

Interfacial engineering to construct IrO_x/WO_3 hetero-structured catalyst for efficient acidic OER catalysis

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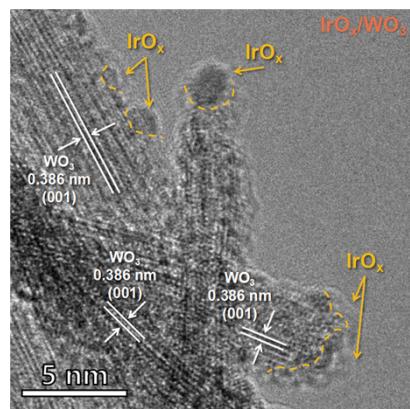


Figure S1. The lattice spacing of $\text{IrO}_x/\text{WO}_3 \cdot x\text{H}_2\text{O}$.

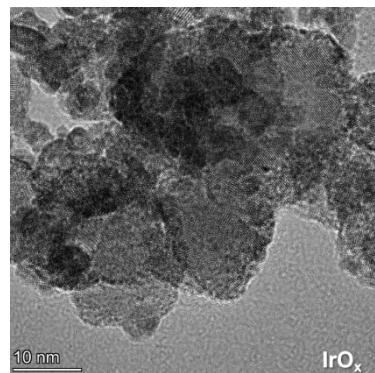


Figure S2. The HAADF-STEM image of IrO_x.

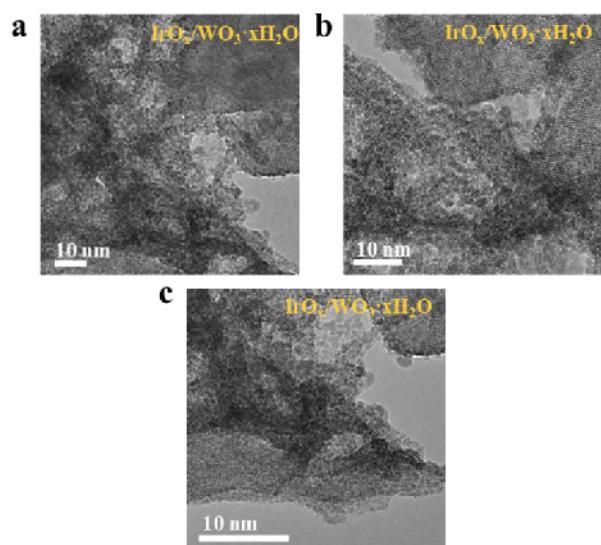


Figure S3. The HR-TEM pictures of $\text{IrO}_x/\text{WO}_3\cdot\text{xH}_2\text{O}$.

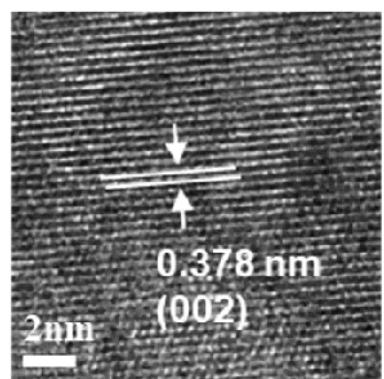


Figure S4. The (002) lattice spacing of $\text{IrO}_x/\text{WO}_3 \cdot x\text{H}_2\text{O}$.

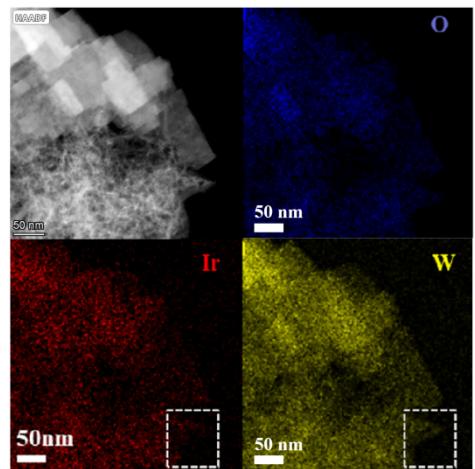


Figure S5. The elements mapping images of $\text{IrO}_x/\text{WO}_3 \cdot x\text{H}_2\text{O}$.

