

**Supporting information**

**Novel 2D/2D 1T-MoS<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>T<sub>x</sub> Heterostructures for High-Voltage Symmetric Supercapacitors**

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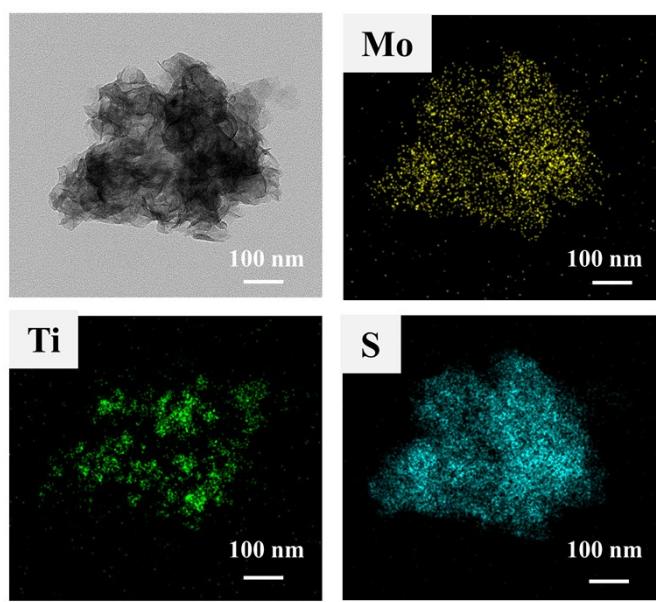


Figure S1 The elemental mapping images of Mo, Ti, and S elements of 1T-MoS<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>T<sub>Z</sub>

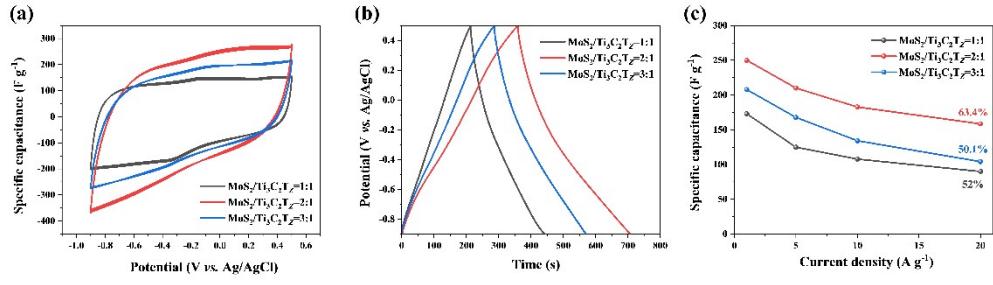


Figure S2 (a)-(b) The CV curves, GCD curves of the 1T-MoS<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>T<sub>z</sub>-1:1, 1T-MoS<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>T<sub>z</sub>-2:1 and 1T-MoS<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>T<sub>z</sub>-3:1; (c) The  $C_g$  of the 1T-MoS<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>T<sub>z</sub>-1:1, 1T-MoS<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>T<sub>z</sub>-2:1 and 1T-MoS<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>T<sub>z</sub>-3:1 electrodes plotted as a function of current densities.

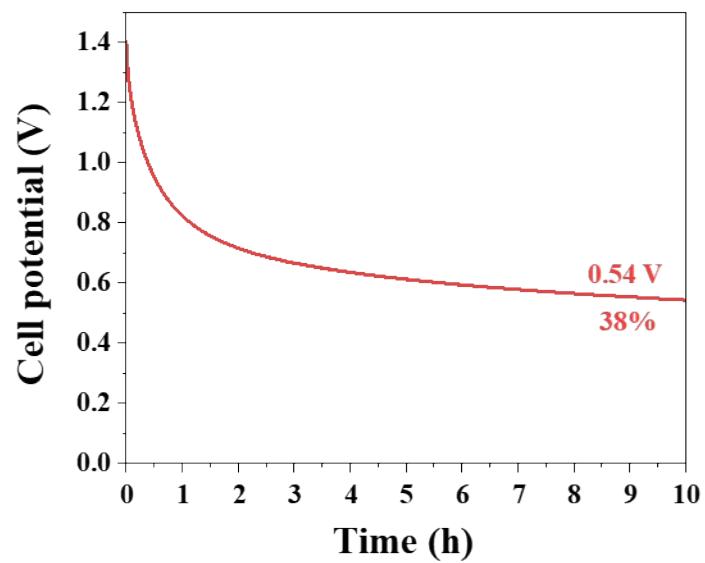


Figure S3 Self-discharge curves of SSCs.

