

Photodegradation of Methylene Blue Using $\text{Ce}_x\text{Zr}_y\text{O}_2$ Nanocomposites

Prepared via Non-stoichiometry Method

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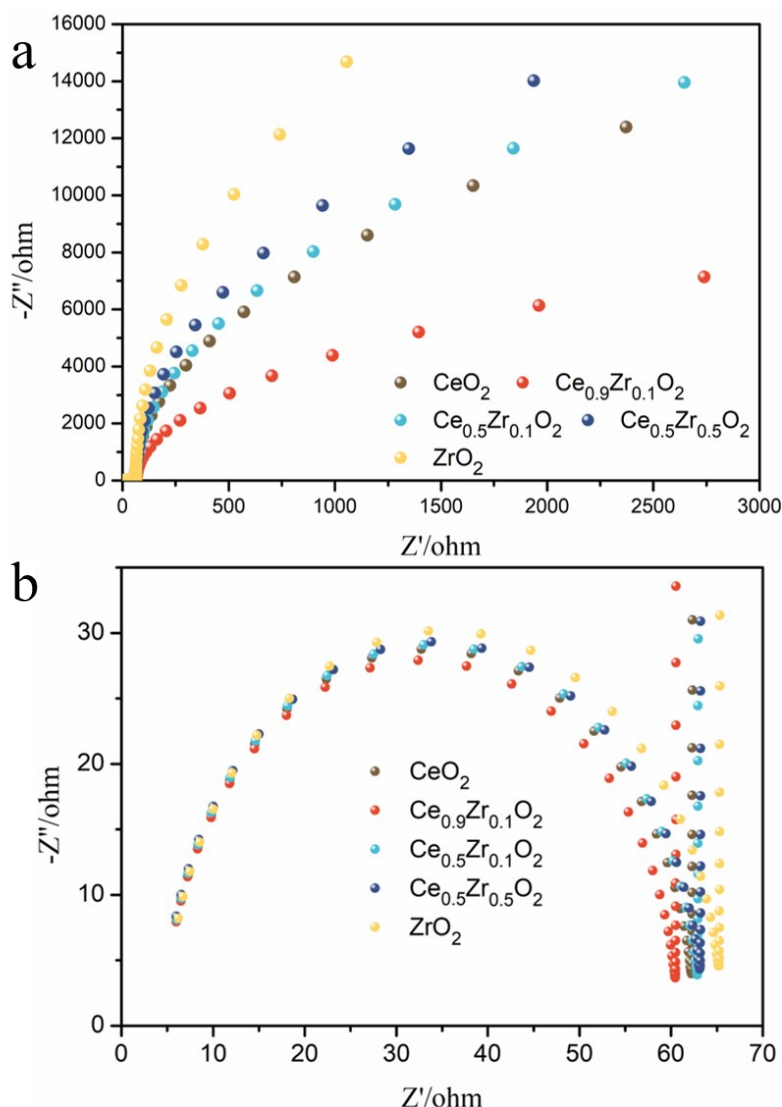


Fig. S1 (a) and (b) Nyquist plots of the various $\text{Ce}_x\text{Zr}_y\text{O}_2$ nanocomposites, CeO_2 and ZrO_2 .

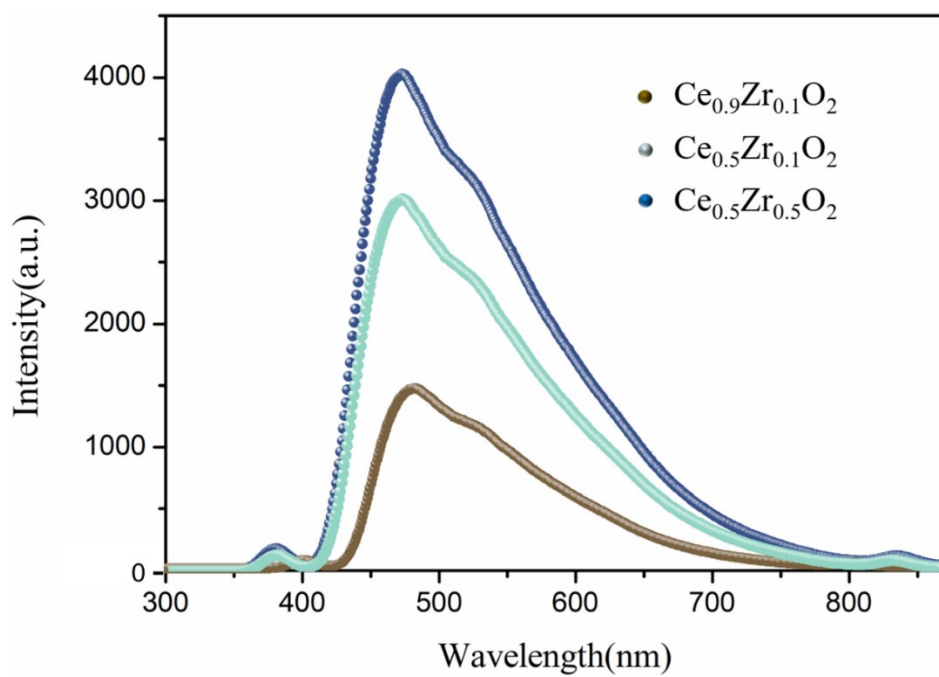


Fig. S2 The PL spectral data graph of $\text{Ce}_x\text{Zr}_y\text{O}_2$.

