

Supporting Information for  
**DNA-Modulated Single-atom Nanozyme with Enhanced  
Enzyme-Like Activity for Ultrasensitive Detection of Dopamine**

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## Materials and methods

### Chemical and materials.

All oligonucleotide sequences were synthesized by Sangon Biotechnology Co. Ltd. (Shanghai, China) and shown in Table S1. All sequences were rapidly denatured at 95°C for 5 min and then cooled to room temperature for the following studies. Zinc nitrate hexahydrate ( $\text{Zn}(\text{NO}_3)_2 \cdot 6\text{H}_2\text{O}$ ), Hemin (95%), 2-methylimidazole (2-MI, 98%), 3,3',5,5'-tetramethylbenzidine (TMB), hydrogen peroxide ( $\text{H}_2\text{O}_2$ , 30%), 2,2'-azinobis (3-ethylbenzothiazoline-6-sulfonic acid) diammonium salt (ABTS), ascorbic acid, *L*-cysteine, *L*-phenylalanine were purchased from Aladdin (Shanghai, China). DA and *L*-tyrosine were purchased from Meryer (Shanghai, China). Glutathione, glucose, lactose, and uric acid were purchased from Energy Chemical (Shanghai, China). All reagents were used directly without further purification. Ultrapure water ( $\geq 18 \text{ M}\Omega$ , Millipore) was used in the whole experiment.

### Apparatus and characterization.

Transmission electron microscopy (TEM) images and element mappings were gained from a JEM-2100F with an accelerating voltage of 200 kV. Scanning electron microscope (SEM) images were performed on a TESCAN MIRA LMS (Czech). X-ray photoelectron spectroscopy (XPS) measurement was performed on a Thermo Scientific K-Alpha with Al K $\alpha$  X-ray radiation as the X-ray source for excitation. The crystal structure of samples was identified by X-ray diffractometer (XRD) using Cu K $\alpha$  radiation (Rigaku SmartLab SE, Japan). Zeta potential of the materials was measured on a nano-particle size and Zeta potential analyzer (Zetasizer nano ZS90, Malvern Panalytical). Ultraviolet-visible (UV-vis) absorption spectra was collected on a UV-vis-NIR spectrophotometer (UV-3600 Plus, Shimadzu, Japan). Fluorescence analysis was carried out on a FS05 steady-state transient fluorescence spectrometer (Thermo Field, USA).

### Synthesis of N-C

The synthetic route of N-C is similar to that of Fe-N-C, except that hemin is not added in the first step to obtain ZIF-8.

### **Kinetic measurements.**

The steady-state kinetic analysis was carried out using 0.5  $\mu\text{g mL}^{-1}$  Fe-N-C or DNA/Fe-N-C in reaction volume of acetate buffer (20 mM, pH 4.0) at room temperature with diverse concentrations of  $\text{H}_2\text{O}_2$  (TMB) and 0.5 mM TMB (5 mM  $\text{H}_2\text{O}_2$ ). According to the typical Michaelis-Menten equation:  $v = V_{\text{max}} [C] / (K_M + [C])$ , where  $v$  is the initial velocity,  $V_{\text{max}}$  is the maximal reaction velocity,  $[C]$  is the concentration of substrate (TMB or  $\text{H}_2\text{O}_2$ ), and  $K_M$  is Michaelis constant.

### **Selectivity analyses of DNA/Fe-N-C for dopamine (DA).**

To verify the selectivity of the DNA/Fe-N-C-based sensing platform for DA, various amino acids and small molecules such as cysteine, phenylalanine, tyrosine, uric acid, lactose, glucose, glutathione, ascorbic acid were similarly tested replaced with 40  $\mu\text{M}$  of DA.

### **Detection of $\cdot\text{OH}$**

Typically, Fe-N-C or DNA/Fe-N-C solution (10  $\mu\text{g mL}^{-1}$ ), terephthalic acid (TA) solution (0.5 mM) and  $\text{H}_2\text{O}_2$  solution (10 mM) were mixed in acetate buffer (pH 4.0, 20 mM) until the final volume reached 1 mL. After incubating for 3 h, the solutions were used for the detection of  $\cdot\text{OH}$ .

