

High Value-added Fuel additive production from Waste bio-glycerol over a versatile nanohybrid catalyst

Vijendra Singh,^{a&b} Selvamani Arumugam,^{*a&b} Deepak Joshi,^a Mahesh Kumar,^{a&b} Sadhna Semalty,^c and Nagabhatla Viswanadham,^{*a&b}

^aLight Stock Processing Division, Council of Scientific & Industrial Research-Indian Institute of Petroleum, Dehradun 248005, Uttarakhand, India.

^bAcademy of Scientific and Innovative Research (AcSIR), Ghaziabad-201002, India.

^cPt Lalit Mohan Sharma Campus Rishikesh, Sridev Suman Uttarakhand University Badshahithol 249199, Uttarakhand, India.

*Corresponding Author E-mail: Vijendranegi8@gmail.com

*Corresponding Author E-mail: nvish@iip.res.in

*Corresponding Author E-mail: a.selvamani@iip.res.in

Tel: 91-135-2525856/827; Fax: 91-135-2660202

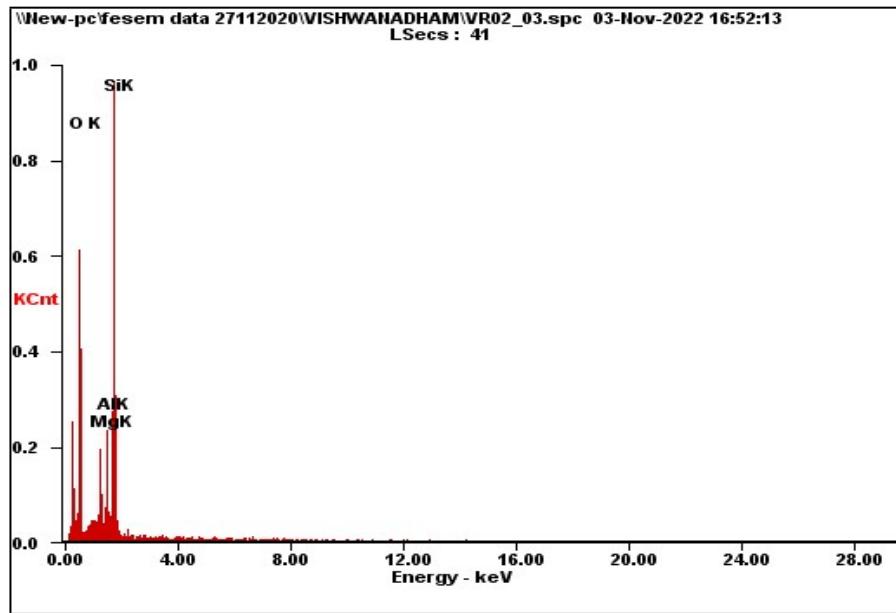


Fig. S1 EDX spectrum of MAZ hybrid catalyst sample

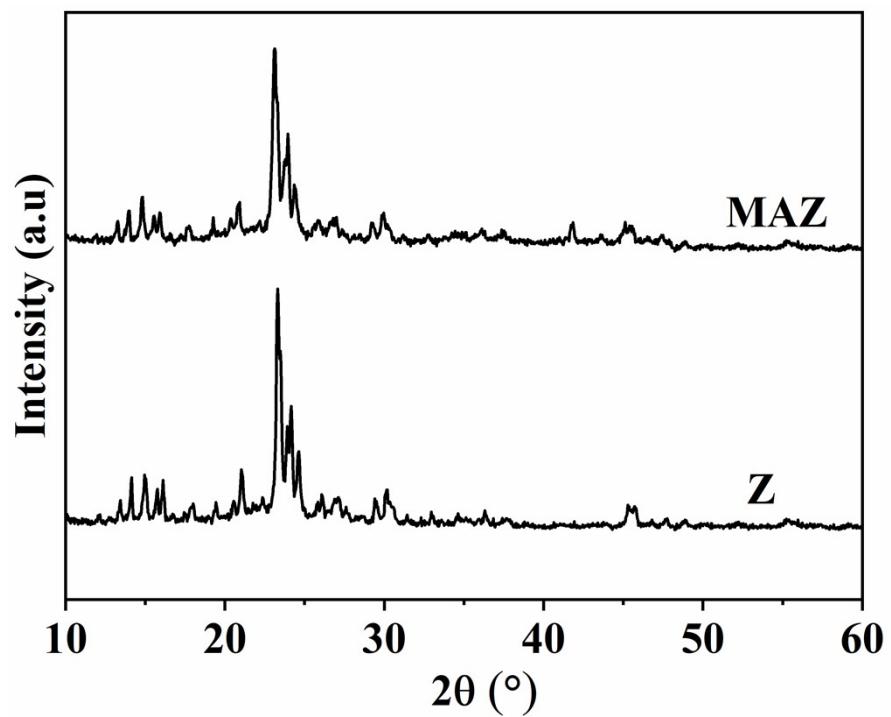


Fig. S2 XRD diffractogram of catalyst samples

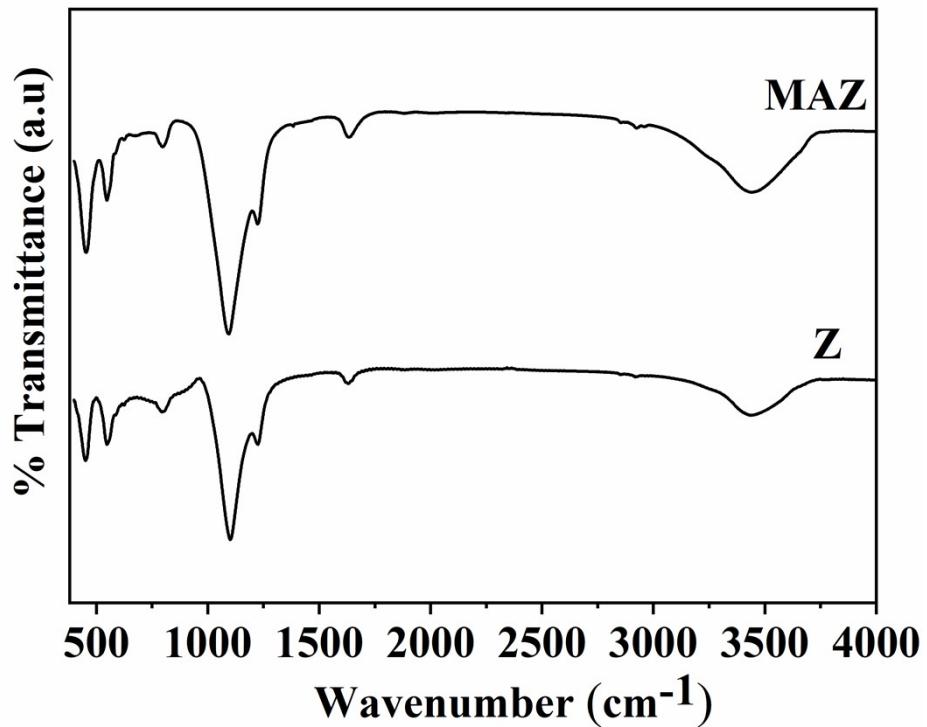


Fig. S3 FT-IR patterns of catalyst samples

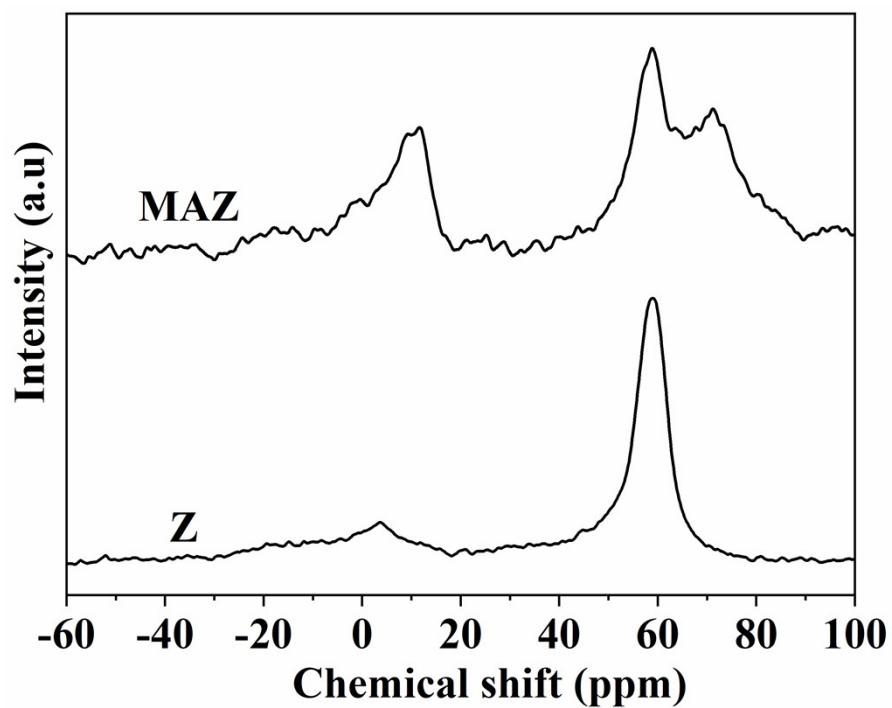


Fig. S4 ^{27}Al NMR spectra of catalyst samples

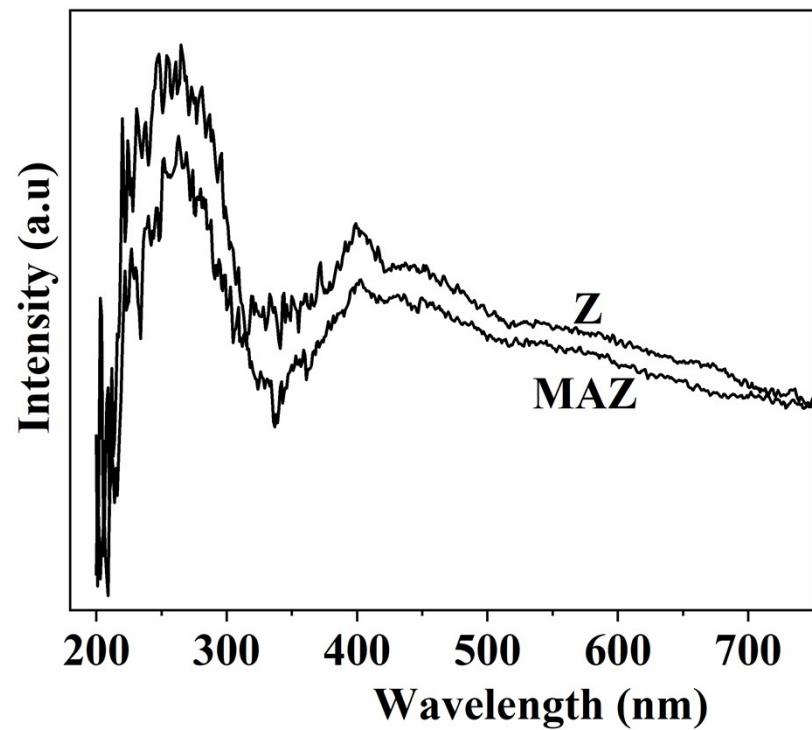


Fig. S5 UV-Vis spectra of catalyst samples

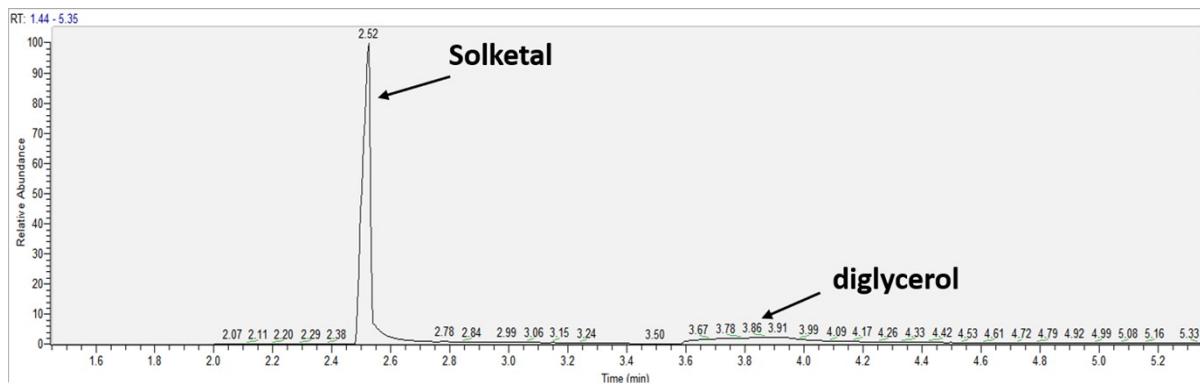


Fig. S6 GC-MS spectrum of a liquid product obtained with MAZ nanohybrid catalyst

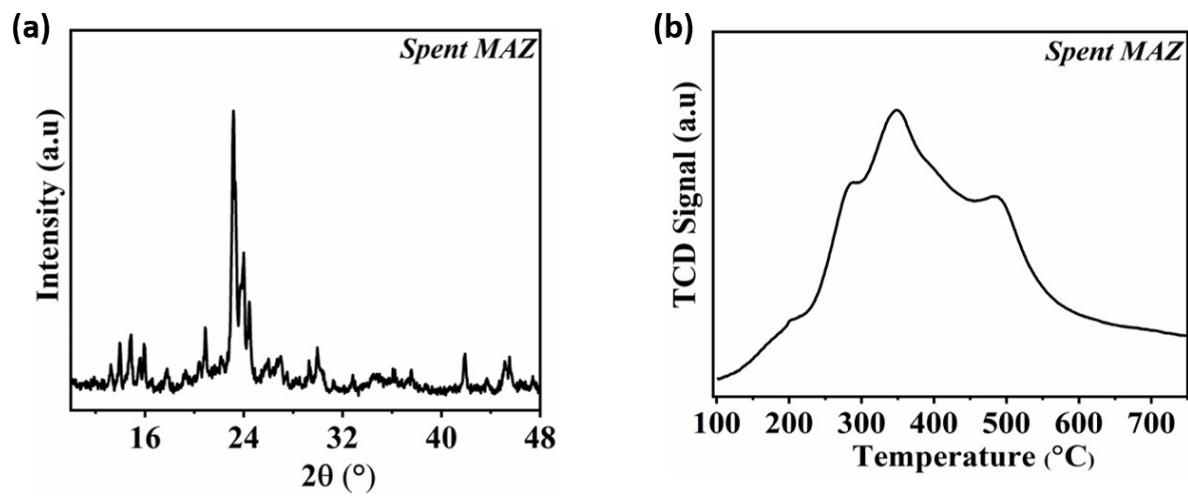


Fig. S7 Spent catalyst results (A) XRD patterns, and (B) NH₃-TPD profile

Table S1 Comparison of literature reports with the present study

S. No	Catalyst	Reaction conditions (Temperature/time)	Conversion (%)	Yield (%)	Ref.
