging	[4]
	fluorescent					
TADF	Fluorescent	Turn off	/	7.0-8.6	Alkaliphiles detection	[5]
AlkaP-1	Fluorescent	Turn off	8.01	7.0-9.0	Ewing's sarcoma cells	[6]
					imaging	
BODIPY based probes	Colorimetric and	Ratiometric	7.33	9.4-9.9	1	[7]
	fluorescent					
CADB	Fluorescent	Turn on	10.62	9.65-11.68	Zebrafish imaging	[8]
hemicyanine-	Colorimetric and	Ratiometric	/	5-12	paper sensor	[9]
naphthalene-based	fluorescent					
fluorescent sensor						
PTZ-aminopyrazole	Colorimetric and	Ratiometric	/	11.1-13.8	paper sensor	[10]
	fluorescent					
PN-SP	fluorescent	Turn on	10.25	9.41–11.30	HeLa cells imaging	[11]
DDPB	Colorimetric and	Ratiometric	9.33	7.00-10.23	HK-2 cells imaging and	This work
	fluorescent				paper sensor	

Table S2 An overview on recently reported alkaline optical pH sensors.

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