

Supporting Information

Tetraalkylammonium based Dicationic Ionic Liquid (IL) for CO₂
Capture

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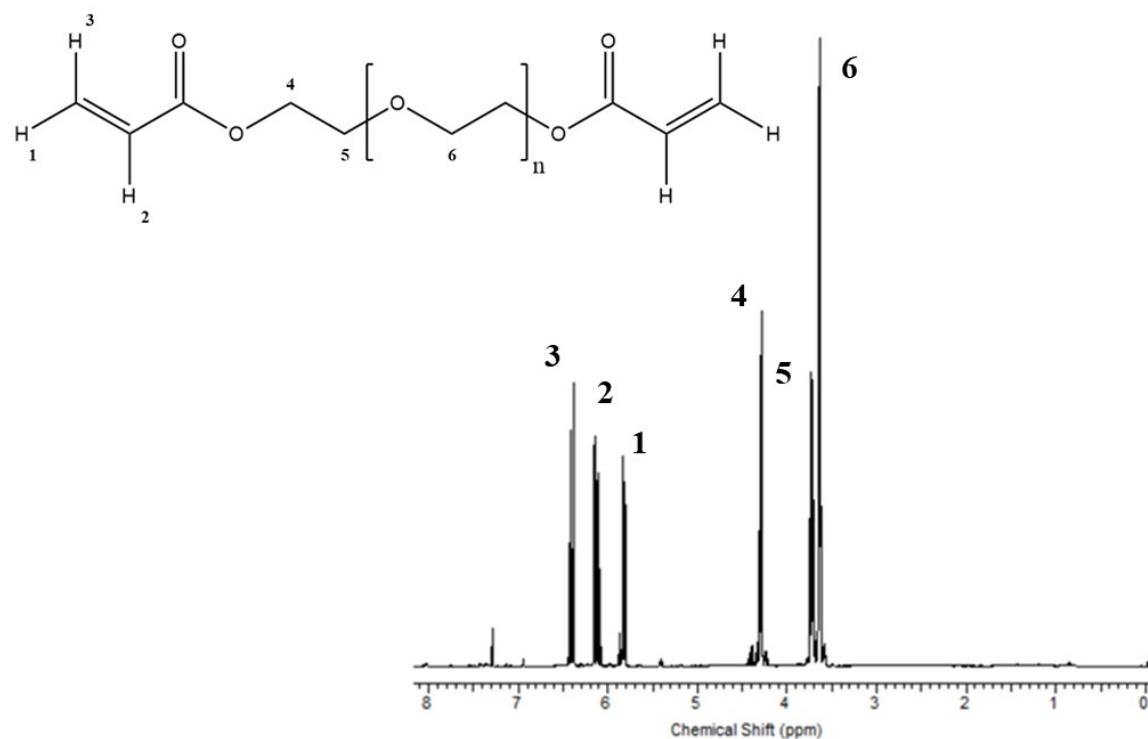


Figure S1: ¹H NMR spectra of PEGDA.