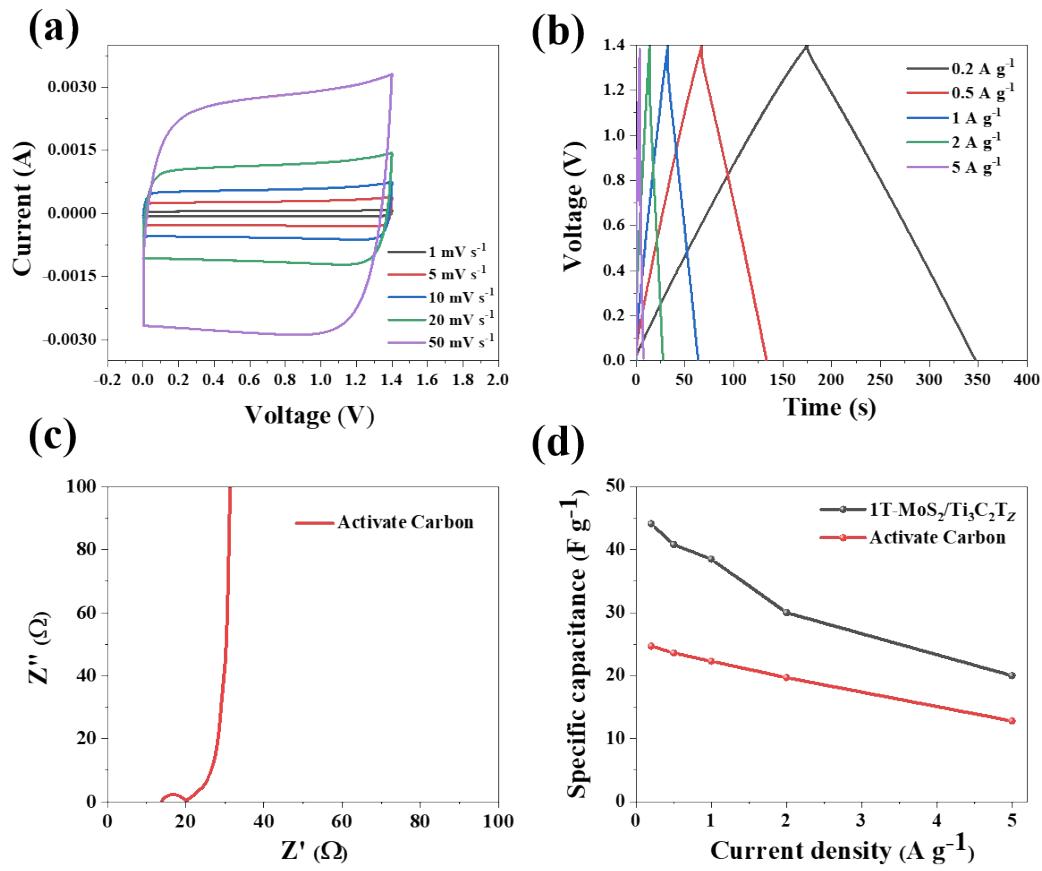


Figure S4 Electrochemical performance of AC SSCs: (a) CV curves at different scan rates from 1 to 50 mV s<sup>-1</sup>. (b) GCD curves at different current densities from 0.2 to 5 A g<sup>-1</sup>. (c) Nyquist plots of the AC SSCs. (d) The gravimetric capacitance ( $C_g$ ) of the 1T-MoS<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>T<sub>2</sub> and AC electrodes plotted as a function of current densities.

Table S1. Comparison of  $R_s$ ,  $R_{ct}$ , CPE-T, and CPE-P element values in equivalent circuits for  $Ti_3C_2T_Z$ , 1T-MoS<sub>2</sub>, and 1T-MoS<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>T<sub>Z</sub> EIS data.