1	NbO ₂ (OH)	70 °C/60 min	42	-	[1]
2	Hydrophobized NbO ₂ (OH)	70 °C/60 min	65	-	[1]
3	Carbon catalyst (GC-1:2)	RT/180 min	80	-	[2]
4	Carbon catalyst (GC-1:3)	RT/180 min	80	-	[2]
5	Ar-SBA-15	70 °C/30 min	82	-	[3]
6	Amberlyst-15	70 °C/30 min	86	-	[3]
7	Nb ₅ -HUSY	40 °C/180 min	66	65	[4]
8	Nb ₁₅ -HUSY	40 °C/180 min	61	60	[4]
9	Ni-MOR	100 °C/15 min	93	-	[5]
10	Cu-MOR	100 °C/15 min	95	-	[5]
11	H-ZSM-5	50 °C/60 min, methanol (solvent)	85	-	[6]
12	H-Beta	50 °C/60 min, methanol (solvent)	86	-	[6]
13	H-USY	50 °C/60 min, methanol (solvent)	69	-	[6]
14	HR/Y zeolite	50 °C/120 min	93	72	[7]
15	HR/Y-W ₂₀ zeolite	50 °C/120 min	98	94	[7]
16	UiO-SO ₃ H-0.2	60 °C/240 min	70.3	70.1	[8]
17	ZSM-5	70 °C/35 min	22	-	[9]
18	p-toluene-sulfonic acid	70 °C/35 min	80	-	[9]