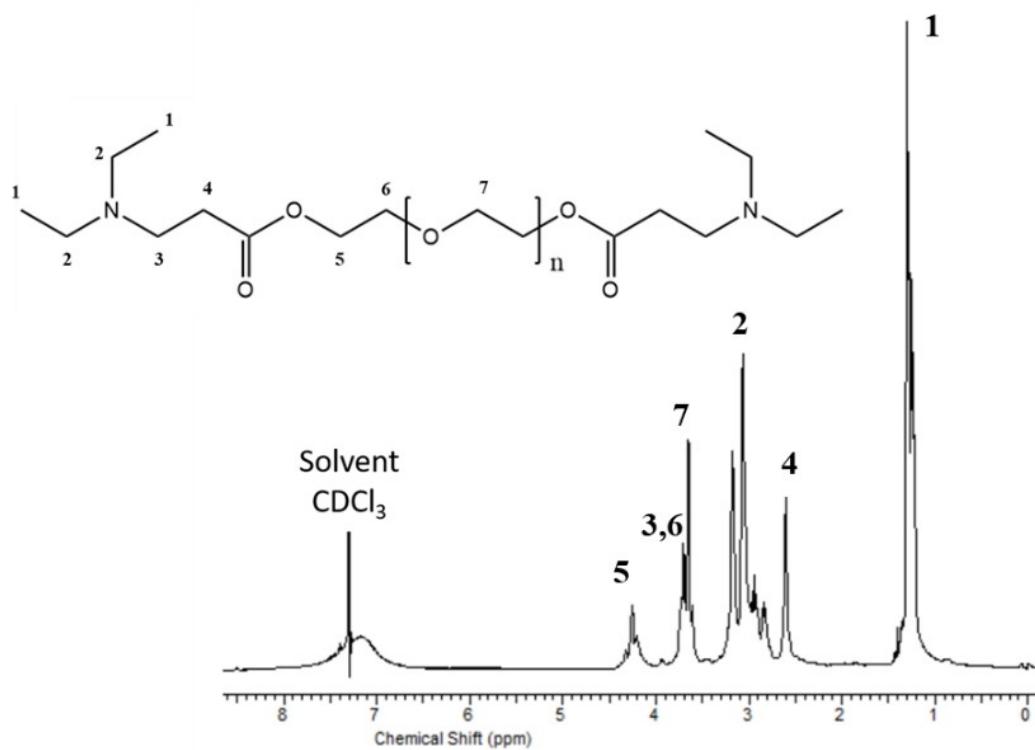


Figure S2: ¹H NMR spectra of PEGDA-DEA product (PD).

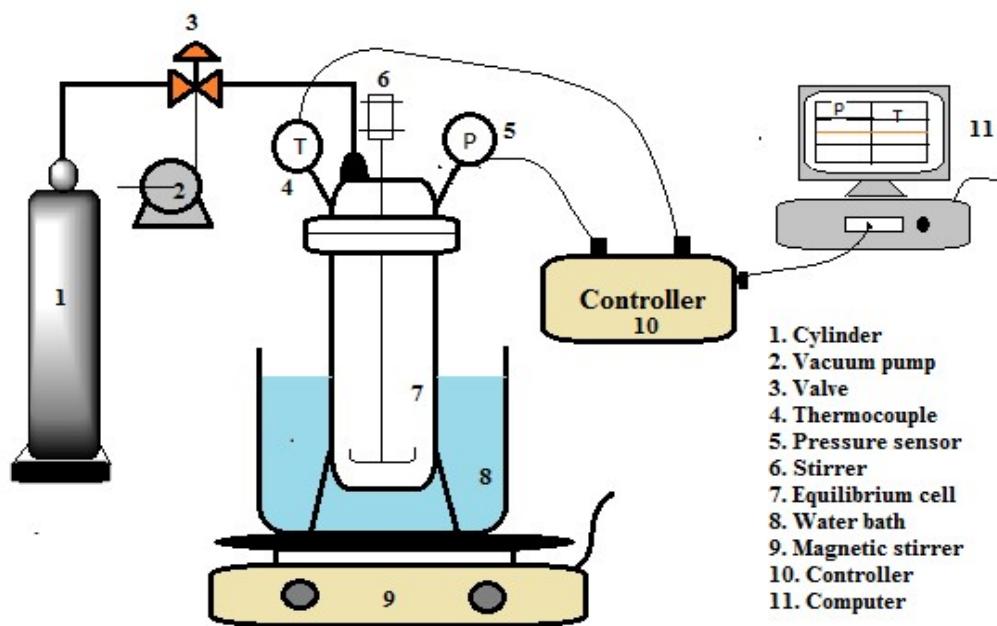


Figure S3: Schematic representation of fabricated CO₂ absorption apparatus.