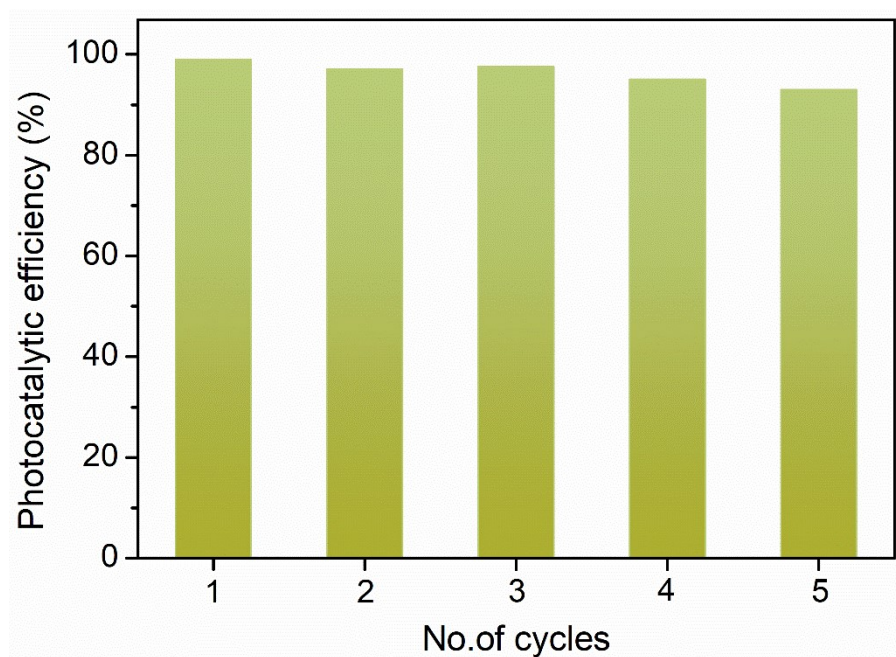


Fig. S3 Repeated photocatalytic tests of $\text{Ce}_{0.9}\text{Zr}_{0.1}\text{O}_2$.

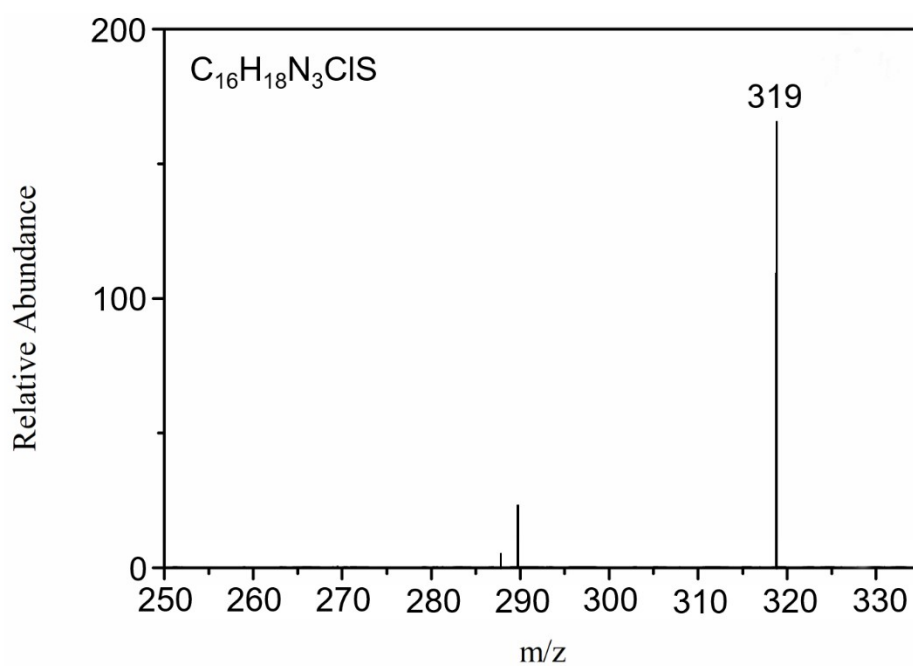


Fig. S4 Mass spectrum of MB during photocatalytic degradation reaction.

