## Supplementary Material

## Multi-heteroatom-doping promotes molecular oxygen activation on polymeric carbon nitride for simultaneous generation of $H_2O_2$ and

## degradation of oxcarbazepine

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Fig. S1. The N 1s high-resolution XPS spectra of PCN and K, P, O-CN<sub>5</sub>.



**Fig. S2.** TEM images of (a) PCN, (b) K, P, O-CN<sub>5</sub>, (c) The elements mapping images of K, P, O-CN<sub>5</sub>.



Fig. S3. UV–vis absorption spectra of  $H_2O_2$  produced with IPA in photocatalytic stage of K, P, O-CN<sub>1</sub> at different times.



Fig. S4. UV–vis absorption spectra of  $H_2O_2$  produced with IPA in photocatalytic stage of K, P, O-CN<sub>3</sub> at different times.



Fig. S5. UV–vis absorption spectra of  $H_2O_2$  produced with IPA in photocatalytic stage of K, P, O-CN<sub>5</sub> at different times.



Fig. S6. UV–vis absorption spectra of  $H_2O_2$  produced with IPA in photocatalytic stage of K, P, O-CN<sub>15</sub> at different times.



Fig. S7. UV–vis absorption spectra of  $H_2O_2$  produced with IPA in photocatalytic stage of PCN at different times.



**Fig. S8.** The first-order reaction kinetic constants of OXC degradation on PCN as well as K, P, O-CN<sub>5</sub>.



Fig. S9. Photocatalytic  $H_2O_2$  production of K, P, O-CN<sub>5</sub> in the presence of oxcarbazepine with different concentrations.



**Fig. S10.** Degradation efficiency of K, O, P-CN<sub>5</sub> in oxcarbazepine solutions of different concentrations.



Fig. S11. Mott-Schottky plots collected at different frequencies of PCN.



Fig. S12. Mott-Schottky plots collected at different frequencies of K, P, O-CN<sub>5</sub>.



Fig. S13. EPR signals of PCN and K, P, O-CN<sub>5</sub>.







Fig. S14. HPLC-MS images of the degradation of OXC.

	m/z	Molecular structure	Molecular formular
OXC	252		$C_{15}H_{12}N_2O_2$
P1	268		$C_{15}H_{12}N_2O_3$
Р2	225		C <sub>14</sub> H <sub>11</sub> NO <sub>2</sub>
Р3	207		C <sub>14</sub> H <sub>9</sub> NO
P4	179		C <sub>13</sub> H <sub>9</sub> N
Р5	268		C <sub>15</sub> H <sub>12</sub> N <sub>2</sub> O <sub>3</sub>
P6	284		C <sub>15</sub> H <sub>12</sub> N <sub>2</sub> O <sub>4</sub>
Р7	266		$C_{15}H_{10}N_2O_3$
P8	282		C <sub>15</sub> H <sub>10</sub> N <sub>2</sub> O <sub>4</sub>

**Table. S1.** Possible degradation intermediates of OXC.

Р9	223	O T T T T T T T T T T T T T	C <sub>14</sub> H <sub>9</sub> NO <sub>2</sub>
P10	223		C <sub>14</sub> H <sub>9</sub> NO <sub>2</sub>
P11	223		C <sub>14</sub> H <sub>9</sub> NO <sub>2</sub>
P12	195		C <sub>13</sub> H <sub>9</sub> NO
P13	195	© ↓ ↓ ↓ ↓ ↓ ↓ ↓ ↓ ↓ ↓ ↓ ↓ ↓ ↓ ↓ ↓ ↓ ↓ ↓	C <sub>13</sub> H <sub>9</sub> NO
P14	211		C <sub>13</sub> H <sub>9</sub> NO <sub>2</sub>