19	Amberlyst-15	70 °C/35 min	98	-	[9]
20	HZSM-5	50 °C/120 min	-	15	[10]

21	NaOH-pretreated HZSM-5	50 °C/120 min	-	30	[10]
22	HZSM-5 film capillary	50 °C/120 min	-	29	[10]
23	ZSM-5	60 °C/180 min	83	36.2	[11]
24	Z-beta	60 °C/180 min	89	52.2	[11]
25	Z-mordenite	60 °C/180 min	93	68.1	[11]
26	HZSM-5	60 °C/90 min	40	-	[12]
27	WO ₃ /Nb ₂ O ₅	RT/150 min	92.3	-	[13]
28	MoO ₃ -ZrO ₂	60 °C/10 min	90	84	[14]
29	H-Beta-1 zeolite	RT/ 60 min	86	84.7	[15]
30	(C ₃ H ₇) ₄ N ⁺ /PWA	30 °C /120 min	94	-	[16]
31	Hf-TUD-1	80 °C/360 min	52	-	[17]
32	H-BEA Zeolite	60 °C/60 min	72.6	-	[18]
33	BEA Zeolite	30 °C/30 min	76	-	[19]
34	1.0 M SABEA	30 °C/30 min	84	-	[19]
35	γ-Al ₂ O ₃ -IPA	25 °C/30 min	29.1	0	[20]
36	0.2 mol% FeCl ₃ /γ-Al ₂ O ₃ -IPA	25 °C/30 min	99.89	98.25	[20]
37	<i>MAZ nanohybrid</i>	60 °C/120 min	100	>99	<i>This Work</i>