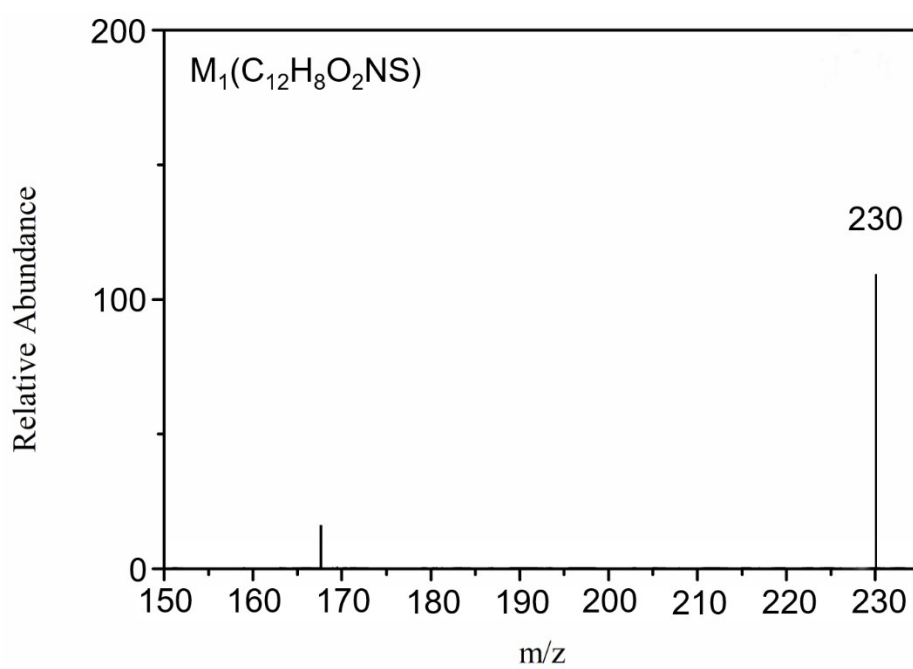


Fig. S5 Mass spectrum of reaction products M_1 during photocatalytic degradation of MB.

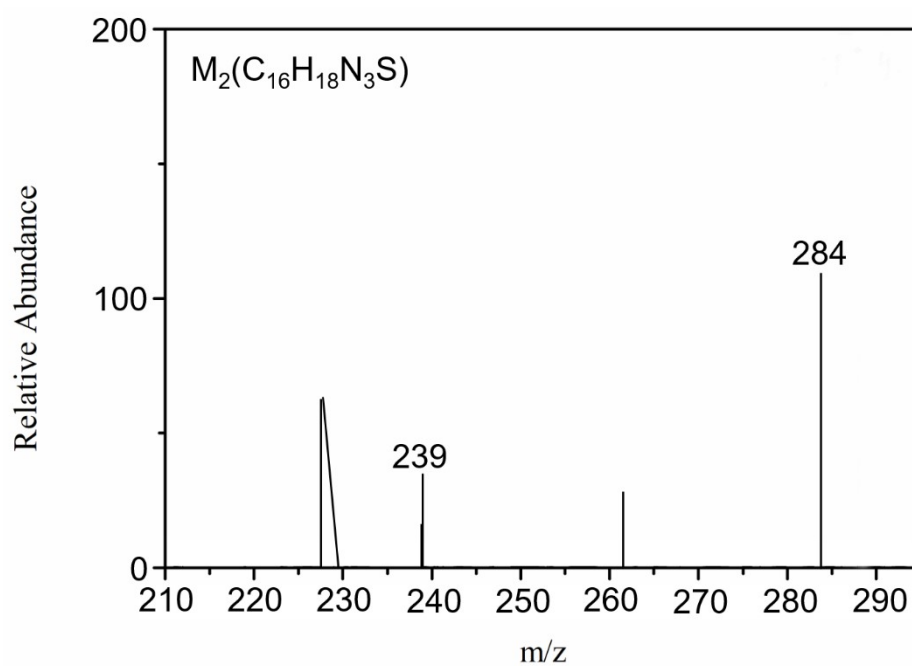


Fig. S6 Mass spectrum of reaction products M_2 during photocatalytic degradation of MB.

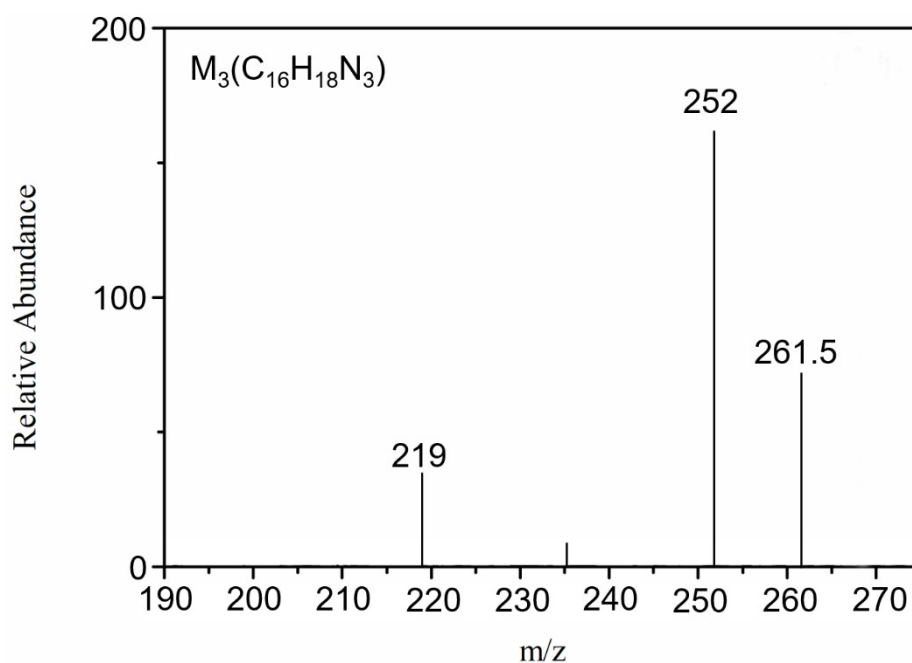


Fig. S7 Mass spectrum of reaction products M_3 during photocatalytic degradation of MB.