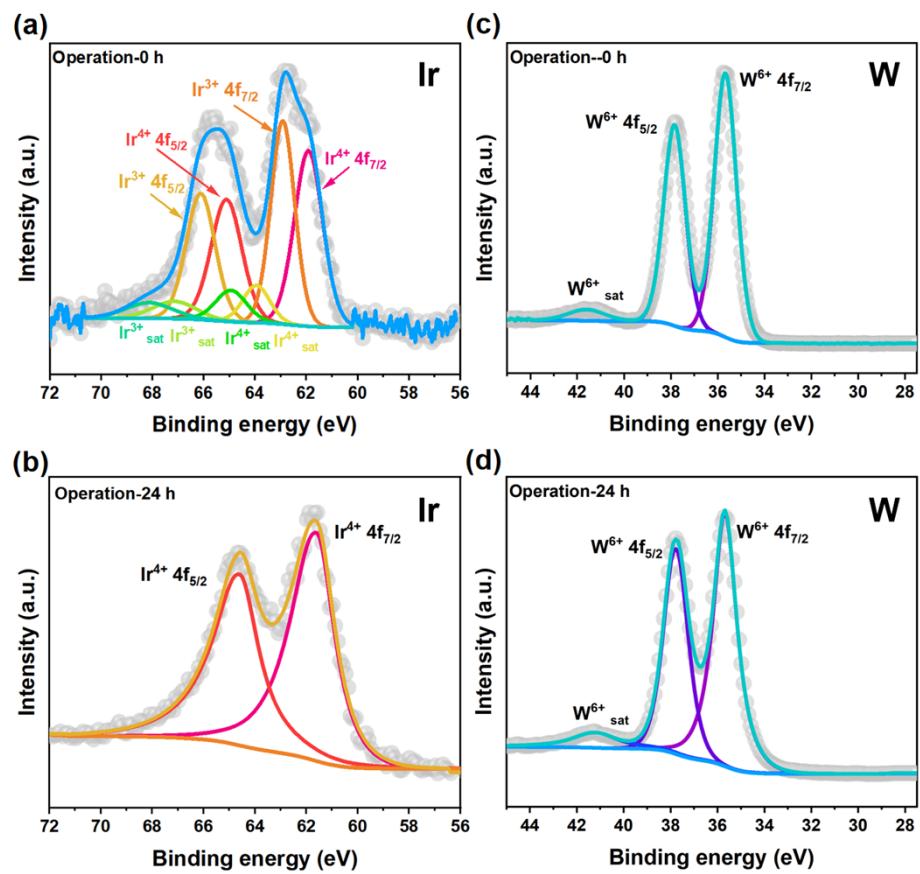


Figure S6. a-d) The XPS analyses of Ir and W for the surface structure during the OER chronopotentiometric test.

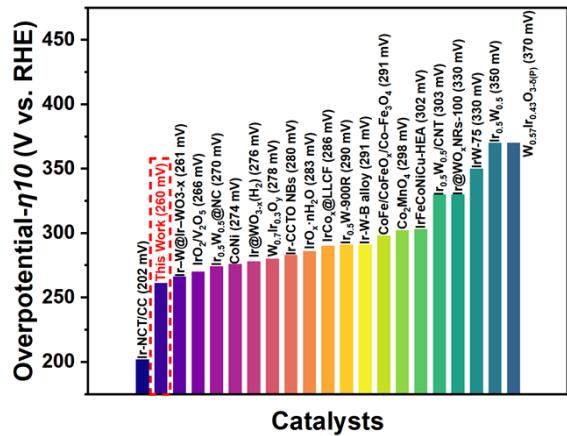


Figure S7. The comparison of overpotentials at 10 mA cm⁻² for IrO_x/WO₃ with latest reported OER catalysts.¹⁻²⁰

Table S1 The element content composition of the IrO_x/WO_3 catalyst.

element	mass fraction (wt%)	atom fraction (at%)
Ir	18.1	8.9
W	72.8	37.7

Table S2 The corresponding parameter of the fitting equivalent circuit.

Samples	$R_s(\Omega)$	$R_{ct}(\Omega)$
IrO_x/WO_3	1.30	5.35
IrO_2	1.32	36.1
$\text{IrO}_x/\text{WO}_3 \cdot x\text{H}_2\text{O}$	1.31	70.6
WO_3	1.30	132.1
$\text{WO}_3 \cdot x\text{H}_2\text{O}$	1.42	117.9

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