	$R_s$ (ohm cm <sup>-2</sup> )	$R_{ct}$ (ohm cm <sup>-2</sup> )	CPE-T (F s <sup>(a-1)</sup> )	CPE-P (F s <sup>(a-1)</sup> )
Ti <sub>3</sub> C <sub>2</sub> T <sub>Z</sub>	0.84	4.01	$6.57 \times 10^{-7}$	1.06
1T-MoS <sub>2</sub>	1.27	4.53	$1.26 \times 10^{-6}$	1.03
1T-MoS <sub>2</sub> /Ti <sub>3</sub> C <sub>2</sub> T <sub>Z</sub>	0.90	3.94	$7.27 \times 10^{-7}$	1.04

Table S2. Comparative study of MoS<sub>2</sub>-based electrode materials for specific capacitance.

Electrode material	Electrolyte	Specific capacitance for single electrode	Cycle stability	Ref
1T-MoS <sub>2</sub> /Ti <sub>3</sub> C <sub>2</sub> T <sub>Z</sub>	20 m LiCl	250 F g <sup>-1</sup> at 1 A g <sup>-1</sup>	82.3% for 5000 cycles	our work
MoS <sub>2</sub> film on Cu foil	0.5 M H <sub>2</sub> SO <sub>4</sub>	33 mF cm <sup>-2</sup> at 25.47 mA cm <sup>-2</sup>	97% for 5000 cycles	1
MoS <sub>2</sub>	1 M Na <sub>2</sub> SO <sub>4</sub>	138 F g <sup>-1</sup> at 1 Ag <sup>-1</sup>	86% for 5000 cycles	2
1T/2H hybrid MoS <sub>2</sub>	1 M KCl	259 F g <sup>-1</sup> at 5 mV s <sup>-1</sup>	104.7% for 1000 cycles	3
1T-MoS <sub>2</sub> hydrogel	1 M H <sub>2</sub> SO <sub>4</sub>	147 F g <sup>-1</sup> at 0.1 A g <sup>-1</sup>	90% for 5000 cycles	4
MoS <sub>2</sub> /graphene	1 M Na <sub>2</sub> SO <sub>4</sub>	270 F g <sup>-1</sup> at 0.1 A g <sup>-1</sup>	89% for 1000 cycles	5
1T-MoS <sub>2</sub> /Ti <sub>3</sub> C <sub>2</sub> MXene	1 M Na <sub>2</sub> SO <sub>4</sub>	206 F g <sup>-1</sup> at 1 A g <sup>-1</sup>	94.4% for 10000 cycles	6
1T-MoS <sub>2</sub> /Ti <sub>3</sub> C <sub>2</sub>	1 M H <sub>2</sub> SO <sub>4</sub>	303.8 F g <sup>-1</sup> at 1 A g <sup>-1</sup>	82% for 10000 cycles	7
MoS <sub>2</sub> @CNT/rGO	PVA/H <sub>2</sub> SO <sub>4</sub>	126.8 mF cm <sup>-2</sup> at 0.1 mA cm <sup>-2</sup>	94.6% for 10000 cycles	8
PPy/MoS <sub>2</sub>	1 M KCl	384 F g <sup>-1</sup> at 0.5 A g <sup>-1</sup>	96.47% for 1000 cycles	9
MoS <sub>2</sub> -Cu <sub>3</sub> N	1 M Na <sub>2</sub> SO <sub>4</sub>	215.47 F g <sup>-1</sup> at 0.5 A g <sup>-1</sup>	90% for 2000 cycles	10
MoS <sub>2</sub> /MoO <sub>2</sub> @CNT	1 M KOH	228.4 F g <sup>-1</sup> at 0.5 A g <sup>-1</sup>	187.1% for 25000 cycles	11
Fe <sub>2</sub> O <sub>3</sub> /MoS <sub>2</sub>	3 M KOH	266 F g <sup>-1</sup> at 0.5 A g <sup>-1</sup>	82% for 2000 cycles	12
MoS <sub>2</sub> nanosheets on Mo foil	1 M Na <sub>2</sub> SO <sub>4</sub>	192.7 F g <sup>-1</sup> at 1 mA cm <sup>-2</sup>	98% for 1000 cycles	13

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