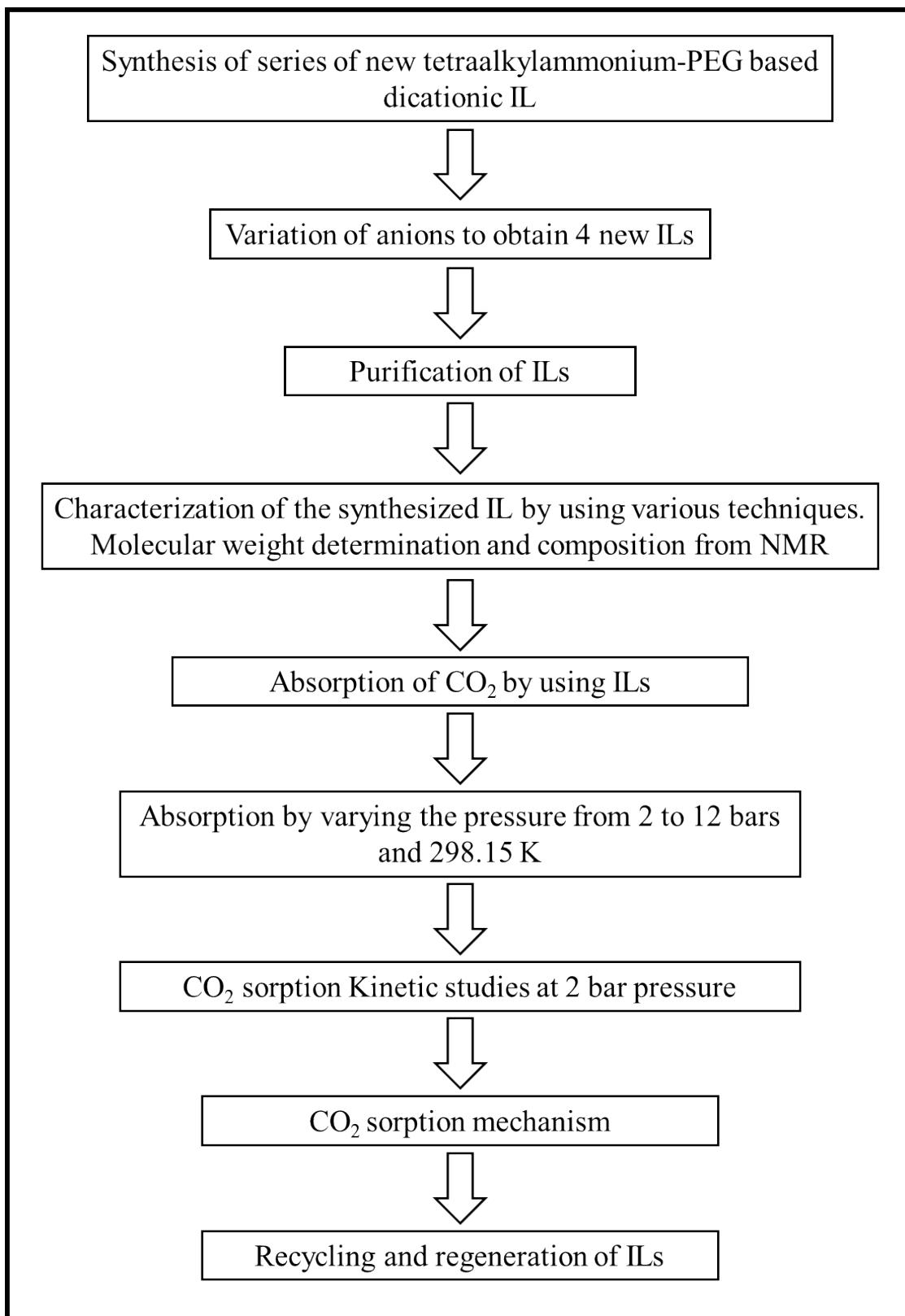


Figure S4: Flowchart - absorption of CO₂ by using tetraalkylammonium dicationic IL.