## Supporting tables and figures

**Table S1** Sequences of synthesized DNA

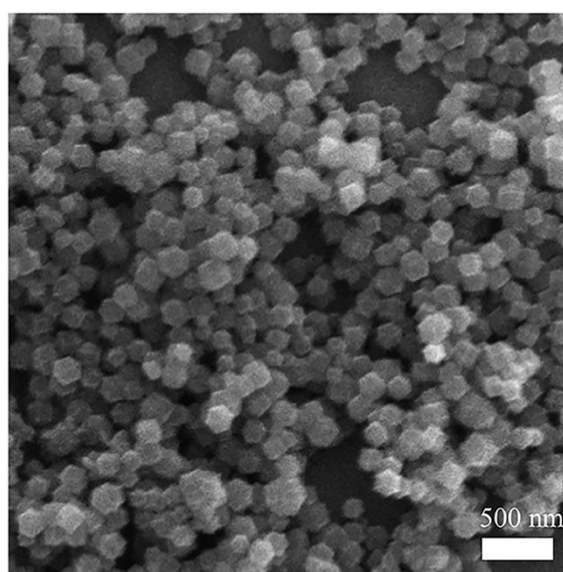
Name	Sequences (5'-3')
G <sub>14</sub>	GGGGGGGGGGGGGGG
T <sub>14</sub>	TTTTTTTTTTTTTTT
C <sub>14</sub>	CCCCCCCCCCCCCC
A <sub>14</sub>	AAAAAAAAAAAAAAAA
A <sub>20</sub>	AAAAAAAAAAAAAAAAAAAA
A <sub>30</sub>	AAAAAAAAAAAAAAAAAAAAAAAAAAAA
A <sub>40</sub>	AAAAAAAAAAAAAAAAAAAAAAAAAAAAAAAAAAAA
Random DNA	TACCCTGTAGAACCGAATTTGTG
Random FAM-DNA	TACCCTGTAGAACCGAATTTGTG, FAM at the 5' terminus

**Table S2** Comparison of the apparent Michaelis-Menten constant ( $K_M$ ) and maximum reaction rate ( $V_{\max}$ ) of the catalytic reactions.

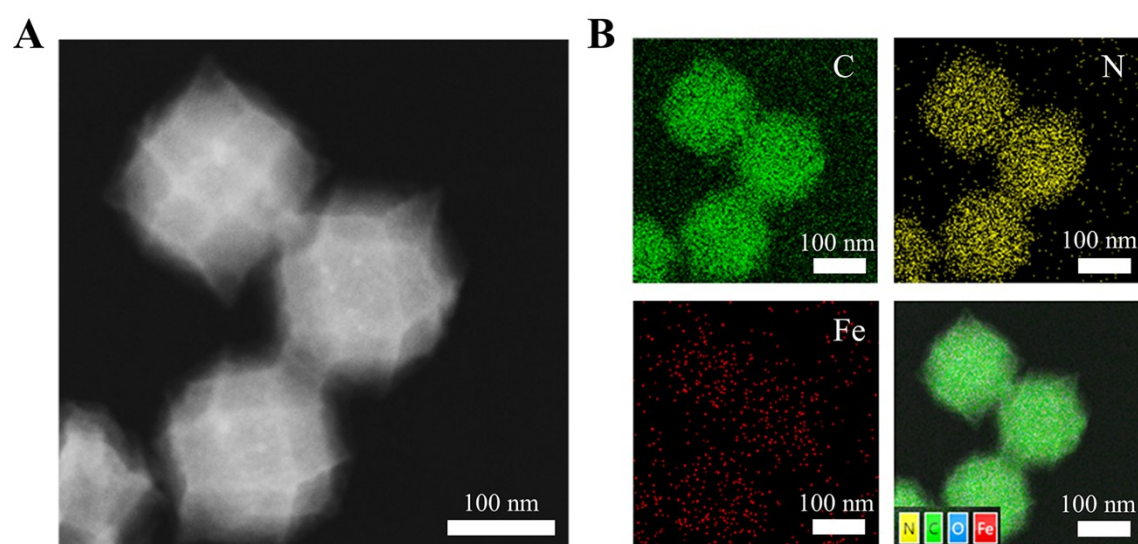
Catalyst	Substrate	$K_M$ (mM)	$V_{\max}$ ( $10^{-8}$ M s $^{-1}$ )	References
Fe-N-C	TMB	0.15	8.20	This work
	H <sub>2</sub> O <sub>2</sub>	1.60	7.61	
DNA/Fe-N-C	TMB	0.08	7.75	This work
	H <sub>2</sub> O <sub>2</sub>	3.03	11.69	
HRP	TMB	0.43	10.00	1
	H <sub>2</sub> O <sub>2</sub>	3.70	8.70	