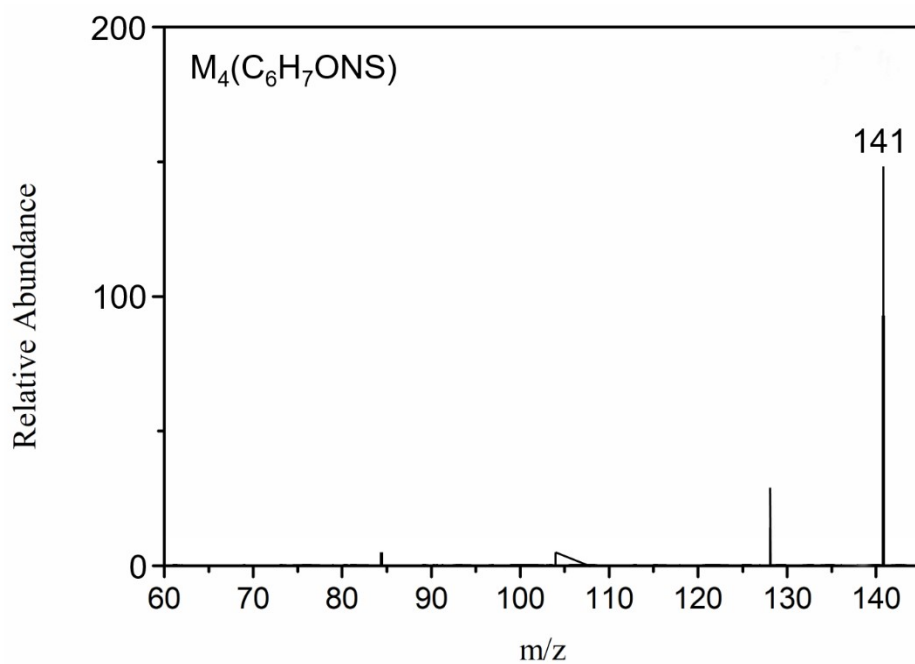


Fig. S8 Mass spectrum of reaction products M_4 during photocatalytic degradation of MB.

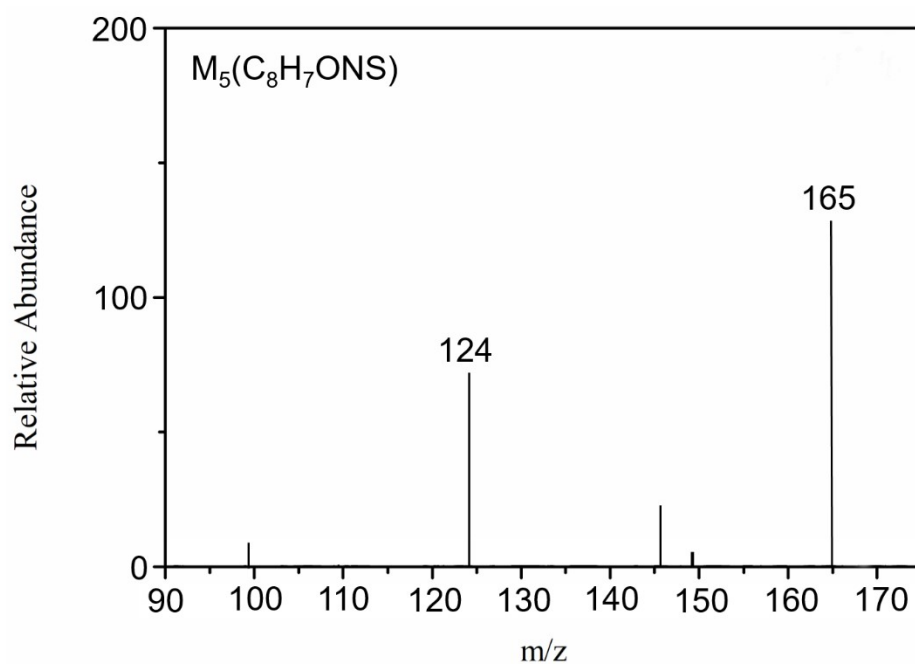


Fig. S9 Mass spectrum of reaction products M_5 during photocatalytic degradation of MB.

