

Supporting Information

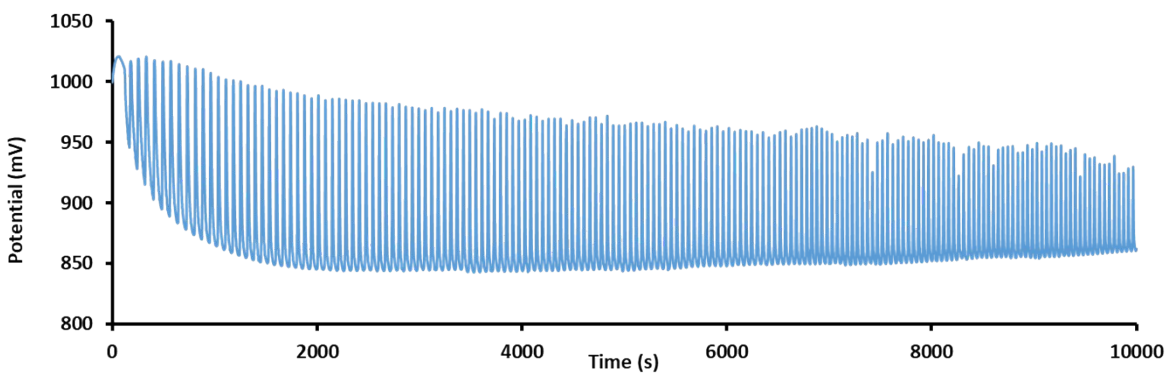


Figure S1. Oscillation of pure B-Z reaction.

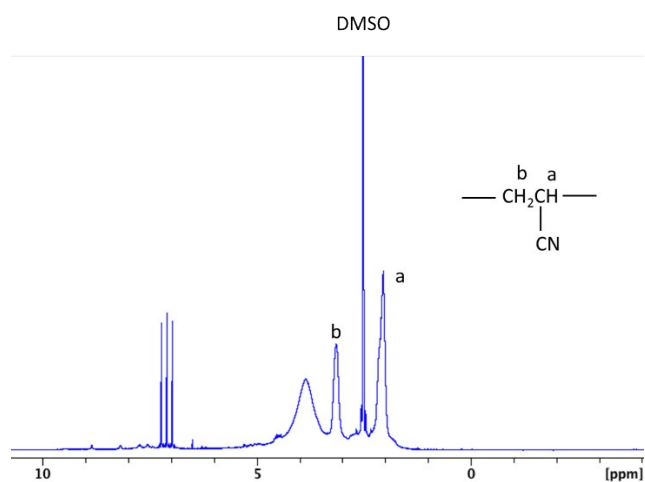


Figure S2. ^1H NMR of polyacrylonitrile.

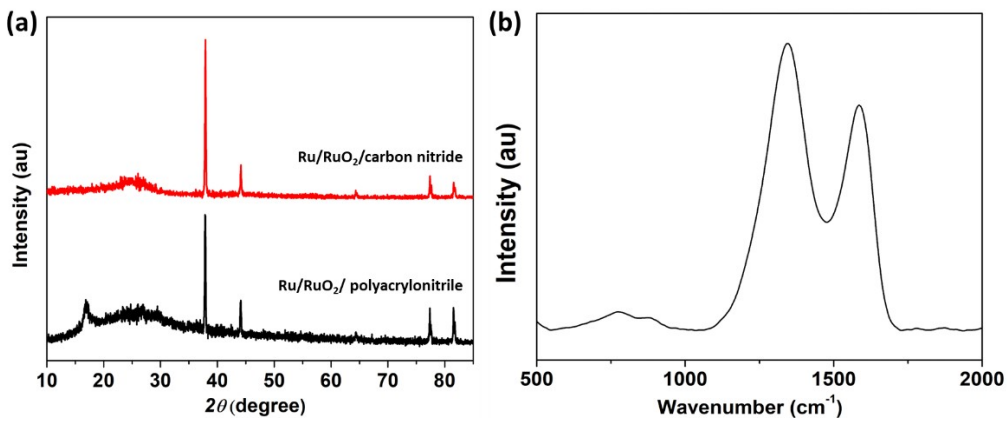


Figure S3. XRD and Raman spectra of carbon nitride.

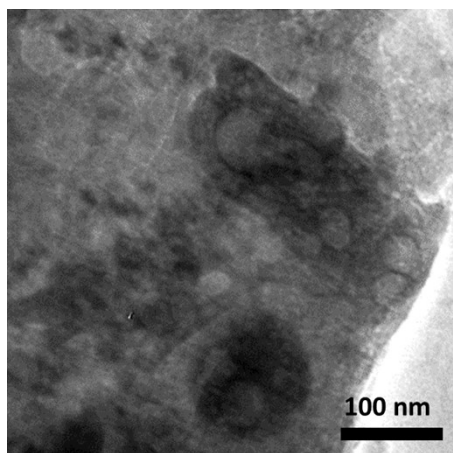


Figure S4. TEM Images of polyacrylonitrile polymer.

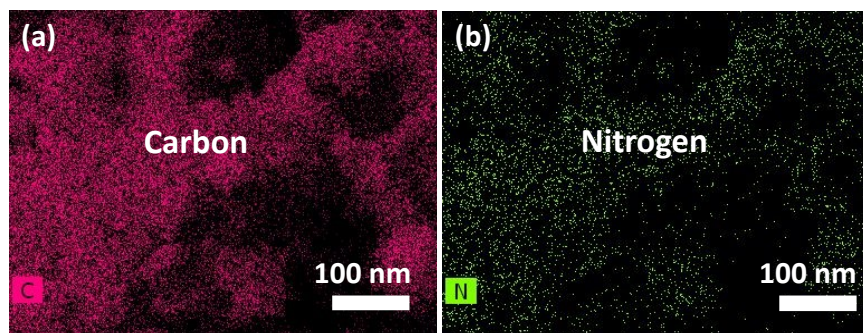


Figure S5. SEM EDX of carbon nitride.

Table S1: Supercapacitor performance of RuO₂ based materials

Electrode materials	Synthesis method	morphology	Electrolyte	Specific capacitance	Ref.
RuO ₂ @Ru	Hydrothermal	Core-shell nanostructure	1 mol/L H ₂ SO ₄	318.5 F/g at 0.5 A/g	[1]
RuO ₂ /C	Hydrothermal and carbonization	Irregular particles	1M H ₂ SO ₄	326.95 F/g at 0.1 A/g	[2]
RuO ₂ /C	Hydrothermal	Nanoparticles	2 M H ₂ SO ₄	155 F/g at 5 mV/s	[3]
RuO ₂ /TiO ₂	Hydrothermal	Nanotube	1 M KOH	46 F/g at 15 mA/cm ²	[4]
MoO ₃ -RuO ₂ /Ti	Annealing	Irregular particles	0.5 M H ₂ SO ₄	639 F/g at 1 A/g	[5]
rGO/RuO ₂	Ultrasonic bath	Nanoparticles	3 M H ₂ SO ₄	328.6 F g ⁻¹ at 0.5 A/g	[6]
GO/RuO ₂ /MWCNT	Hummers	Nanoparticles and nanosheets	3 M H ₂ SO ₄	514.9 F/g at 0.5 A/g	[7]
RuO ₂ @Co ₃ O ₄ /N-GO	Hydrothermal	Nanoparticles and nanosheets	2 M KOH	472 F/g at 0.5 A/g	[8]
C ₃ N ₄ /Ru/RuO ₂	Belousov-Zhabotinsky (B-Z) reaction	Irregular particles	3 M KOH	763 F/g at 0.3 A/g	Our work

References

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