**Table S3** Comparison of different methods for DA detection.

Methods	Materials	Linear range ( $\mu\text{M}$ )	LOD (nM)	References
Fluorescence	GNC-Ce <sup>3+</sup>	0.1-1000	10.85	2
Fluorescence	BSA-CuNCs	0.50-50	280	3
ECL	CdSeTe/ZnS QDs	3.75-450	100	4
Electrochemistry	CAuNE	1-100	5830	5
Colorimetric	CuFe <sub>2</sub> O <sub>4</sub> /Cu <sub>9</sub> S <sub>8</sub> / PPy h-CuS	2-20	1000	6
Colorimetric	Bi <sub>2</sub> Fe <sub>4</sub> O <sub>9</sub>	0.15-50	50	7
Colorimetric	Pt/hBNNSs-5	2-55	760	8
Colorimetric	h-CuS NCs	2-150	1670	9
Colorimetric	CGNPs	1-20	1150	10
Colorimetric	FeN <sub>3</sub> -PtN <sub>4</sub> -SAzyme	1-10	109	11
Colorimetric	Fe-N-C	5-100	4170	This work
Colorimetric	DNA/Fe-N-C	0.01-4 5-100	9.56	This work



**Fig. S1** SEM image of Fe-N-C.



**Fig. S2** (A) HAADF-STEM image of Fe-N-C. (B) the elemental mapping images of C (green), N (yellow), Fe (red) and their overlapping image.

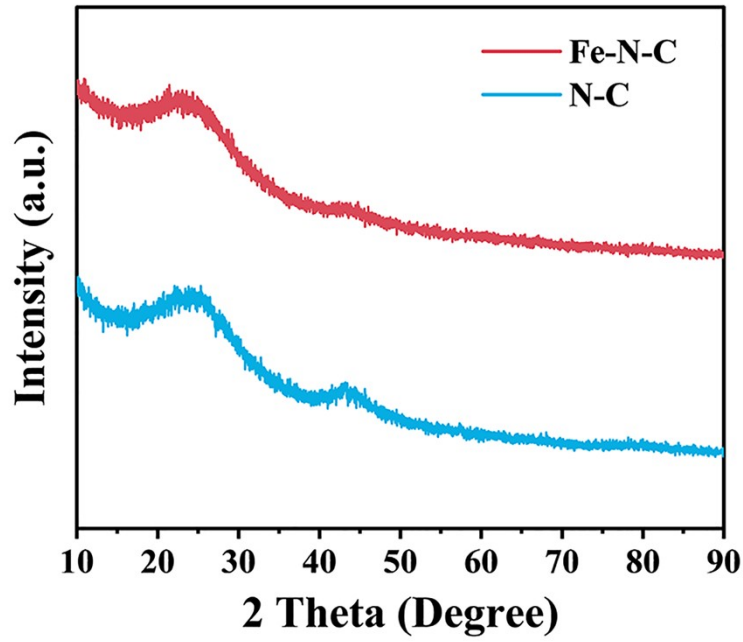


Fig. S3 XRD pattern of N-C and Fe-N-C.