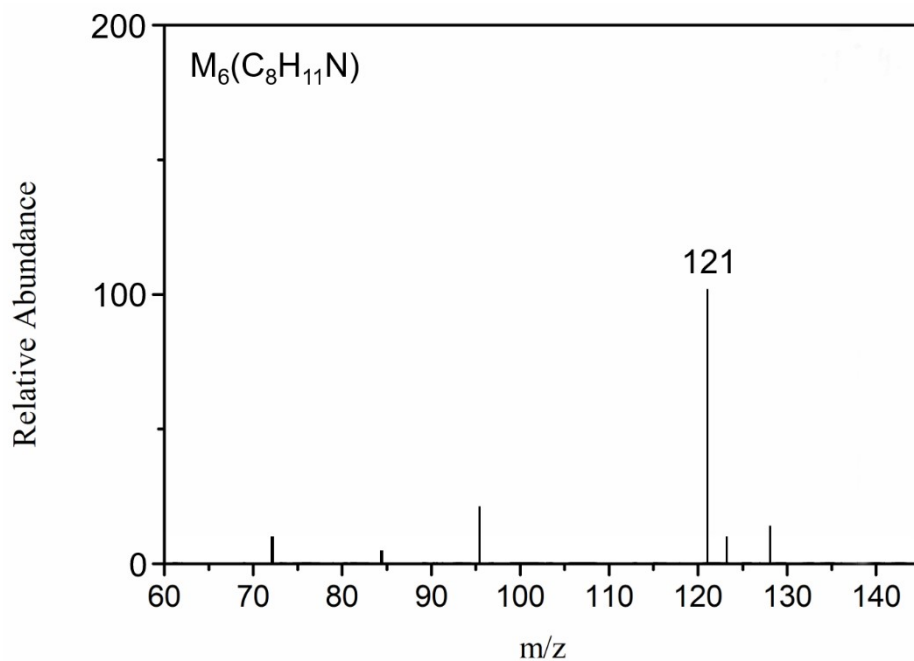


Fig. S10 Mass spectrum of reaction products M_6 during photocatalytic degradation of MB.

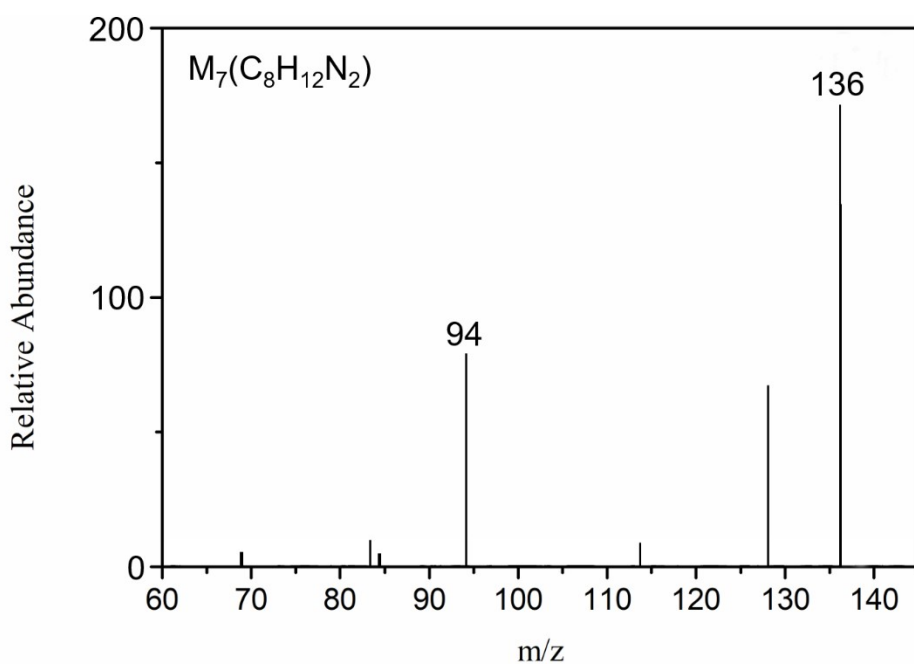


Fig. S11 Mass spectrum of reaction products M_7 during photocatalytic degradation of MB.

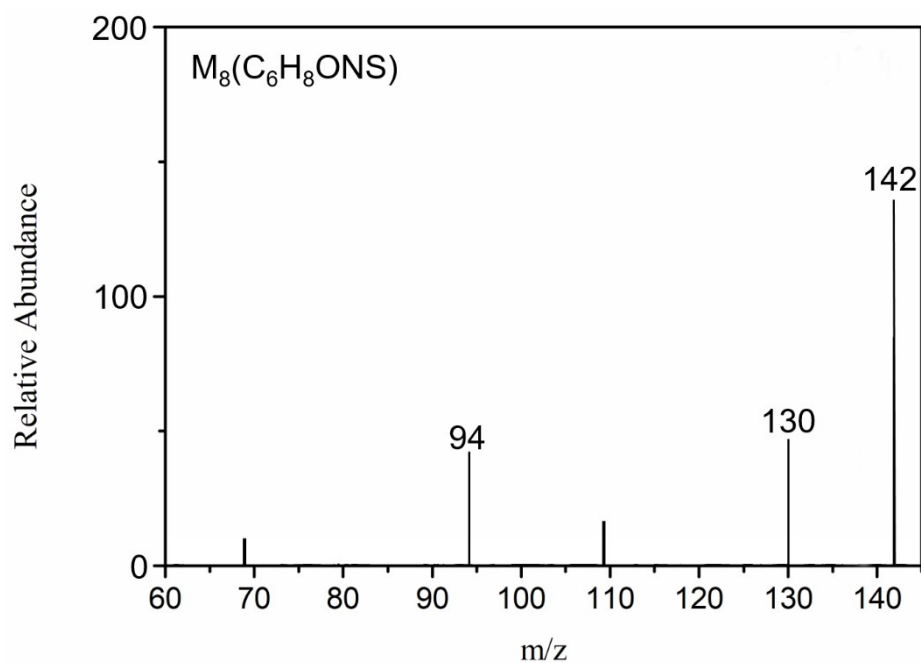


Fig. S12 Mass spectrum of reaction products M_8 during photocatalytic degradation of MB.

