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## Supplementary Data

Boosting photocatalytic activity of  $ZnIn_2S_4$ -based photocatalyst for  $H_2$  evolution using porous  $ZnWO_4$ nanoflakelets as a cocatalyst

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Fig. S1. HRTEM image of ZIS/PZW (5%).



**Fig. S2.** (A) SEM image of ZIS/PZW (5%) and (B, C, D, E and F) corresponding element mappings (Zn, W, O, S and In).



Fig. S3. XPS survey spectrum of ZIS/PZW (5%).

Supplementary Table S1

Table S1. Photocatalytic activity of the  $ZnIn_2S_4$ -based photocatalysts reported recently

| Photocatalysts  | H <sub>2</sub> generation rate (mmol g <sup>-1</sup> h <sup>-1</sup> ) | Ref.      |
|---|--|-----------|
| ZIS/PZW (5%)  | 8.86   | this work |
| $ZnIn_2S_4$ nanosheets on FeWO <sub>4</sub> flowers                     | 3.53   | 1         |
| $NiWO_4/ZnIn_2S_4$ heterojunction                                       | 30.51  | 2         |
| $NiS/ZnIn_2S_4/AgIn(WO_4)_2$ nanocomposite                              | 4.82   | 3         |
| Ag <sub>2</sub> S modified ZnIn <sub>2</sub> S <sub>4</sub> nanosheets  | 1.00   | 4         |
| ZnIn <sub>2</sub> S <sub>4</sub> /black TiO <sub>2</sub> hollow spheres | 5.56   | 5         |



Fig. S4. Photocatalytic activity of ZIS/PZW (5%) for  $H_2$  evolution when the sacrificial agent is (a) methyl alcohol, (b) glycol, (c) triethanolamine, (d)  $Na_2S/Na_2SO_3$  or (e) lactic acid (photocatalyst 15 mg; temperature 10 °C; irradiation time 4 h).



Fig. S5. UPS spectra of (A) the porous  $ZnWO_4$  nanoflakelets and (B) the flower-like  $ZnIn_2S_4$  microspheres.



Fig. S6. Mott-Schottky curves of (A) the flower-like  $ZnIn_2S_4$  microspheres and (B) the porous  $ZnWO_4$  nanoflakelets.



Fig. S7. Fluorescence spectra of (a) the  $ZnIn_2S_4$  microspheres and (b) ZIS/PZW (5%) when the excitation wavelength is 360 nm.



**Fig. S8.** Time-resolved fluorescence spectra of (A) the ZnIn<sub>2</sub>S<sub>4</sub> microspheres and (B) ZIS/PZW (5%).

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