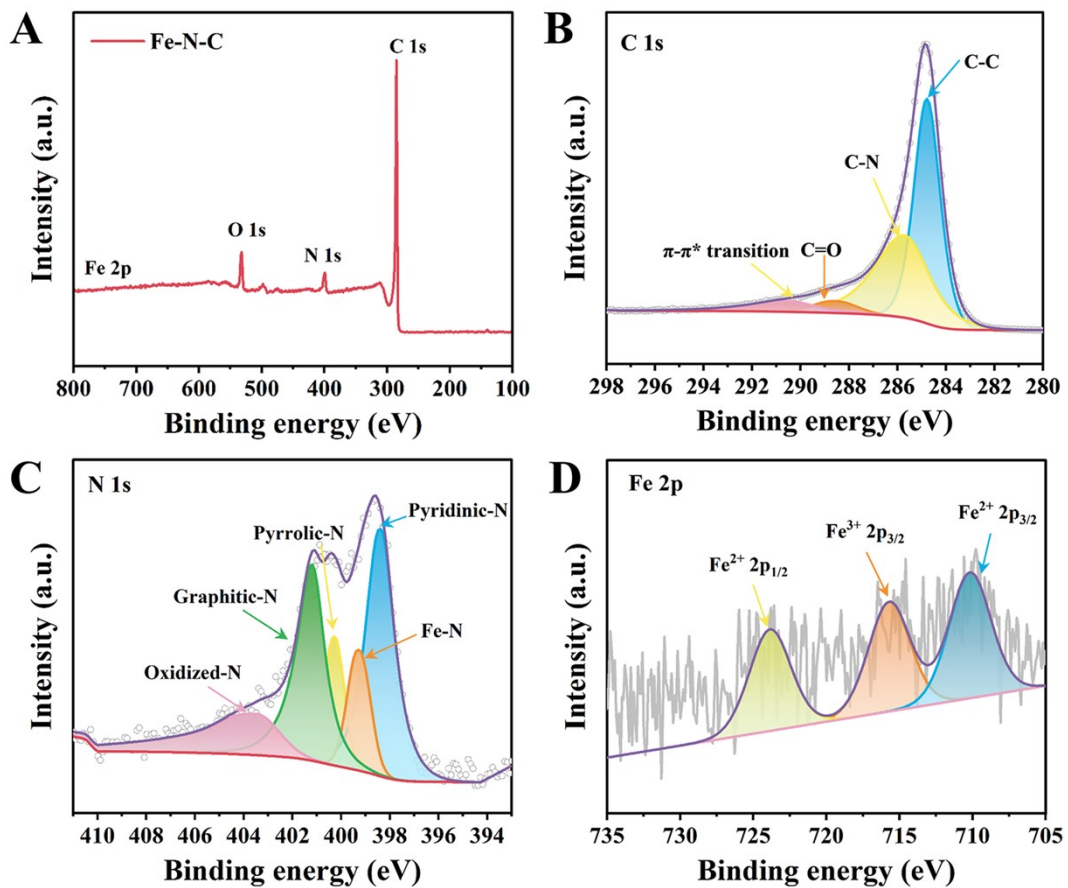
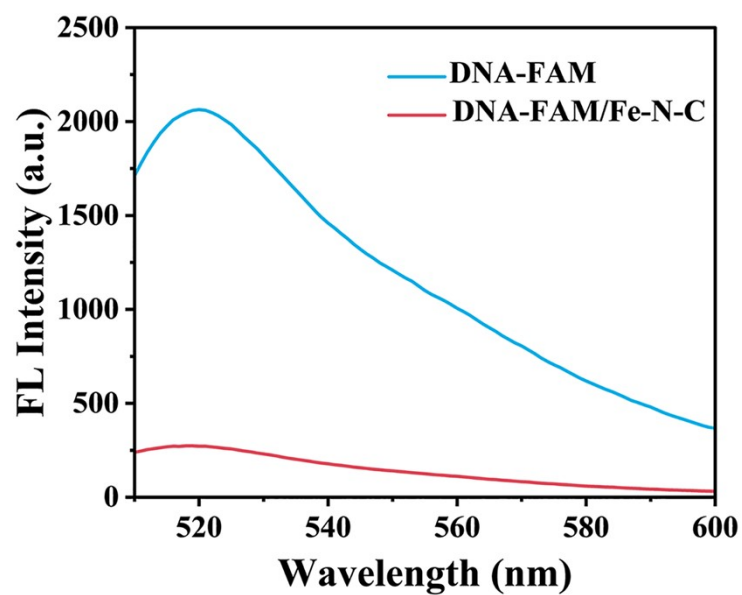
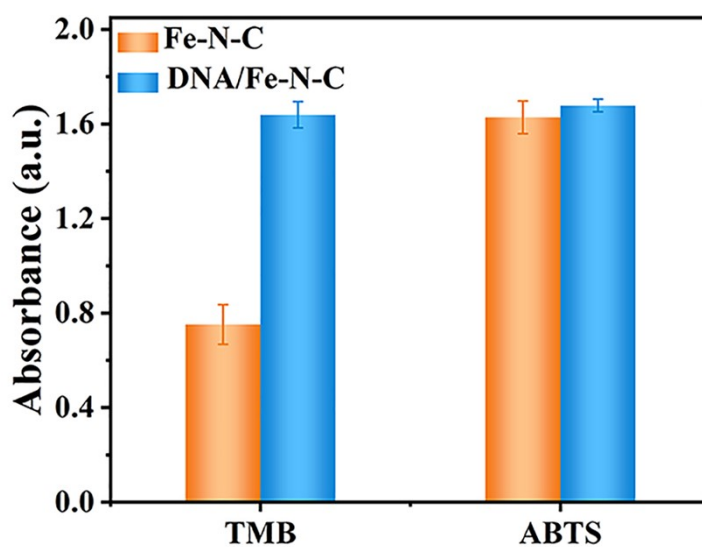