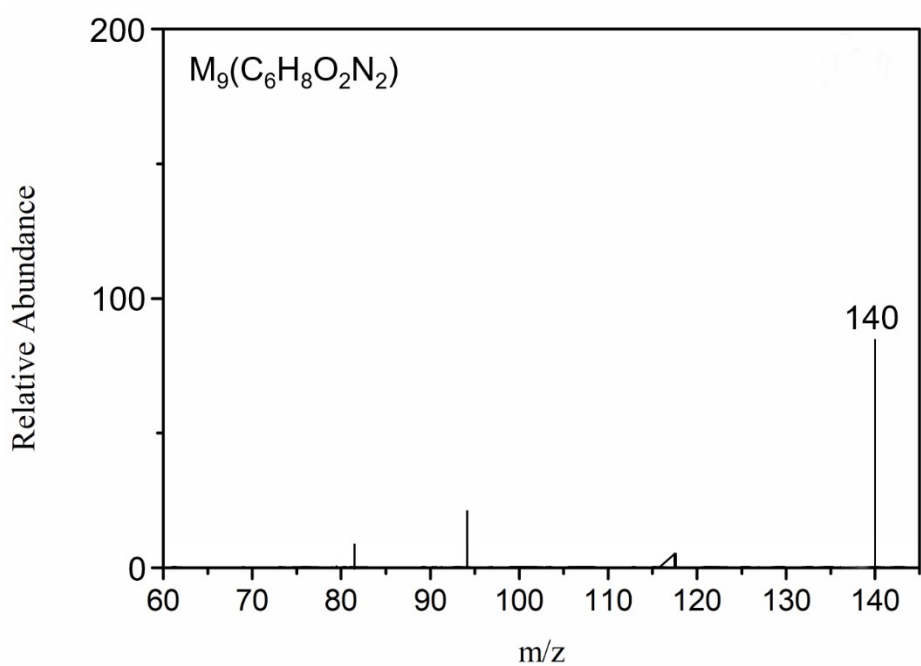


Fig. S13 Mass spectrum of reaction products M_9 during photocatalytic degradation of MB.

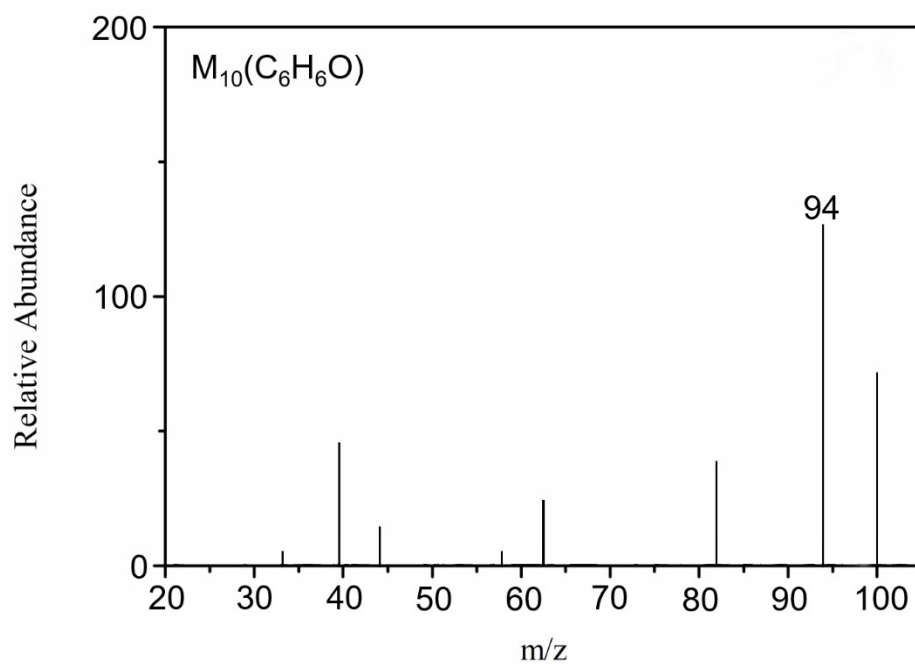


Fig. S14 Mass spectrum of reaction products M_{10} during photocatalytic degradation of MB.

