

Cobalt oxide confined in mesoporous SBA-15 as a highly efficient catalyst for enhanced degradation of sulfamethoxazole

Tao Liu^{a,b}, Tingting Jiang^a, Jinqi Zhu^b, Tingwei Rui^b, Shourong Zheng^a, Zhixin Hu^{b*}

^a School of the Environment, Nanjing University, Nanjing 210046, P.R. China

^b School of Environmental Engineering, Nanjing Institute of Technology, Nanjing 211167, P.R. China

Material characterization

The morphology of materials was determined by scanning electron microscope (SEM, Quanta FEG 250, USA). The TEM images of the samples were determined on a TF20 electron microscope using an operating voltage of 200 kV. The XRD patterns of the materials were obtained on an D/max-RA powder diffraction meter (Bruker D8 Advance, Germany), operating with Cu K α radiation (40 kV, 40 mA). The contents of Co were measured by inductively coupled plasma atomic emission spectrometry (ICP-AES, J-A1100, Jarrell-Ash, Franklin, USA). The XPS of the materials was conducted on a ESCALAB 250 X-ray Photoelectron Spectroscopy (Thermo, Co., USA) equipped with a monochromatized Al K α excitation source ($h\nu = 1486.6$ eV).

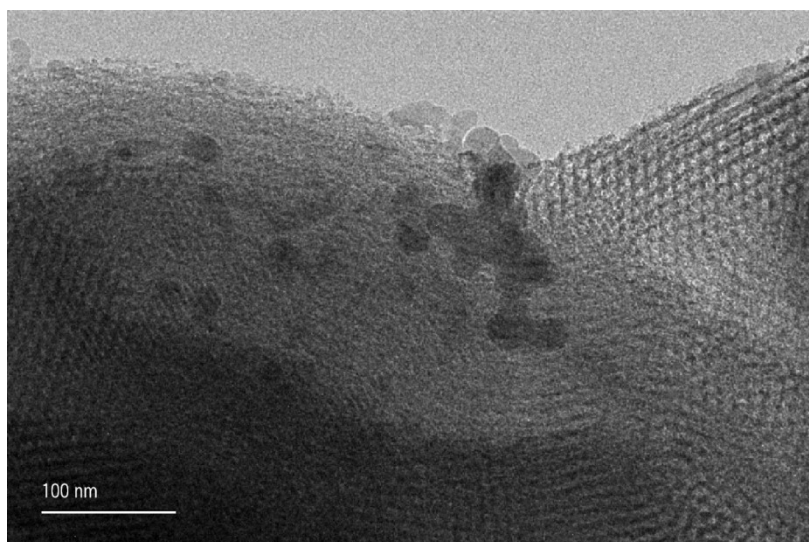


Fig. S1 TEM image of Co₃O₄/SBA-15

Table S1 Surface elemental composition of catalysts determined from XPS

Samples	Co (atom %) ^a	Co ²⁺ /Co ³⁺ ^b
Co ₃ O ₄ (I)	12.57	1.13
Co ₃ O ₄ (II)	12.78	0.72
Co ₃ O ₄ (III)	23.11	0.61
Co ₃ O ₄ /SiO ₂	1.49	0.78
Co ₃ O ₄ /SBA-15	1.32	0.95
Co ₃ O ₄ @SBA-15(I)	0.45	1.19
Co ₃ O ₄ @SBA-15(II)	0.88	0.93
Co ₃ O ₄ @SBA-15(III)	1.05	0.73

^a Atomic percentages of surface elements are obtained from XPS measurements;

^b Calculated by the peak area of Co 2p in XPS spectrum.

Fig. S2 Co 2p_{3/2} XPS spectra of catalysts