Fig. S4 (A) XPS survey spectrum of Fe-N-C. (B) High-resolution XPS spectra of C 1s, (C) N 1s, and (D) Fe 2p.

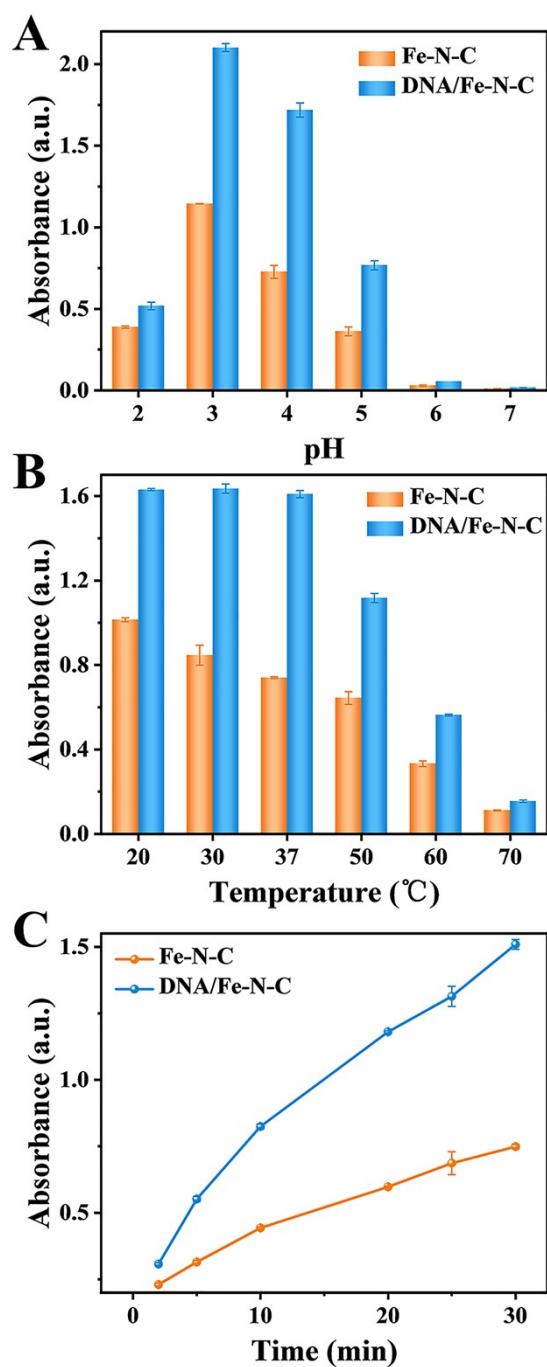




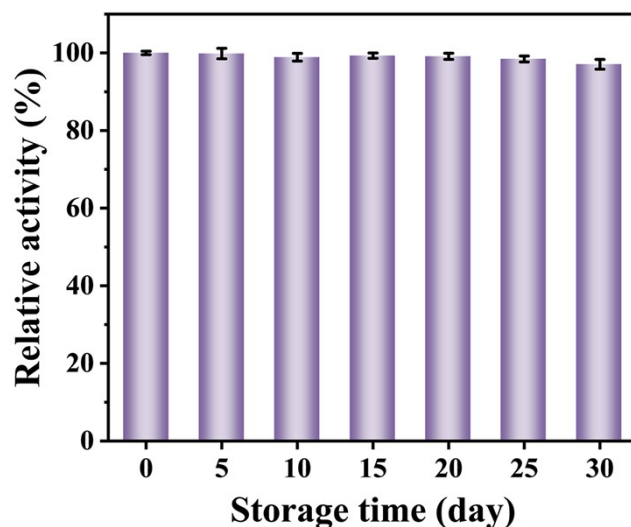
**Fig. S5** Fluorescence spectra of FAM-DNA responding to  $5 \mu\text{g mL}^{-1}$  Fe-N-C.