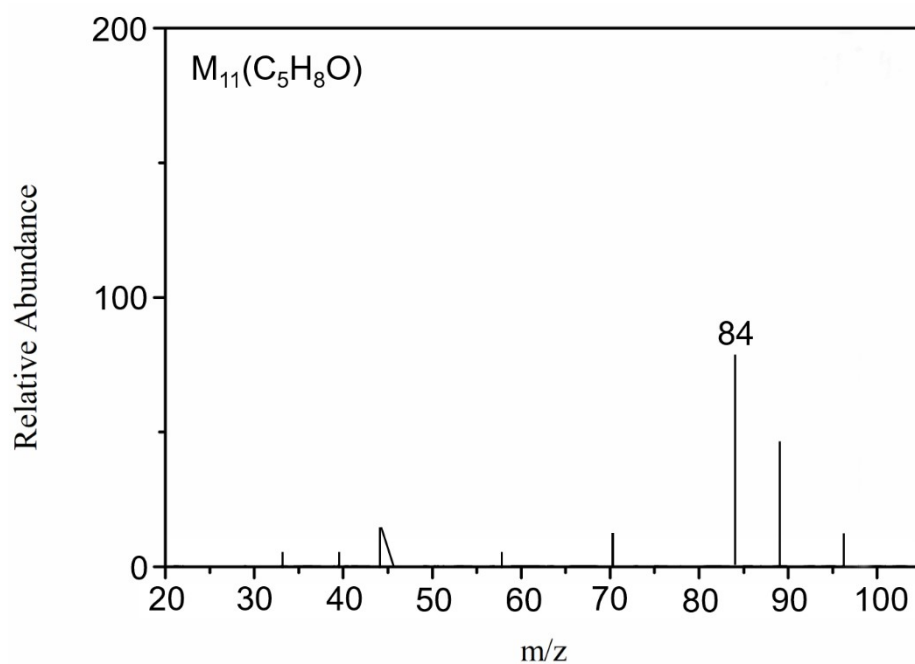


Fig. S15 Mass spectrum of reaction products M_{11} during photocatalytic degradation of MB.

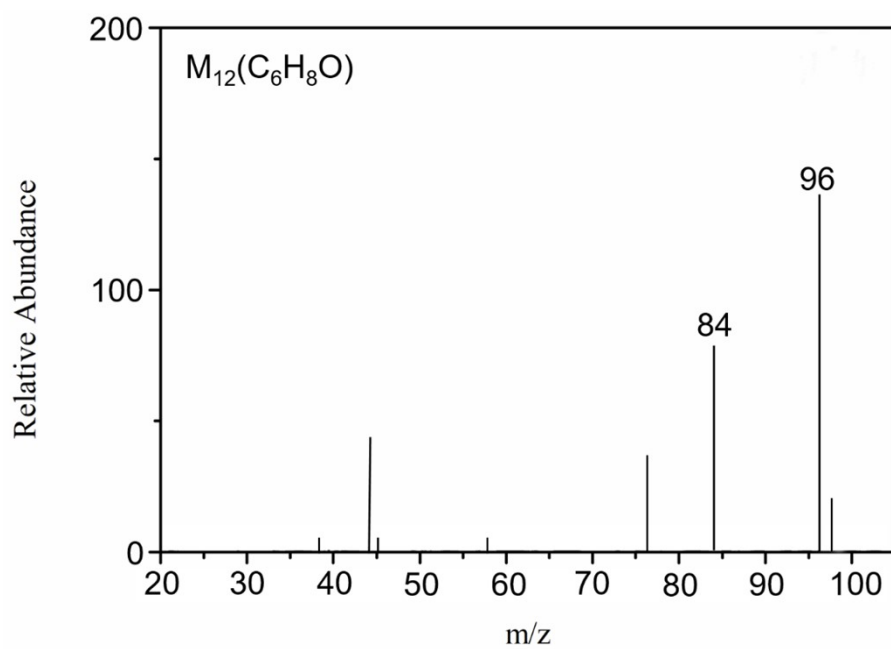


Fig. S16 Mass spectrum of reaction products M_{12} during photocatalytic degradation of MB.

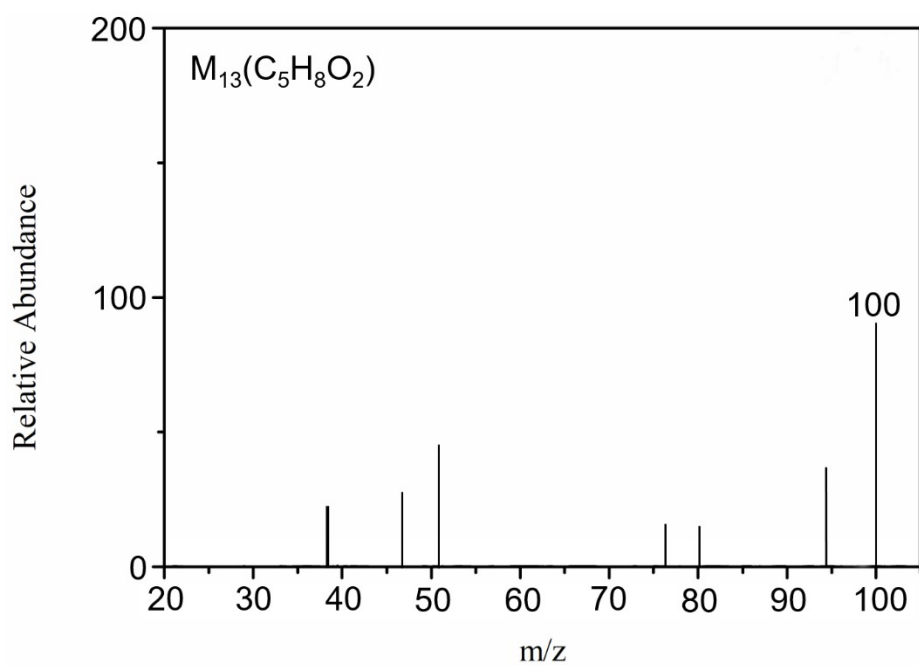


Fig. S17 Mass spectrum of reaction products M_{13} during photocatalytic degradation of MB.