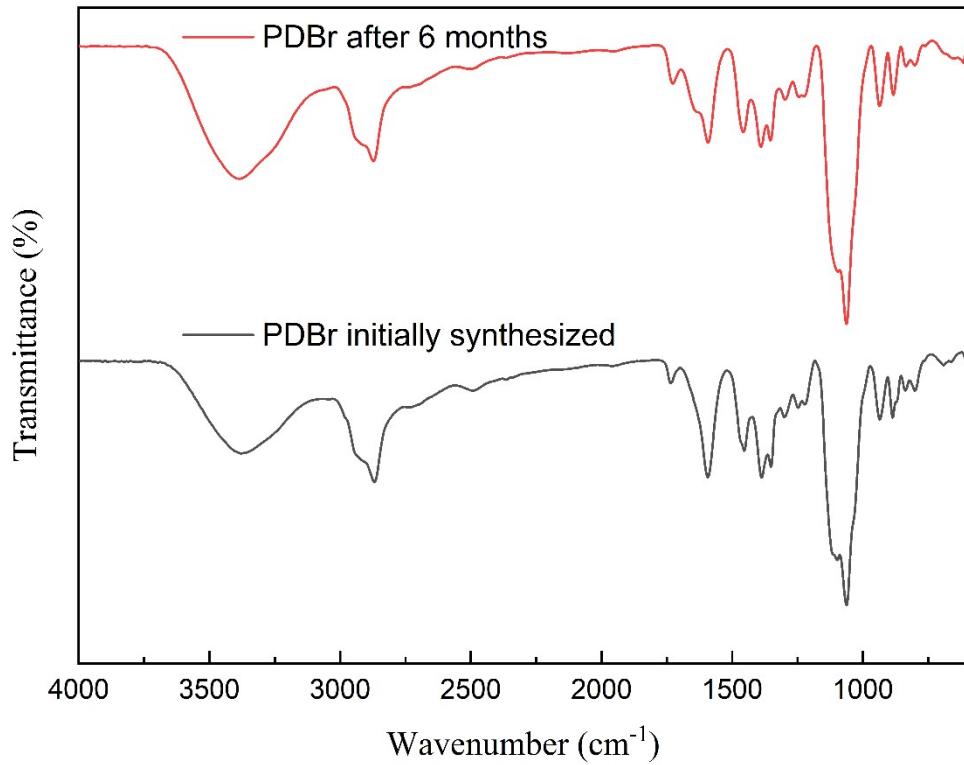


Figure S5: Chemical stability of PDBr IL after 6 months, as investigated by FTIR.

Table S1: Design of experiment with input parameters and operating conditions.

Ionic Liquid (IL)	Operating Conditions					Outcomes	
	Pressure bar	Temper- ature K	Molecular Weight (g.mol ⁻¹)	Viscosity (mm ² .s ⁻¹) (1 bar and 298.15 K)	Density (g.cm ⁻³) (1 bar and 298.15 K)	x_{CO_2}	mol CO ₂ / mol IL
PDBF ₄	2	298.15	764.48	198.40	1.09	0.128	0.147
	4					0.193	0.239
	6					0.222	0.287
	8					0.368	0.584
	10					0.559	1.273
	12					0.577	1.365
PDNTf ₂	2	298.15	1151.17	84.74	1.40	0.095	0.105
	4					0.211	0.268
	6					0.334	0.502
	8					0.385	0.626
	10					0.443	0.797
	12					0.471	0.892
PDPF ₆	2	298.15	880.80	44.66	1.29	0.167	0.201
	4					0.181	0.222
	6					0.227	0.294
	8					0.283	0.396
	10					0.421	0.727
	12					0.459	0.849
PDBr	2	298.15	750.67	80.33	1.23	0.039	0.041
	4					0.160	0.191
	6					0.232	0.303
	8					0.243	0.321
	10					0.334	0.503
	12					0.362	0.568