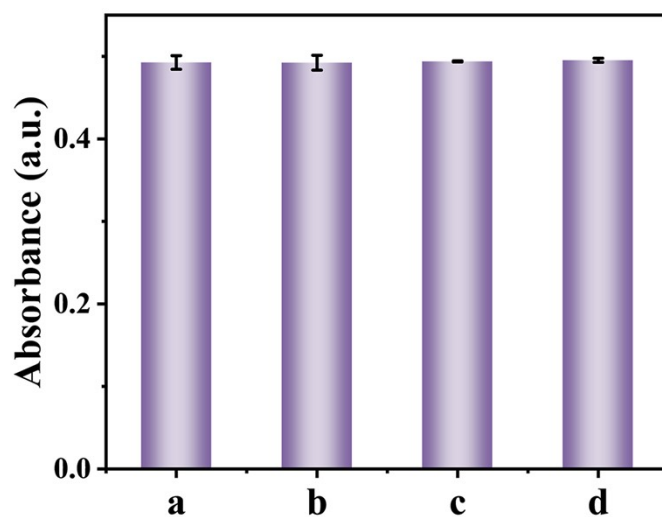
**Fig. S6** POD-like activity of Fe-N-C using ABTS, and TMB as peroxidase substrates before and after adding DNA. Absorbance was determined at 652 nm. The error bars are standard deviations obtained from three independent measurements.



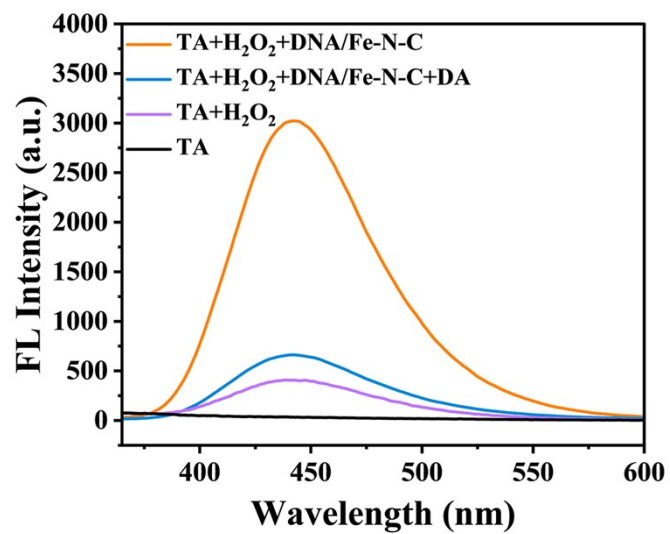
**Fig. S7** The absorbance variation of Fe-N-C and DNA/Fe-N-C depending on pH (A) and temperature (B) and time (C). Absorbance was determined at 652 nm. The error bars are standard deviations obtained from three independent measurements.



**Fig. S8** Stability tests of DNA/Fe-N-C for 30 days. Condition:  $0.5 \mu\text{g mL}^{-1}$  Fe-N-C, 10 nM DNA, 0.4 mM TMB, 4 mM  $\text{H}_2\text{O}_2$ , HAc-NaAc buffer (20 mM, pH 4.0), a reaction time of 20 minutes and a reaction temperature of  $37^\circ\text{C}$ . The error bars are standard deviations obtained from three independent measurements.



**Fig. S9** The absorbance of DNA/Fe-N-C in different conditions of (a) pH=7; (b) pH=4; (c) pH=7,  $\text{H}_2\text{O}_2$ ; (d) pH=4,  $\text{H}_2\text{O}_2$ . Absorbance was determined at 260 nm. The error bars are standard deviations obtained from three independent measurements.



**Fig. S10** Fluorescence spectra were measured under excitation at 315 nm for the interaction of TA with DNA/Fe-N-C, DNA/Fe-N-C+DA, and H<sub>2</sub>O<sub>2</sub>, respectively.

## References

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