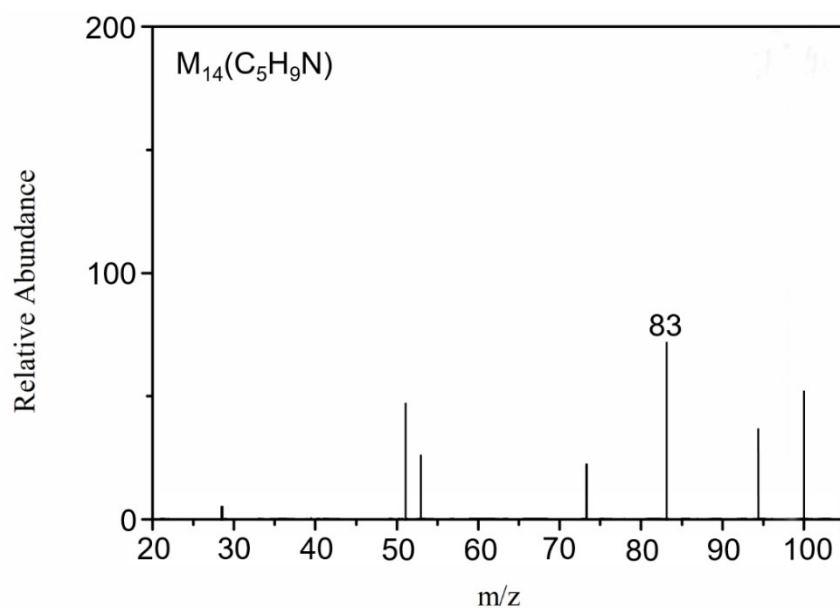


Fig. S18 Mass spectrum of reaction products M_{14} during photocatalytic degradation of MB.

Table S1. Photocatalytic degradation of methylene blue (MB) performance data comparison.

Materials	Response time	Degradation rate	Ref.
$Ce_{0.9}Zr_{0.1}O_2$	120 min	99.04%	This work
B-TiO ₂ /MIL-100(Fe)	60 min	91.12%	1
ZnS/Cu-2%	360 min	56%	2
Cr/CeO ₂	100 min	59%	3
Pd/TiO ₂	120 min	96.90%	4
Ag/CoFe ₂ O ₄	30 min	>95%	5
TM-2-d	720 min	96%	6
SiO ₂ commercial ceramic supports	300 min	94%	7
0.1% Ag-ZnO	210 min	92.90%	8
Bi _{0.9} Gd _{0.07} La _{0.03} FeO ₃	90 min	95%	9
3at% Tb-doped	150 min	98.20%	10

References

1. Liu, L.; Liu, Y.; Wang, X.; Hu, N.; Li, Y.; Li, C.; Meng, Y.; An, Y., Synergistic effect of B-TiO₂ and MIL-100(Fe) for high-efficiency photocatalysis in methylene blue degradation. *Applied Surface Science* **2021**, *561*.
2. Mehrabian, M.; Esteki, Z., Degradation of methylene blue by photocatalysis of copper assisted ZnS nanoparticle thin films. *Optik* **2017**, *130*, 1168-1172.
3. Habib, I. Y.; Burhan, J.; Jaladi, F.; Lim, C. M.; Usman, A.; Kumara, N. T. R. N.; Tsang, S. C. E.; Mahadi, A. H., Effect of Cr doping in CeO₂ nanostructures on photocatalysis and H₂O₂ assisted methylene blue dye degradation. *Catalysis Today* **2021**, *375*, 506-513.
4. Nguyen, C. H.; Fu, C.-C.; Juang, R.-S., Degradation of methylene blue and methyl orange by palladium-doped TiO₂ photocatalysis for water reuse: Efficiency and degradation pathways. *Journal of Cleaner Production* **2018**, *202*, 413-427.
5. Feng, X.; Ma, L.; Cai, F.; Sun, C.; Ding, H., Ag/CoFe₂O₄ as a Fenton-Like Catalyst for the Degradation of Methylene Blue. *ChemistrySelect* **2022**, *7* (22).
6. Zhao, T.; Liu, R.; Lu, J.; Zhu, X.; Zhu, X.; Lu, K.; Zhu, H., Photocatalytic degradation of methylene blue solution by diphenylanthrazoline compounds. *Journal of Physical Organic Chemistry* **2017**, *30* (12).
7. Zarazúa-Morín, M. E.; Galindo-Luna, A. S.; Gallegos-Sánchez, V. J.; Juárez-Ramírez, I., Chemical exfoliation of silica filters used on methylene blue degradation by photocatalysis. *Chemical Papers* **2022**, *76* (9), 5627-5642.
8. Kwon, D.; Kim, J., Silver-doped ZnO for photocatalytic degradation of methylene blue. *Korean Journal of Chemical Engineering* **2020**, *37* (7), 1226-1232.
9. Mani, A. D.; Li, J.; Wang, Z.; Zhou, J.; Xiang, H.; Zhao, J.; Deng, L.; Yang, H.; Yao, L., Coupling of piezocatalysis and photocatalysis for efficient degradation of methylene blue by Bi_{0.9}Gd_{0.07}La_{0.03}FeO₃ nanotubes. *Journal of Advanced Ceramics* **2022**, *11* (7), 1069-1081.
10. Chen, W.-S.; Wu, M.-H.; Wu, J.-Y., Effects of Tb-Doped BiVO₄ on Degradation of Methylene Blue. *Sustainability* **2023**, *15* (8).