Table S2 Composition of crude-bio-glycerol produced from bio-diesel process

Glycerol % (w/w)	86
Water % (w/w)	13
Density (g/ml)	1.23
Fatty acid esters % (w/w)	0.05
Ash % (w/w)	0.005

References

- [1] T. E. Souza, M. F. Portilho, P. M. T. G. Souza, P. P. Souza, L. C. A. Oliveira, *ChemCatChem* **2014**, *6*, 2961–2969.
- [2] M. Gonçalves, R. Rodrigues, T. S. Galhardo, W. A. Carvalho, *Fuel* **2016**, *181*, 46–54.
- [3] G. Vicente, J. A. Melero, G. Morales, M. Paniagua, E. Martín, *Green Chem.*, **2010**, *12*, 899.
- [4] C. Ferreira, A. Araujo, V. Calvino-Casilda, M. G. Cutrufello, E. Rombi, A. M. Fonseca, M. A. Bañares, I. C. Neves, *Microporous Mesoporous Mater.*, **2018**, *271*, 243–251.
- [5] S. S. Priya, P. R. Selvakannan, K. V. R. Chary, M. L. Kantam, S. K. Bhargava, *Mol. Catal.*, **2017**, *434*, 184–193.
- [6] J. Kowalska-Kuś, A. Held, K. Nowińska, *Chem. Eng. J.*, **2020**, *401*, 126143.
- [7] A. Talebian-Kiakalaieh, S. Tarighi, *J. Ind. Eng. Chem.*, **2019**, *79*, 452–464.
- [8] Y. Jiang, R. Zhou, B. Ye, Z. Hou, *J. Ind. Eng. Chem.*, **2022**, *110*, 357–366.
- [9] C. X. A. da Silva, V. L. C. Gonçalves, C. J. A. Mota, *Green Chem.*, **2009**, *11*, 38–41.
- [10] X. Huang, G. Zhang, L. Zhang, Q. Zhang, *ACS Omega* **2020**, *5*, 20784–20791.
- [11] B. K. Raja, N. Mohindra, U. Goswami, B. Modhera, *Journal of The Institution of Engineers (India): Series E* **2022**, *103*, 145–148.
- [12] M. Alsawalha, *Front Chem* **2019**, *7*, DOI 10.3389/fchem.2019.00799.
- [13] S. B. Putla, P. Subha, B. Swapna, N. Singh, P. Sudarsanam, *Catal Commun* **2024**, *186*, 106827.
- [14] H. Huang, J. Mu, M. Liang, R. Qi, M. Wu, L. Xu, H. Xu, J. Zhao, J. Zhou, Z. Miao, *Mol. Catal.*, **2024**, *552*, 113682.
- [15] P. Manjunathan, S. P. Maradur, A. B. Halgeri, G. V. Shanbhag, *J Mol Catal A Chem.*, **2015**, *396*, 47–54.
- [16] S. Sandesh, A. B. Halgeri, G. V. Shanbhag, *J Mol Catal A Chem.*, **2015**, *401*, 73–80.
- [17] L. Li, T. I. Korányi, B. F. Sels, P. P. Pescarmona, *Green Chem.*, **2012**, *14*, 1611.
- [18] V. Rossa, Y. da S. P. Pessanha, G. Ch. Díaz, L. D. T. Câmara, S. B. C. Pergher, D. A. G. Aranda, *Ind Eng Chem Res.*, **2017**, *56*, 479–488.
- [19] N. J. Venkatesha, Y. S. Bhat, B. S. Jai Prakash, *RSC Adv* **2016**, *6*, 18824–18833.
- [20] T. Zhang, X. Xin, S. Liao, H. Liu, Y. Yu, B. Xu, *ChemistrySelect* **2023**, *8*, DOI 10.1002/slct.202205005.